

“Complementi di Fisica”

Lecture 16

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In this lecture

- **Contents**
 - Introduction to the semi-classical approximation:
 - orders of magnitude for “internal” and “external” electric fields in crystals;
 - use of “effective mass” in a classical example
 - Semi-classical equations of motion for electrons in crystals (“Bloch wave-packets”)
 - Effective mass of electrons
 - Examples of electrons confined in energy bands and comparison with a free electron: “Bloch oscillations”
- **Reference textbooks**
 - D.A. Neamen, **Semiconductor Physics and Devices**, McGraw-Hill, 3rd ed., 2003, p.71-78
 - Ch.Kittel, **Introduction to Solid State Physics**, J.Wiley & Sons, 7th ed., p. 203-212 (“equations of motion, effective mass”)
 - W.Ibach & H.Luth, **Solid-State Physics**, Springer, 3rd ed., p. 231-235 (“9.1: Motion of Electrons in Bands and the Effective Mass”)

Electric fields in materials

- “Internal”, microscopic (atomic level) $q = 1.6 \times 10^{-19} C$, $r \approx 10^{-10} m$

$$E_x = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2} \approx 9 \times 10^9 \frac{Vm}{C} \times \frac{1.6 \times 10^{-19}}{(10^{-10})^2} \frac{C}{m^2} \approx 1.4 \times 10^{11} V/m$$

- “External”, macroscopic (for instance, the potential difference set by an external electric circuit over a junction)

$$|E_x| = \frac{dV}{dx} \approx \frac{1}{10^{-4}} \frac{V}{m} \approx 10^4 V/m$$

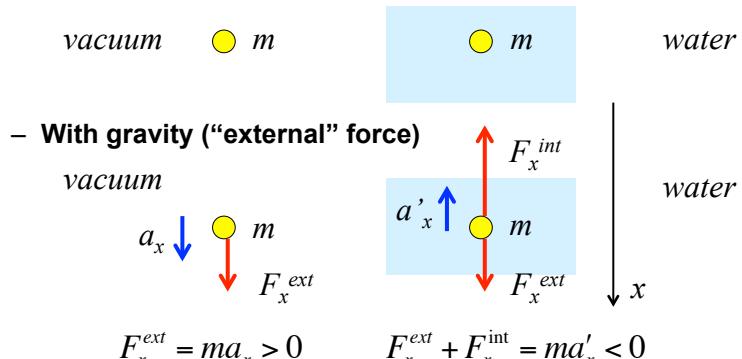
- Different by many orders of magnitude: can be treated separately!

– **Microscopic**: quantum mechanical description, periodic potential

– **Macroscopic**: semi-classical description, in terms of (group) velocity, acceleration, inertia (effective mass, that may be negative!!!)

“Effective mass”: classical example

- Tennis ball in a vacuum, or immersed in a fluid (water)
 - without gravity (“external” force)



$$F_x^{ext} = ma_x > 0 \quad F_x^{ext} + F_x^{int} = ma'_x < 0$$

- “effective mass” (consider only the external force) $m^* = F_x^{ext}/a_x$

$$m^* = m > 0 \quad m^* = \frac{F_x^{ext}}{a'_x} < 0 \quad (m^* \neq m)$$

“Effective mass”: classical example

- In this example:
the “effective mass”, defined as $m^* = F_x^{ext}/a_x$
is negative in water, because:
Gravity (“external force”) pulls the ball downwards, but:
Archimede’s force (“internal”, interaction with the fluid),
pushes upwards;
The net force and the acceleration are directed upwards
- Similar effects may be expected for electrons in crystals, if we consider separately
 - “external” forces (macroscopic electric or magnetic fields)
 - “internal” interactions with the periodic crystal fields

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Semi-classical equation of motion

- Electron in a crystal with an external field:
 - The “quantum” interaction with the crystal periodic potential:
 - Separable wave functions and energy E eigenvalues
 - Dispersion relation $E(k)$ with k “crystal wave number”
 - Bloch wave packets, propagating with group velocity dE/dk
 - The “semi-classical” interaction with the external field changes:
 - the electron energy E and wave number k (quantum description)
 - the group velocity (acceleration: classical description)
 - Example in real space, electric field $E_x > 0$

$$F_x^{ext} = -|q|E_x \quad \begin{array}{c} \xleftarrow{\hspace{-1cm} F_x^{ext}} \text{---} |q| \\ \xrightarrow{\hspace{-1cm} E_x } \end{array}$$

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In other words:

- An electron, partially localised in a crystal:

- is described by a Bloch wave packet:

$$\Psi(x, t) = \int_{k_0-\Delta k}^{k_0+\Delta k} a(k) u_k(x) e^{i\left(kx - \frac{E(k)}{\hbar}t\right)} dk \quad v_g = \frac{1}{\hbar} \frac{dE}{dk}$$

- The net force on the electron has 2 contributions:

- “internal” interaction with the crystal, described by dE/dk
 - “external” interaction with a macroscopic field, that determines changes in E , k , and in v_g

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Interaction with the external field

- Work theorem: variation of (kinetic ...) energy E

$$dx = v_g dt = \frac{1}{\hbar} \frac{dE}{dk} dt \quad \delta E = \frac{dE}{dk} dk \quad \Rightarrow \quad F_x^{\text{ext}} \frac{1}{\hbar} \frac{dE}{dk} dt = \frac{dE}{dk} dk$$

$$\Rightarrow \quad \hbar \frac{dk}{dt} = F_x^{\text{ext}}$$

- Kinematics: acceleration as derivative of group velocity

$$a_x = \frac{dv_g}{dt} = \frac{d}{dt} \left(\frac{1}{\hbar} \frac{dE}{dk} \right) = \frac{1}{\hbar} \frac{d}{dk} \left(\frac{dE}{dk} \right) \frac{dk}{dt} = \frac{1}{\hbar^2} \frac{d^2 E}{dk^2} \hbar \frac{dk}{dt} = \frac{1}{\hbar^2} \frac{d^2 E}{dk^2} F_x^{\text{ext}}$$

$$\Rightarrow \quad m^* = \frac{F_x^{\text{ext}}}{a_x} = \hbar^2 \left/ \left(\frac{d^2 E}{dk^2} \right) \right. \quad \text{“effective mass”}$$

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Semi-classical equation of motion

- Summary: we should remember, for an electron in a crystal interacting also with an external field:

Semi-classical equation of motion: time-evolution of k

$$\hbar \dot{k}_x \equiv \hbar \frac{dk_x}{dt} = F_x^{ext}$$

$$m^* = \hbar^2 \left/ \left(\frac{d^2 E}{dk_x^2} \right) \right.$$

similar to $\frac{dp_x}{dt} = F_x^{ext}$

Effective mass (may be negative!)

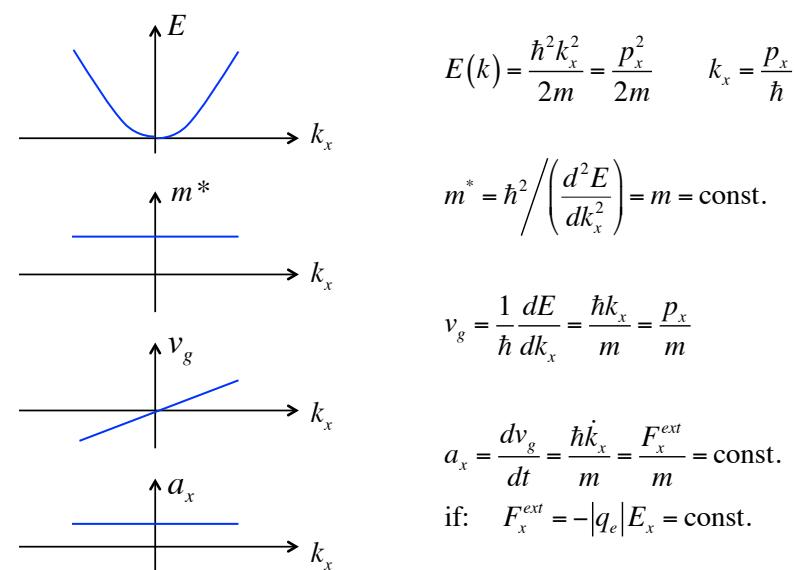
Exercise: application to three different dispersion relations $E(k)$:
“free particle”, “valence band”, “conduction band”

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Free electron in uniform electric field

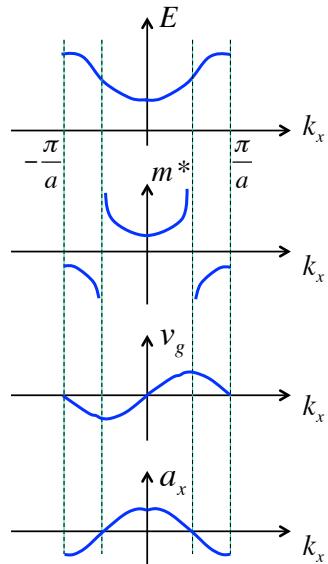


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Electron in “conduction band”



$$E(k)$$

$$m^* = \hbar^2 / \left(\frac{d^2 E}{dk_x^2} \right)$$

$$v_g = \frac{1}{\hbar} \frac{dE}{dk_x}$$

$$a_x = \frac{dv_g}{dt} = \frac{F_x^{ext}}{m^*}$$

even if: $F_x^{ext} = -|q_e|E_x = \text{const.}$

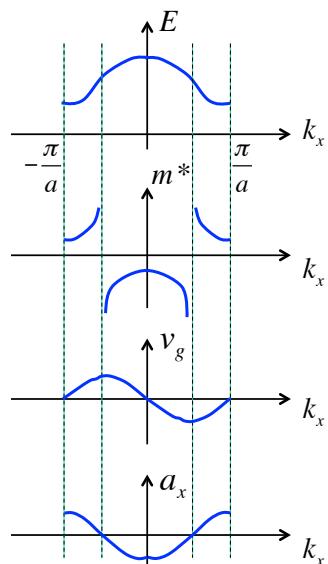
a_x is NOT constant!

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Electron in “valence band”



$$E(k)$$

$$m^* = \hbar^2 / \left(\frac{d^2 E}{dk_x^2} \right)$$

$$v_g = \frac{1}{\hbar} \frac{dE}{dk_x}$$

$$a_x = \frac{dv_g}{dt} = \frac{F_x^{ext}}{m^*}$$

even if: $F_x^{ext} = -|q_e|E_x = \text{const.}$

a_x is NOT constant!

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Consequences

- “Bottom” of conduction band
 - Positive effective mass
 - Acceleration: follows the external force, as usual
 - The electron energy increases
- “Top” of valence band
 - Negative effective mass!
 - Acceleration *contrary* to the external force!
 - The electron energy *decreases*!
 - Who takes the energy? The crystal...
- “Bloch oscillations” in a band in an ideal crystal
 - Experimentally observed in a periodic “potential”