from Lect.VII

- correlations
- (small or large) fluctuations

Correlations

[a very simple calculation if the array points(n) is stored]

```
do j = 0, jmax
si=0._dp ; si2=0._dp ; sij=0._dp
  do i = 1, n-j
    si = si + points(i)
    si2 = si2 + points(i)**2
    sij = sij + points(i)*points(i+j)
  end do
    si = si/(n-j)
    si2 = si2/(n-j)
    sij = sij/(n-j)
```

```
C(j) = \frac{\langle x_{i+j}x_i \rangle - \langle x_i \rangle^2}{\langle x_i^2 \rangle - \langle x_i \rangle^2}
```

write(2,*), j, (sij-si**2)/(si2-si**2)

end do

Correlations - Box-Muller algorithm



Correlations - Metropolis algorithm



Correlations - Metropolis algorithm



Correlations - Metropolis algorithm



Metropolis algorithm

by construction, Metropolis algorithm has **short-range correlations**. But it is very general!

and particularly useful for applications in the

canonical ensemble : fix N,V,T (system in equilibrium with a thermal bath)

Metropolis algorithm in the canonical ensemble

Microstates & macrostates

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The canonical ensemble

(N,V,T) fixed. The probability P_s for a system to be in the microstate *s* depends only on the energy E_s of the microstate and the temperature *T* (common to the system and the thermal bath):

$$P_s = \frac{1}{Z} e^{-\beta E_s}, \text{(canonical distribution)}$$

where $\beta = 1/kT$, and Z is a normalization constant.

(in the canonical ensemble the energy is a characteristic of the **microstate**, **not** of the **macrostate**) The higher is the energy of the state, the lower is the probability of finding the system in that state.

Because $\sum P_s = 1$, $Z = \sum_{s=1}^{M} e^{-E_s/kT}$

(canonical partition function)

(M: all accessible microstates of the system, having the same N, V, T)

Averages in the canonical ensemble

The ensemble average of a physical quantity A can be calculated using the canonical distribution function:

$$\langle A \rangle = \sum_{s=1}^{M} A_s P_s = \frac{1}{Z} \sum_{s=1}^{M} A_s e^{-\beta E_s} \qquad (M: total poss)$$

(M: total number of possible microstates)

In practice, approximating the sums over a finite number m of the total number M of accessible microstates:

$$\langle A \rangle \approx A_m = \frac{\sum_{s=1}^m A_s e^{-\beta E_s}}{\sum_{s=1}^m e^{-\beta E_s}}$$

(Note: *m*, not *M* !)

Averages in the canonical ensemble Due to the functional form of $e^{-\beta E_s}$, this formulation is highly inefficient (many microstates have very small probability):

$$\langle A \rangle \approx A_m = \frac{\sum_{s=1}^m A_s e^{-\beta E_s}}{\sum_{s=1}^m e^{-\beta E_s}}$$

=> much better to use the **importance sampling,** i.e. simple averages of the desired physical quantity over microstates *s* generated according to the proper distribution: $1 \int_{1}^{m}$

$$A_m = \frac{1}{m} \sum_{s=1}^m A_s$$

Averages in the canonical ensemble

How to generate microstates properly distributed ? => generate a Markov chain with the transition matrix:

$$T_{s,trial} = min\left[1, \frac{P_{trial}}{P_s}\right] = min\left[1, \frac{e^{-\beta E_{trial}}}{e^{-\beta E_s}}\right] = min\left[1, e^{-\beta\Delta E}\right]$$

where
$$\Delta E = E_{trial} - E_s$$

Peculiarities w.r.t. the application of the Metropolis algorithm already discussed:

 $e^{-\beta E_s}$ is a monotonic decreasing function of $E_{s,s}$ therefore we accept all the moves decreasing the energy.

Metropolis algorithm in the canonical ensemble

- 1. Establish an initial microstate.
- 2. Make a random trial change in the microstate. For example, choose a spin at random and flip it. Or choose a particle at random and displace it a random distance.
- 3. Compute $\Delta E \equiv E_{\text{trial}} E_{\text{old}}$, the change in the energy of the system due to the trial change.
- 4. If ΔE is less than or equal to zero, accept the new microstate and go to step 8.
- 5. If ΔE is positive, compute the quantity $w = e^{-\beta \Delta E}$.
- 6. Generate a random number r in the unit interval.
- 7. If $r \leq w$, accept the new microstate; otherwise retain the previous microstate.
- 8. Determine the value of the desired physical quantities.
- 9. Repeat steps (2) through (8) to obtain a sufficient number of microstates.
- 10. Periodically compute averages over microstates.

Metropolis algorithm in the canonical ensemble

A few remarks:

1) Because it is necessary to evaluate only the ratio $P_j/P_i = e^{-\beta\Delta E}$, it is not necessary to normalize the probability.

 $(P_j/P_i = \pi_j/\pi_i)$

2) ERGODICITY implicitly assumed!

3) TEMPERATURE:

If $E_B > E_A$, accept the new (higher energy) configuration with probability $p = e^{-(E_B - E_A)/T}$. This means that when the temperature is high, we don't mind taking steps in the "wrong" direction, but as the temperature is lowered, we are forced to settle into the lowest configuration we can find in our neighborhood.

Metropolis algorithm in the canonical ensemble

Don't confuse P_s with $P(E_s)$, since $\Omega(E_s)$ microstates with the same energy E_s can exist, in general (*unless in particular cases where* E_s *uniquely identifies the microstate*). Therefore:

$$P(E_s) = \frac{\Omega(E_s)}{Z} e^{-\beta E_s}$$

where the **degeneracy of the energy level is accounted for,** whereas:

$$P_s = \frac{1}{Z} e^{-\beta E_s}$$

4)

Some programs:

on **\$/home/peressi/comp-phys/VIII-canonical/** [do: \$cp /home/peressi/.../VIII-canonical/* .]

boltzmann_metropolis.f90

and also (see later): box.f90 entropy.f90 simulated_annealing.f90

Boltzmann distribution in the canonical ensemble

The Metropolis algorithm really produces microstates with the Boltzmann distribution: application to ideal classical ID gas (ex. n. 4 Lect.VII)

1 free particle: Energy: $E = \frac{1}{2}mv^2$ in this case, velocity or energy labels a microstate

(the energy with a factor of 2, due to +/- sign of v);

we generate different microstates by random variations of the velocity and we accept/reject with Metropolis

Important quantities are the probabilities: P(v)dv that the system has a velocity between v and v+dv or P(E)dE that the system has an energy between E and E+dE

ideal classical ID gas

A particle moving randomly has in each direction a distribution of the component of the velocity:

$$f(v_x) = \left(\frac{m}{2\pi k_B T}\right)^{1/2} e^{-mv_x^2/2k_B T}$$
(1)

$$\langle v_x^2 \rangle = \int_{-\infty}^{+\infty} v_x^2 f(v_x) dv_x = \frac{k_B T}{m}$$
(2)

In 1D:

$$f(v)2dv = P(E)dE$$

that gives: $P(E) = \frac{1}{(\pi k_B T)^{1/2}} \frac{1}{\sqrt{E}} e^{-E/k_B T}$

In 3D:

$$\boldsymbol{P}(E) = \frac{2}{\sqrt{\pi}} \frac{1}{(k_B T)^{3/2}} \sqrt{E} \exp\left(-\frac{E}{k_B T}\right)$$
(3D)

Boltzmann distribution in the canonical ensemble

#	<e0></e0>	:	.000000
#	<v0></v0>	:	.000000
#	dvmax	:	2.00000
#	deltaE	:	5.00000E-02
#	nbin	:	79
==>	boltzma	nn.	1K <==
#	nMCsteps	:	1000
#	<e></e>	:	.501263
#	<v></v>	:	7.456664E-02
#	accept.	:	.692000
#	sigma	:	.713780
==>	boltzma	nn.	10K <==
#	nMCsteps	:	10000
#	<e></e>	:	.507580
#	<v></v>	:	3.366172E-02
#	accept.	:	.707700
#	sigma	:	.726145
==>	boltzma	nn.	1M <==
#	nMCsteps	:	1000000
#	<e></e>	:	.500138
#	<v></v>	:	1.833840E-04
#	accept.	:	.693837
#	sigma	:	.707472

T

$$T = 1 \rightarrow \langle E \rangle (expected) = 0.5$$
 $(m = 1)$

 $\sigma/\sqrt{n} = 0.022$

(σ is the variance of the energy)

 $\sigma/\sqrt{n} = 0.007$

 $\sigma / \sqrt{n} = 0.0007$

NOTE:

- Accuracy of ~ 1% on <E> and 10% on <v> : NMCS=1000 is enough
- NOT ENOUGH to well reproduce the BOLTZMANN DISTRIBUTION! (1M needed!)
- ACCEPTANCE RATIO: constant, depends only on dvmax
- SIGMA also

Boltzmann distribution in the canonical ensemble

many particles: Energy: $E = \sum_{i=1}^{N} \frac{1}{2}m_i v_i^2$ in this case, the energy is NOT a label of a microstate (there are several microstates with the same total energy)

Note: the energy histogram is NOT the distribution of microstates!

$$P(E) = \sum_{\substack{\text{states } s \\ \text{with } E_s = E}} P_s \text{ with } P_s = \frac{1}{Z} e^{-\beta E_s}$$
$$P(E) \propto = e^{\frac{-(E - \langle E \rangle)^2}{2\sigma^2}} \text{ with } \langle E \rangle \text{ average over all the microstates}$$

What is P(E)? (exercise n. 4 Lec.VII - many particles)



Statistical averages and stochastic fluctuations

Averages



But in practice T is finite, and $\langle G \rangle_T$ oscillates(varying T): divide T into intervals $A, B, C \dots$ of length L and sum(block averages):

$$\langle G \rangle_T = \frac{L}{T} \sum_{I=A,B,C,\dots} \langle G \rangle^{(I)} \qquad (\sigma_s/\sqrt{s}) \qquad (\sigma_n/\sqrt{n})$$

$$\Delta \langle G \rangle_T = \left[\frac{L}{T} \sum_{I} \left(\langle (G^{(I)})^2 \rangle - (\langle G^{(I)} \rangle^2 \right) \right]^{1/2} \xrightarrow{\longrightarrow} \sim \frac{1}{\sqrt{T}}$$

<u>Note:</u> not always $\Delta < G >_T$ is a good indicator of the actual error! (remind "ergodicity")

Stochastic fluctuations

Fluctuations are always present, due to the nature of the system, also when evolving towards equilibrium.

A simple example: non-interacting classical particles in a box (gas diffusion)



A box is divided into two parts communicating through a small hole. One particle randomly can pass through the hole per unit time, from the left to the right or viceversa.

 $N_{\text{left}}(t)$: number of particles present at time t in the left side Given $N_{\text{left}}(0),$ what is $N_{\text{left}}(t)$?



Stochastic fluctuations



How to reduce fluctuations?

- more particles
- average over many simulation runs

What can we do with fluctuations?

Intrinsic energy fluctuations in the canonical ensemble - I

Remind:
$$\langle E \rangle = \frac{1}{Z} \sum_{s} E_{s} e^{-\beta E_{s}}$$
 and $Z = \sum_{s} e^{-\beta E_{s}}$, therefore: $\langle E \rangle = -\frac{\partial}{\partial \beta} \ln Z$
Consider the thermal capacity:
 $C_{v} = \frac{\partial \langle E \rangle}{\partial T} = \dots = -\frac{1}{kT^{2}} \frac{\partial \langle E \rangle}{\partial \beta}$
we have
 $\frac{\partial \langle E \rangle}{\partial \beta} = -\frac{1}{Z^{2}} \frac{\partial Z}{\partial \beta} \sum_{s} E_{s} e^{-\beta E_{s}} - \frac{1}{Z} \sum_{s} E_{s}^{2} e^{-\beta E_{s}} = \langle E \rangle^{2} - \langle E^{2} \rangle = \langle (\delta E)^{2} \rangle$
where $\delta E \equiv E - \langle E \rangle$
Result:
 $C_{v} = \frac{\langle (\delta E)^{2} \rangle}{k_{B}T^{2}}$

The thermal capacity (or specific heat if considered for each particle) is related to the intrinsic stochastic energy fluctuations

Intrinsic energy fluctuations in
the canonical ensemble - IISince: $C_v = \frac{\langle (\delta E)^2 \rangle}{k_B T^2}$

if *N* is the number of particles, we have:

$$\frac{\sqrt{\langle (\delta E)^2 \rangle}}{\langle E \rangle} = \frac{\sqrt{k_B T^2 C_v}}{\langle E \rangle} \propto \frac{\sqrt{N}}{N} \sim \frac{1}{\sqrt{N}}$$

i.e., the relative energy fluctuations reduce when N is large

(correct; in the thermodynamic limit: $E \rightarrow const.$, macro ~ micro)

Macrostates, microstates and entropy at equilibrium

Macroscopic systems towards equilibrium

Again the simple example of non-interacting classical

particles in a box (gas diffusion)



A box is divided into two parts communicating through a small hole. One particle randomly can pass through the hole per unit time, from the left to the right or viceversa.

 $N_{\rm left}(t)$: number of particles present at time t in the left side Given $N_{\rm left}(0),$ what is $N_{\rm left}(t)$?

microcanonical ensemble : fix E, N,V (isolated closed system)

Microcanonical ensemble -a fundamental postulate-

Given an isolated system in equilibrium, it will assume with equal probability each of its accessible microstates

(i.e., a system in equilibrium does not have any preference for any of its available microstates)

Example: suppose a macrostate defined by (N,V,E); if $\Omega(N,V,E)$ is the # of microstates with energy $E \div dE$, for the microstate s the probability of occurrence is:

$$P_s = \begin{cases} 1/\Omega, & \text{if } s \text{ is accessible} \\ 0, & \text{otherwise.} \end{cases}$$

Consequence: for a system at equilibrium, the thermodynamic state (macrostate) which could result from the **largest number of microstates** is also the **most probable macrostate** of the system.

Macroscopic systems towards equilibrium



Another version: particles blue/red in both sides

(interdiffusion of two gases):

per unit time, one from each side is picked at random and put in the other side: $N_{left}^{blue}(t) + N_{left}^{red}(t) = constant; N_{left}^{red}(t) = ?$

Approach to equilibrium



A box divided into left/right; total number N of non-interacting particles. macrostate: specified by the **number** of particles n on the left side; microstate: specified by the **specific list** of the n particles on the left side

Equilibrium and entropy

The macrostate is specified by the number of particles on the left side, *n*, and the corresponding microstates are:

number of microstates
$$= \frac{N!}{n!(N-n)!} = \binom{N}{n} = \binom{N}{N-n}$$

The total number of microstates for N=10 is $2^{10}=1024$

n	# of microstates	۳.)».». ه ار با ه ار با
0	_	0.00
	10	2.30
2	45	3.81
3	120	4.79
4	210	5.35
5	252	5.53
6	210	5.35
7	120	4.79
8	45	3.81
9	10	2.30
10		0.00

n=5 is the macrostate with the largest number of microstates!

- (n=5 is the most probable macrostate)
- the most "random"!
 Equilibrium =
 Maximum number of
 possible microstates =
 Maximum entropy

Equilibrium and entropy

We could simply generate microstates and count them...

but also some smart algorithm (more suitable for computer simulations) are possible....

Entropy: Coincidence method

(S.K. Ma, J. Stat. Phys. 26, 221 (1981))

Equilibrium = Maximum entropy = Maximum number of possible microstates

Too much effort to enumerate all of them!

Alternative procedure (good for computing):

A system evolving in time will duplicate a microstate, before or later... The longer it takes for duplication, the fewer are the microstates in the corresponding macrostate. Hence, the lower is the entropy. Idea: measure the ratio of the number of pairs of duplicated microstates to the total number of possible pairs; entropy is the log of the inverse ratio.

E.g.: suppose as in the previous slide N=10, and the **macrostate n=1**; consider 20 different **microstates** labelled with the "name" of the particle: **8 7 5 10 7 2 4 6 2 10 3 4 3 9 6 5 2 9 2 4** Possible pairs: 20*(20-1)/2=190. Here: 6 pairs for particle "2"; 1 pair with particle "10" etc etc... Sum all of them: get 15. Ratio = 15/190 , Entropy: $S \propto \log(190/15) \sim 2.5$ Remind the definition of entropy:

$$S = -k_B \sum_{s} P_s \ln P_s$$
 in the canonical ensemble

 $S = k_B \log \Omega$ in the microcanonical ensemble, where all the microstates corresponding to a macrostate have the same energy (Ω is the number of microstates) Metropolis method in the canonical ensemble and the simulated annealing

Metropolis and simulated annealing - I

- •Stochastic search for global minimum. Monte Carlo optimization.
- •The concept is based on the manner in which liquids freeze or metals recrystallize. Sufficiently high starting temperature and slow cooling are important to avoid freezing out in metastable states.

Metropolis and simulated annealing - II

usual Metropolis procedure in the canonical ensemble

•Thermodynamic system at temperature T, energy E.

•Perturb configuration (generate a new one).

- •Compute change in energy dE. If dE is negative the new configuration is accepted. If dE is positive it is accepted with a probability given by the Boltzmann factor : exp(-dE/kT).
- The process is repeated many times for good sampling of configuration space.
 - •then the temperature is slightly lowered and the entire procedure repeated, and so on, until a frozen state is achieved at T = 0.

Example

in **simulated annealing.f90:** minimization of

f(x)=(x+0.2)*x+cos(14.5*x-0.3)
considered as an energy function and
using a fictitious temperature



Function to be minimized: f(x); Starting point: x, fx=f(x)

Annealing schedule:	initial (high) temperature: annealing temperature reduction factor: number of steps per block: 'ad hoc' parameter for trial move:	temp tfactor (<1) nsteps scale
DO WHILE (temp >	1E-5) ! anneal cycle	
DO istep = 1, 1 CALL RANDOM_1	nsteps NUMBER(rand) ! generate 2 random nu	mbers; dimension(2)
$x_new = x + s$	<pre>scale*SQRT(temp)*(rand(1) - 0.</pre>	5) ! stochasti

```
:: rand
                                                            c move
  fx new = func(x new) ! new object function value
  IF (EXP(-(fx new - fx)/temp) > rand(2)) THEN ! success, save
    fx = fx new
    x = x new
  END IF
  IF (fx < fx min) THEN
    fx min = fx
    x \min = x
    PRINT '(3ES13.5)', temp, x_min, fx_min
  END IF
END DO
temp = temp * tfactor ! decrease temperature
```

```
END DO
```



initial T: 10 (K_B units)
initial x: 1.000000
initial f(x): 1.137208



final T: 2.50315E-01 final x: -1.95067E-01 final f(x):-1.00088E+00