

Introduction to Computational Systems Biology

Luca Bortolussi

DMG, Università di Trieste, IT

Modelling and Simulation Group, Saarland University, DE

The Holy Grail

Theory

Cell

A Whole-Cell Computational Model Predicts Phenotype from Genotype

Jonathan R. Karr,^{1,4} Jayodita C. Sanghvi,^{2,4} Derek N. Macklin,² Miriam V. Gutschow,² Jared M. Jacobs,² Benjamin Bolival, Jr.,² Nancyra Assad-Garcia,³ John I. Glass,³ and Markus W. Covert^{2,*}

¹Graduate Program in Biophysics

²Department of Bioengineering

Stanford University, Stanford, CA 94305, USA

³J. Craig Venter Institute, Rockville, MD 20850, USA

⁴These authors contributed equally to this work

*Correspondence: mcovert@stanford.edu

<http://dx.doi.org/10.1016/j.cell.2012.05.044>

Mycobacterium with 600 genes.
Scaling to Eucaryotes is highly non-trivial.

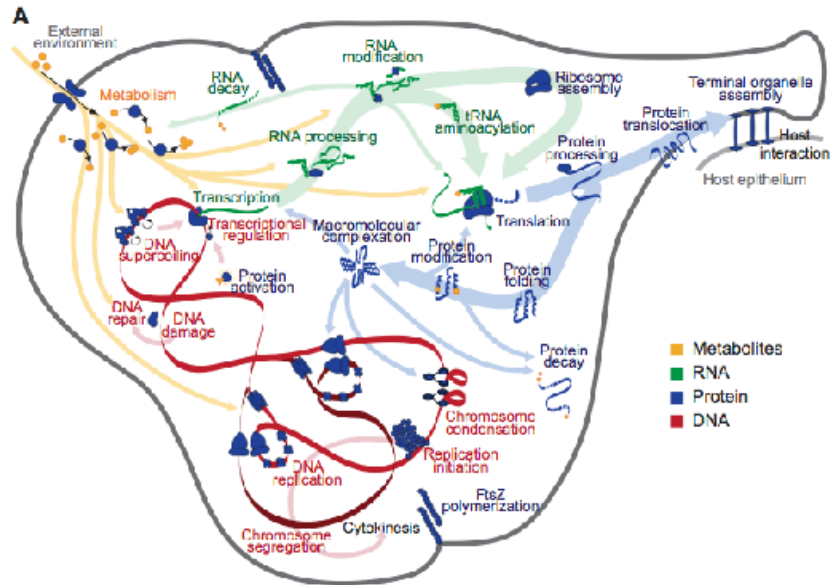
Biological systems

A cell is made of many subsystems, performing different tasks and interacting among them.

We have several *classes* of subsystems

- sensor networks
- signalling networks
- gene networks
- transport networks
- metabolic networks

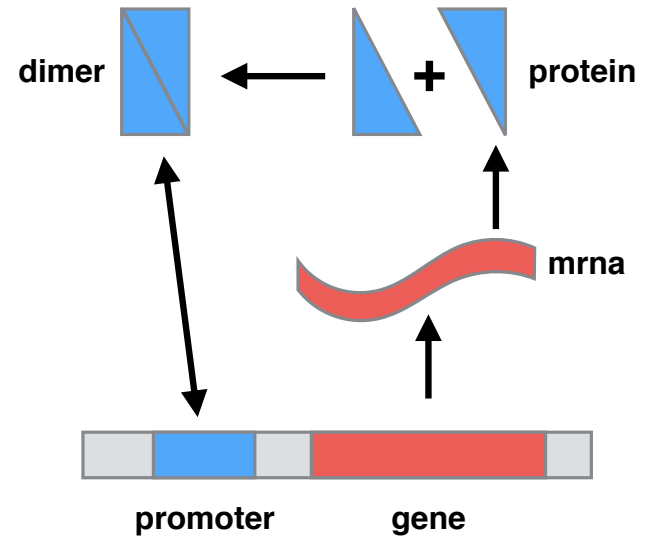
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Bio-chemical networks

Most biological systems can be described as a set of bio-chemical reactions, to be intended as a modelling language.

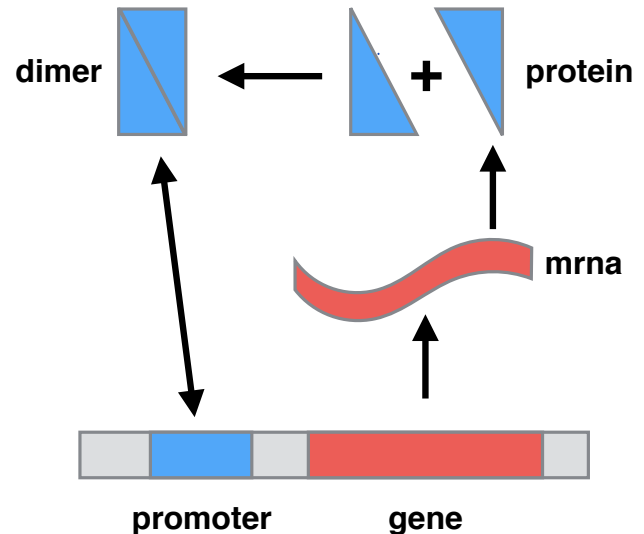
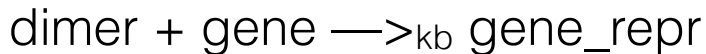
(warning: not suited for systems involving large protein complexes)



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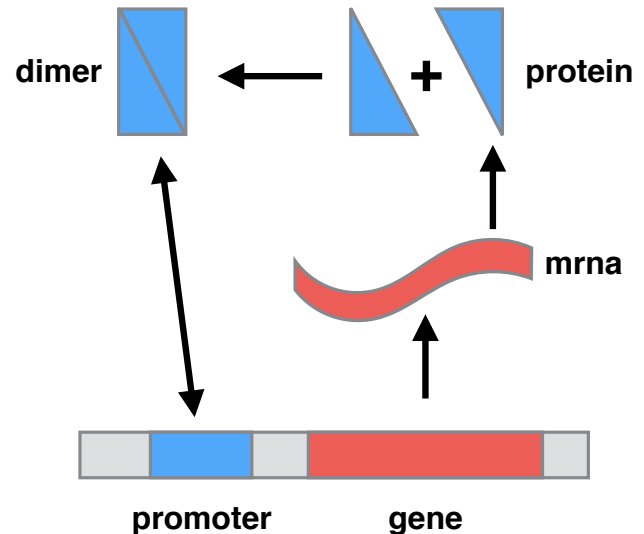
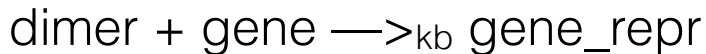
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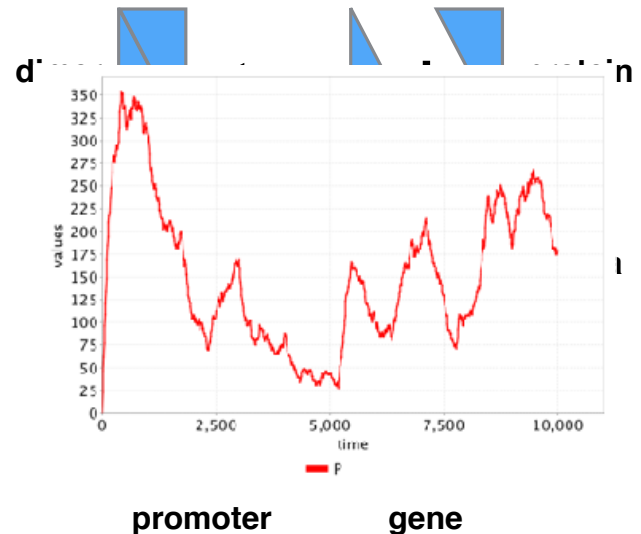
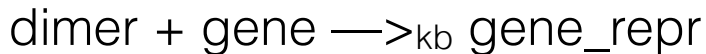


We are typically interested in the dynamic behaviour.

Kinetic constants are crucial for this, but are hard to measure or infer.

Bio-chemical networks

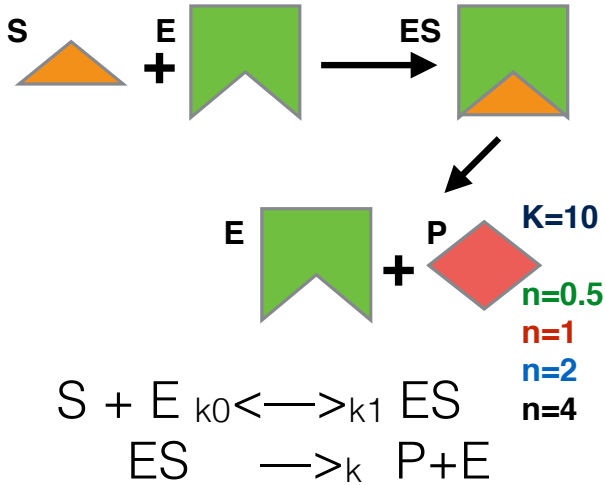
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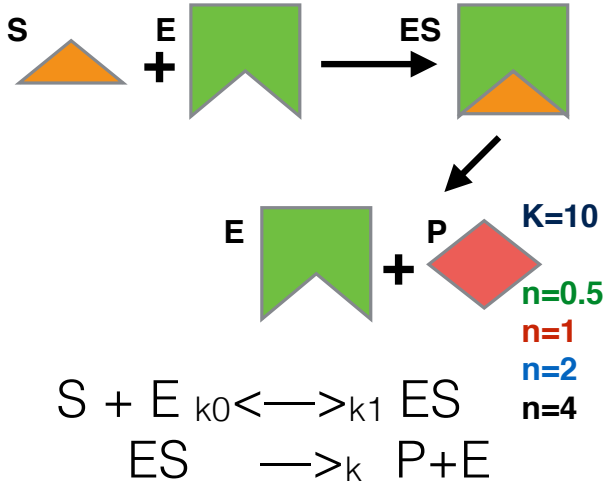
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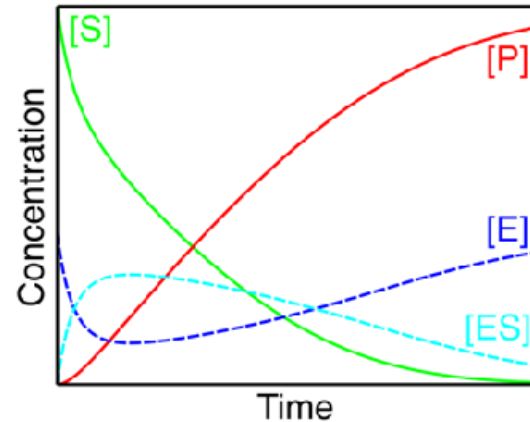
Dynamic Modelling



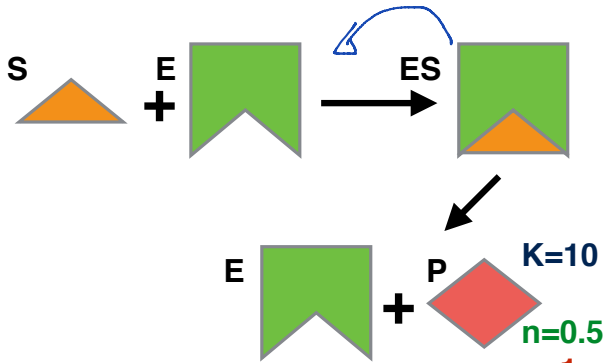
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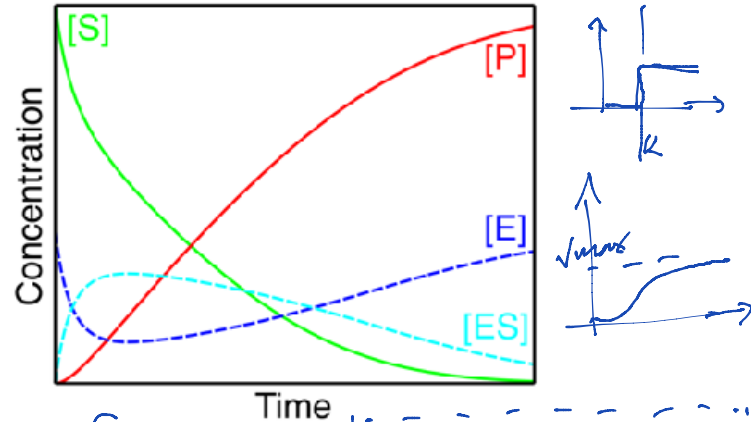
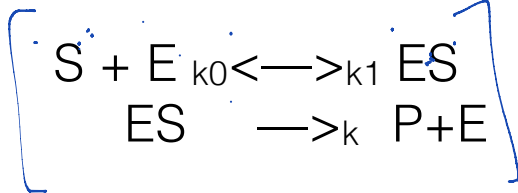
$$\begin{aligned}
 d[S]/dt &= -k_1[S][E] + k_0[ES] \\
 d[E]/dt &= -k_1[S][E] + k_0[ES] + k[ES] \\
 d[ES]/dt &= -k[ES] - k_0[ES] + k_1[S][E] \\
 d[P]/dt &= k[ES]
 \end{aligned}$$



Dynamic Modelling



$$\begin{cases} d[S]/dt = -k_1[S][E] + k_0[ES] \\ d[E]/dt = -k_1[S][E] + k_0[ES] + k[ES] \\ d[ES]/dt = -k_0[ES] + k_1[S][E] \\ d[P]/dt = k[ES] \end{cases}$$



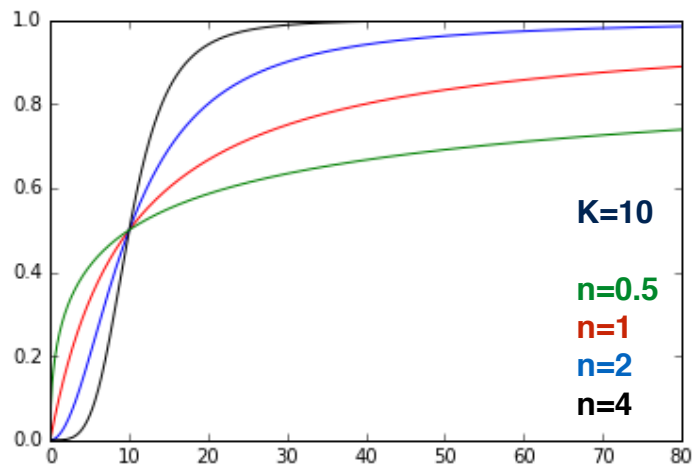
$$d[P]/dt = V_{\max} [S]/(K + [S])$$

Under time-scale separation, we can assume $d[ES]/dt = 0$, getting the classic Michaelis Menten kinetics:

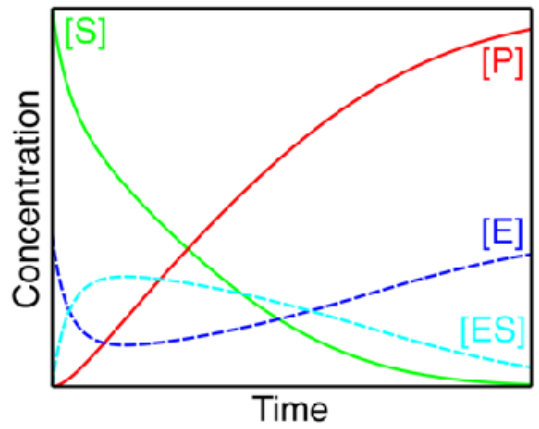
Cooperation/competition between enzyme and substrate results in the Hill kinetics:

$$d[P]/dt = V_{\max} [S]^n / (K^n + [S]^n)$$

Dynamic Modelling



$$\begin{aligned} d[S]/dt &= -k_1[S][E] + k_0[ES] \\ d[E]/dt &= -k_1[S][E] + k_0[ES] + k[ES] \\ d[ES]/dt &= -k[ES] - k_0[ES] + k_1[S][E] \\ d[P]/dt &= k[ES] \end{aligned}$$



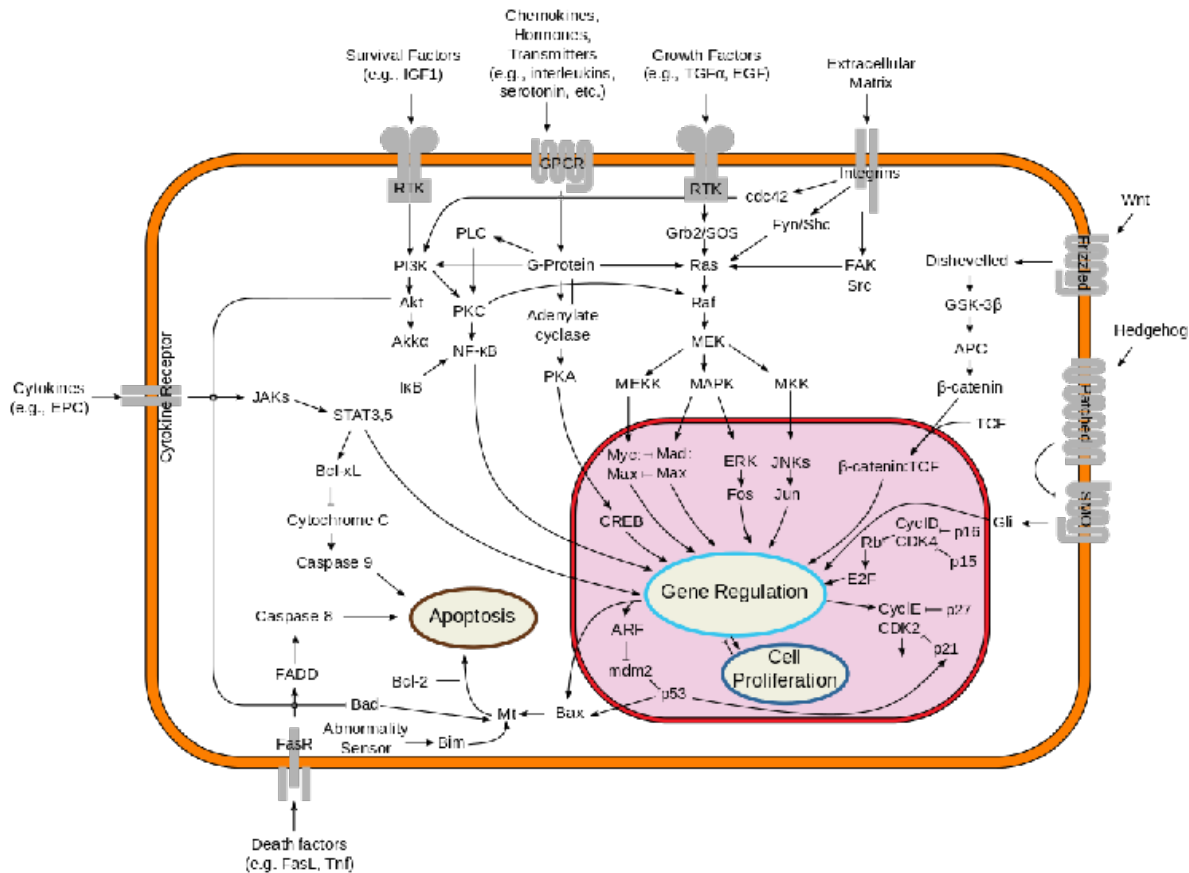
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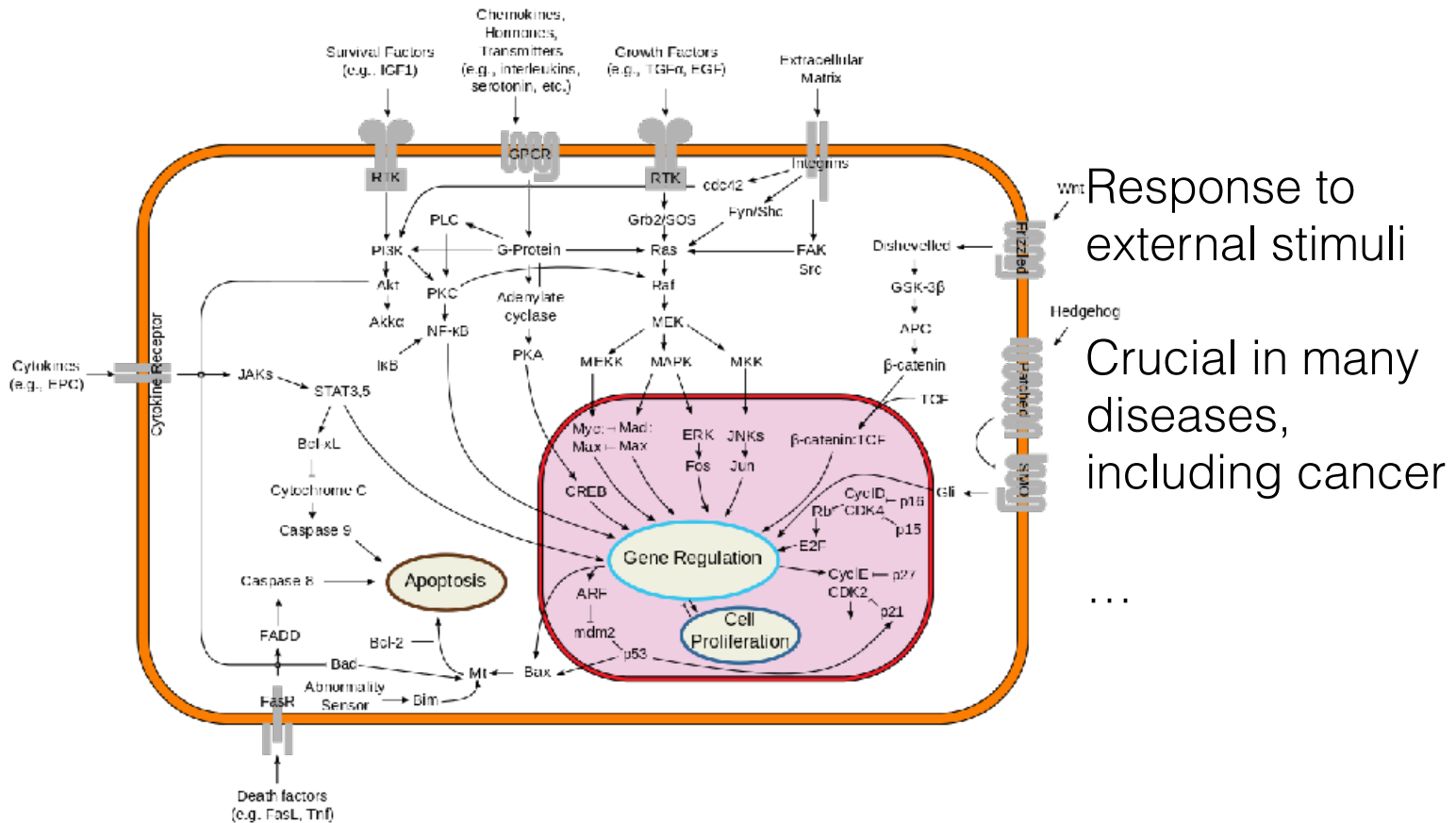
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Signal transduction networks



Signal transduction networks



Response to external stimuli

Crucial in many diseases, including cancer

...

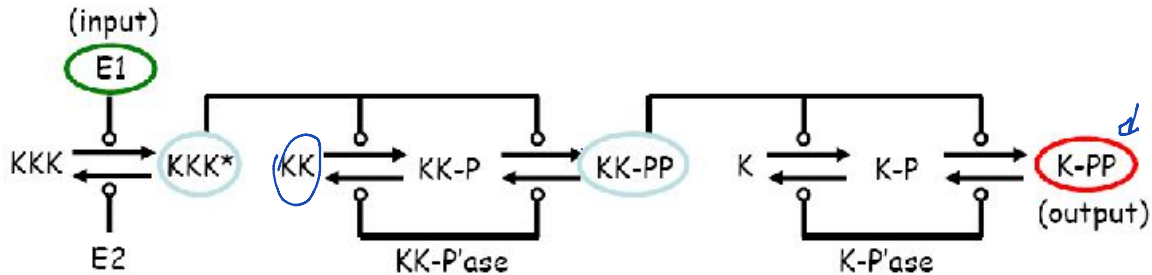
Signal transduction networks

Proc. Natl. Acad. Sci. USA
Vol. 93, pp. 10078-10083, September 1996
Biochemistry

MAPK

Ultrasensitivity in the mitogen-activated protein kinase cascade

CHI-YING F. HUANG AND JAMES E. FERRELL, JR.†

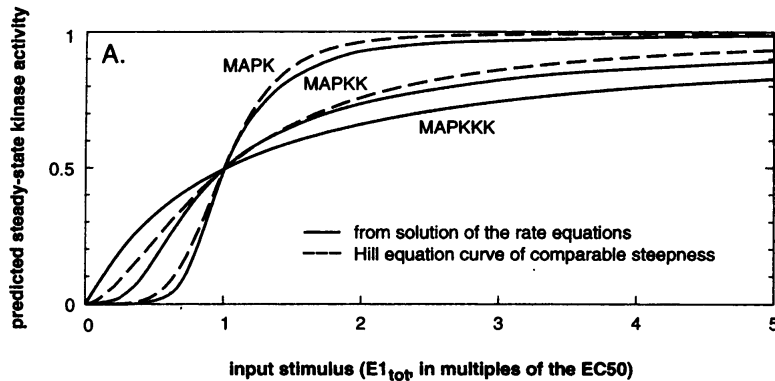
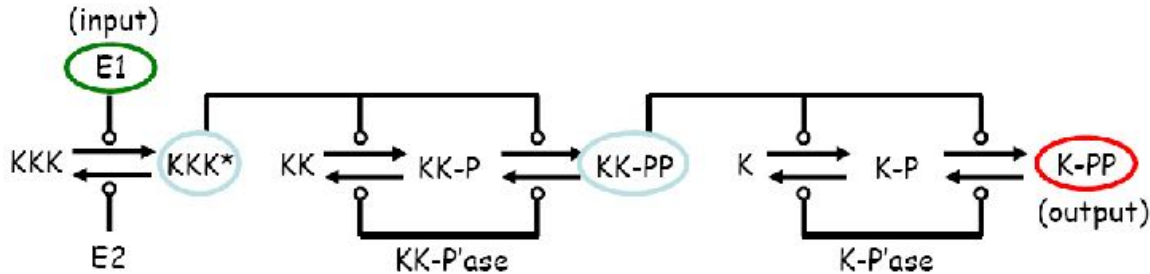


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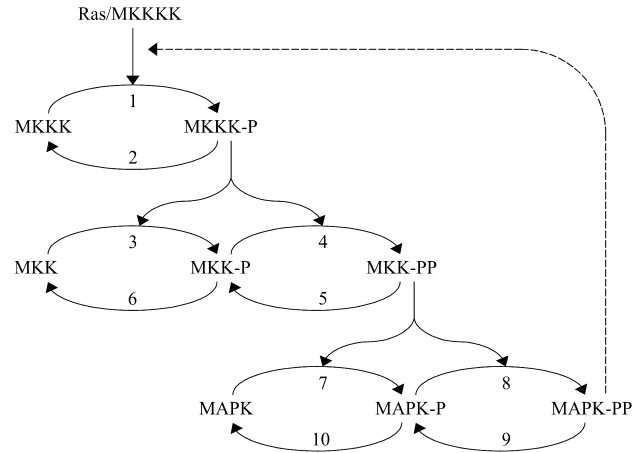
The activation cascade gives MAPK an ultra-sensitive response to variations in input

Signal transduction networks

Negative feedback and ultrasensitivity can bring about oscillations in the mitogen-activated protein kinase cascades

Boris N. Kholodenko

Department of Pathology, Anatomy and Cell Biology, Thomas Jefferson University, Philadelphia, PA, USA



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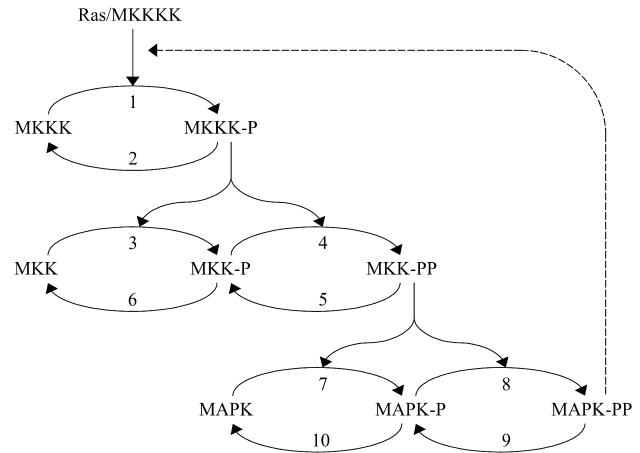


Table 1. Kinetic equations comprising the computational model of the MAPK cascade.

$$d[\text{MK4}]/dt = v_2 - v_1$$

$$d[\text{MK4-P}]/dt = v_1 - v_2$$

$$d[\text{MK3}]/dt = v_6 - v_3$$

$$d[\text{MK3-P}]/dt = v_3 + v_5 - v_4 - v_6$$

$$d[\text{MK3-PP}]/dt = v_4 - v_5$$

$$d[\text{MAPK}]/dt = v_{10} - v_7$$

$$d[\text{MAPK-P}]/dt = v_7 + v_9 - v_8 - v_{10}$$

$$d[\text{MAPK-PP}]/dt = v_8 - v_9$$

Moiety conservation relations:

$$[\text{MK4}]_{\text{total}} = [\text{MK4}] + [\text{MK4-P}]$$

$$[\text{MK3}]_{\text{total}} = [\text{MK3}] + [\text{MK3-P}] + [\text{MK3-PP}]$$

$$[\text{MAPK}]_{\text{total}} = [\text{MAPK}] + [\text{MAPK-P}] + [\text{MAPK-PP}]$$

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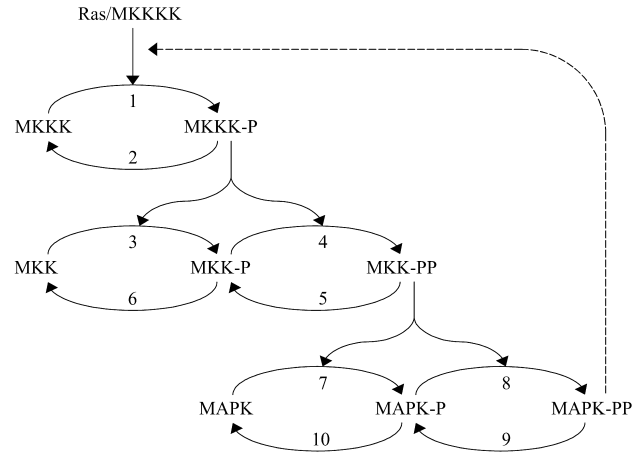
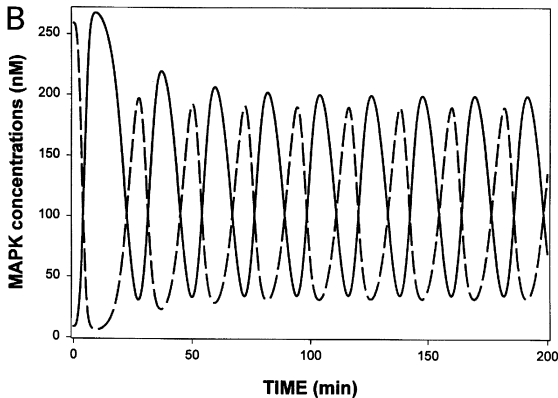
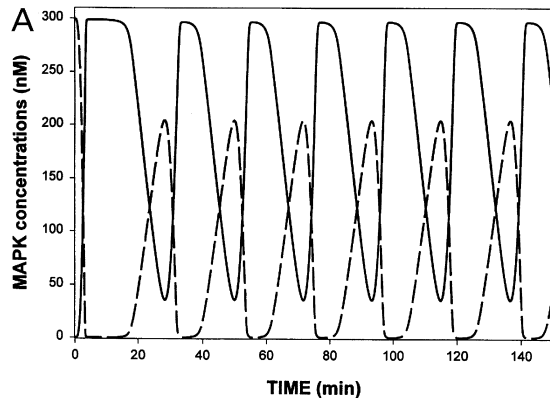


Table 1. Kinetic equations comprising the computational model of the MAPK cascade.

$$\begin{aligned}
 d[\text{MK444}]/dt &= v_2 - v_1 \\
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 d[\text{MK444}]/dt &= v_6 - v_3 \\
 d[\text{MK444-PP}]/dt &= v_3 + v_5 - v_4 - v_6 \\
 d[\text{MK444-PPP}]/dt &= v_4 - v_5 \\
 d[\text{MAPK}]/dt &= v_{10} - v_7 \\
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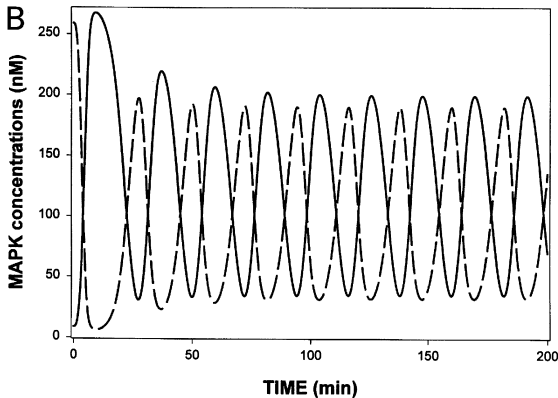
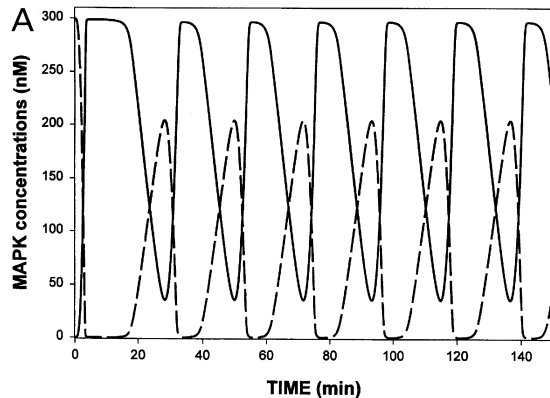
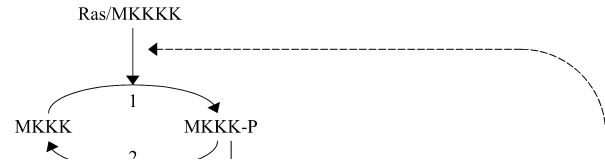
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 [\text{MK444}]_{\text{total}} &= [\text{MK444}] + [\text{MK444-P}] \\
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Reaction number	Rate equation	Parameter values
1	$V_1 \cdot [\text{MKKK}][I] / (K_1 + [\text{MAPK-PP}][K_1] + [\text{MKKK}][I])$	$V_1 = 2.5; a = 1; K_1 = 9; K_1 = 10;$
2	$V_2 \cdot [\text{MKKK-P}] / (K_2 + [\text{MKKK-P}])$	$V_2 = 0.25; K_2 = 8;$
3	$k_3 \cdot [\text{MKKK-P}] \cdot [\text{MKKK}] / (K_3 + [\text{MKKK}])$	$k_3 = 0.025; K_3 = 15;$
4	$k_4 \cdot [\text{MKKK-P}] \cdot [\text{MKKK-P}] / (K_4 + [\text{MKKK-P}])$	$k_4 = 0.025; K_4 = 15;$
5	$V_5 \cdot [\text{MKK-PP}] / (K_5 + [\text{MKK-PP}])$	$V_5 = 0.75; K_5 = 15;$
6	$V_6 \cdot [\text{MKK-P}] / (K_6 + [\text{MKK-P}])$	$V_6 = 0.75; K_6 = 15;$
7	$k_7 \cdot [\text{MKK-P}] \cdot [\text{MAPK}] / (K_7 + [\text{MAPK}])$	$k_7 = 0.025; K_7 = 15;$
8	$k_8 \cdot [\text{MKK-PP}] \cdot [\text{MAPK-P}] / (K_8 + [\text{MAPK-P}])$	$k_8 = 0.025; K_8 = 15;$
9	$V_9 \cdot [\text{MAPK-PP}] / (K_9 + [\text{MAPK-PP}])$	$V_9 = 0.5; K_9 = 15;$
10	$V_{10} \cdot [\text{MAPK-P}] / (K_{10} + [\text{MAPK-P}])$	$V_{10} = 0.5; K_{10} = 15;$
Total concentrations: $[\text{MKKK}]_{\text{total}} = 100; [\text{MKK}]_{\text{total}} = 300; [\text{MAPK}]_{\text{total}} = 300$		

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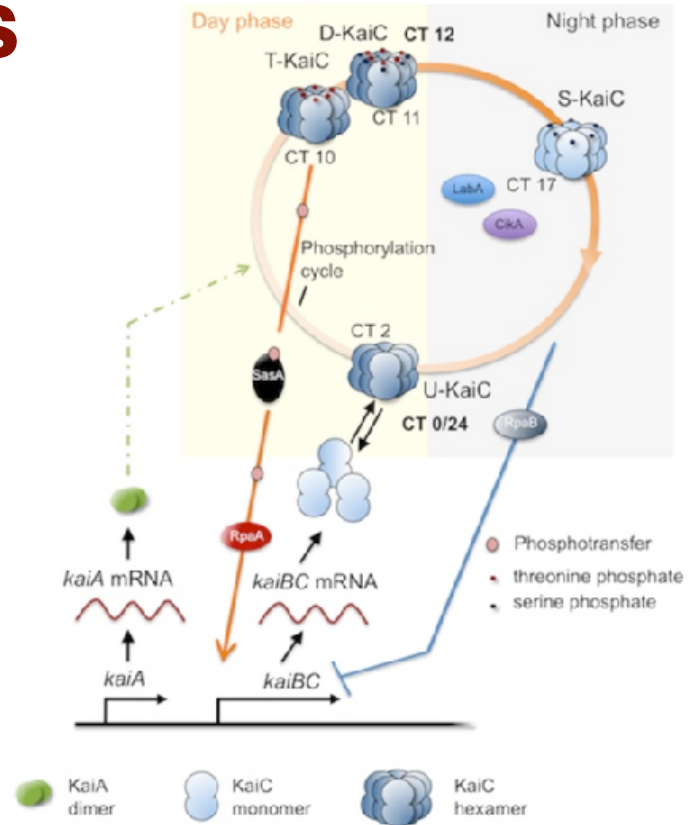
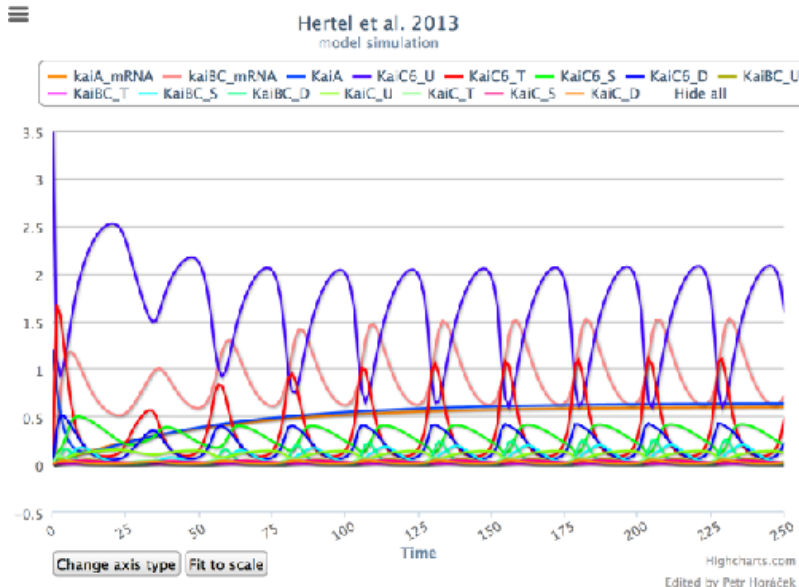
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 \end{aligned}$$

Genetic Networks

A typical example of genetic regulatory network is the circadian clock (here in cyanobacteria, peculiar), an oscillatory module regulated by alternation of light and dark.



Revealing a Two-Loop Transcriptional Feedback Mechanism in the Cyanobacterial Circadian Clock

Stefanie Hertel, Christian Bretschneider, Ilka M. Axmann

Published: March 14, 2013 • <http://dx.doi.org/10.1371/journal.pcbi.1002966>

A Noisy Life

A Noisy Life



Stochastic Gene Expression in a Single Cell

Michael B. Elowitz, *et al.*
Science **297**, 1183 (2002);
DOI: 10.1126/science.1070

Molecular interactions and gene expression in single cells are **random events**, the fewer the molecules involved, the more the effect of **noise**.

Models have to account for this.

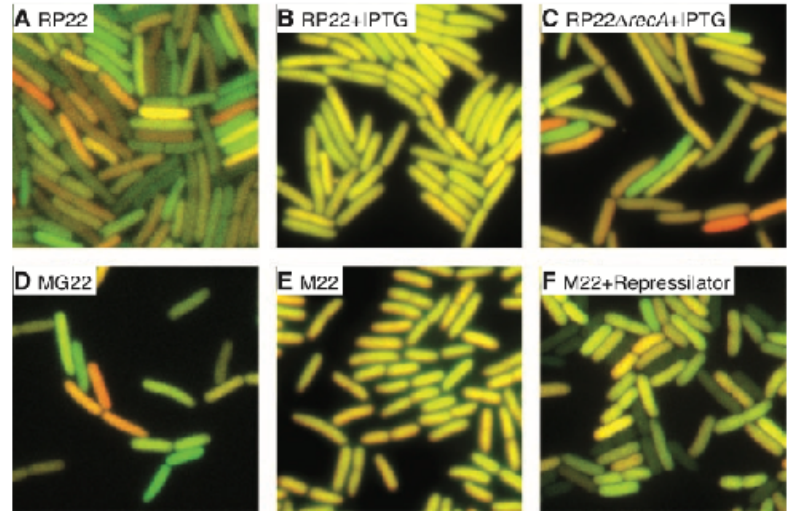


Fig. 2. Noise in *E. coli*. CFP and YFP fluorescence images were combined in the green and red channels, respectively. (A) In strain RP22, with promoters repressed by the wild-type *lacI* gene, red and green indicate significant amounts of intrinsic noise. (B) RP22 grown in the presence of lac inducer, 2 mM IPTG. Both fluorescent proteins are expressed at higher levels and the cells exhibit less noise. (C) As in (B), except the *recA* gene has been deleted, increasing intrinsic noise. (D) Another wild-type strain, MG22, shows noise characteristics similar to those of RP22. (E) Expression levels and noise in unrepresed *lacI* strain M22 are similar to those in *lacI* strains induced with IPTG (B). (F) M22 cells regulated by the Repressator (16), an oscillatory network that amplifies intrinsic noise.

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What are the **sources of noise** in cells?

Intrinsic and extrinsic contributions to stochasticity in gene expression

Peter S. Swain^{*†‡}, Michael B. Elowitz^{*§}, and Eric D. Siggia^{*}

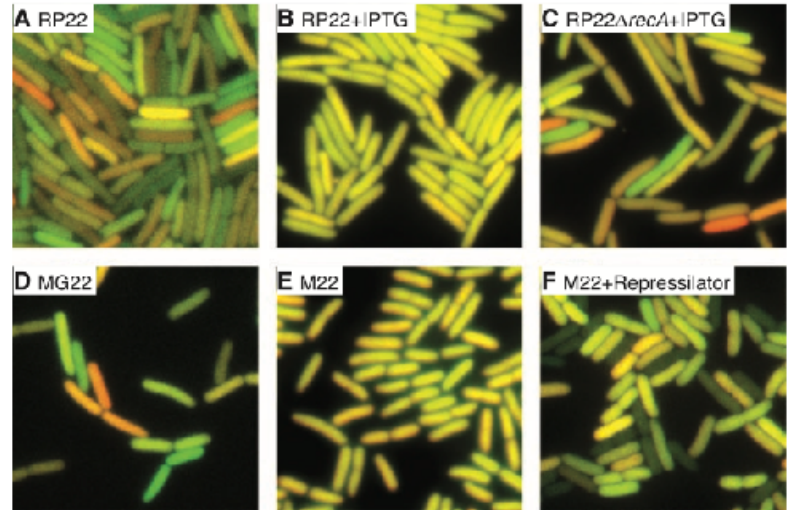


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Is Noise Always Detrimental?

What is **the role of noise** in cells?

Is it a nuisance to cope with, or it
has also been exploited by Nature?

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Stochasticity and Cell Fate
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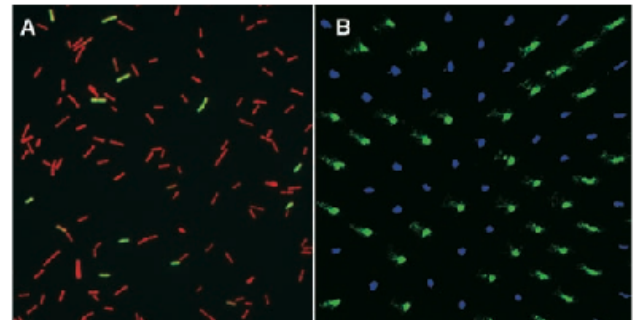


Fig. 1. Stochastic distribution of cell fates in bacteria and in insect photoreceptors. **(A)** Fluorescence micrograph of *B. subtilis* cells containing the coding sequence for GFP fused to the promoter for a gene under the control of the competence regulator ComK. The cells were visualized with a red stain; the green fluorescence reveals the subpopulation of cells that are ON for ComK. The cells are 1 to 2 μm in length. **(B)** Photograph of a whole adult *Drosophila* retina whose R8 photoreceptors were stained with antibodies to the green-sensitive photopigment Rh6 (green) and the blue-sensitive photopigment Rh5 (blue). The horizontal distance between photoreceptors is about 10 μm .

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Cells can randomly switch to different operating modes (multi-stability). This fosters exploration of surviving strategies at the population level.

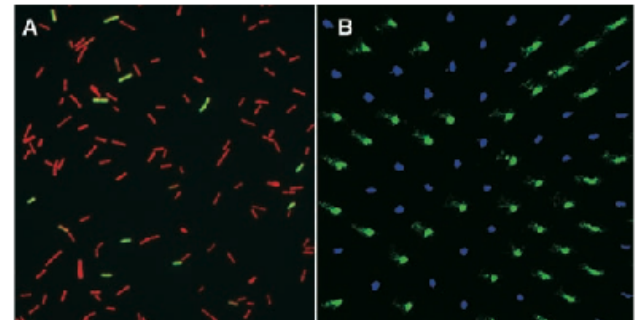
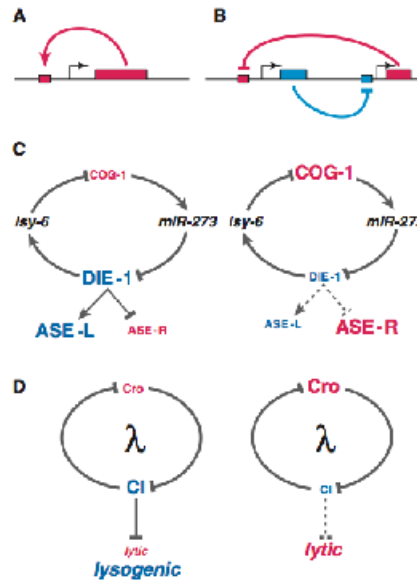


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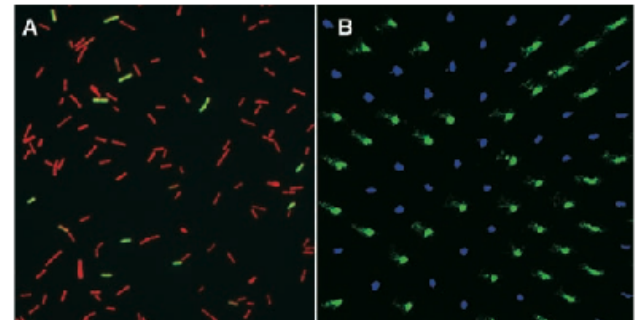
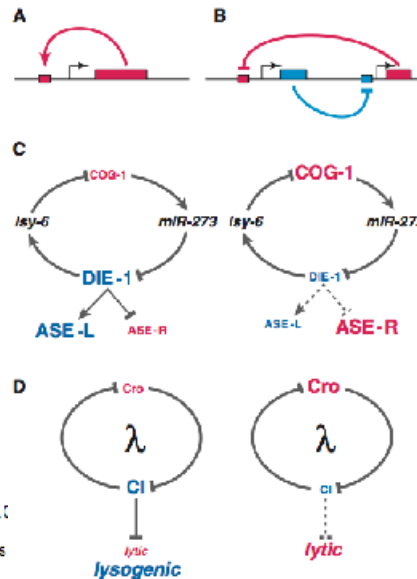


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Molecular Systems Biology 5: Article number 326; doi:10.1093/msb/nbn036
 Citation: Molecular Systems Biology 5:326
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 www.molecular-systemsbiology.com

REVIEW

Strategies for cellular decision-making

Theodore J Perkins¹ and Peter S Swain^{2*}

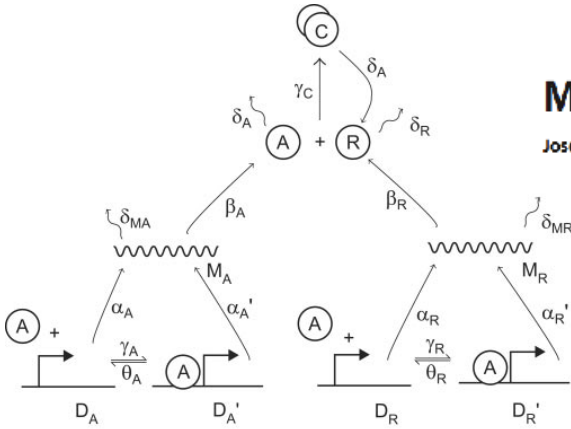
extracellular enviro

Noise-induced oscillations

Noise-induced oscillations

Mechanisms of noise-resistance in genetic oscillators

José M. G. Vilar^{*†}, Hao Yuan Kueh^{*}, Naama Barkai[‡], and Stanislas Leibler^{**§}



$$dD_A/dt = \theta_A D'_A - \gamma_A D_A A$$

$$dD_R/dt = \theta_R D'_R - \gamma_R D_R A$$

$$dD'_A/dt = \gamma_A D_A A - \theta_A D'_A$$

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$$dM_A/dt = \alpha'_A D'_A + \alpha_A D_A - \delta_{M_A} M_A$$

$$dA/dt = \beta_A M_A + \theta_A D'_A + \theta_R D'_R - A(\gamma_A D_A + \gamma_R D_R + \gamma_C R + \delta_A)$$

$$dM_R/dt = \alpha'_R D'_R + \alpha_R D_R - \delta_{M_R} M_R$$

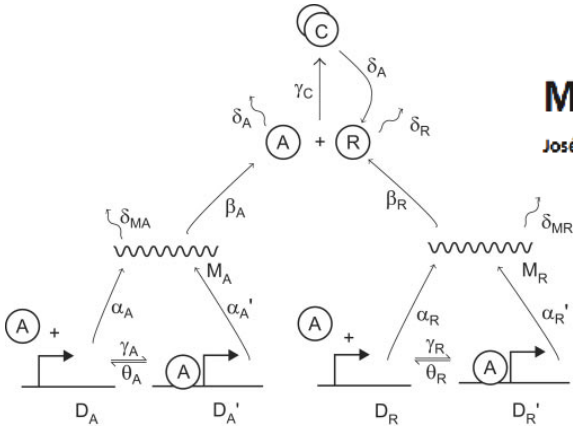
$$dR/dt = \beta_R M_R - \gamma_C A R + \delta_A C - \delta_R R$$

$$dC/dt = \gamma_C A R - \delta_A C,$$

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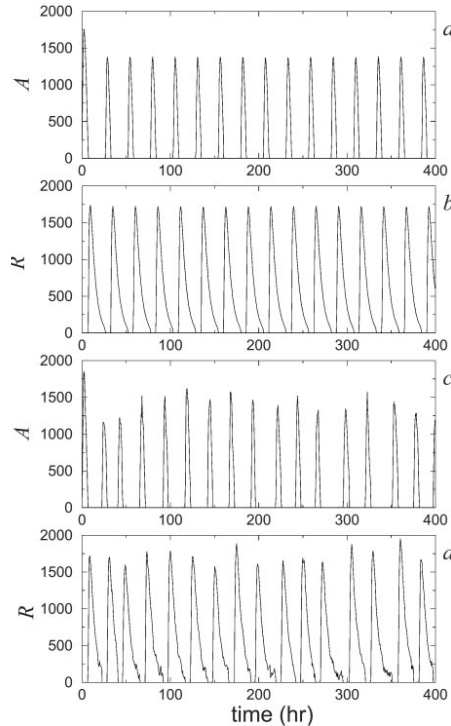
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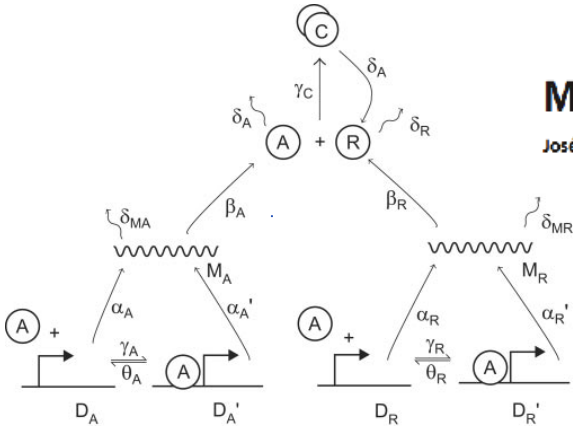
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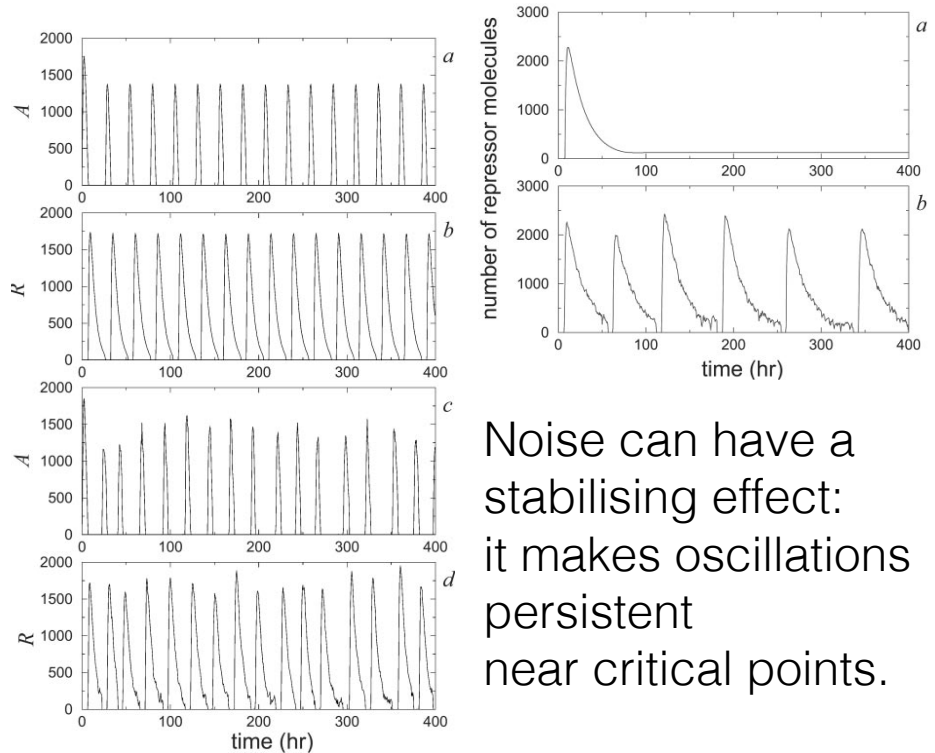
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$$\begin{aligned}
 dD_A/dt &= \theta_A D'_A - \gamma_A D_A A \\
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 dD'_R/dt &= \gamma_R D_R R - \theta_R D'_R \\
 dM_A/dt &= \alpha'_A D'_A + \alpha_A D_A - \delta_{M_A} M_A \\
 dA/dt &= \beta_A M_A + \theta_A D'_A + \theta_R D'_R \\
 &\quad - A(\gamma_A D_A + \gamma_R D_R + \gamma_C R + \delta_A) \\
 dM_R/dt &= \alpha'_R D'_R + \alpha_R D_R - \delta_{M_R} M_R \\
 dR/dt &= \beta_R M_R - \gamma_C A R + \delta_A C - \delta_R R \\
 dC/dt &= \gamma_C A R - \delta_C C,
 \end{aligned}$$



Noise can have a stabilising effect: it makes oscillations persistent near critical points.

Stochastic Modelling

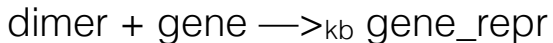
Chemical Reaction Networks can be modelled as **Markov Population Processes**. Variables count the amount of molecules per each species. Update vectors are defined by reactions. Rates depend on the total population (mass action, Hill).

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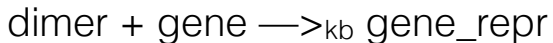
$$\nu_i: (0, +1, 0, \dots, -1), f_i(x) = \frac{k_p}{k_d} x_{\text{gene}}$$



Stochastic Modelling

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Counting variables:

$X_{\text{gene}}, X_{\text{gene_repr}}, X_{\text{mrna}}, X_{\text{protein}}, X_{\text{dimer}}$

Propensity of a reaction (expected frequency)

follows the mass action law:

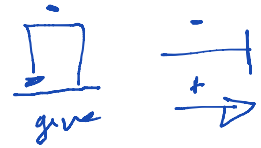
$$a_1(\mathbf{x}) = k_p X_{\text{gene}}; \quad a_5(\mathbf{x}) = k_b X_{\text{dimer}} X_{\text{gene}};$$

$$a_3(\mathbf{x}) = k_1 X_{\text{protein}} (X_{\text{protein}} - 1)/2;$$

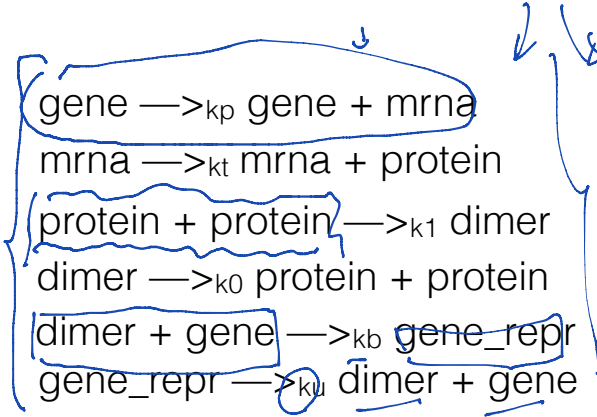
Update of a reaction: net variation of each species

$$\mathbf{v}_1 = (0,0,1,0,0), \quad \mathbf{v}_3 = (0,0,0,-2,1), \quad \mathbf{v}_5 = (-1,1,0,0,-1)$$

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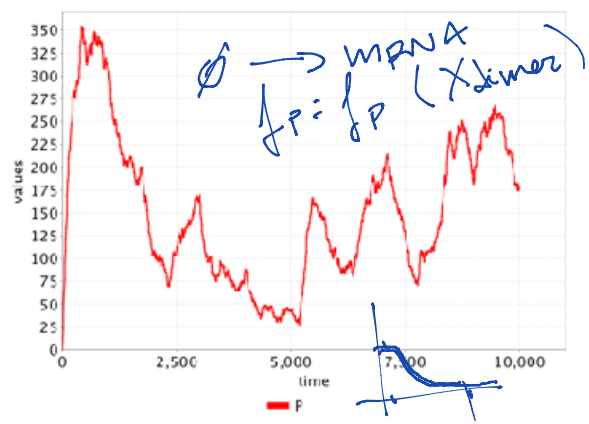
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Update of a reaction: net variation of each species

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Typical rate functions

- **Mass Action:** rate proportional to concentration/ numbers. The only one having a physical interpretation.



- **Hill Kinetics.** Typically used for enzymatic reactions or to implicitly model gene expression.

$f_p(X_d) = k_p \cdot \frac{1}{k^n + X_d^n}$

$$\frac{dx}{dt} = g(x) \quad \frac{1}{V} \frac{dX}{dt} = g(x) \quad \frac{dX}{dt} = \gamma g(x)$$

Rates and Scaling

$$x = \frac{X}{N_A \cdot V} \quad N_A \cdot V = \gamma$$

$$V \approx 10^{-15} \quad N_A \cdot V \approx 10^9$$

$$X = x/V$$

Biochemical reactions happen in a volume V . We can convert molecule numbers into concentrations (often micro or nano-molar) dividing by V .

\approx counting molecules

Molecule numbers: variables X count the number of molecules. Updates are integers.

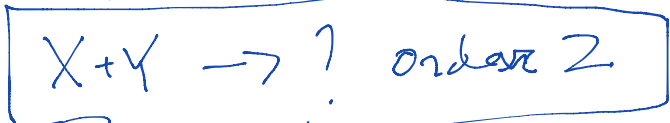
Concentrations: variable x are concentrations. Updates are multiple of $1/V$.

How do rates change while passing from numbers to concentrations?

$f(\vec{X}) = \bar{c} \cdot X \cdot Y$ $\bar{g}(\vec{x}) = \bar{c} \cdot x \cdot y$ $x = \frac{X}{V}$
 \uparrow \uparrow \uparrow
 number speed $\bar{c} = \frac{\bar{k}}{V^2}$
 does not change

$f(\vec{X}) = \bar{g}(\vec{x}) \implies \bar{c} \cdot X \cdot Y = \bar{c} \cdot \frac{X}{V} \cdot \frac{Y}{V} \implies \bar{c} = \frac{\bar{k}}{V^2}$

Example: dimerisation (P monomer, P₂ dimer)



$$c = \frac{k}{V}$$

$X \rightarrow ?$ order 1

$$cX \equiv \gamma \cdot k \cdot X$$

$$cX = \gamma \cdot k \cdot \frac{X}{V} \implies c = k$$

$$c := \frac{k}{V^2}$$

$$g(x) = k \cdot x \cdot y$$

$$\bar{g}(\vec{x}) = \gamma \cdot k \cdot x \cdot y = \gamma \cdot g(x)$$

$$\emptyset \rightarrow X : c = \gamma \cdot k$$

Rates and Scaling

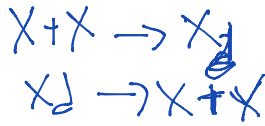
If we express the model in terms of concentrations, by multiplying rate and update vector of each transition and adding them up, we obtain the standard deterministic model of chemical kinetic, as a set of ODEs, the **reaction rate equations**.

$$(V_j, g_j(x))$$

$$c \begin{pmatrix} x^2 \\ x \\ 1 \end{pmatrix} = \gamma \begin{pmatrix} x^2 \\ x \\ 1 \end{pmatrix} \rightarrow c = \frac{2k}{\gamma}$$

$$\frac{dx}{dt} = \sum_j V_j g_j(x)$$

Example: dimerisation.



$$f(x) = k \cdot x^2$$

$$g(x) = h \cdot x_2$$

$$f(x) = \gamma g(x) \Rightarrow c = h$$

$$\frac{dx}{dt} = \sum_j \frac{V_j}{\gamma} g_j(x)$$

Relation between stochastic and deterministic rate constants

$$\frac{dx}{dt} = -2 \cdot k \cdot x^2 + 2 \cdot h \cdot x_2$$

$$\frac{dx_2}{dt} = k \cdot x^2 - h \cdot x_2$$

$$2x_2 + x = \text{const}$$

~~multiply by?~~
const seen.

$$= \sum_j V_j g_j(x)$$