

METALLI IN MEDICINA

A.A. 2016-2017

PARTE 8

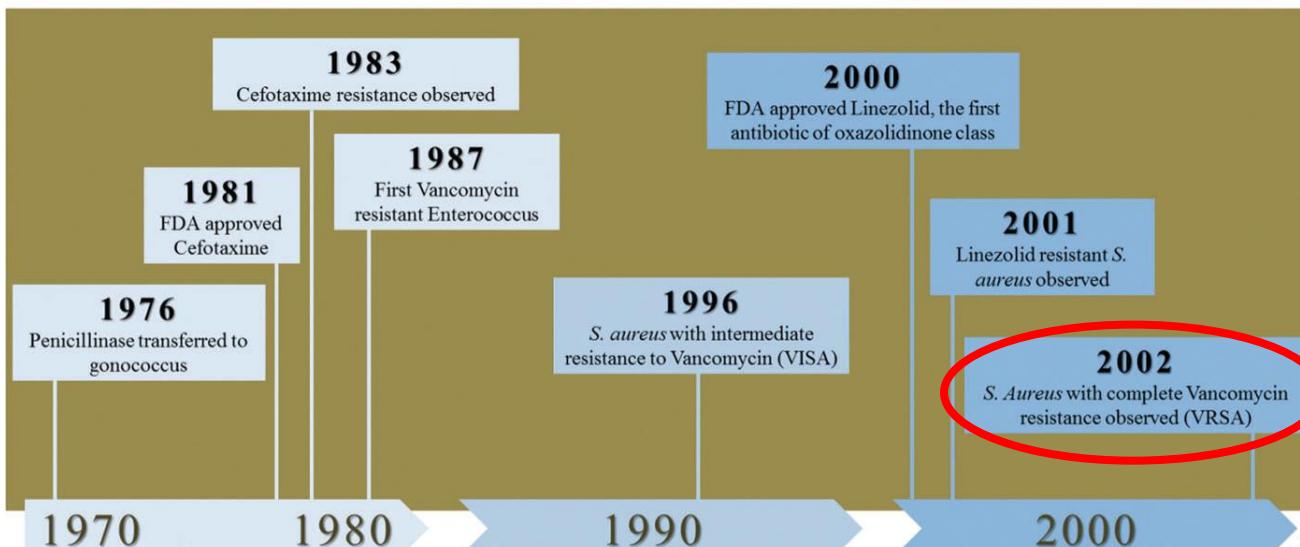
Enzo Alessio

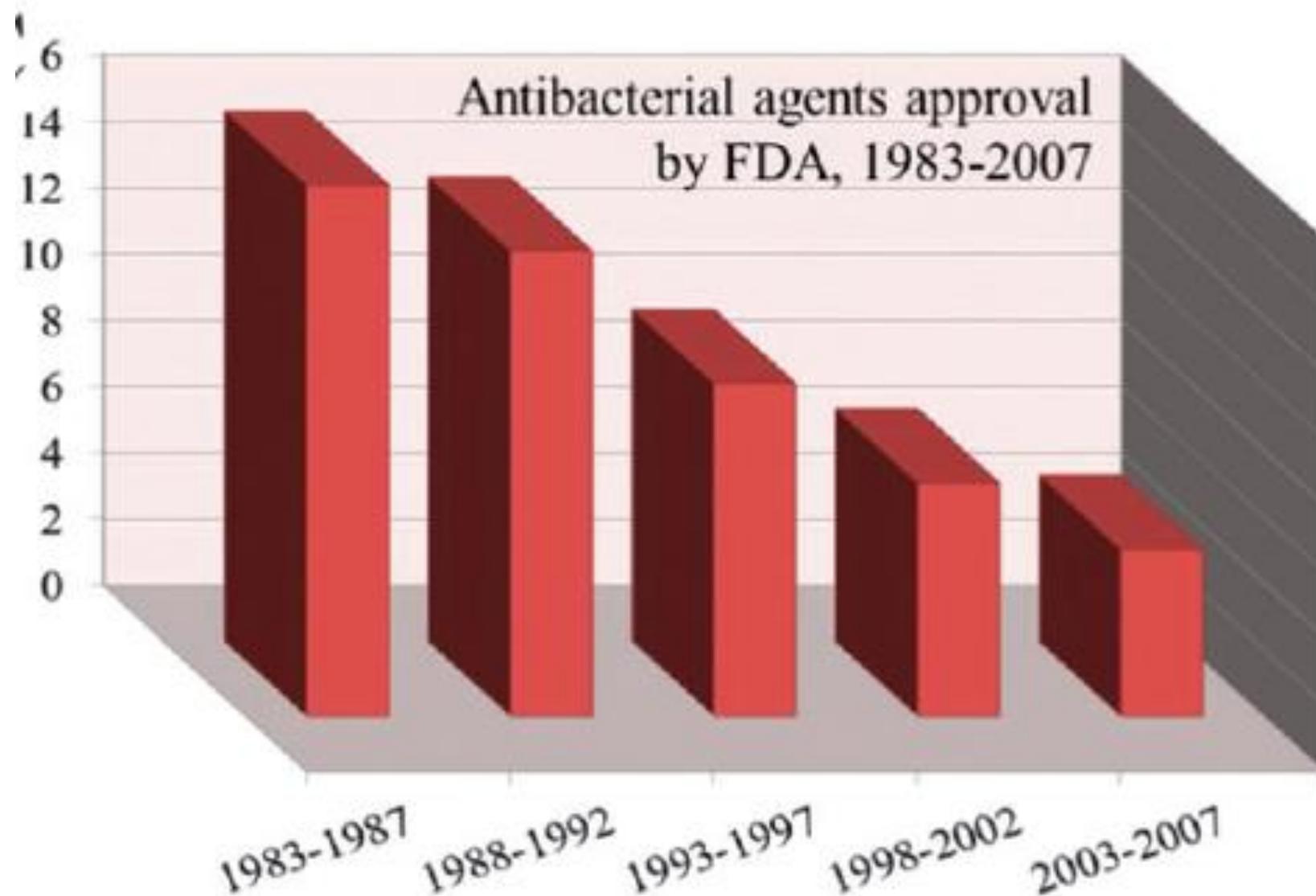
alessi@units.it



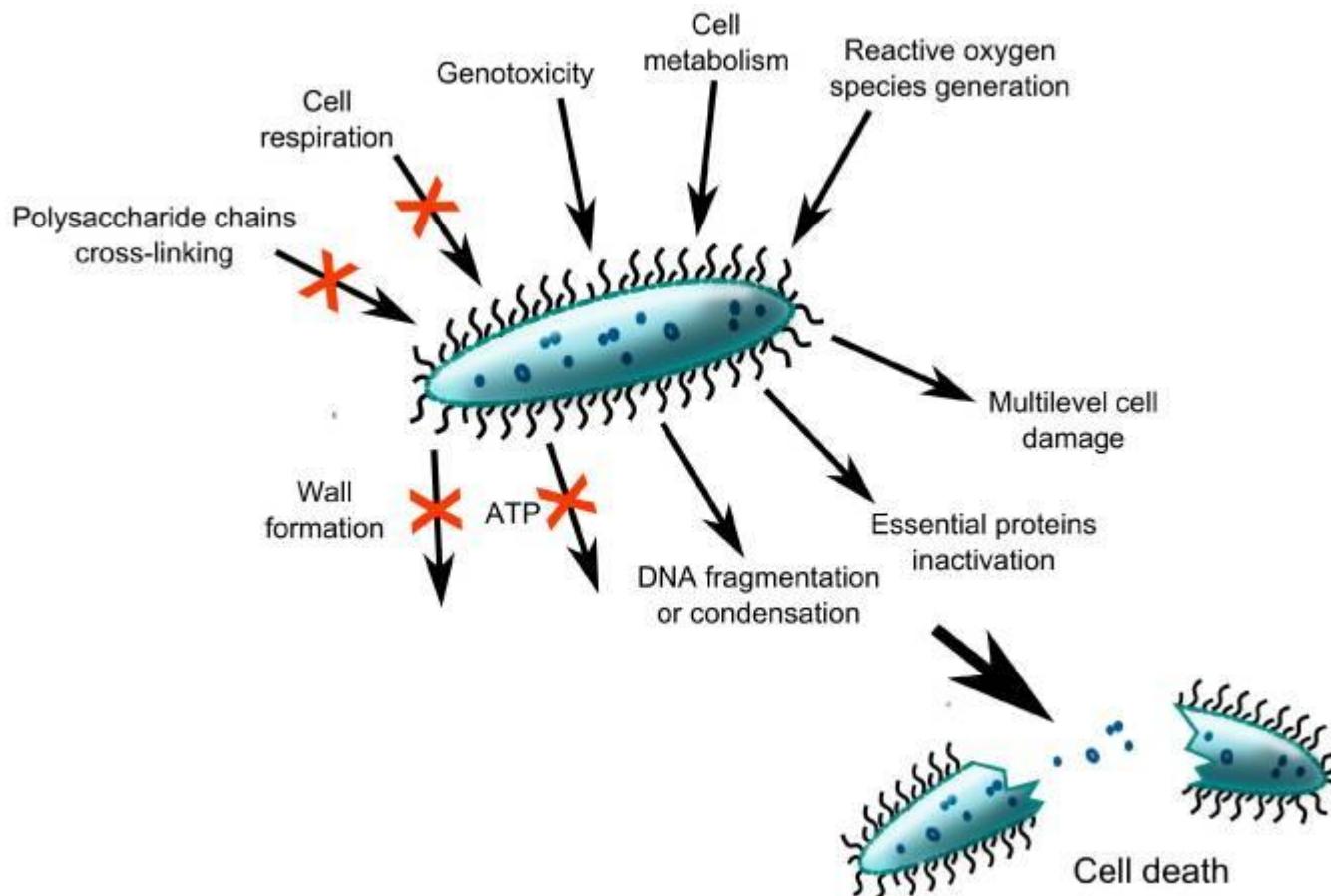


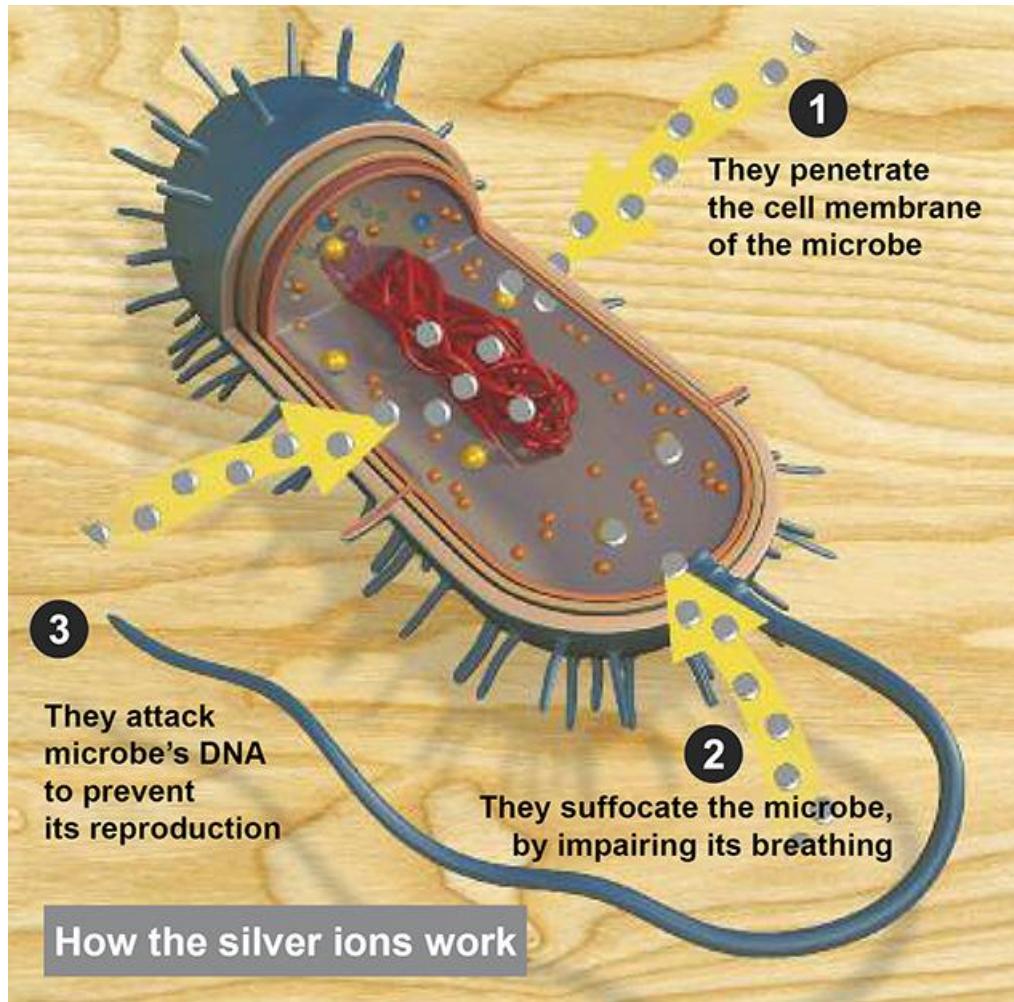
Antibacterials





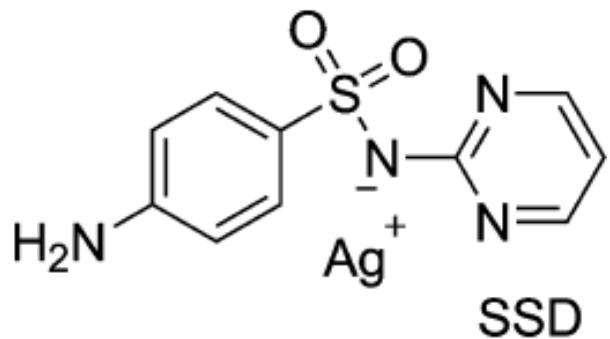
Multiple mechanism of action of Ag⁺ ions







Silver salts



silver sulfadiazine



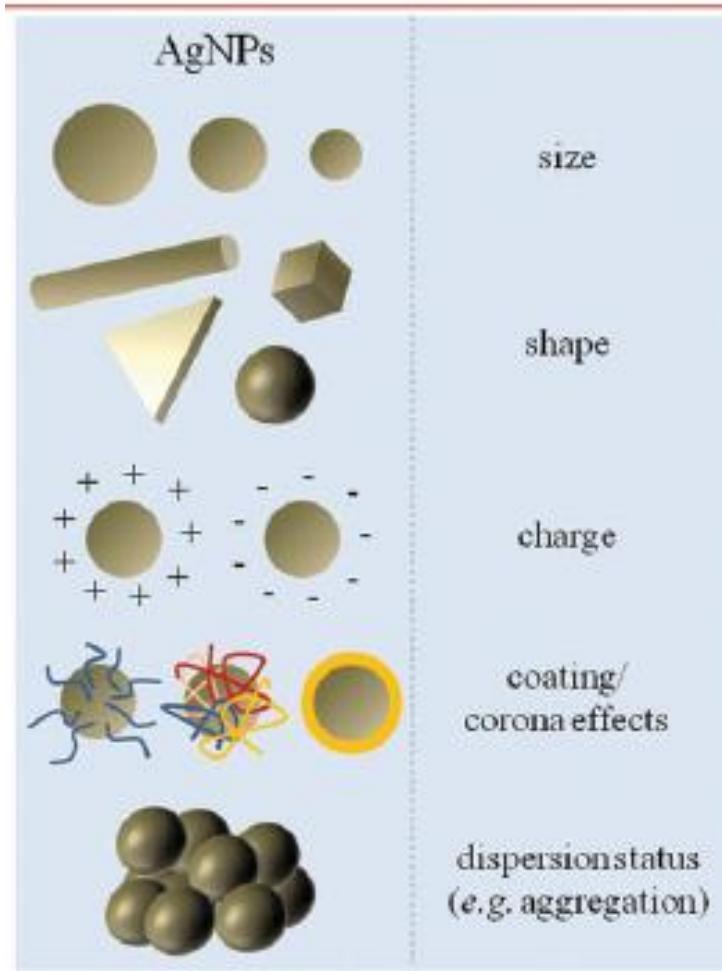


Silver nanoparticles (AgNPs)



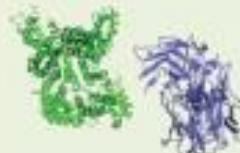
Estimated 2014 production of commercial AgNPs: 320 t

Ag⁺ release from AgNPs



Assay-like conditions

Media components



peptides/
proteins



carbohydrates



salts

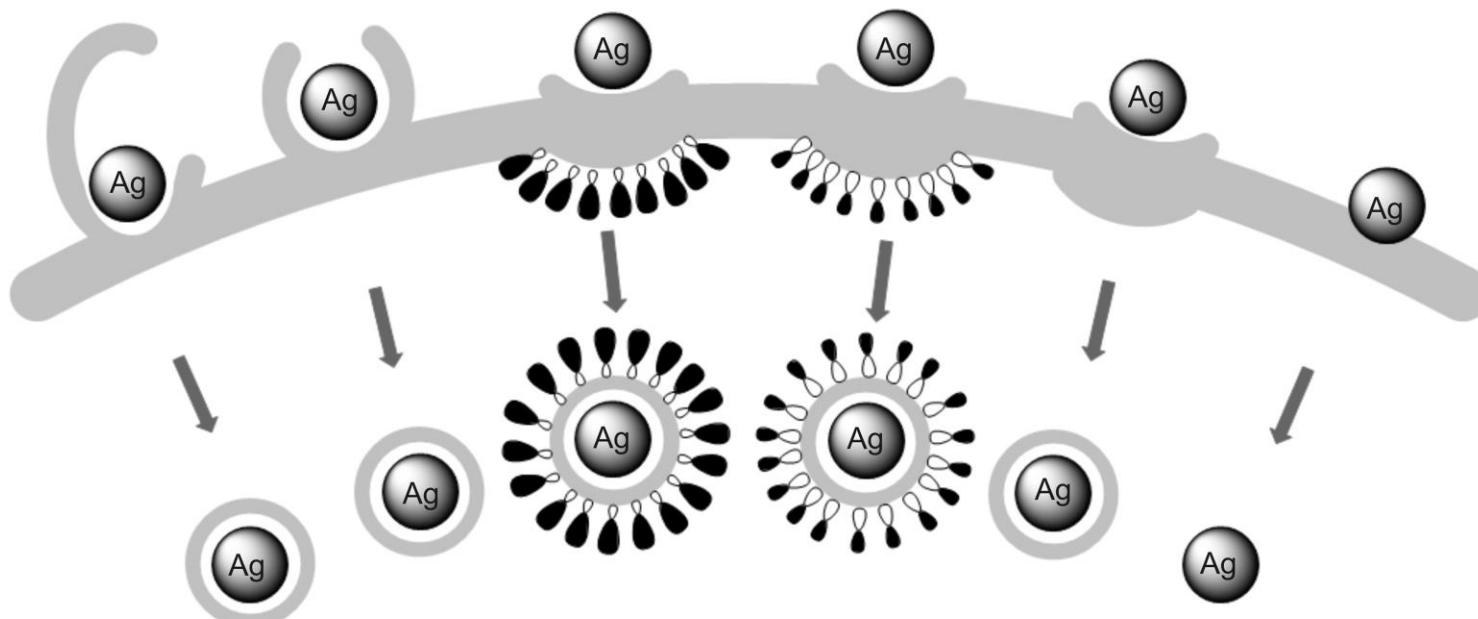
Bacterial strain
(strain-dependent acidification)



Ag⁺ release depends on

Uptake of AgNPs

- a) Macropinocytosis
- b) Phagocytosis
- c) Clathrin-mediated Endocytosis
- d) Caveolin-mediated Endocytosis
- e) Clathrin-independent and Caveolin-independent Endocytosis
- f) Diffusion



Legend:



AgNP

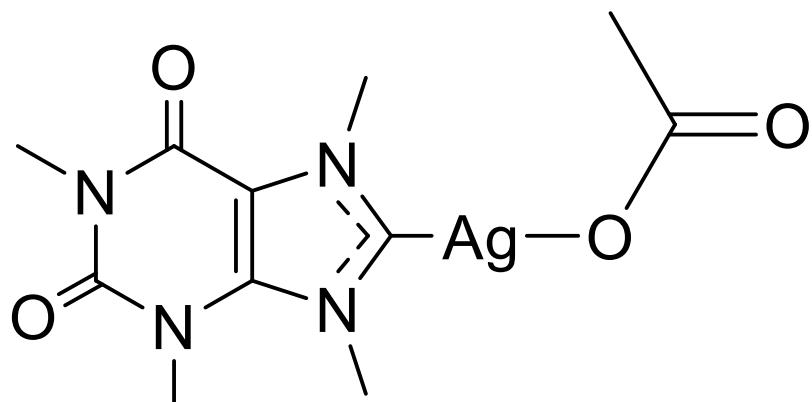


Clathrin



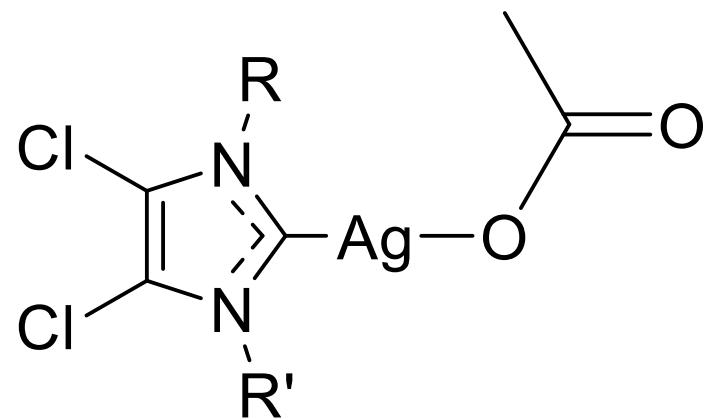
Caveolin

Antibacterial Ag-NHC compounds



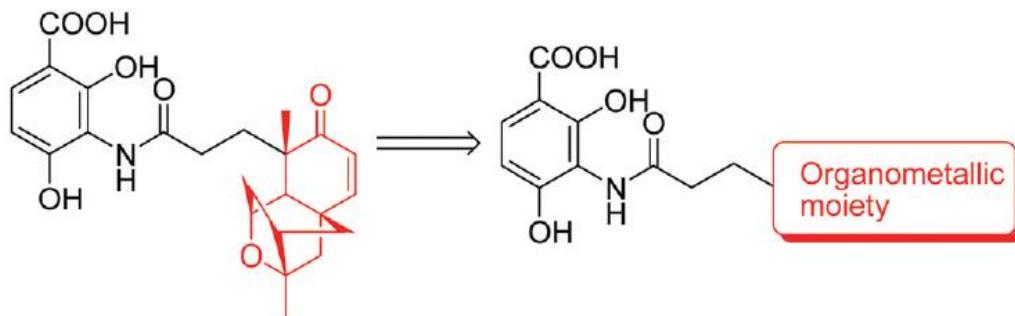
Silvamist®

Drug candidate: high activity against tobramycin-resistant pathogenic bacteria *in vitro* as well as *in vivo*.

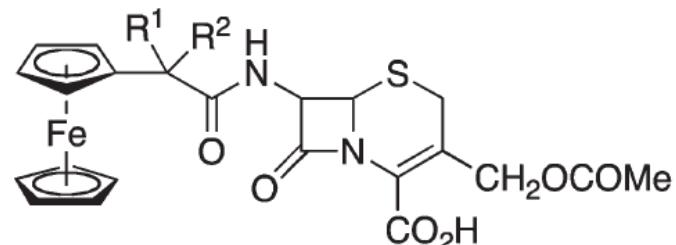
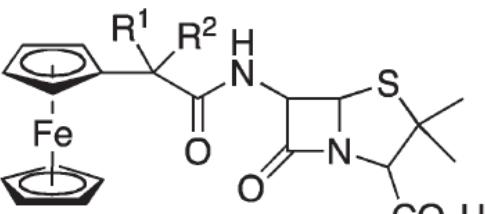
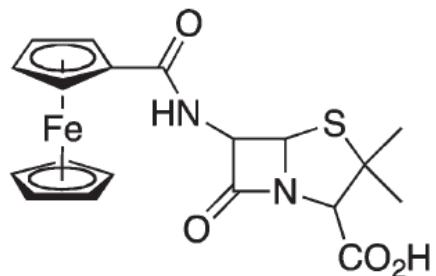


Improved stability to hydrolysis due to the electron-withdrawing Cl substituents that pull electron density from the carbene

Other strategies: metal modification of known antibiotics

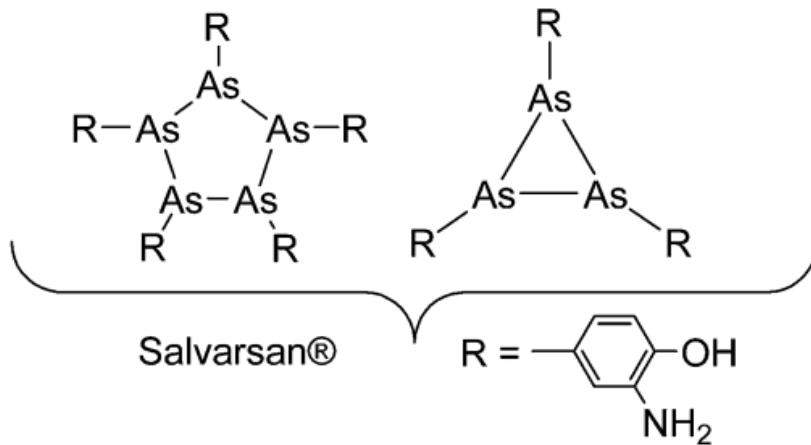


Platensimycin mimics





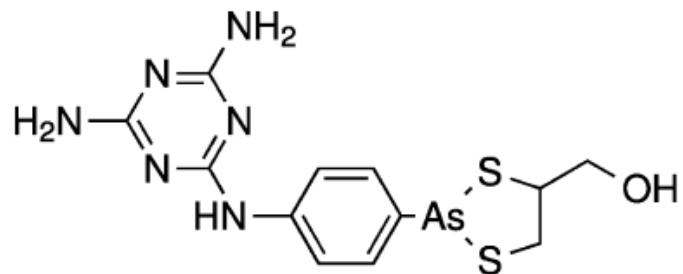
Other metals: As, Sb, Bi, Hg



Antimicrobial agent introduced in early 1900 for the treatment of the deadly bacterial infection *Syphilis*. Later replaced by modern antibiotics



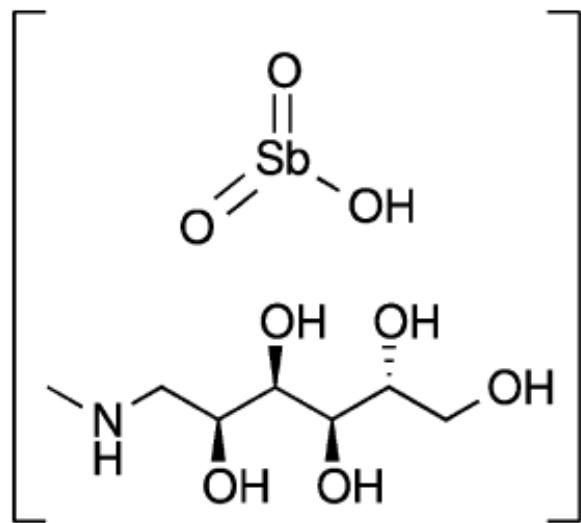
Antiparasitic compounds



melarsoprol

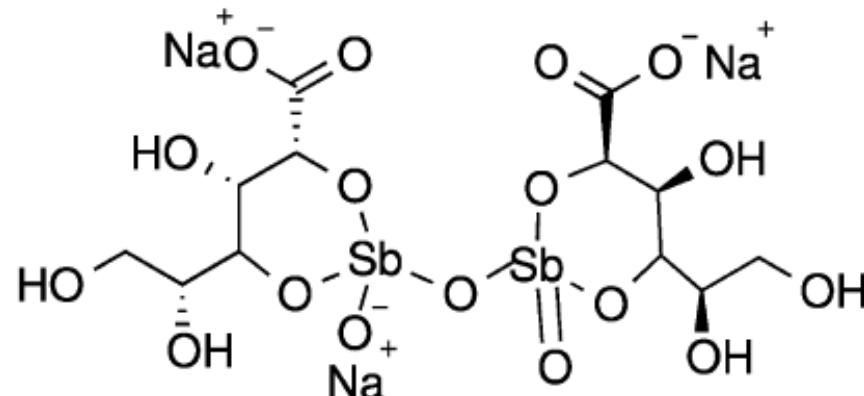
treatment of sleeping sickness (*African trypanosomiasis*)

Anti-leishmaniasis compounds



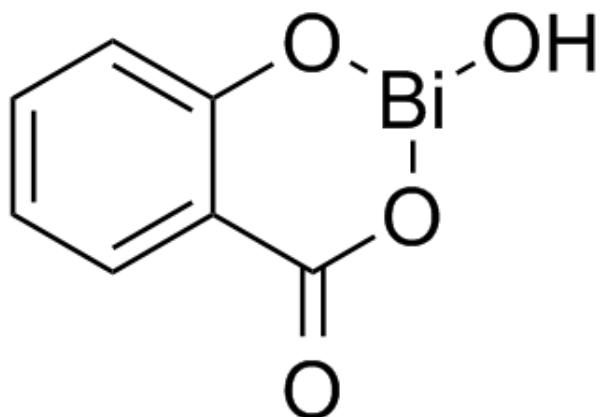
meglumine antimoniate

Veterinary use



sodium stibogluconate

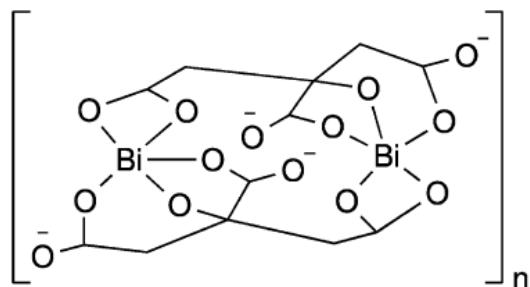
Sb, reduced to Sb(III), inhibits *trypanothione reductase*, an essential enzyme of the parasite



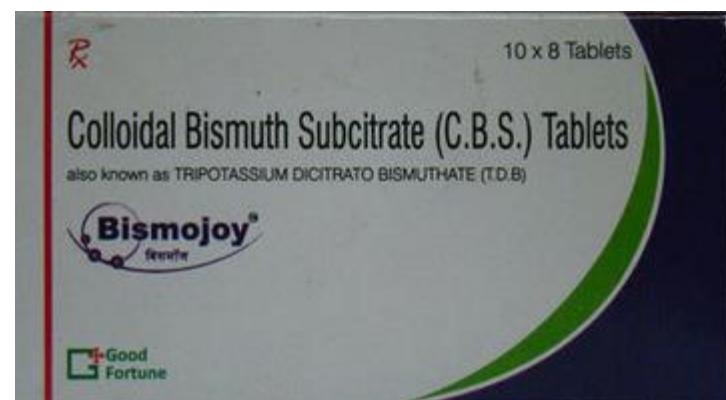
bismuth subsalicylate



The pink stuff (introduced 1901)

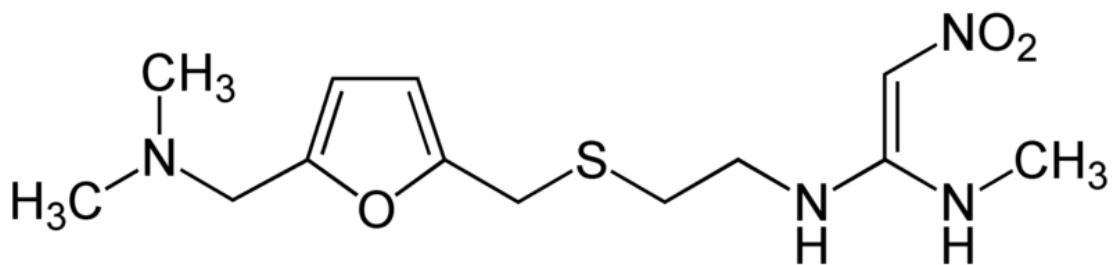


CBS





Helicobacter pylori



ranitidine bismuth citrate





Bismuth-ethanedithiol (BisEDT)

UNIVERSITY OF PENNSYLVANIA

Almanac

Tuesday
November 13, 2012
Volume 59 Number 12
www.upenn.edu/almanac

\$2.5 Million Grant for Penn Researchers to Investigate Anti-Infection Drug

A team of researchers led by Dr. Samir Mehta, chief of the Orthopaedic Trauma & Fracture Service at the Perelman School of Medicine at the University of Pennsylvania, has received a \$2.5 million grant from the Congressionally Directed Medical Research Program (CDMRP), provided through the US Department of Defense (DoD), to begin Phase 2 human trials of a study that examines the effective treatment of post-surgical orthopedic infections using Microbion Corporation's topical BisEDT drug. The University of Pennsylvania will work with a team of researchers from Microbion and the University of California, San Francisco, on the trial, set to begin pending FDA approval.

"We're honored to be given this award from



Samir Mehta

forms of surgery (2.8 percent) as a result of the high-energy nature of the injury. With approximately 2.6 million orthopaedic devices implanted annually in the United States, approximately 4.3 percent of patients (112,000) will suffer from a post-operative infection.

Orthopaedic extremity injuries also constitute the majority

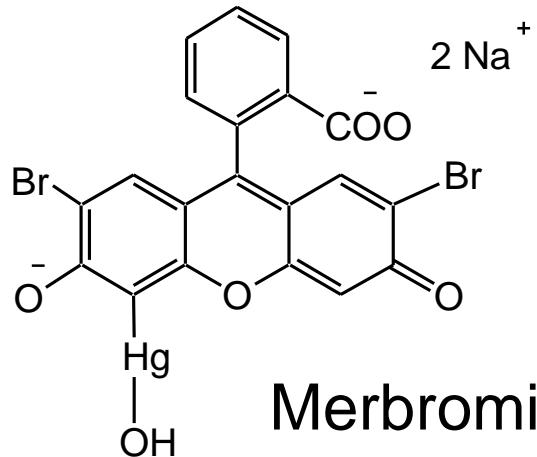
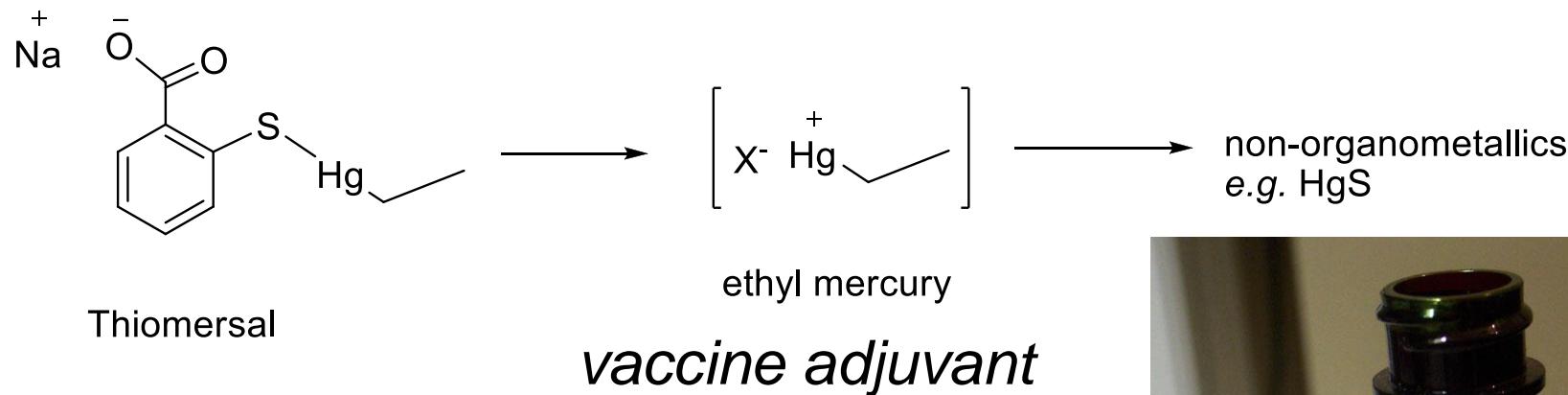
Debate at Penn: Have We Lost the Spirit of Compromise in America?

Will a hard-fought US election, replete with record spending and ever more divisive rhetoric, really change anything in Washington?

A panel including University of Pennsylvania President Amy Gutmann and moderated by Michael X. Delli Carpini, dean of Penn's Annenberg School for Communication, will debate these questions and more today at Penn. The event is sponsored by Penn's Annenberg School for Communication, the department of political science and the philosophy, politics & economics program.

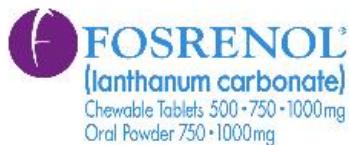
Panelists participating in *Time to Deal: Reawakening the Spirit of Compromise in American Politics* will address how we can best avoid

Antibacterial mercury compounds





Fosrenol™: a success story



FOSRENOL
Chewable Tablets

FOSRENOL
Oral Powder

Phosphorus
Burden in ESRD

Patient
Support

Conferences
and Resources

To reduce serum phosphate in patients with
end-stage renal disease (ESRD)

**HELP IT FALL
WITH FOSRENOL***
(lanthanum carbonate)

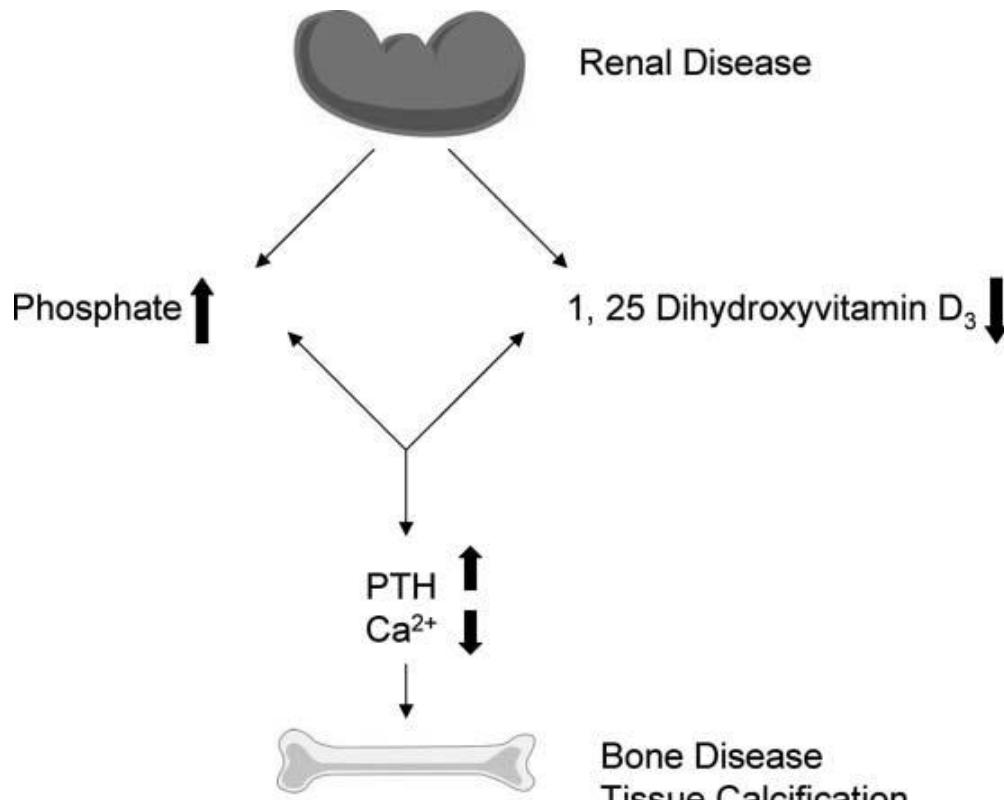
*Phosphorus reductions maintained for up to 3 years in patients
remaining on therapy (n=46)¹⁻³

- **FOSRENOL Chewable Tablets:** Approved in 2004 and used in US
clinical practices for more than a decade^{4,5} [LEARN MORE ▶](#)
- **FOSRENOL Oral Powder:** Available since May 2015, offering you
another approved administration option⁶ [LEARN MORE ▶](#)



Approved by FDA in 2004 for the treatment of
hyperphosphatemia (increased phosphate levels in serum) in
patients with end stage renal disease.

Phosphate metabolism is intimately linked with calcium metabolism, and is regulated by parathyroid hormone (PTH) and vitamin D



Pathological consequences of hyperphosphatemia: cardiac and vascular tissue calcification, bone malformations in the joints



*The **ideal phosphate binder** should have a high affinity for phosphate and should be able to bind dietary phosphate rapidly in the guts. It should have low solubility and little or no systemic absorption. It should be non-toxic, available as a palatable oral dosage form, with a low pill burden.*

Calcium phosphate binders are effective....however, calcium can be absorbed, resulting in hypercalcemia and increased risk of cardiovascular calcification.



Fosrenol™: a success story

Among the many lanthanide salts screened, $\text{La}_2(\text{CO}_3)_3 \cdot 4\text{H}_2\text{O}$ possessed the best phosphate binding properties:

- Optimal binding at pH 3–5, but retains binding activity in the full pH range of 1–7
- It is very insoluble and the La^{3+} cation does not cross biological membranes (when given by the oral route, >90% excreted in the feces, and <0.001% absorbed)
- No toxicity observed in animal studies, in particular no direct effects on calcium, vitamin D, or PTH metabolism



FosrenolTM represents a significant improvement in treatment options for patients with end-stage renal disease.

- in the acidic environment of the stomach lanthanum carbonate dissociates sufficiently to allow formation of a highly insoluble phosphate.
- It has the required pharmacokinetic properties, it is poorly absorbed, with both the parent molecule and the phosphate product being eliminated in the feces.
- Because of the lack of absorption it has no systemic toxicity, it has no detrimental effect on calcium, vitamin D or PTH metabolism, and is safe and well tolerated.
- Its effectiveness as a phosphate binder results in a lower pill burden for patients, an advantage over competing medications.