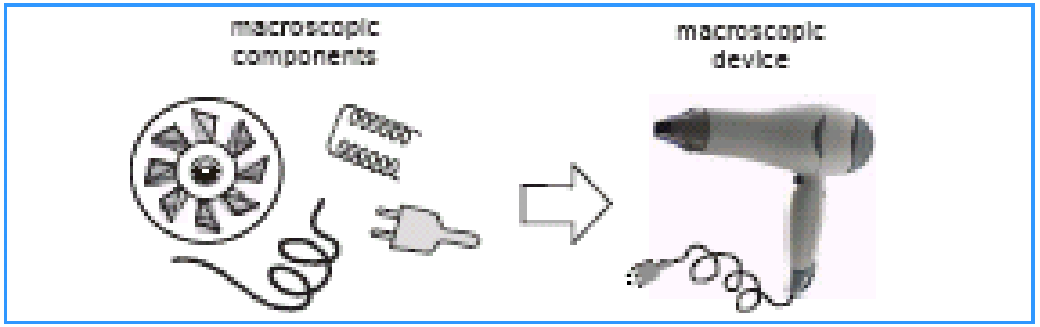
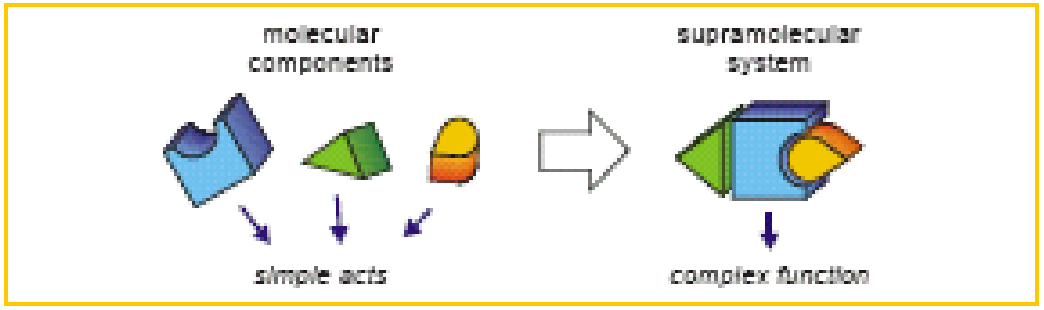
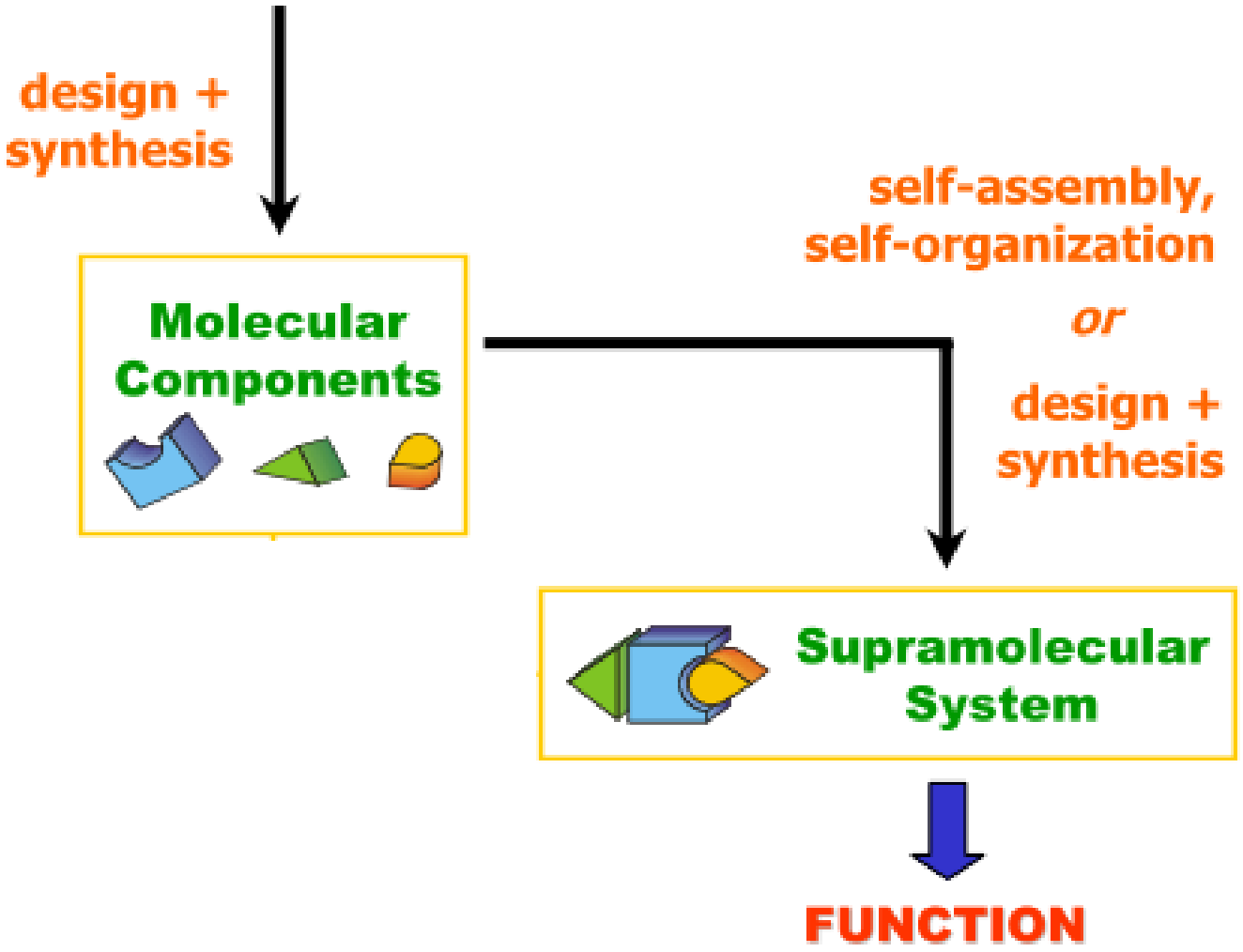
Dispositivi e Macchine Molecolari

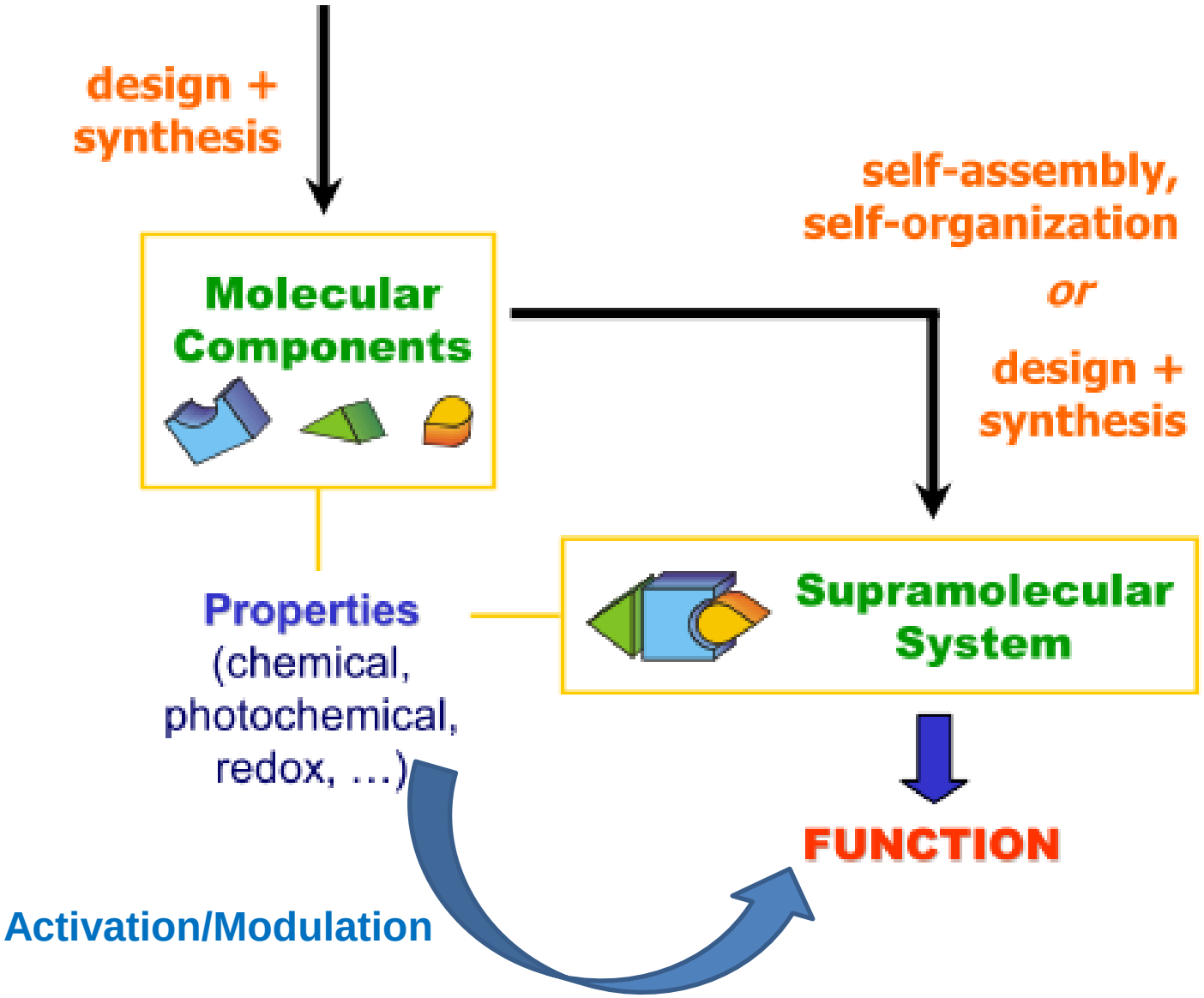
Macroscopic device



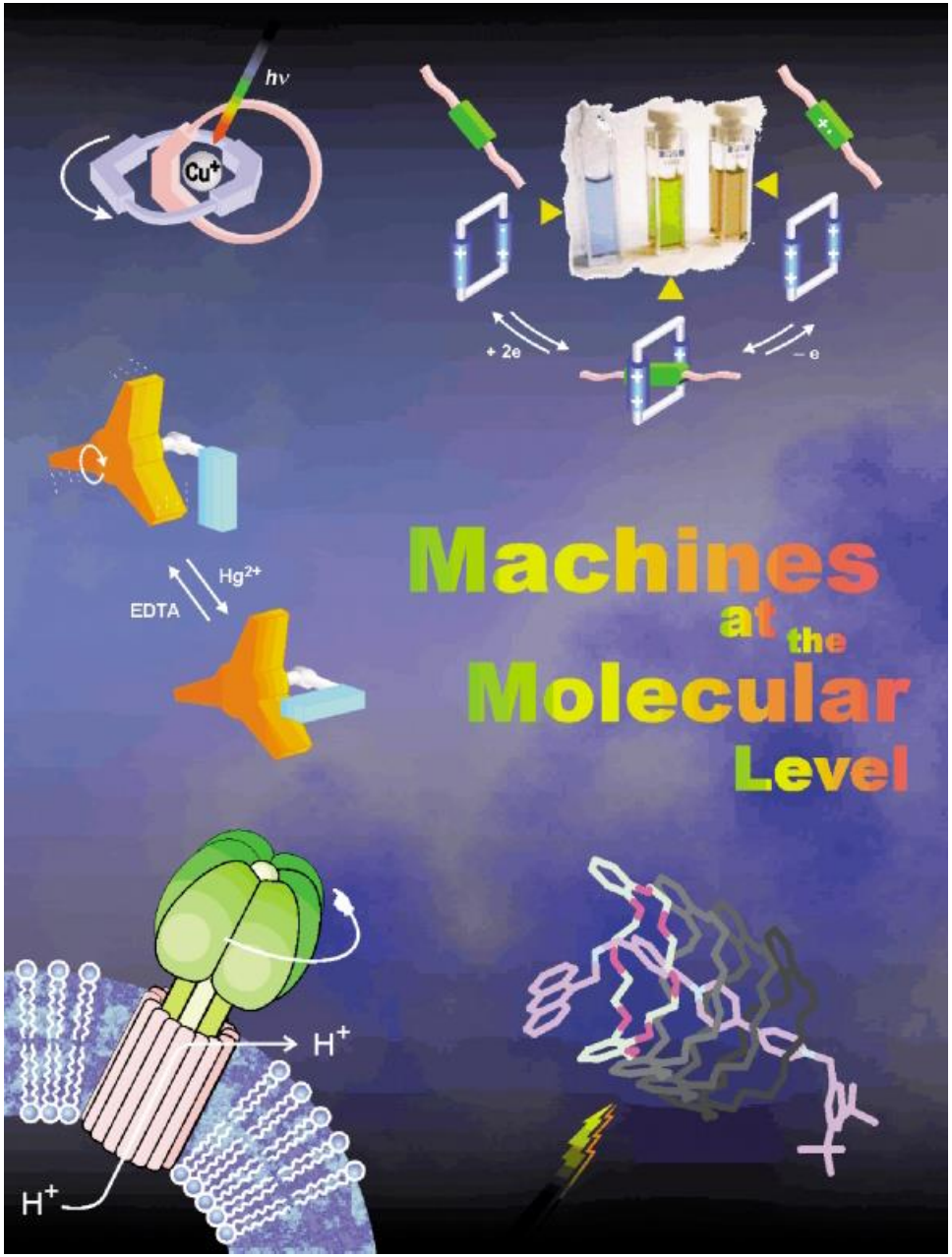
Molecular-level device




design +
synthesis
Molecular
Components
self-assembly,
self-organization
or
design +
synthesis
Supramolecular
System
FUNCTION


design +
synthesis
Molecular
Components
self-assembly,
self-organization
or
design +
synthesis
Supramolecular
System
Properties
(chemical,
photochemical,
redox, …)
FUNCTION
Activation/Modulation

- tipo di energia (chimica, fotoni, elettroni)
- monitoraggio (tecniche fotofisiche, elettrochimiche)
- processo ciclico
- tempo (picosecondi-minuti)
- funzione


hν
Cu+
+ 2e
– e
EDTA
Hg2+
H+
H+
Machines
at the
Molecular
Level

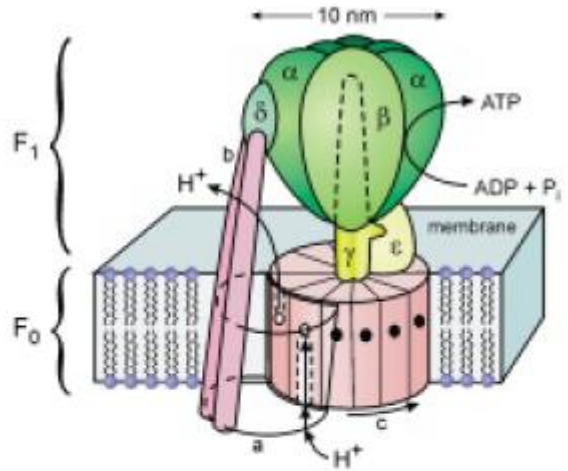


Figure 1. The structure of F_0F_1ATP synthase.[18] The catalytic region is composed of the subunits $\alpha-\varepsilon$. The proton channels lie at the interface between the subunits a and c (dashed lines indicate the putative inlet and outlet channels). Proton flow through the channels develops torque between the a and c subunits. This torque is transmitted to F_1 via the γ shaft and the ε subunit, where it is used to release ATP sequentially from the catalytic sites in F_1. The c subunit consists of 9–12 twin α-helices arranged in a central membrane-spanning array. The a subunit consists of 5–7 membrane-spanning α-helices and is connected to F_1 by the b and δ subunits. Reprinted by permission from ref. [16f] (Copyright© Macmillan Magazines Ltd 1998).

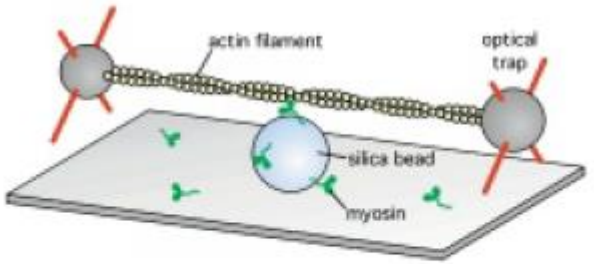
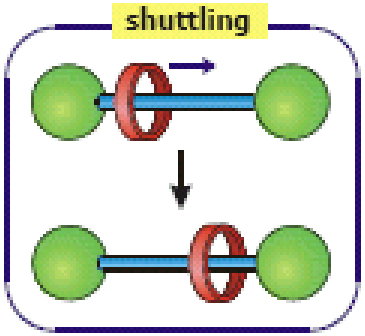
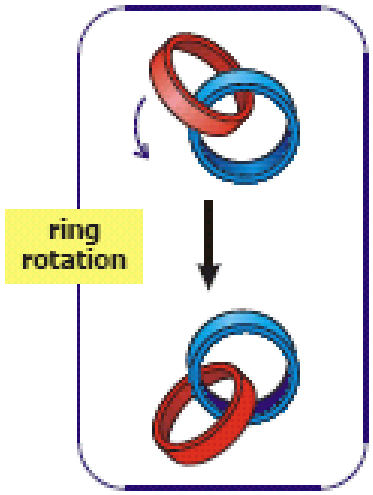
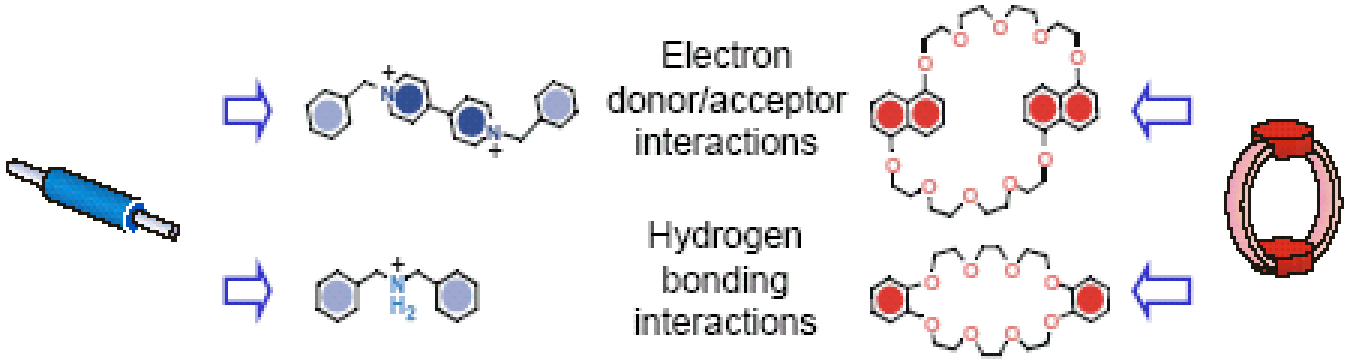
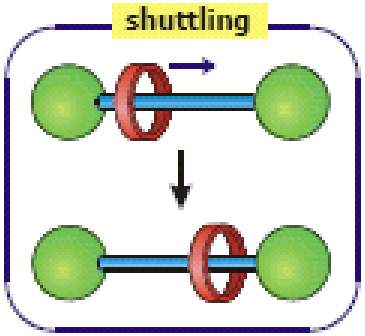


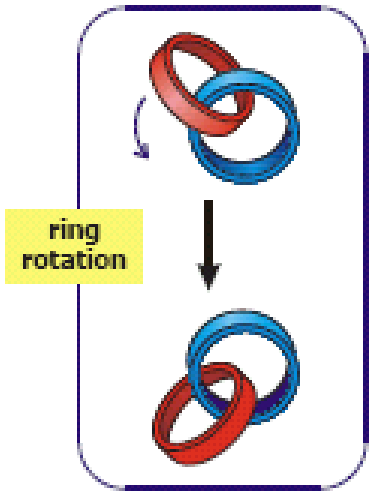
Figure 2. Experimental geometry used[19] to observe single myosin molecules binding and pulling an actin filament. The filament was attached at either end to a trapped bead. These beads were used to stretch the filament taut and move it near surface-bound silica beads that were decorated sparsely with myosin molecules. Adapted with permission from ref. [19] (Copyright© Macmillan Magazines Ltd 1994).

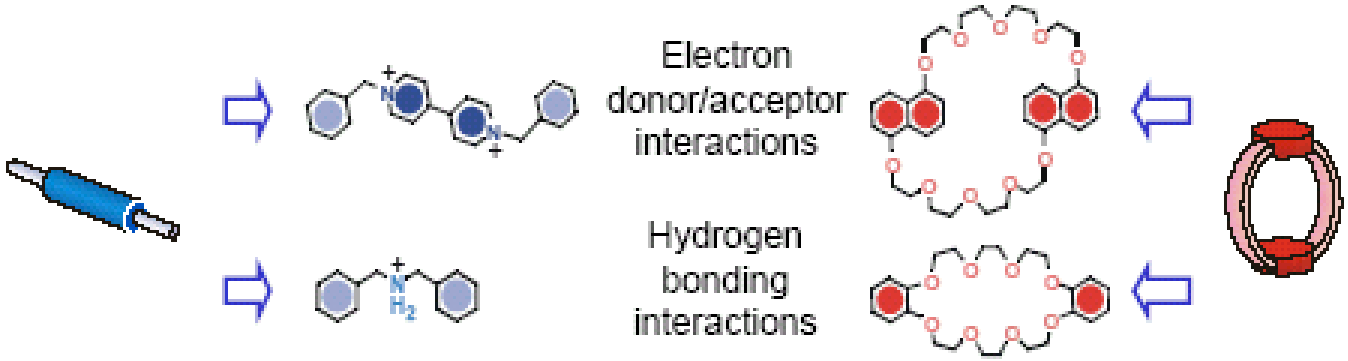

shuttling


ring
rotation

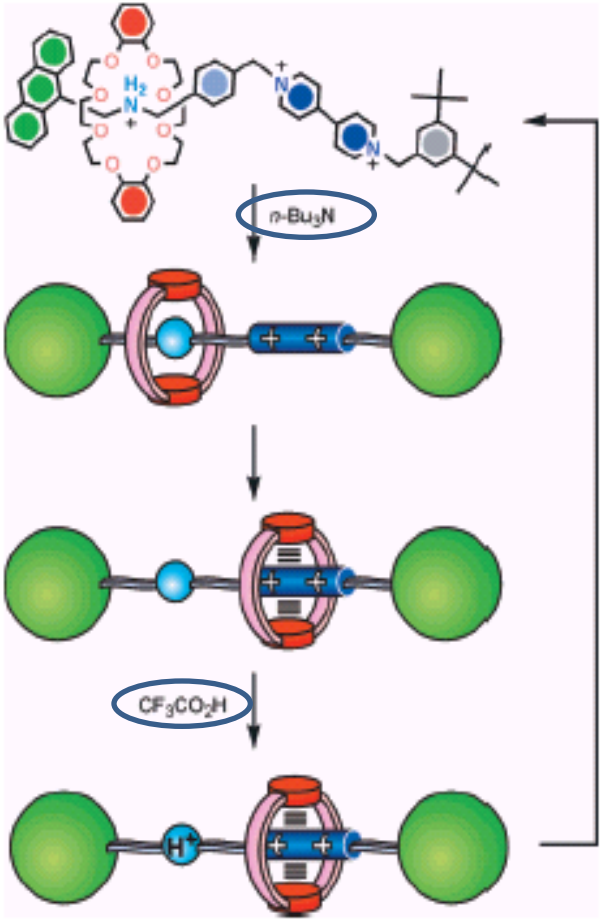

Electron
donor/acceptor
interactions
Hydrogen
bonding
interactions

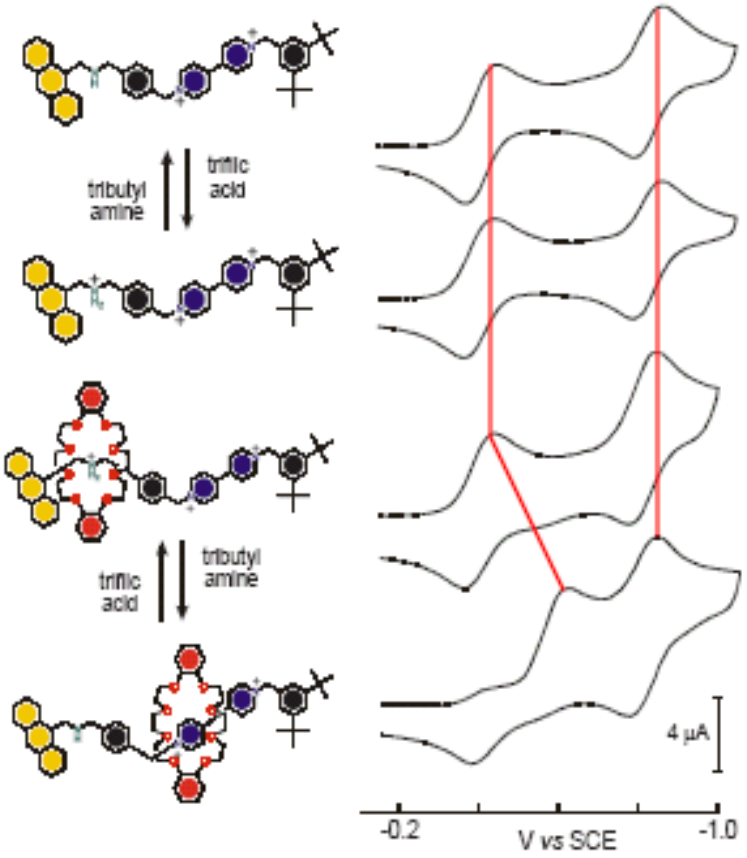

shuttling


ring
rotation

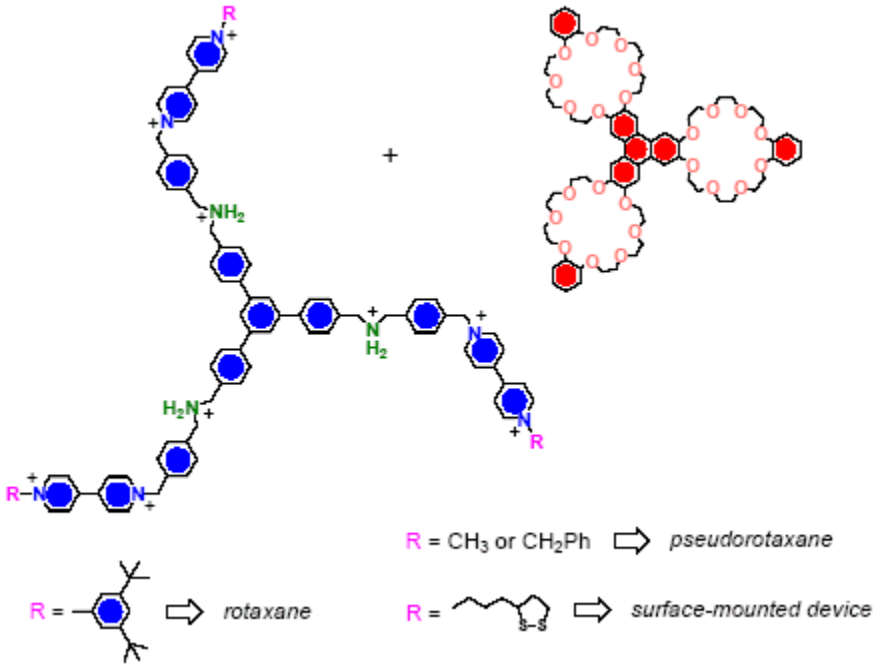

Electron
donor/acceptor
interactions
Hydrogen
bonding
interactions

Input chimico





Ascensore molecolare



Strong N^+–H····O

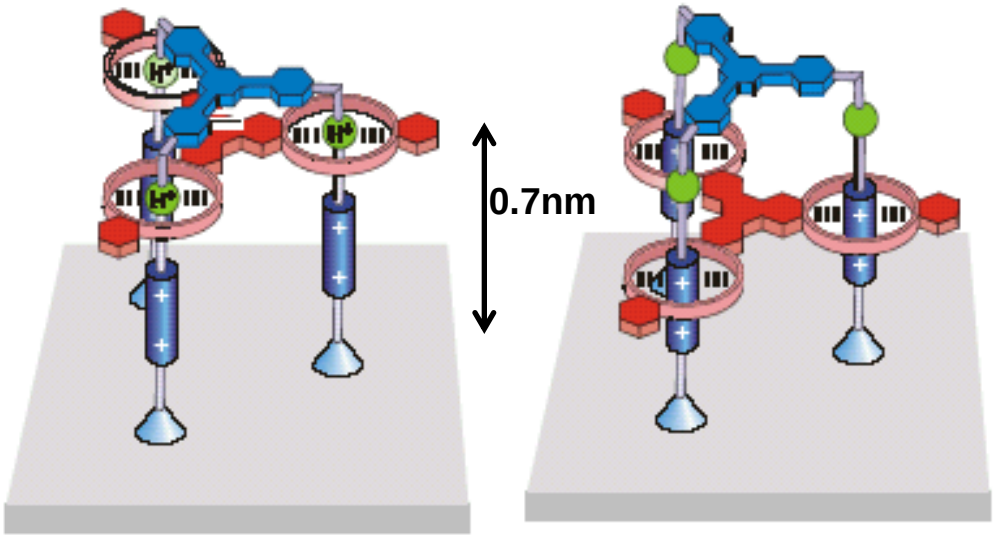
π–π

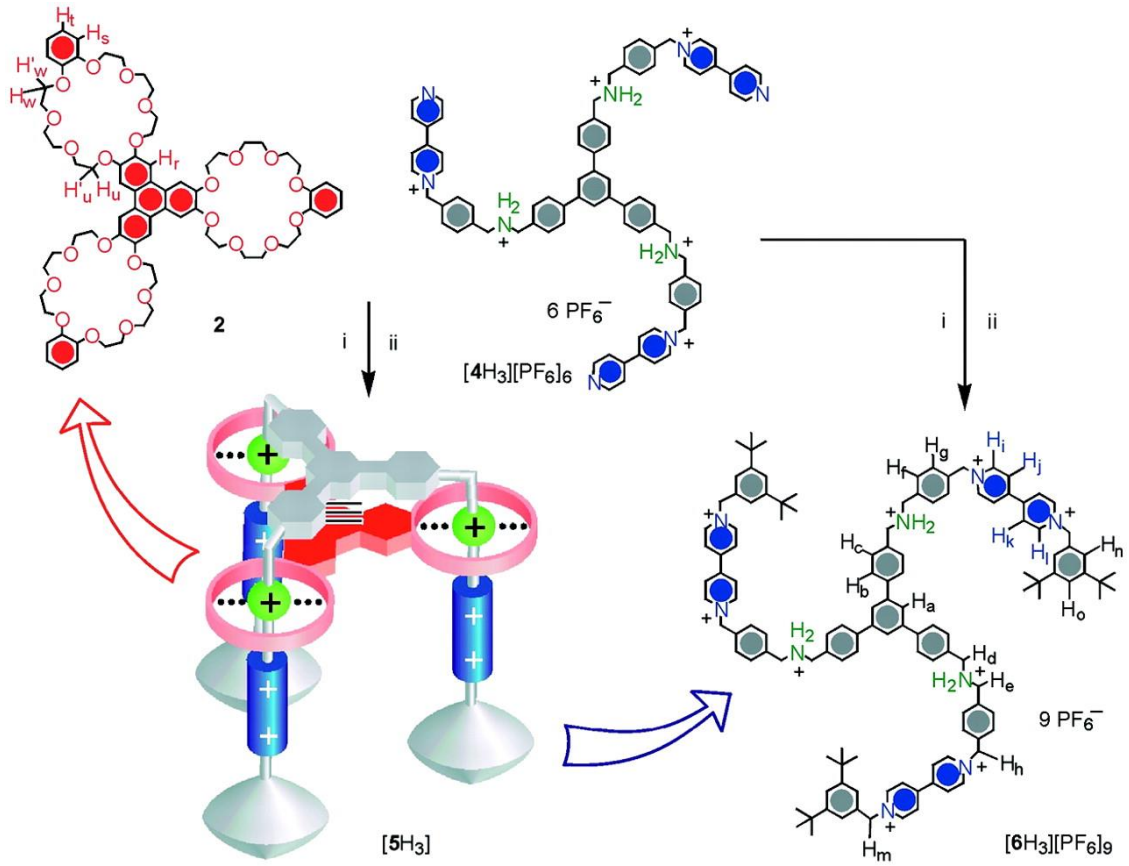
acid

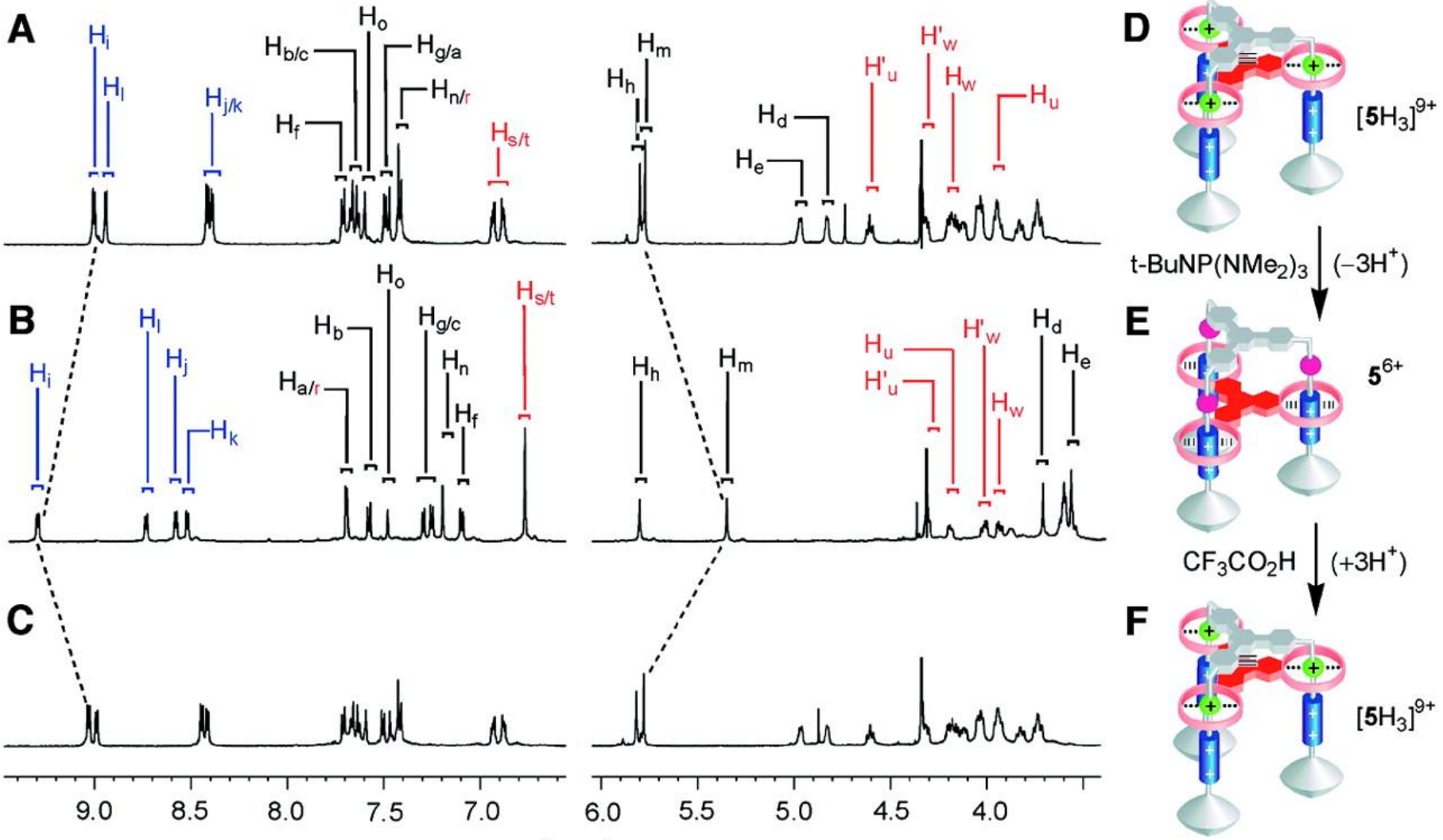
Tertiary amine

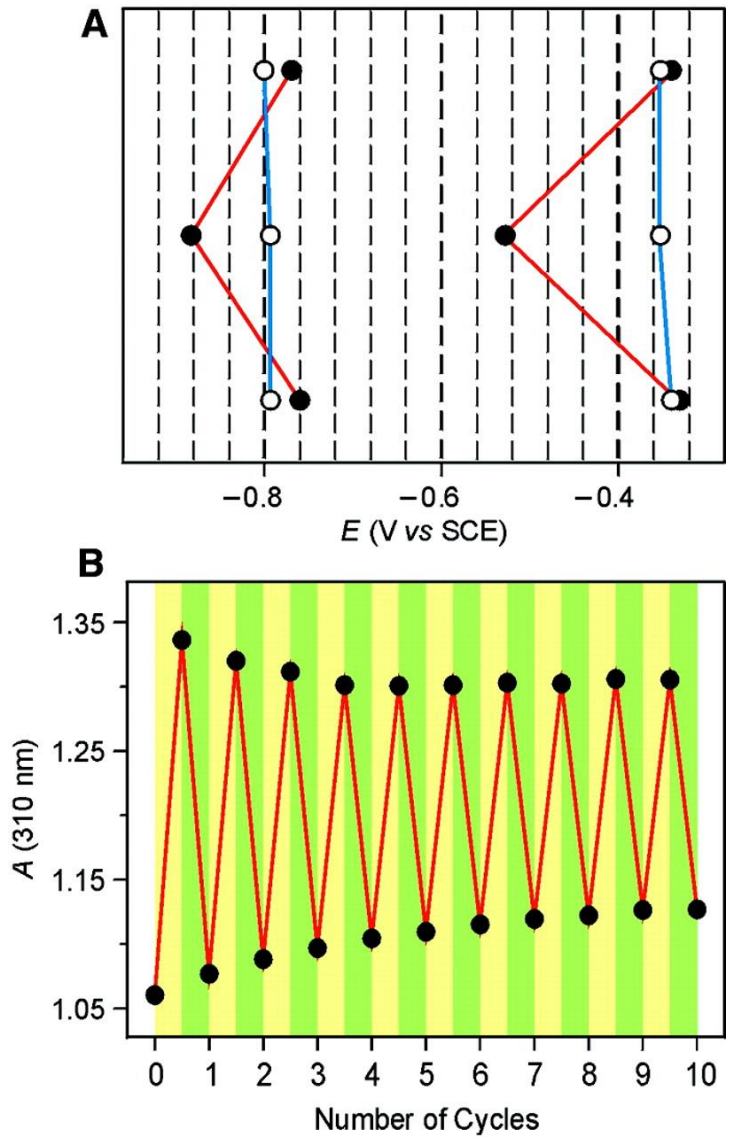
Weak C–H····O

π–π




2
i
ii
6 PF6−
[4H3][PF6]6
[5H3]
i
ii
9 PF6−
[6H3][PF6]9


A
B
C
D
E
F
H_i
H_l
$H_{j/k}$
$H_{b/c}$
H_o
$H_{g/a}$
$H_{n/r}$
H_f
$H_{s/t}$
H_m
H_h
H_d
H_e
H'_u
H'_w
H_w
H_u
H_j
H_k
H_b
$H_{a/r}$
$H_{g/c}$
H_n
9.0
8.5
8.0
7.5
7.0
6.0
5.5
5.0
4.5
4.0
(ppm)
$[\mathbf{5}H_3]^{9+}$
t-BuNP(NMe$_2$)$_3$
($-3H^+$)
$\mathbf{5}^{6+}$
CF_3CO_2H
($+3H^+$)


A
−0.8
−0.6
−0.4
E (V vs SCE)
B
1.35
1.25
1.15
1.05
A (310 nm)
0
1
2
3
4
5
6
7
8
9
10
Number of Cycles

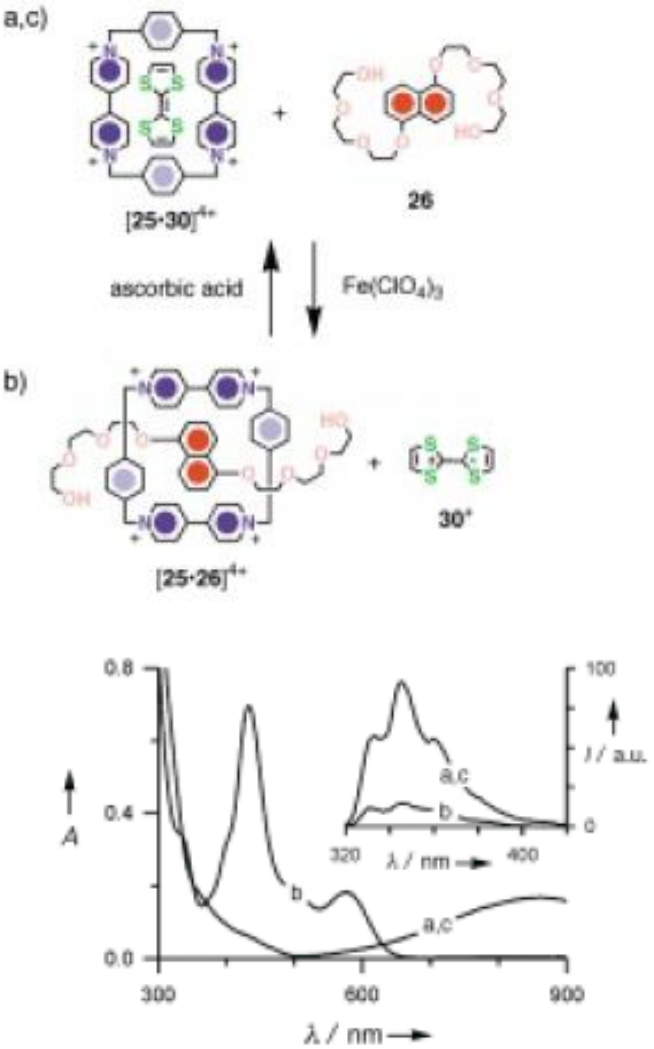
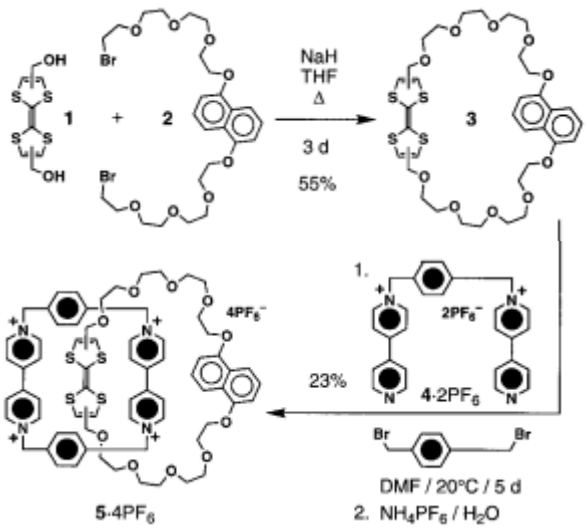
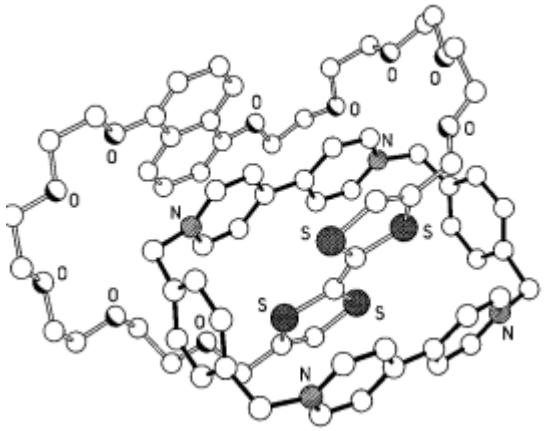
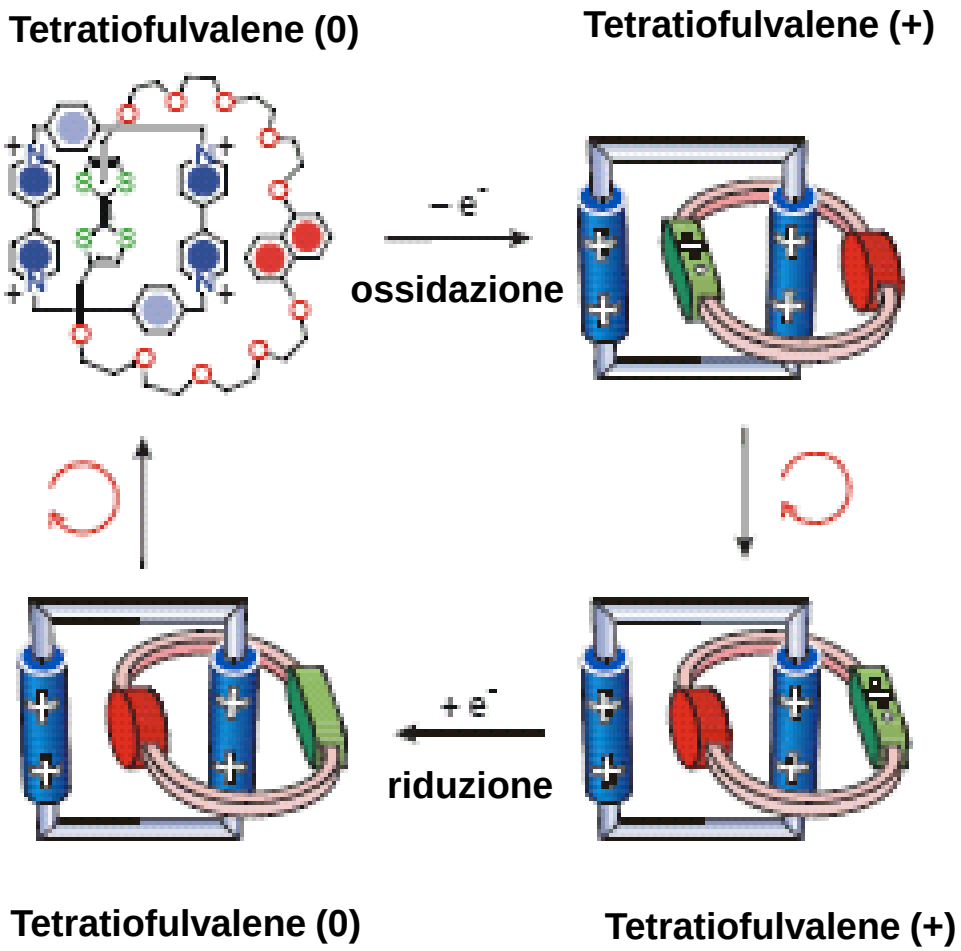


Figure 4. Top: the chemically induced interchange of guests **26** and **30** into the cavity of cyclophane $\mathbf{25}^{4+}$.[93] Bottom: absorption and (inset) fluorescence ($\lambda_{exc} = 295$ nm) spectra of a) a 5×10^{-5} M aqueous solution (298 K) of $[\mathbf{25 \cdot 30}]^{4+}$ and **26;** b) the same solution after addition of one equivalent of $Fe(ClO_4)_3$; c) solution b) after addition of one equivalent of ascorbic acid.


NaH
THF
Δ
3 d
55%
1
2
3
1.
2PF6−
4·2PF6
23%
4PF6−
5·4PF6
DMF / 20°C / 5 d
2. NH4PF6 / H2O



Input elettrochimico



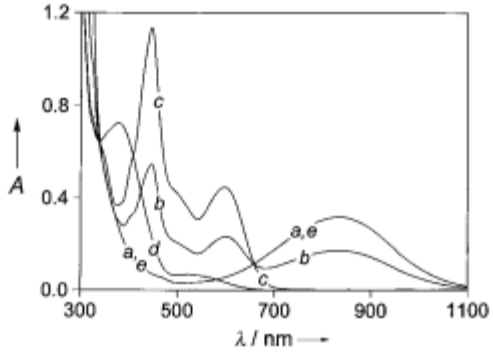
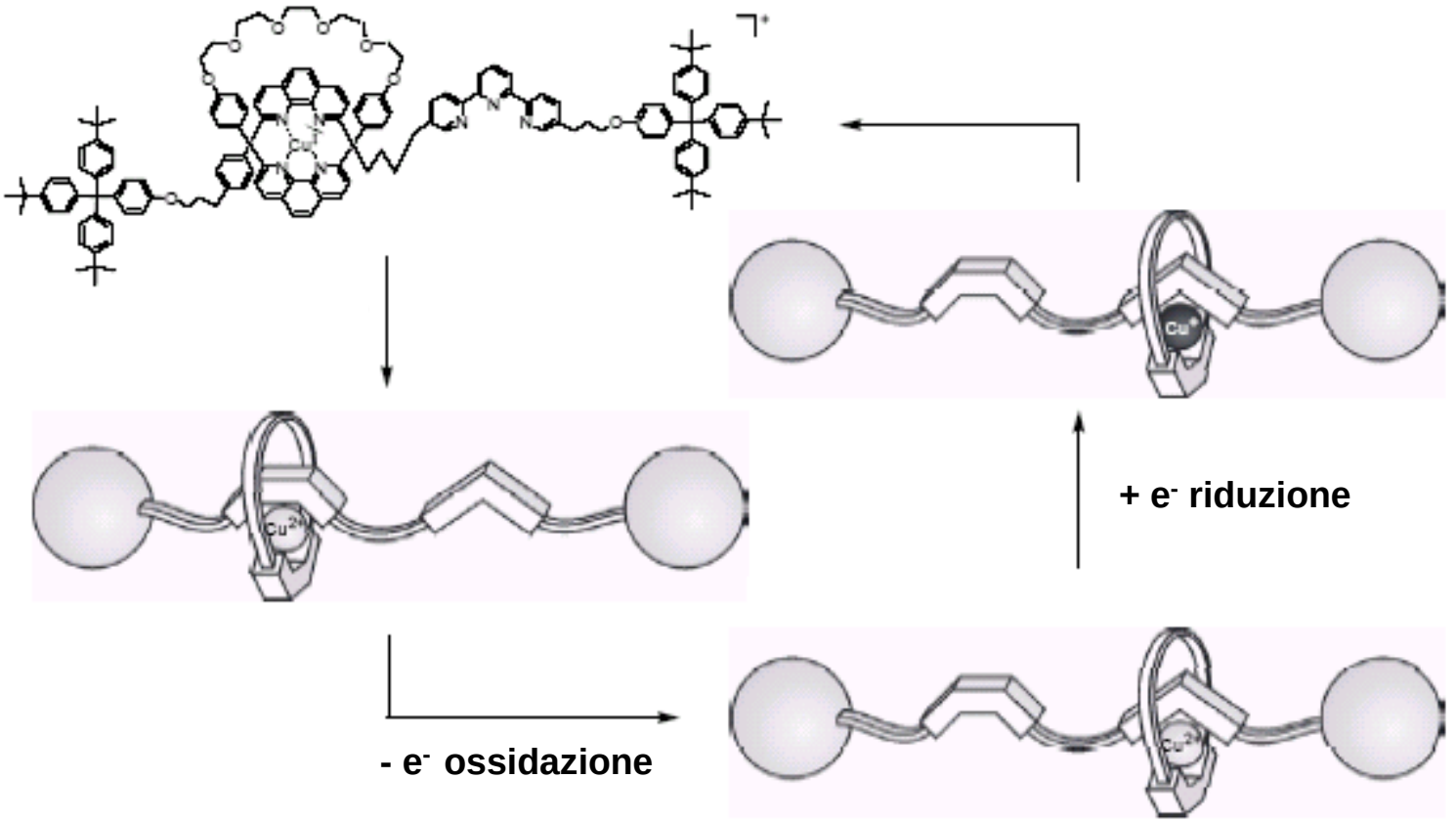
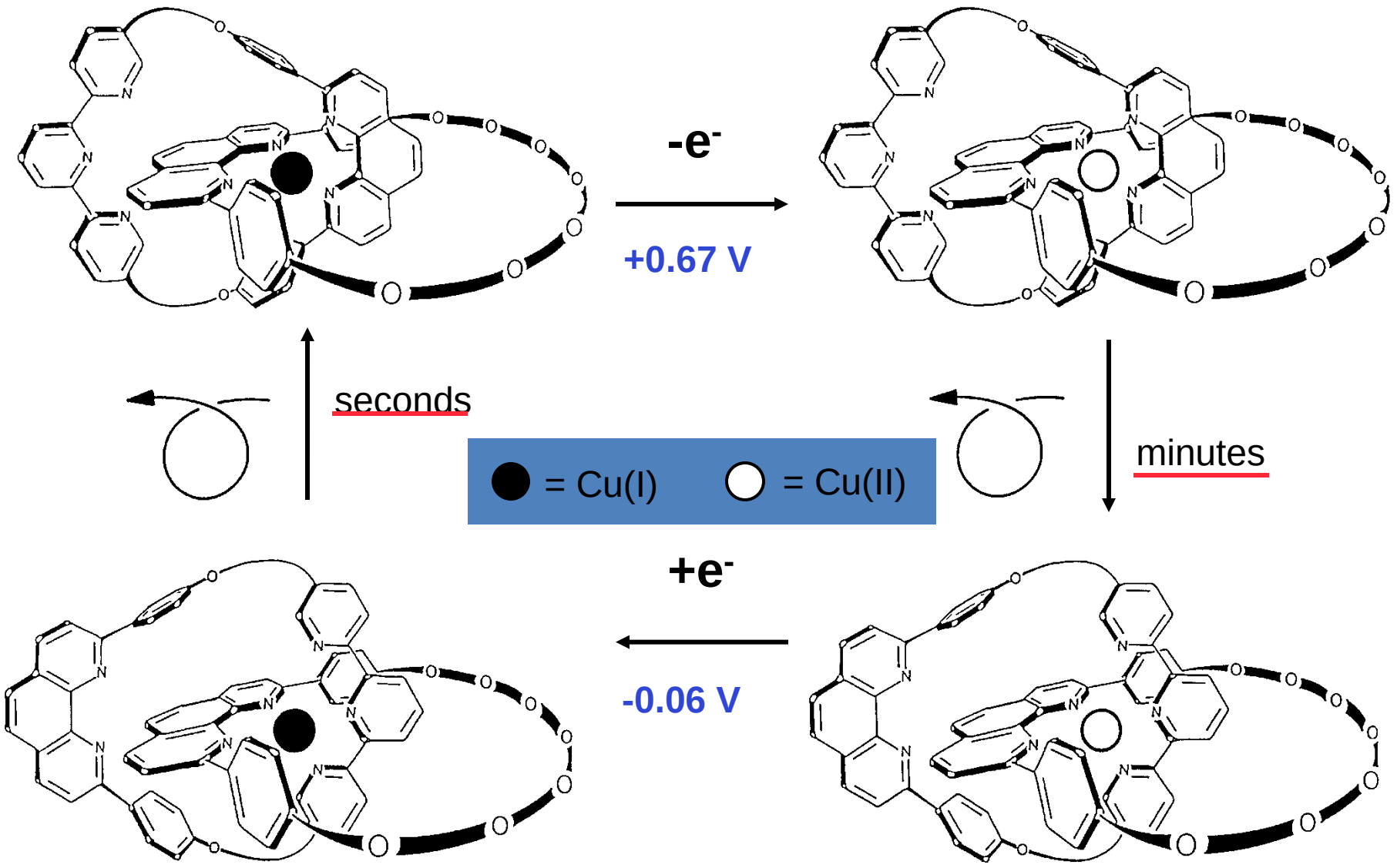


Figure 2. Absorption spectra (MeCN, 298 K) of a 9×10^{-5} M solution of [2]catenane $\mathbf{5}^{4+}$ (curve a) and of the same solution after addition of 0.4, 1.0, and 2.0 equiv of $Fe(ClO_4)_3$ (curves b–d). Addition of 2 equiv of ascorbic acid gives back the original spectrum (curve e).

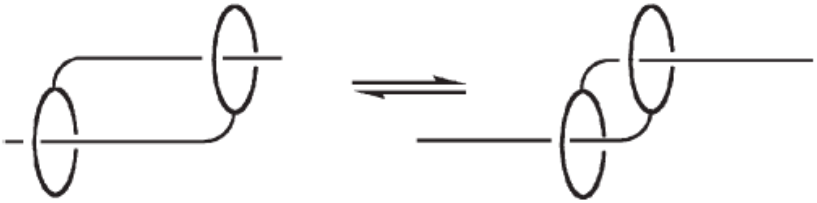
Input elettrochimico

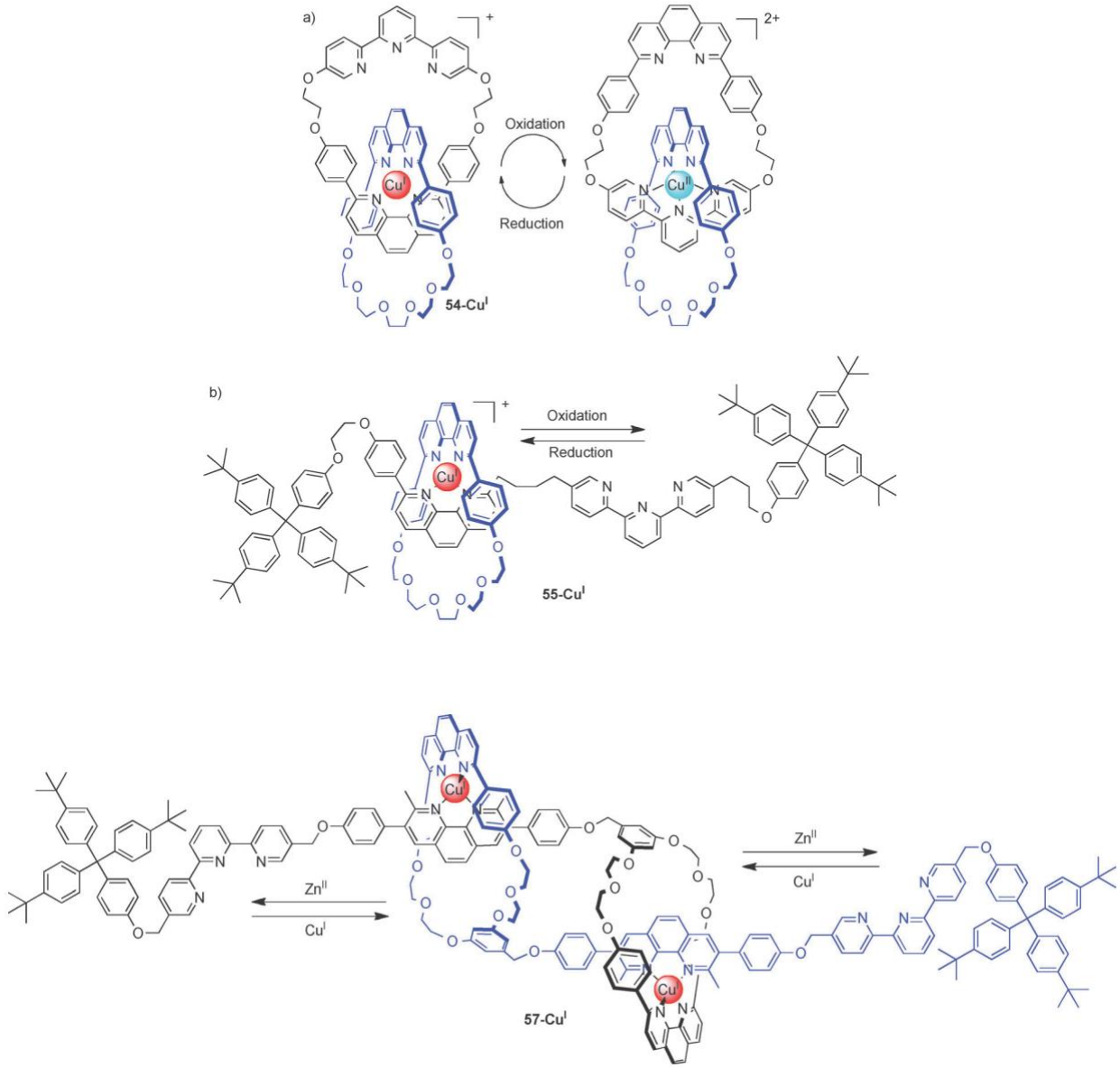


Input elettrochimico



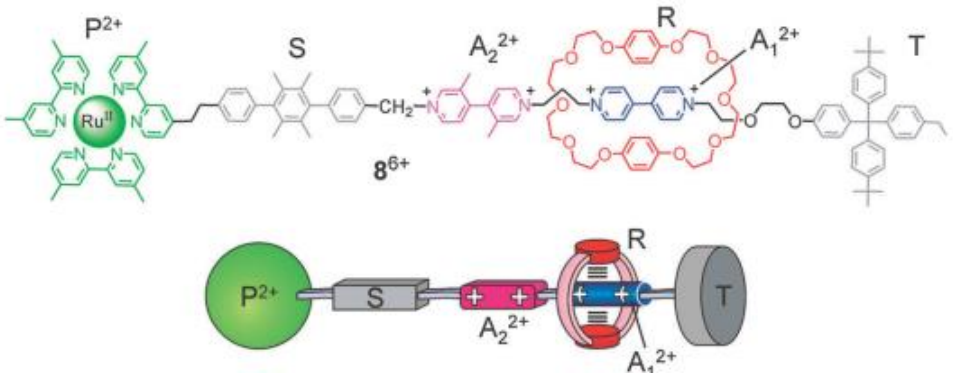
Input chimico



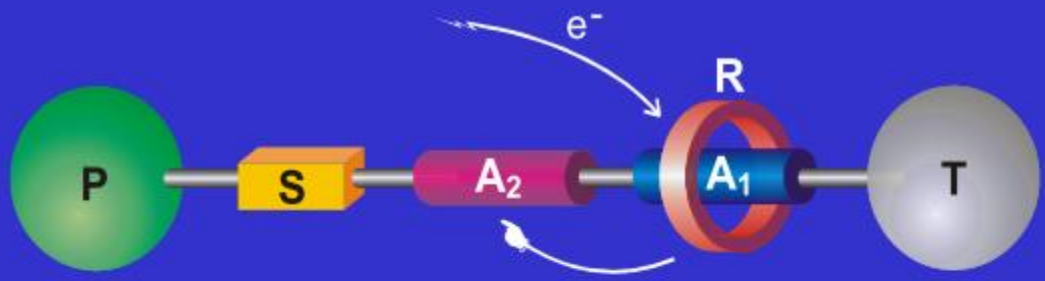

a)
Oxidation
Reduction
54-Cu^{I}
b)
Oxidation
Reduction
55-Cu^{I}
Zn^{II}
Cu^{I}
Zn^{II}
Cu^{I}
57-Cu^{I}

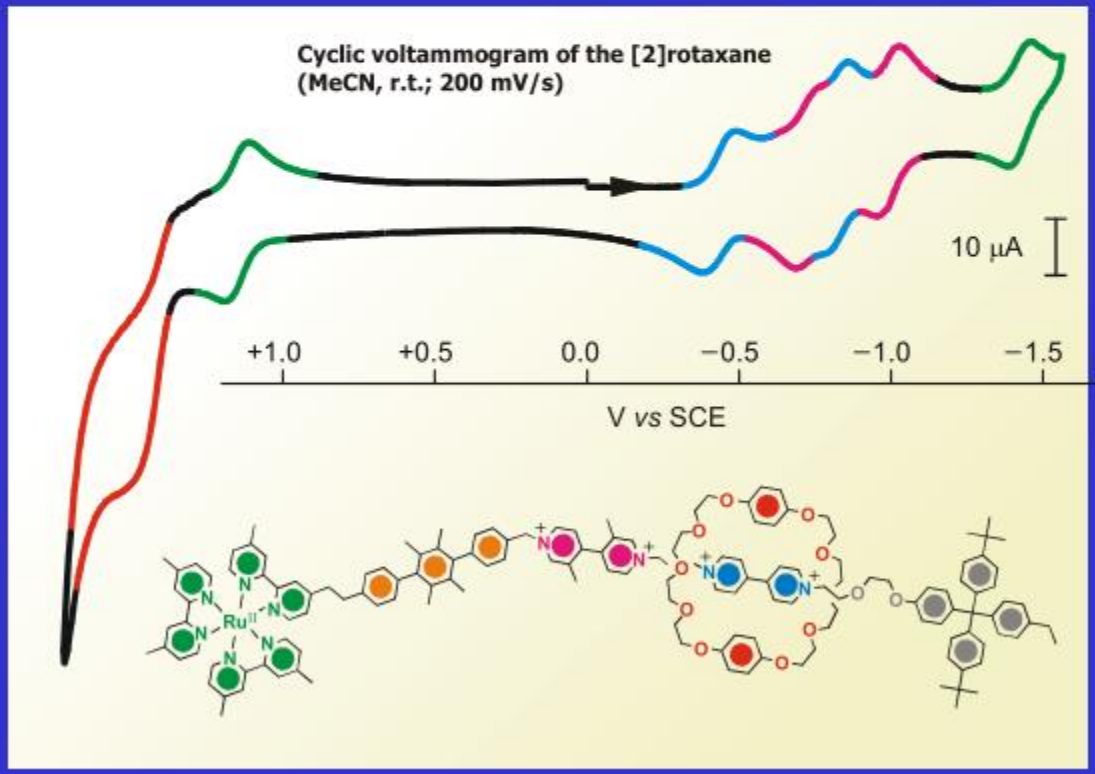
Input fotochimico

Ru(II)polypyridine complex (P^{2+})
p-terphenyl-type rigid spacer (S)
4,4'-bipyridinium (A_1^{2+})
3,3'-dimethyl-4,4'-bipyridinium (A_2^{2+})
Tetraarylmethane group (T)
Six PF_6^- counterions



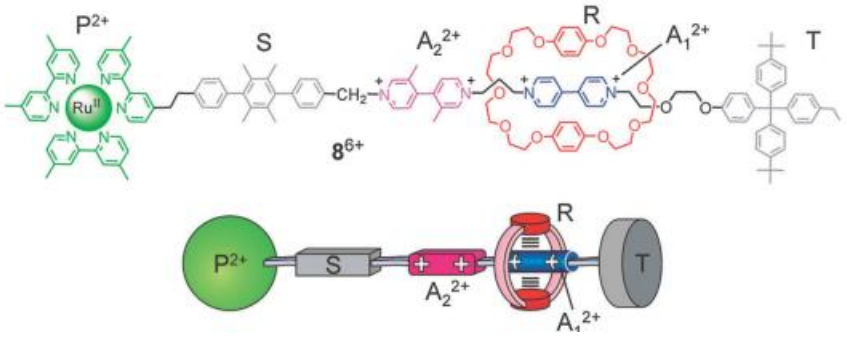
a) Redox-induced ring motion

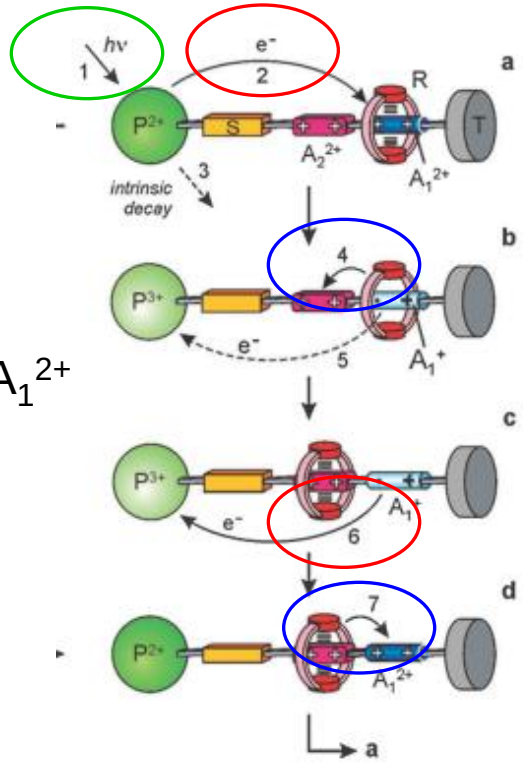




Input fotochimico

Ru(II)polypyridine complex (P^{2+})
p-terphenyl-type rigid spacer (S)
4,4'-bipyridinium (A_1^{2+})
3,3'-dimethyl-4,4'-bipyridinium (A_2^{2+})
Tetraarylmethane group (T)
Six PF_6^- counterions





1 $P^{2+} \longrightarrow {}^*P^{2+}$

2 ${}^*P^{2+} + A_1^{2+} \longrightarrow P^{3+} + A_1^{1+}$
-0.8 V -0.44 V

3 ${}^*P^{2+} \longrightarrow P^{2+}$

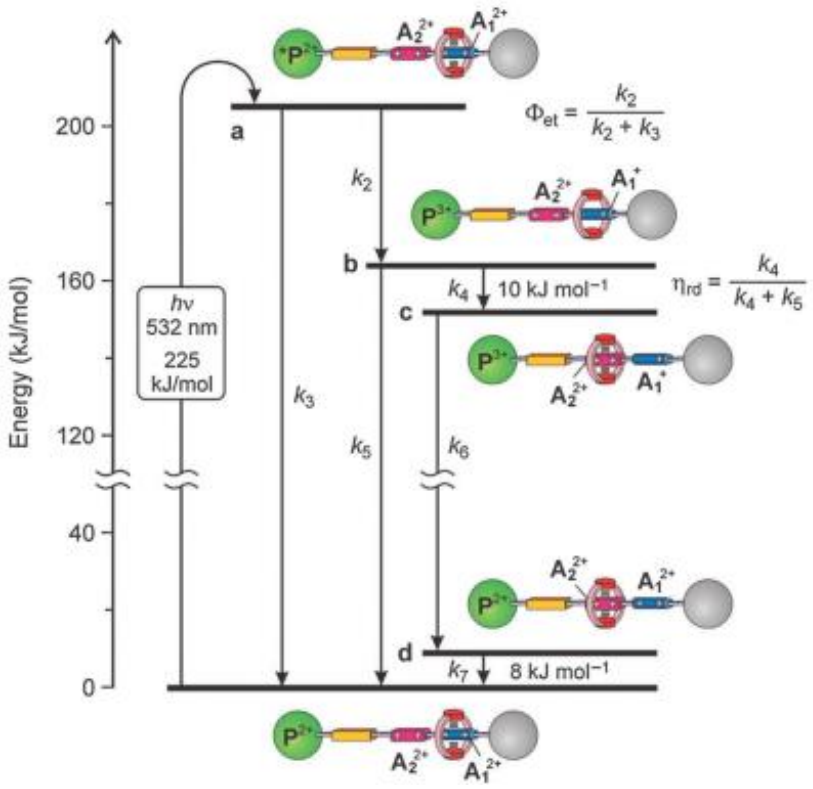
4 Shuttling (5 nm)

5 $P^{3+} + A_1^{1+} \longrightarrow {}^*P^{2+} + A_1^{2+}$

6 $P^{3+} + A_1^{1+} \longrightarrow P^{2+} + A_1^{2+}$

7 Shuttling (5 nm)

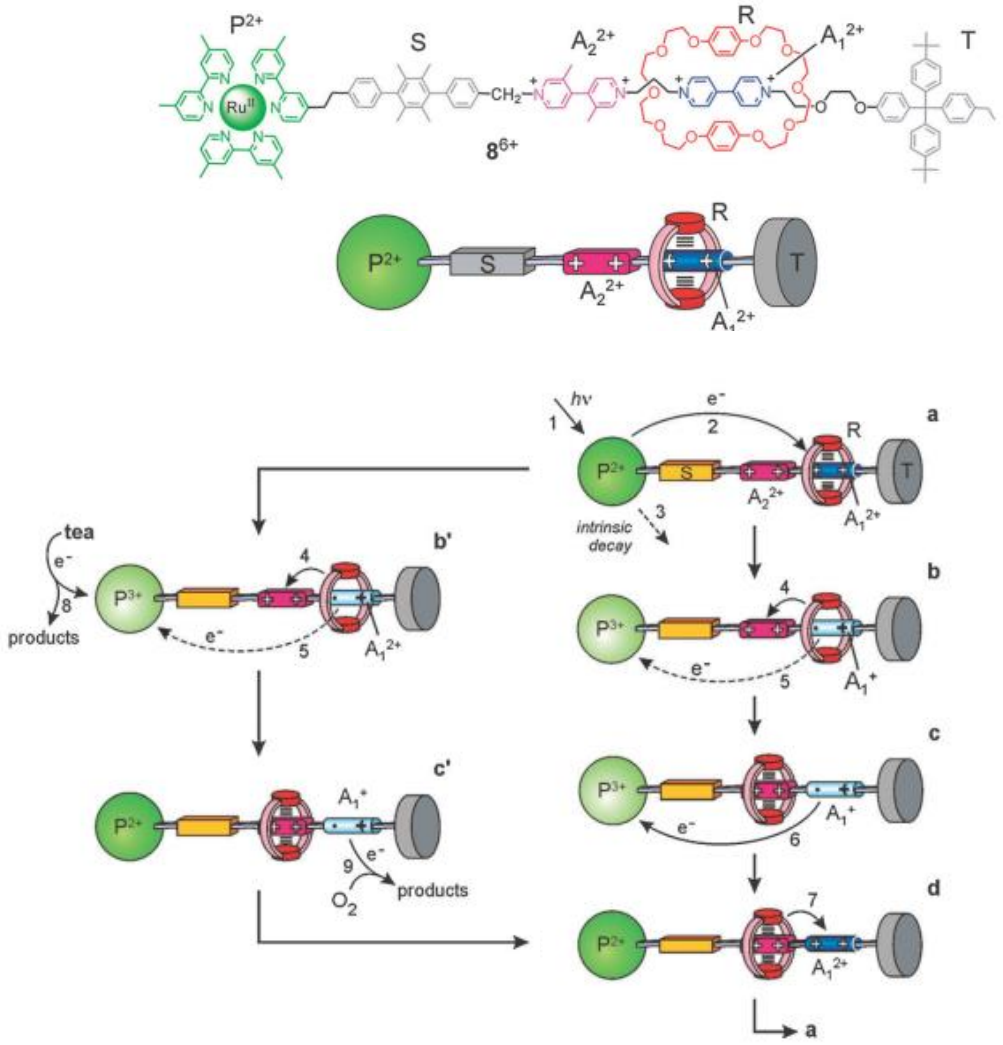
Input fotochimico



Hence, the fraction F of the excited state energy (205 kJ mol^{-1}) used for the motion of the ring amounts to *ca.* 10%, and the overall efficiency of the machine is $\eta = \Phi_{sh} \times F = 0.2\%$.

This somewhat disappointing result is compensated by the fact that the investigated system gathers together the following features: (i) it is powered by visible light (in other words, sunlight); (ii) it exhibits autonomous behaviour (*i.e.*, like natural molecular motors, it operates automatically in a constant environment as long as the energy source is available); (iii) it does not generate waste products; (iv) its operation can rely only on intramolecular processes, allowing in principle operation at the single-molecule level; (v) it can be driven at a frequency of about 1 kHz; (vi) it works in mild environmental conditions (*i.e.*, fluid solution at ambient temperature); and (vii) it is stable for at least 10^3 cycles.

Input fotochimico e chimico (agenti sacrificali TEA e O_2)

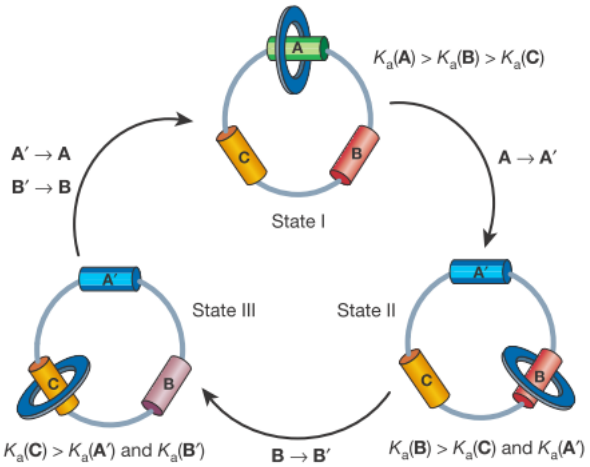


Input fotochimico

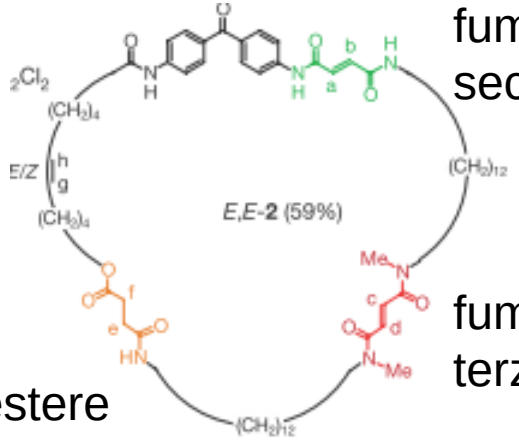
Unidirectional rotation in a mechanically interlocked molecular rotor

David A. Leigh*, Jenny K. Y. Wong*, François Dehez† & Francesco Zerbetto†

** School of Chemistry, University of Edinburgh, The King's Buildings, West Mains Road, Edinburgh EH9 3JJ, UK*
† Dipartimento di Chimica 'G. Ciamician', Universita degli Studi di Bologna, via F. Selmi 2, 40126 Bologna, Italy



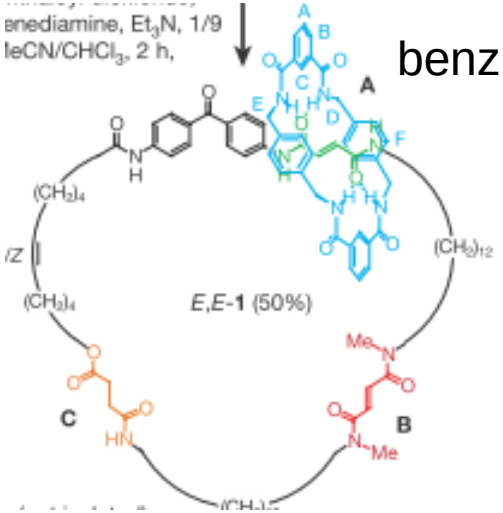
benzofenone



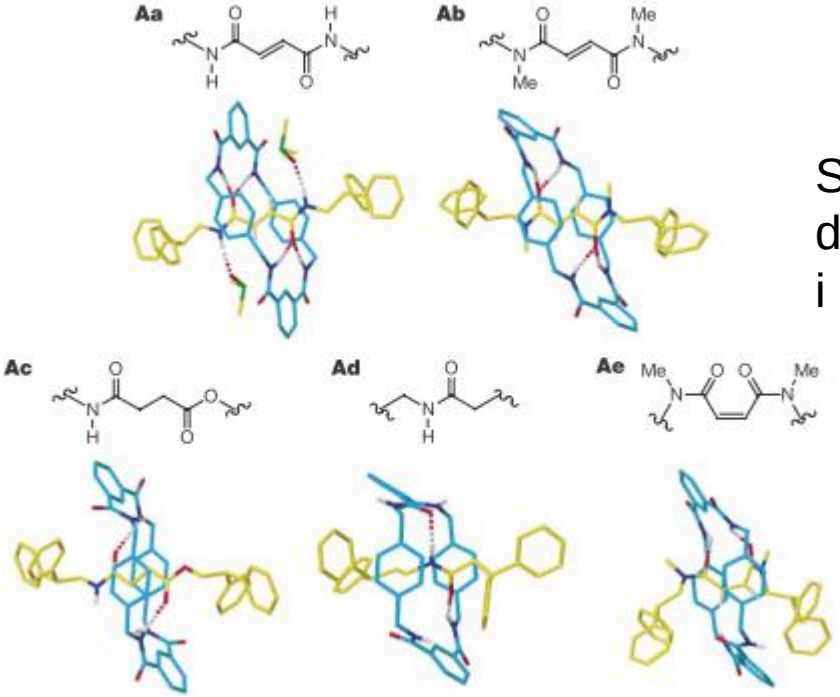
fumaramide secondaria *E*

fumaramide terziaria *E*

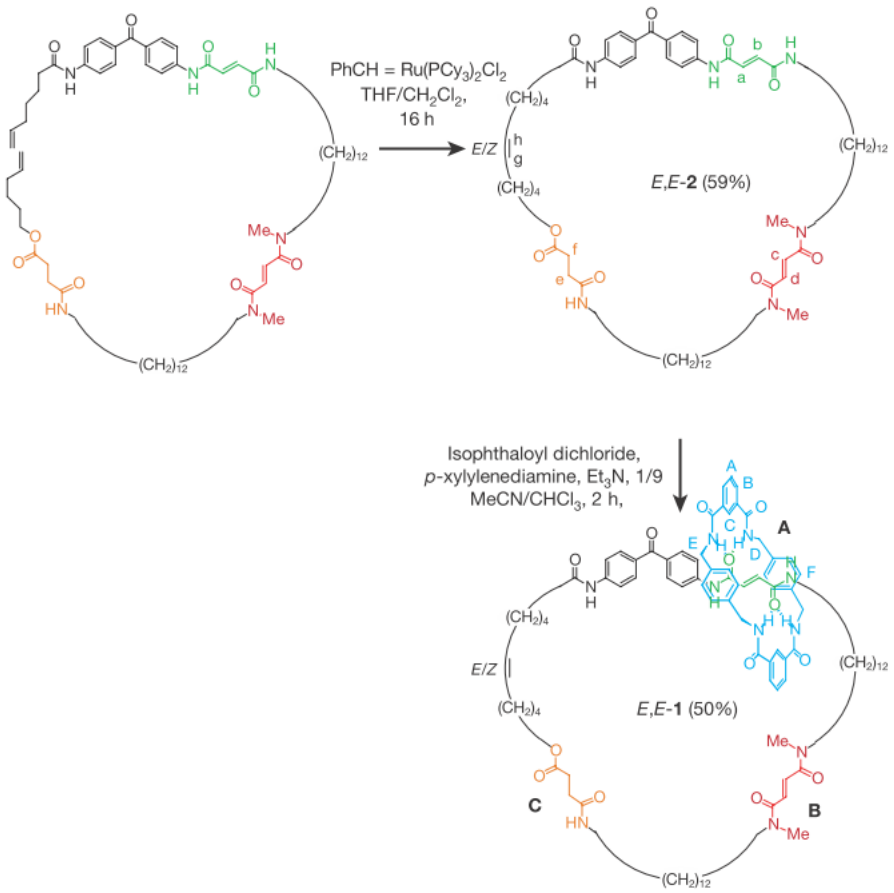
estere succinamidico

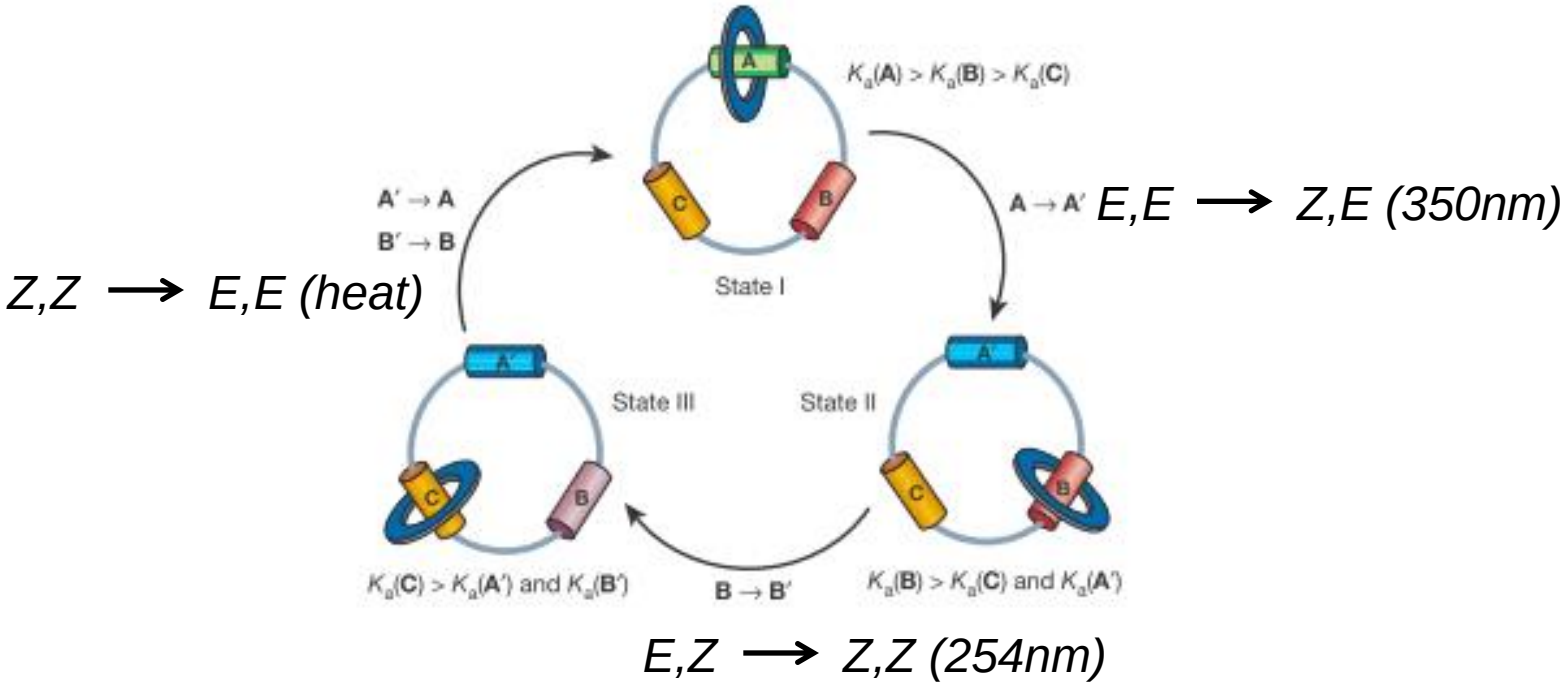


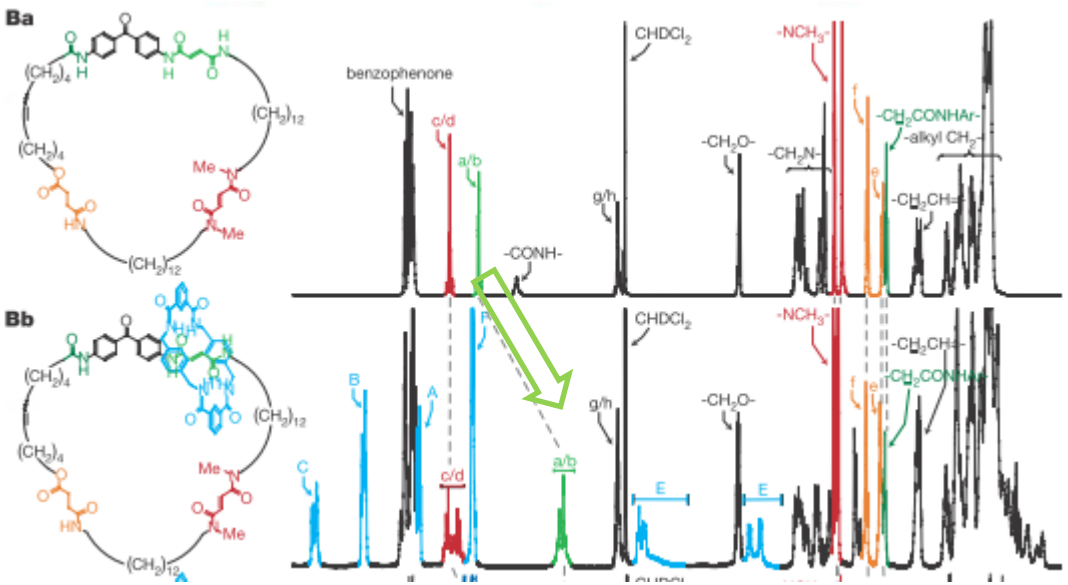
benzilamide

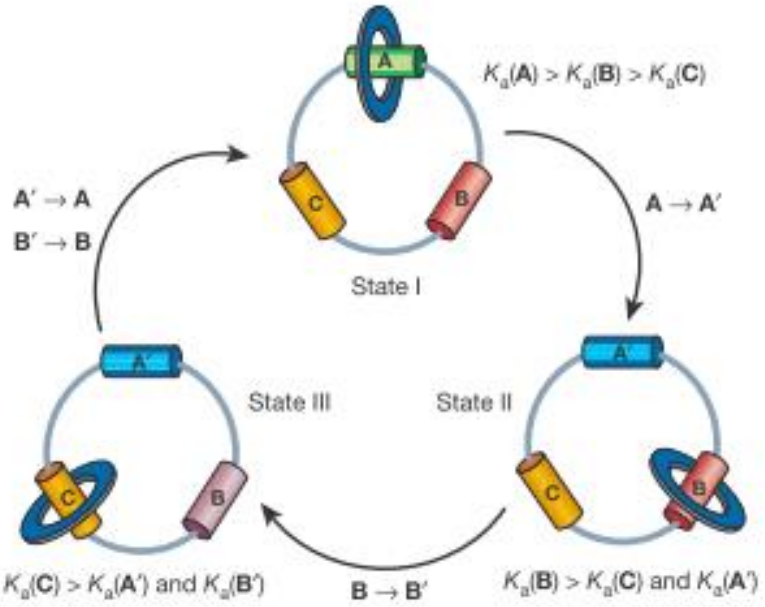


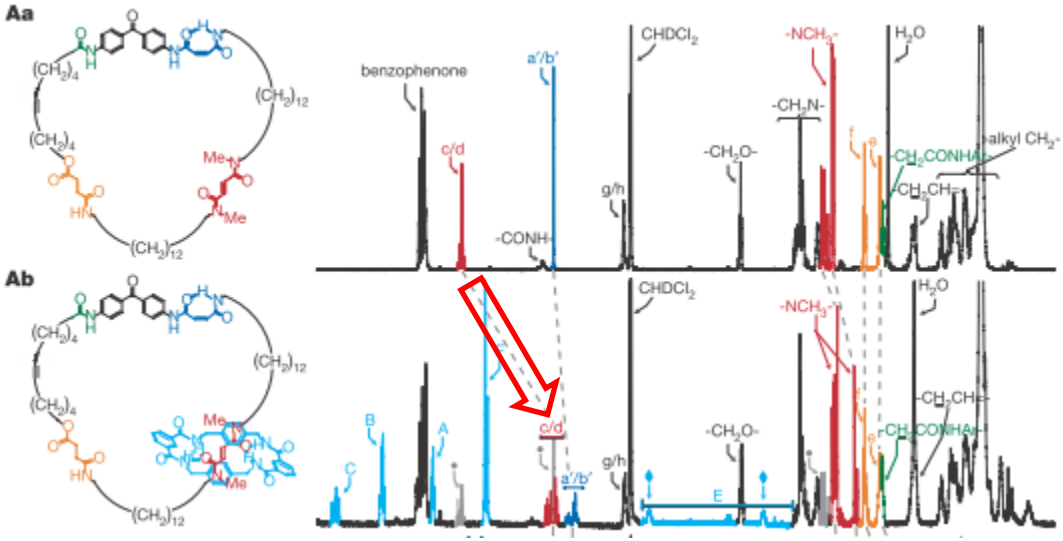
Strutture ai raggi X dei siti di binding dei modelli [2] rotaxani con indicati i legami idrogeno (in ordine di affnità)

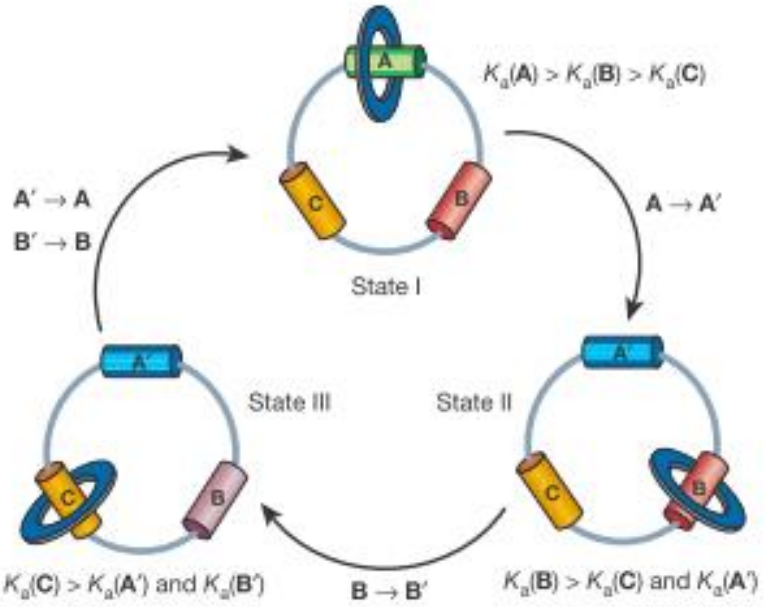

PhCH = Ru(PCy3)2Cl2
THF/CH2Cl2,
16 h
E/Z
E,E-2 (59%)
Isophthaloyl dichloride,
p-xylylenediamine, Et3N, 1/9
MeCN/CHCl3, 2 h,
E/Z
E,E-1 (50%)
A
B
C
(CH2)12
(CH2)4
Me
HN

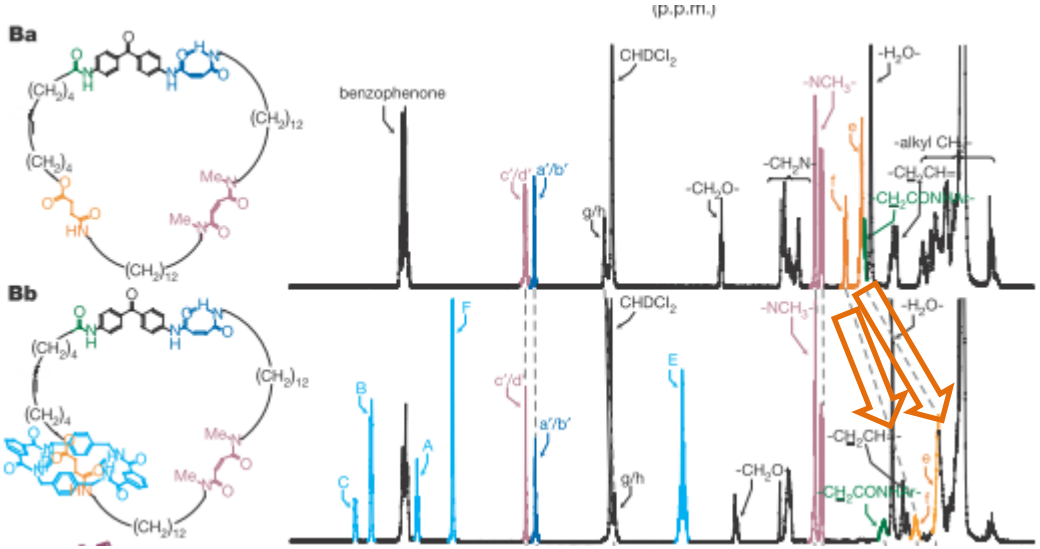

Z,Z ⟶ E,E (heat)
A′ → A
B′ → B
K_a(A) > K_a(B) > K_a(C)
State I
A → A′ E,E ⟶ Z,E (350nm)
State II
K_a(B) > K_a(C) and K_a(A′)
B → B′
State III
K_a(C) > K_a(A′) and K_a(B′)
E,Z ⟶ Z,Z (254nm)

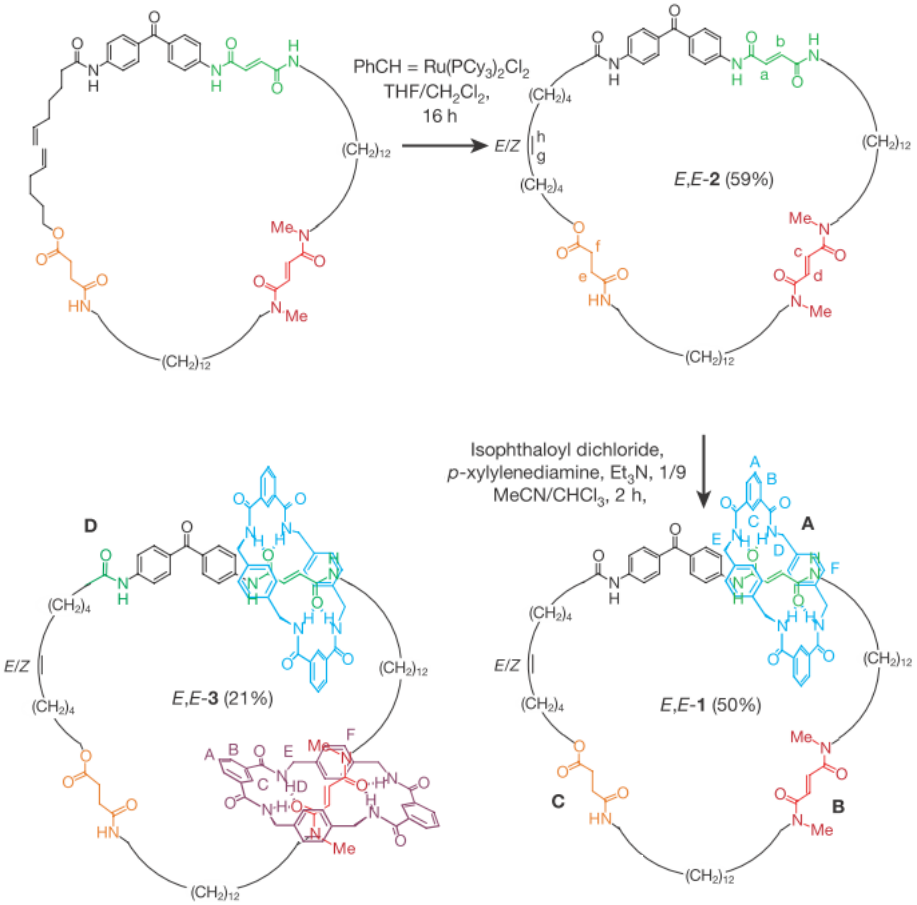

Ba
Bb
benzophenone
c/d
a/b
-CONH-
g/h
CHDCl2
-CH2O-
-NCH3-
-CH2N-
-CH2CONHAr-
-alkyl CH2-
-CH2CH-
A
B
C
E


K_a(A) > K_a(B) > K_a(C)
A → A′
B′ → B
A′ → A
B′ → B
State I
State II
State III
K_a(C) > K_a(A′) and K_a(B′)
B → B′
K_a(B) > K_a(C) and K_a(A′)


Aa
Ab
benzophenone
a′/b′
CHDCl2
-NCH3-
H2O
-CH2N-
c/d
-CH2O-
g/h
-CH2CONHAr-
-alkyl CH2-
-CH2CH2-
-CONH-
A
B
C
E

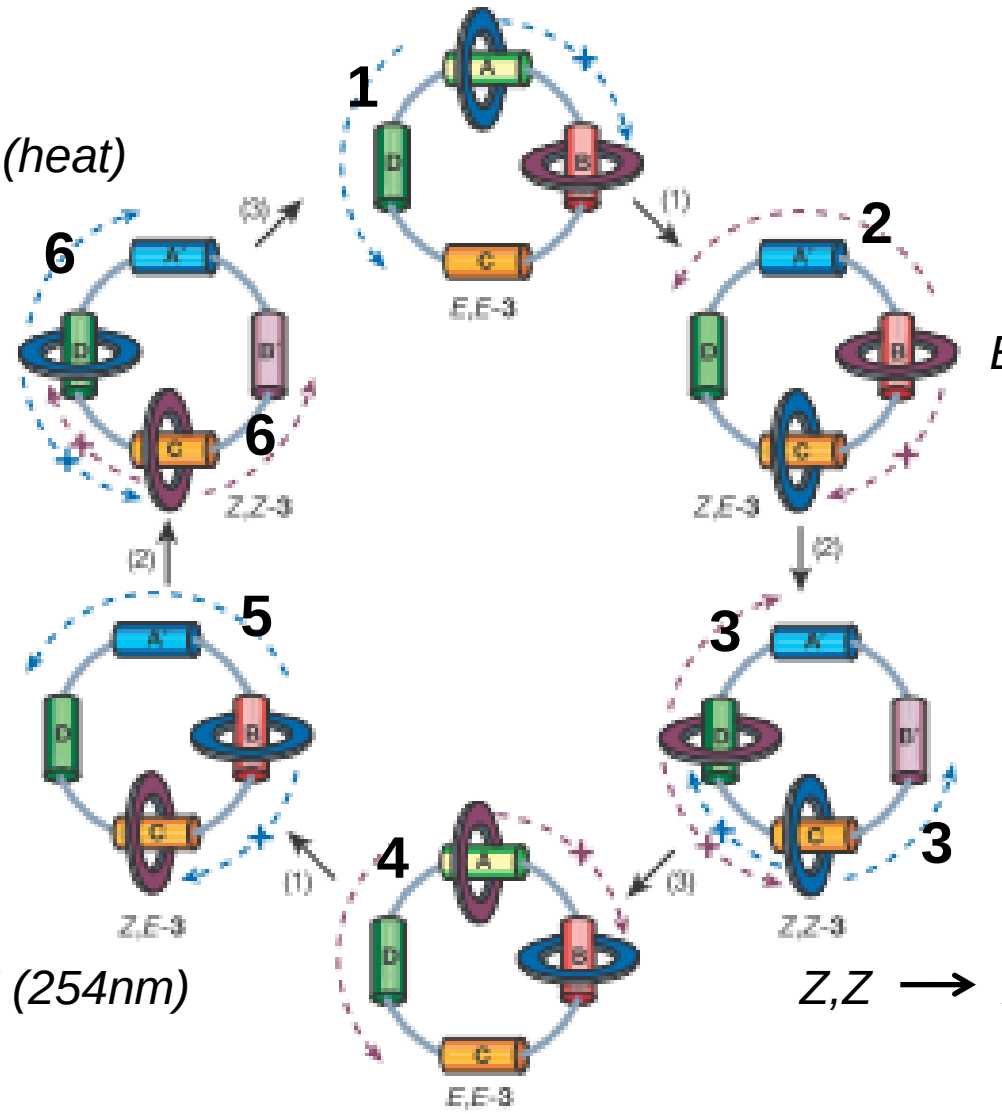

$K_a(A) > K_a(B) > K_a(C)$
A
C
B
State I
$A \rightarrow A'$
State II
$B \rightarrow B'$
State III
$A' \rightarrow A$
$B' \rightarrow B$
$K_a(B) > K_a(C)$ and $K_a(A')$
$K_a(C) > K_a(A')$ and $K_a(B')$


Ba
Bb
(p.p.m.)
benzophenone
$CHDCl_2$
-NCH_3-
-H_2O-
-CH_2O-
-CH_2N-
-alkyl CH_2-
-CH_2CH=
-CH_2CONHN-
c'/d'
a'/b'
g/h
e
f
A
B
C
E
F
(CH2)4
(CH2)12
Me
HN
O
N
H


PhCH = Ru(PCy3)2Cl2
THF/CH2Cl2,
16 h
E/Z
E,E-2 (59%)
Isophthaloyl dichloride,
p-xylylenediamine, Et3N, 1/9
MeCN/CHCl3, 2 h,
E,E-3 (21%)
E,E-1 (50%)
A
B
C
D
(CH2)12
(CH2)4
Me
HN

E,E ⟶ *Z,E (350nm)*

Z,Z ⟶ *E,E (heat)*



E,Z ⟶ *Z,Z (254nm)*

E,Z ⟶ *Z,Z (254nm)*

Z,Z ⟶ *E,E (heat)*

E,E ⟶ *Z,E (350nm)*