



# Adhesive Selection & Characterization

*What You Don't Know Can Kill  
You*

Blain Olbert

University of Arizona

Steward Observatory Mirror Lab

# Objective

- ★ Present a general overview of adhesive bonding issues to raise the designer's "awareness level."
  - ★ Strength & longevity of polymer adhesive bonds is the focus topic.
- ★ Discuss recent findings that show how conventional design philosophy fails.
- ★ Describe test facilities at UASO-ML for adhesive bond characterization.

# Scope

- ★ The presentation is aimed at practicing designers with some understanding of damage micromechanics, polymer chemistry & polymer physics.

# Scope limitations

- ☀ This is not a detailed recipe for selection & characterization of adhesive.
  - No such recipe exists!

# Adhesive fundamentals

- ★ The basic function of adhesive is to join parts.
  - Adhesives offer alternatives to conventional joining methods such as soldering, welding & bolting.
  - Some materials need adhesively bonded joints for optimum or adequate performance.
    - Prime example: joining of notch sensitive “brittle” materials that are intolerant of bolt holes (e.g., engineering composites, ceramics, glass).

# Adhesive fundamentals

- ★ Adhesives offer a convenient or cost-effective way to join parts.
  - But don't assume adhesives are a cure-all solution for every critical joining problem.
  - When long-term reliability is paramount, bolting may be the best option after the full magnitude of the problem is understood. This may hold true even though bolted connections may be difficult & expensive to implement and may significantly reduce the ultimate load bearing capacity of a structure.

# Adhesive fundamentals

- ★ Some common adhesive material classes:
  - Polymers
  - Metals (solders)
  - Inorganic glass powders (“frits”, solder glasses)
- ★ Polymer adhesives are the most commonly used bonding agents.
  - Epoxies (contain the epoxide functional group)
  - Acrylics (contain the aryl functional group)
  - Urethanes (contain the urethane functional group)
  - Silicones (contain the siloxane functional group)

# Adhesive fundamentals

- ✱ Adhesive properties can show broad variation between & within material classes.
  - The primary information source for material property data is the technical data sheet from the manufacturer.
  - Scientific & engineering literature may contain a wealth of additional information on certain adhesives used in high-risk applications (aerospace, defense). Do a literature search.
- ✱ Cost of adhesive varies widely.
  - Some of the better-characterized engineering adhesives are relatively inexpensive.



# Adhesive fundamentals

- ★ Polymer adhesive technology advances rapidly.
  - Survey current offerings before you pick adhesive for a new application or review an old application.
  - Do not assume “rules” forever hold true, such as “epoxies are brittle.” New nano processing technologies, such as polymer chain engineering at the molecular level, are re-writing the rules.
  - Even polymer experts do not fully understand the mechanical properties of adhesives. Designer beware!

# Chemistry fundamentals

## ☀ Polymer structure

- ☀ The fundamental molecular building blocks of a polymer are monomers. Polymer means “many mers.”
- ☀ Monomers generally consist of several atoms that are chemically bonded to one another.
- ☀ One or more atoms of the monomer are chemically reactive and will bond to other monomer atoms under certain conditions.

# Chemistry fundamentals

## ☀ Polymer structure

- ☀ The characteristic 3D molecular chain structure of a polymer is built-up as the monomers link together in a controlled chemical reaction.
- ☀ The degree of polymerization of monomers (i.e., molecule chain length between crosslinks & number of crosslinks between chains) impacts the mechanical properties.

# Chemistry fundamentals

## ☀ Polymer structure

- ☀ The polymerization chemistry of commercial adhesives is usually complex.
  - A initiator (cross-linking agent and/or a catalyst) is purposely introduced to trigger the polymerization reaction of the “base” material & “cure” the adhesive. There are two popular cure systems:
    - “two-part” system: the initiator is mixed into the base just before use.
    - “one-part” system: the initiator is water vapor that’s introduced to base by exposure to ambient air.

# Chemistry fundamentals

## ☀ Polymer structure

- ☀ Adhesive base material is usually short chain molecules that were prepared by controlled monomer reactions.
  - Since chain properties (overall length, entanglement tendency, etc.) control viscosity of the base material, the same adhesive may come in a range of viscosities.

# Chemistry fundamentals

## ☀ Curing characteristics

- One-part systems cure by diffusion of initiator thru the bond. Years may be needed for a thick bond to fully cure!
- Two-part systems cure uniformly throughout the bulk and so are preferred for thick section bonds.
- Both one-part & two-parts systems nearly always emit reaction products
  - Reaction products may be corrosive (e.g., acetic acid) or flammable (e.g., methanol). Read the data sheet!
  - If the reaction products are not removed by diffusion, the curing reaction will slow down & possibly reverse itself (depolymerization or “reversion”).

# Chemistry fundamentals

## ☀ Bulk properties

- Bulk physical properties (e.g., mechanical properties, thermal properties, etc.) depend on monomer properties & chain structure.
- Bulk properties, by definition, are not influenced by the chemistry & physics of interfaces.

# Chemistry fundamentals

## ☀ Interface properties

- ☀ Adhesion is an interface property that influences the fundamental bond strength between substrate & adhesive.
- ☀ Adhesion is controlled by surface chemistry. During cure, full adhesion properties typically develop hours or days after bulk properties.
- ☀ Surface chemistry is notoriously susceptible to trace quantities of molecular compounds. Pay attention to purity of cleaning chemicals & solvents; chemical compatibility of storage, mixing & applicator materials.



# Chemistry fundamentals

## ☀ Interface properties

- ☀ Adhesion, like any property that depends on surface or interface chemistry & physics, can possess bewildering complexity.
- ☀ Commercial adhesives often contain small quantities of molecular compounds to promote adhesion.
- ☀ Some adhesive formulations, when used with certain substrates, require adhesion promoters (primers) - the property data sheet should tell you this.

# Chemistry fundamentals

## ☀ Interface properties

- ☀ Precisely why adhesion promoters work is rarely understood - formulation of promoters is a quasi-empirical, black art. Their use must be tested in your specific application.
  - Temperature & humidity generally play important roles in primer activation, and must be monitored & controlled if possible.
- ☀ Surface cleanliness is critical to adhesion. A consistent surface preparation technique, executed by process procedures, is essential.

# Mechanics fundamentals

## ★ Strength

- ★ Strength is defined here as stress at rupture.
- ★ Rupture is due to propagation of damage through the bond. When damage accumulates to a critical level, the bond ruptures.
- ★ Because strength depends on the rupture condition, its fate is tied to the details of damage evolution.

# Mechanics fundamentals

## ★ Strength

- ★ The rupture condition for a given material cannot be predicted from the laws of mechanics alone. Experimental data are needed (e.g., toughness – a material property, flaw size – usually a specimen property that can vary significantly).

# Mechanics fundamentals

## ☀ Damage

- ☀ “Damage” as used here is an umbrella term that includes many different specific mechanisms such as:
  - Initiation & growth of quasi-homogeneously distributed damage such as microcracks or microcavities.
  - Initiation & growth of a discrete crack.
- ☀ Damage initiation begins at the atomic level and so is fundamentally governed by quantum physics, not classical physics.

# Mechanics fundamentals

## ☀ Damage

- ☀ When the spatial extent of damage is sufficiently large & homogeneous, classical field equations & the laws of continuum mechanics can be used to characterize damage.
- ☀ Classical fracture mechanics is a special damage case where evolution of a discrete, well-defined crack is described by the laws of continuum mechanics & constitutive laws for sub-critical crack growth (e.g., chemically-assisted crack growth).

# Mechanics fundamentals

## ☀ Damage

- ☀ Critical damage usually appears first at corner or edge fillets due to localized stress concentrations plus direct exposure to corrosive species in the environment.

# Mechanics fundamentals

## ☀ Damage

- ☀ Fractography (analysis of fracture surfaces) often gives valuable insight into the damage evolution process. Preserve your test specimens.
- ☀ Accurate & full interpretation of fracture surfaces is an expert, specialized skill usually practiced by consultants. A few textbooks discuss fractography.



# Mechanics fundamentals

## ☀ Damage

- ☀ The precise way that critical damage accumulates in a given material varies from specimen to specimen even with specimens of the same geometry under the same loading condition.
  - Hence strength, which depends on the critical damage level, is not a true material constant.
    - The apparent exception of note is the (ultimate) strength of ductile engineering alloys with or without notches. But the constant strength approximation breaks down in extreme cases where a notch is present in a very large specimen.

# Mechanics fundamentals

## ★ Basic damage modes

- ★ Cohesive failure results from critical damage in the bulk region of the bond.
- ★ Adhesive failure results from critical damage along the interface between substrate & adhesive.
- ★ We implicitly assume the substrate does not fail (cohesive or adhesive).
  - ★ If the substrate fails, then the strongest possible bond has been achieved.

# Mechanics fundamentals

## ☀ Cohesive failure

- ☀ Always associated with the strongest possible bond.
- ☀ Goal is to achieve cohesive failure every time by proper adhesive-primer selection & development of good bond preparation technique.

# Mechanics fundamentals

## ☀ Adhesive failure

- ☀ Usually occurs from poor bonding of adhesive to substrate. Strength depends on interface properties & interface mechanics - not bulk properties per se.
- ☀ Strength is very sensitive to surface preparation method due to the importance of interface chemistry.

# Mechanics fundamentals

## ☀ Adhesive failure

- Strength may be increased by changing surface preparation method and by using a proper adhesion promoter or “primer.”
  - Most primers are fussy & substrate specific – they must be used with care & consistency.
- Damage predictions based on bulk material behavior are usually erroneous.
  - Interface properties may have no relation to measured bulk properties!
  - Do not confuse bulk strength with adhesive bond strength. Sometimes both values are quoted on data sheets.

# Design guidelines

- ☀ The “golden rule” for design with adhesive:

- ☀ *Never conceptualize a design that requires a critical adhesive bond without addressing the bond problem early on.*

# Design guidelines

## ☀ General recommendations:

- ☀ Always use pedigreed engineering adhesive.
- ☀ Do not use adhesive if very-high, quantifiable bond reliability is required for a long time ( $>10$  years) unless hard data or past experience prove that the specific adhesive meets the requirements under all conditions.

# Design guidelines

## ★ Problem background work

- ★ In addition to detailed stress analysis, obtain a *complete* description of the problem such as:
  - Operating environment details
    - Temperature, humidity, chemical species present.
    - Extreme environmental conditions & the time spent at those conditions.
  - Substrate particulars
    - Standard metallic alloys: specify alloy designation (AISI, SAE), heat treatment condition & other form/physical conditions.
    - Glasses & ceramics: specify chemistry, physical properties, anneal condition.
    - Polymers: specify chemistry & physical properties.



# Design guidelines

## ☀ Problem background work

- ☀ Pay close attention to the operational temperature range.
  - Polymers become brittle & show remarkably different properties when taken below their glass transition temperature ( $T_g$ ). Find  $T_g$  for your adhesive!
  - Polymers creep at widely different rates. If bond dimensional stability is important, be aware of creep.
    - Creep rate usually increases exponentially with temperature and sometimes applied stress. Creep rate also increases with humidity level for most adhesives.

# Design guidelines

## ☀ Problem background work

### ☀ Bond longevity

- What is the reliability requirement?
  - <1% failure probability over life?
- What is the average lifetime requirement?
  - 1 yr? 50 yr?

# Design guidelines

## ☀ Collate data

- ☀ Compare all of the forgoing requirements to published adhesive property data.
  - Does the adhesive meet all requirements under all operating conditions?
  - What vital adhesive property information is missing?
  - If the known property database does not address all requirements, you have a potentially serious problem.
    - It's time to get help & put on the lab coat, or find a different adhesive that's better characterized.

# Processing guidelines

## ☀ Before you make a bond:

- ☀ Collect all existing processing procedures.
  - Scour the technical data sheet for processing information (i.e., the details about how the bonds were prepared).
  - Request processing information from the technical representative.
  - Search the literature.
- ☀ If there is no processing procedure, you must prepare one to raise the odds of making consistent bonds & to provide a future reference benchmark.

# Processing guidelines

## ☀ Before you make a bond

- ☀ Develop a process control punchlist. List would callout items such as:
  - Critical data to log (adhesive batch numbers, batch constituent weights, date, time, temperature, humidity, etc., and name of worker).
  - Vacuum outgass adhesive to reduce entrapped air bubbles & dissolved gasses that can precipitate as bubbles when the adhesive cures. Bubble defects decrease strength.
  - Witness sample preparation scheme. Always make witness samples and keep leftover batch!

# Design pitfalls

- ✱ Do not assume that the strength factor-of-safety (FOS) design strategy applies to adhesive.
  - FOS was developed & proven for ductile engineering alloys in static, non-corrosive, non-creep conditions.
  - A static (time-independent) FOS cannot be applied unless the adhesive is below its damage threshold.
    - Examples of materials with no damage threshold: superalloys used at high temperature; silicate glass (controversial subject)

# Design pitfalls

## ☀ Stress analysis

- ☀ Analyze the exact bond geometry in detail, especially stresses near corners & fillets. Critical damage will nearly always initiate in the region of highest stress.
- ☀ Analysis must include all driving forces such as:
  - ☀ Residual stresses caused by bulk expansion/contraction of adhesive during curing.
  - ☀ Differential thermal stresses.

# Design pitfalls

## ☀ Fracture mechanics & strength

- Nearly all polymers (including elastomers) behave as brittle materials, and so linear elastic fracture mechanics can be used to estimate strength.
  - This implies that a constant toughness serves to characterize energy dissipation in the fracture process zone. Note this implicitly assumes that the fracture process zone is small compared to specimen dimensions.
- In bonds, the substrate constrains the free-development of damage compared to bulk. Interface mechanics become important.
- Fracture mechanics estimates of bond strength based on cracks in bulk material are still useful for benchmark comparisons.



# Design pitfalls

- ★ The implicit assumption that materials exist which meet every imagined requirement is poor design practice.
  - ✱ Keep the design envelope within easy reach of existing materials. Use your mechanical ingenuity to overcome material limitations.

# Facts of life

- ☀ Lifetime is the fundamental design metric. Nothing physical lasts forever. No bond lasts forever.
  - Lifetime depends on strain level & environment (temperature, humidity, chemical species present, etc..)
  - Probabilistic lifetime is the most useful design parameter but is rarely ever addressed explicitly.

# Facts of life

- ★ Most time-delayed bond failures observed in the field are adhesive failures.
  - This can happen even though short-time lab tests always showed cohesive failure!
  - Such is the insidious nature of interface damage processes.

# Facts of life

- ☀ Damage may propagate in time through the bulk adhesive, along the adhesive-substrate interface, or a combination of bulk & interface (mixed failure).
- ☀ Sometimes the substrate suffers critical damage over time by corrosion or weathering that undermines the adhesive bond.
  - ☛ Propagating damage may therefore change mode from cohesive to adhesive over time.

# Facts of life

- ★ There are no specific rules, formulas or procedures to predict adhesive bond lifetime.
  - ★ Barring quality lifetime data from the lab, field data are the best source of information.
- ★ Determination of accurate bond lifetime predictor equations requires extensive experimental testing & fitting to theoretical or empirical models.
  - ★ Such data is rarely available to the design engineer.

# Accelerated testing

- ☀ Accelerated testing is the favorite red herring of lifetime prediction novices.
- ☀ There are serious questions associated with accelerated testing:
  - Precisely what damage mechanism/s are being accelerated? Are you accelerating the mechanism that's active at nominal conditions and only that mechanism?
  - What is the appropriate acceleration variable?
    - Elevated temperature? Elevated stress? Elevated humidity? Elevated everything?

# Accelerated testing

- ☀ The challenge of accelerated testing:
  - ☀ How do you scale lifetime under accelerated conditions back to nominal conditions?
  - What is the exact functional form of the lifetime scaling equation?
    - Does life decrease exponentially with increasing stress level?
    - Does life obey time-temperature superposition?

# Accelerated testing

- ★ Unless the damage mechanism/s are understood in detail, and physical models exist that predict damage evolution with the accelerated variable, it's impossible to accurately fit data to a life predictor model.
- ★ With no fundamental understanding, the only option is quasi-empirical or purely empirical modeling.
  - But there is no clear way to validate an empirical model unless you compare lifetime measured under nominal conditions to the predicted lifetime from accelerated testing. It's usually not feasible to verify lifetime of many years – if it were, you probably wouldn't be doing accelerated testing.



# What can a designer do?

- ☀ The knowledge & lab skills needed to characterize an adhesive bond lifetime is not part of formal skill training of most designers.
  - Be absolutely certain that you must use an uncharacterized adhesive before you start an expensive, time consuming test & evaluation program.
  - Seek help if you must press on & don't know how to proceed.

# The lifetime problem signature

- ★ Two simple tests will show if longevity is a potential issue:
  - A dependence of strength on testing speed indicates that a time-dependent damage mechanism is active.
  - Failure after some time under static load is a sure-fire indication of a time-dependent damage mechanism.

# The next step

- ✦ If the forgoing simple tests show there's a potential longevity issue, further investigation is warranted.

# The fatal mistake

- ✱ In spite of pitfalls, may engineers push ahead with a simple FOS or statistical strength design.
- ✱ We offer two recent case studies to prove that hidden details can kill you.
  - ✱ We hope these examples may save a soul. Good luck!

# Example #1: puck bond strength

## ★ Background history

- ★ Dow-Corning Q3-6093 RTV silicone has been used by UASO since 1997 as a compliant adhesive to bond high-expansion steel pucks to the backplate of low expansion mirrors.
- ★ The mirror support system bolts into the pucks (see photos, page 54). If the bonds fail, the mirror may be destroyed & serious injury may occur.

# Example #1: puck bond strength

## ★ Background history

- ★ Initial work by P. Gray showed an average witness puck bond shear strength of 0.8 MPa with a standard deviation of ~10% of the mean.
  - ★ An average strength of 0.8 MPa exceeds the highest nominal shear stress by ~10X.

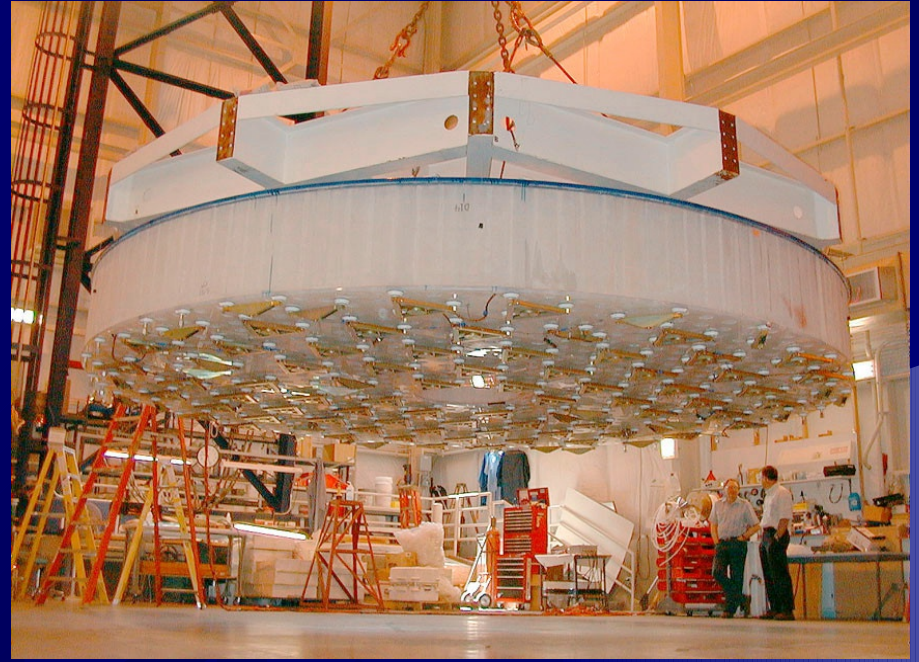
# Example #1: puck bond strength

## ☀ Background history

- ☀ At a LOTIS SRR review in 2001, B. Cuerden reported empirical data that indicated LOTIS collimator puck bonds would fail in ~15 years due to an age-weakening process.
- ☀ A study was ordered by Lockheed-Martin MPE to evaluate this critical safety threat in more detail. The foregoing slides are selected results from that study.



# Mirror Support System



Left: Bonding of pucks to the mirror backplate; note the squeeze-out bead of Q3-6093 RTV around the perimeter of each puck (white disks). Right: Pucks & mirror support loadspreader frames on the backplate of an 8.4m honeycomb mirror.



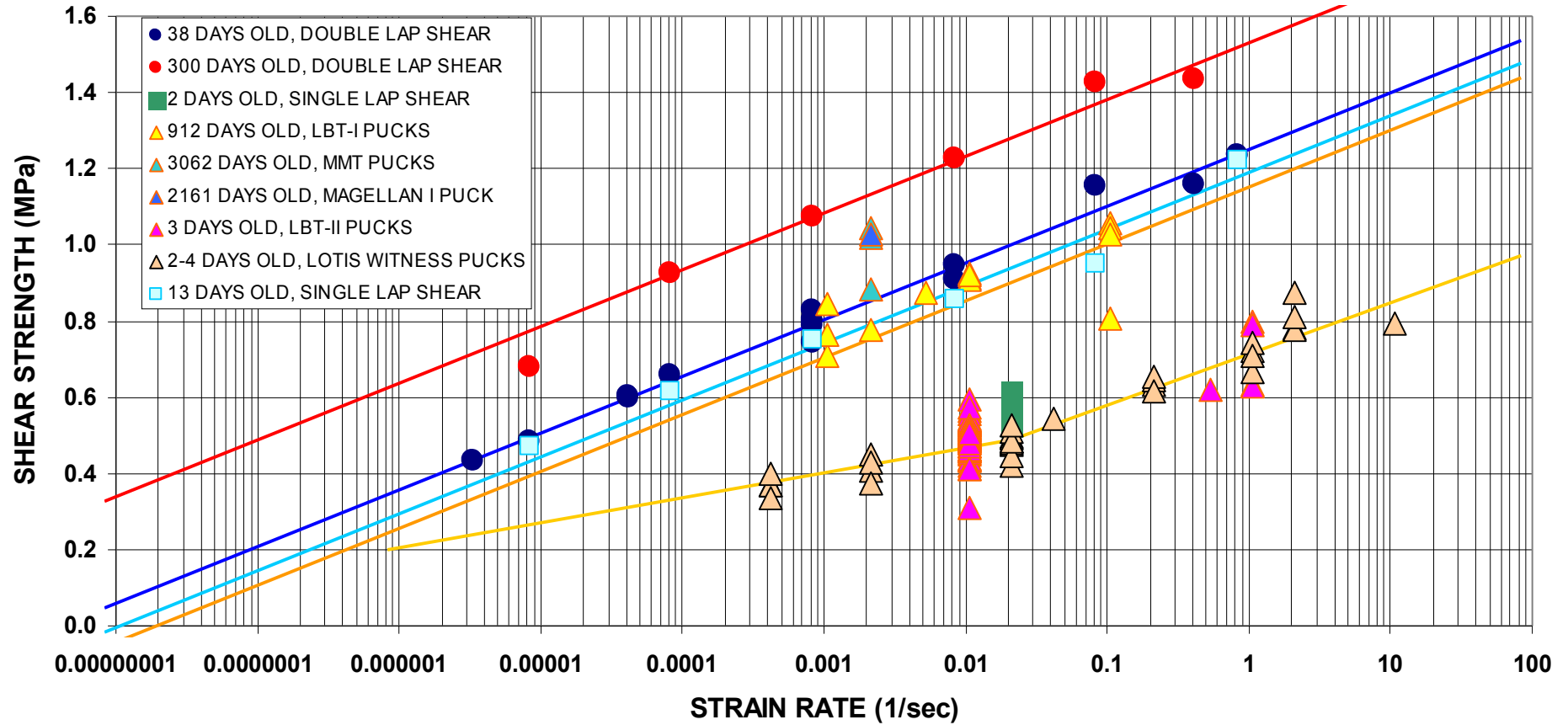
# RTV bond tests at constant speed

- ✱ Shear strength depends on testing speed & bond geometry.
- ✱ Shear strength depends on bond age initially, but becomes independent of age sometime between 2.5-6 yr.
- ✱ Shear strain at rupture (“critical strain”) depends on testing speed & age initially, but becomes independent of speed & age sometime between 1-2.5 yr, reaching a constant value of ~100%.

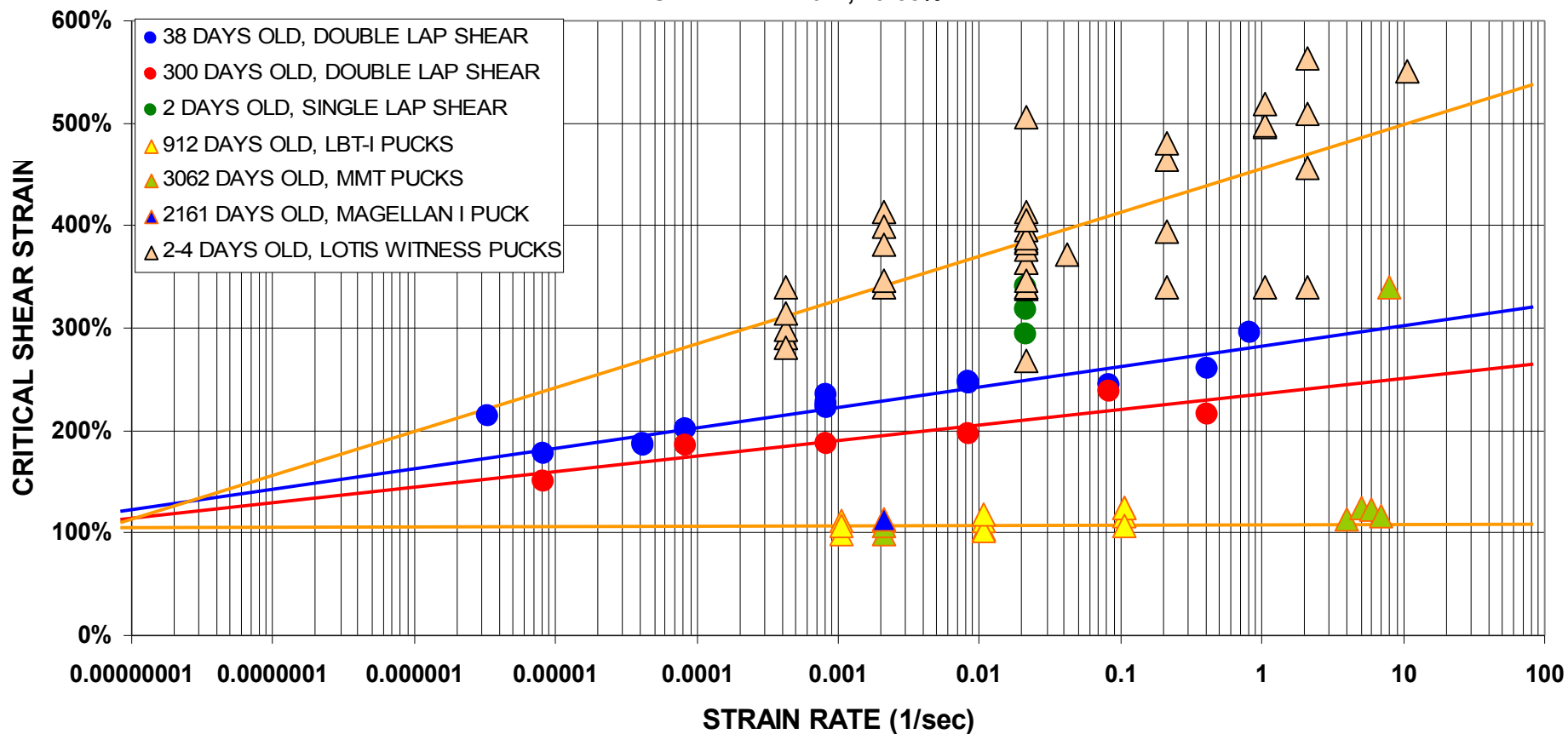
# RTV bond tests at constant speed

- ★ Inspection of fracture surfaces showed 100% cohesive failure occurred in all cases.

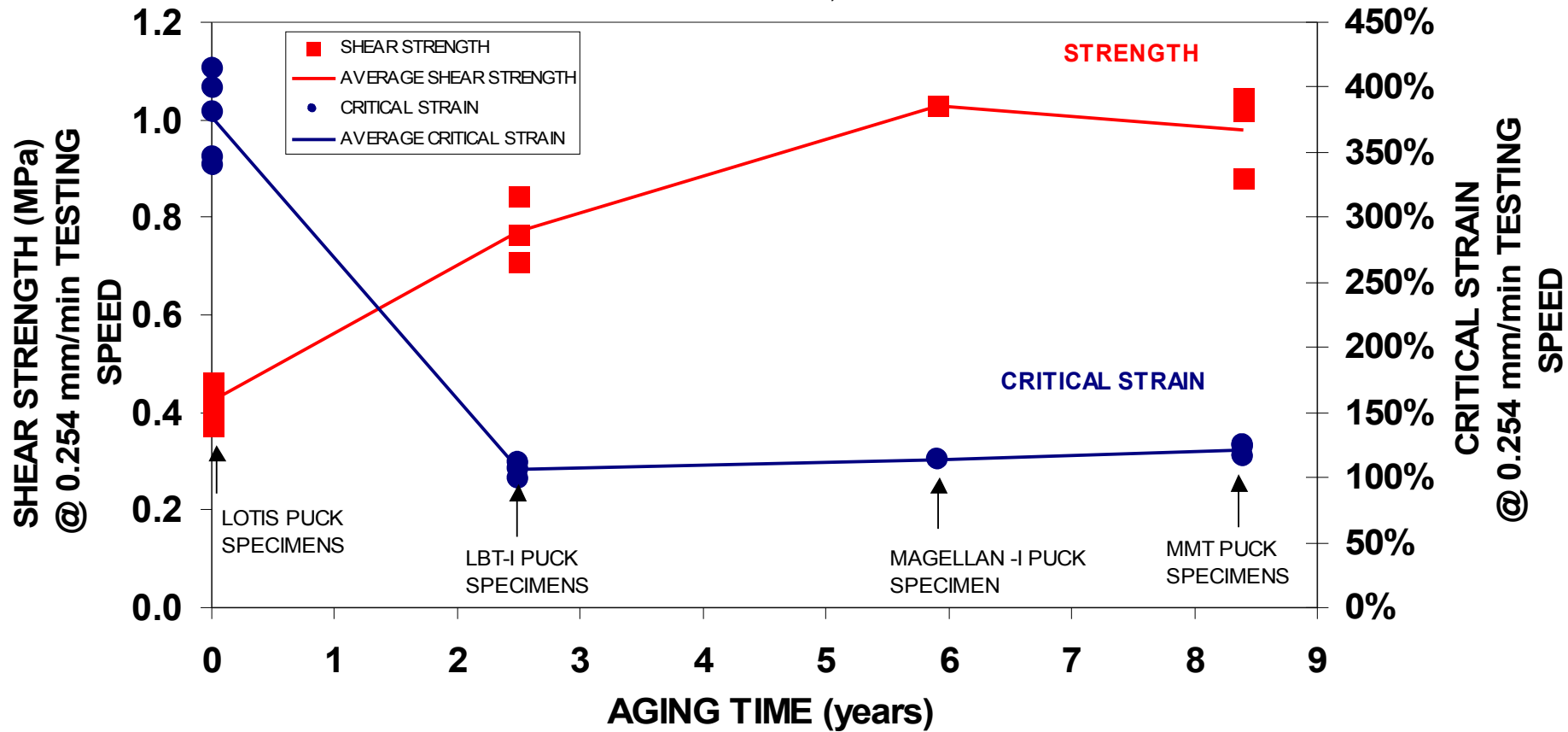
**DOW-CORNING Q3-6093 RTV SILICONE BOND STRENGTH  
AGED AT LABORATORY AMBIENT CONDITIONS BEFORE TESTING  
TESTED AT 17-20 C, 29-38% RH**



**DOW-CORNING Q3-6093 RTV SILICONE BOND**  
**AGED AT LABORATORY AMBIENT CONDITIONS BEFORE TESTING**  
**TESTED AT 17-20 C, 29-38% RH**



**WITNESS PUCK SPECIMENS, DOW-CORNING Q3-6093 RTV SILICONE BOND**  
BONDS AGED VARIOUS LENGTHS OF TIMES AT LABORATORY AMBIENT CONDITIONS BEFORE TESTING  
TESTED AT 22 C, 34-43% RH



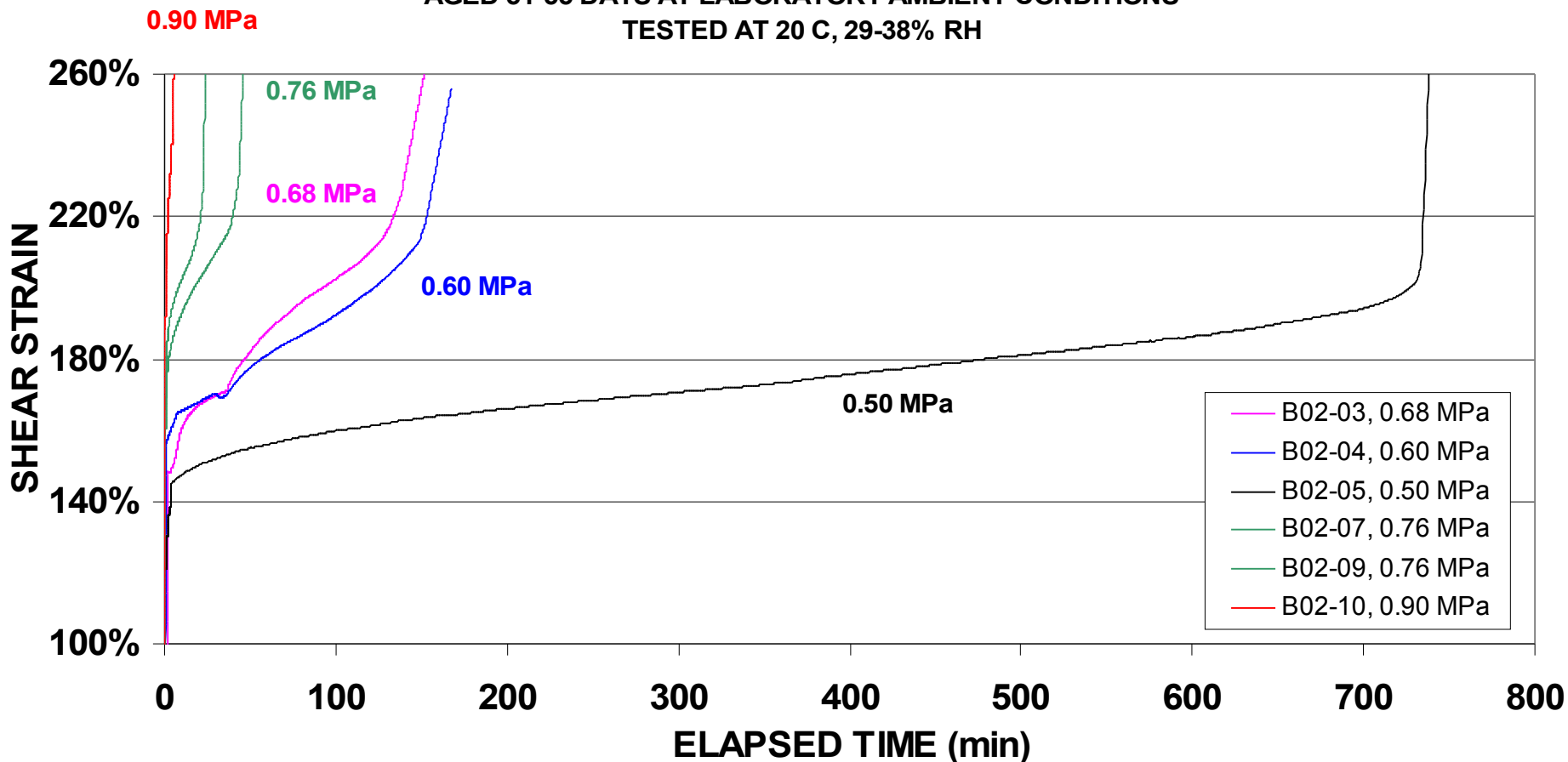
# RTV tests under static load

- ★ Young lap shear bonds (8 wk old) show power law creep-rupture behavior in the shear stress range 0.5-0.9 MPa.
- ★ Aged witness puck bonds (5+ years old) show linear creep in the shear stress range 0-0.08 MPa.
  - ★ Assuming that pucks rupture when they reach a critical shear strain of ~100%, the lifetime extrapolated from the measured creep curve is ~3.4 years at 0.08 MPa (0.08 MPa the nominal shear stress level for a fixed horizon-pointing mirror).

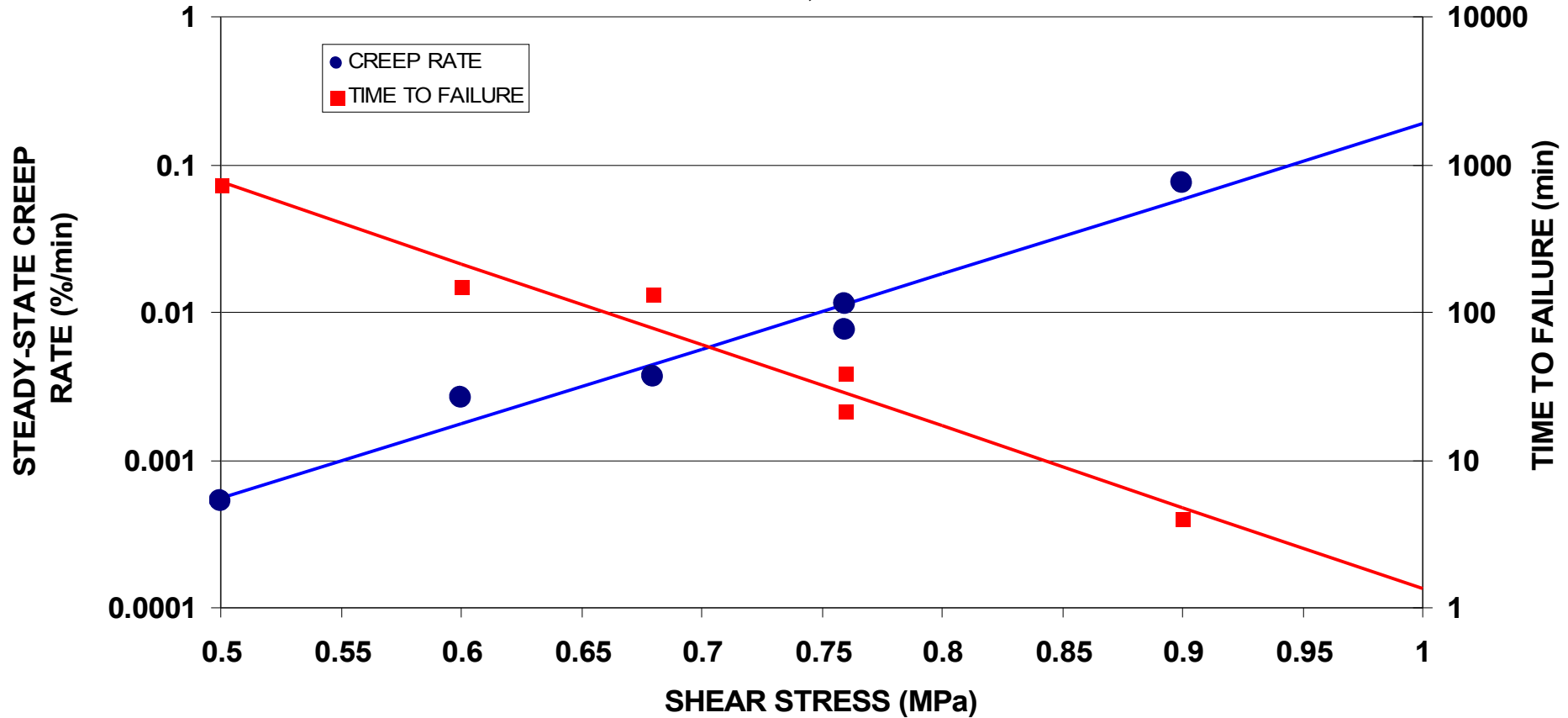
# DOUBLE LAP SHEAR SPECIMENS, DOW-CORNING Q3-6093 RTV SILICONE BOND

AGED 51-55 DAYS AT LABORATORY AMBIENT CONDITIONS

TESTED AT 20 C, 29-38% RH

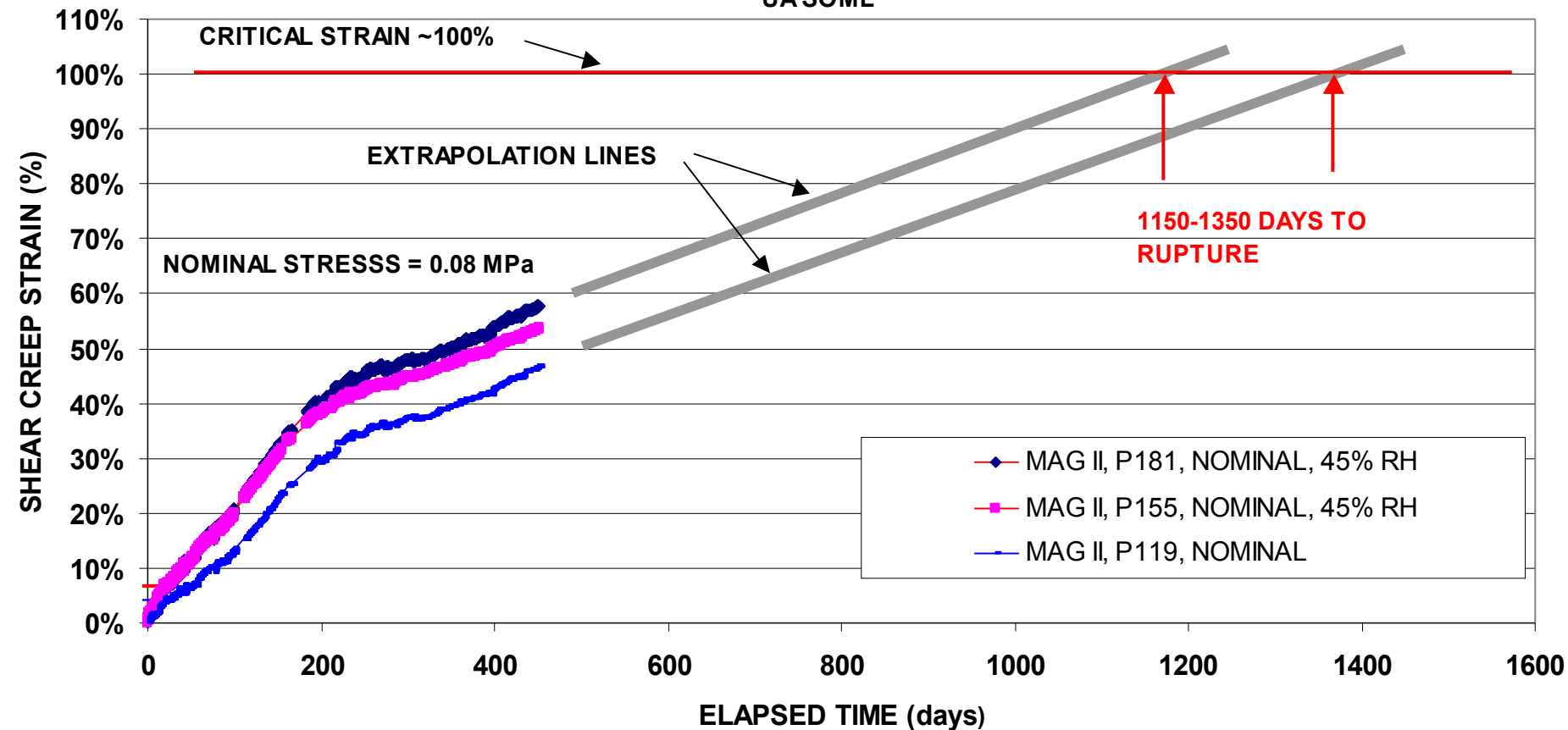


**DOUBLE LAP SHEAR SPECIMENS, DOW-CORNING Q3-6093 RTV SILICONE BOND**  
AGED 51-55 DAYS AT LABORATORY AMBIENT CONDITIONS  
TESTED AT 20 C, 29-38% RH





**DOW CORNING Q3-6093 PUCK BONDS**  
**CREEP UNDER LATERAL SHEAR LOAD, AMBIENT CONDITIONS**  
**UA SOML**



# Simple conclusions

- ★ Gray's original bond strength tests at uncontrolled high speed failed to capture the significant time-dependent nature of bond strength.
  - Although strength tests were carried out to prove good adhesion properties of the RTV, they were of insufficient scope to identify longevity problems.
  - Now UASO is faced with a serious problem:
    - **What should be done to three 6.5m & two 8.4m telescope mirrors that have Q3-6093 puck bonds? Will they fail in the 50 year design lifetime? We don't yet know!**
    - **Note that a bond life problem could have been identified early in the design phase by just a few more simple strength tests.**

# Simple conclusions

- ★ Although subsequently shown to be insufficient, Gray's work went far beyond what a typical designer would do.

# Simple conclusions

- ✱ A typical designer would probably use conventional statistical strength (e.g., 99% CI on strength) or FOS design based on data in the manufacturer's product data sheet or in-house data collected at one testing speed.
- ✱ For Q3-6093, the strength decreases with time, eating away the design margin as time passes.
  - ✱ Note the product data sheet doesn't mention a strength-speed or strength-time dependence. Designer beware!

# Simple conclusions

- ☀ The bond also creeps under load & the creep rate depends on humidity. There is no mention of creep in the product data sheet.
- ☀ At fixed testing speed, strength depends on specimen geometry. The exact bond geometry must be tested to reveal this fact.
  - ☀ Note this “fact” is universally recognized by experienced designers even though no explicit mention of it is made in the product data sheet. Designer beware!

# Simple conclusions

- ★ Strength vs. testing speed cannot be directly related to time-to-failure under static load by the laws of continuum mechanics alone.

# Data unification

- ✱ Can strength vs. testing speed and lifetime under static load be related & folded into a universal lifetime prediction model?
  - Yes - if the damage mechanism is quantitatively understood.
  - The challenge is developing & verifying the damage model.
    - Attempts to unify the RTV test data by assuming the backbone molecule is cut by a hydrolytic scission mechanism has failed.
    - The next approach is to assume that damage is proportional to a strain that's viscoelastic in nature. If this proves to be true, bond life should obey time-temperature superposition & accurate life extrapolations are possible.

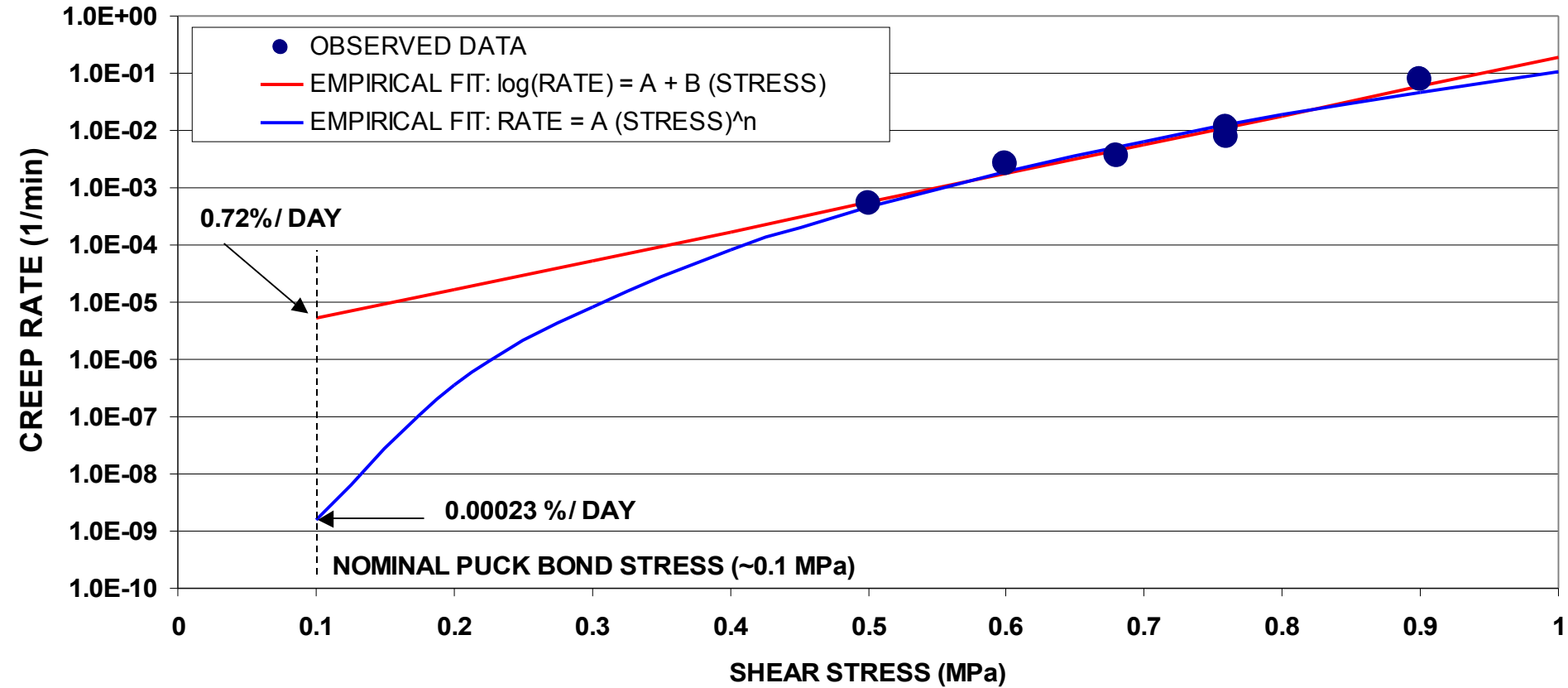
# Empirical modeling

- ★ Time-to-failure & creep rate under static loads can, in principle, be extrapolated from high stress levels (where measurements are fast & easy) to low stress levels (where measurements take years & so there's little or no data).
  - But accurate extrapolation requires a predictor equation that obeys the true damage physics.
  - Oftentimes, theory or purely empirical curve fitting offer competing models. What is the correct model?
  - Extrapolations far from the last observation can give remarkably bad predictions if the model is inaccurate. And without an observation at low stress levels, it is impossible to empirically discriminate between competing models!



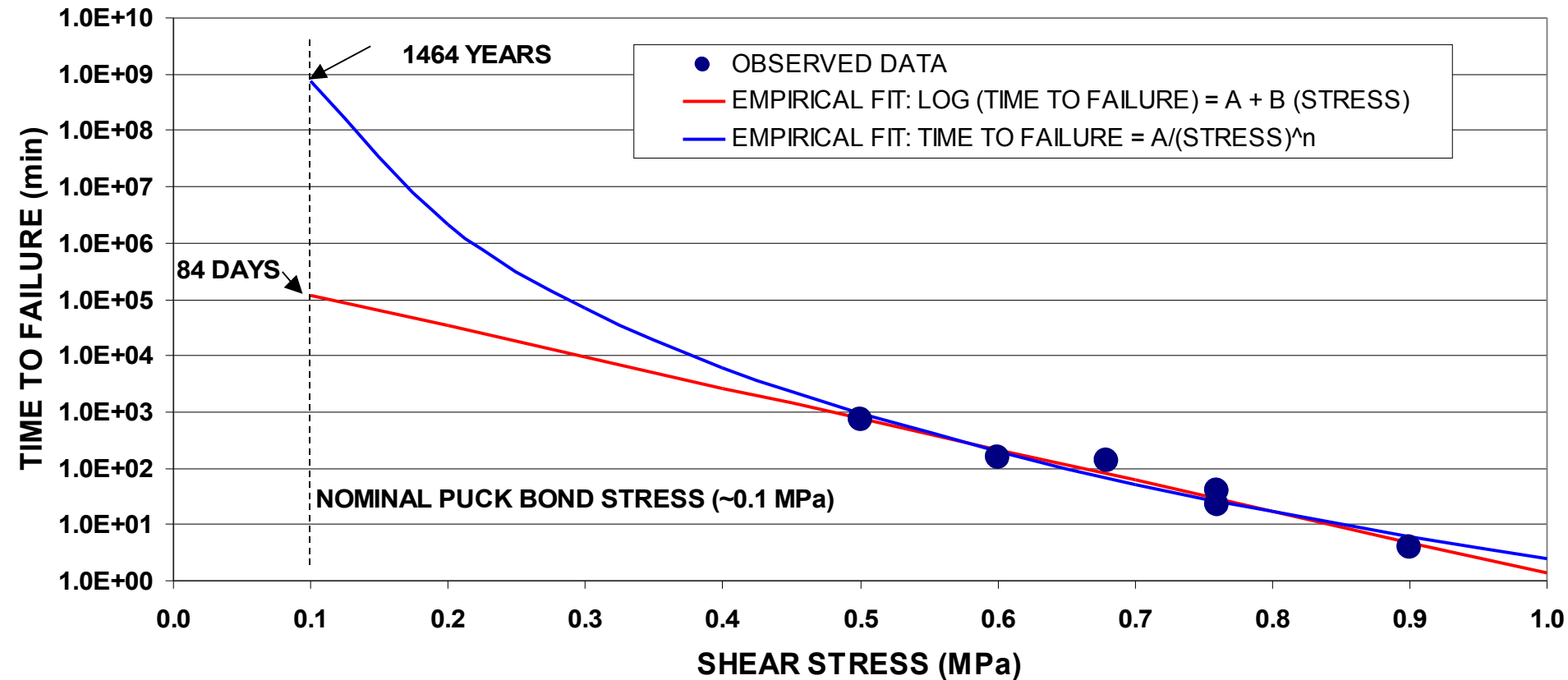
# Which one is the true model?

DOUBLE LAP SHEAR SPECIMENS, DOW-CORNING Q3-6093 RTV SILICONE  
BOND AGED 51-55 DAYS AT LABORATORY AMBIENT CONDITIONS.  
TESTED AT 20 C, 29-38% RH



# Which one is the true model?

**DOUBLE LAP SHEAR SPECIMENS, DOW-CORNING Q3-6093 RTV SILICONE**  
**BOND AGED 51-55 DAYS AT LABORATORY AMBIENT CONDITIONS.**  
**TESTED AT 20 C, 29-38% RH**



# Example #2: 3M Scotchweld 2216 epoxy bonds

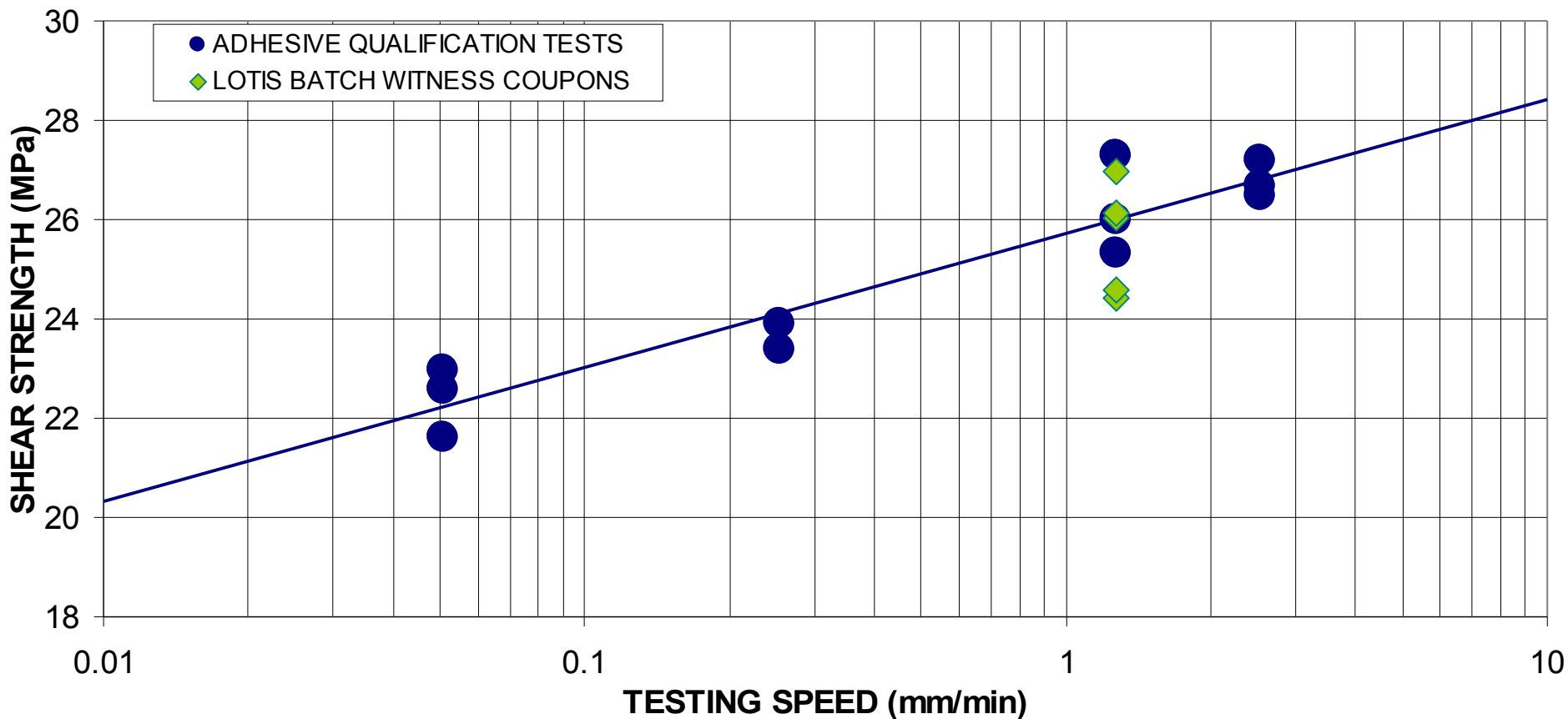
- ★ Scotchweld 2216 is a moderately compliant two-part epoxy often used to bond hardware to glass.
  - ★ A substrate primer is recommended for optimum bond strength.
- ★ The longevity of 2216 bonds is not discussed in the manufacturer's data sheet or in the literature.

# Scotchweld 2216

## ☀ Tests showed that:

- ☀ Bond shear strength depends on testing speed.
- ☀ The bond creeps under load.
- ☀ Bonds under moderately high shear stress (18 MPa) failed < 2 hours.
- ☀ Critical shear strain ~200%.
- ☀ Predominantly adhesive failure was observed (<10% cohesive).

**LMSSC-MSO LOTIS COLLIMATOR PROJECT**  
SCOTCH WELD 2216 EPOXY, 3901 PRIMER, CURED 150 F FOR 2 HRS, AVERAGE BOND THICKNESS 0.18 m m  
SINGLE LAP SHEAR, ABRADED/PRIMED ALUMINUM SUBSTRATE, TESTED AT 71 F, 24% RH

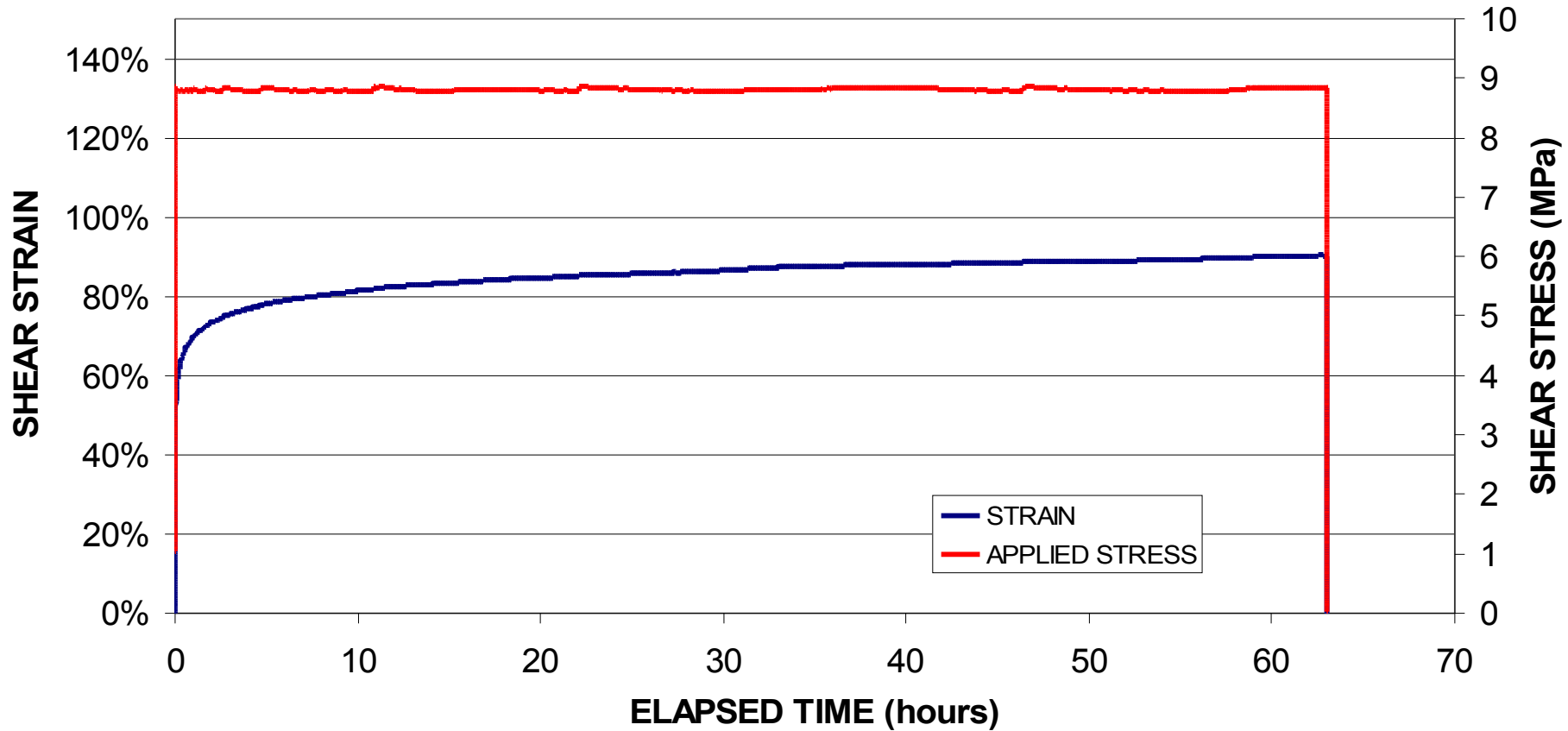


# CRITICAL STRAIN ~200%

## LMSSC-MSO LOTIS COLLIMATOR PROJECT

SCOTCH WELD 2216 EPOXY, CURED 150 F FOR 2 HRS, SPEC L2216-19, BOND THICKNESS 0.18 mm, 254 mm<sup>2</sup> AREA

SINGLE LAP SHEAR, ABRADED/PRIMED ALUMINUM SUBSTRATE, TESTED AT 19.5 C, 30% RH



# Scotchweld 2216

- ★ Preliminary results show that longevity concerns are warranted even though field data indicate no bond life problems.
- ★ Much more work is needed to develop a lifetime predictor model!

# UASO-ML testing capabilities

## ☀ Specialized test equipment:

- Universal electromechanical (screw driven) testing machines:
  - Instron 4501, 1100 lbf load capacity. Instron Series IX & Merlin software control package.
  - Instron 1125, 20000 lbf load capacity. Instron Series IX & Merlin software control package.
- Olympus BH-2 polarizing microscope with optical petrographic analysis accessories.
- Cahn C-31 microbalance, 0.1 micro-gram sensitivity.
- General data acquisition PC running DasyLab software.



# UASO-ML testing capabilities

## ☀ Services

- ☀ UASO does not provide general-use testing services.
- ☀ Equipment is purchased & maintained by specific projects. This is expensive equipment whose cost to maintain, operate & replace is not managed by a UASO overhead account.
- ☀ Use of equipment by outside projects is done on a courtesy basis free-of-charge provided it does not interfere with principal program needs & does not consume more than a few man hours of labor.

# UASO-ML testing capabilities

## ☀ Services

- ☀ Outside requestors must provide their own labor or a charge number for extended use of ML labor.
- ☀ Users must demonstrate competence or must be trained prior to equipment operation. Users must provide specialized tooling & their own test plan.
- ☀ Cost to repair damage caused by user negligence will be billed to project.

# Puck bond references

- ★ Peter Gray, “6.5m MMT Mirror Support Load Spreader Gluing: Summary of Problems, Investigations and Solutions,” MMT Observatory Technical Report #29, July 1997.
- ★ Amar Brar, Brian Cuerden, Kurt Kenagy, Blain Olbert and Steve Warner, “*LMSSC-MSO LOTIS Collimator Project Mirror Puck Bond Test Plan*,” LOTIS Project Technical Memo, 14 Nov. 02. File: Puck Bonding Plan R03.doc
- ★ Blain H. Olbert, “*LMSSC-MSO LOTIS Collimator Project: Physics & Rationale of Mirror Puck Test Plan*,” LOTIS Project Technical Memo, 12 Dec. 2002, File: Puck Bonding Plan Theory R03.doc
- ★ Blain H. Olbert, “LMSSC-MSO LOTIS COLLIMATOR PROJECT, MIRROR PUCK TEST PLAN SUMMARY,” PowerPoint Summary, 1 Feb. 2003, File: LOTIS\_PUCKTEST\_SUM\_01. Ppt

# Puck bond references

- ★ Blain Olbert, “LMSSC-MSO LOTIS COLLIMATOR PROJECT, MIRROR PUCK & RTV TESTING: PLAN SUMMARY, STATUS & RESULTS TO-DATE,” PowerPoint Summary, 02/27/03, File: LOTIS\_PUCKTEST\_UPDATE\_02.ppt
- ★ Blain Olbert, “LMSSC-MSO LOTIS COLLIMATOR PROJECT, MIRROR PUCK BOND & RTV TESTING, STATUS REPORT & RESULTS SINCE 03/04/03 GROUP TELECON,” PowerPoint Summary, 04/25/03, File: LOTIS\_PUCKTEST\_UPDATE\_03.ppt
- ★ Blain Olbert, “LMSSC-MSO LOTIS COLLIMATOR PROJECT, MIRROR PUCK BOND & RTV TESTING, STATUS REPORT & RESULTS SINCE 05/01/03 GROUP TELECON,” PowerPoint Summary, 04/25/03, File: LOTIS\_PUCKTEST\_UPDATE\_04.ppt
- ★ Brian Cuerden and Steven Bauman, “Safety Clip System Design Concept, ” 07/01/03, File: LOTIS 070103-SOLM.ppt

# Puck bond references

- ✱ **Blain H. Olbert, “Proposal to Explore the Long-Term Potential for Catastrophic Failure of the AFRL 3.5m SOR Primary Mirror Support,” Prepared for AFRL Starfire Optical Range, Kirtland AFB, Albuquerque, NM.**