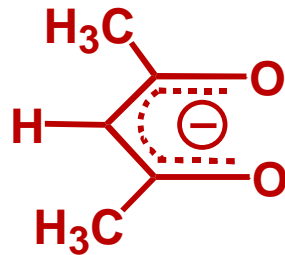
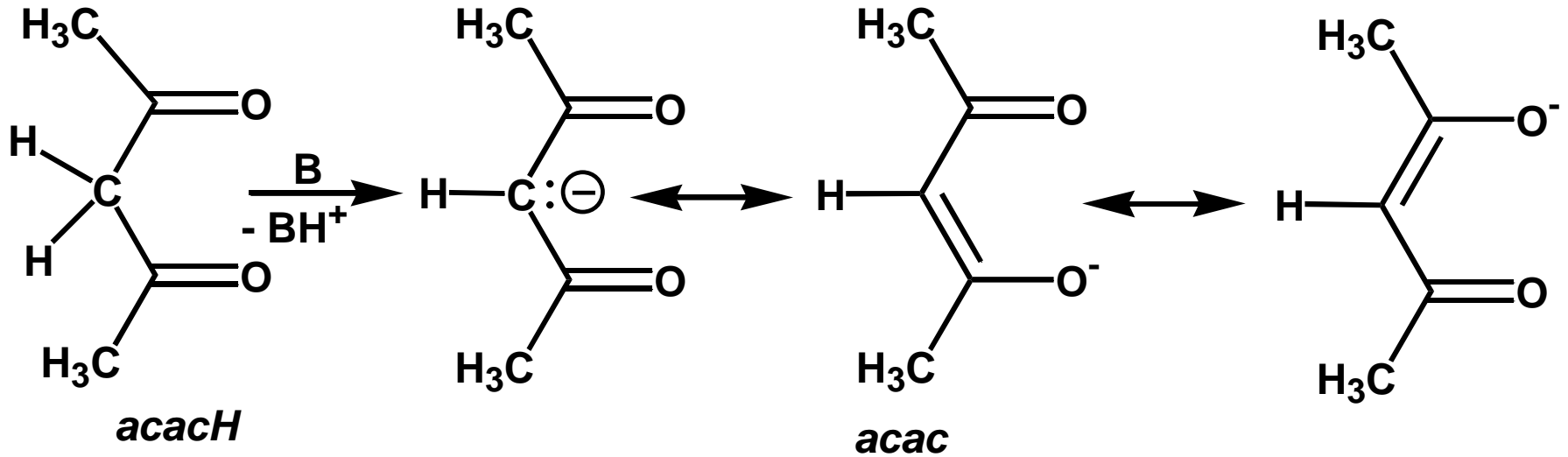


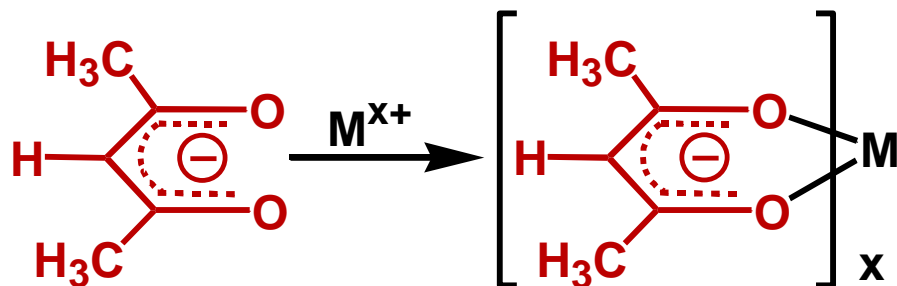
ESPERIENZA 1

Sintesi di acetilacetato complessi di metalli di transizione della prima serie

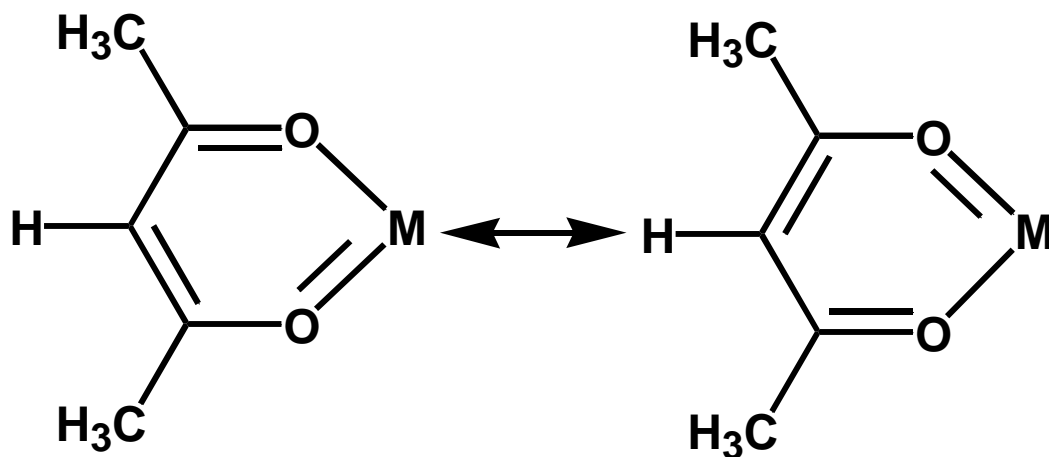
Il comportamento acido/base del 2,4-pentandione



La **coordinazione** dello ione acetilacetonato ai metalli di transizione



Altre **forme di risonanza** dovute alla coordinazione



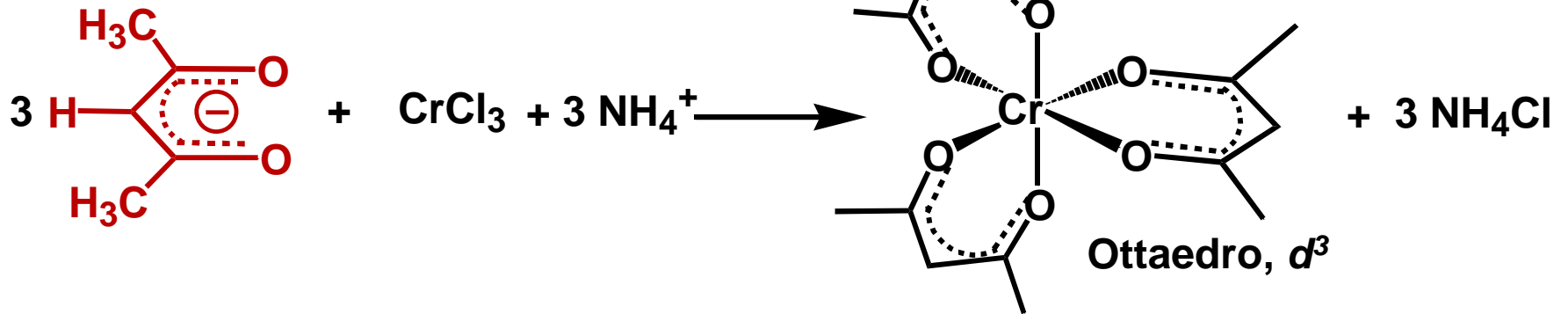
Distanze di legame a due a due **uguali**: $M-O$, $C-O$, $C-C$

Sintesi di $[\text{Cr}(\text{acac})_3]$

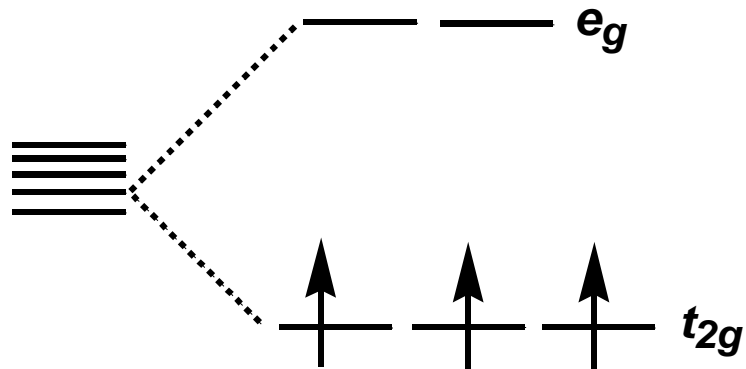
Idrolisi dell'urea



Sintesi del complesso

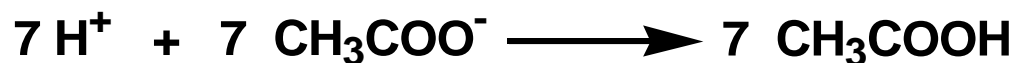
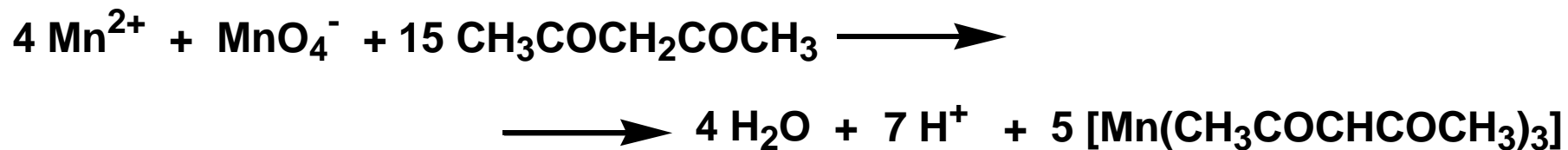
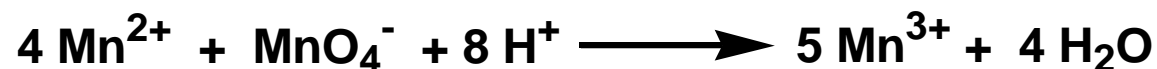


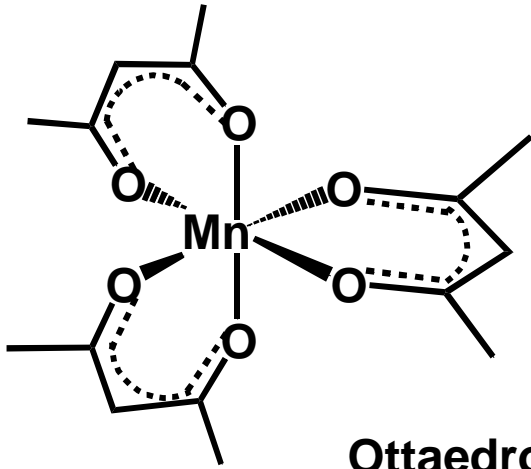
paramagnetico



Sintesi di $[\text{Mn}(\text{acac})_3]$

Sintesi del complesso



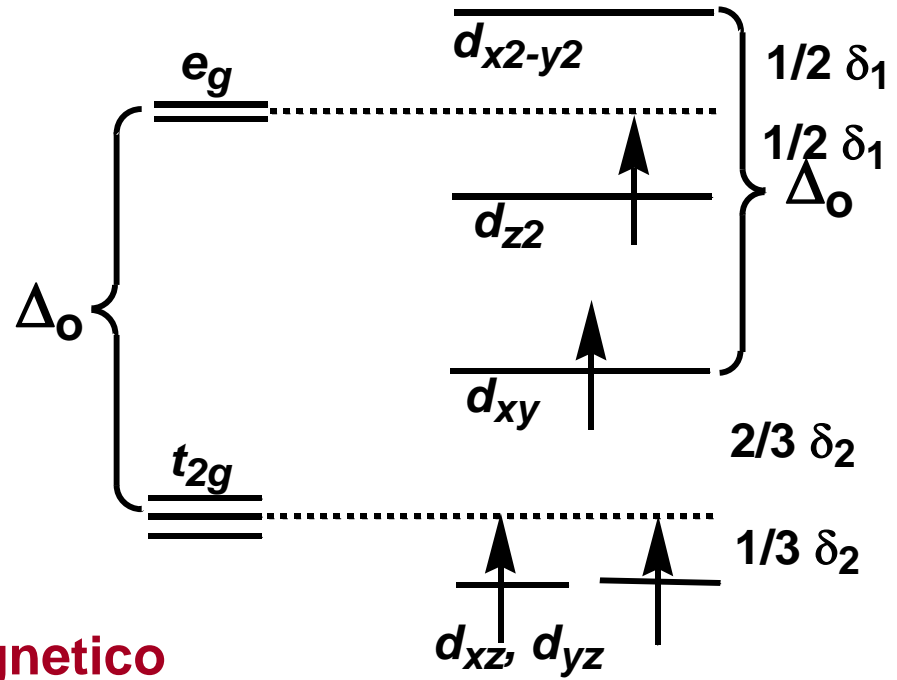
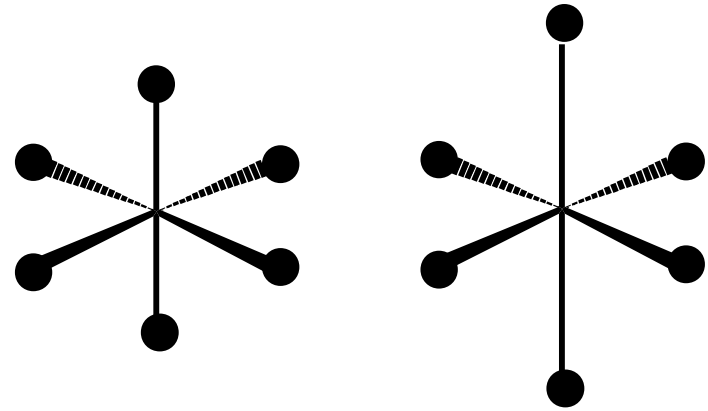


Ottaedro, d^4

Distorsioni per effetto Jahn-Teller:

Allungamento: 2 Mn-O 2.12 Å;
4 Mn-O 1.93 Å;

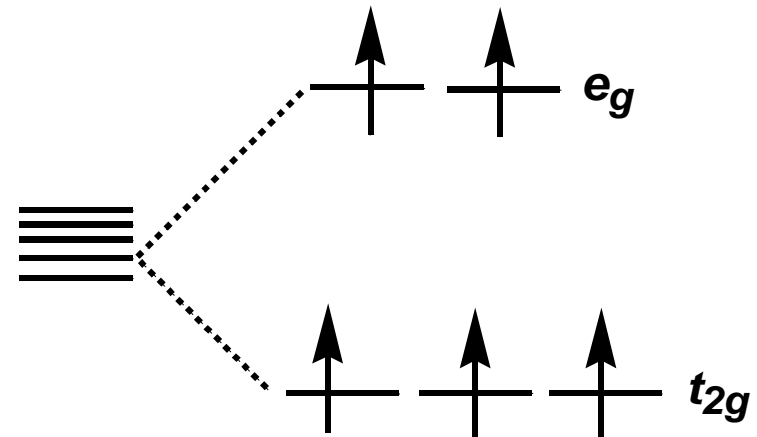
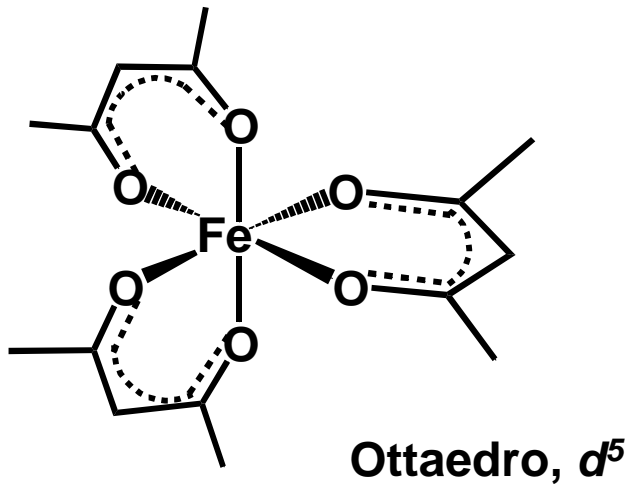
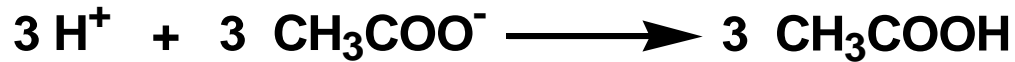
Compressione: 2 Mn-O 1.95 Å;
4 Mn-O 2.00 Å.



paramagnetico

Sintesi di $[\text{Fe}(\text{acac})_3]$

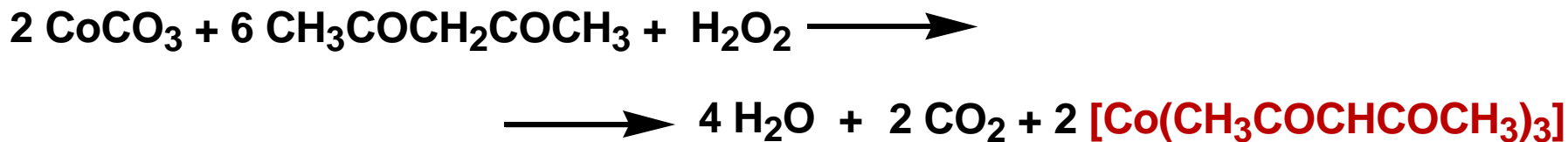
Sintesi del complesso



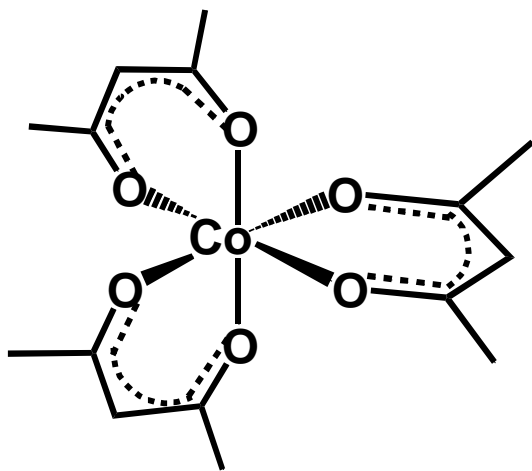
paramagnetico

Sintesi di $[\text{Co}(\text{acac})_3]$

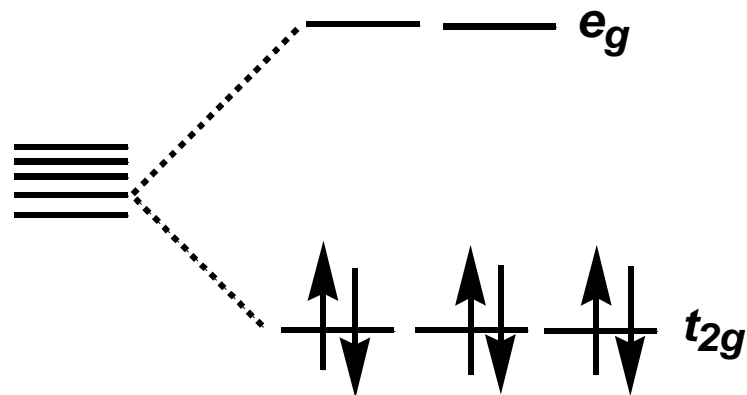
Sintesi del complesso



Da bilanciare



Ottaedro, d^6



diamagnetico

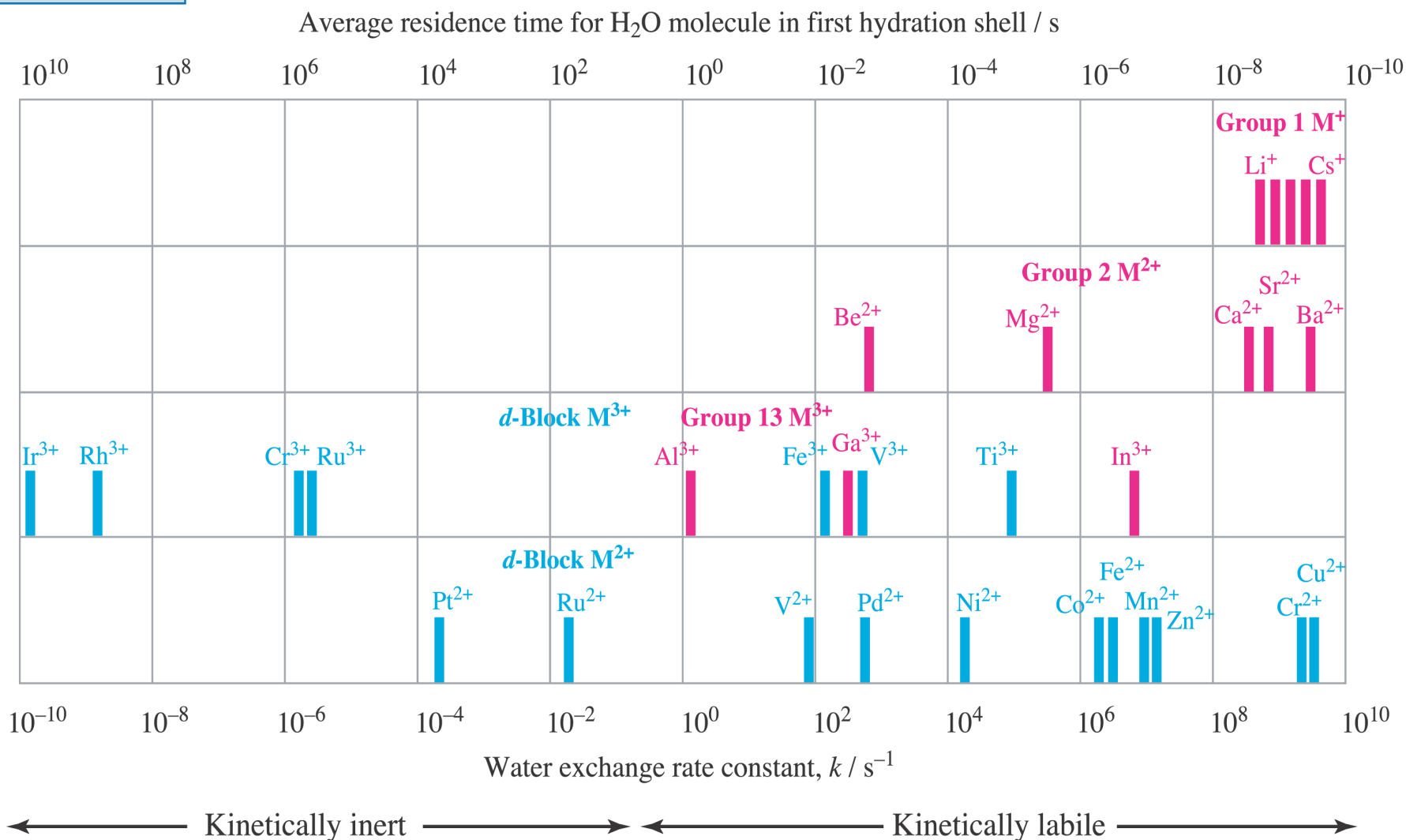
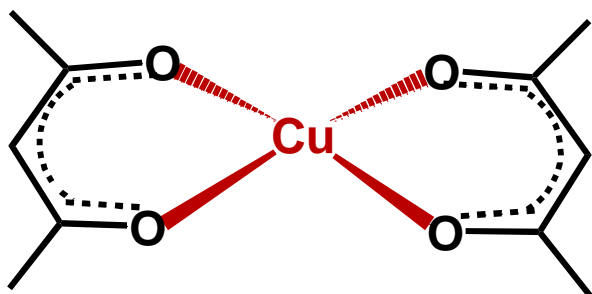
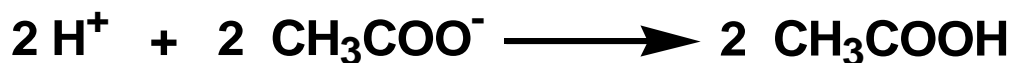
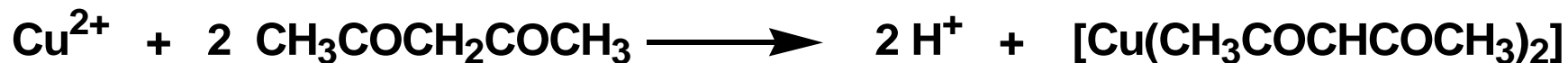


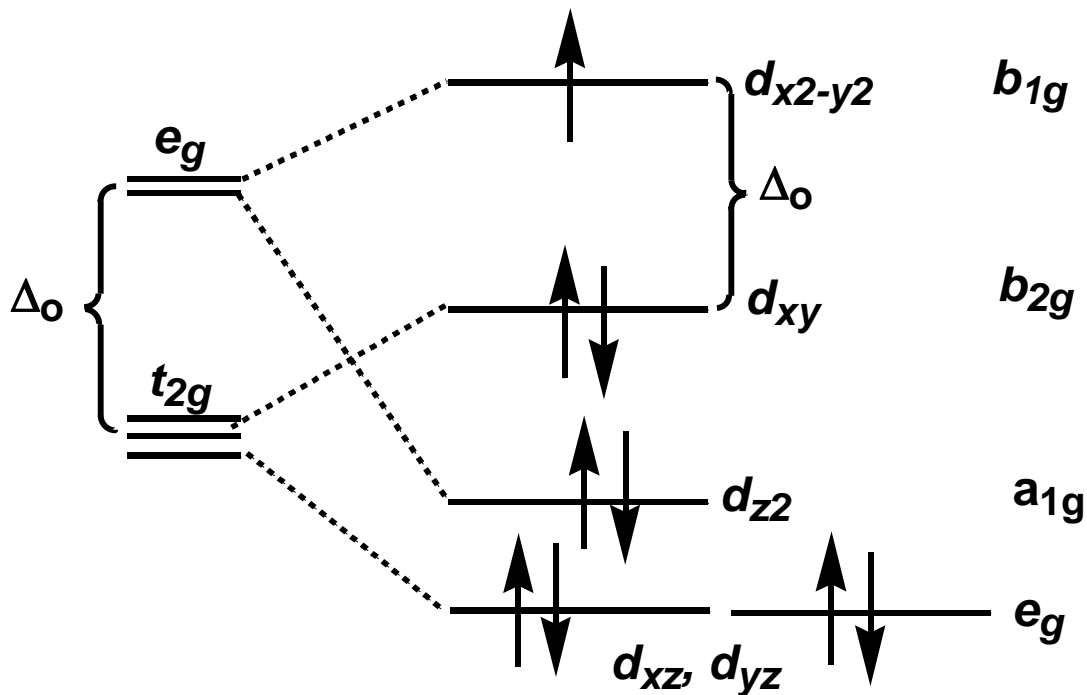
Fig. 26.1 Water exchange rate constants and average residence times for water molecules in the first coordination sphere of aquated metal ions at 298 K. Group 1, 2 and 13 metal ions are shown in pink, and *d*-block metal ions in blue. [Based on S.F. Lincoln (2005) *Helv. Chim. Acta*, vol. 88, p. 523 (Figure 1).]

Sintesi di $[\text{Cu}(\text{acac})_2]$

Sintesi del complesso



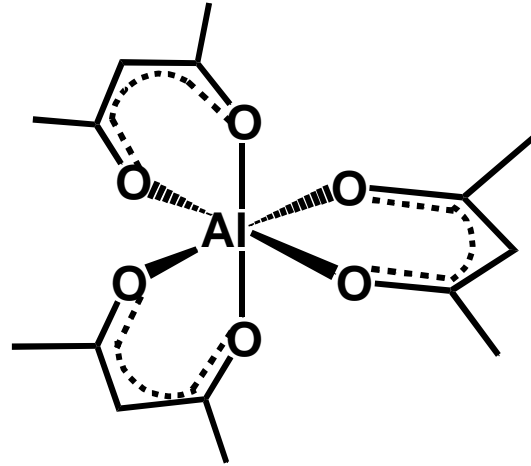
planare quadrato, d^9



paramagnetico

Sintesi di $[Al(acac)_3]$

Sintesi del complesso



ottaedro, **no elettroni *d***

diamagnetico