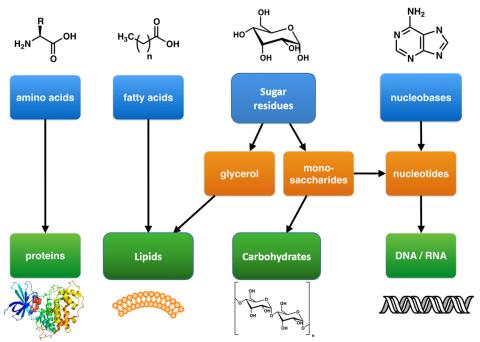
Prof. Sabrina Pricl

A.Y. 2020-2021

Lesson 1 Recognizing Macromolecules

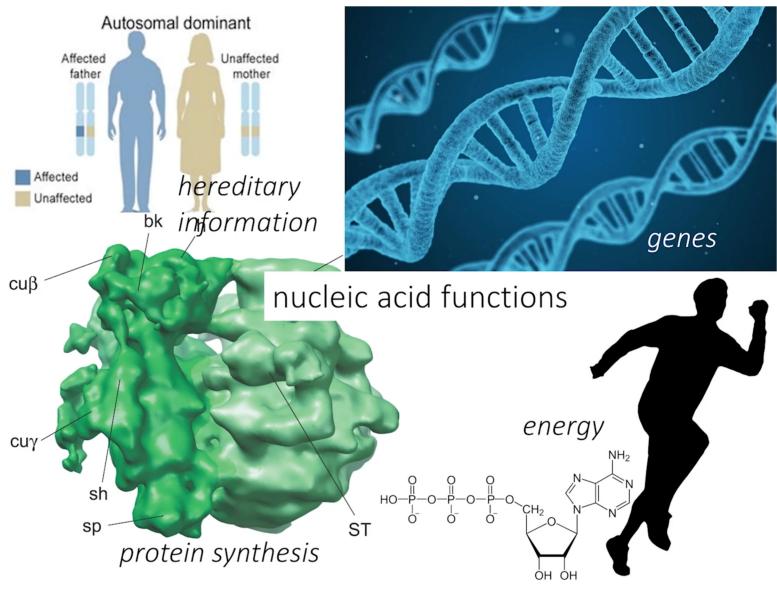


Nucleic acids

- Hereditary information (genes), energy
- Monomer M = nucleotides
- Polymers = DNA or RNA
- The nucleotide structure is stereotypical:

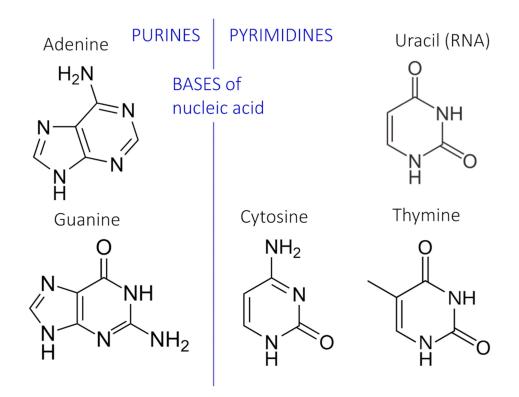
Phosphate-sugar-base (P-S-B)

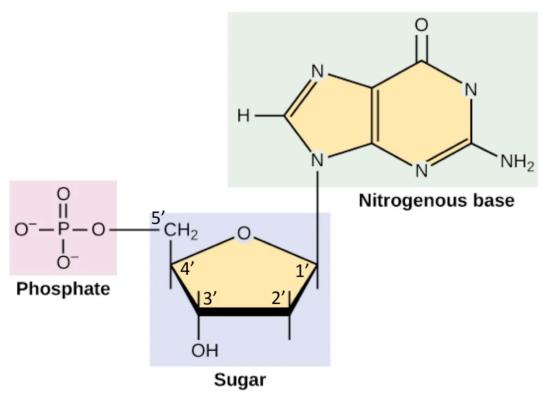
- Sugar = (5C) ribose (RNA) or deoxyribose (DNA)
- 5 bases:
 - Adenine (A), Guanine (G) = PURINES
 - Cytosine (C), Thymine (T), Uracil (U) = PYRIMIDINES
- A,G,C,**T** = DNA
- A,G,C,**U** = RNA



Nucleic acids

Nucleotide





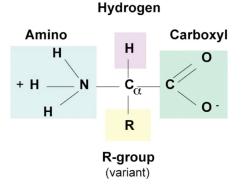
Note that the carbon atoms of the sugar are called C1', C2' etc C1' is where the base attaches

C2' can have attached a H (as here, for DNA) or a OH group (for RNA) C3' is very important and where the next nucleotide will join C5' is where the phosphate group attaches

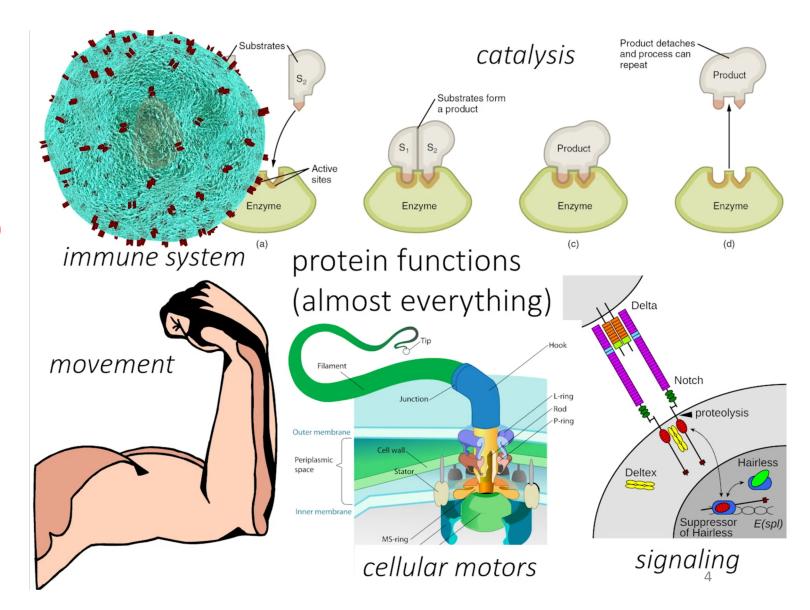
Proteins

- Everything except genes
- Monomer M = amino acids
- 20 natural common amino acids
- Polymer = protein
- Amino acid structure:

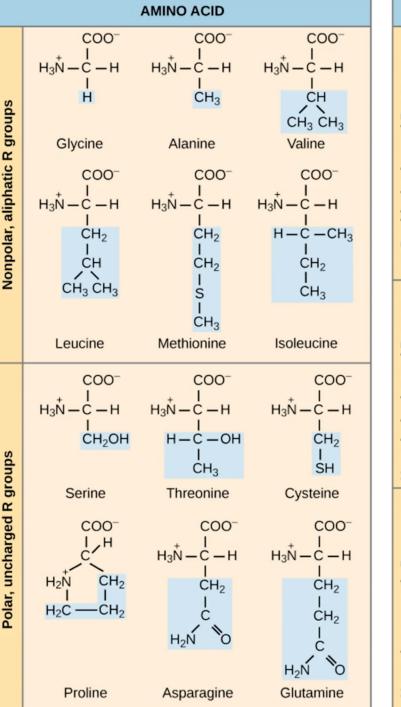
Alpha C + NH₂ + COOH + R (side group)

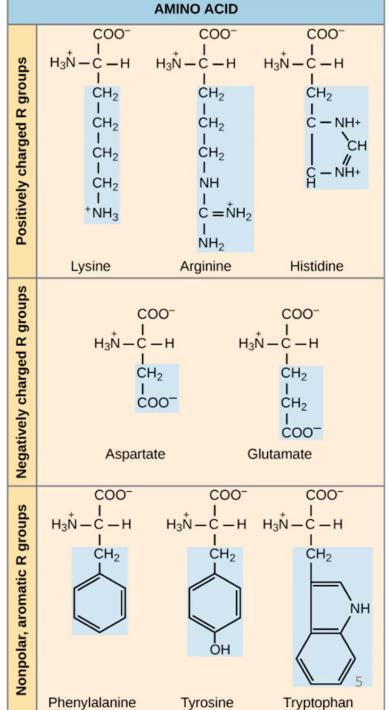


- R can be polar, non-polar, charged, uncharged
- Three or one letter code:
 - Valine = VAL
 - Valine = V



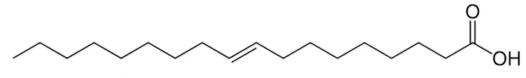
Proteins



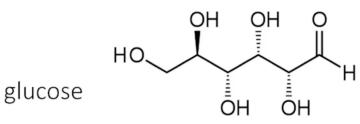


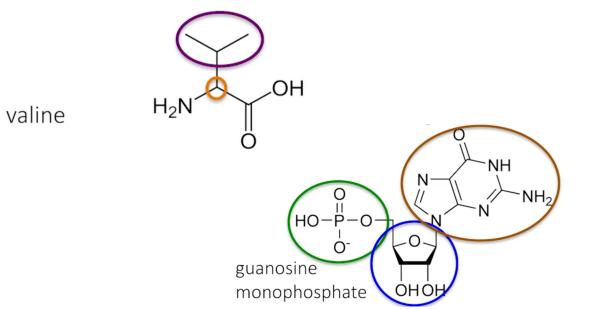
Elements of Chemical and Molecular Biology – Lesson 1

Recognizing macromolecules - review



trans-oleic acid





Lipids: nonpolar/hydrophobic

Carbohydrates: CH₂O formula/polar

Proteins: M = amino acid = α C bonded to NH₂, COOH and R, R = polar/nonpolar

Nucleic acids: M = nucleotide = phosphate + sugar + base, polar

Recognizing macromolecules

• Take assignment 1: Recognizing Macromolecules