

Solution Properties of the new Fusion Protein HUG



CASO STUDIO 2:
Ricerche svolte dal 2010 ad oggi

HELP (*Human Elastin-like Polypeptide*), a synthetic polypeptide based on the **VAPGVG** hexapeptidic motif that is found in the human elastin repetition domains, shows very interesting chemical-physical properties in solution and in particular it has the peculiar thermal behavior defined as the **reverse phase transition**:

- at temperatures below the transition temperature (T_t), the biopolymer is soluble in aqueous solutions where the free chains of HELP exist in a disordered and completely hydrated state.
- at temperatures above T_t , these chains show a more orderly structure (β -spiral) stabilized by intramolecular hydrophobic interactions that favor their association and the formation of an amorphous solid phase.

HELP

HELP

MW 44885.7 536 aa, Theoretical pi: 11.68

MRGSHHHHHGSAAAAAAA**KAAKAAQFGLVPGVGVAPGVGVAPGVGLAPGVGVAPGV**
GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGLAPGVGVAPGV
GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGLAPGVGVAPGV
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GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGVAPGVGVAPGVGLAPGVGVAPGV
GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGVAPGVGVAPGVGLAPGVGVAPGV
GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGVAPGVGVAPGVGLAPGVGVAPGV
GVAPGVGVAPGIAPAAAAAKAAKAAQFGLVPGVGVAPGVGVAPGVGVAPGVGVAPGVGVAPGVGVAPGV

Alpha helix	(Hh)	:	137	is	25.56%
β_{10} helix	(Gg)	:	0	is	0.00%
Pi helix	(Ii)	:	0	is	0.00%
Beta bridge	(Bb)	:	0	is	0.00%
Extended strand	(Ee)	:	0	is	0.00%
Beta turn	(Tt)	:	0	is	0.00%
Bend region	(Ss)	:	0	is	0.00%
Random coil	(Cc)	:	399	is	74.44%
Ambiguous states (?)	:	:	0	is	0.00%
Other states	:	:	0	is	0.00%

Total number of negatively charged residues (Asp + Glu): 0
Total number of positively charged residues (Arg + Lys): 17

HELP

Target Sequence:

10 20 30 40 50 60 70
MRGSHHHHHH GSAAAAAAAAA KAAAKAAQFG LVPGVGVAPG VGVAPGVGVA PGVGLAPGVG VAPGVGVAPG

80 90 100 110 120 130 140
VGVAPGIAPA AAAAAAKAAAK AAQFGLVPGV GVAPGVGVAP GVGVAPGVGL APGVGVAPGV GVAPGVGVAP

150 160 170 180 190 200 210
GIAPAAAAAAA KAAAKAAQFG LVPGVGVAPG VGVAPGVGVA PGVGLAPGVG VAPGVGVAPG VGVAPGIAPA

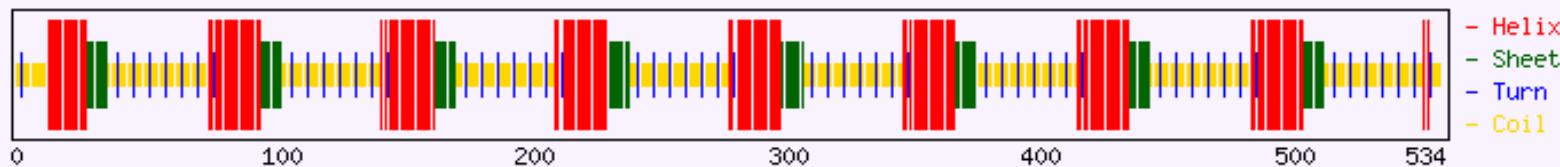
220 230 240 250 260 270 280
AAAAAAKAAAK AAQFGLVPGV GVAPGVGVAP GVGVAPGVGL APGVGVAPGV GVAPGVGVAP GIAPAAAAAAA

290 300 310 320 330 340 350
KAAAKAAQFG LVPGVGVAPG VGVAPGVGVA PGVGLAPGVG VAPGVGVAPG VGVAPGIAPA AAAAAAKAAAK

360 370 380 390 400 410 420
AAQFGLVPGV GVAPGVGVAP GVGVAPGVGL APGVGVAPGV GVAPGVGVAP GIAPAAAAAAA KAAAKAAQFG

430 440 450 460 470 480 490
LVPGVGVAPG VGVAPGVGVA PGVGLAPGVG VAPGVGVAPG VGVAPGIAPA AAAAAAKAAAK AAQFGLVPGV

500 510 520 530
GVAPGVGVAP GVGVAPGVGL APGVGVAPGV GVAPGVGVAP GIAP



In **2013**, a new protein, called **UnaG**, was identified and isolated from the muscle of the Japanese eel (*Anguilla japonica*) (N:B: è la prima proteina fluorescente derivata dai vertebrati).

This free-fatty acid binding protein binds unconjugated bilirubin (Br) in a highly specific and selective way, emitting strong fluorescence (Kumagai et al., 2013).

The UnaG gene was been cloned to the C-terminal of the gene encoding the HELP polypeptide obtaining a new functionalized synthetic polypeptide called **HUG**, an acronym that indicates **HELP-UnaG** sequence.

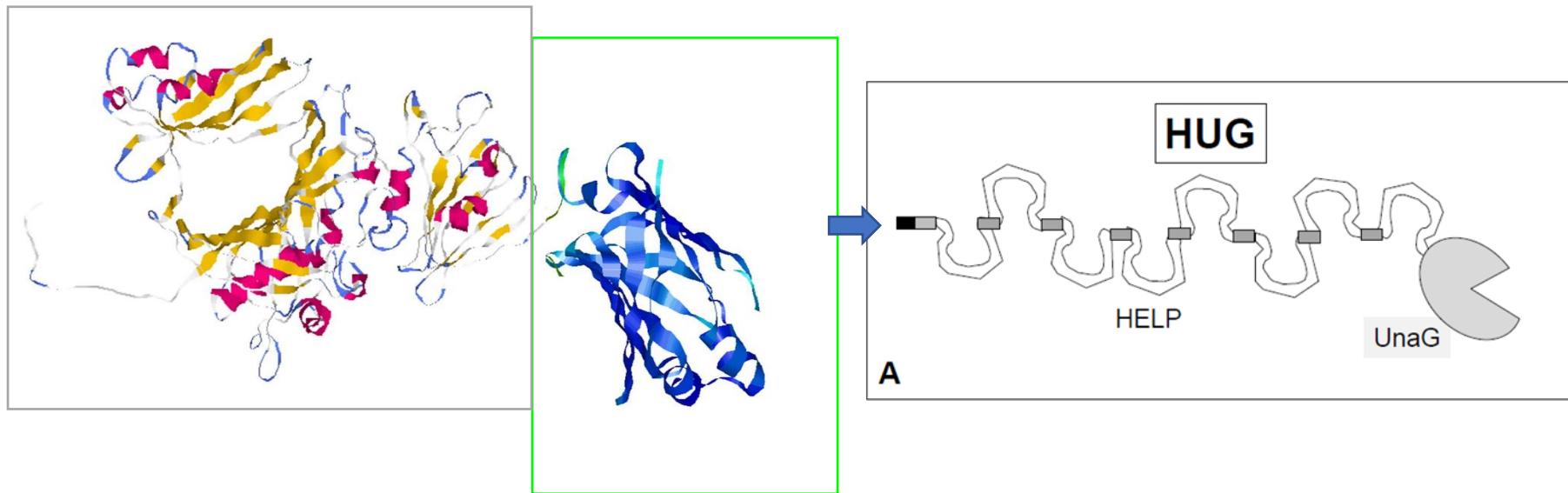
HUG (HELP-UnaG) was been produced from a synthetic gene of the HELP polypeptide fused with the **139** amino acids coding sequence of the UnaG bilirubin-binding protein

Its molecular property appears highly interesting because unconjugated bilirubin (indirect bilirubin) is hardly measured with traditional colorimetric assay.

HUG

A bi-functional, synthetic protein

HUG is composed of a bilirubin-binding domain (UnaG) fused with a scaffold (HELP)



HUG is the acronym of **HELP-UnaG**

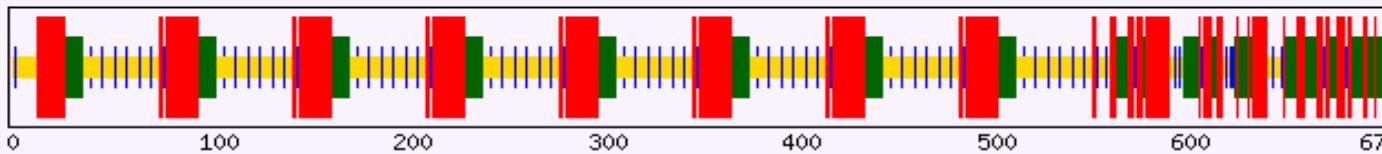
HELP-UnaG (clon. /10/2014) Number of amino acids: 675, Molecular weight: 60406.47, Theoretical pI: 9.88

Alpha helix	(Hh)	:	198	is	22.84%
β_{10} helix	(Gg)	:	0	is	0.00%
Pi helix	(Ii)	:	0	is	0.00%
Beta bridge	(Bb)	:	0	is	0.00%
Extended strand	(Ee)	:	45	is	5.19%
Beta turn	(Tt)	:	0	is	0.00%
Bend region	(Ss)	:	0	is	0.00%
Random coil	(Cc)	:	624	is	71.97%
Ambiguous states (?)	(?)	:	0	is	0.00%
Other states	(Oo)	:	0	is	0.00%

Total number of negatively charged residues
(Asp + Glu): 22
Total number of positively charged residues
(Arg + Lys): 39

HUG

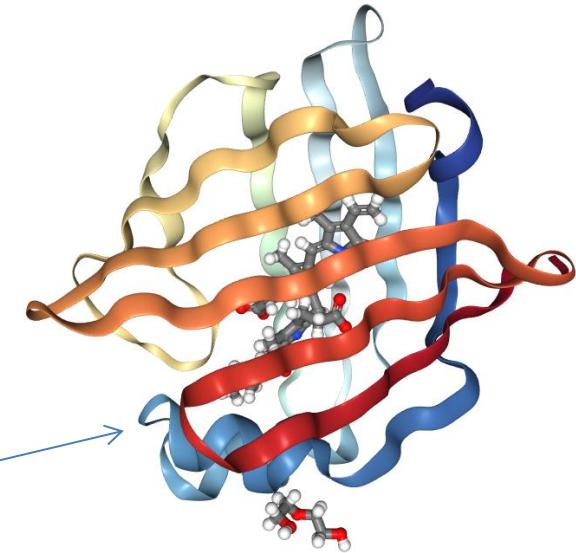
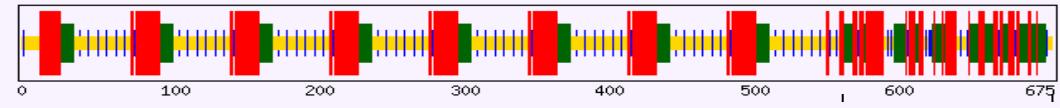
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80	90	100	110	120	130	140
VGVAPGIAPA	AAAAAKAAA	AAQFGLVPGV	GVAPGVGVAP	GVGVAPGVGL	APGVGVAPGV	GVAPGVGVAP
150	160	170	180	190	200	210
GIAPAAAAAA	KAAAKAAQFG	LVPGVGVAPG	VGVAPGVGVA	PGVGLAPGVG	VAPGVGVAPG	VGVAPGIAPA
220	230	240	250	260	270	280
AAAAAKAAA	AAQFGLVPGV	GVAPGVGVAP	GVGVAPGVGL	APGVGVAPGV	GVAPGVGVAP	GIAPAAAAAA
290	300	310	320	330	340	350
KAAAKAAQFG	LVPGVGVAPG	VGVAPGVGVA	PGVGLAPGVG	VAPGVGVAPG	VGVAPGIAPA	AAAAAKAAA
360	370	380	390	400	410	420
AAQFGLVPGV	GVAPGVGVAP	GVGVAPGVGL	APGVGVAPGV	GVAPGVGVAP	GIAPAAAAAA	KAAAKAAQFG
430	440	450	460	470	480	490
LVPGVGVAPG	VGVAPGVGVA	PGVGLAPGVG	VAPGVGVAPG	VGVAPGIAPA	AAAAAKAAA	AAQFGLVPGV
500	510	520	530	540	550	560
GVAPGVGVAP	GVGVAPGVGL	APGVGVAPGV	GVAPGVGVAP	GIAPGGMVEK	FVGTWKIADS	HNFGGEYLKAI
570	580	590	600	610	620	630
GAPKELSDGG	DATTPTLYIS	QKDGDKMTVK	IENGPPTFLD	TQVKFKLGEE	FDEFPSDRRK	GVKSVVNLVG
640	650	660	670			
EKLVYVQKWD	GKETTYVREI	KDGKLVVTLT	MGDVVAVRSY	R RATE		



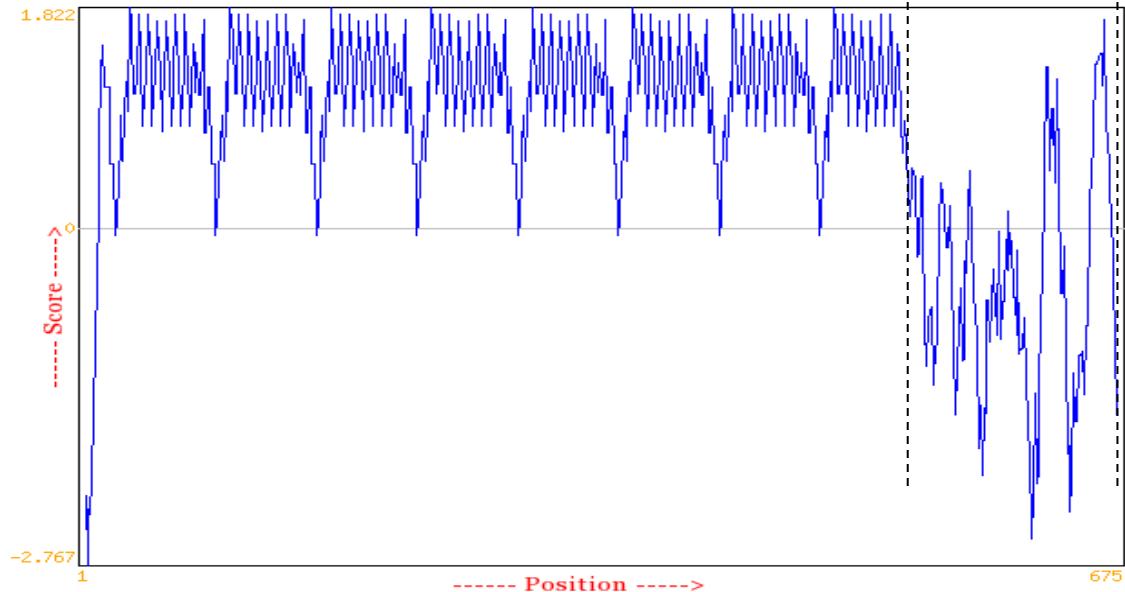
- Helix
- Sheet
- Turn
- Coil

10 GSAAAAAAAAA 20 KAAAKAAQFG 30 LVPGVGVAAPG 40 VGVAPGVGVA 50 PGVGLAPGVG 60 VAPGVGVAPG
 80 VGVAPGIAPA 90 AAAAAKAAAK 100 AAQFGLVPGV 110 GVAPGVGVAP 120 GVGAPGVGL 130 APGVGVAPGV 140 GVAPGVGVAP
 150 GIAPAAAAAA 160 KAAAKAAQFG 170 LVPGVGVAAPG 180 GVAPGVGVVA 190 PGVGLAPGVG 200 VAPGVGVAPG 210 VGVAPGIAPA
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 430 LVPGVGVAAPG 440 VGVAPGVGV 450 PGVGLAPGVG 460 VAPGVGVAPG 470 VGVAPGIAPA 480 AAAAAKAAAK 490 AAQFGLVPGV
 500 GVAPGVGVAP 510 GVGAPGVGL 520 APGVGVAPGV 530 GVAPGVGVAP 540 GIAPGGMVEK 550 FVGTWKIADS 560 HNFGEYLKAT
 570 GAPKELSDGG 580 DATTPTLYIS 590 QKDGDKMTVK 600 IENGPPTFLL 610 TQVKFKLGEE 620 FDEFPSDRRK 630 GVKSVVNLVG
 640 EKLVYVQRWD 650 GKTETYVREI 660 KDGKLVVVLT 670 MGDVVAVRSY RRATE

UNAG



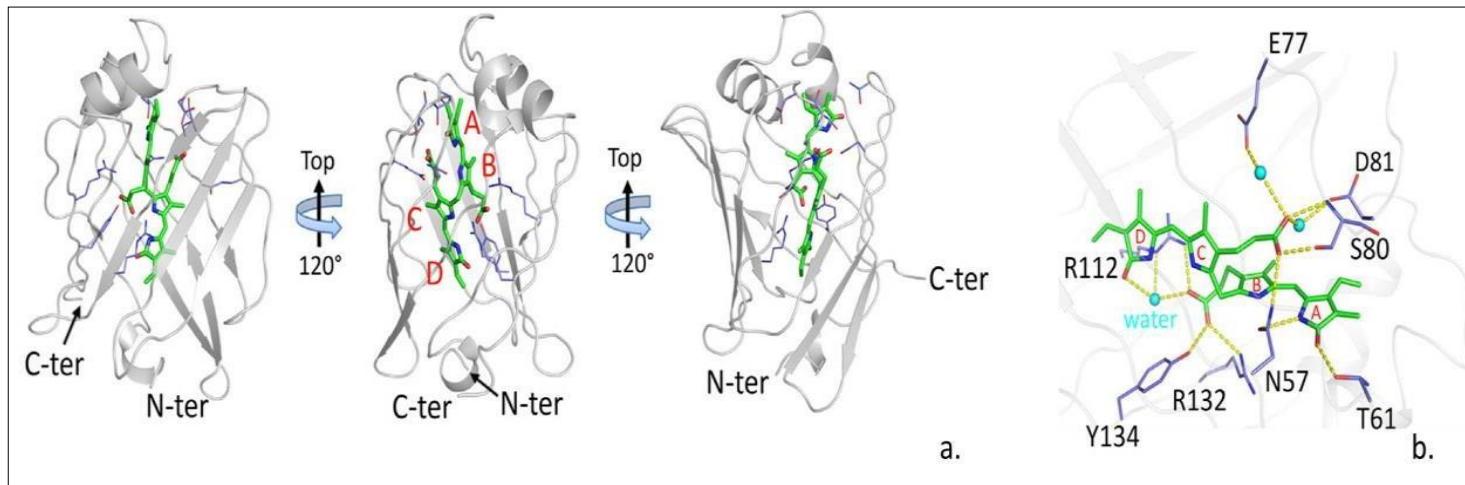
Kyte & Doolittle Hydrophobicity Plot



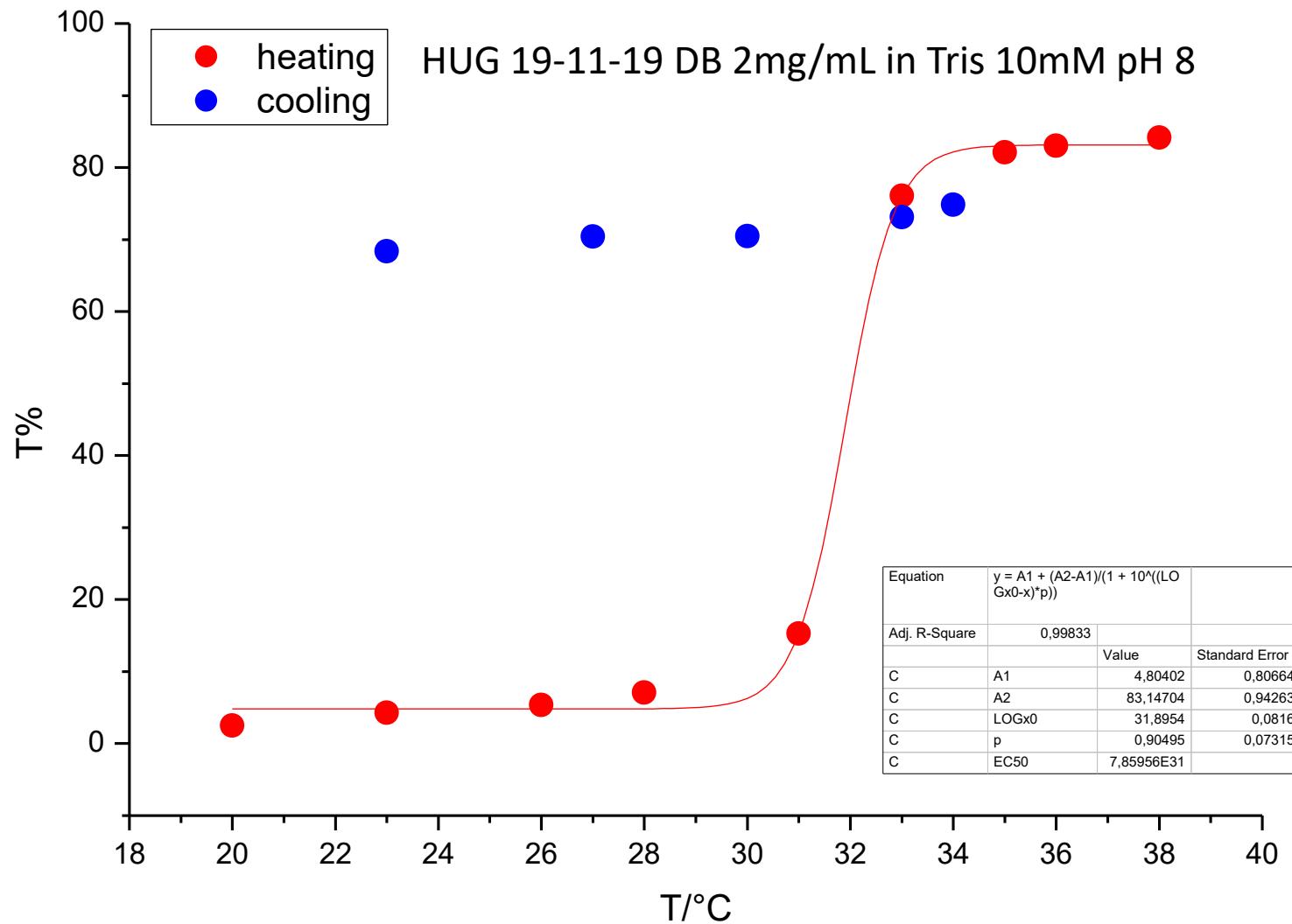
The **biosensor HUG** for the quantitative analysis of unconjugated bilirubin was developed in our laboratory (Dr. Antonella Bandiera and co-workers):

This new fusion protein preserves:

- 1) the reverse phase transition property of HELP
- 2) the highly specific capacity of UnaG to bind bilirubin and to become fluorescent.



The reverse thermal transition of HUG



CHARACTERIZATION OF HUG

- Molecular mechanics and dynamic simulation**
- Circular Dichroism (CD)**
- Differential Scanning Calorimetry (DSC)**
- Potentiometric titration of HUG and HELP**

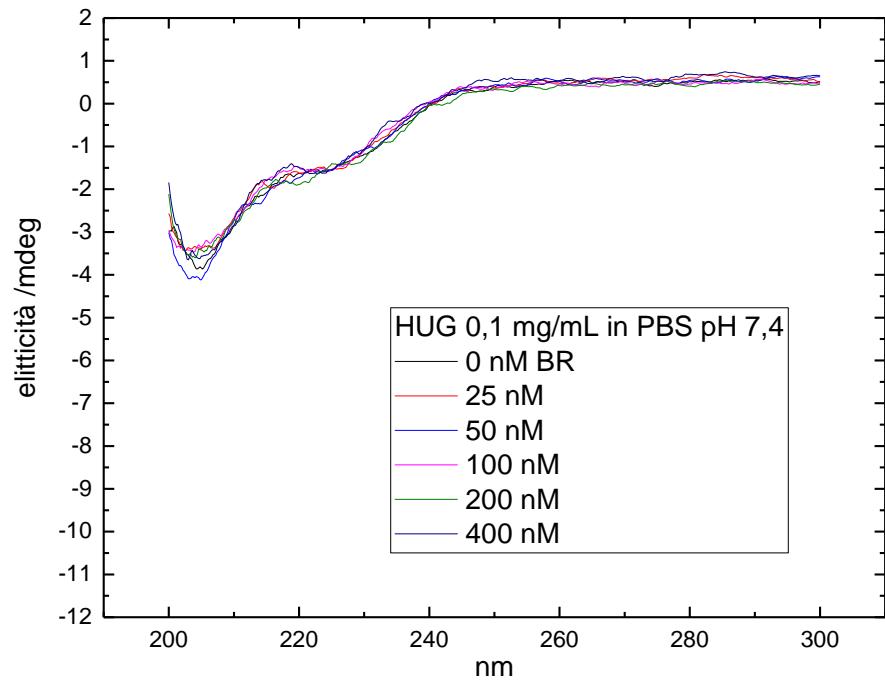
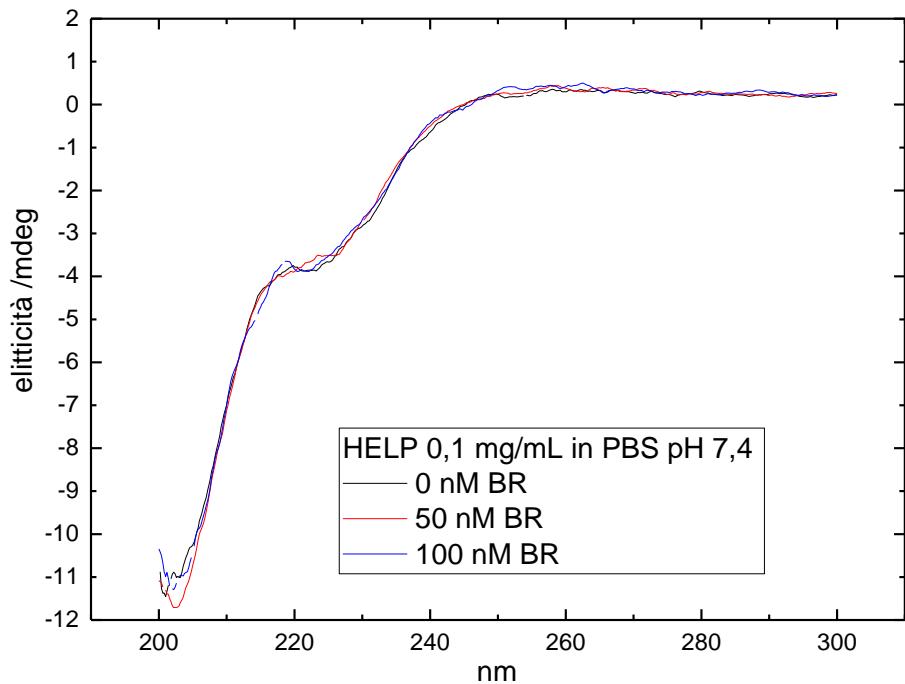
The physical and chemical characterization of the new fluorescent biosensor HUG. This step is very important for optimization of liquid assay of free bilirubin in biological samples

Analyzing quantitatively the ligand binding to macromolecule it is possible to focus on the application and importance of HUG biopolymer for bilirubin quantitative assessment in biological systems

Fluorescence spectroscopy is one of the most convenient methods for the evaluation of binding processes. Since the intensity and wavelength are very sensitive to the change of the environment due to the ligand binding, variation in fluorescence intensity as a function of ligand concentration provide information about the strength of the protein-ligand interaction.

CIRCULAR DICHROISM

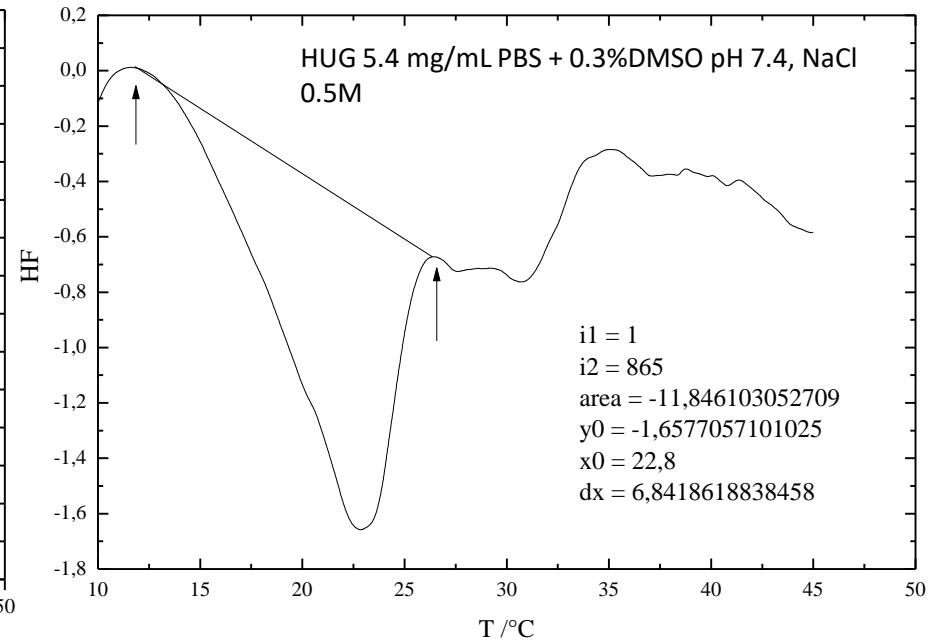
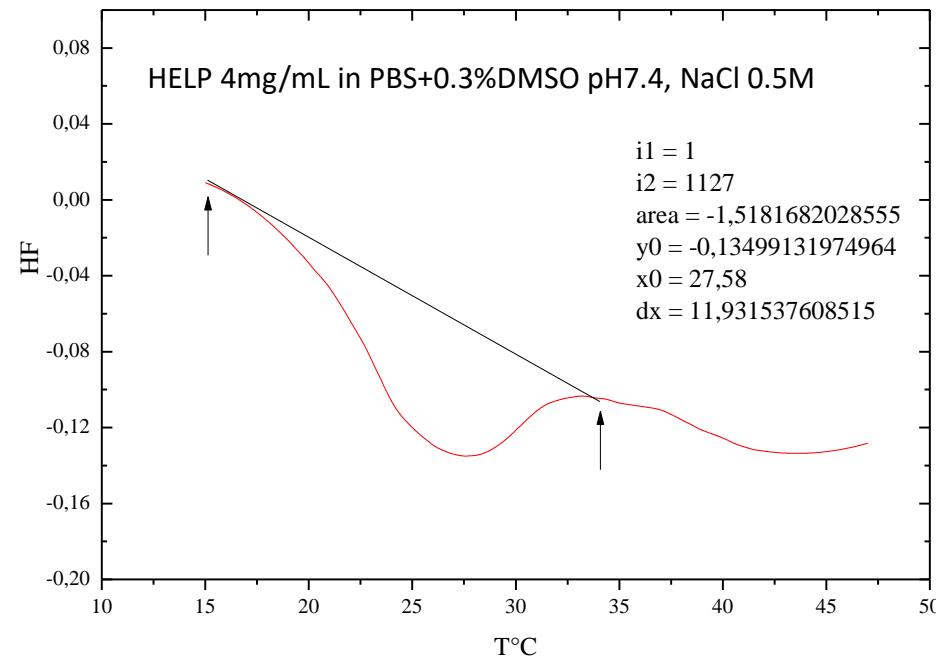
CD is an excellent tool for rapid determination of the secondary structure and folding properties of proteins that have been obtained using recombinant techniques. The most widely used applications of protein CD are to determine whether an expressed protein is folded. In addition, it can be used to study protein-ligand interactions.



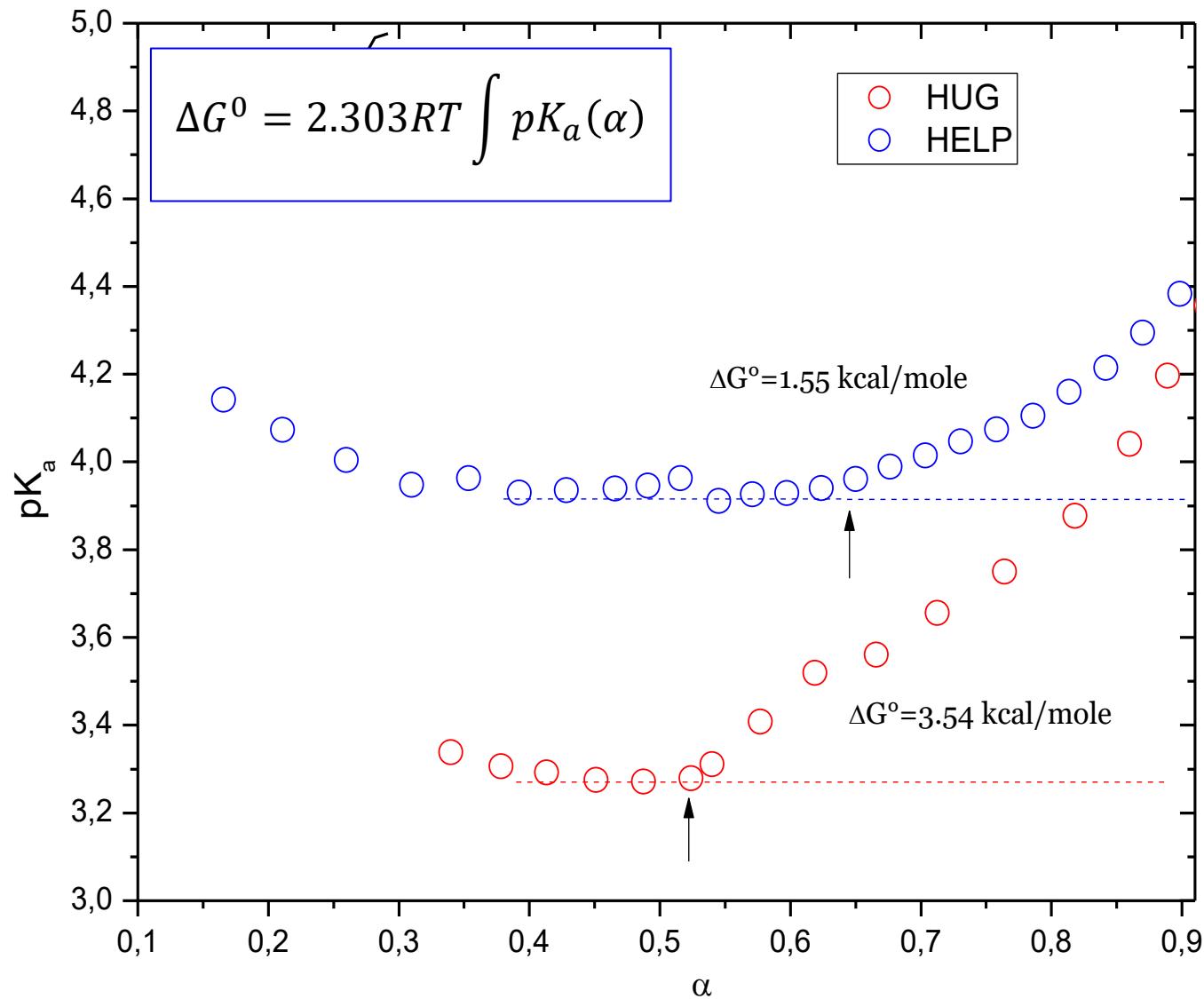
DSC SETARAM (preliminary results)

The DSC approach employs a reference cell and a sample cell heated at a controlled rate. It can detect some of transitions such as melts, glass transitions, phase changes.

Calorimetric methods demonstrate an advantage over other methods in their ability to provide a relatively full picture of thermodynamic parameters during binding reactions, including the equilibrium binding constants, the enthalpy of binding reactions (ΔH) and the entropy change (ΔS).

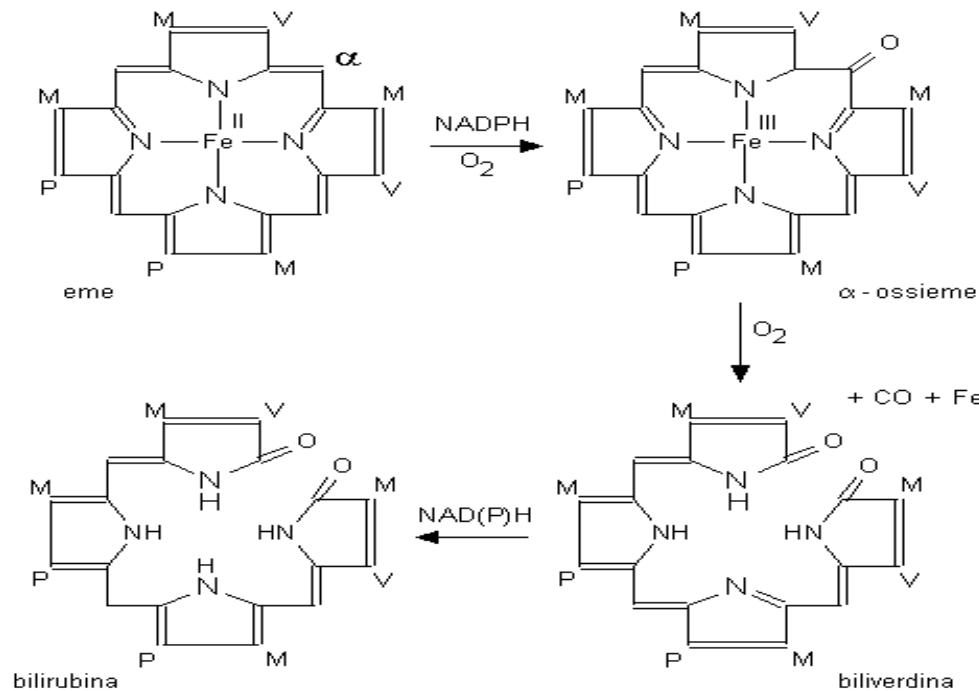


Potentiometric titration of HUG and HELP



La bilirubina (dal latino bilis = bile e ruber = rosso) è un pigmento di colore rosso-arancione che deriva per l'80% dal catabolismo dell'eme dell'emoglobina (Hb) e per il restante 20% da altre emoproteine, come le mioglobine e i citocromi.

La bilirubina si ottiene dal processo di degradazione della ferroporfirina, o gruppo eme, dell'emoglobina, rilasciata durante la distruzione dei globuli rossi, o eritrociti, nella milza.



il metabolismo della bilirubina non coniugata avviene a livello epatico, dopo essere stata trasportata nel sangue da una proteina di trasporto, l'albumina, in virtù della sua insolubilità in acqua (liposolubilità);

BILIRUBIN

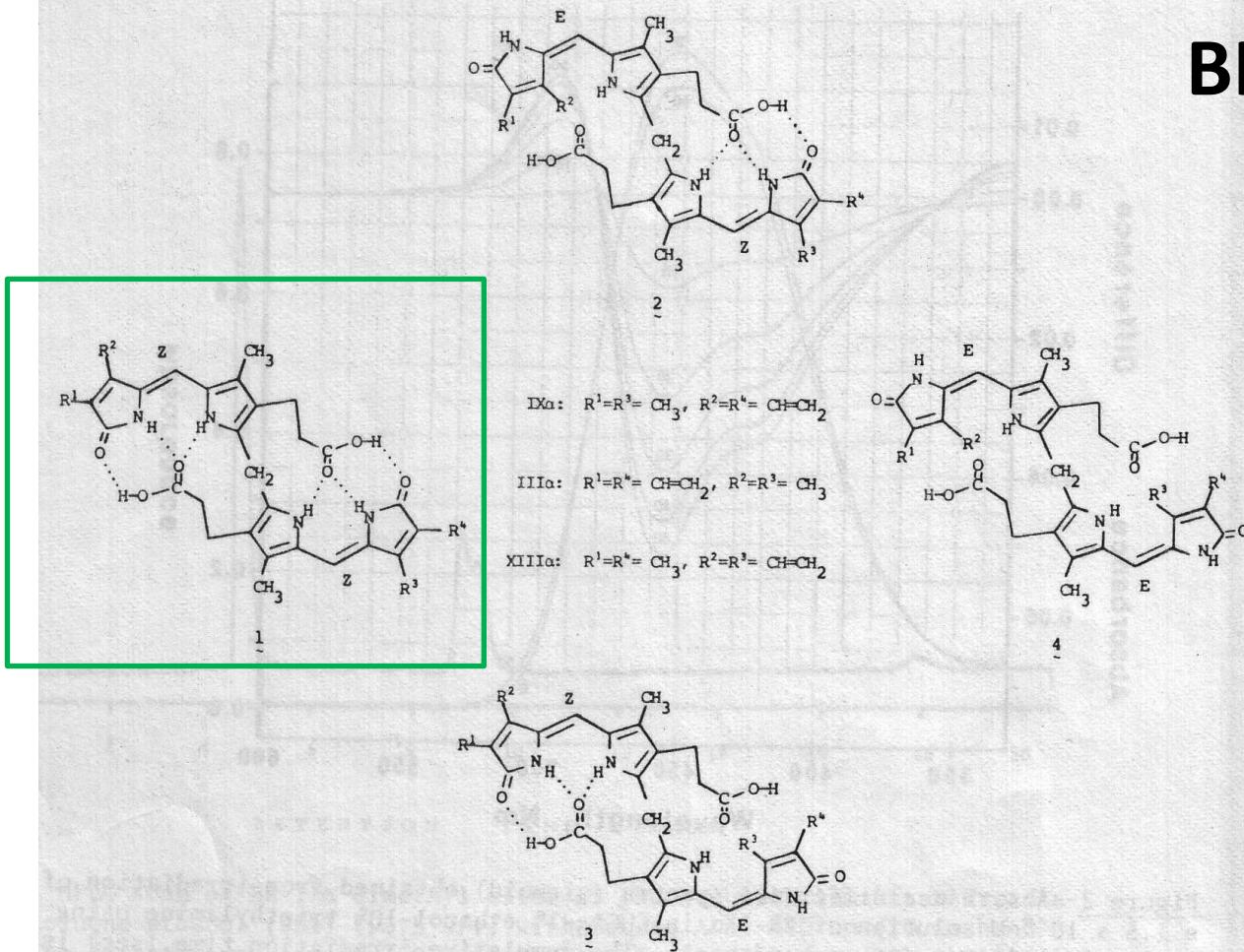
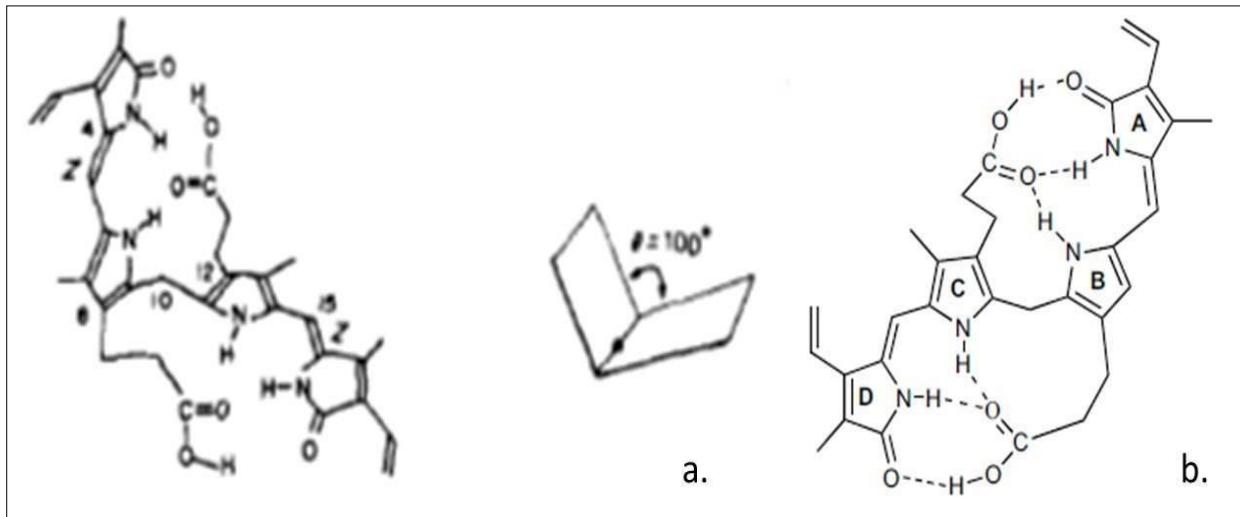


Figure 1 Configurational isomers of BR showing geometric isomerism about the 5,15 *meso* double bonds: 1 = Z-Z, 2 = E-Z, 3 = Z-E and 4 = E-E. The Z-Z configuration (1) is the stable ground state structure of BR (1). The E-Z (2), Z-E (3) and E-E (4) structures are the photochemically accessible isomers which, in the case of BR-IX α , we call collectively PBR. The E-Z (2) and Z-E (3) isomers are identical for (symmetrical) BR-III α and BR-XIII α but not for BR-IX α . In the corresponding BR dimethyl esters, the propionic acid groups become methyl propionate groups.



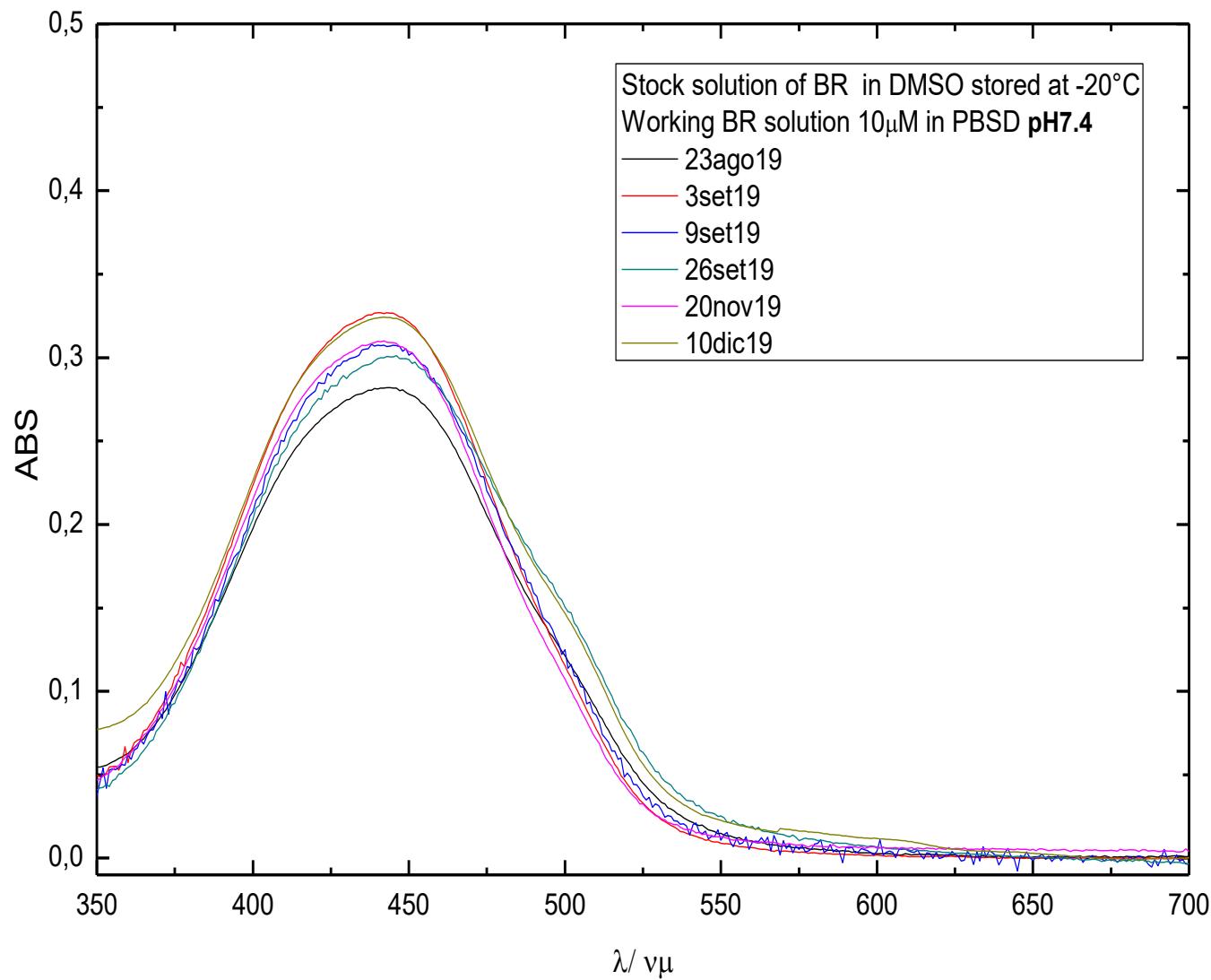
we assessed the bilirubin stability at given experimental conditions:

BR stability			
pH	Temperature	Solvent	Time
7.4	-20°C	DMSO	0 min
	4°C	PBSD	120 min mesi

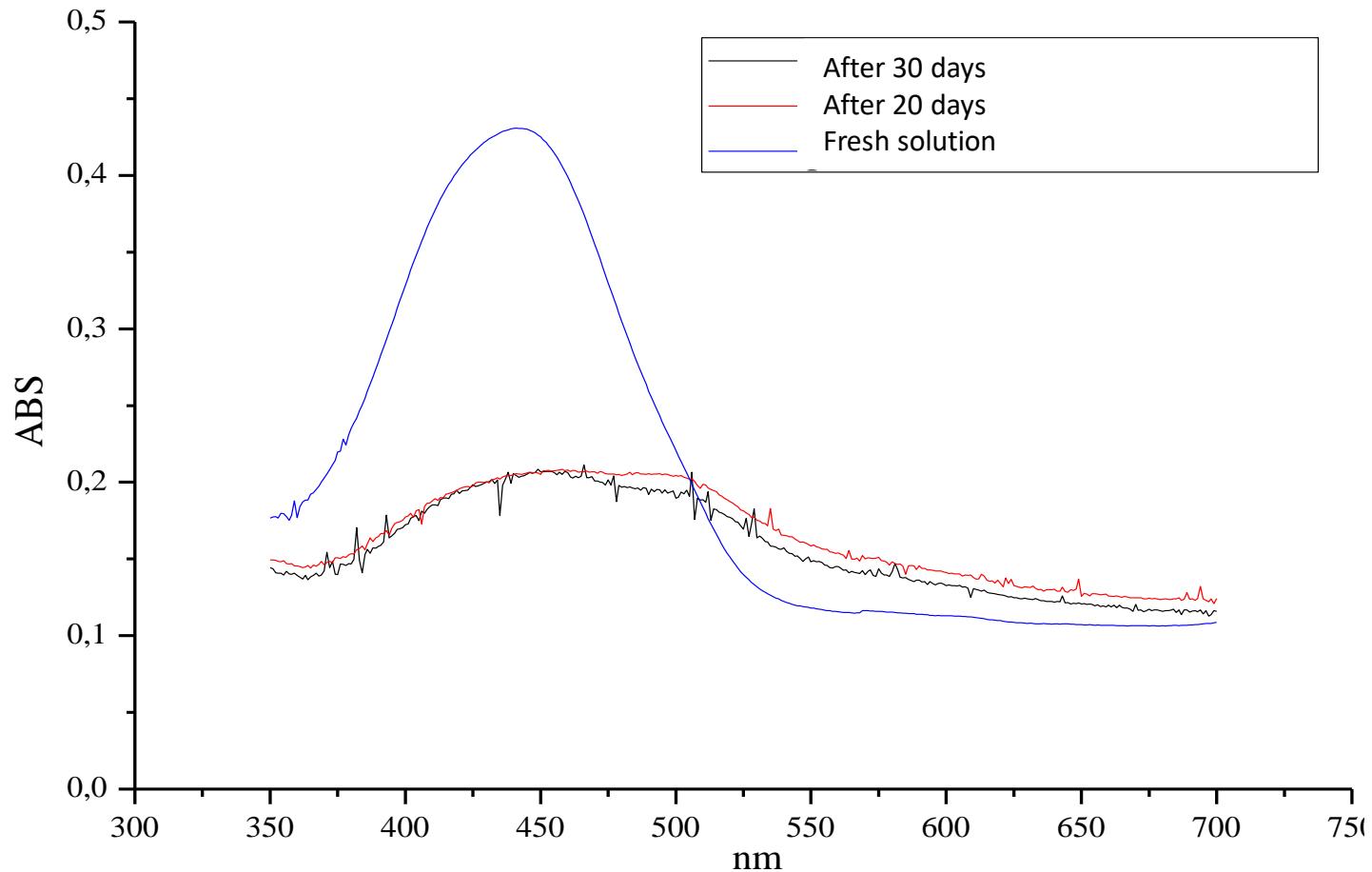
Stock solution of BR is prepared in 100% DMSO and it is stored up to -20°C. Stability assay was done on working bilirubin solution (**BR 10µM in PBS + 0.3%DMSO**). This mother solution is used to prepare the standard solutions.

NB “To prevent the denaturation of UnaG, the final concentrations of DMSO were at most 0.2% (vol/vol) in all experiments.” Shitashima 2017

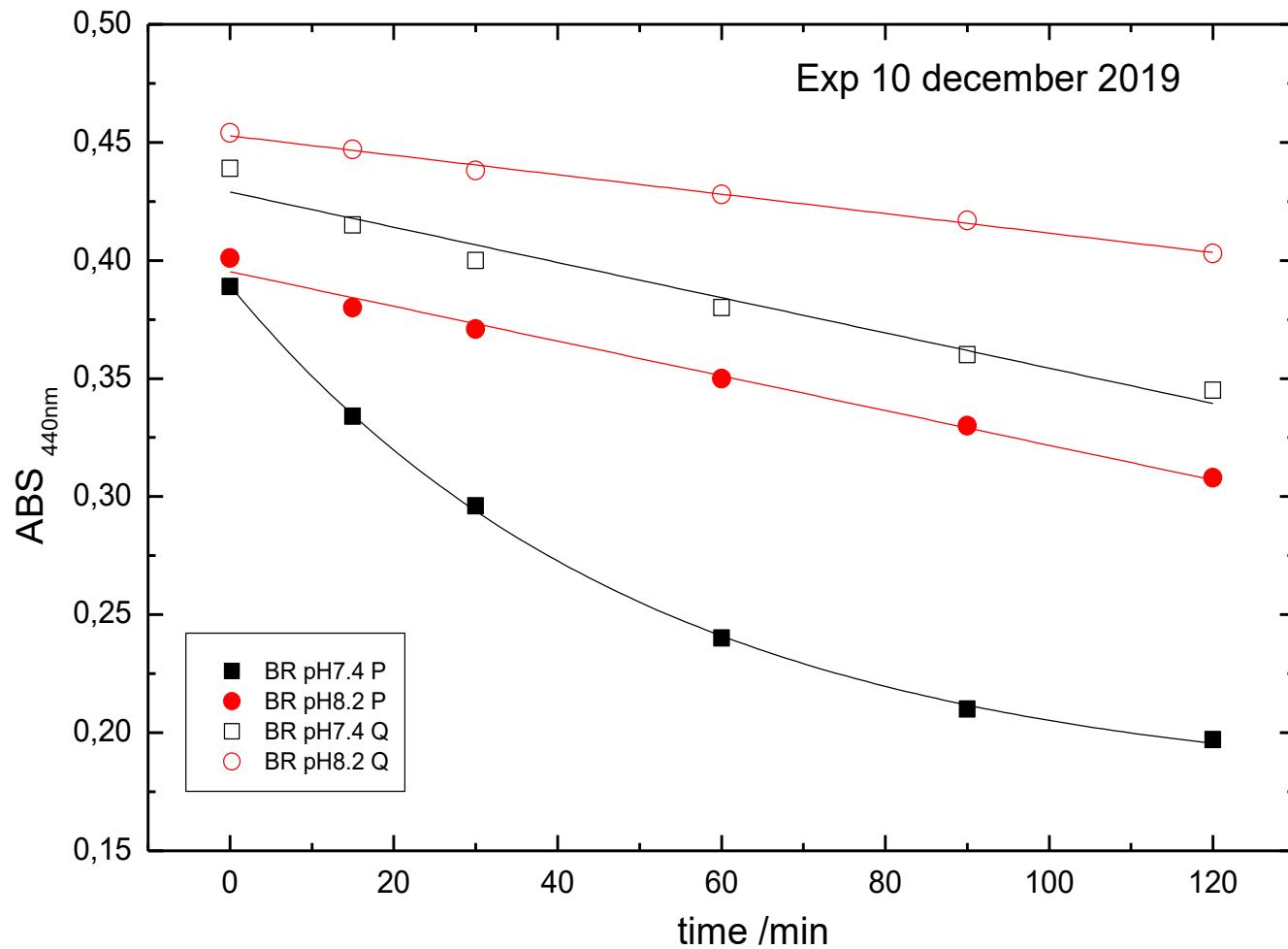
BR is stable when the solution is stored at -20°C



Working solution (PBSD pH 7.4) is not stable if it is stored at -20°C

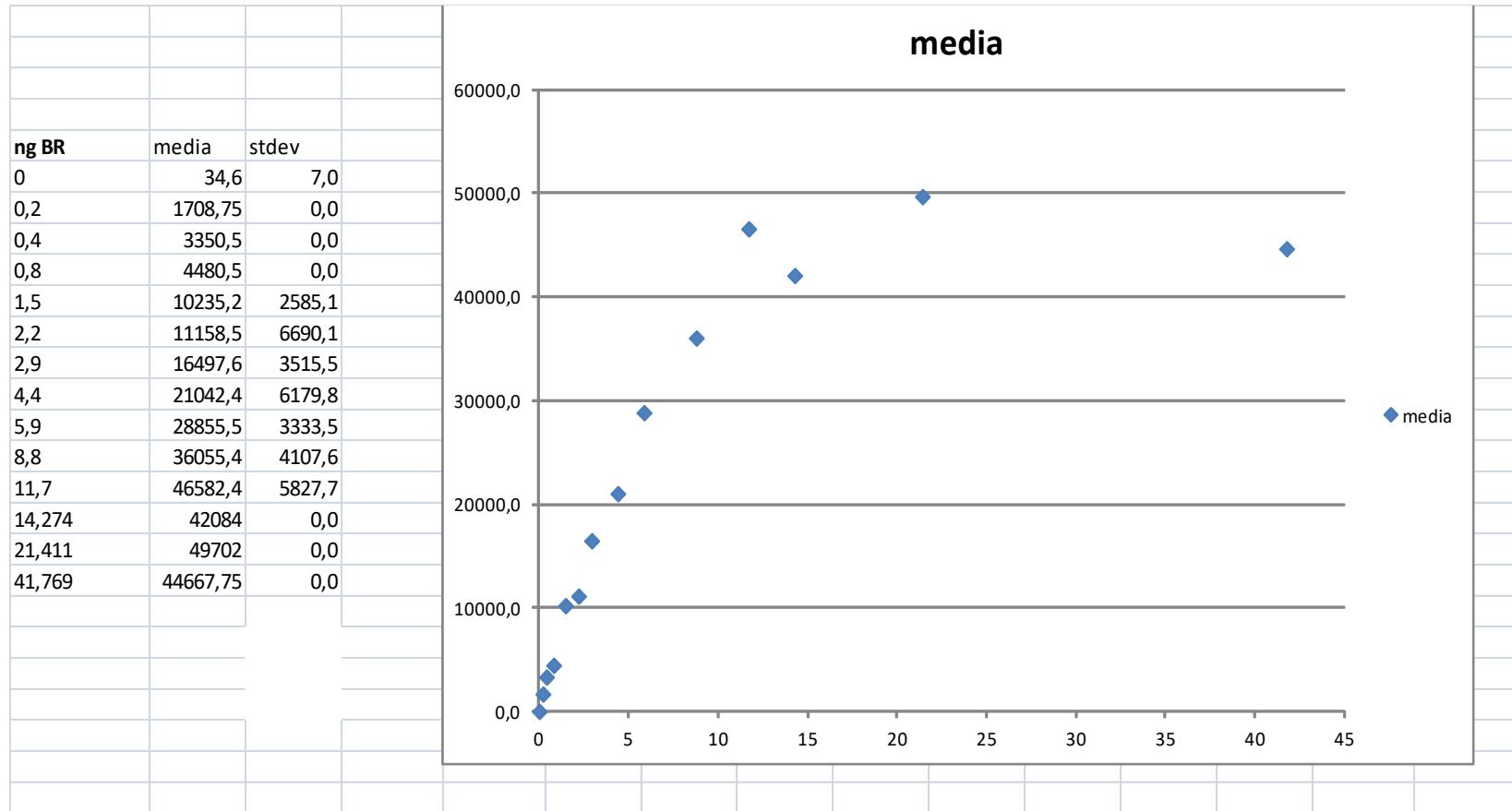


BR stability is influenced by surface effects (plastic or quartz cuvettes)



HUG-BR interaction

The estimate of unconjugated bilirubin concentration has been performed on 96-wells plate in liquid or solid phase (PBSD pH 7.4)



Whole Saliva

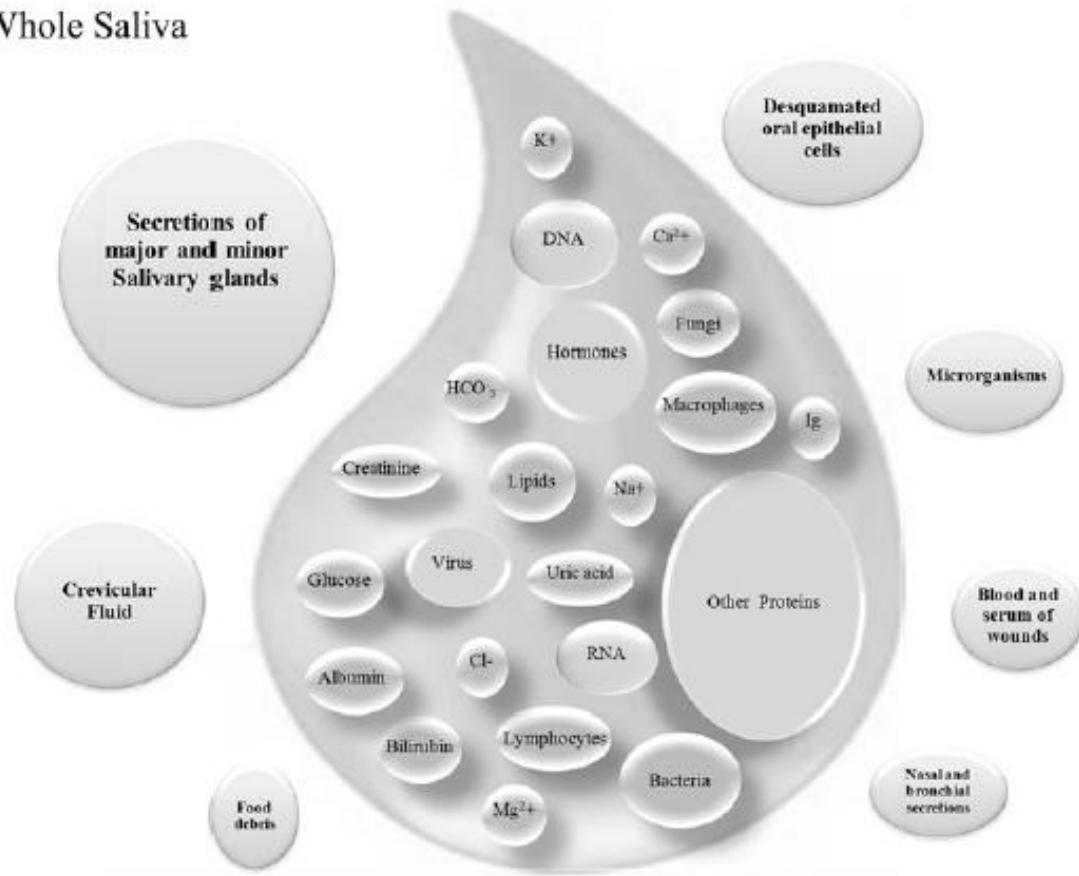


FIG. 1. Schematic representation of the components of whole saliva. The size of each individual droplet is an approximate representation of its concentration in whole saliva.

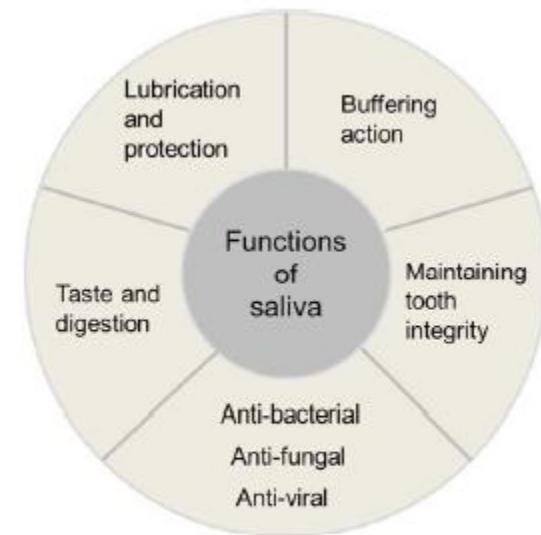


Figure 1: Main functions of saliva

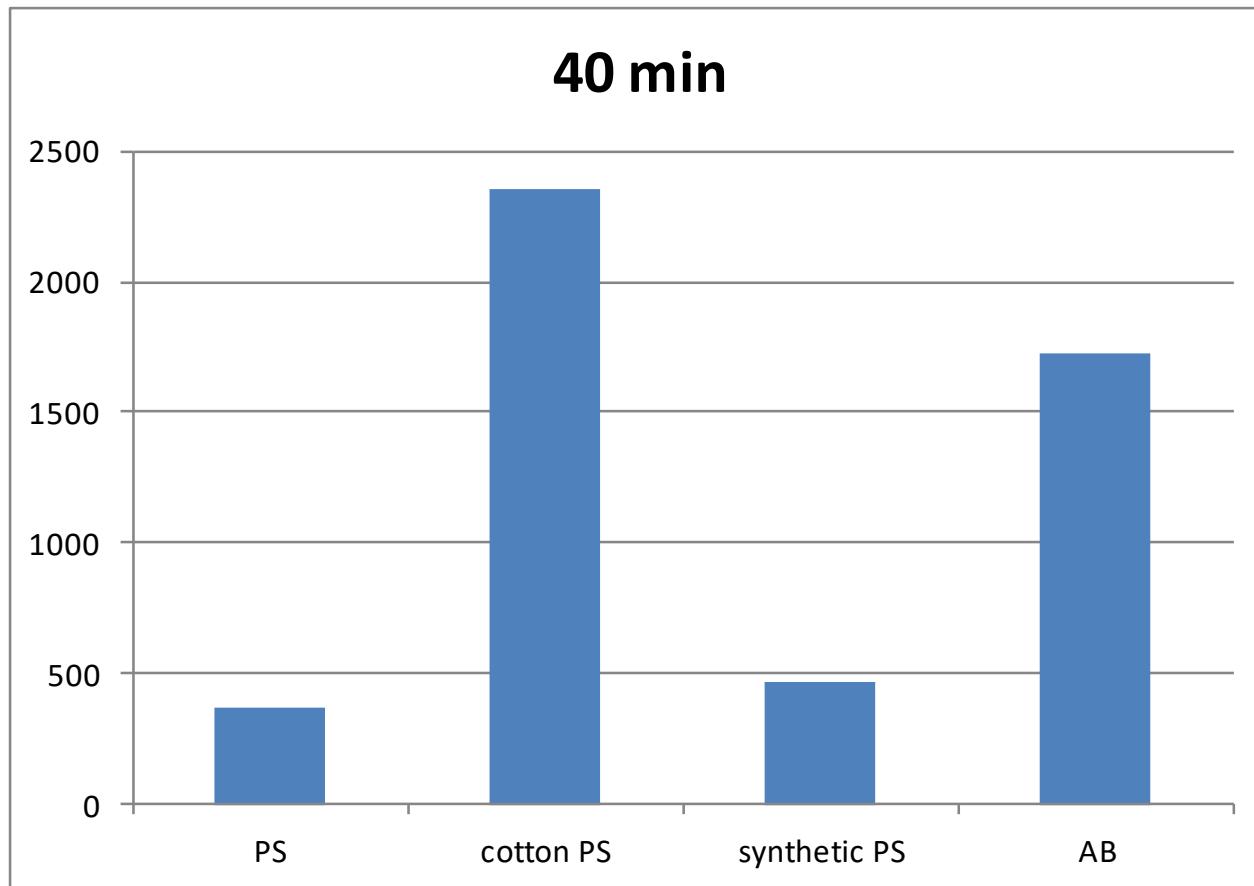
Table 1 Components of authentic human saliva and a comparison of the normal range of the concentrations between saliva and other biological fluids

Real saliva compositions		Normal range		Ref
		Saliva	Other biological fluids	
1. Inorganic compounds	Na ⁺	20-80 mmol/L	Plasma 145 mmol/L	11
	K ⁺	20 mmol/L	4 mmol/L	
	Ca ²⁺	1-4 mmol/L	2.2 mmol/L	
	Cl ⁻	30-100 mmol/L	120 mmol/L	
	HCO ₃ ⁻	15-80 mmol/L	25 mmol/L	
	Phosphate	4 mmol/L	1.2 mmol/L	
	Mg ²⁺	0.2 mmol/L	1.2 mmol/L	
	SCN ⁻	2 mmol/L	< 0.2 mmol/L	
	NH ₃	3 mmol/L	0.05 mmol/L	
2. Organic compounds (non-protein and lipids)	Uric acid	3.38 ± 0.21 mg/dL 217.2±110.3 mol/L 0.1-7.5 mg/dL	Serum 6.31±0.24 mg/dL	21-23
	Bilirubin	0.5-5.0 μmol/L	Serum 0.2-1.2 mg/dL	
	Creatinine	0.12 ± 0.06 mg/dL 0.05-0.2 mg/dL	Serum 0.89 ± 0.17 mg/dL Serum 0.6-1.5 mg/dL	
	Glucose	91.3±10.1 mg/dL 4-13 mg/dL	Plasma 80-120 mg/dL	
	Cholesterol	0.02-5.46 μmol/L	Serum <5 mmol/L	
	Lactate	0.3-1.8 mM 0.1 to 2.5 mmol/L	Serum 0.5-1.0 mM	
	a-Amylase	19-308 U/mL* 93±62 U/L * 2.64±1.8 mg/mL	Serum 0.05-0.125 U/mL* 93±62 U/L *	23, 30
	Albumin	0.2±0.1 mg/mL	Serum 3.5-5.5 g/dL	
	Secretory-IgA	80-717 mg/dL 124.3-333.5 μg/mL	Serum 70-400 mg/dL	
	Mucins group	MUC5B: 2.4±1.7 U/mL 1.19±0.17 mg/mL	Serum 9.9 ± 0.8 ng/ml	
	Lysozyme	3-50 μg/mL 59.7 to 1062.3 μg/ml	Serum 7.4 ± 1.8 mg/mL Serum 4-9 μg/mL	
3. Protein/Polypeptide compounds	Total proteins	7.1-223.2 mg/dL 0.9±0.2 mg/mL	Serum 6-8 g/dL	23, 31
	Cortisol	3.5-27.0 mg/dL	Serum 2-25 mg/dL	
	Testosterone	32-55 pg/mL	Serum 320-600 ng/dL	
	Progesterone	Luteal phase 436 ±34 pmol/L Follicular phase 22.1±2.7 pmol/L	Serum Male: < 1 ng/mL Serum Female: 0.1-20 ng/mL	
	Estrogen(Estra diol)	Luteal phase 20.6 ±0.4 pmol/L	Serum Male: 15-60 pg/mL Serum Female: 15-370 pg/mL	
4. Hormones				35, 36, 37

*U/mL: enzymatic activity per unit (mL) of saliva

Preliminary tests on saliva

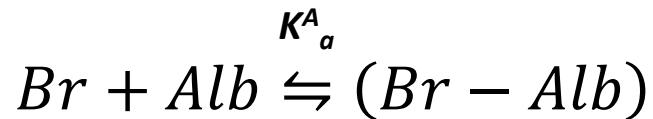
Test with 10 uL of HUG and 150 uL of saliva



ATTENZIONE: non e' stata considerata la fluorescenza spontanea della saliva!!!

Binding of bilirubin to HUG protein

The Br-HUG interaction was studied by mean of the fluorescence titration technique assessing the intensity enhancement during ligand addition

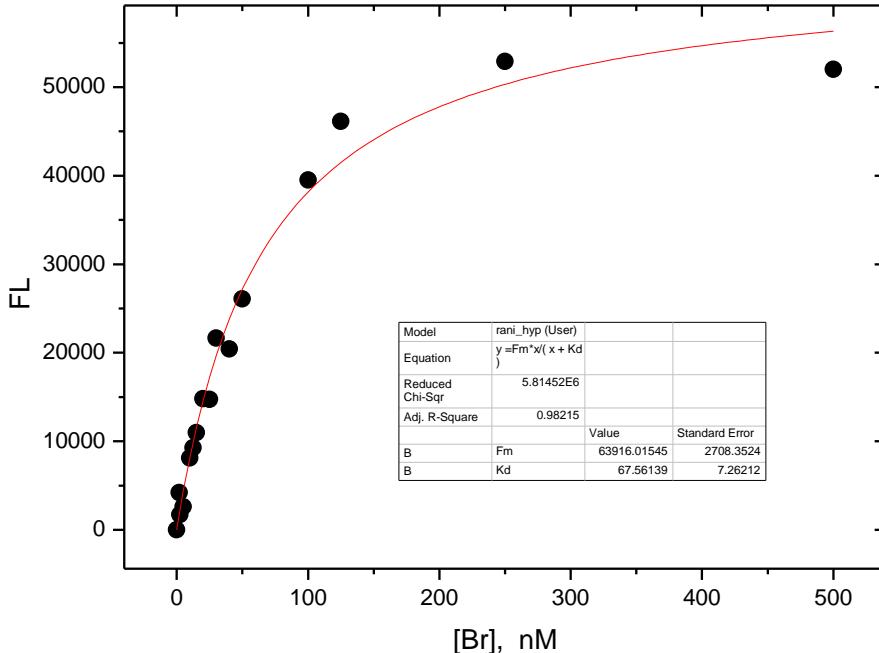


+

HUG

$$\Downarrow K_a = 1/K_D$$

$(Br - HUG)$ **fluorescence**



For each fluorescence value (F), the fractional enhancement (Y) was computed by the equation:

$$Y = \frac{F}{F_0}$$

where Y is the fractional saturation related to the extent of binding and F_0 the fluorescence intensity at the Br/HUG ratio greater than 1 (the asymptotic value).

$$K_D = \frac{[HUG][Br]}{[Br - HUG]}$$

During the binding titration the Br concentration is increased so that saturation Y is expressed in terms of the Br-HUG complex concentration, $[Br - HUG]$ as:

$$Y = \frac{[Br - HUG]}{[HUG]_T} \quad (1)$$

where $[HUG]_T$ is the total HUG concentration used for the measurement. Then:

$$[Br - HUG]_{\square} = \frac{[HUG][Br]}{K_D}$$

$$[HUG]_T = [HUG] + [Br - HUG] = P_T$$

$$Y = \frac{[HUG][Br]/K_D}{[HUG] + [HUG][Br]/K_D}$$

$$Y = \frac{[Br]}{K_D + [Br]}$$

Since in the binding measurement instead of free concentrations the total protein and ligand concentrations are known, an expression of Y as a function of total quantities is derived as follows:

$$[Br] = [Br]_T - [Br - HUG] = L_T - [Br - HUG]$$

$$[HUG] = P_T - [Br - HUG]$$

then:

$$K_D = \frac{(L_T - [Br - HUG])(P_T - [Br - HUG])}{[Br - HUG]}$$

$$K_D [Br - HUG] = P_T L_T - (P_T + L_T) [Br - HUG] + [Br - HUG]^2$$

$$[Br - HUG]^2 - (P_T + L_T + K_D) [Br - HUG] + P_T L_T = 0$$

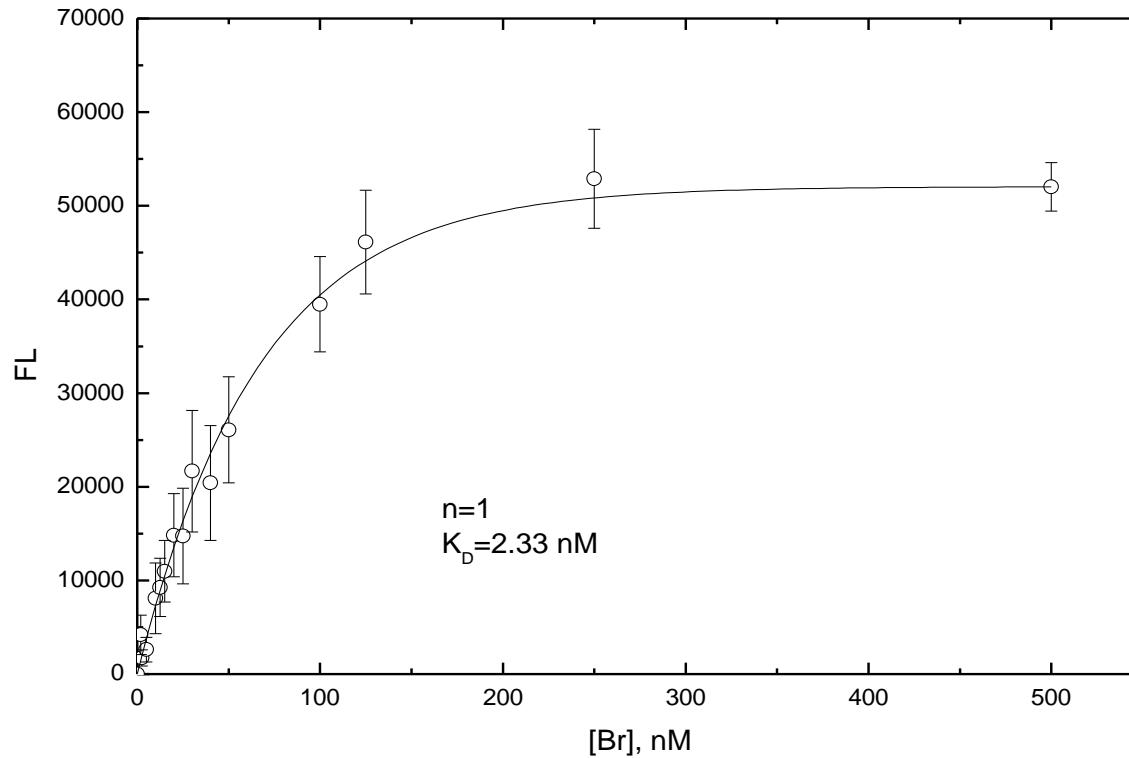
The root of the equation is:

$$[Br - HUG] = \frac{(P_T + L_T + K_D) - \sqrt{\{(-(P_T + L_T + K_D))^2 - 4P_T L_T\}}}{2}$$

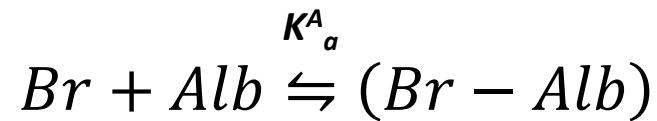
By substitution in the equation (1):

$$Y = \frac{(P_T + L_T + K_D) - \sqrt{\{(-(P_T + L_T + K_D))^2 - 4P_T L_T\}}}{2P_T} \quad (2)$$

The plot of $F=F_o Y$ versus $L_T = [Br]$ is shown in Figure where average values of all the results obtained by several experiments are reported.



By nonlinear least-squares fitting of the hyperbolic curve the best evaluation of equilibrium dissociation constant ($K_D=2.33 \text{ nM}$) and of the maximal fluorescence value of the ligand-bound protein ($F_o=56300$) was obtained



+

HUG

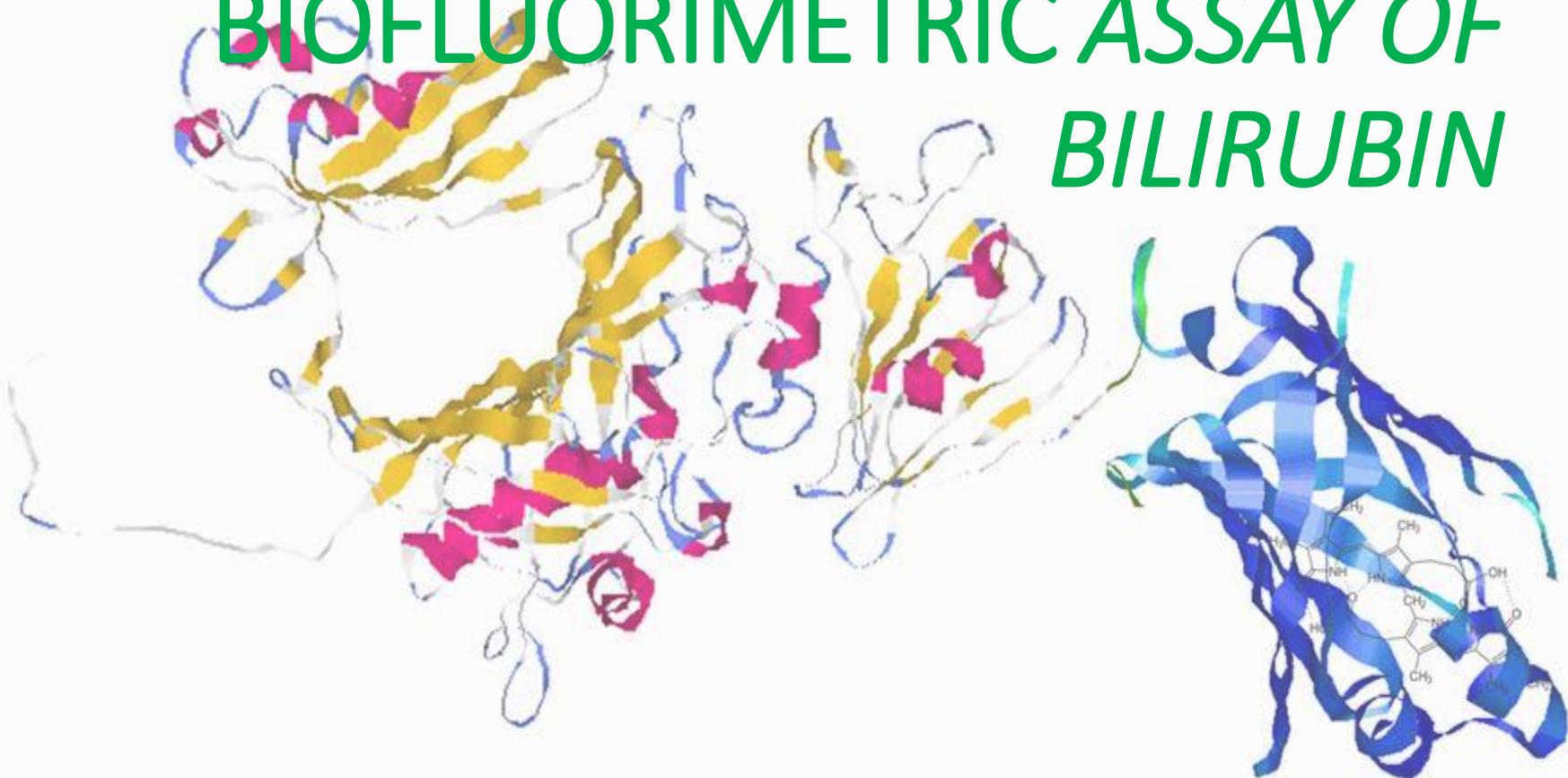
$$\downarrow \uparrow K_a = 1/K_D$$

$(Br - HUG)$ fluorescence

$K_a = 1/K_D = 4.3 \times 10^8 \text{ M}^{-1}$ per $Br-HUG$

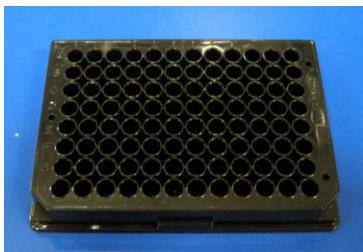
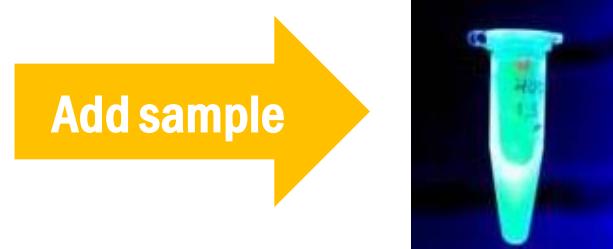
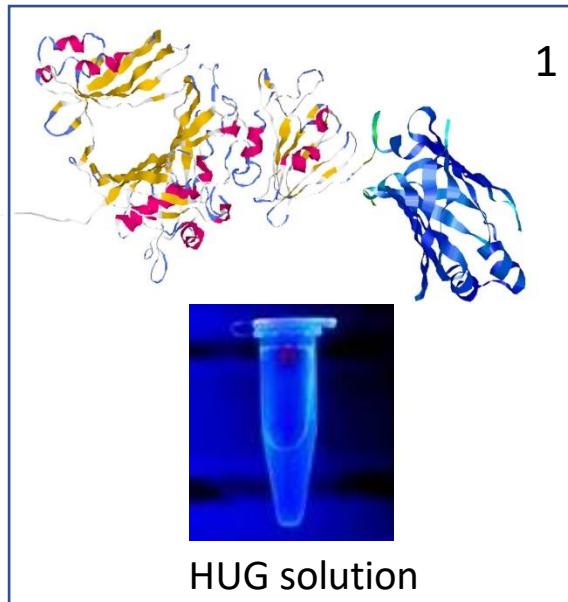
$K_a^A = 1/K_D = \text{ca } 10^6 \text{ M}^{-1}$ per $Br-Alb$

BIOFLUORIMETRIC ASSAY OF *BILIRUBIN*



BIOFLUORIMETRIC ASSAY OF SERUM BILIRUBIN

- Direct analysis of indirect bilirubin by HUG



Future goals:

1. Molecular Dynamics simulation of the single protein (HUG and HELP),
2. Molecular Dynamics of binding process (HUG-BR),
3. A thorough study of the reverse thermal transition of HUG in comparison to HELP (which is a critical property for a future upscale of biopolymer production),
4. Bilirubin assay on fish serum samples.

