

CONTROLLED FISSION REACTIONS AND REACTORS

i.e. How can we use the fission to produce energy in a controlled way.

Let's consider a infinitely large mass of Uranium, which for now we assume to be of natural isotopic composition ($0,72\% {}^{235}\text{U}$, $99,28\% {}^{238}\text{U}$)

A single fission will produce 2,5 neutrons on average.

Each "second-generation" neutron is capable to produce other fissions producing each time more neutrons. This is called "chain reaction"

Each fission releases $\approx 200 \text{ MeV}$ in form of kinetic energy and radiation.

It can be convenient to define a neutron reproduction factor k_{∞} (for infinite medium \Rightarrow ignoring the loss of neutron at the surface)

Each thermal neutron produces k_{∞} new thermal neutron.

For a reaction chain to continue we must have $k_{\infty} \geq 1$.

Although for each reaction there 2.5n, there are fast neutrons

Neutron lose energy in elastic collisions with nuclei. A popular choice for moderator is Carbon in form of graphite blocks.
(Why? C is solid, inexpensive and easy to handle).

A lattice of blocks of U alternated with graphite is called "chain-react pile", and the first one was constructed by Fermi in 1942.

If $K = 1$ \Rightarrow The pile is critical \longrightarrow Steady release of energy
 $k < 1$ \Rightarrow The pile is subcritical
 $k > 1$ \Rightarrow " supercritical

Let's assume we have N thermal neutrons at the present generation. Even if we will not have vN fast fission neutrons immediately available some will be absorbed by many processes, mainly (n, γ) in both ^{235}U and ^{238}U .

M is defined as mean number of fission neutron produced per original thermal neutron. $M < v$ in general

If σ_f is ^{the} fission cross-section, ^{and} σ_a the cross-section for absorptive processes the relative probability for a neutron to cause fission is

$$P = \frac{\sigma_f}{\sigma_f + \sigma_a}$$

and then

$$M = v \frac{\sigma_f}{\sigma_f + \sigma_a}$$

For ^{235}U $\sigma_f = 58 \text{ b}$ $\sigma_a = 97 \text{ b}$ $\Rightarrow M = 2.08$ fast neutrons are produced per thermal neutron

^{238}U is not fissionable so $\sigma_f = 0$ and $\sigma_a = 2.75 \text{ b}$. For a natural mixture of Uranium:

$$\sigma_f = \frac{97.2}{100} \sigma_f(235) + \frac{99.28}{100} \sigma_f(238) = 4.20 \text{ b}$$

$$\Sigma_a = \frac{0.72}{100} \Sigma_a(235) + \frac{99.28}{100} \Sigma_a(238) = 3.65 \text{ b}$$

$$\Rightarrow M_{\text{NAT}} = 1.33$$

Very close to 1, so we should minimize the neutrons to be lost in order to obtain a critical reactor.

If enriched Uranium is used ($^{235}\text{U} = 3\%$) then $M = 1.84$, allowing more neutron to be lost by other means to maintain the criticality conditions.

At this point the N thermal neutrons have been partially absorbed and now there are $M N$ neutrons to slow down to thermal energies.

These fast neutrons can fission with ^{238}U ($\Sigma \approx 1 \text{ b}$), increasing the number of neutrons \rightarrow A factor ϵ (fast fission factor) has to be taken into account $\rightarrow M \epsilon N$ neutrons are present ($\epsilon = 1.03$ for natural Uranium). Moderation of n is accomplished by light moderator.

It takes ≈ 100 collisions ^{in carbon} to moderate a MeV neutron down to thermal energy.

But, in the process, n must pass through the resonance region with high Σ (up to $1000 \text{ b} \gg \Sigma_{\text{fission}}$) \Rightarrow If we want thermal neutrons we have to avoid resonance capture at all.

If C and U are mixed in powder this is impossible to achieve, but if there is enough C n can scatter inside it avoiding resonance region. About 20 cm of C are needed to moderate MeV n down to 10^{-3} eV .

Nevertheless there will be a probability p that some n are captured.

$p \approx 0.9$. The # of U now is $n_{\text{heat}} = M \epsilon p N$

Now that n are thermalized they must interact with ^{235}U .

Note that n can still interact with C ($\Sigma_{\text{fission}} = 0.0036 \text{ b}$), because there is

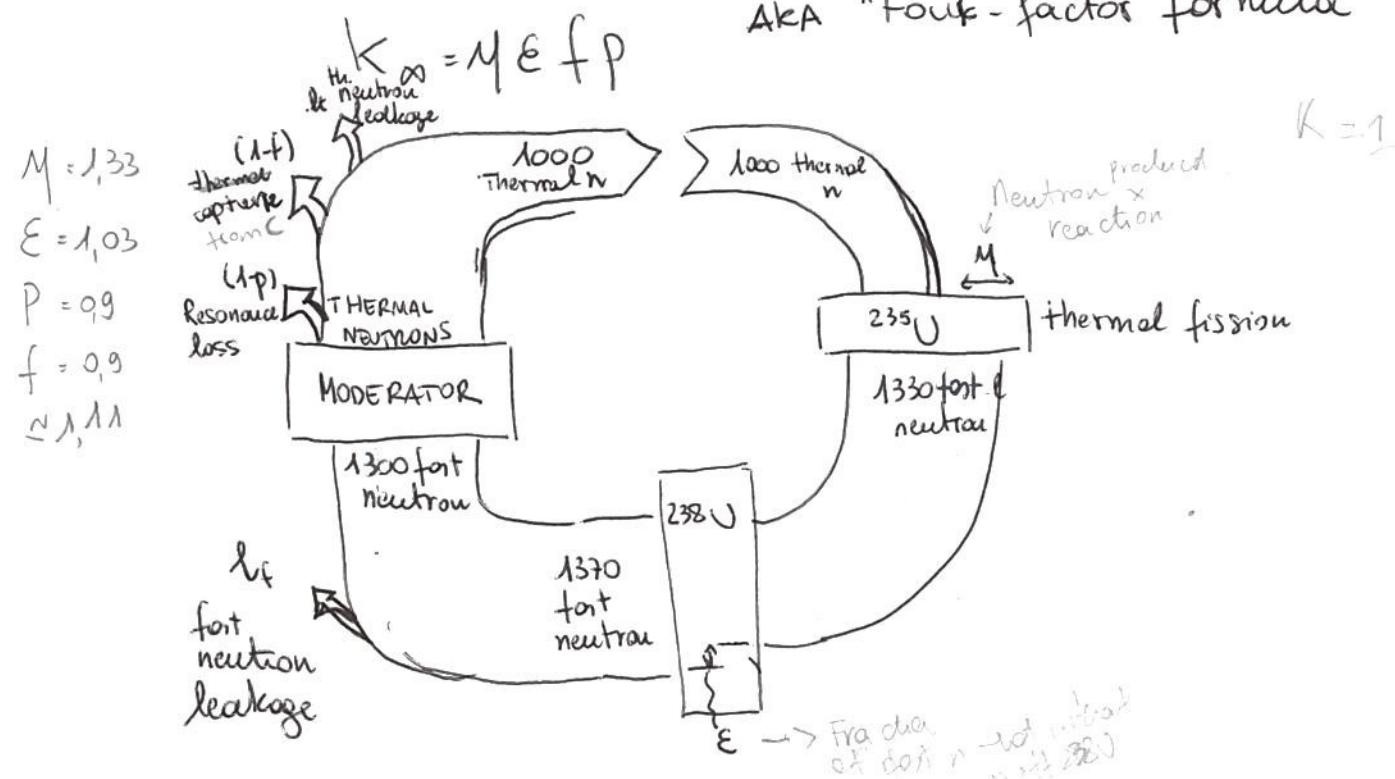
a lot of it. \Rightarrow An additional term must be included.

f = thermal utilization factor = fraction of thermal neutron available to ^{235}U and ^{238}U . $f \approx 0,9$

Find number of n $N_{\text{neut}} = M \epsilon p f N$ whether this is $>$ or $<$ than N determines the criticality of reactor.

The reproduction factor is

AKA "Four-factor formula"



$k_{\infty} \approx 1,11 \rightarrow$ which is still not appropriate because we ignored leakage of n at the surface $\Rightarrow l_f$ and l_t are fraction of leakage of fast and thermal neutrons

$$k_{\infty} = M \epsilon f (1 - l_f)(1 - l_t)$$

The larger the pile, the smaller is Surface/Volume ratio \Rightarrow the smaller is the fraction of neutrons that leak

If l_f, l_t are small $\rightarrow k_{\infty} - k \approx k(l_f + l_t)$

the total leakage decreases as the surface increases.

Leakage will increase if neutrons have to travel for a long distance, called MITIGATION LENGTH M .

M includes 2 contributes diffusion length L_d for thermal neutrons and the slowing distance L_s

$$M = (L_d^2 + L_s^2)^{1/2}$$

For graphite $L_s = 18.7 \text{ cm}$ $L_d = 50.8 \text{ cm}$

If the pile has a dimension R (radius or side length)

$$(k_\infty - k) \propto R^{-2} \quad \text{and} \quad k_\infty - k \propto M$$

Are reasonable assumptions

$$\rightarrow k_\infty - k \propto \frac{M^2}{R^2}$$

So there will be a critical size R_c which corresponds to $k=1$

$$R_c \propto \frac{M}{\sqrt{k_\infty - 1}}$$

From spherical arrangement

$$R_c = \frac{\pi M}{\sqrt{k_\infty - 1}}$$

For neutral uranium-graphite reactor $R_c = 5 \text{ m}$. For $R = 5 \text{ m}$ the "reactor" goes critical. How to decrease the size? Surround the pile with a reflector material for neutron.

Let's now evaluate now the time needed for the processes.

Neutrons are characterized by a time constant γ which includes the time to moderate ($\approx 10^{-6} \text{ s}$) and a diffusion time ($\approx 10^{-3} \text{ s}$).

If the reproducing factor is k and there are N neutrons at time t , on average there are kN neutrons at time $t + \gamma$ and $k^2 N$ at time $t + 2\gamma$

In a short interval dt

$$dN = (kN - N) \frac{dt}{\gamma}$$

$$\Rightarrow N(k) = N_0 e^{(k-1)t/\gamma}$$

If $k = 1 \rightarrow N = \text{constant}$. If $k < 1$ the number of neutrons decreases exponentially. If $k > 1$ N increases exponentially, with a time constant $\tau/k-1$.

Control: the number of neutrons is essential and can be achieved by inserting some bars of materials able to absorb neutrons (like Cadmium).

BASICS OF NUCLEAR REACTORS

We will cover only the general properties and categories of reactors.