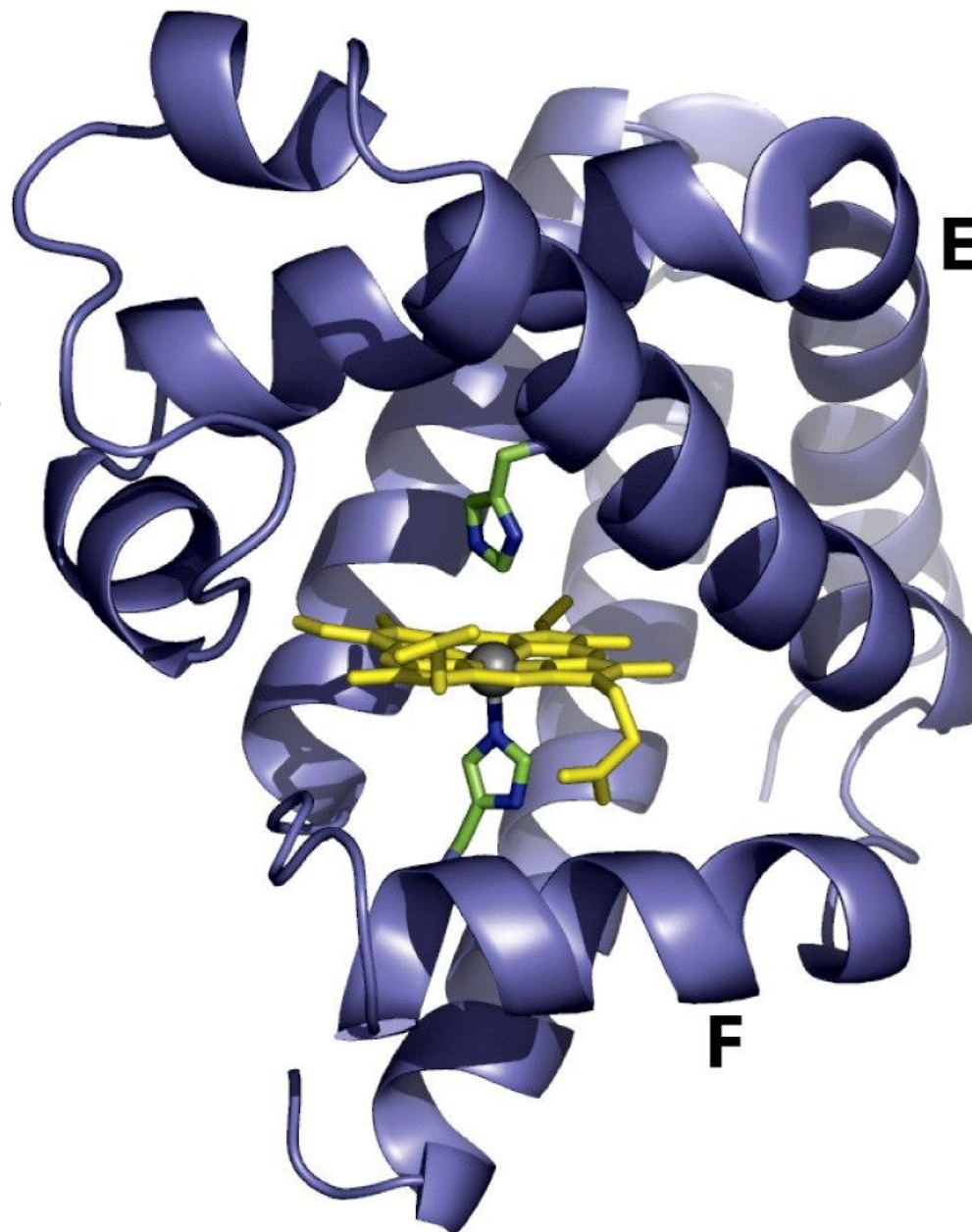


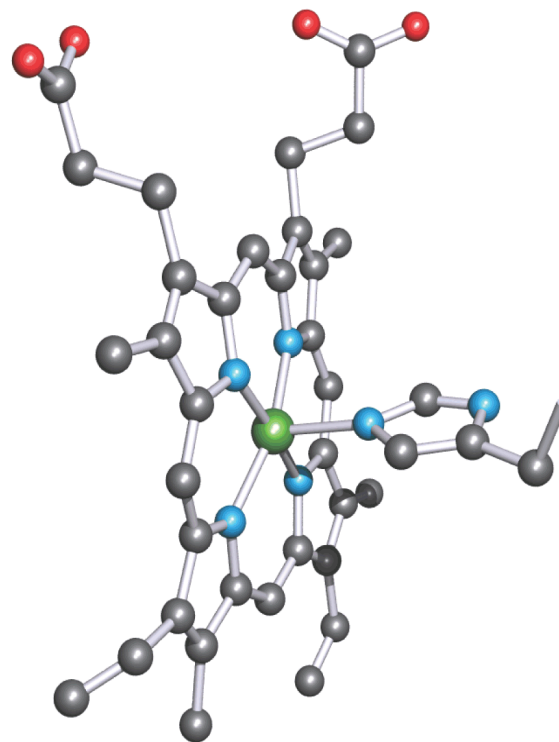
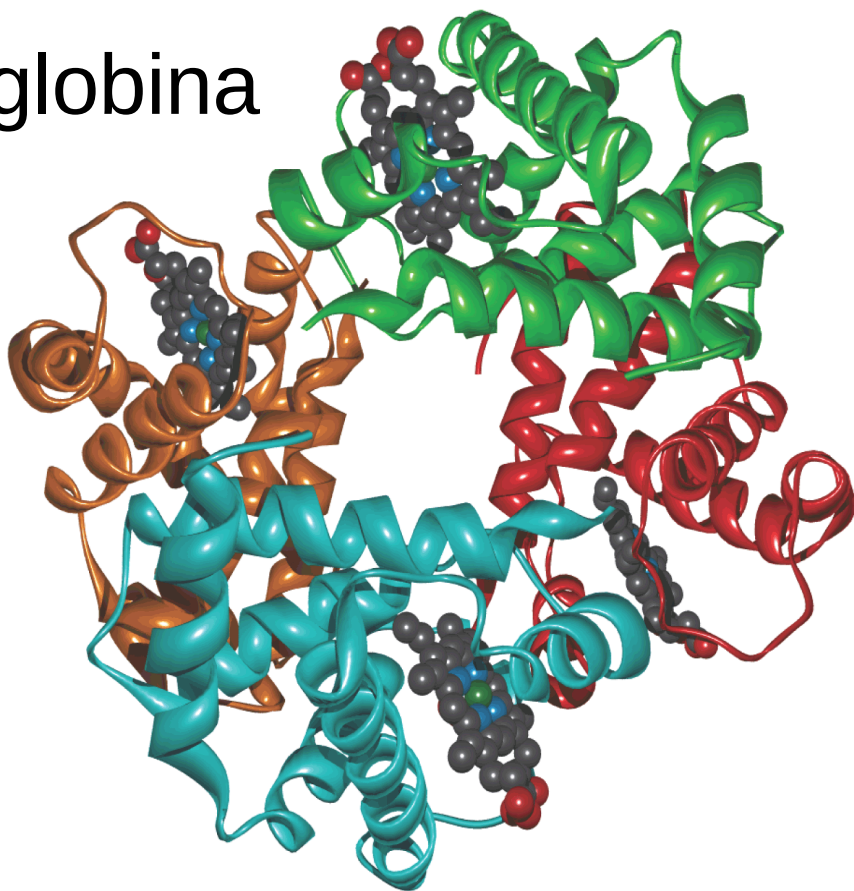
Mioglobina

17.8 kDa



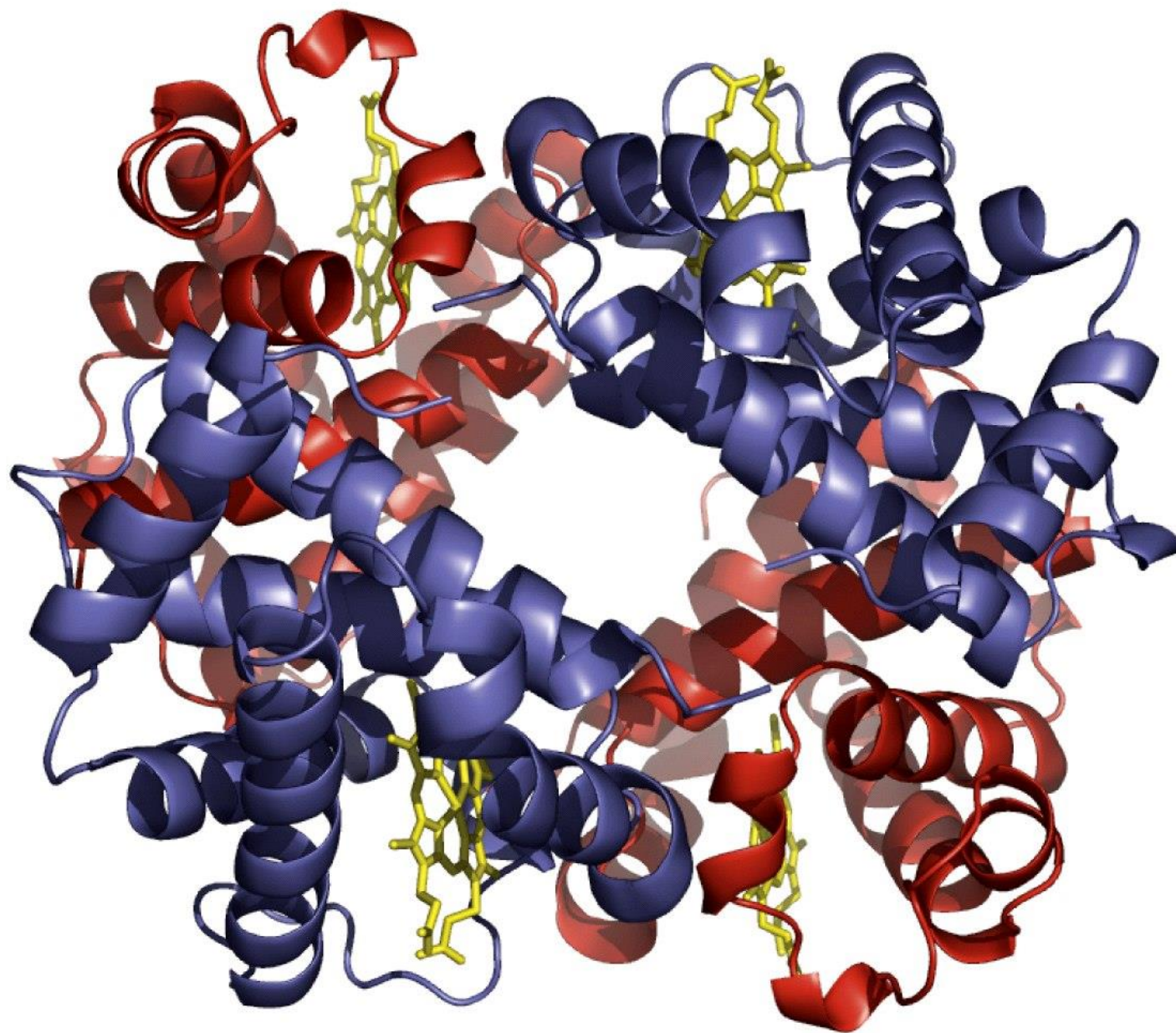
ca. 6% Fe_{tot}

Emoglobina



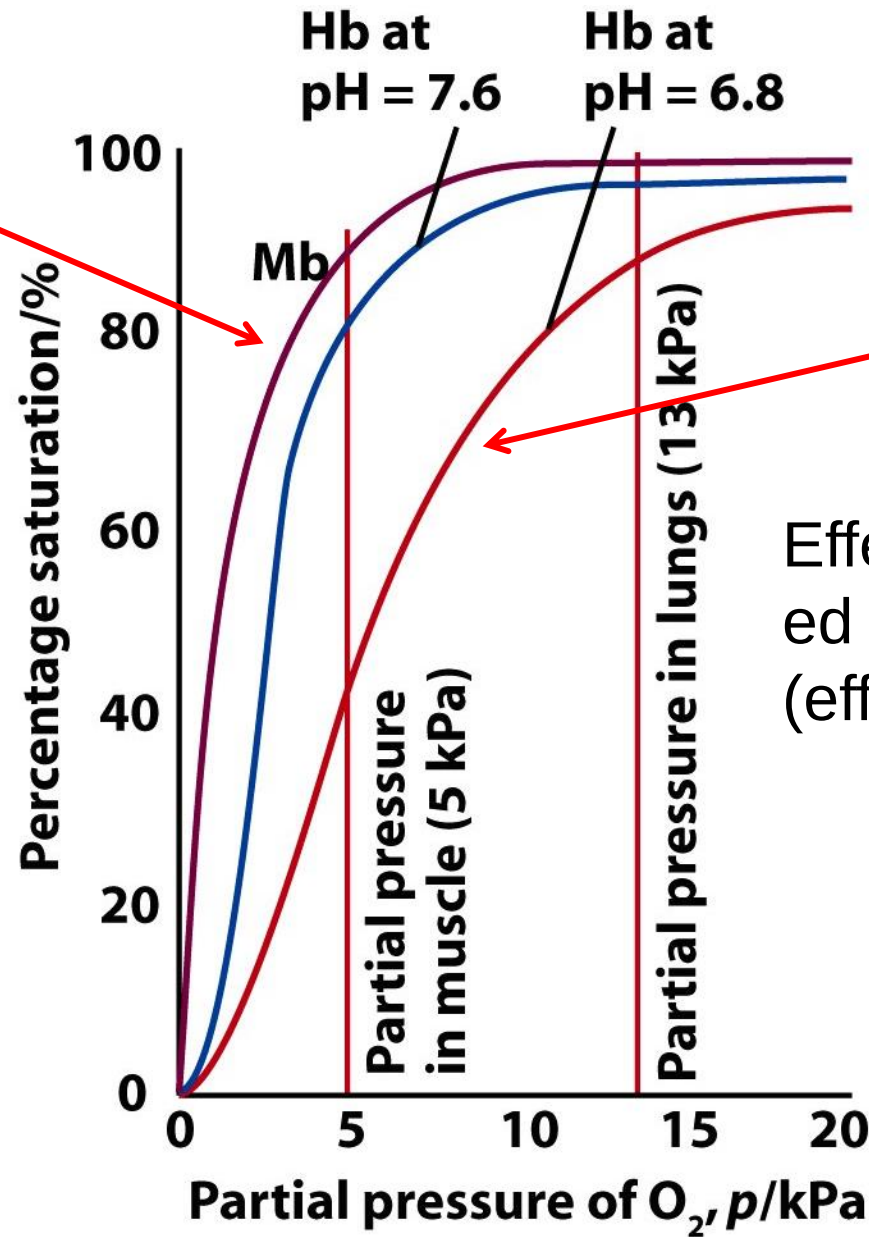
tetramero $\alpha_2\beta_2$ 141 e 146 a.a. 64.5 kDa

ca. 65% Fe_{tot}



tetramero $\alpha_2\beta_2$

Curva di saturazione
iperbolica



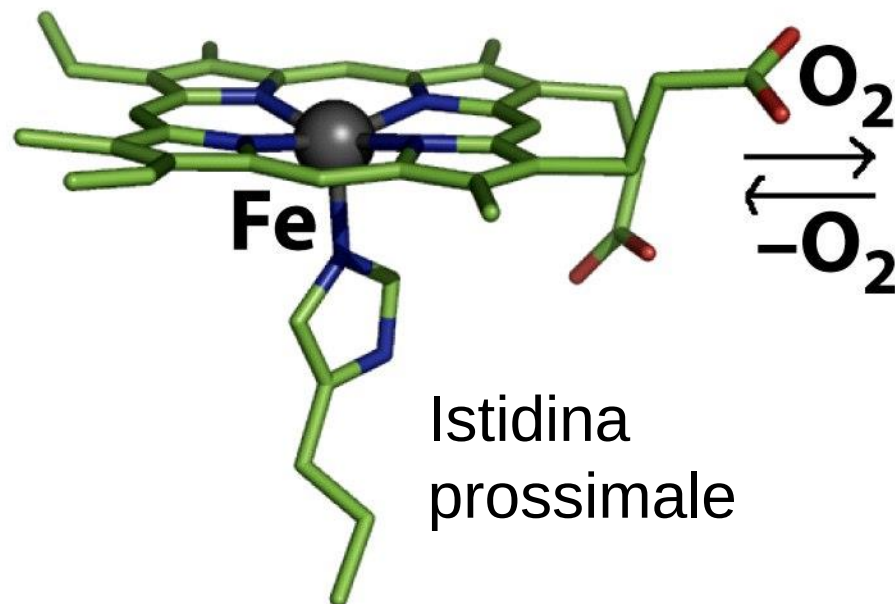
Curva di saturazione
sigmoidale, pH
dipendente

Effetto cooperativo
ed effetto Bohr
(effetto allosterico)

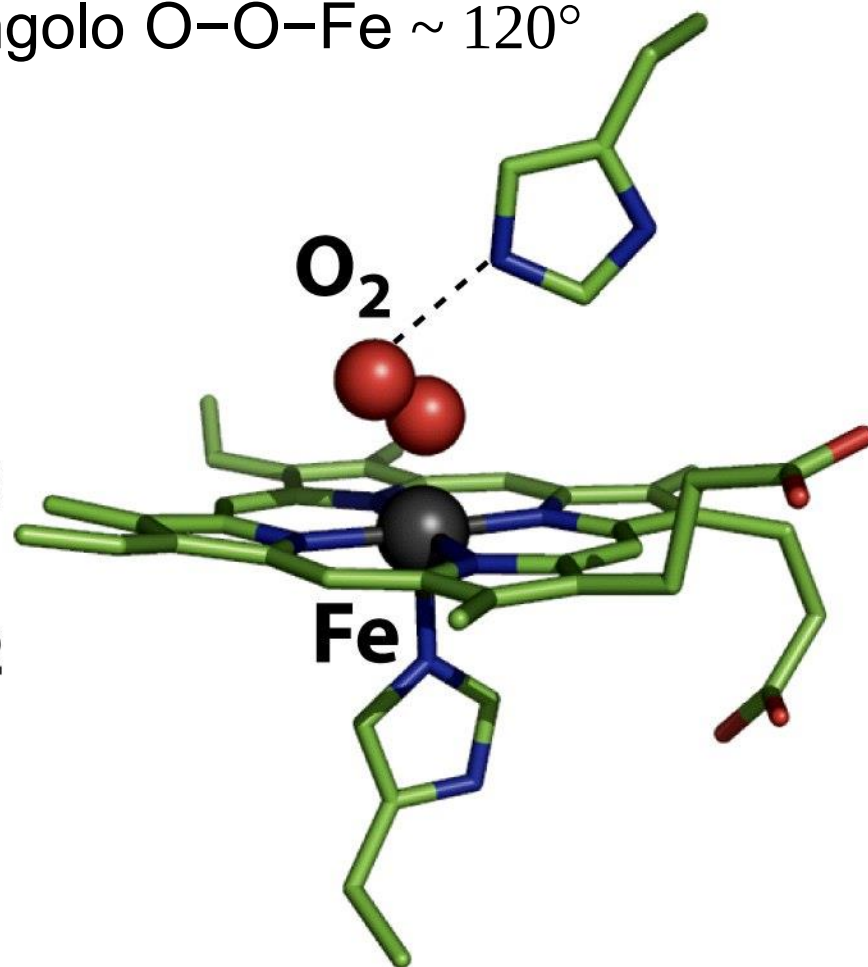
La coordinazione di CO_2 e protoni (pH) alla catena proteica dell'emoglobina causano variazioni conformazionali che influenzano l'affinità dell'eme per O_2

Coordinazione O_2 *end-on*
Angolo O-O-Fe $\sim 120^\circ$

Istidina
distale



Fe(II), d⁶, h.s.

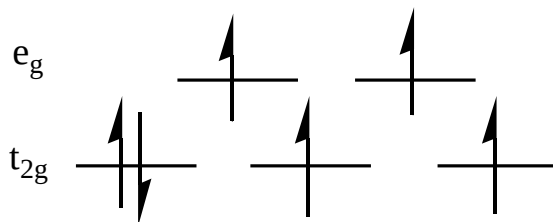


S = 0

desossi:

$S = 2$

paramagnetica



Fe(II), d^6 paramagnetico

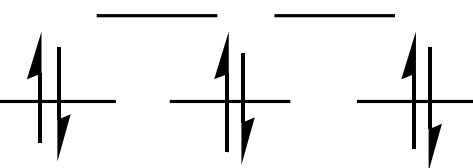
Pauling

Weiss

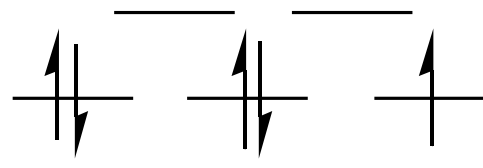
ossi:

$S = 0$

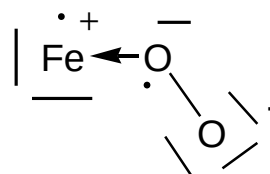
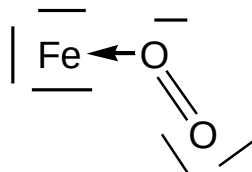
diamagnetica



Fe(II) d^6 basso spin
+ 1O_2 legato



Fe(III) d^5 basso spin
+ $^2O_2^{\cdot-}$ legato



Pauling

$\text{Fe}^{\text{II}} / \text{O}_2$

$S = 0$



McClure/Goddard

$\text{Fe}^{\text{II}} / \text{O}_2$

$S = 1$



Weiss

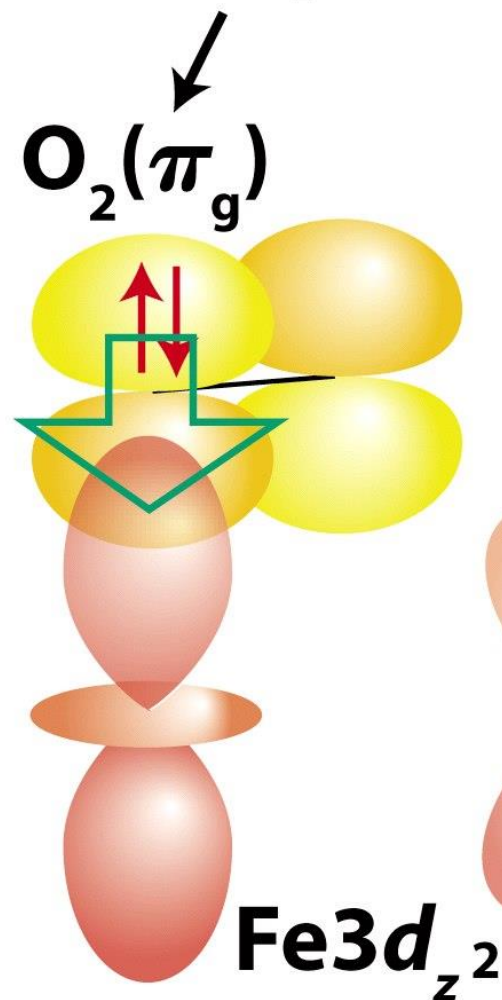
$\text{Fe}^{\text{III}} / \text{O}_2^-$

$S = 1/2$

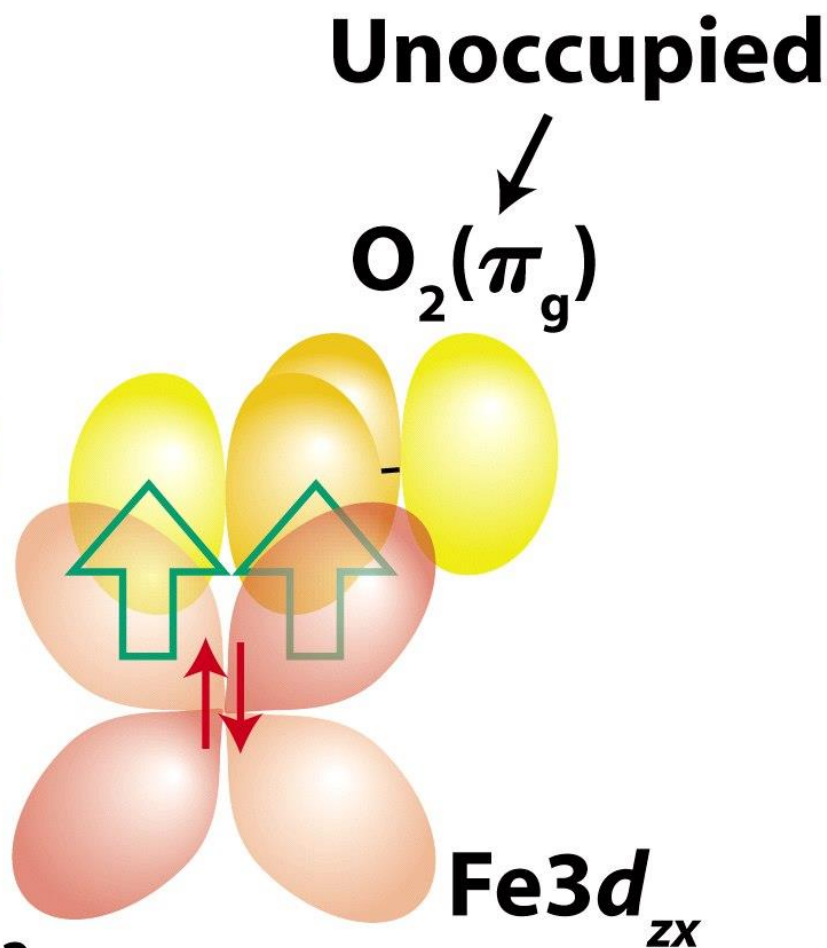


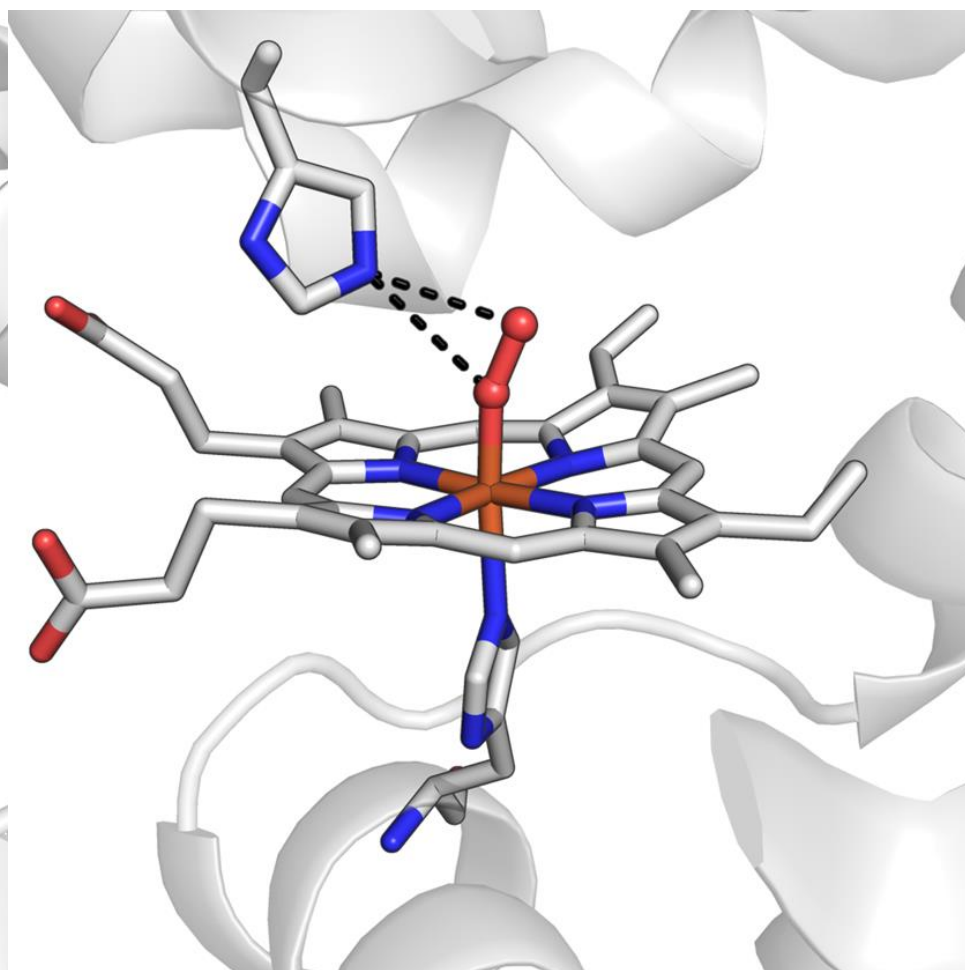
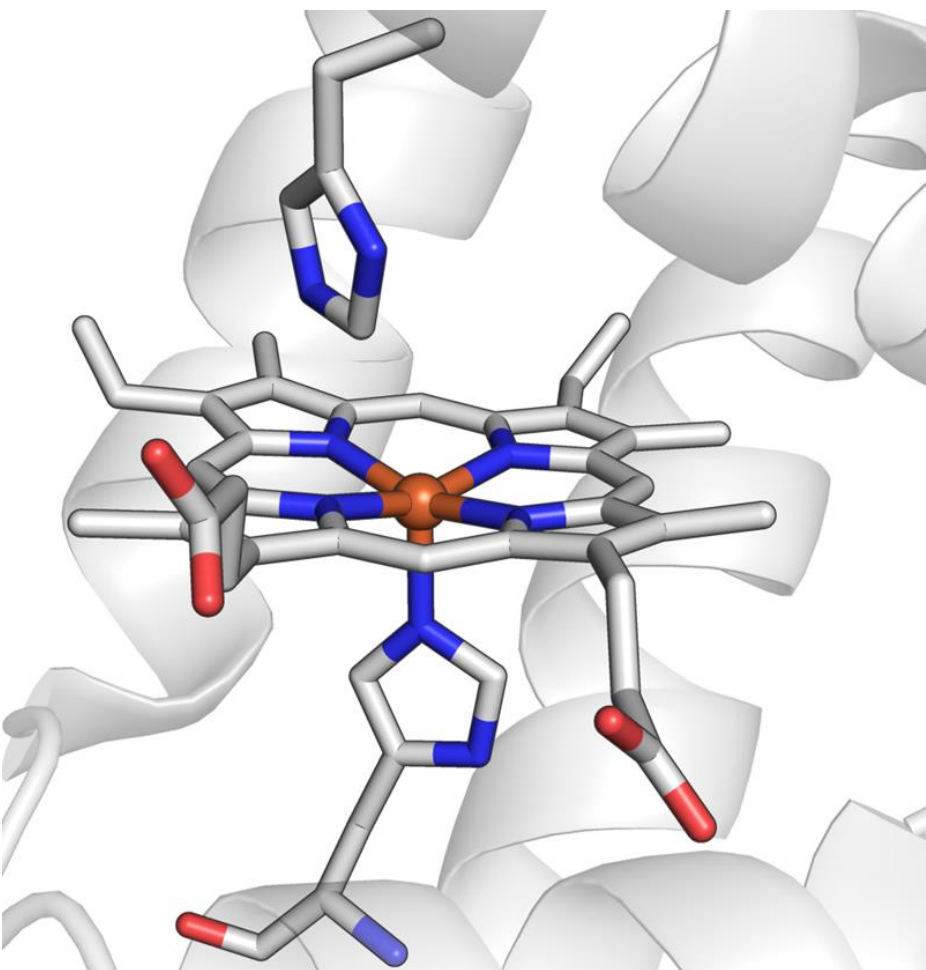
Modello di Pauling: Fe(II) I.s. + $^1\text{O}_2$

(a) Occupied

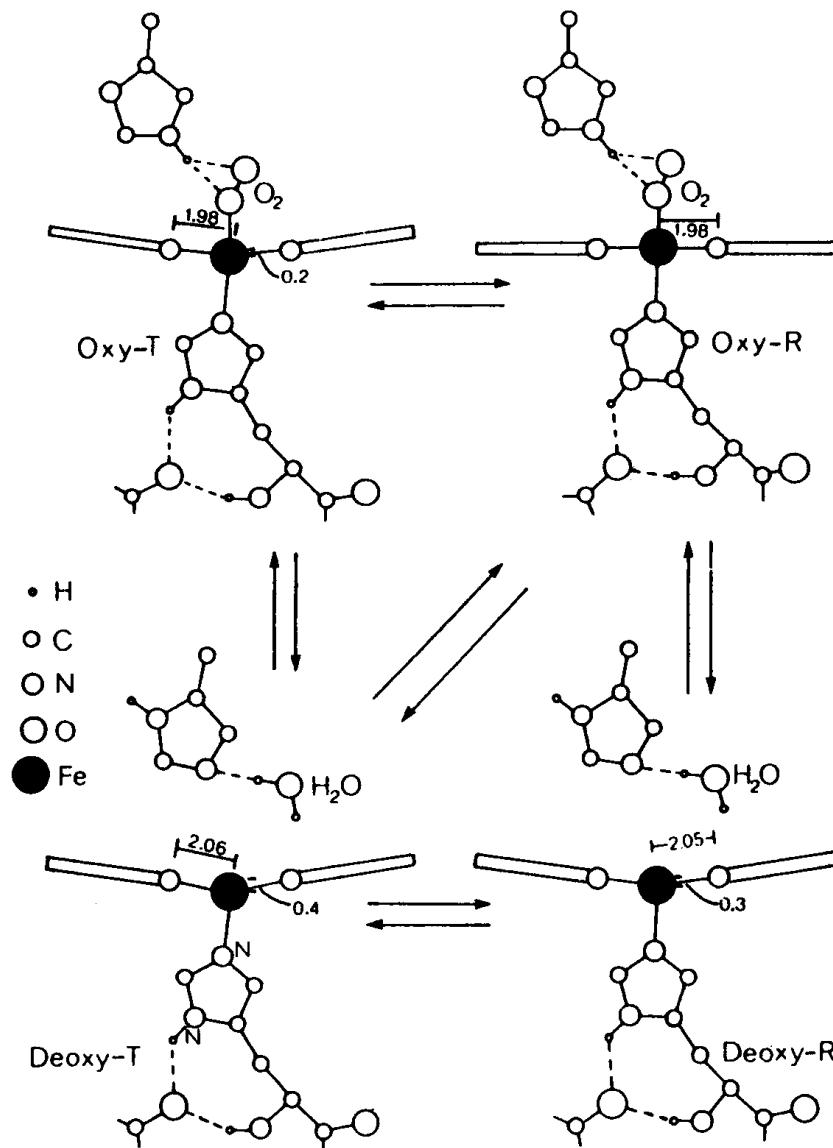


(b)



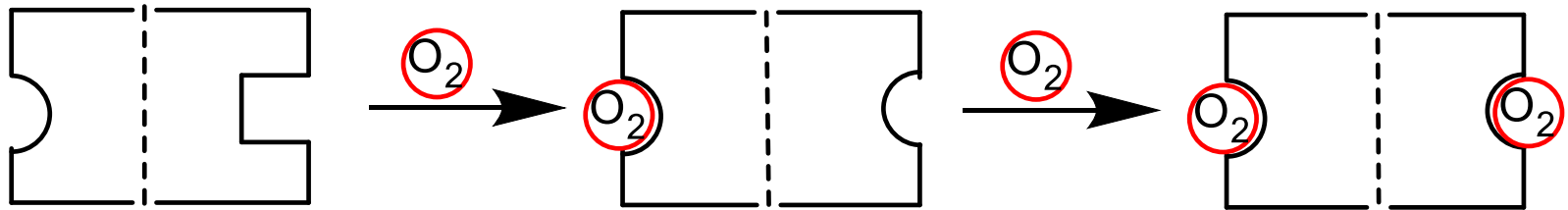


Fe(II) h.s.: 0.92 Å
Fe(II) l.s.: 0.75 Å
Fe(III) l.s.: 0.55 Å

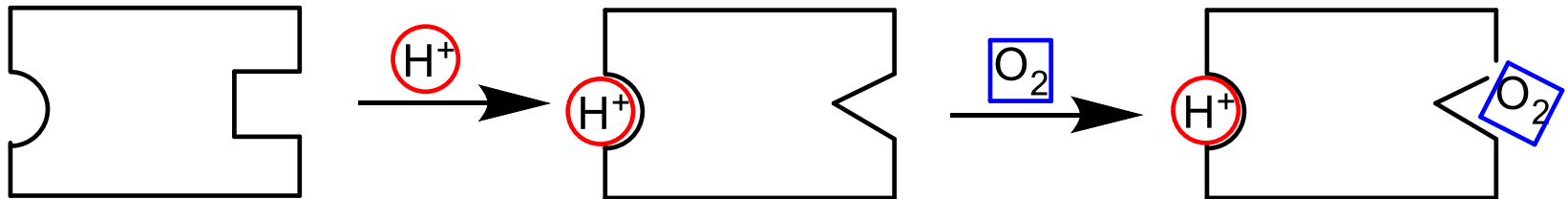


l'affinità della forma R per O_2 è circa la stessa delle catene α e β isolate, cioè della Mb, mentre quella della forma T è circa 12 – 14 kJ/mole inferiore

allosterismo = il cambio di attività e conformazione di un enzima in seguito al legame di un composto (*effettore*) ad un sito dell'enzima diverso da quello attivo

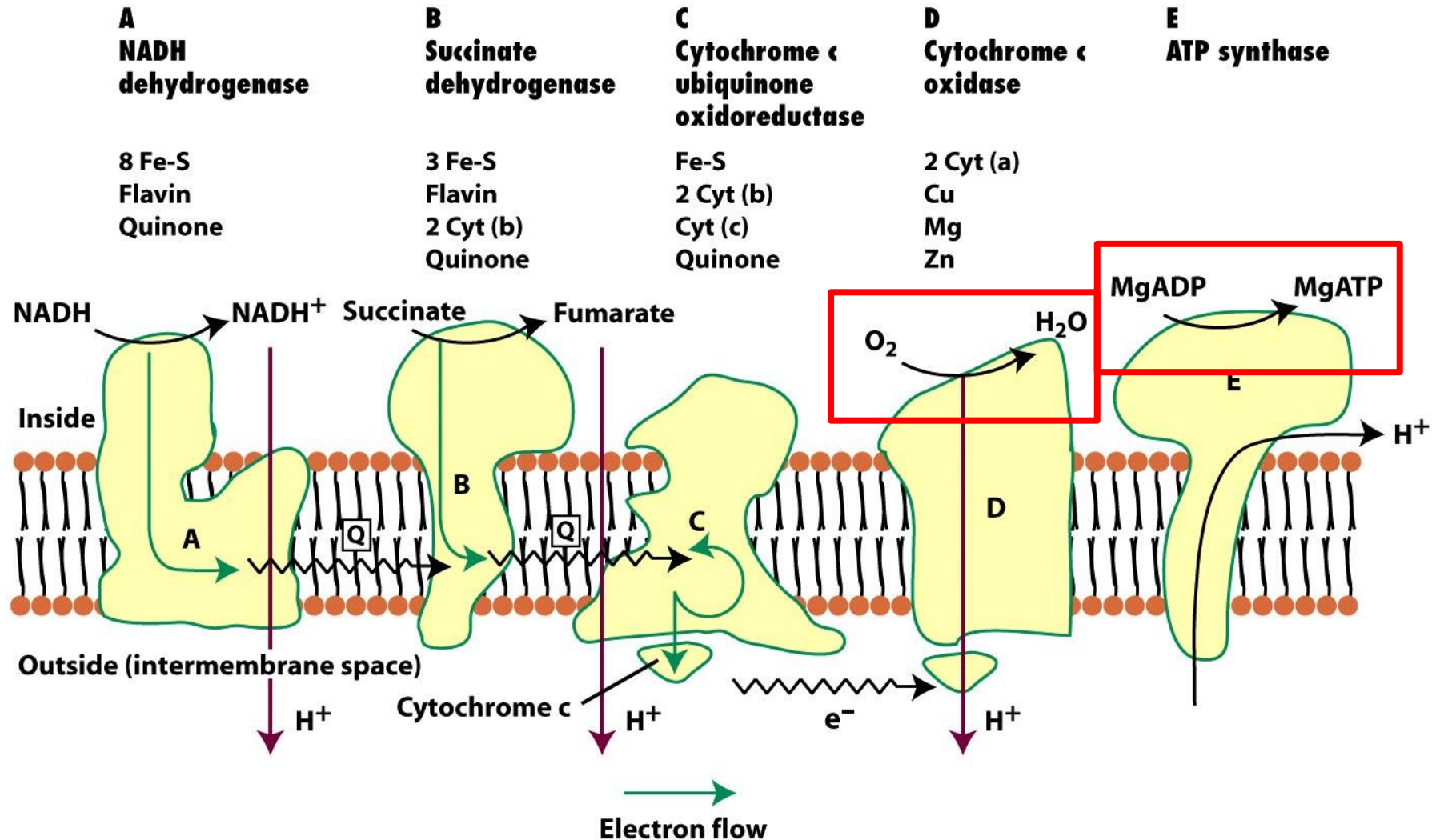


Allosterismo Omotropico



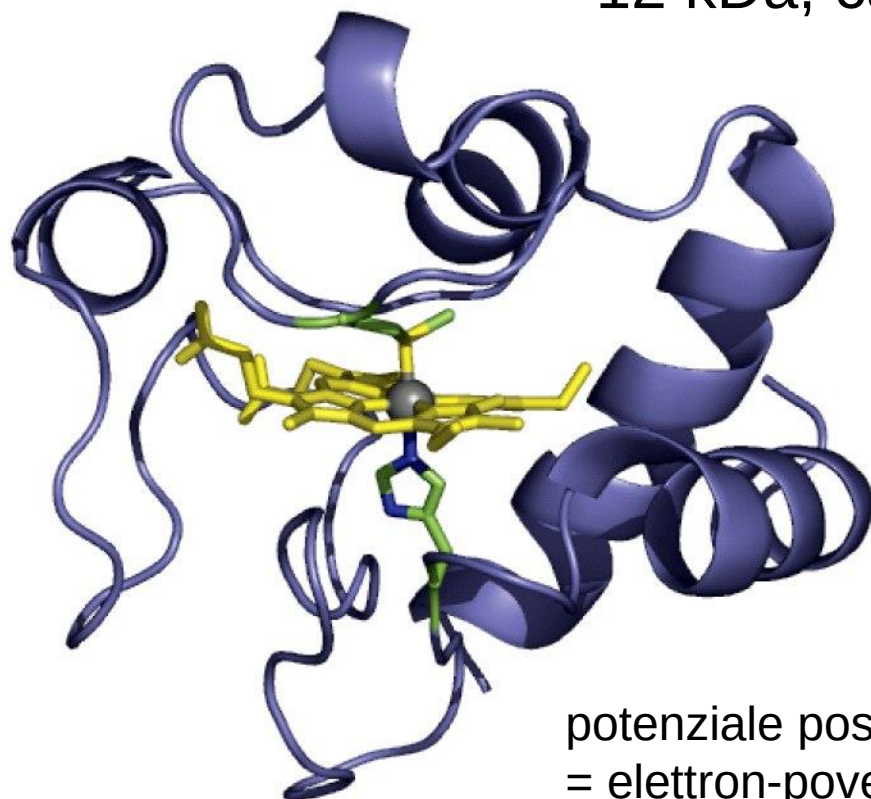
Allosterismo Eterotropico

Catena della respirazione cellulare

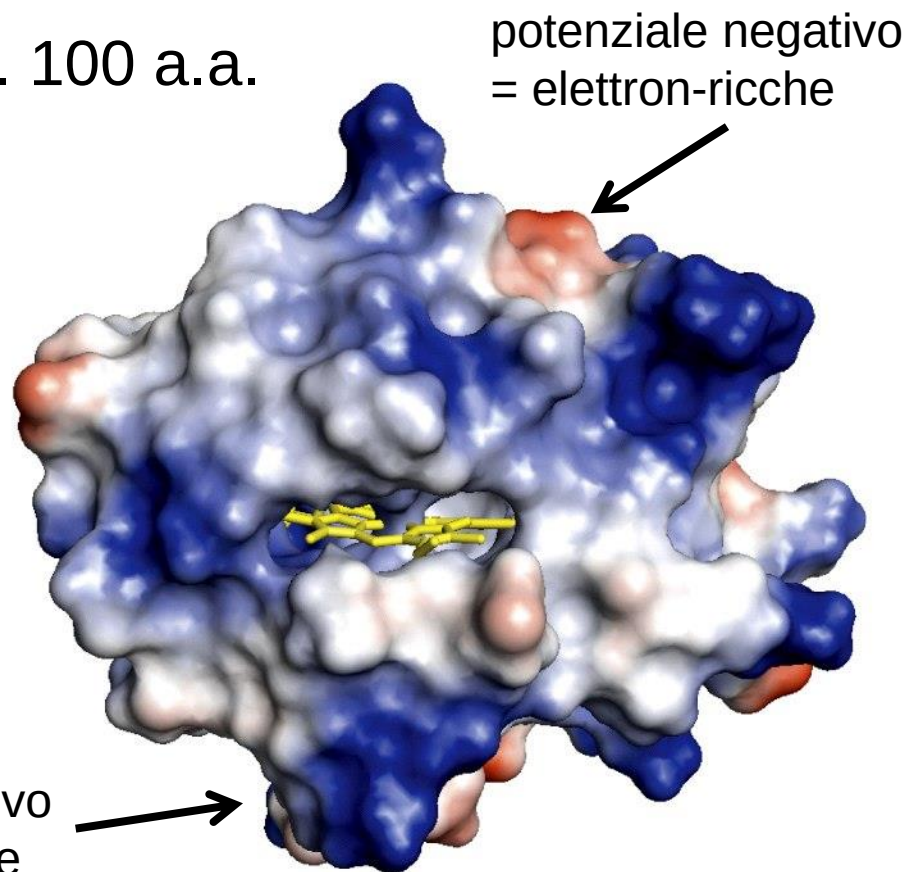


Citocromo c (da cuore di cavallo)

12 kDa, ca. 100 a.a.



potenziale positivo
= elettron-povere

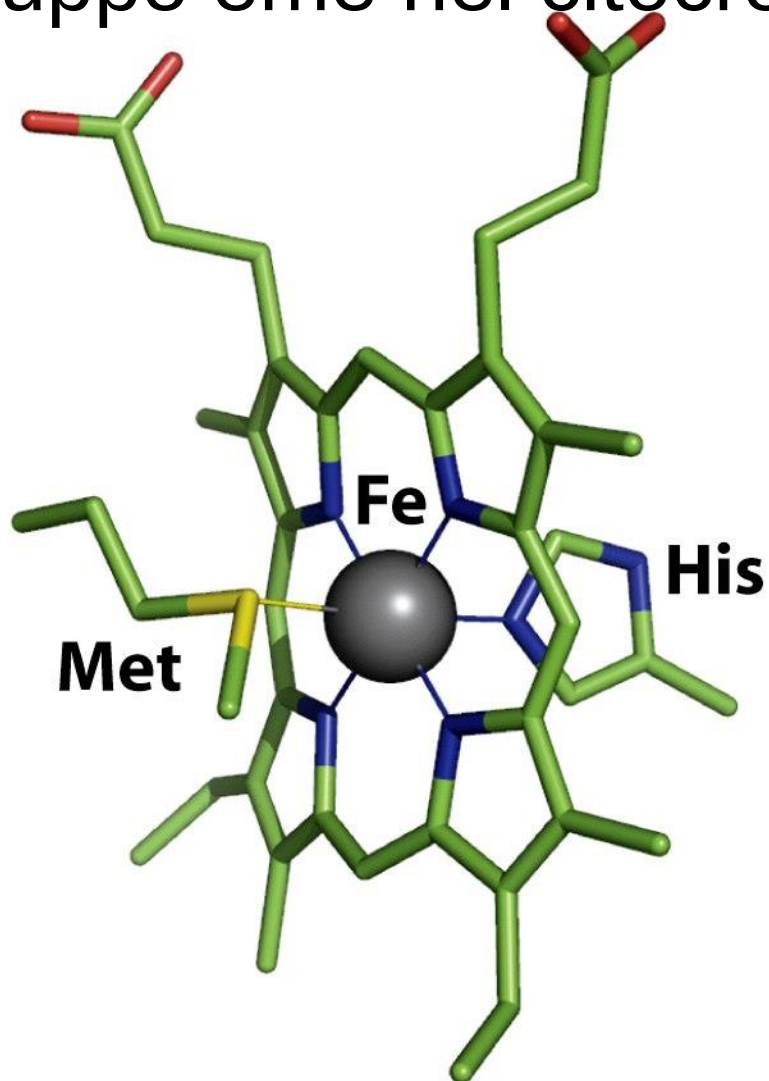


potenziale negativo
= elettron-ricche

mappa del potenziale elettrostatico

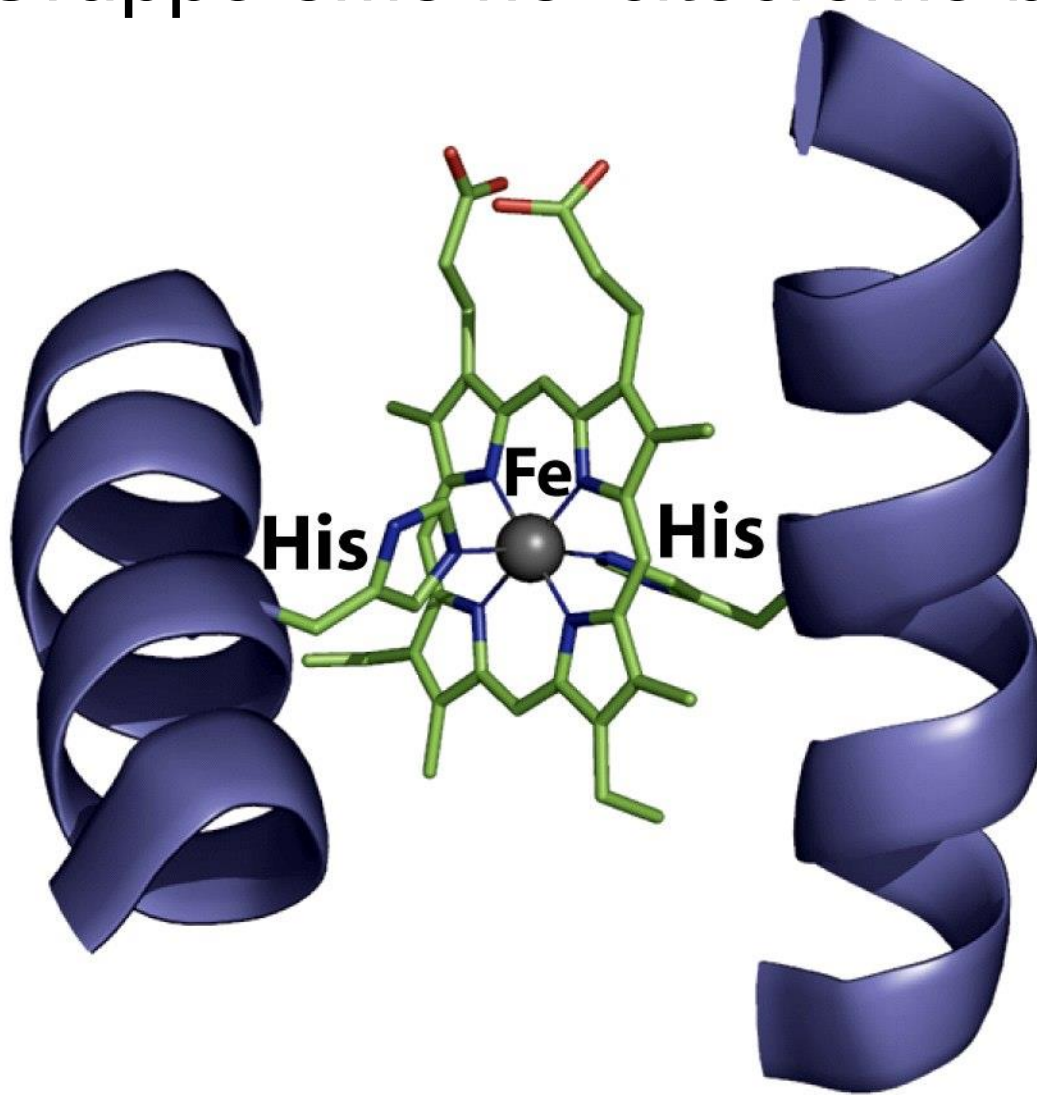
$$E_{\text{Fe(III)/Fe(II)}} = +260 \text{ mV}$$

Gruppo eme nel citocromo c



$$E_{\text{Fe(III)/Fe(II)}} = +260 \text{ mV}$$

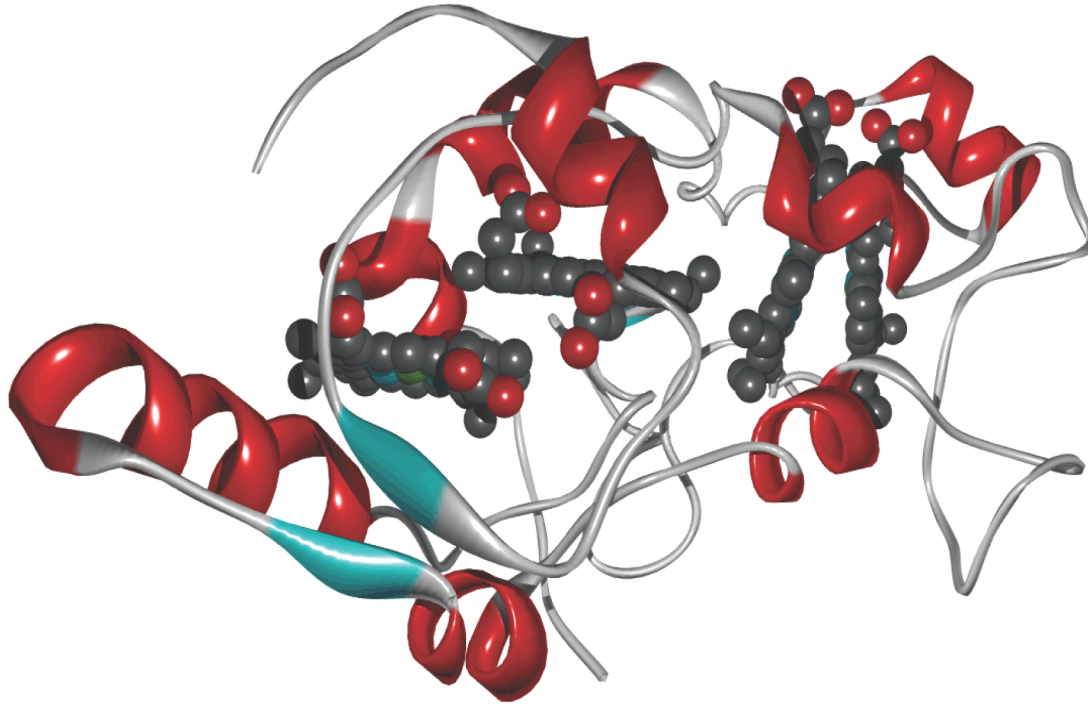
Gruppo eme nel citocromo b



$$E_{\text{Fe(III)/Fe(II)}} = +20 \text{ mV}$$

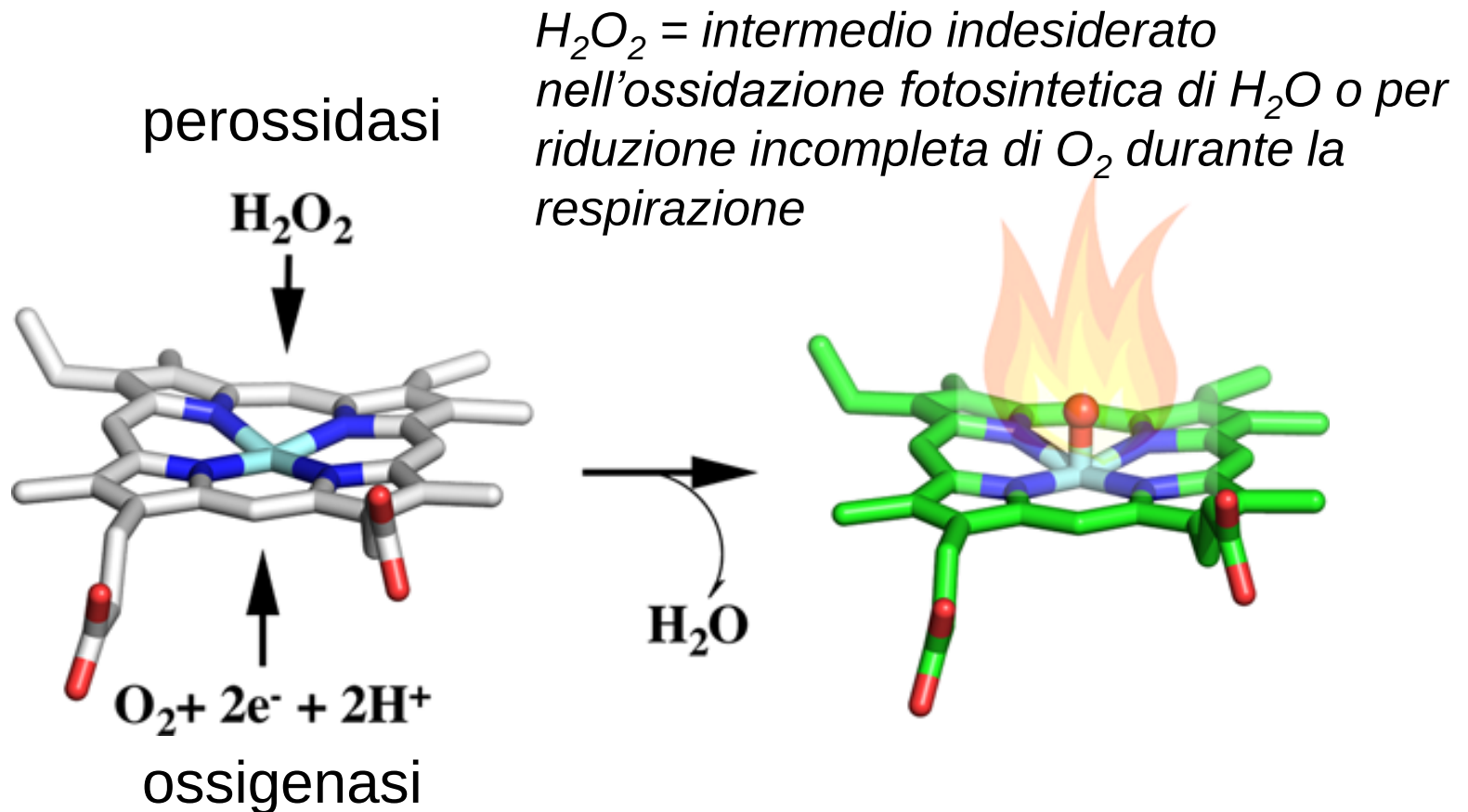
Citocromo c554

(nitrificazione catalizzata da batteri)



Distanze $\text{Fe} \cdots \text{Fe} \approx 950, 1220, 920 \text{ pm}$

Ossidazioni catalizzate da eme-proteine

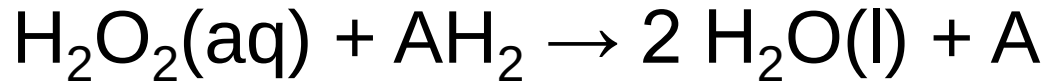


ossidare non sempre implica anche ossigenare i substrati

Perossidasi e Catalasi

e.g. *lignina perossidasi*, *horse-radish perossidasi (HRP)*,
Citocromo c perossidasi, *ascorbato perossidasi*...

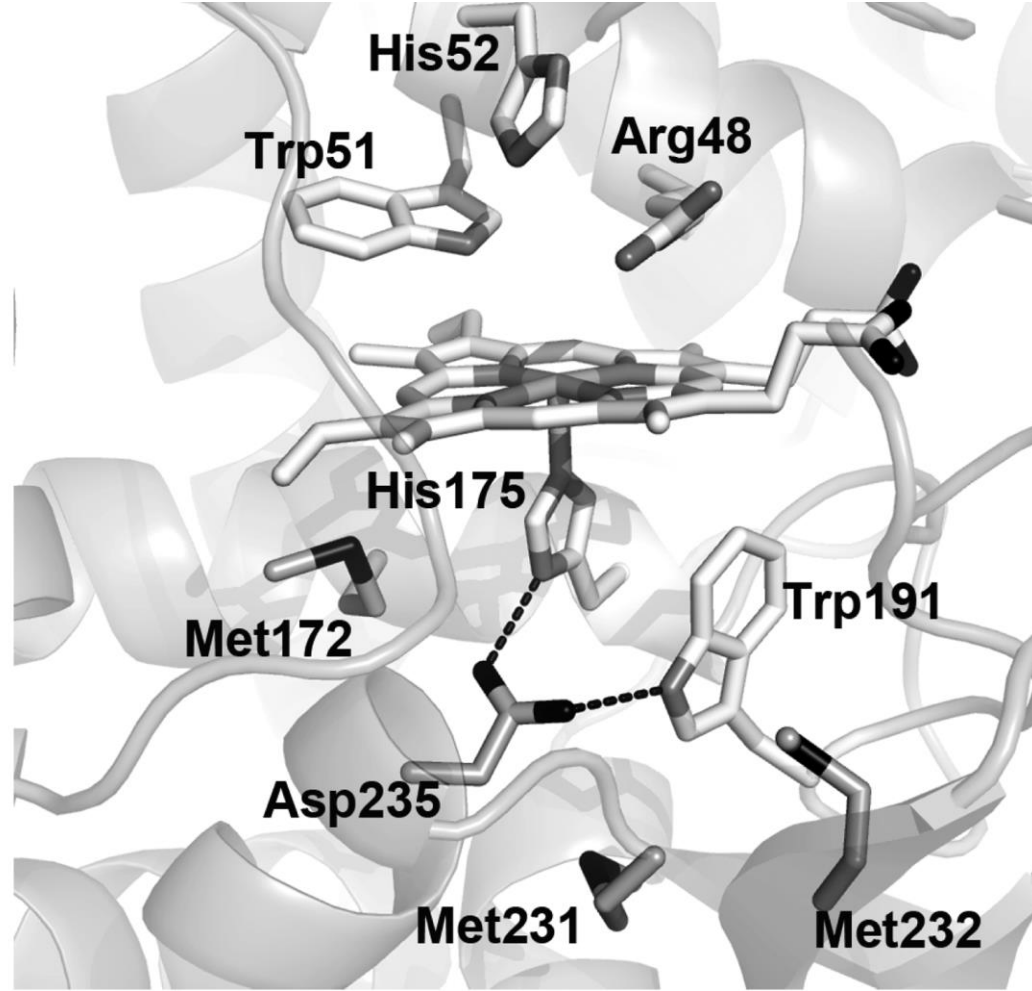
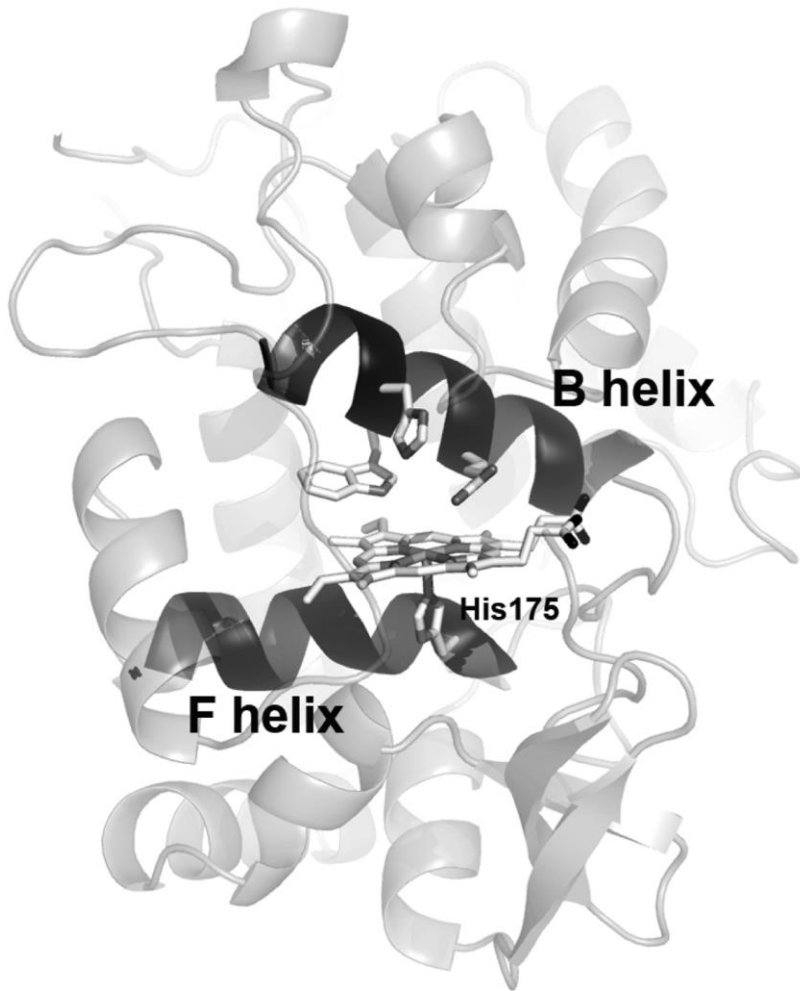
sono anche enzimi detossificanti



Substrati: acidi grassi, ammine, fenoli, tossine xenobiotiche,....

Citocromo c Perossidasi

(lievito, batteri)



Sito attivo della
Citocromo c Perossidasi
(lievito, batteri)

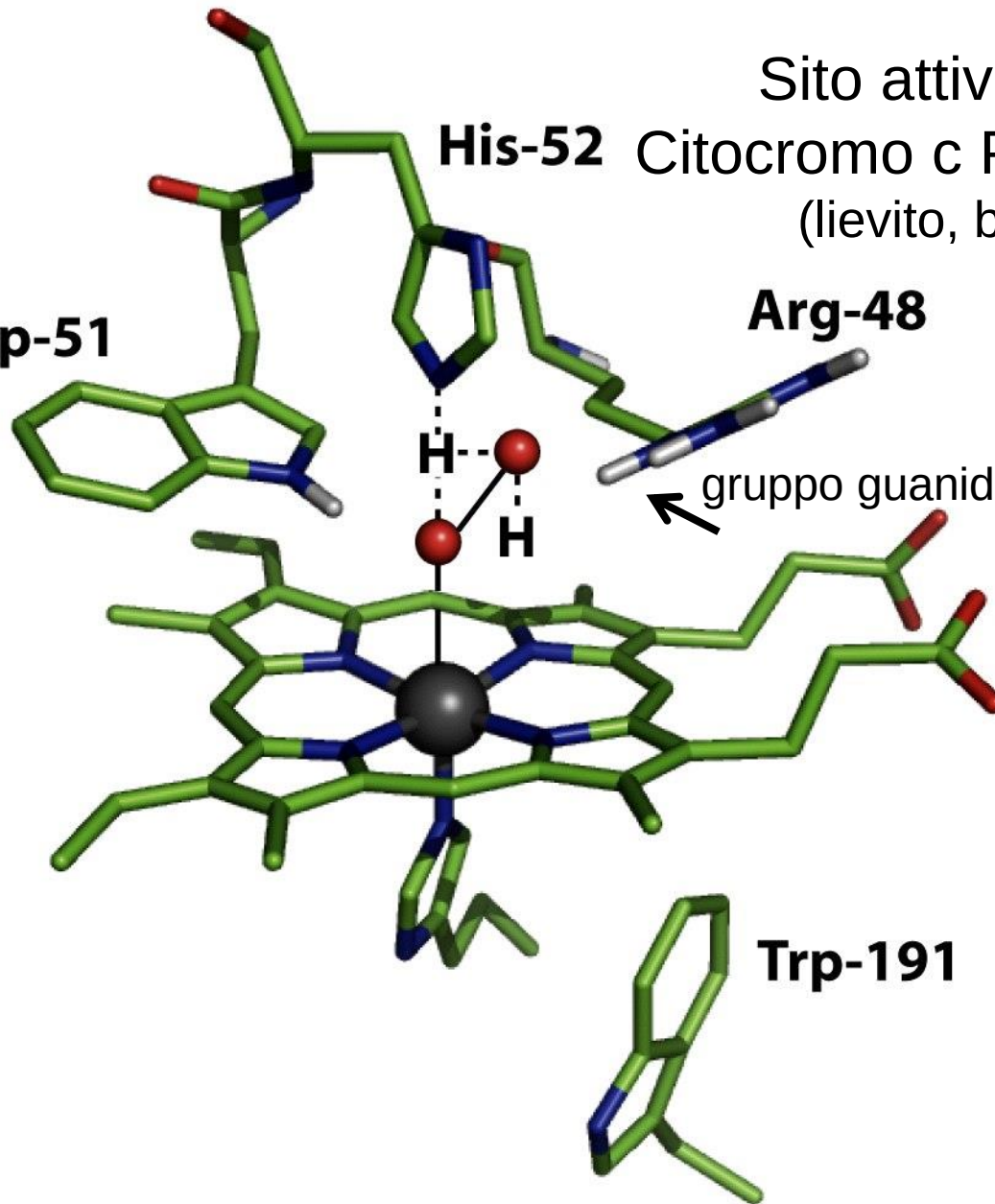
Trp-51

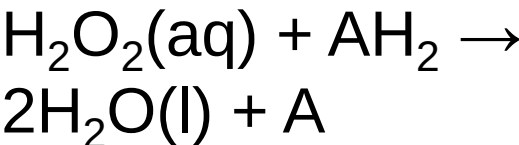
His-52

Arg-48

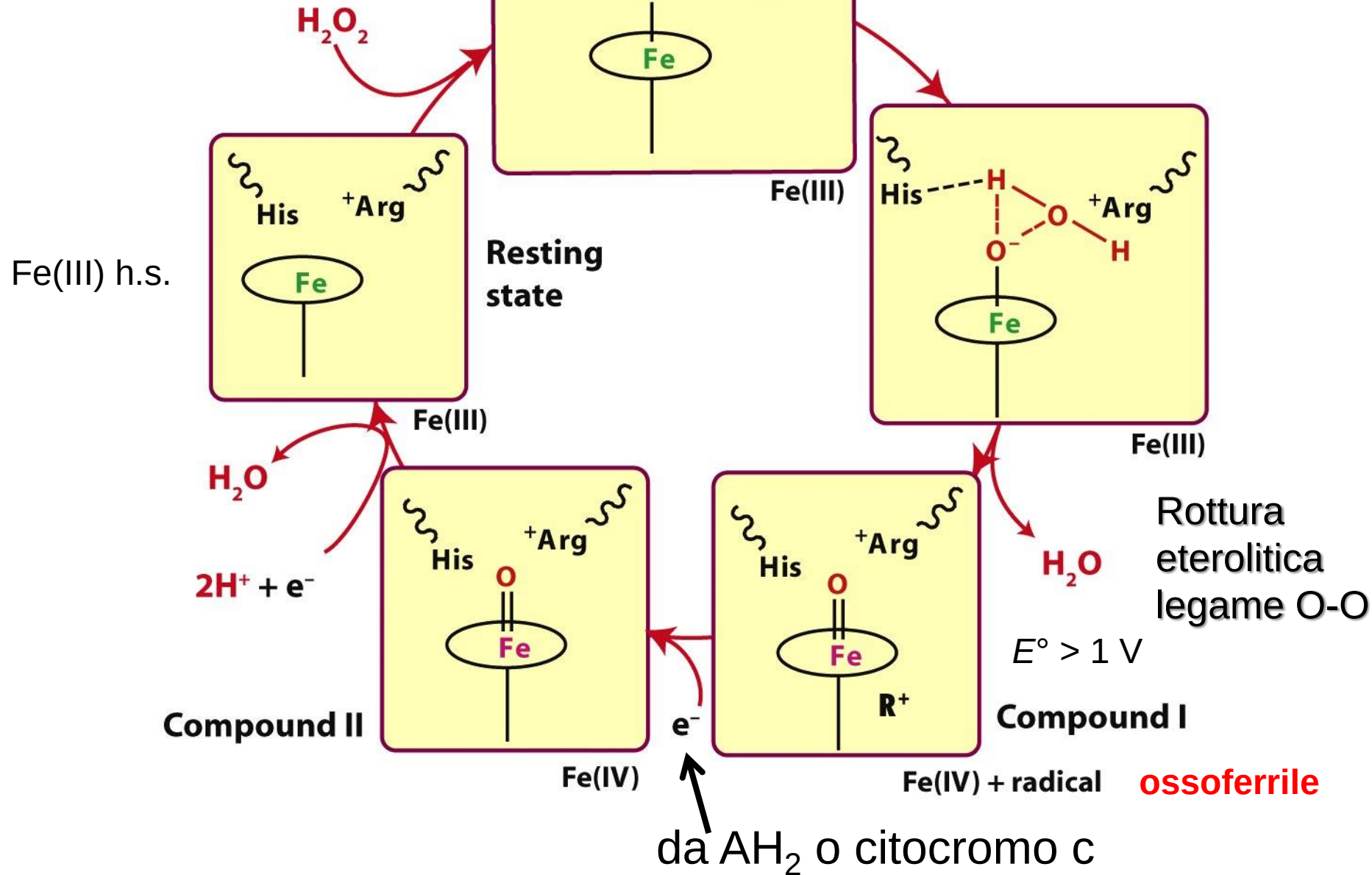
gruppo guanidinio

Trp-191

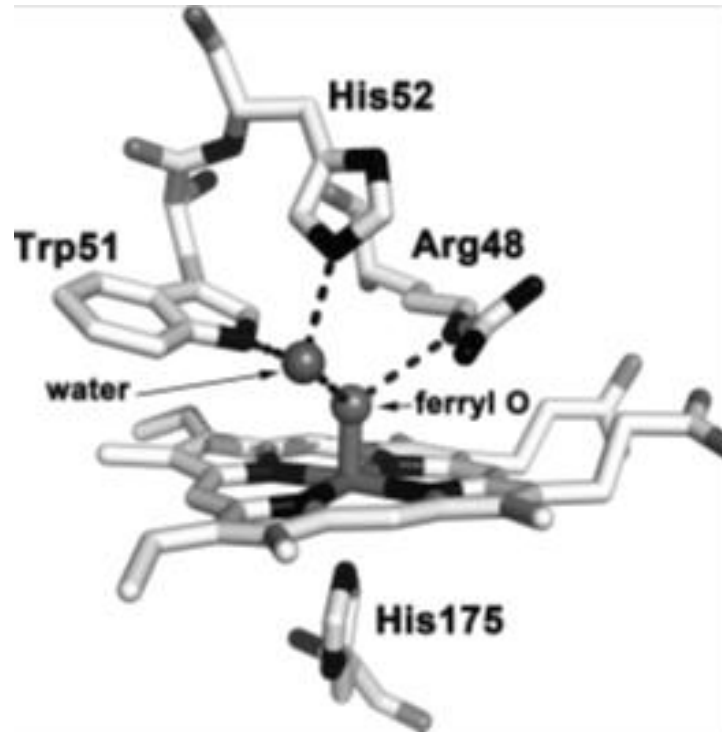




Ciclo catalitico di una perossidasi

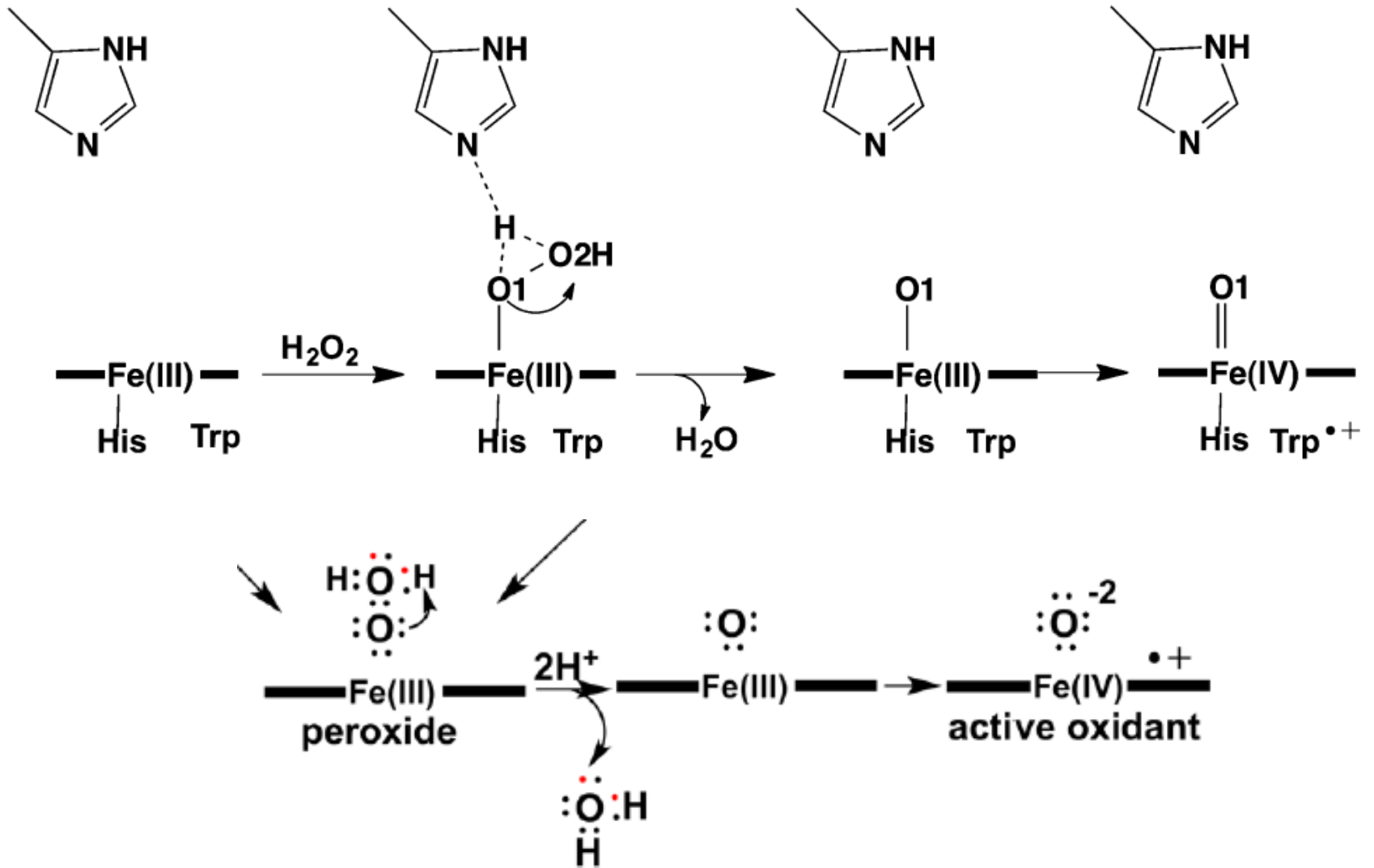


Struttura ai raggi X dell'osso-ferrile

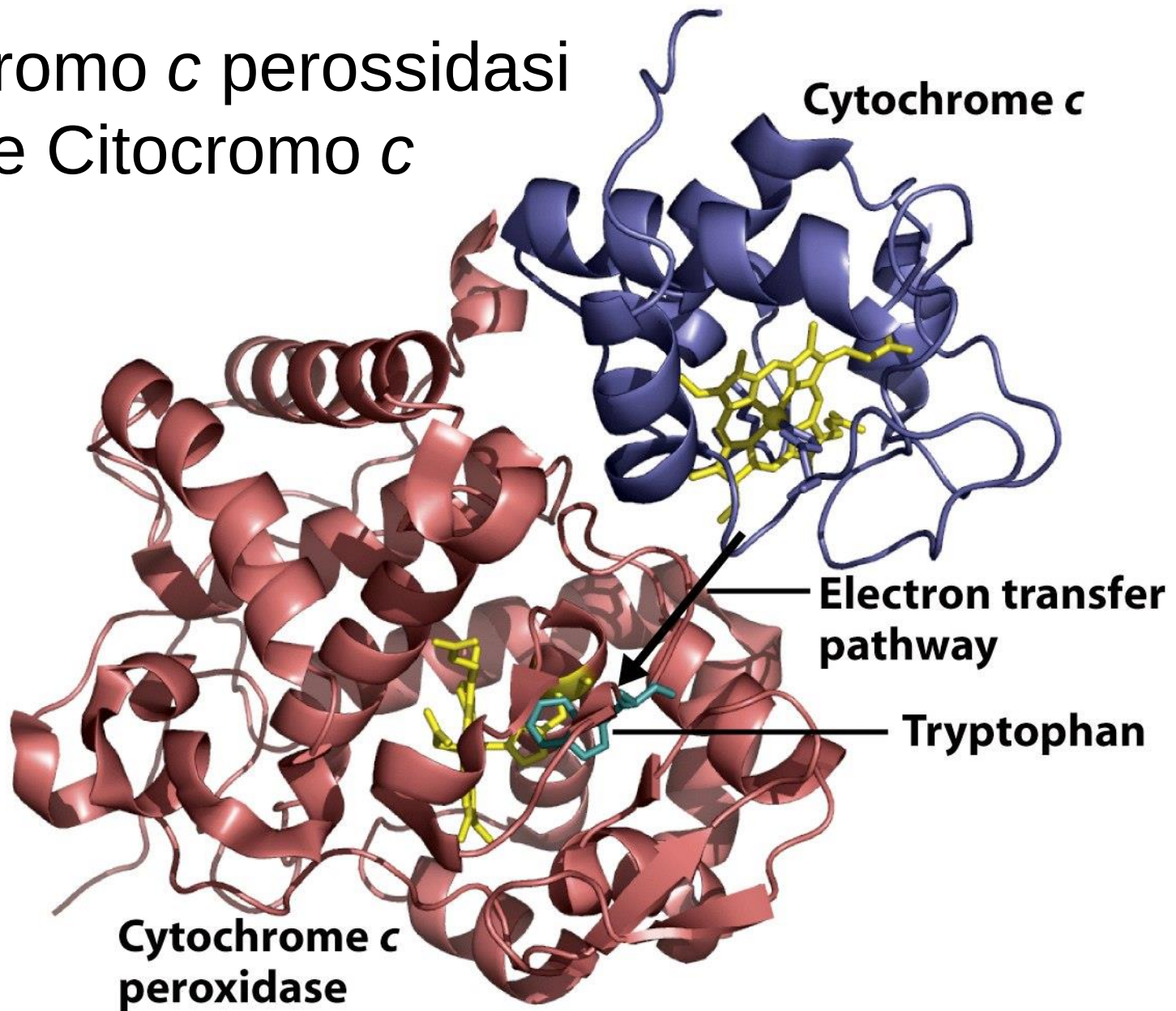


Formazione dell'osso-ferrile

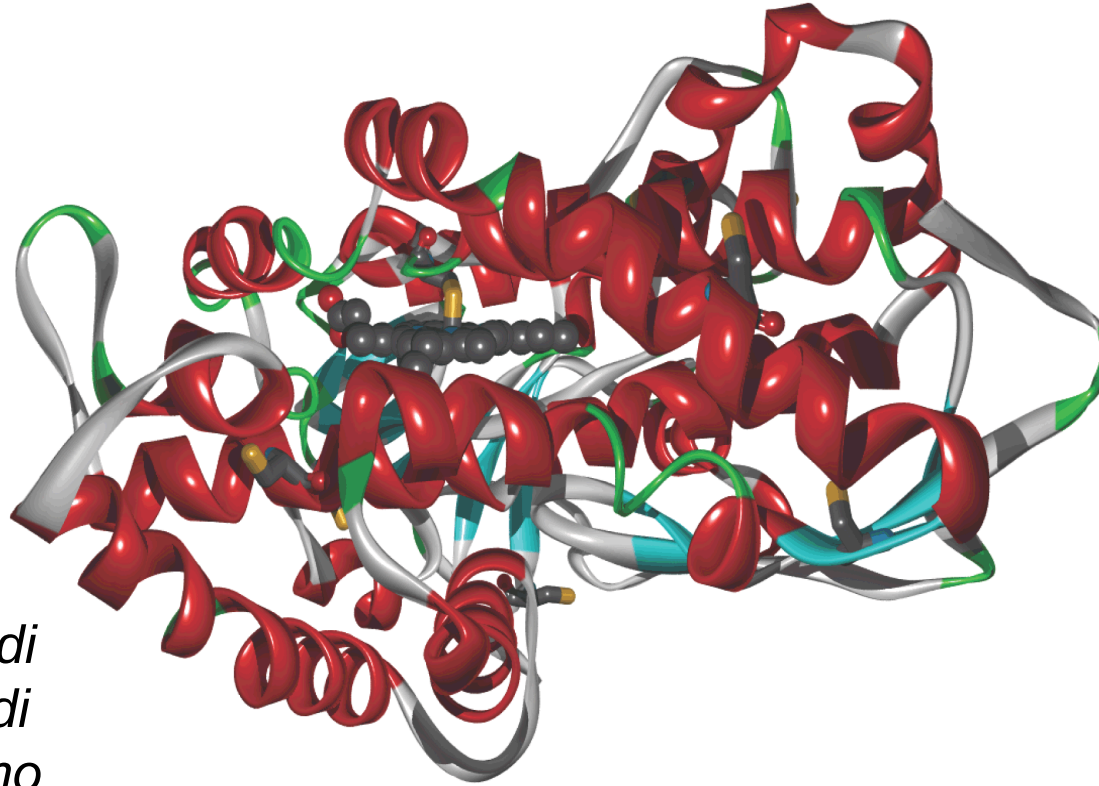
Original Mechanism



Citocromo c perossidasi e Citocromo c



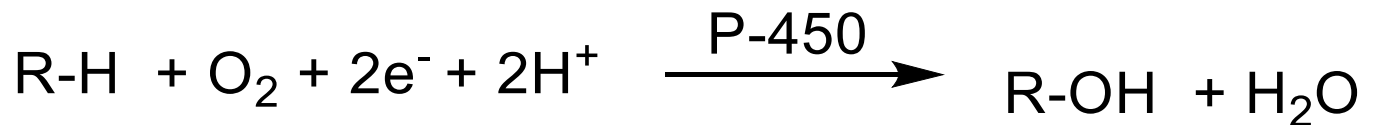
Monoossigenasi



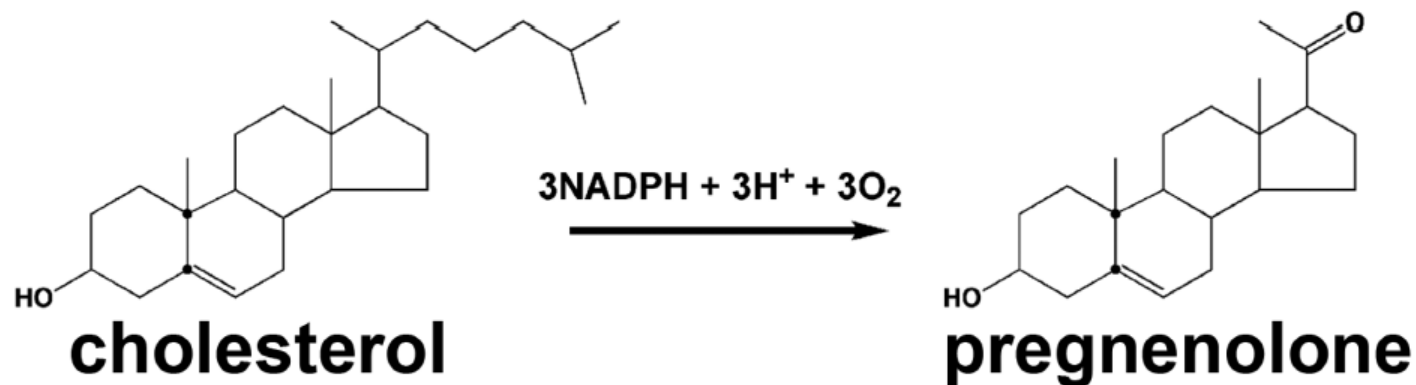
*individuati più di
18.000 P450, di
cui 57 nell'uomo*

Citocromo P-450

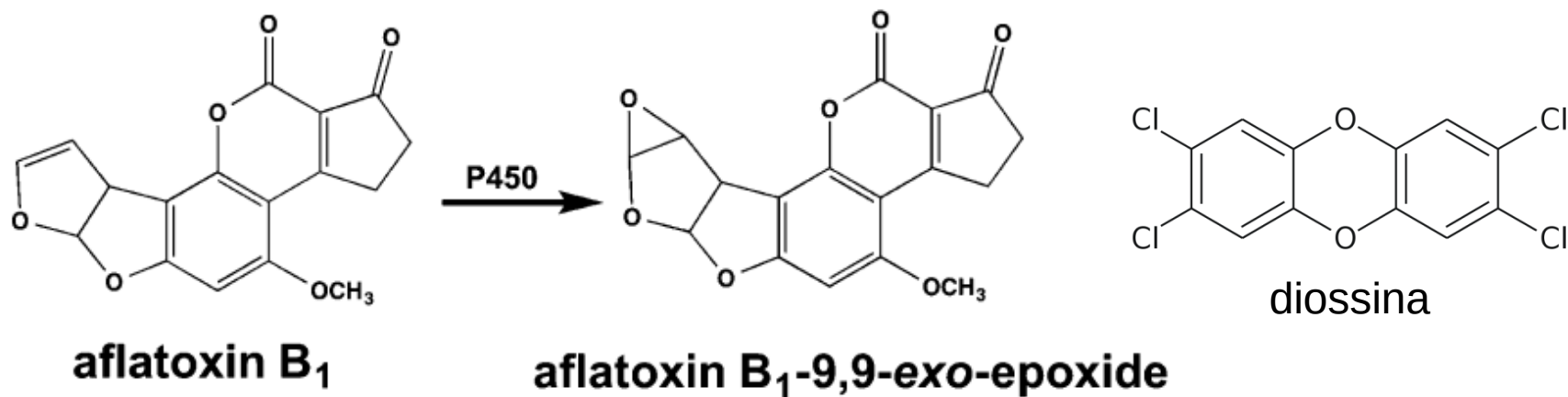
(da *Pseudomonas putida*) 50 kDa



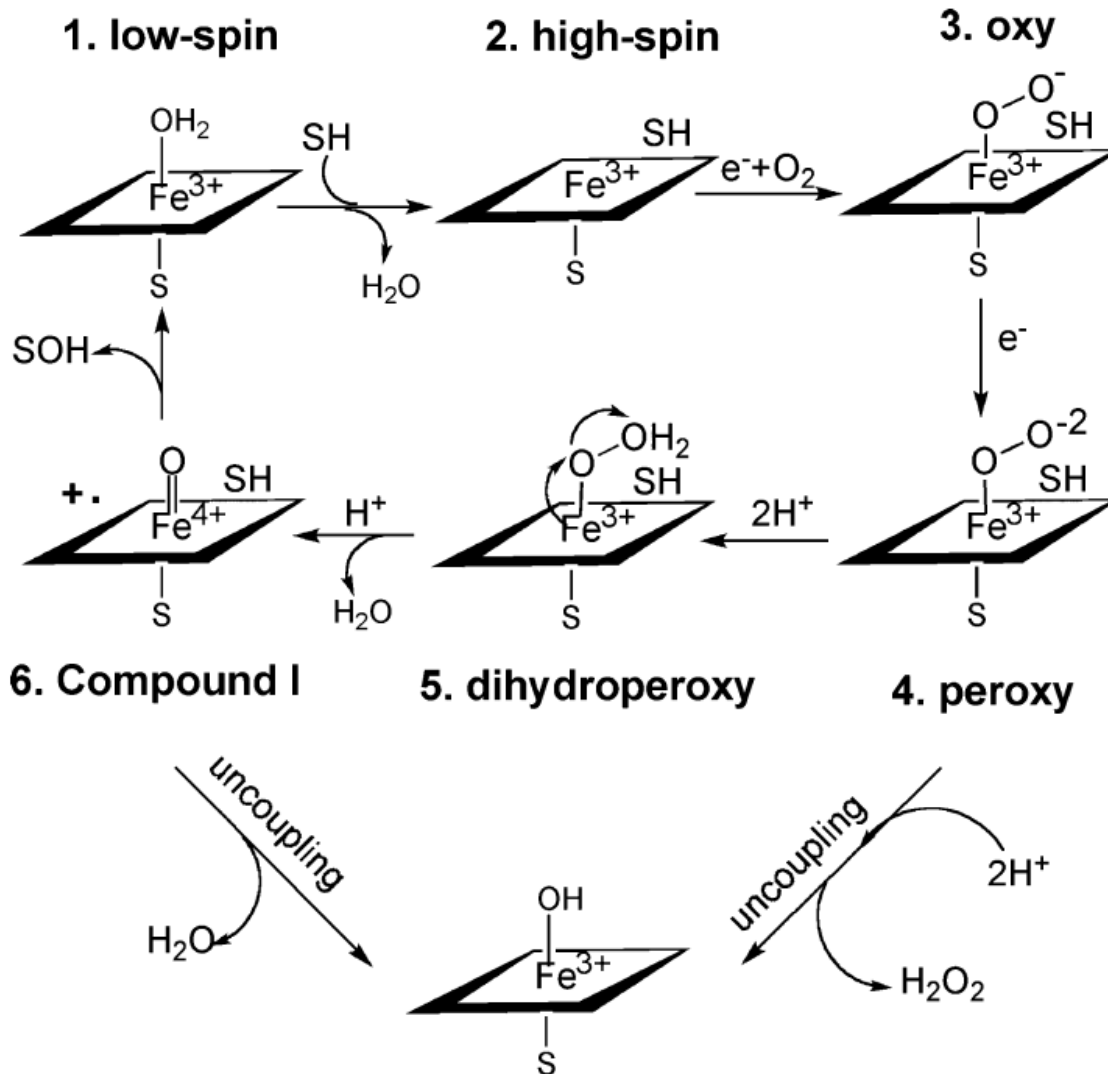
Ossigenazione di substrati specifici da parte di P450



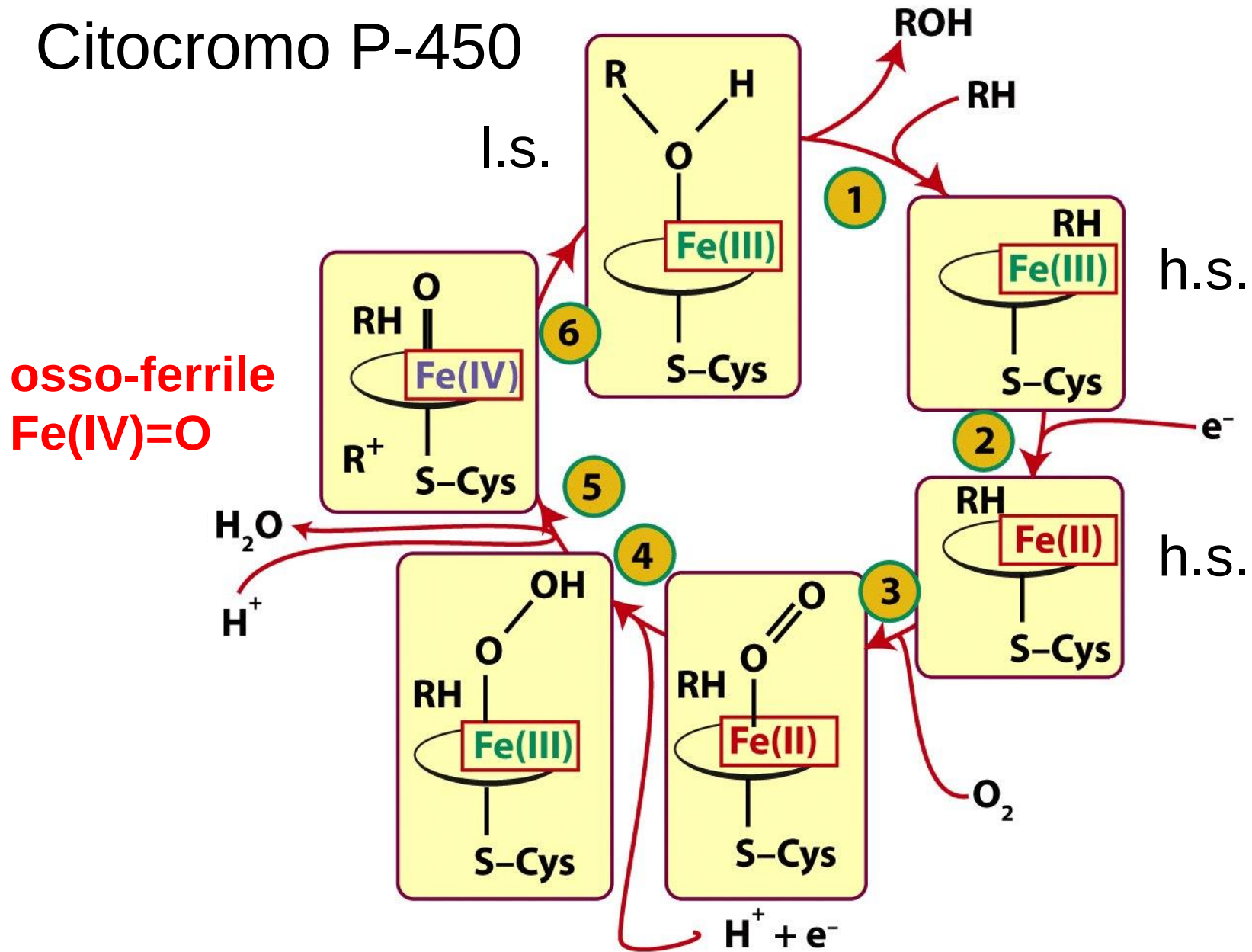
Attivazione (involontaria) di substrati da parte di P450



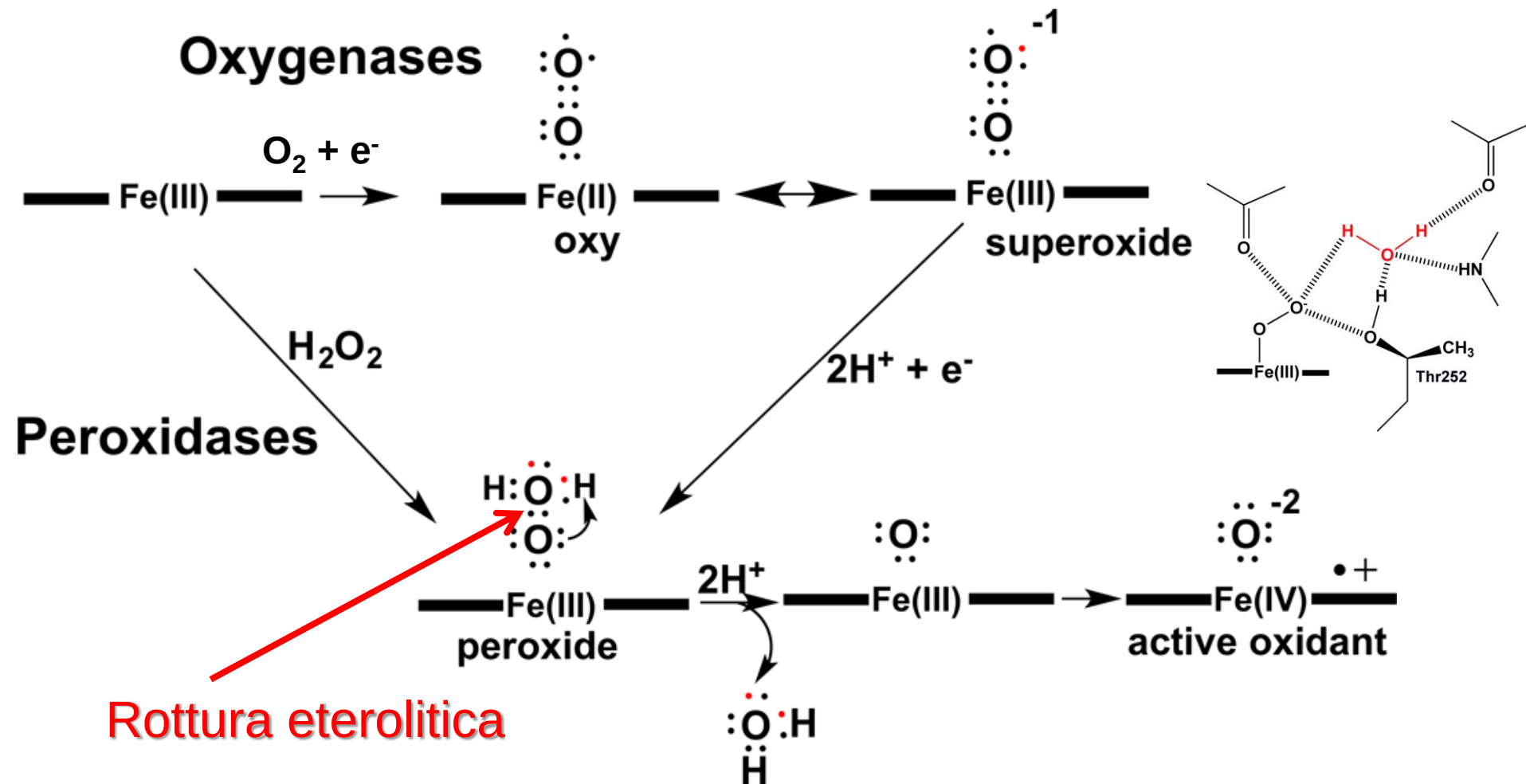
Citocromo P-450



Citocromo P-450

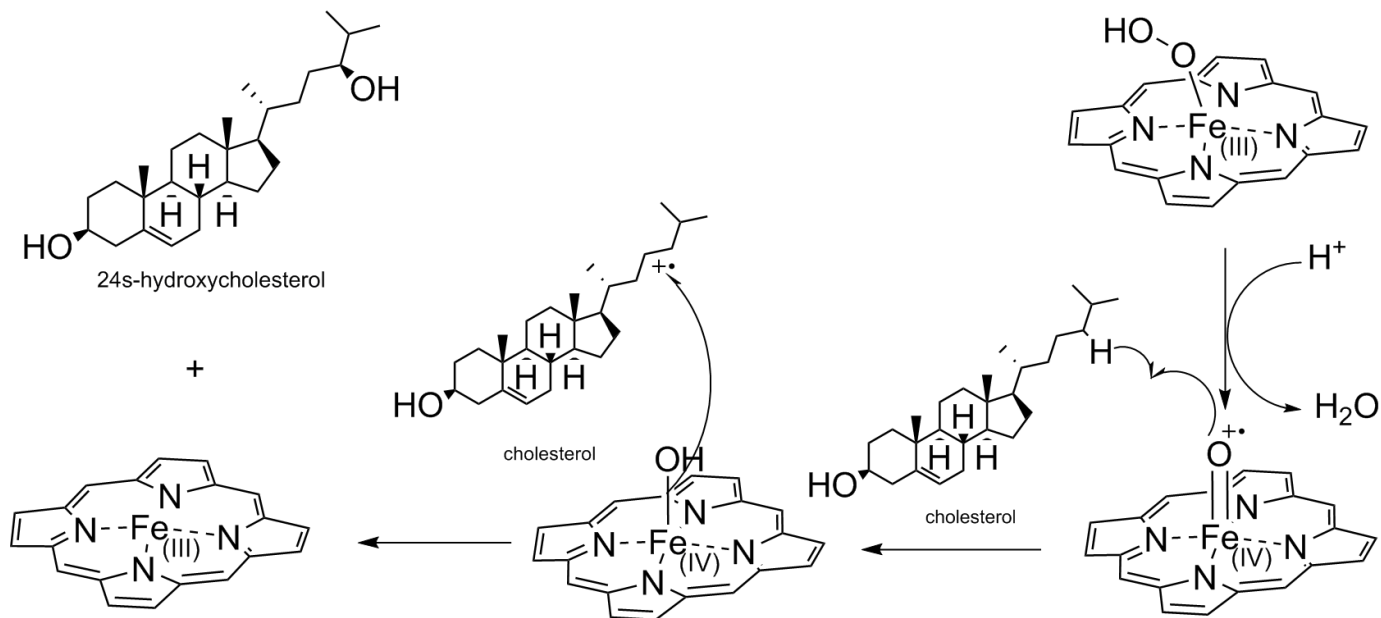
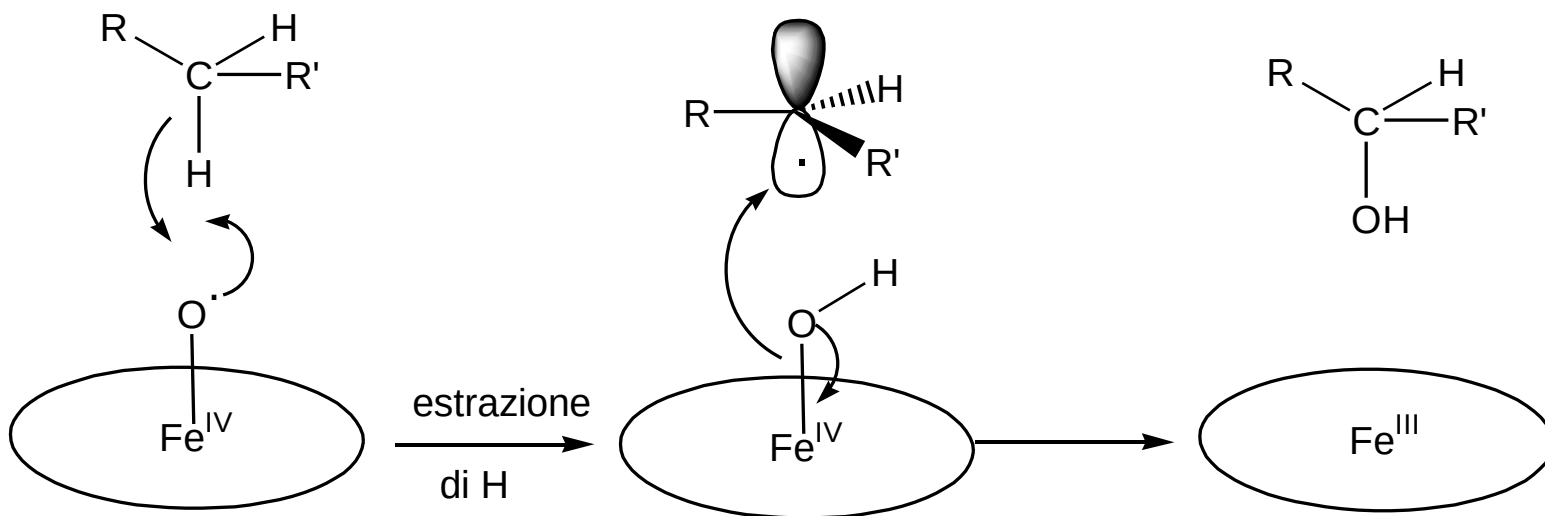


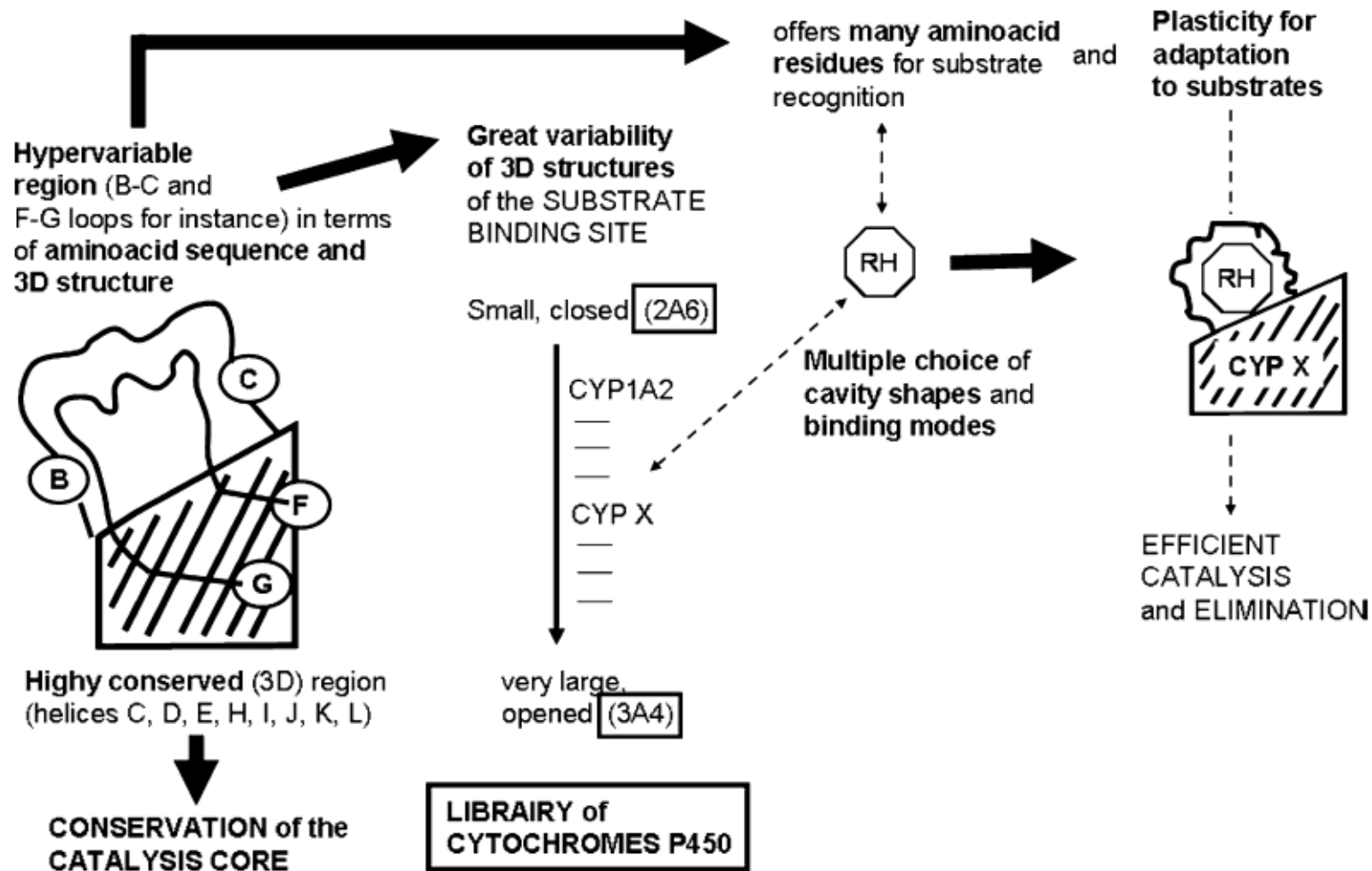
Rottura eterolitica del legame O–O e protonazione dell'ossigeno distale



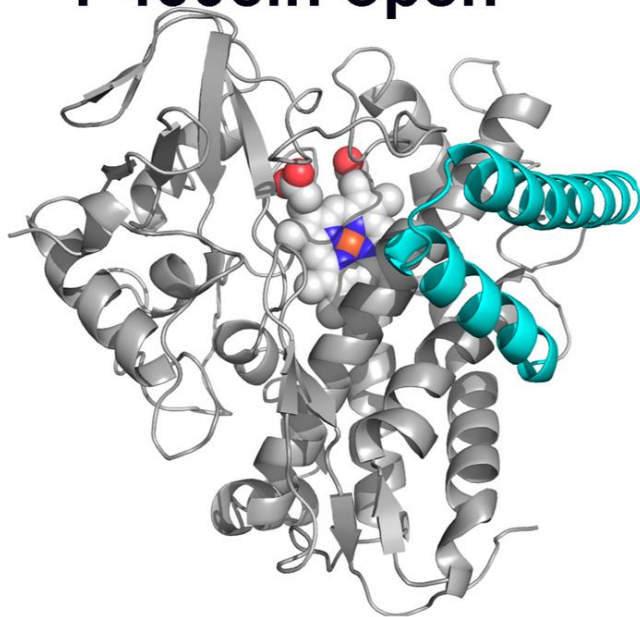
Meccanismo di ossidazione del substrato

Oxygen Rebound Mechanism

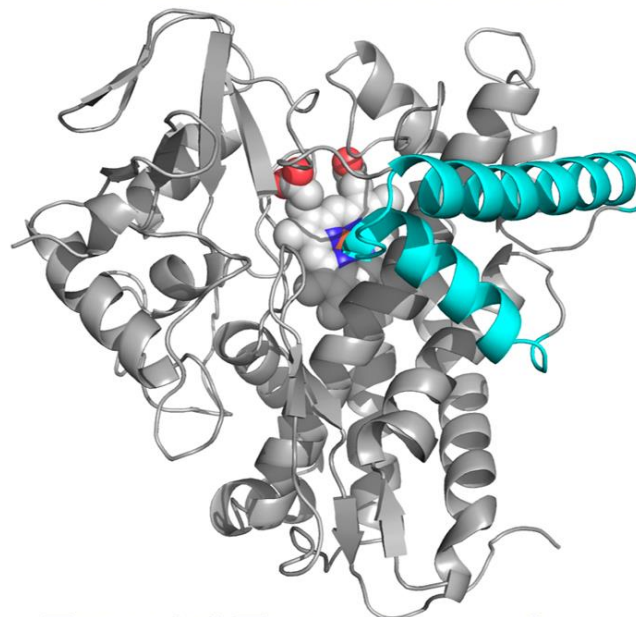




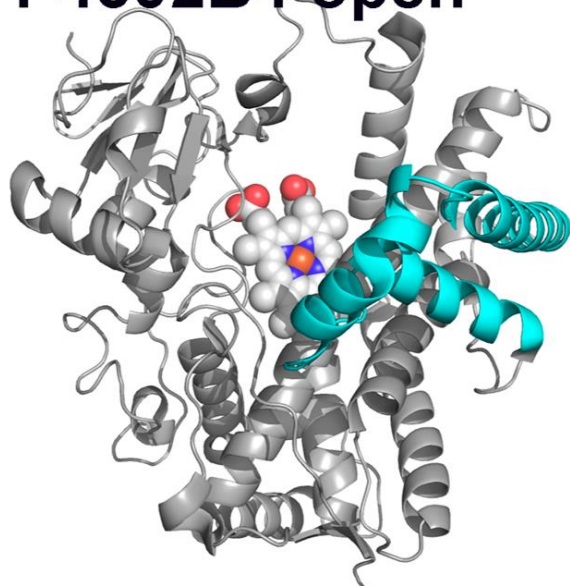
P450cin open



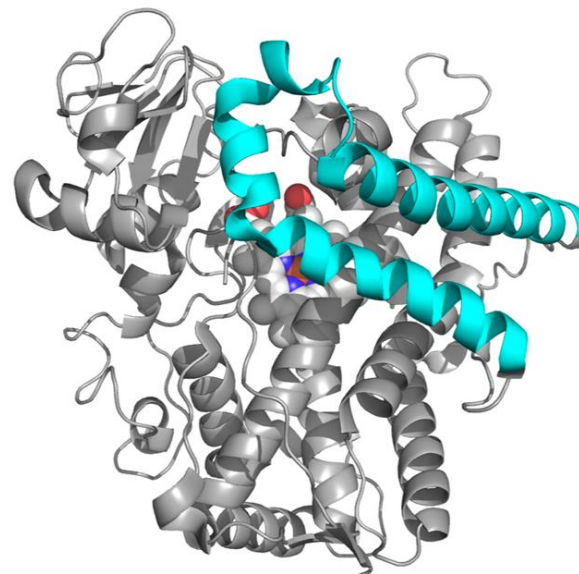
P450cin closed



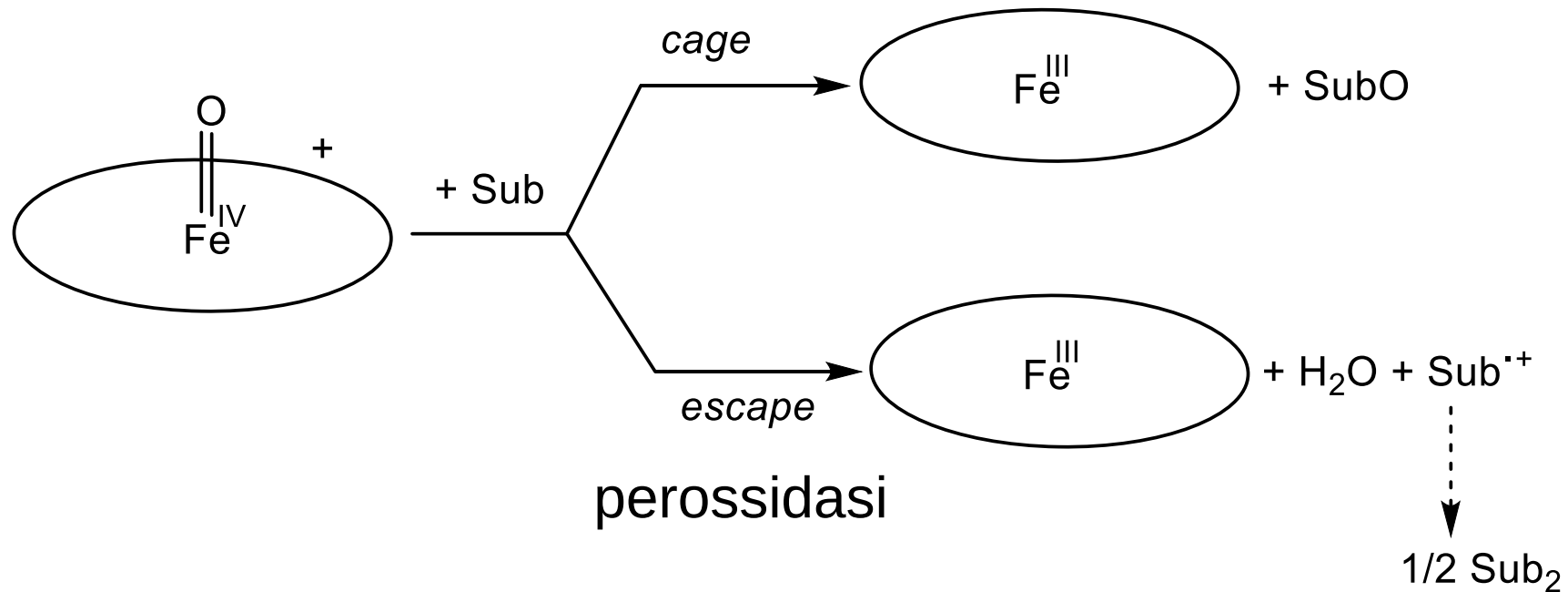
P4502B4 open



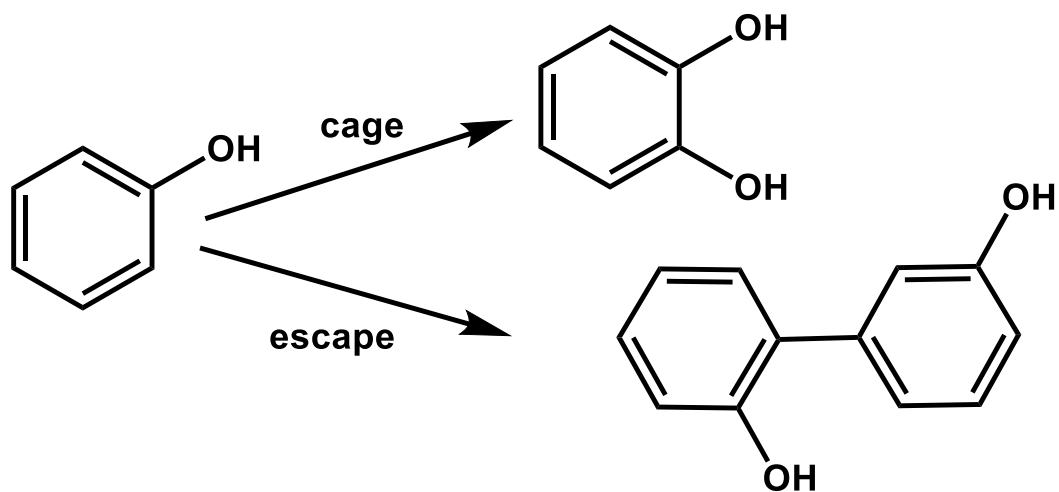
P4502B4 closed

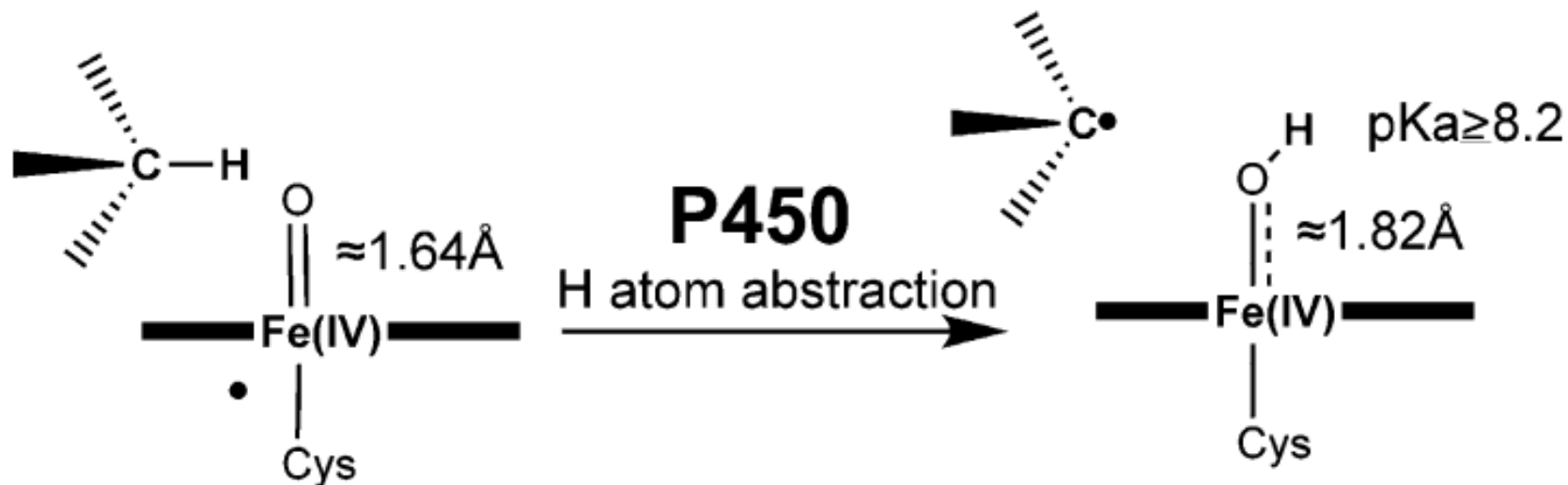


P450



perossidasi

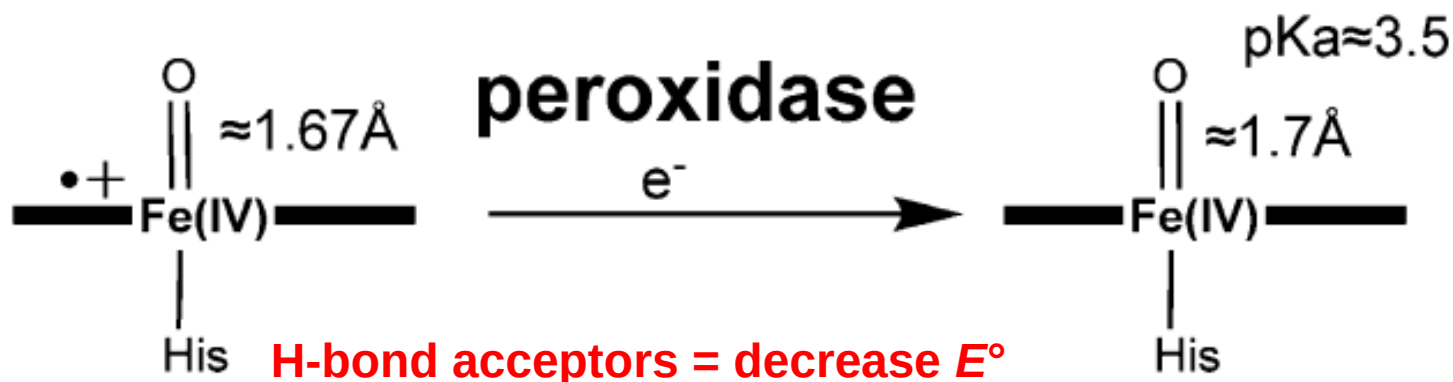




H-bond donors = increase E°

Compound I

Compound II



H-bond acceptors = decrease E°