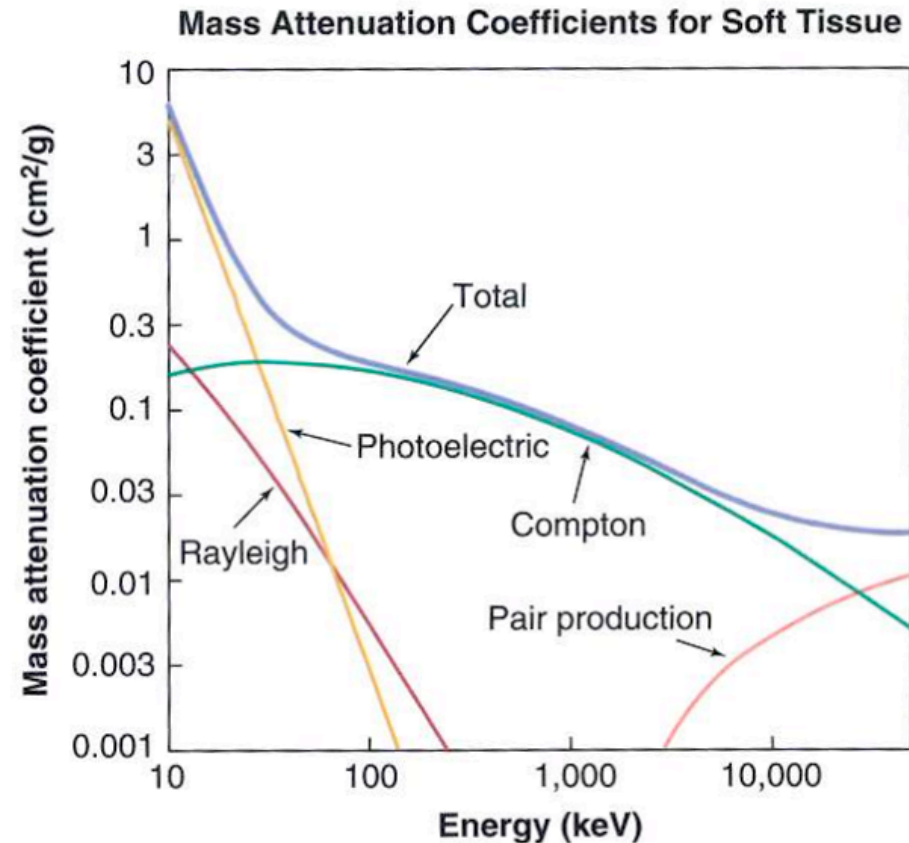


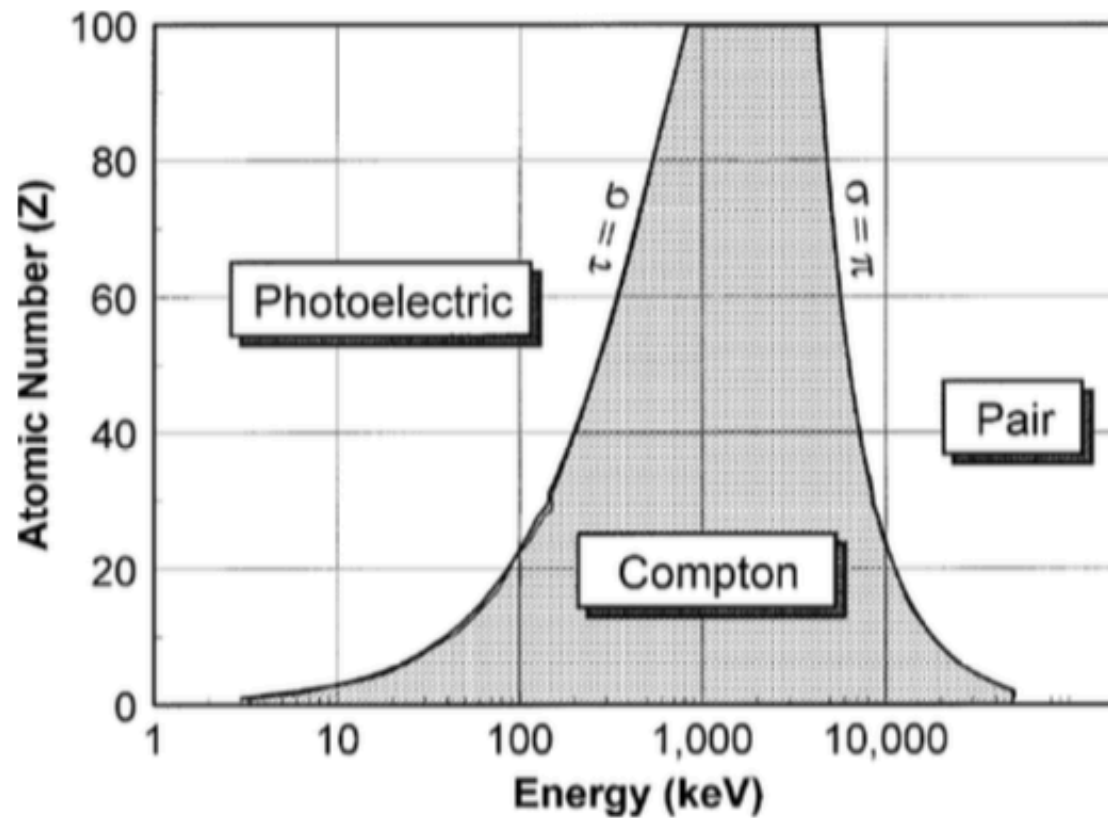
## X-ray interaction with matter

The 4 major interactions of x-ray and gamma-ray photons with matter are:

- Photoelectric effect
- Rayleigh scattering
- Compton scattering
- Pair production (above 1.022 MeV)

■ **FIGURE 3-13** Graph of the Rayleigh, photoelectric, Compton, pair production, and total mass attenuation coefficients for soft tissue ( $Z \approx 7$ ) as a function of photon energy.





**Figure 1.27:** The region where each x-ray interaction process is most likely is shown as a function of atomic number and x-ray energy. The transition zones between regions correspond to the two cross sections being equal ( $\tau = \sigma$  and  $\sigma = \pi$ ).

## Linear attenuation coefficient

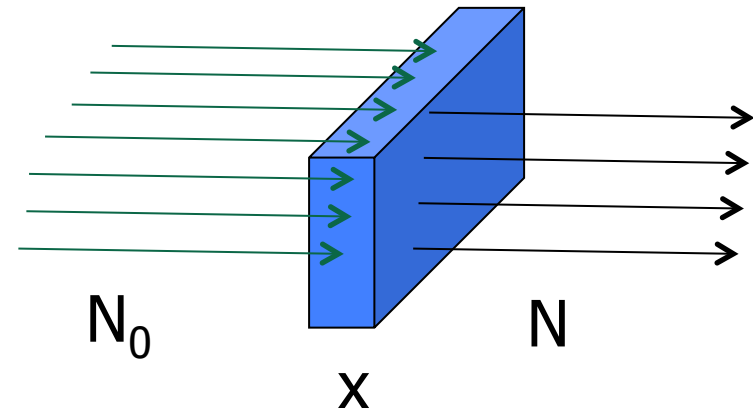
The fraction of photons removed from a monoenergetic beam of x-rays or gamma rays per unit thickness of material is called the *linear attenuation coefficient* ( $\mu$ ), typically expressed in units of inverse centimeters ( $\text{cm}^{-1}$ ). The number of photons removed from the beam traversing a very small thickness  $\Delta x$  can be expressed as

$$n = \mu N \Delta x \quad [3-4]$$

where  $n$  = the number of photons removed from the beam, and  $N$  = the number of photons incident on the material.

The relationship for thin slab  $\Delta X$ , can be integrated. If  $N_0$  is the number of the incident photons and  $N$  the transmitted photons through a thickness  $x$  without any interaction (primary):

$$N = N_0 e^{-\mu x}$$



## Linear attenuation coefficient mathematical model

Let's consider monoenergetic photons through an absorber of infinitesimal thickness  $dx$  of the same material and the same density

If  $n$  is the number of photons removed from the beam:  $n = -dN$

where  $-dN$  is the variation in photon number; minus sign indicating that the intensity is reduced by the absorber

$$-dN = \mu N dx$$

Dividing by  $N$

$$-\frac{dN}{N} = \mu dx$$

This equation describes the situation for thin absorber with thickness  $dx$ .  
For the thickness  $x$  of an absorber we integrate the above equation.  
From thickness 0 to  $x$ , the radiation intensity will decrease from  $N_0$  to  $N$ :

$$-\int_{N_0}^N \frac{dN}{N} = \mu \int_0^x dx \qquad \ln\left(\frac{N}{N_0}\right) = -\mu x$$

$$\frac{N}{N_0} = \exp(-\mu x)$$

## Linear attenuation coefficient

The linear attenuation coefficient is the sum of the individual linear attenuation coefficients for each type of interaction:

$$\mu = \mu_{\text{Rayleigh}} + \mu_{\text{photoelectric effect}} + \mu_{\text{Compton scatter}} + \mu_{\text{pair production}} \quad [3-6]$$

In the diagnostic energy range, the linear attenuation coefficient decreases with increasing energy except at absorption edges (e.g., K-edge). The linear attenuation coefficient for soft tissue ranges from approximately  $0.35$  to  $0.16 \text{ cm}^{-1}$  for photon energies ranging from  $30$  to  $100 \text{ keV}$ .

For a given thickness of material, the probability of interaction depends on the number of atoms the x-rays or gamma rays encounter per unit distance. The density ( $\rho$ , in  $\text{g/cm}^3$ ) of the material affects this number. For example, if the density is doubled, the photons will encounter twice as many atoms per unit distance through the material. Thus, the linear attenuation coefficient is proportional to the density of the material, for instance:

$$\mu_{\text{water}} > \mu_{\text{ice}} > \mu_{\text{water vapor}}$$

## Mass attenuation coefficient

For a given material and thickness, the probability of interaction is proportional to the number of atoms per volume. This dependency can be overcome by normalizing the linear attenuation coefficient for the density of the material. The linear attenuation coefficient, normalized to unit density, is called the *mass attenuation coefficient*.

$$\begin{aligned} \text{Mass Attenuation Coefficient } (\mu / \rho) [\text{cm}^2 / \text{g}] \\ = \frac{\text{Linear Attenuation Coefficient } (\mu) \text{cm}^{-1}}{\text{Density of Material } (\rho) [\text{g}/\text{cm}^3]} \quad [3-7] \end{aligned}$$

The linear attenuation coefficient is usually expressed in units of  $\text{cm}^{-1}$ , whereas the units of the mass attenuation coefficient are usually  $\text{cm}^2/\text{g}$ .

The mass attenuation coefficient is *independent* of density. Therefore, for a given photon energy,

$$\mu_{\text{water}} / \rho_{\text{water}} = \mu_{\text{ice}} / \rho_{\text{ice}} = \mu_{\text{water vapor}} / \rho_{\text{water vapor}}$$

However, in radiology, we do not usually compare equal masses. Instead, we usually compare regions of an image that correspond to irradiation of adjacent volumes of tissue. Therefore, density, the mass contained within a given volume, plays an important role. Thus, one can radiographically visualize ice in a cup of water due to the density difference between the ice and the surrounding water

## Mass attenuation coefficient

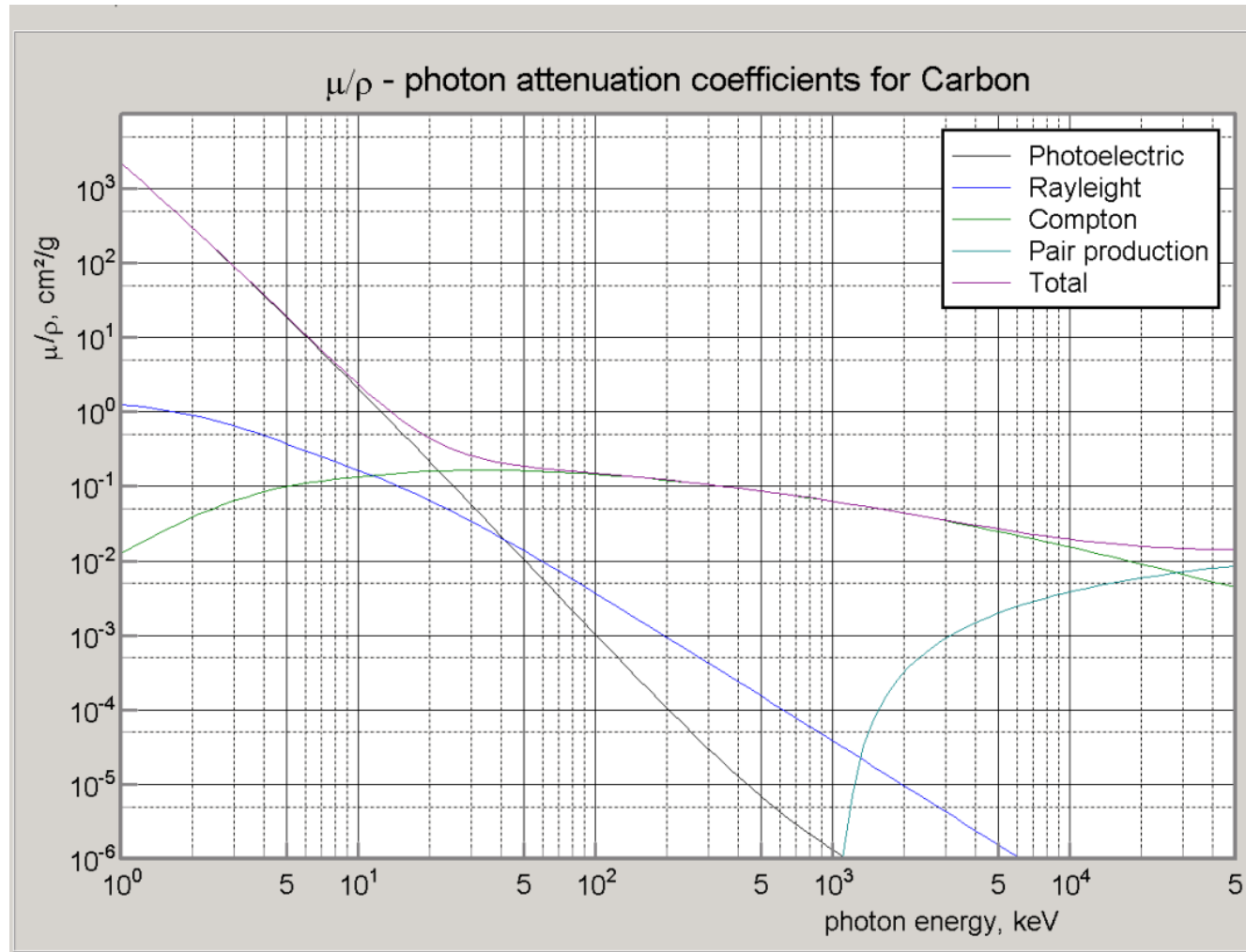
To calculate the linear attenuation coefficient for a density other than 1 g/cm<sup>3</sup>, the density  $\rho$  of the material is multiplied by the mass attenuation coefficient to yield the linear attenuation coefficient. For example, the mass attenuation coefficient of air, for 60-keV photons, is 0.186 cm<sup>2</sup>/g. At typical room conditions, the density of air is 0.00129 g/cm<sup>3</sup>. Therefore, the linear attenuation coefficient of air under these conditions is

$$\mu = (\mu/\rho_0)\rho = (0.186 \text{ cm}^2/\text{g}) (0.00129 \text{ g/cm}^3) = 0.000240 \text{ cm}^{-1}$$

To use the mass attenuation coefficient to compute attenuation, Equation can be rewritten as

$$N = N_0 e^{-\left(\frac{\mu}{\rho}\right)\rho x}$$

# Total photon attenuation coefficient



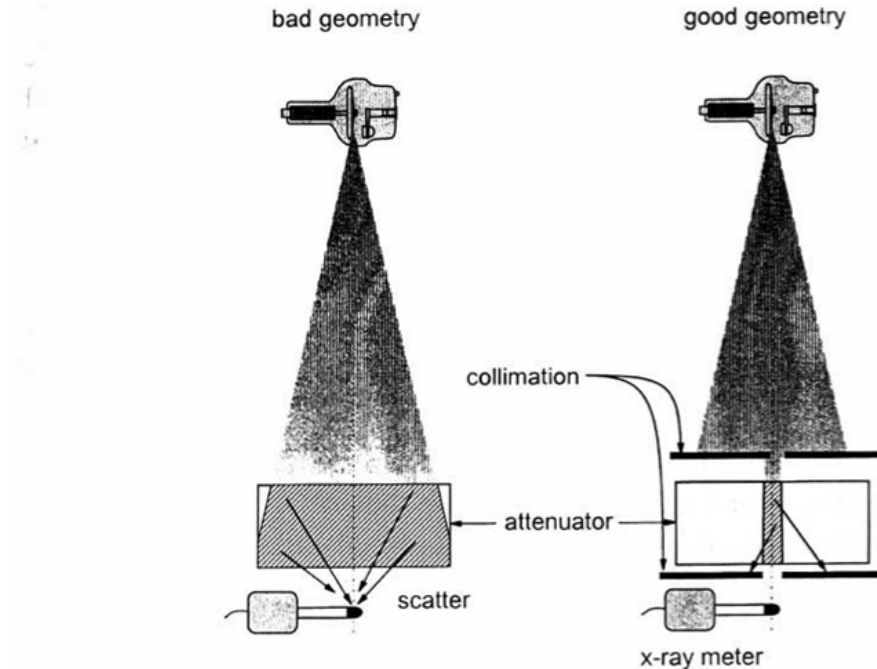


## Linear attenuation coefficient

**TABLE 3-1. MATERIAL DENSITY, ELECTRONS PER MASS, ELECTRON DENSITY, AND THE LINEAR ATTENUATION COEFFICIENT (AT 50 keV) FOR SEVERAL MATERIALS**

Material	Density (g/cm <sup>3</sup> )	Electrons per Mass (e/g) × 10 <sup>23</sup>	Electron Density (e/cm <sup>3</sup> ) × 10 <sup>23</sup>	μ @ 50 keV (cm <sup>-1</sup> )
Hydrogen	0.000084	5.97	0.0005	0.000028
Water vapor	0.000598	3.34	0.002	0.000128
Air	0.00129	3.006	0.0038	0.000290
Fat	0.91	3.34	3.04	0.193
Ice	0.917	3.34	3.06	0.196
Water	1	3.34	3.34	0.214
Compact bone	1.85	3.192	5.91	0.573

## Good and bad geometry

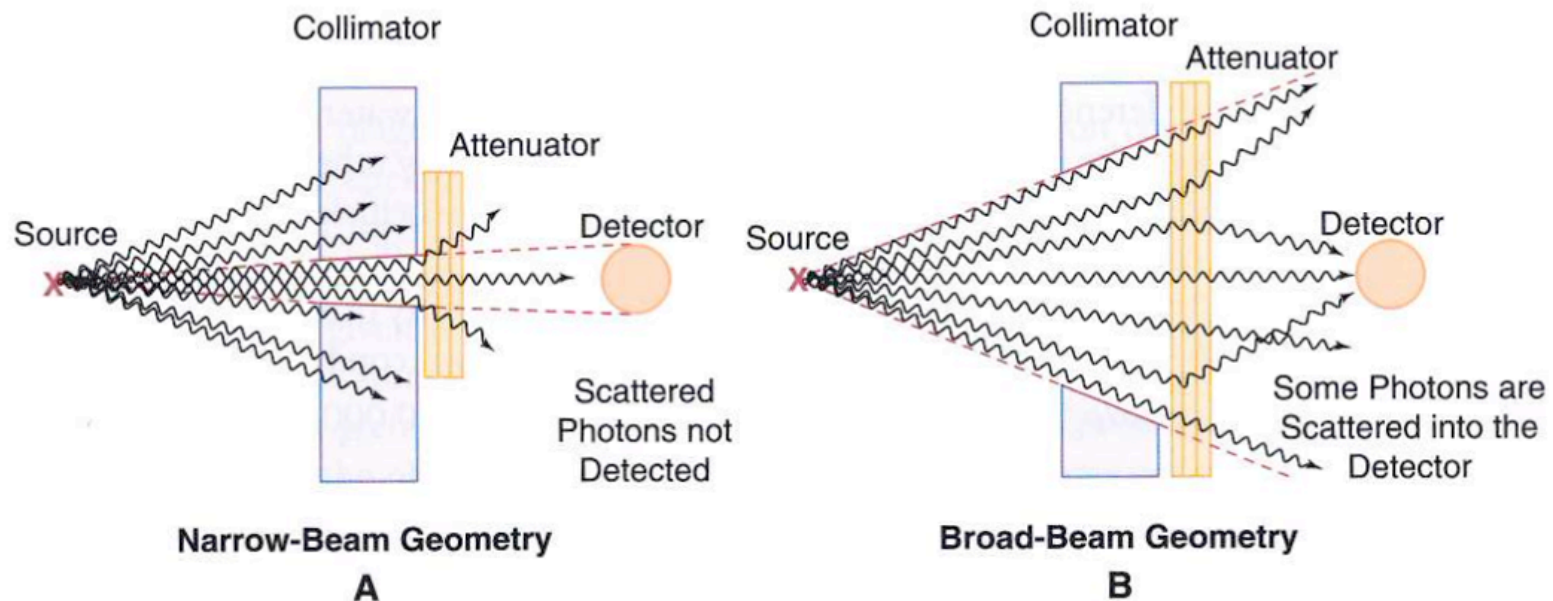


**Figure 1.32:** The so-called *bad geometry* and *good geometry* for making x-ray attenuation measurements are illustrated. Bad geometry exists whenever the exposure measurement includes an appreciable amount of x-ray scatter from the attenuator. The scatter contribution to the measurement can be reduced by using pre-attenuator collimation to limit the x-ray field and post-attenuator collimation to reduce the chance of scatter reaching the x-ray detector.

Attenuation is the removal of photons from the x-ray beam, both by absorption and scattering. To assess attenuation, the x-ray beam that is *not* removed by attenuation is what is actually measured (the primary x-ray beam). It is important that scattered photons not be included in this measurement. Because scattered x rays tend to fly about in all directions near an object being exposed to an x-ray beam, it is important to use a measurement geometry which excludes the measurement of scattered x-ray photons, to the extent possible.

## Half-value layer

The half-value layer (HVL) is defined as the thickness of material required to reduce the intensity (e.g., air kerma rate) of an x-ray or gamma-ray beam to one half of its initial value. The HVL of a beam is an indirect measure of the photon energies (also referred to as the *quality*) of a beam, when measured under conditions of *narrow-beam geometry*. Narrow-beam geometry refers to an experimental configuration that is designed to exclude scattered photons from being measured by the detector (Fig. 3-15A). In *broad-beam geometry*, the beam is sufficiently wide that a substantial fraction of scattered photons remain in the beam.



■ **FIGURE 3-15** **A.** Narrow-beam geometry means that the relationship between the source shield and the detector is such that almost no scattered photons interact with the detector. **B.** In broad-beam geometry, scattered photons may reach the detector; thus, the measured attenuation is less compared with narrow-beam conditions.

Underestimation of attenuation means overestimation of HVL

## Half-value layer

The HVL of a diagnostic x-ray beam, measured in millimeters of aluminum under narrow beam conditions, is a measure of the penetrability of the x-ray spectrum.

### Relationship between $\mu$ and HVL

N is equal to  $N_0/2$  when the thickness of the absorber is 1 HVL. Thus, for monoenergetic beam:

$$\begin{aligned}N_0/2 &= N_0 e^{-\mu(\text{HVL})} \\1/2 &= e^{-\mu(\text{HVL})} \\ \ln(1/2) &= \ln e^{-\mu(\text{HVL})} \\ -0.693 &= -\mu(\text{HVL}) \\ \text{HVL} &= 0.693/\mu\end{aligned}$$

## Half-value layer

For a monoenergetic incident photon beam, the HVL can be easily calculated from the linear attenuation coefficient, and vice versa. For example, given

1.  $\mu = 0.35 \text{ cm}^{-1}$

$$\text{HVL} = 0.693/0.35 \text{ cm}^{-1} = 1.98 \text{ cm}$$

2.  $\text{HVL} = 2.5 \text{ mm} = 0.25 \text{ cm}$

$$\mu = 0.693/0.25 \text{ cm} = 2.8 \text{ cm}^{-1}$$

The HVL and  $\mu$  can also be calculated if the percent transmission is measured under narrow-beam geometry.

**EXAMPLE:** If a 0.2-cm thickness of material transmits 25% of a monoenergetic beam of photons, calculate the HVL of the beam for that material.

**STEP 1.**  $0.25 = e^{-\mu(0.2 \text{ cm})}$

**STEP 2.**  $\ln 0.25 = -\mu(0.2 \text{ cm})$

**STEP 3.**  $\mu = (-\ln 0.25)/(0.2 \text{ cm}) = 6.93 \text{ cm}^{-1}$

**STEP 4.**  $\text{HVL} = 0.693/\mu = 0.693/6.93 \text{ cm}^{-1} = 0.1 \text{ cm}$

## Half-value layer

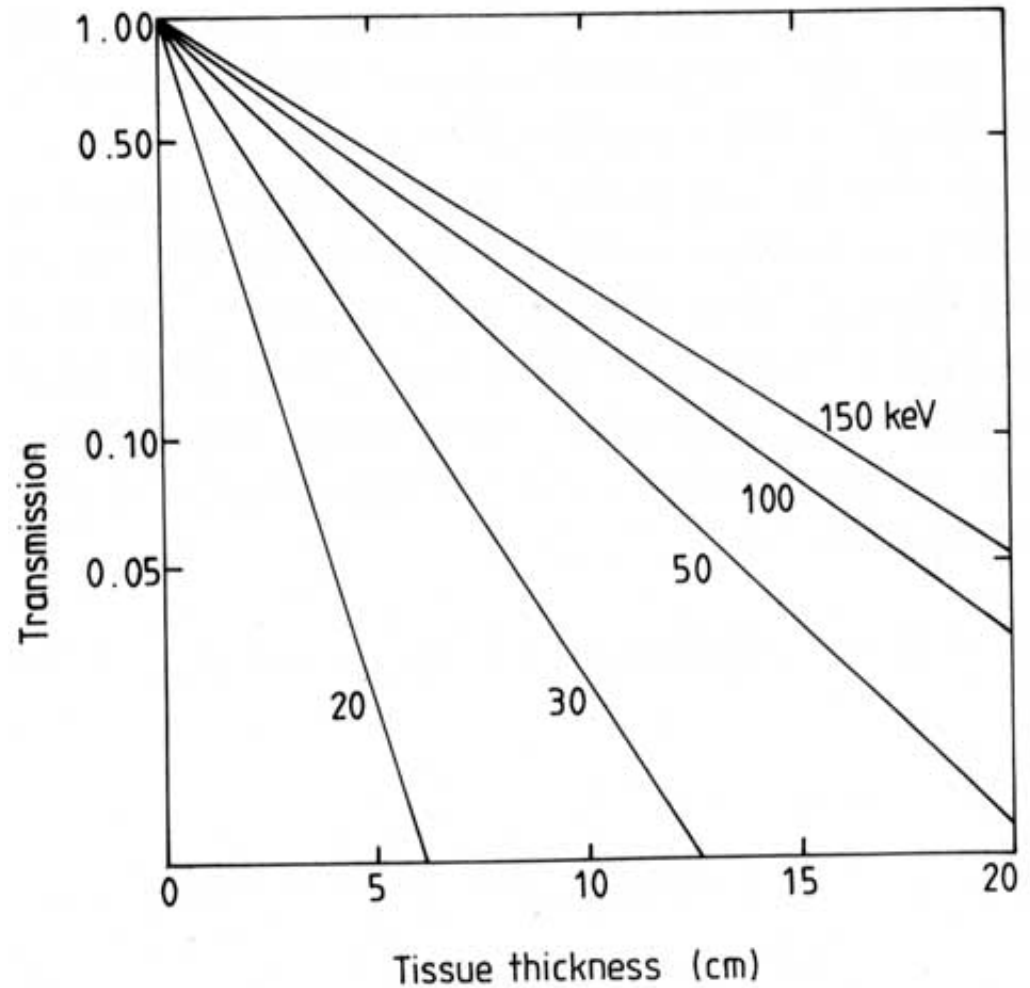
For monoenergetic photons under narrow-beam geometry conditions, the probability of attenuation remains the same for each additional HVL thickness placed in the beam. Reduction in beam intensity can be expressed as  $(1/2)^n$  where  $n$  equals the number of half value layers. For example, the fraction of monoenergetic photons transmitted through 5 HVLs of material is

$$1/2 \times 1/2 \times 1/2 \times 1/2 \times 1/2 = (1/2)^5 = 1/32 = 0.031 \text{ or } 3.1\%$$

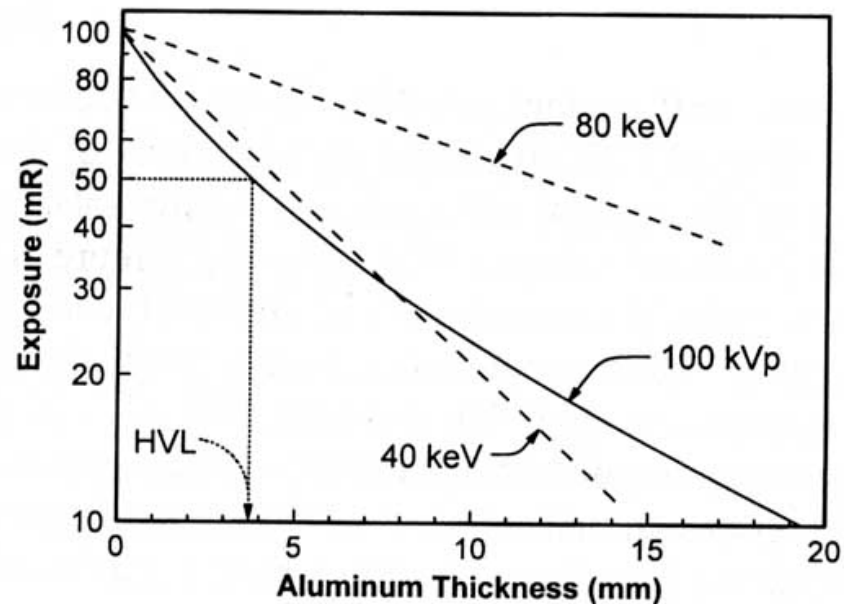
Therefore, 97% of the photons are attenuated (removed from the beam). The HVL of a diagnostic x-ray beam, measured in millimeters of aluminum under narrow-beam conditions, is a surrogate measure of the average energy of the photons in the beam.

## Monochromatic beam attenuation

- Semi-logarithmic graph: transmission decreases linearly
- Smaller the energy, higher the attenuation: the slope is higher
- Beam attenuation for monoenergetic beams for different tissue thickness (in cm)



## Polychromatic beam attenuation



**Figure 1.33:** Attenuation profiles (exposure as a function of thickness) are shown for aluminum and for three different x-ray beams. The dashed lines are for monoenergetic x-ray beams at 40 keV and 80 keV. Higher-energy x rays are more *penetrating*, so the curve for 80-keV x rays is less steep than the curve for 40-keV x rays. On this semi-logarithmic plot, the attenuation curves for monoenergetic x-ray beams appear as straight lines. For a 100-kVp x-ray spectrum, the attenuation curve demonstrates curvature which is representative of *beam hardening*. The half-value layer (HVL) is the thickness of aluminum required to reduce the exposure of the x-ray beam by 50%. The HVL for the 100-kVp attenuation curve shown is approximately 3.7 mm Al.

The homogeneity coefficient is the ratio of the first to the second HVL and describes the polyenergetic character of the beam. The first HVL is the thickness that reduces the incident intensity to 50% and the second HVL reduces it to 25% of its original intensity [i.e.,  $(0.5)(0.5) = 0.25$ ]. A monoenergetic source of gamma rays has a homogeneity coefficient equal to 1.



## Average energy

The conventional notation *for photon fluence* (photons/mm<sup>2</sup>) at a given energy E is  $\phi(E)$ .

The corresponding *energy fluence* (joules/mm<sup>2</sup> or  $\phi(E) \times E$ ) is  $\psi(E)$

In case of polyenergetic beams the concept is replaced by fluence spectrum and energy fluence spectrum, differential in energy E (fluence/rate if per unit of time )

$$\Phi_E(E) \equiv \frac{d\Phi}{dE}(E) \quad \Psi_E(E) \equiv \frac{d\Psi}{dE}(E) = \frac{d\Phi}{dE}(E)E$$

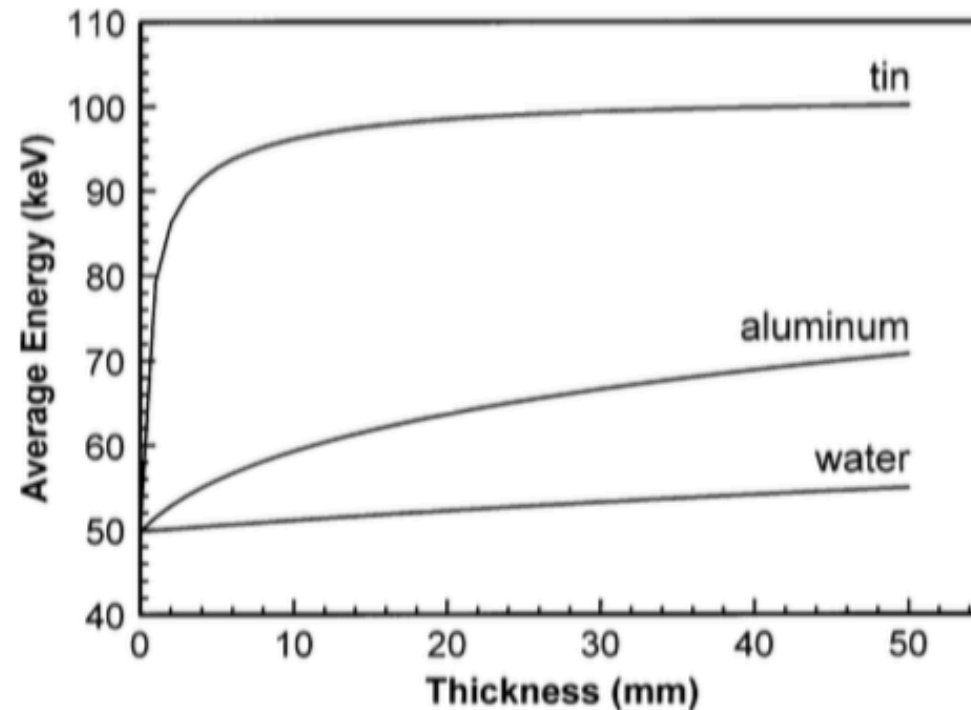
$$\bar{E} = \frac{\int_0^{E_{\max}} \phi(E) E dE}{\int_0^{E_{\max}} \phi(E) dE} \quad \bar{E}_{att} = \frac{\int_0^{E_{\max}} \phi(E) e^{-\mu(E)x} E dE}{\int_0^{E_{\max}} \phi(E) e^{-\mu(E)x} dE}$$

We can calculate the average energy of a given spectrum and the average energy of an attenuated spectrum after x mm of Al.

$$\bar{E} = \frac{\sum_i \phi_i(E_i) E_i}{\sum_i \phi_i(E_i)} \quad \bar{E} = \frac{\sum_i \phi_i(E_i) e^{-\mu(E_i)x} E_i}{\sum_i \phi_i(E_i) e^{-\mu(E_i)x}}$$

$\phi_i(E_i)$  is the total photon fluence in an i-th energy interval with bin width  $\Delta E$

## Beam hardening



**Figure 1.36:** The average x-ray energy in a 100-kVp x-ray spectrum is illustrated as a function of thickness for three different attenuators. Fifty millimeters of water produces only a modest increase in average x-ray energy, whereas 50 mm of aluminum causes a more noticeable increase in average energy. Because of the high atomic number and density of tin ( $Z = 50$ ,  $\rho = 7.3 \text{ g/cm}^3$ ), it has a profound beam-hardening effect.

# Radiological Units

**Table 6**  
**Radiologic Units**

Quantity	Description	Conventional Unit*
Fluence	Number of photons per unit area	1/centimeter <sup>2</sup> [1/meter <sup>2</sup> ]
Flux (fluence rate)	Fluence per unit time	1/(centimeter <sup>2</sup> · second) [1/(meter <sup>2</sup> · second)]
Intensity (energy fluence)	Number of photons times photon energy per unit area	kiloelectron volt/centimeter <sup>2</sup> [joule/meter <sup>2</sup> ]
Exposure (X)	Charge produced per unit mass of air from x and gamma rays	roentgen [coulomb/kilogram] <sup>†</sup>
Kerma (K)	Kinetic energy released in matter per unit mass	rad [joule/kilogram or gray] <sup>‡</sup>
Dose (D)	Energy absorbed per unit mass	rad [joule/kilogram or gray] <sup>‡</sup>

\*SI units are given in brackets.

<sup>†</sup> 1 roentgen =  $2.58 \times 10^{-4}$  coulomb/kilogram.

<sup>‡</sup> 100 rad = 1 gray.

# Exposure

ICRU (1980) International Commission on Radiation Units and Measurements

Exposure is defined as  $dQ/dm$ , where  $dQ$  is the sum of the electrical charges of all ions of one sign produced in air when all the electrons and positrons liberated by photons in a volume of air whose mass is  $dm$  are completely stopped in air.

$$X = dQ/dM$$

*The ionization arising from the absorption of bremsstrahlung or annihilation radiation emitted by the electrons is not to be included in  $dQ$ .*

*Owing to the difficulty of measurement, exposure is not normally used when the photon energy exceeds 3 MeV.*

Unita' di misura:

Sistema Internazionale C kg<sup>-1</sup>

Roentgen R = 2.58 10<sup>-4</sup> C kg<sup>-1</sup>

*NB: Average atomic number of air and soft tissue is about the same !*

## Exposure and absorbed dose in air

How do we measure them X rays? X-ray beam can be measured using the ionization it produces in air. An *ionization chamber* is an air-filled chamber surrounded by electrodes (a positive and negative electrode). X rays ionize the molecules of air present in the chamber, and the electrons follow the electric field lines and are collected on the positive electrode, while the positive ions are collected on the negative electrode. The net charge is collected on the electrodes. The unit of exposure is the roentgen (R), where:

$$1R = 2.58 \times 10^{-4} \text{ C/kg}$$

A typical ionization chamber for general diagnostic measurements has a volume of approximately 6 cm<sup>3</sup>. At standard temperature and pressure, the mass of air in 6 cm<sup>3</sup> is about 7.8 mg. A 1-R exposure will liberate a charge of  $2.0 \times 10^{-9}$  coulombs inside the chamber, corresponding to  $1.2 \times 10^{10}$  ions.

The roentgen is defined only in air, and under conditions of electron equilibrium. Electron equilibrium occurs when the number of energetic ions entering the measurement volume equal those leaving it.

Empirically it takes  $W=33.97$  eV to produce an ion pair in air, equal to 33.97 joules/C. Thus the energy absorbed in air/mass, i.e. *absorbed dose in air*, by a 1-R exposure is:

$$D_{\text{air}} = E_{\text{abs}}/m = 2.58 \times 10^{-4} \text{ C/kg} \times 33.97 \text{ J/C} = 0.00876 \text{ J/kg} = 0.00876 \text{ Gy}$$

Exposure (X) corresponds to *absorbed dose* in air:

$$D_{\text{air}} \text{ (mGy)} = 0.00876 \text{ X(mR)}$$

$$D_{\text{air}} \text{ (mGy)} = 8.76 \text{ X(R)}$$

## Exposure and attenuation calculation

We can calculate the exposure ( $\sim$ charge per unit of mass in air) from the energy absorbed/m

$$\frac{\Delta Q}{\Delta m} = \frac{1}{W} \int_0^{E_{\max}} \phi(E) E \frac{\mu_{en}(E)}{\rho} dE$$

$$D_{air}(E) = \phi(E) E \frac{\mu_{en}(E)}{\rho}$$

$$W = 34 \text{ J/C}$$

For an x-ray spectrum consisting of many different energies (a polyenergetic spectrum), the attenuation curve of the exposure has curvature on the semi-log plot, indicating a slight deviation from exponential falloff.

The attenuation  $A(x)$  of the exposure of a polyenergetic x-ray spectrum from a sheet of aluminum of thickness  $x$  is given by:

$$A(x) = \frac{\int_0^{E_{\max}} \phi(E) \xi(E)^{-1} e^{-\mu(E)_{Al} x} dE}{\int_0^{E_{\max}} \phi(E) \xi(E)^{-1} dE}$$

where  $\phi(E)$  is in photons/mm<sup>2</sup> and the function  $\xi^{-1}(E)$  is in mR per (photon/mm<sup>2</sup>)

$$\xi^{-1}(E) = 6.26 * 10^{-5} \frac{1}{W} E \left( \frac{\mu_{en}(E)}{\rho} \right)_{air}$$

The x-ray fluence per unit exposure is given by:

$$\xi(E) = \frac{5.43 \times 10^5}{E \left( \frac{\mu_{en}(E)}{\rho} \right)_{air}}$$

When the mass energy absorption coefficient for air ( $\mu_{en}/\rho$ ) is in units of cm<sup>2</sup>/g, and  $E$  is in keV, the units of  $\xi(E)$  are photons/mm<sup>2</sup> per mR.

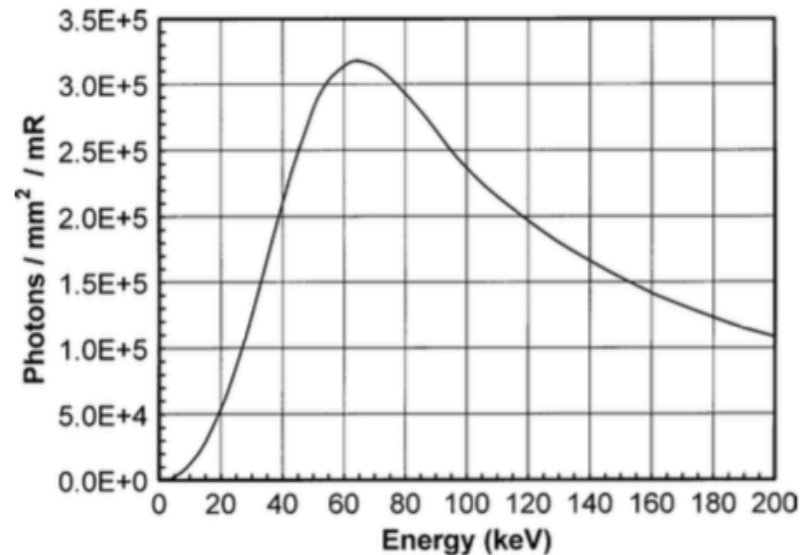
## Fluence/exposure

$\xi(E)$  describes the photon fluence per unit of exposure

For the energy range from 1 to 150 keV, this can be calculated using:

$$\xi(E) = \left[ a + b\sqrt{E}\ln(E) + \frac{c}{E^2} \right]^{-1}$$

where  $a = -5.023 \times 10^{-06}$ ,  $b = 1.81 \times 10^{-07}$ ,  $c = 0.00884$ ,  $E$  is in keV, and  $\xi(E)$  is in the units of photons/mm<sup>2</sup> per mR.



**Figure 1.34:** The photon fluence (photons/mm<sup>2</sup>) per unit exposure (mR) is shown as a function of x-ray energy across the diagnostically relevant energies. X-ray exposure measured in roentgens (or mR) demonstrates energy dependency because the mass energy absorption coefficient for air is energy dependent. This figure illustrates the function  $\xi(E)$  as discussed in the text.

# Exposure and HVL

- An air-ionization exposure meter is a device capable of accurately measuring x-ray exposure. Exposure is a term which relates primarily to the x-ray beam intensity or the *beam quantity*.
- **Measuring the x-ray energy spectrum is much more difficult**, and requires sophisticated equipment that is only available in a handful of laboratories. Nevertheless, some idea of the spectral distribution (*beam quality*) of the x-ray beam. The x-ray attenuation coefficients are energy dependent, and therefore by measuring the exposure attenuation ( $A(x)$ ) of a known material (e.g., aluminum), a parameter relating to the x-ray beam energy distribution ( $\phi(E)$ ) can be assessed.
- The parameter used to characterize polyenergetic beam quality in field **measurements of attenuation is called the *half-value layer (HVL)***. The HVL, usually calculated using aluminum in diagnostic radiology, is the thickness of aluminum required to reduce the exposure of the x-ray beam by a factor of 2 (i.e., to 50% of its unattenuated exposure).
- ***Exposure is defined in air (only), and therefore the HVL is properly measured only using an air ionization exposure meter***; the HVL measured using a solid state x-ray detector system, for example, will be different.



## Effective energy

The HVL of a diagnostic x-ray beam, measured in millimeters of aluminum under narrow beam conditions, is a measure of the penetrability of the x-ray spectrum. Since x-ray beams in radiology are polyenergetic, the determination of HVL is the way of characterizing the penetrability of the x-ray beam for a given kV.

The HVL (in mm of Al) can be converted to a quantity called the **effective energy**.

It is an estimate of the penetration power of the x-ray beam, expressed as the energy of a monoenergetic beam that would exhibit the same "effective" penetrability.

The effective energy from a typical diagnostic x-ray tube is one third to one half the maximum value.

<b>HVL (mm Al)</b>	<b>EFFECTIVE ENERGY (keV)</b>
0.26	14
0.75	20
1.25	24
1.90	28
3.34	35
4.52	40
5.76	45
6.97	50
9.24	60
11.15	70
12.73	80
14.01	90
15.06	100

Al, aluminum.

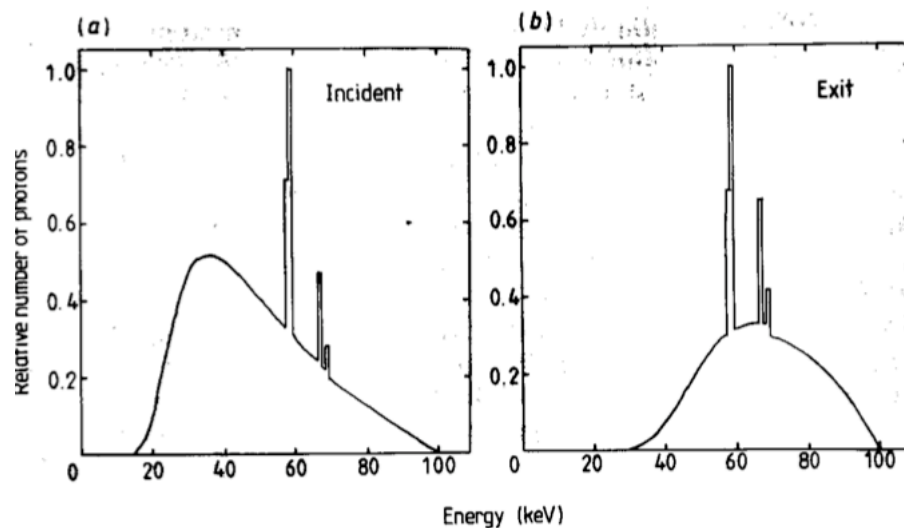
**Table 1.4:** Mass attenuation coefficients of aluminum ( $\rho = 2.699$ )

$E$ (keV)	$(\mu/\rho)_{\text{al}}$	$E$ (keV)	$(\mu/\rho)_{\text{al}}$
10	26.048	50	0.368
11	19.678	55	0.315
12	15.330	60	0.278
13	12.264	65	0.252
14	9.731	70	0.230
15	7.980	75	0.214
16	6.576	80	0.202
17	5.500	85	0.192
18	4.647	90	0.183
19	4.034	95	0.177
20	3.423	100	0.171
25	1.830	105	0.166
30	1.131	110	0.161
35	0.769	115	0.157
40	0.567	120	0.153
45	0.446		

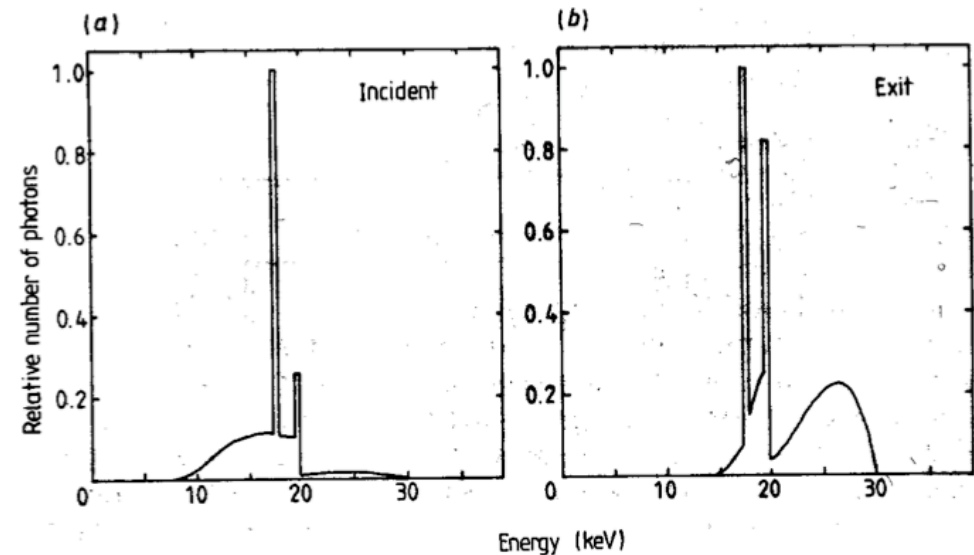
## Beam hardening

The lower energy photons of the polyenergetic x-ray beam will preferentially be removed from the beam while passing through matter. The shift of the x-ray spectrum to higher effective energies as the beam transverses matter is called *beam hardening*

Low-energy (soft) x-rays will not penetrate the entire thickness of the body; thus, their removal reduces patient dose without affecting the diagnostic quality of the exam. X-ray machines remove most of this soft radiation with filters, thin plates of aluminum, copper, or other materials placed in the beam. This added filtration will result in an x-ray beam with a higher effective energy and thus a greater HVL.



**Figure 2.11** X-ray spectra for an x-ray tube with a tungsten target; 100 kV constant potential with 2.5 mm aluminium added. The spectra are shown both before and after attenuation by 18.5 cm soft tissue plus 1.5 cm bone. (The spectra are based on the work of Birch *et al* (1979).)



**Figure 2.12** X-ray spectra for an x-ray tube with a molybdenum target; 30 kV constant potential with 0.03 mm molybdenum filter. The spectra are shown both before and after attenuation by 5 cm tissue. (The spectra are based on the work of Birch *et al* (1979).)

# Photon fluence vs. exposure

- The number of x-ray quanta striking a detector per unit area (the x-ray fluence) is an important experimental parameter. However, the fluence is not directly measurable in most laboratories.
- The function  $\xi(E)$  which is the energy-dependent photon fluence per exposure (photons/mm<sup>2</sup> per mR). Its inverse function,  $\xi^{-1}(E)$ , gives the exposure per fluence (mR per photons/mm<sup>2</sup>). If the spectrum is known and the exposure ( $X$ , in mR) is measured, the photon fluence per exposure for the entire spectrum can be calculated:

$$\hat{\Phi}_{\text{spectrum}} = \frac{\int_{E=0}^{E^{\text{max}}} \Phi(E) dE}{\int_{E=0}^{E^{\text{max}}} \Phi(E) \xi(E)^{-1} dE}, \quad (1.25)$$

where the units of  $\hat{\Phi}_{\text{spectrum}}$  are in photons/mm<sup>2</sup> per mR. The photon fluence,  $\Phi$  (in photons/mm<sup>2</sup>), at a specific exposure level ( $X$ ) is then just:

$$\Phi = X \hat{\Phi}_{\text{spectrum}}. \quad (1.26)$$

- The values of spectrum for a variety of spectra were calculated using Eq. (1.25), and these values are incorporated into Table 1.5. Armed with the kVp, the HVL, and the exposure, the x-ray fluence for a wide range of spectra can be estimated.
- **Note: Energy\_effective < Energy\_average**

**Table 1.5:** Beam-quality parameters (HVL in mm Al,  $E_{ave}$  and  $E_{eff}$  in keV) and photons/cm<sup>2</sup> per R for beams of different kV and added filtration ( $F = \text{mm Al}$ )

kV- $F$	HVL	$E_{ave}$	$E_{eff}$	$\Phi/R$	kV- $F$	HVL	$E_{ave}$	$E_{eff}$	$\Phi/R$
30-0	0.80	22.3	19.9	6.141e + 7	80-0	2.29	41.3	27.0	1.553e + 8
30-1	0.97	23.3	21.5	6.923e + 7	80-1	2.95	43.4	30.3	1.770e + 8
30-2	1.09	24.0	22.5	7.483e + 7	80-2	3.47	44.9	32.8	1.929e + 8
30-3	1.19	24.6	23.3	7.920e + 7	80-3	3.91	46.2	34.7	2.054e + 8
30-4	1.27	25.0	23.9	8.280e + 7	80-4	4.30	47.3	36.5	2.156e + 8
30-5	1.34	25.4	24.4	8.588e + 7	80-5	4.63	48.2	37.9	2.242e + 8
40-0	1.12	26.6	21.9	8.371e + 7	85-0	2.44	42.9	27.4	1.616e + 8
40-1	1.41	28.1	24.1	9.659e + 7	85-1	3.13	45.0	30.9	1.836e + 8
40-2	1.63	29.1	25.6	1.062e + 8	85-2	3.69	46.6	33.5	1.996e + 8
40-3	1.82	29.9	26.8	1.138e + 8	85-3	4.16	47.9	35.5	2.120e + 8
40-4	1.98	30.6	27.7	1.202e + 8	85-4	4.56	49.0	37.2	2.222e + 8
40-5	2.12	31.1	28.6	1.256e + 8	85-5	4.91	49.9	38.8	2.306e + 8
50-0	1.44	30.8	23.7	1.057e + 8	90-0	2.58	44.4	27.8	1.673e + 8
50-1	1.84	32.4	26.3	1.222e + 8	90-1	3.31	46.5	31.4	1.894e + 8
50-2	2.15	33.6	28.0	1.346e + 8	90-2	3.90	48.2	34.1	2.053e + 8
50-3	2.41	34.6	29.5	1.444e + 8	90-3	4.39	49.5	36.2	2.177e + 8
50-4	2.63	35.3	30.7	1.526e + 8	90-4	4.81	50.6	38.0	2.276e + 8
50-5	2.83	36.0	31.6	1.596e + 8	90-5	5.18	51.6	39.6	2.359e + 8
55-0	1.59	32.7	24.4	1.158e + 8	100-0	2.86	47.2	28.7	1.770e + 8
55-1	2.04	34.5	27.0	1.337e + 8	100-1	3.67	49.4	32.4	1.990e + 8
55-2	2.39	35.7	29.0	1.471e + 8	100-2	4.31	51.0	35.1	2.145e + 8
55-3	2.68	36.8	30.6	1.579e + 8	100-3	4.84	52.3	37.4	2.264e + 8
55-4	2.94	37.6	31.8	1.668e + 8	100-4	5.30	53.5	39.5	2.358e + 8
55-5	3.16	38.3	33.0	1.744e + 8	100-5	5.69	54.5	41.3	2.435e + 8
60-0	1.74	34.5	24.9	1.250e + 8	110-0	3.15	49.7	29.4	1.848e + 8
60-1	2.22	36.3	27.8	1.440e + 8	110-1	4.02	51.9	33.4	2.063e + 8
60-2	2.61	37.7	29.8	1.582e + 8	110-2	4.71	53.6	36.3	2.212e + 8
60-3	2.94	38.8	31.5	1.696e + 8	110-3	5.27	54.9	38.7	2.323e + 8
60-4	3.22	39.7	32.9	1.791e + 8	110-4	5.75	56.1	40.9	2.411e + 8
60-5	3.47	40.4	34.1	1.871e + 8	110-5	6.16	57.1	42.8	2.482e + 8

## Attenuation and energy deposition

- Just as the total linear attenuation coefficient  $\mu$  is the sum of the linear attenuation coefficients of the individual interaction types, the total mass attenuation coefficient is the sum of its constituents as well:

$$\left(\frac{\mu}{\rho}\right) = \left(\frac{\tau}{\rho}\right) + \left(\frac{\sigma_{\text{I}}}{\rho}\right) + \left(\frac{\sigma}{\rho}\right) + \left(\frac{\pi}{\rho}\right)$$

- Attenuation is useful in describing the propagation of x rays through a material, but it does not tell the complete story in terms of energy deposition. Energy deposition is important both in the calculation of the radiation dose to a patient, and for the calculation of the total signal generated in an x-ray detector

## Mass energy *transfer* coefficient

- The mass energy *transfer* coefficient is that fraction of the mass attenuation coefficient which contributes to the production of kinetic energy in charged particles. Photons which escape the interaction site do not contribute to the kinetic energy of charged particles.
- For the photoelectric effect the total energy of the incident x-ray photon,  $E_0$ , is transferred to the photoelectron. Part of this energy is used to overcome the binding energy of the atom ( $E_{BE}$ ), and the remaining fraction becomes the kinetic energy  $T$  of the photoelectron:

$$\frac{T}{E_0} = \frac{E_0 - E_{BE}}{E_0}$$

- The ionized atom will either emit one or more characteristic x rays (also called *fluorescent* x rays), which will leave the interaction site, or alternatively a series of nonradiative transitions involving Auger electrons will take place, resulting in the complete local deposition of energy through charged particles. Since the fluorescent yield of the atoms which comprise tissue is negligible, for tissue  $(\tau/\rho) = (\tau_{tr}/\rho)$ . This is not the case for x-ray detector materials.
- For Compton scattering, a large fraction of the incident x-ray energy leaves the site of the interaction in the form of the scattered photon. If the average kinetic energy imparted to electrons during Compton scattering is  $E_k$ , then the Compton mass energy transfer coefficient is given by:

$$\frac{\sigma_{tr}}{\rho} = \frac{\sigma}{\rho} \left[ \frac{E_k}{E_0} \right]$$

## Mass energy absorption coefficient

The mass energy transfer coefficient discussed above describes the fraction of the mass attenuation coefficient that gives rise to the initial kinetic energy of electrons in a small volume of absorber. The mass energy absorption coefficient will be the same as the mass energy transfer coefficient when all transferred energy is locally absorbed. However, energetic electrons may subsequently produce bremsstrahlung radiation (x-rays), which can escape the small volume of interest. Thus, the mass energy absorption coefficient may be slightly smaller than the mass energy transfer coefficient. For the energies used in diagnostic radiology and for low-Z absorbers (air, water, tissue), the amount of radiative losses (bremsstrahlung) is very small. Thus, for diagnostic radiology,

$$\left(\frac{\mu_{en}}{\rho}\right) \cong \left(\frac{\mu_{tr}}{\rho}\right)$$

If  $g$  is the fraction of energy lost in radiative process:  $\left(\frac{\mu_{en}}{\rho}\right) = \left(\frac{\mu_{tr}}{\rho}\right)(1 - g)$

$$\text{Kerma} = \phi E \left(\frac{\mu_{tr}}{\rho}\right)$$

$$\text{Absorbed dose } D = \frac{dE_{abs}}{dm} = \text{Kerma} (1-g) = \phi E \left(\frac{\mu_{ab}}{\rho}\right)$$



# Energy absorption and biological damage

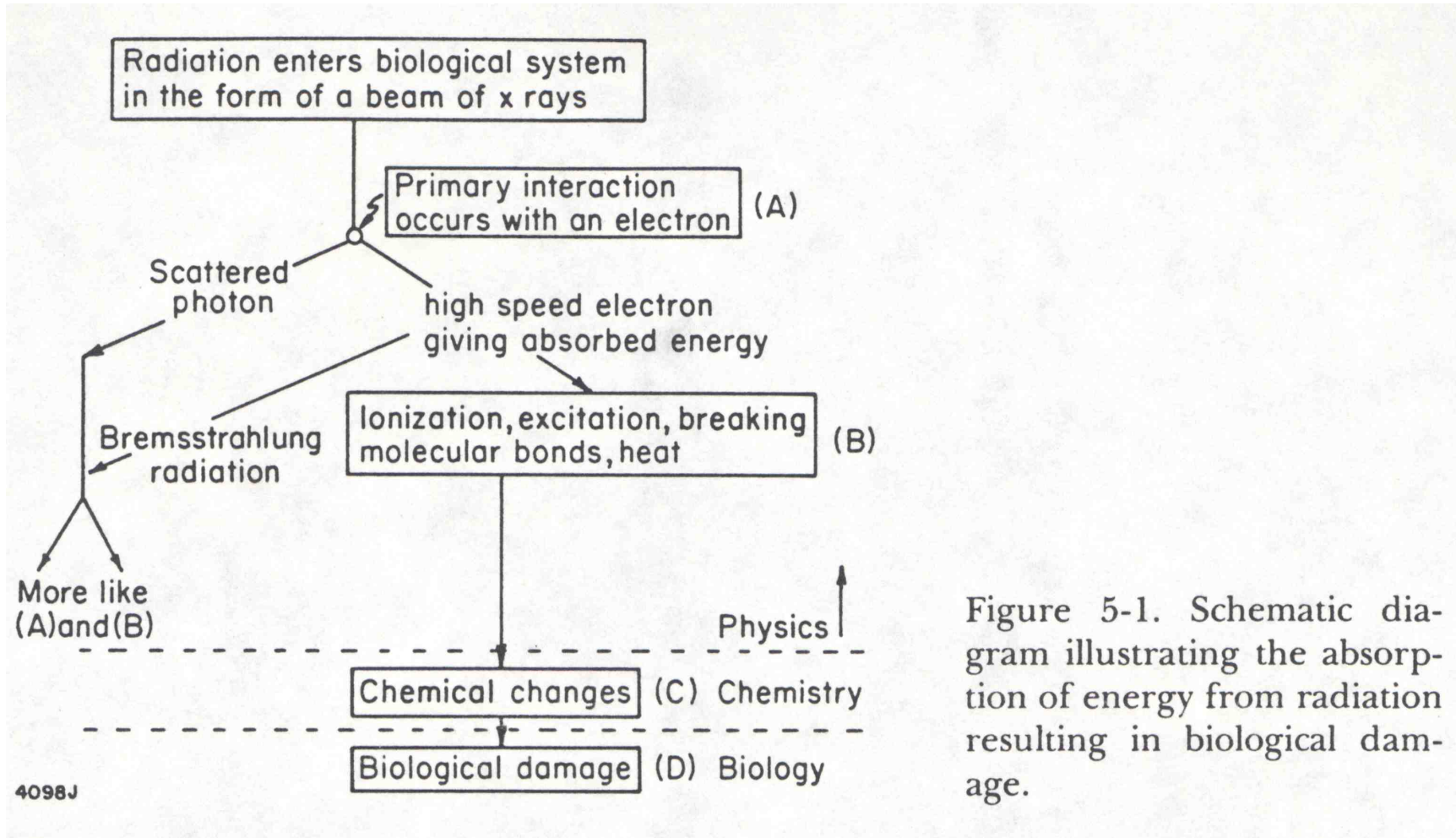


Figure 5-1. Schematic diagram illustrating the absorption of energy from radiation resulting in biological damage.

## **X-ray spectra simulation:**

<https://health.siemens.com/booneweb/index.html>

## **Database NIST – attenuation and absorption coefficients**

<https://www.nist.gov/pml/x-ray-form-factor-attenuation-and-scattering-tables>

<https://physics.nist.gov/PhysRefData/FFast/html/form.html>

<https://www.nist.gov/pml/xcom-photon-cross-sections-database>  
database search form  
(<https://physics.nist.gov/PhysRefData/Xcom/html/xcom1.html>)

## **XMuDat software (not for Mac):**

<https://www-nds.iaea.org/publications/iaea-nds/iaeands-0195.htm>