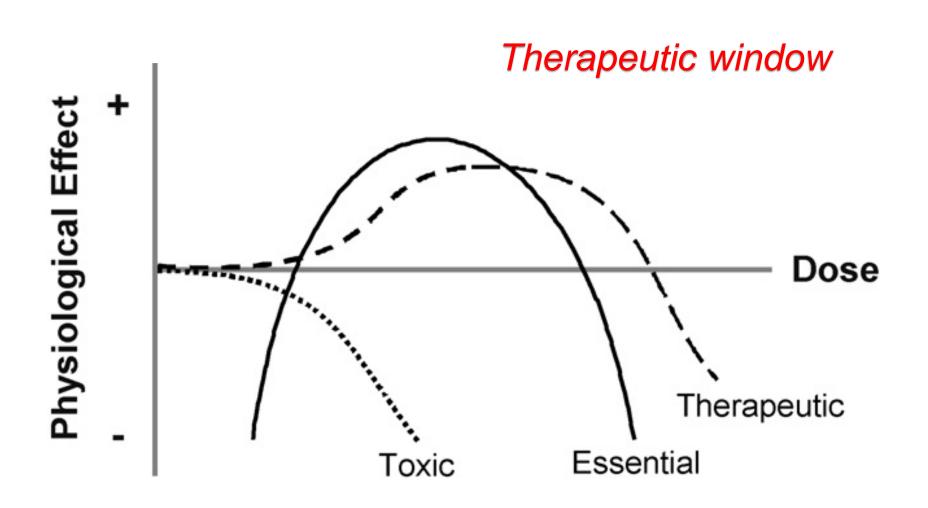


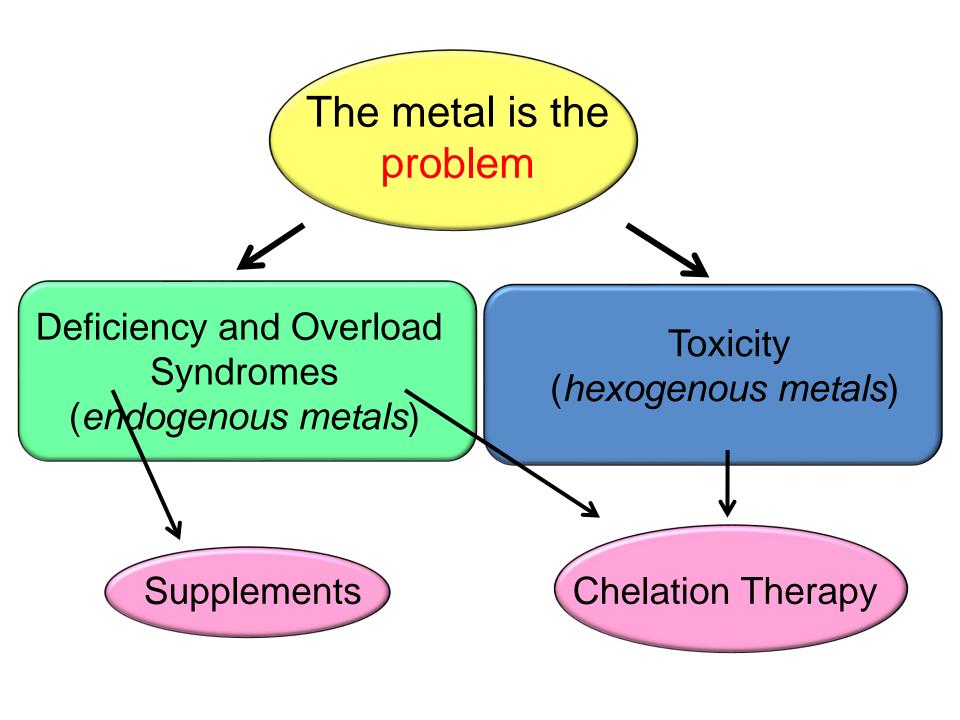
Diagramma di Bertrand

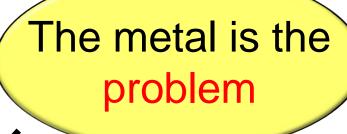


Medicinal Inorganic Chemistry

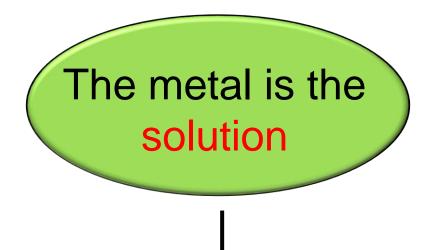
The metal is the problem

The metal is the solution



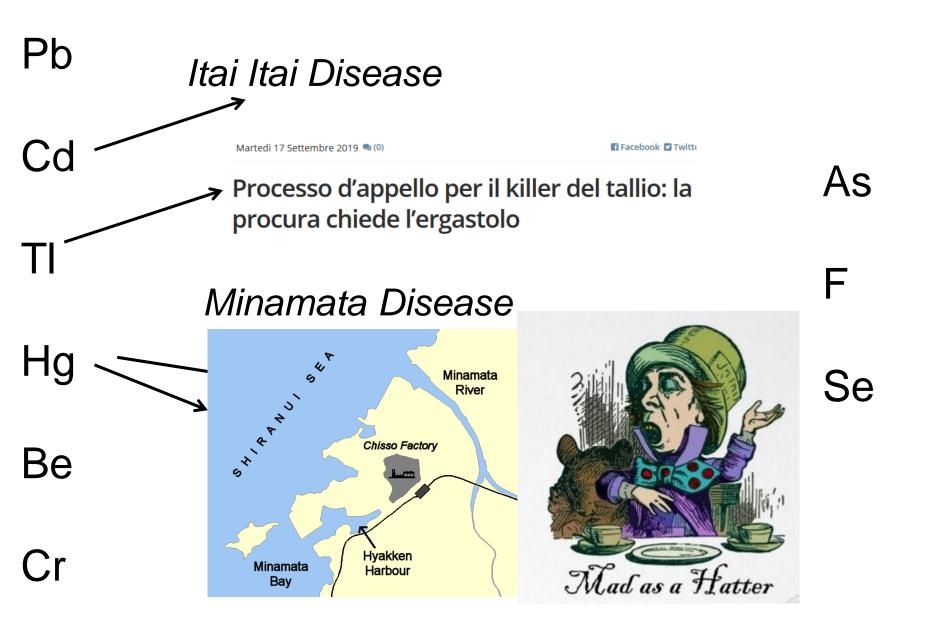


Inhibitors or Analogs of Metalloenzymes

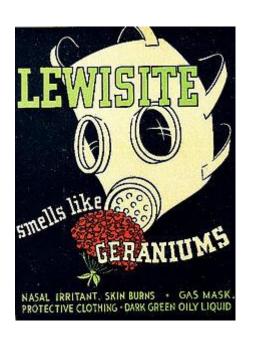


Diagnostic and Therapeutic Agents

Tossicità di metalli esogeni e altri elementi



Chelation Therapy





BAL = British Anti-Lewisite

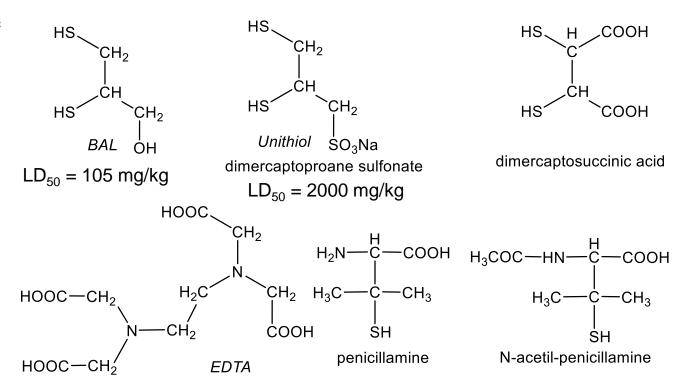
Chelation Therapy

Agent:

- Effective (i.e. match the binding preferences of the ion)
- Selective
- Non toxic
- Resistant to metabolism
- Unexpensive

Adducts:

- Stable
- Non toxic
- Highly soluble in water (rapid clearance)
- Resistant to metabolism



Iron chelation therapy

- Mammals are unable to regulate the export of Fe
- Patients affected by severe forms of anemia (e.g. thalassemia) need frequent blood transfusions
- Transfusions lead to iron overload
- Iron overload, if untreated, leads to premature death Fenton chemistry: Fe²⁺ + H₂O₂ → Fe³⁺ + OH⁻ + OH⁻
- Chelation therapy is essential
 - 1. Efficacia del chelante
 - 2. Tossicità
 - 3. Costo
 - 4. Modo di somministrazione (compliance)

Natural siderophore from Streptomyces pilosus

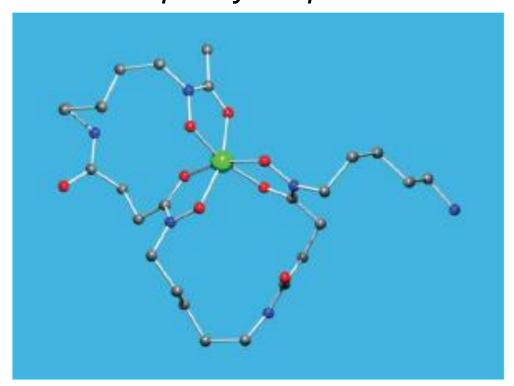
FDA approval: 1968

pFe = 26.6

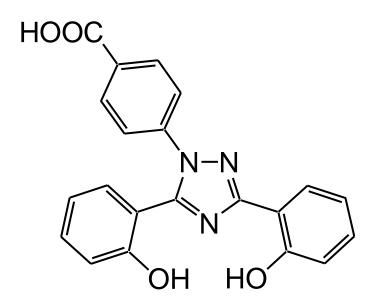
 $pFe = -log[Fe^{n+}]$

Drawback: very long

infusion time: 8 – 12 h



Deferasirox: Orally active



bis-hydroxyphenyl-triazole deferasirox

pFe = 20

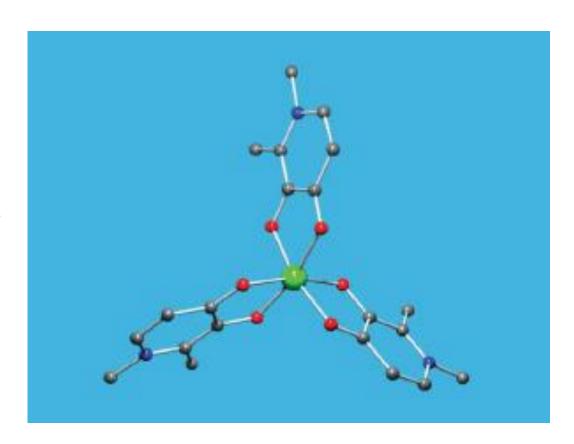
FDA approval: 2005

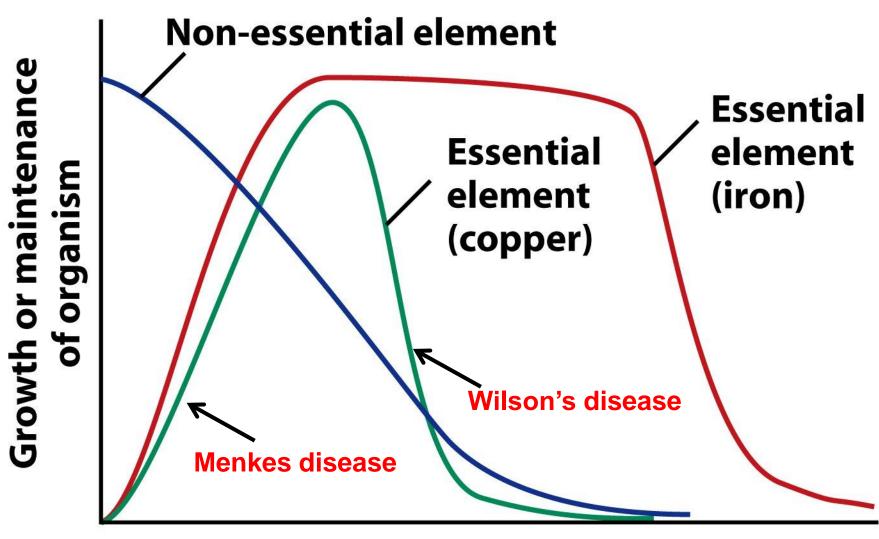
Deferiprone: Orally active

3,4-dihydroxypyridinone *deferiprone*

pFe = 20

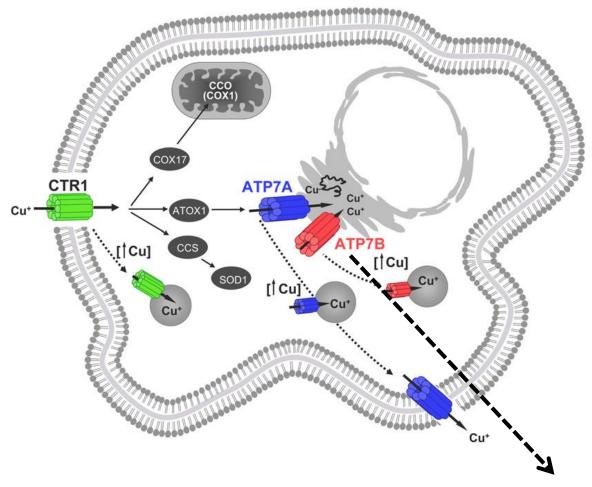
2011 FDA approval as second-line oral drug





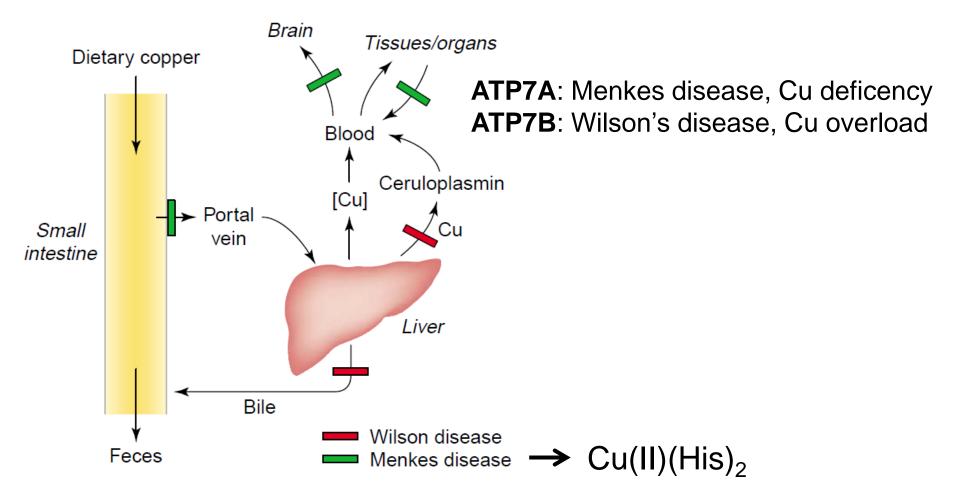
Concentration of element

Copper homeostasis

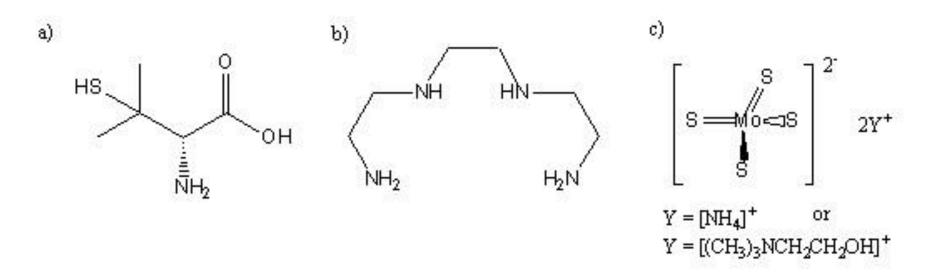


ATP7A: ubiquitous

ATP7B: liver, kidneys, brain



Chelanti per la Sindrome di Wilson (rimozione Cu)



D-penicillamina

Trien (o *Trientina*) (tris-etilenetetrammina)

Tetratiomolibdato

b

$$S = M_0 = S \qquad \left(\begin{array}{c} CH_3 \\ I_4 \\ I_5 \end{array} \right) - OH \right)_2$$

Table 2. Agents for the Treatment of Wilson Disease

| agent | mechanism of action | daily adult dosage |
|---|---|---------------------------------------|
| D-penicillamine ^a | reduction and chelation of copper; urinary excretion of copper by mobilizing copper from organs | 1–2 g orally in divided doses |
| triethylenetetramine (Trien) | copper chelator and urinary | 0.75-1.5 g orally in divided doses |
| zinc salts | inhibits intestinal absorption of copper by induction of intestinal cell metallothionein; may also induce hepatic metallothionein | 150—200 mg orally in divided doses |
| british anti-Lewisite (BAL) | copper chelator | 3 mL of 10% BAL in peanut oil im |
| ${\it tetrathiomolyb} {\it date}^b$ | blocking the intestinal absorption of copper and a copper chelator | Up to 2 mg/kg orally in divided doses |
| ^a Administered with supplementation of | f 25 mg of pyridoxine orally daily. $^{\it b}$ Experimental | |