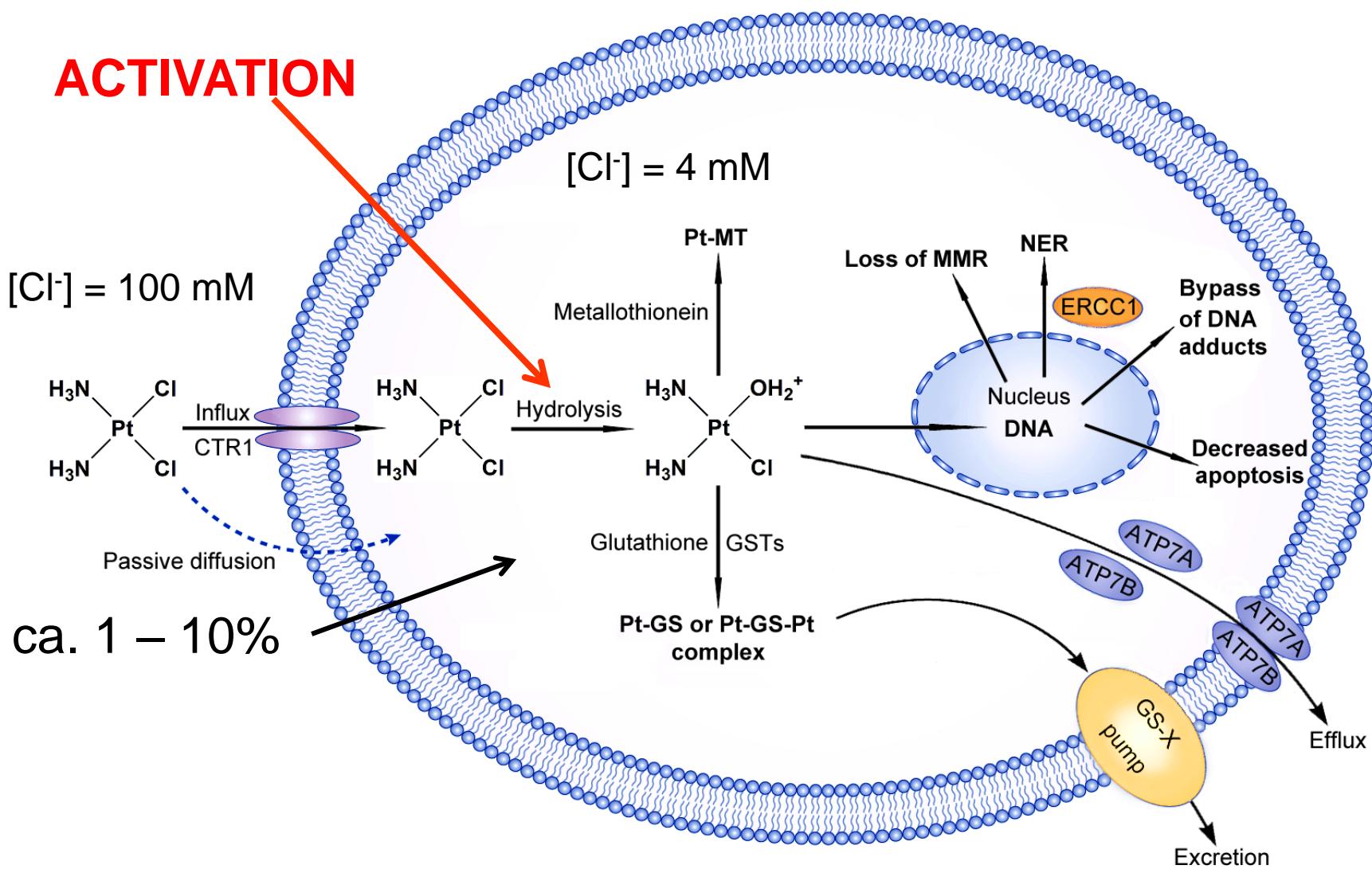
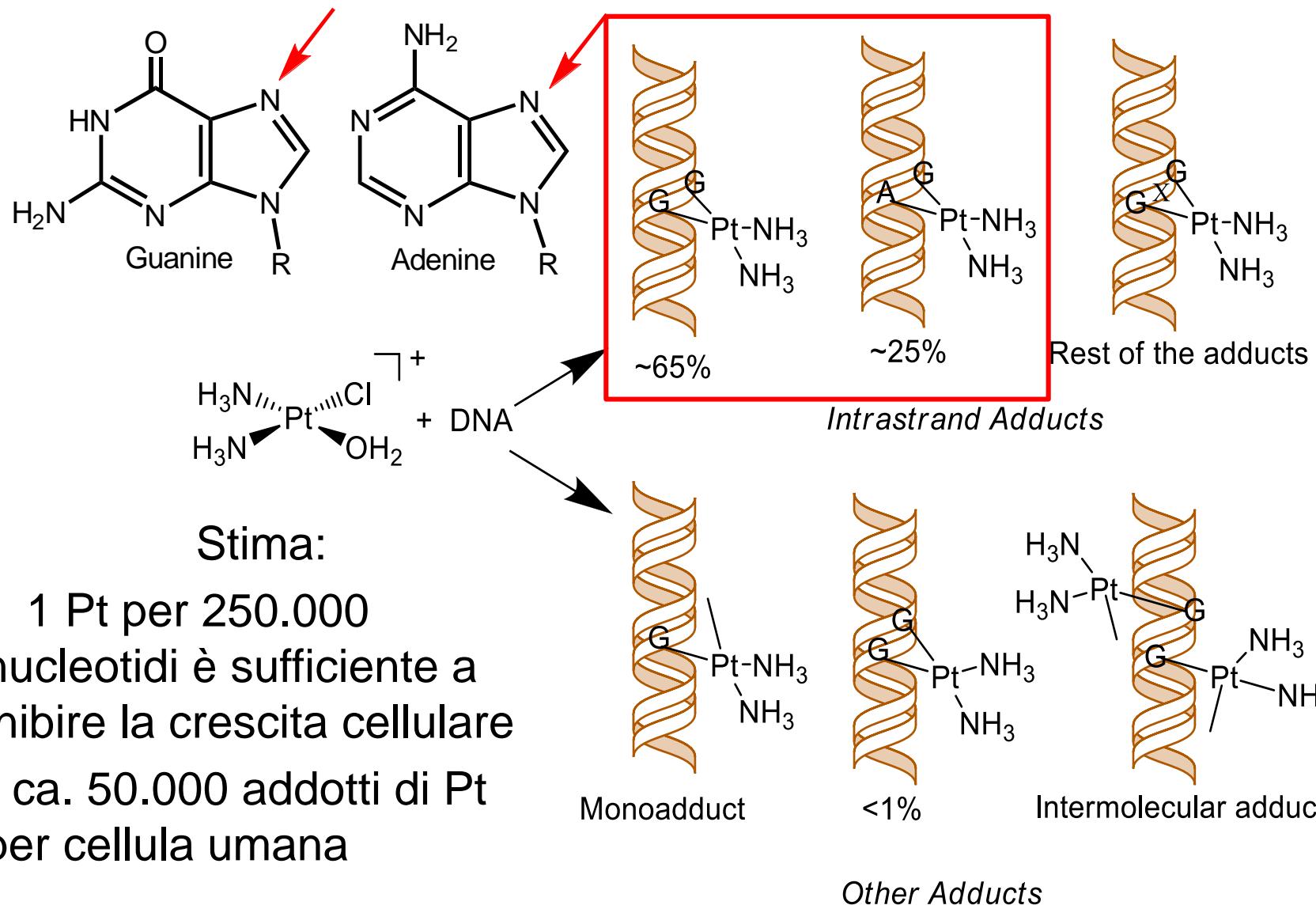


Discovery of cisplatin

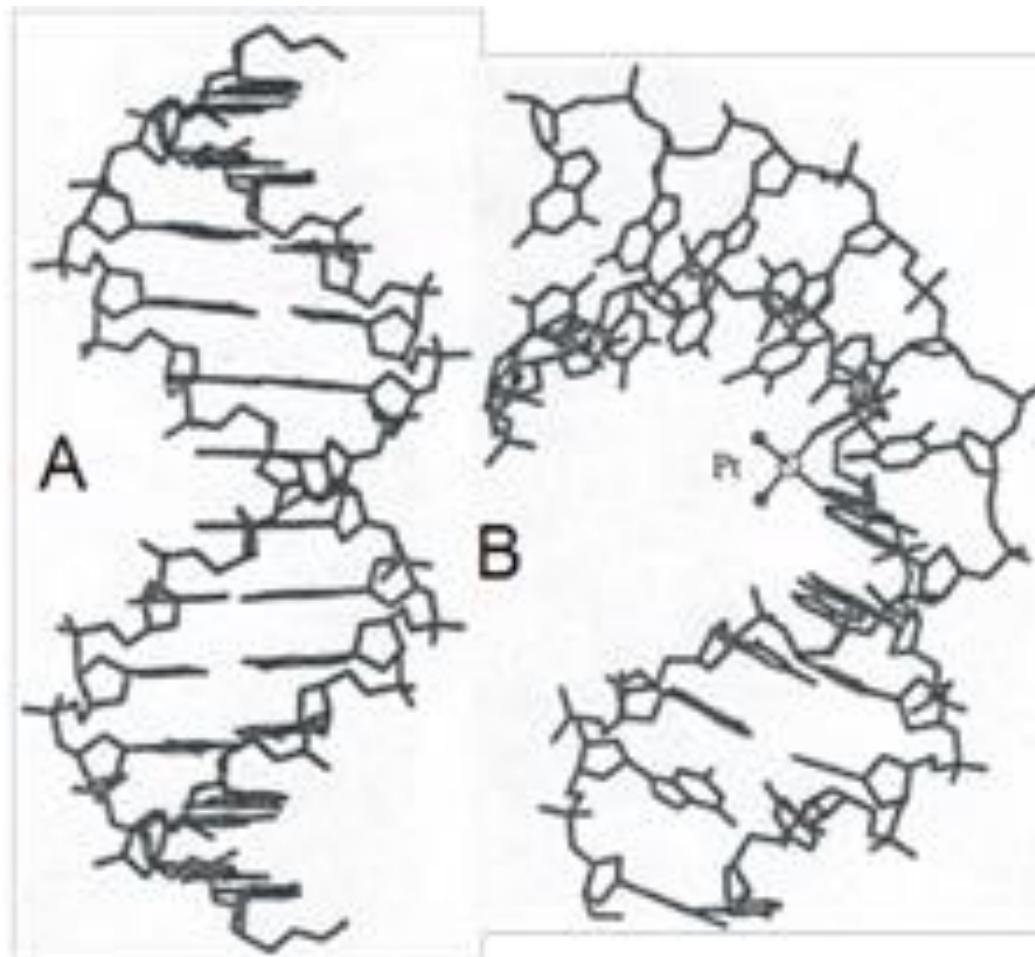
Mechanism of action

Structure – Activity
relationships

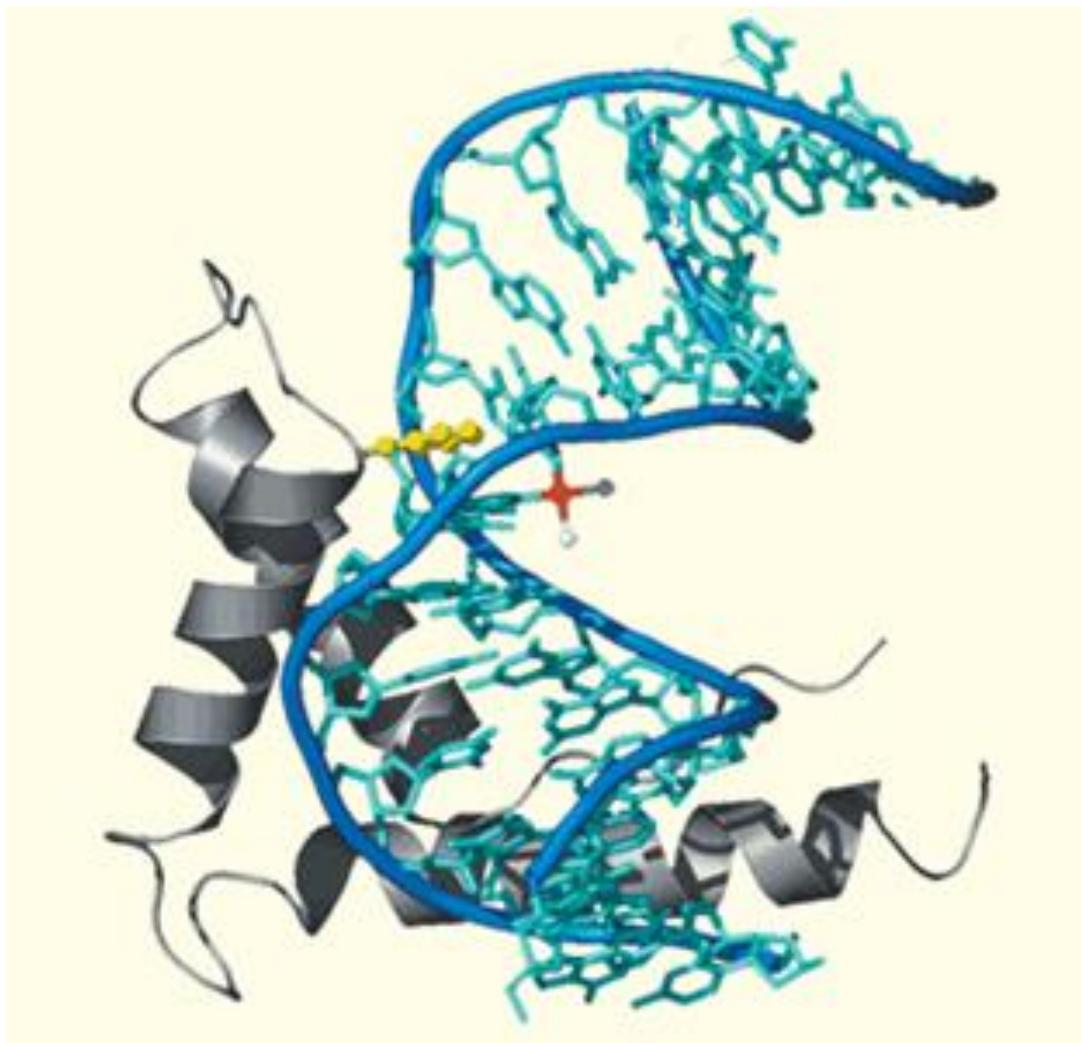




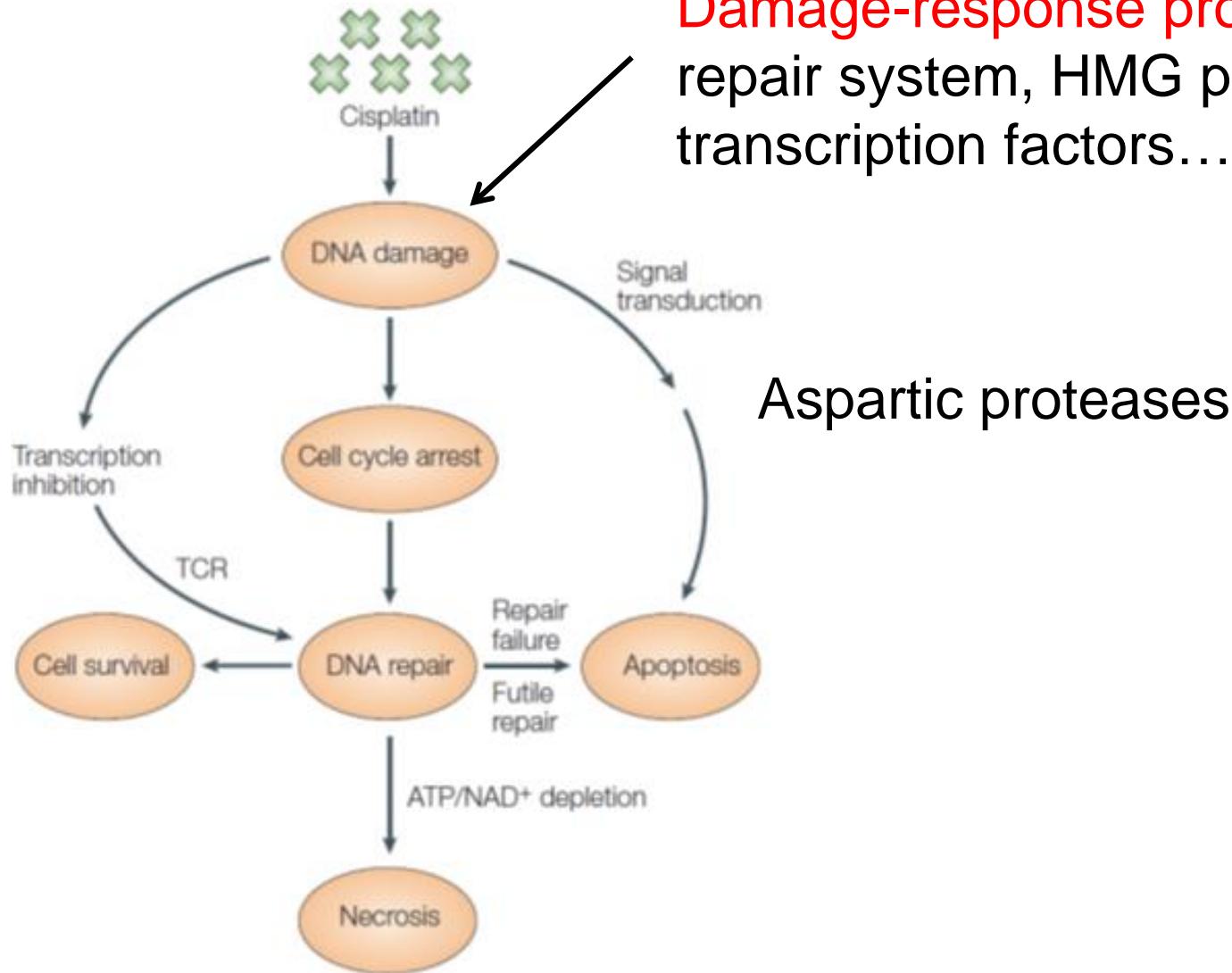
La platinazione induce un piegamento (*kink*) e uno svolgimento (*unwinding*) locale del DNA



Riconoscimento del sito di platinazione del DNA da parte di una proteina HMG

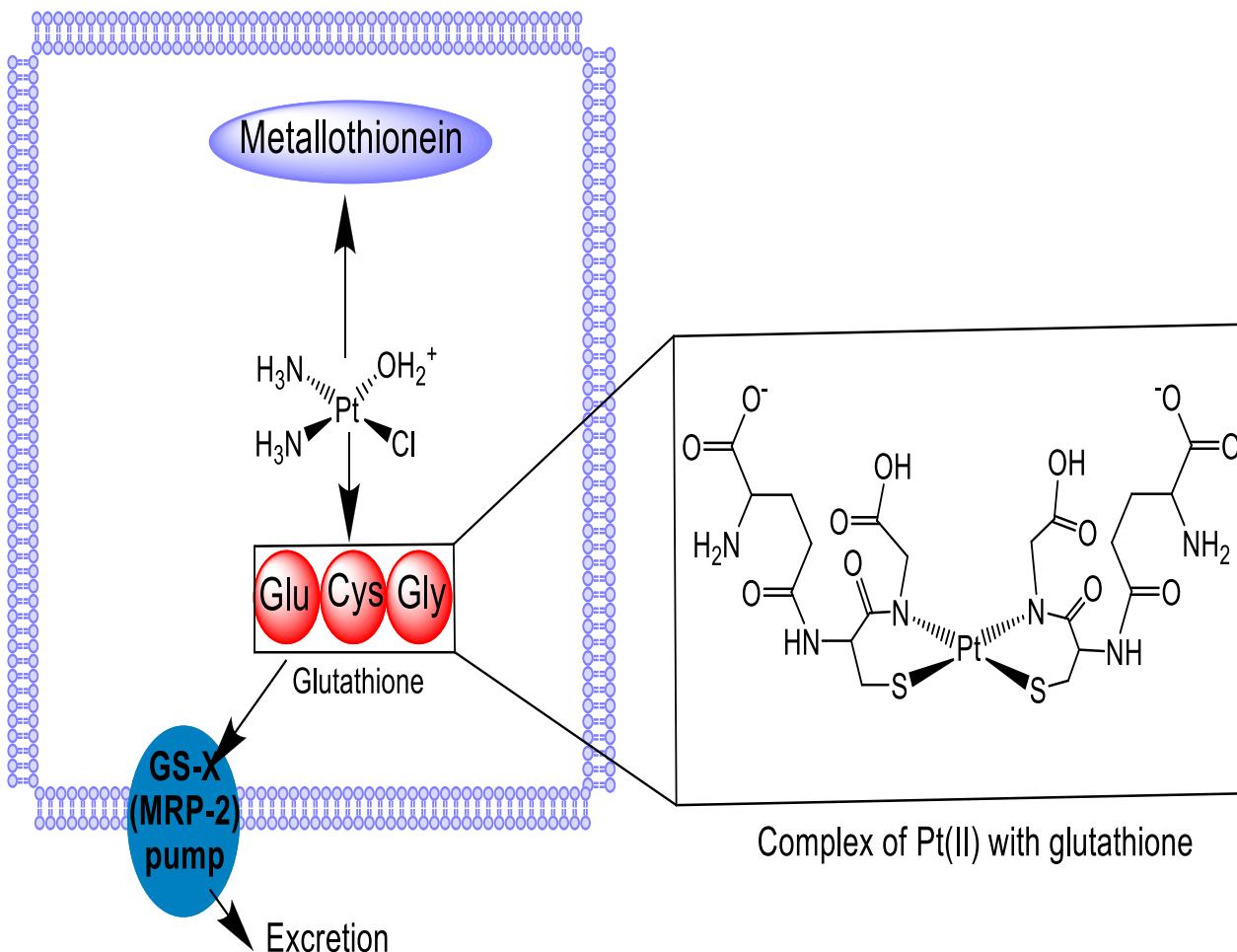


Damage-response proteins: DNA repair system, HMG proteins, transcription factors...

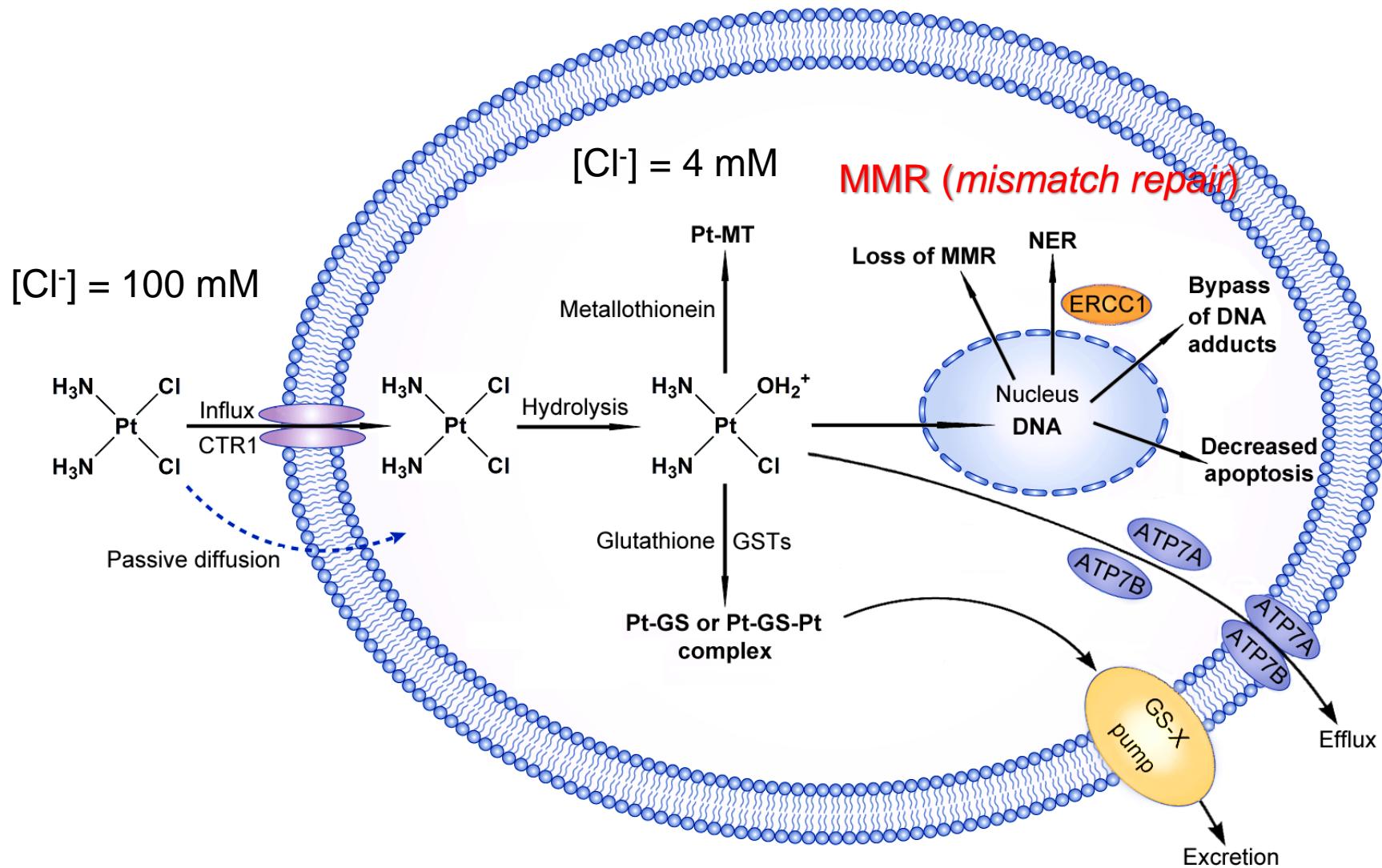


Principali meccanismi di resistenza

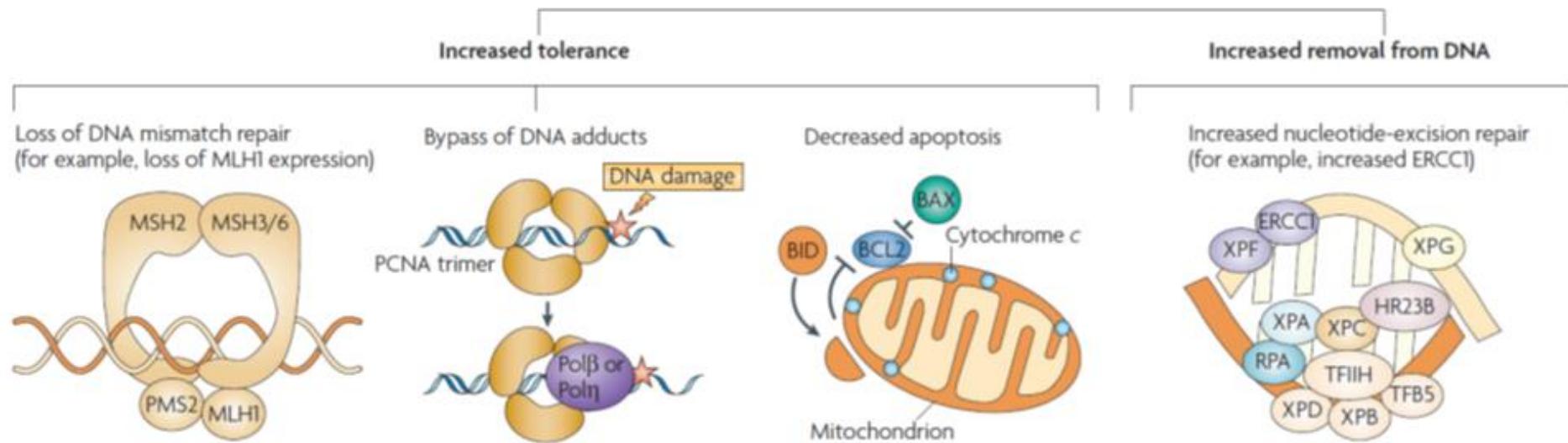
- 1) diminuzione del livello di platino nella cellula;
- 2) aumento del livello di tioli cellulari (glutazione, metallotioneine e altre molecole contenenti zolfo);
- 3) aumento della capacità di riparo del DNA e/o aumento della resistenza al danno;
- 4) cambiamenti nelle catene di segnali che portano alla morte cellulare (*cell-death pathways*), o alla sua sopravvivenza. In particolare, riduzione della risposta apoptotica e attivazione di *survival pathways*.



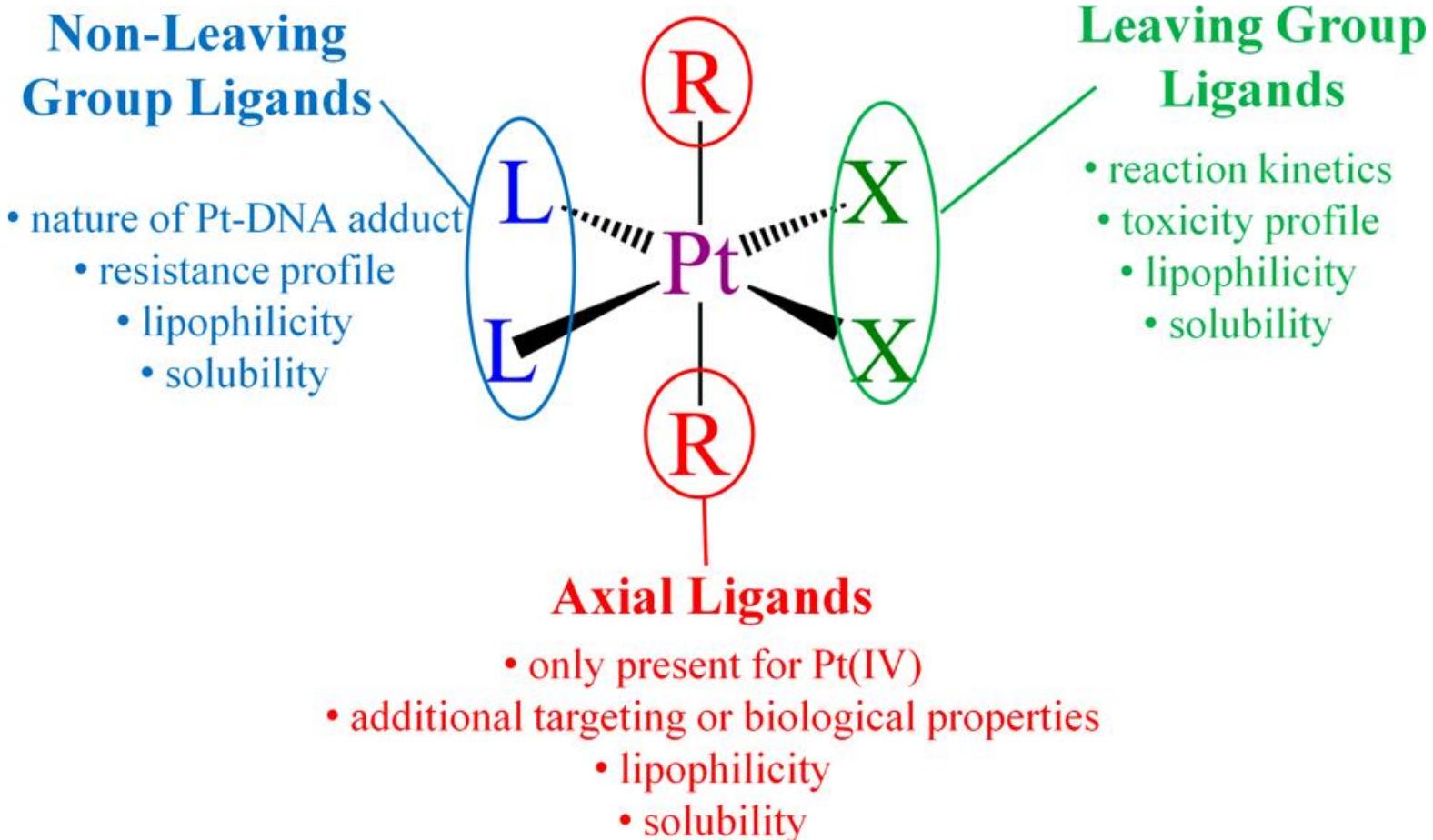
NER (Nucleotides Excision Repair)
ERCC1 (excision repair cross-complementing-1)



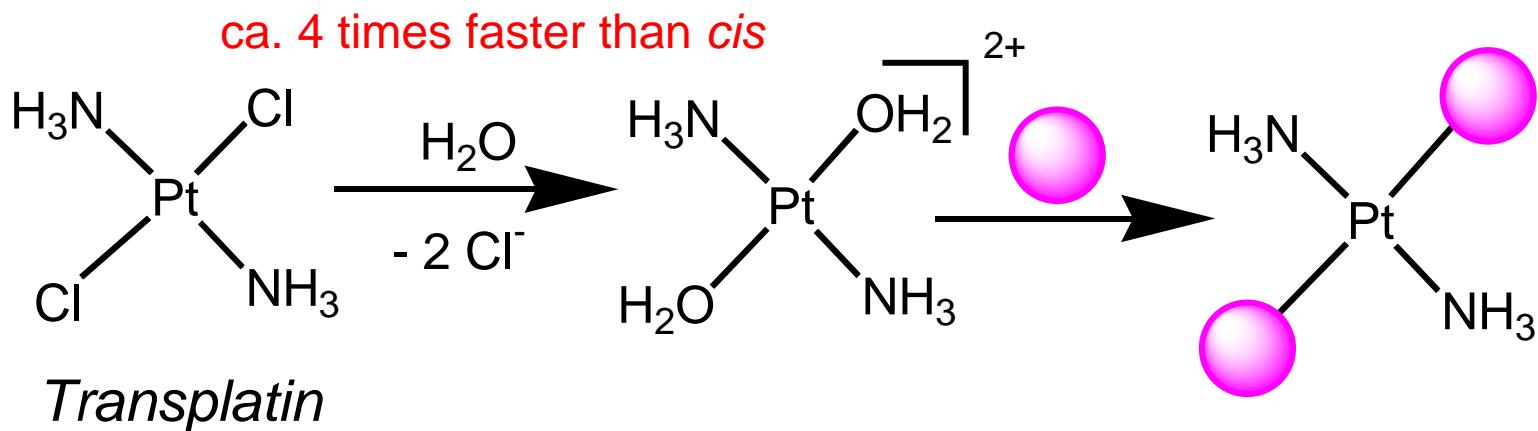
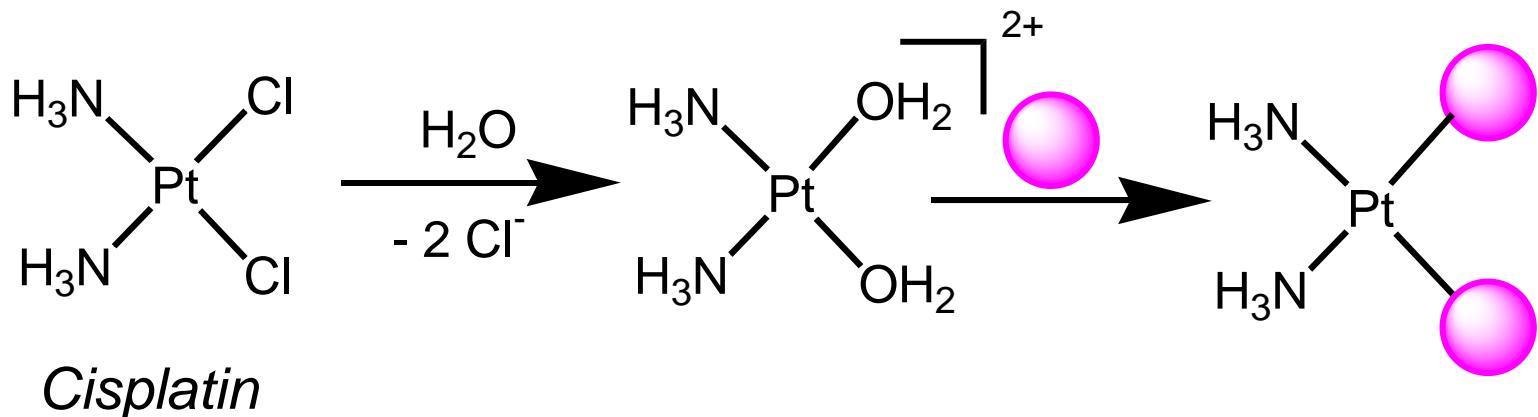
Meccanismi di resistenza



Correlazione struttura – attività

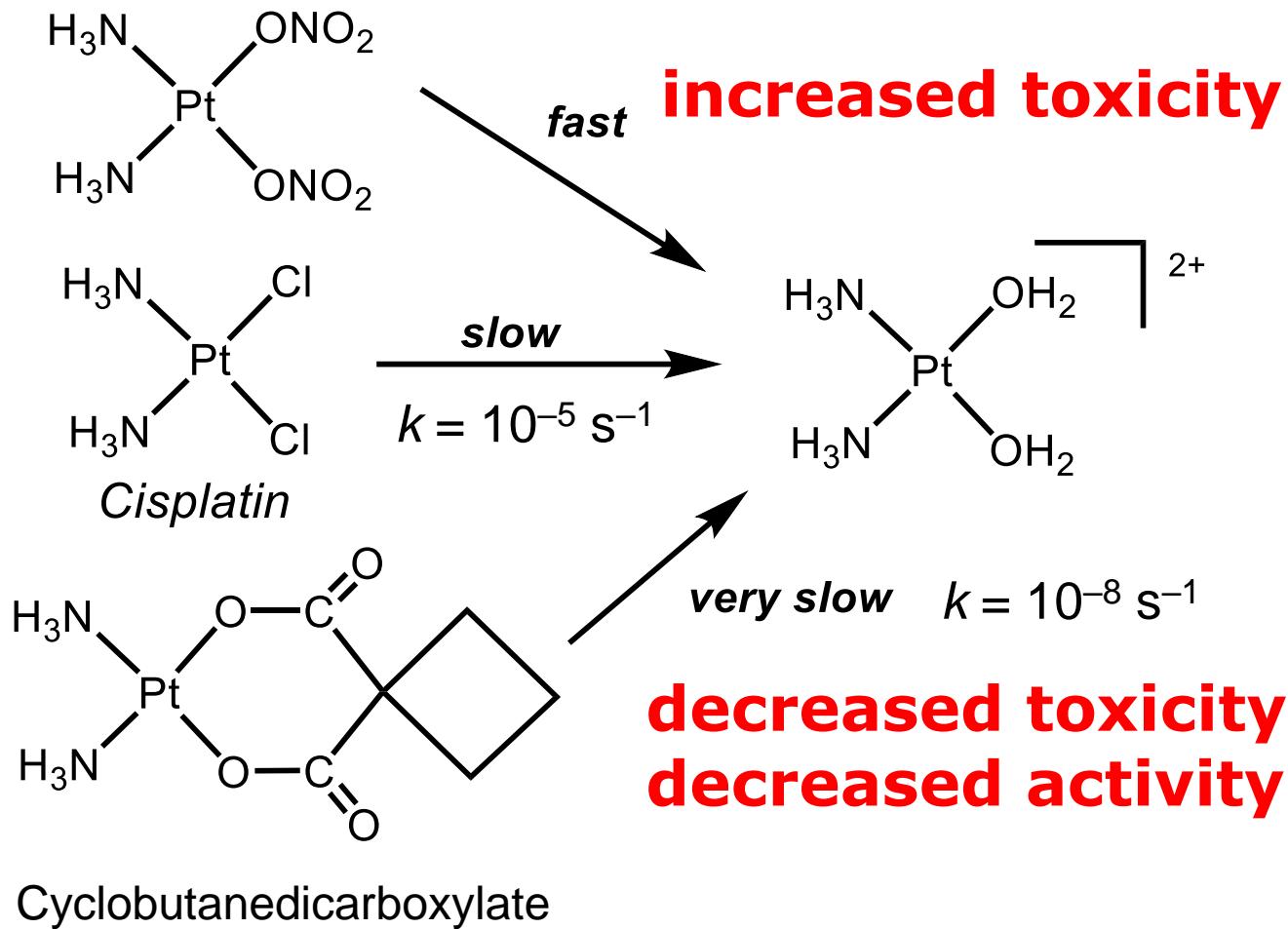


La geometria conta!

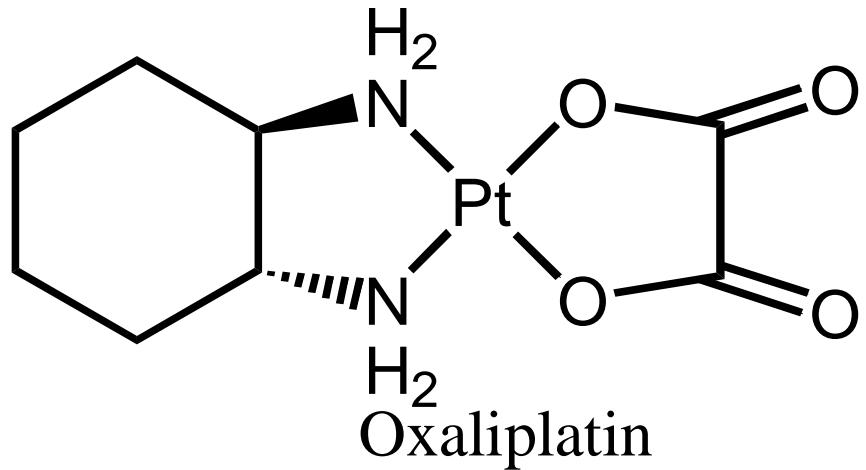
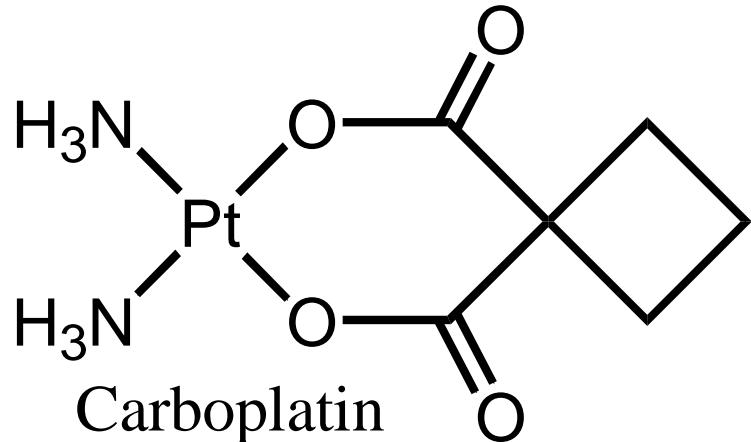


Toxic, but not anticancer active

La cinetica conta!



2nd and 3rd generation Pt(II) drugs: Carboplatin and Oxaliplatin

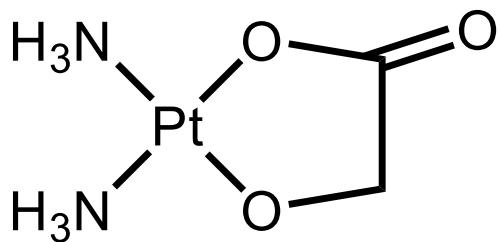


1,1-cyclobutanedicarboxylate
 $t_{1/2}$ aquation = 268h vs 2.4h of cisplatin

300–450 mg/m² vs 20–120 mg/m² of cisplatin

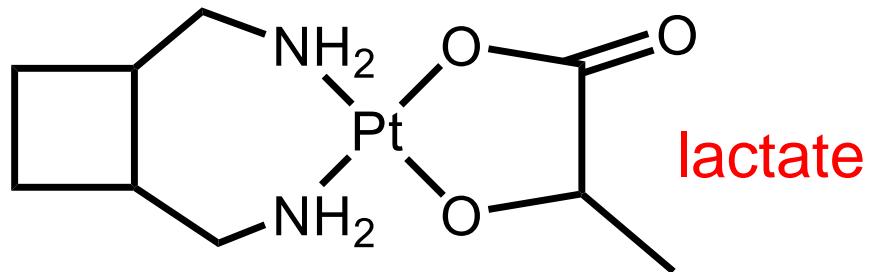
trans-(1*R*,2*R*)-1,2-diaminocyclohexane

2-hydroxyacetate



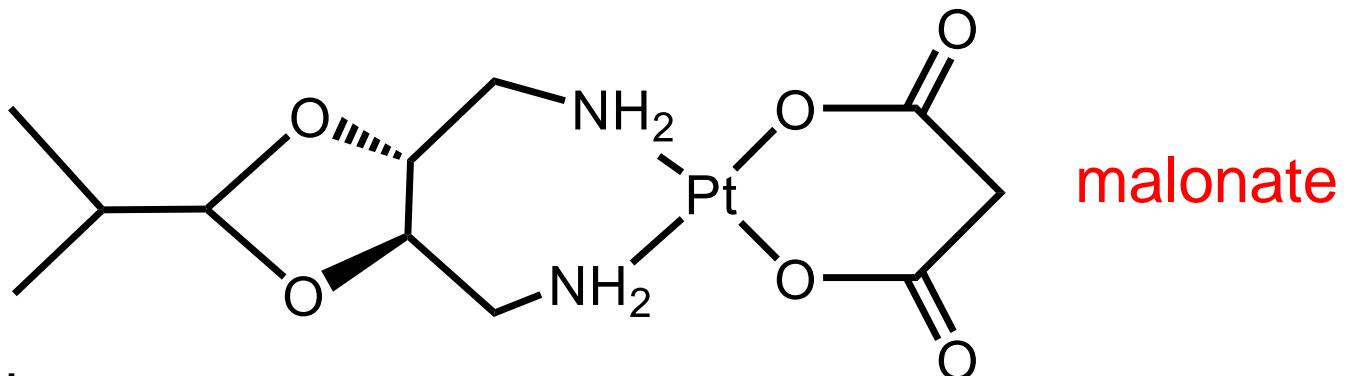
Nedaplatin (Japan)

II generation



Lobaplatin (China)

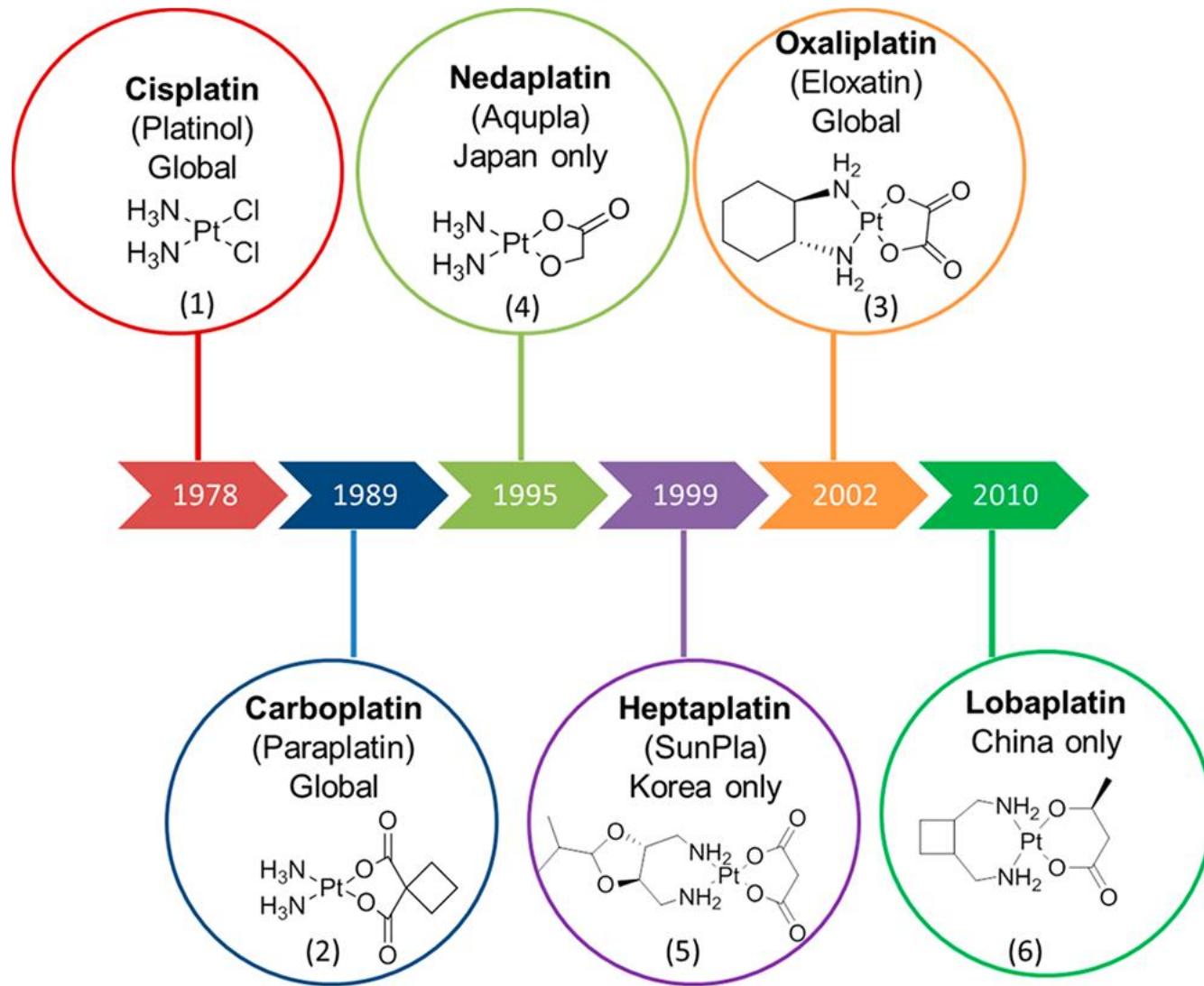
III generation



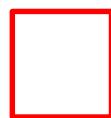
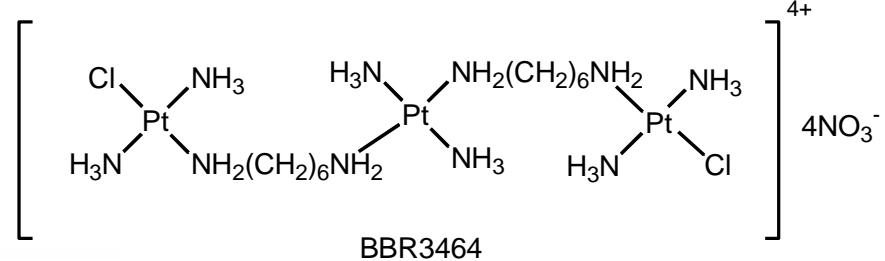
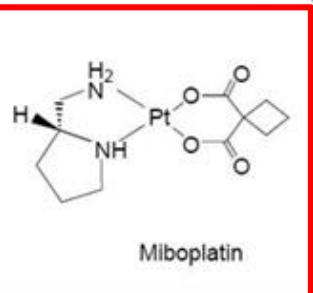
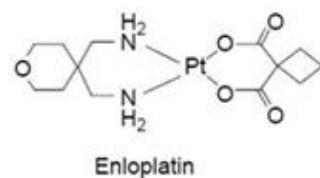
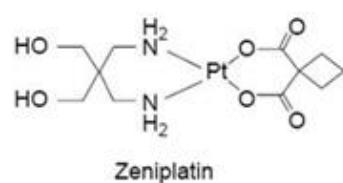
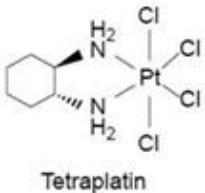
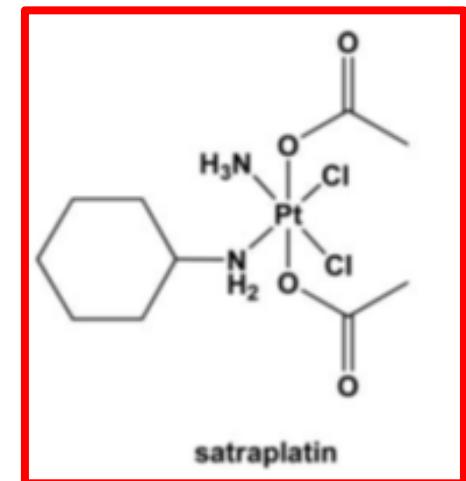
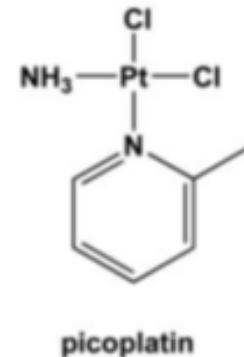
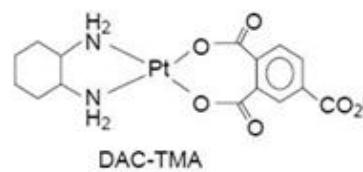
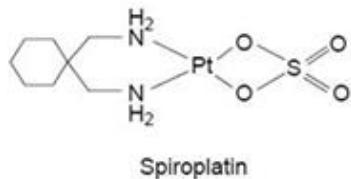
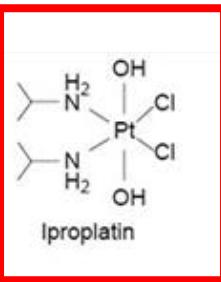
III generation

Heptaplatin (South Korea)

malonate

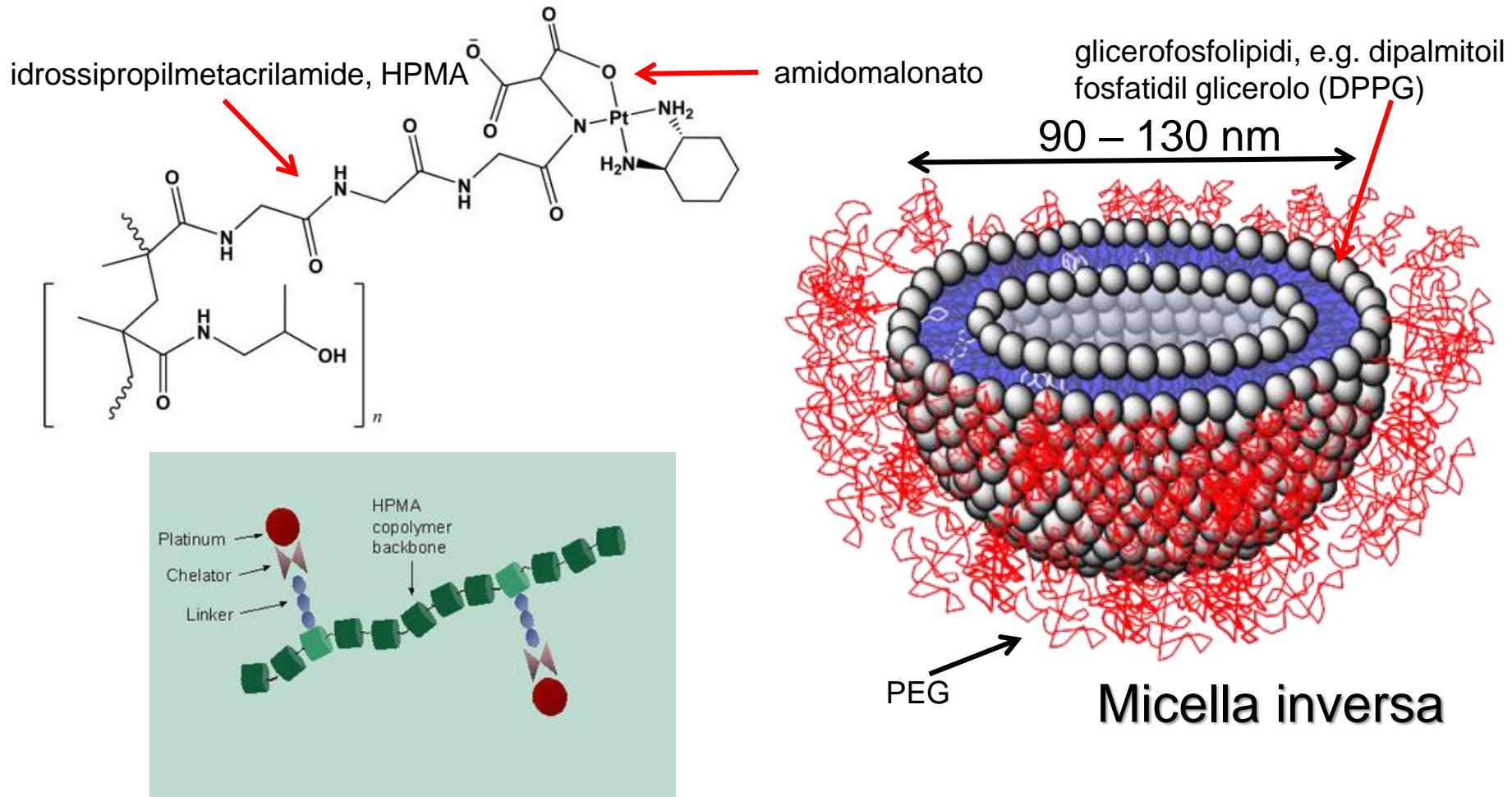


Alcuni dei 23 composti di Pt testati in fase clinica



= fase 3

Alcune formulazioni di Pt in fase clinica: i nano-carrier ProLindacTM e LipoplatinTM



- Pt drugs are actually **prodrugs** (or *functional compounds*) and need an activation step.
- For functional compounds activation occurs typically by hydrolysis, possibly preceded by reduction/oxidation.
- The coordination of the metal to the bio-target is the main interaction responsible for the activity. Additional, less energetic, interactions may be also important.

Functional Compounds

The anticancer activity (e.g. cytotoxicity) of functional compounds will depend on **many parameters**, very often strictly interconnected:

1. on the nature of the metal center (*thermodynamic and kinetic parameters, hard-soft nature, oxidation state*)
2. on the nature of the non-leaving ligands (*lipophylicity, charge, solubility, non-covalent interactions...*);
3. on the kinetics of activation (e.g. *aquation rate*)

Not surprisingly, the few metal anticancer drugs that are in clinical use – all of them functional – were found serendipitously or by rational design from a lead compound (cisplatin → carboplatin → oxaliplatin).

There are apparently two ways for overcoming the limitations of Pt anticancer drugs:

1. Find novel, non-conventional Pt drugs, i.e. Pt compounds that do not follow the established SAR rules;
2. Find new anticancer drugs based on different metals.

Non-platinum active compounds are likely to have thermodynamic and kinetic parameters different from those of Pt drugs and, as a consequence, also different

- mechanism of action
- biodistribution
- toxicity

Among the several metals that are currently being investigated for their anticancer activity, **ruthenium** (among others) occupies a prominent position.

Expectations

Ruthenium drugs are expected:

1. to be active against those human malignancies that are resistant, or have acquired resistance, to Pt drugs.

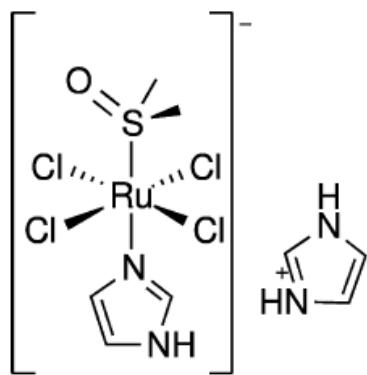
2. to show a lower (or at least different) toxicity compared to Pt drugs.

General features of ruthenium compounds

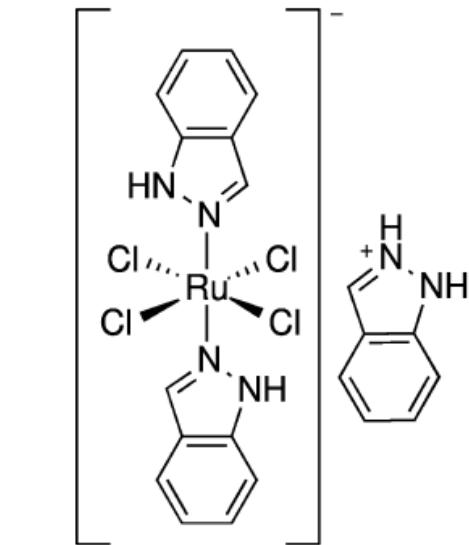
- Six-coordinate, octahedral geometry
- High affinity for nitrogen and sulfur ligands
- Two readily accessible oxidation states in aqueous solution: Ru(III) (d^5 , paramagnetic) and Ru(II) (d^6 , diamagnetic)
- Ru(III) complexes are usually more inert than the corresponding Ru(II) species
- The kinetics of ligand dissociation of Ru compounds are similar to those of Pt compounds (with remarkable exceptions)

Clinically tested anticancer Ru(III) compounds

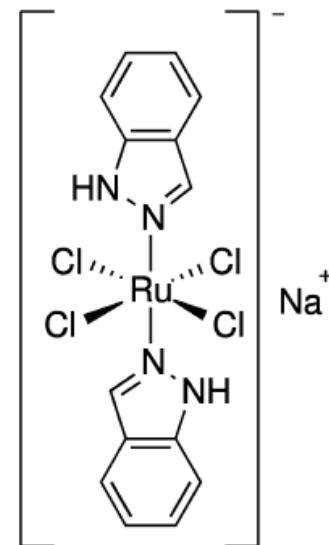
Deceptively similar



NAMI-A



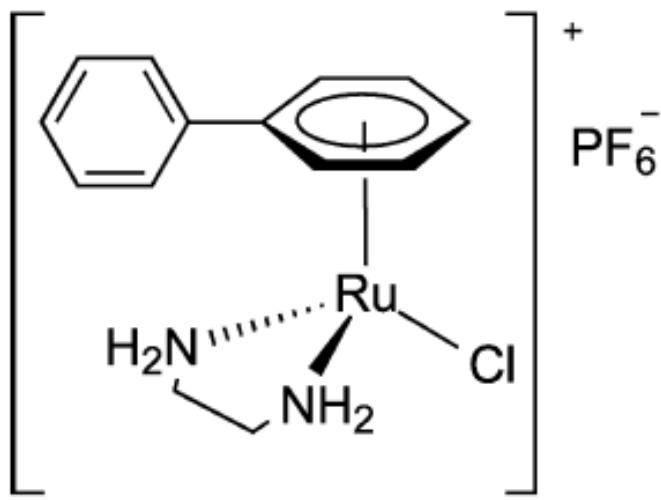
KP1019



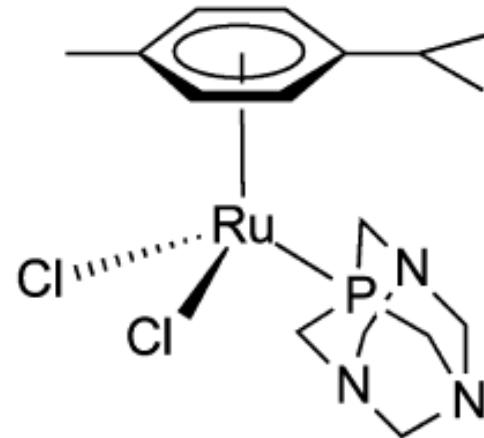
(N)KP1339

Both NAMI-A and KP1019 are **prodrugs**; they are activated through hydrolysis, possibly after reduction to Ru(II).

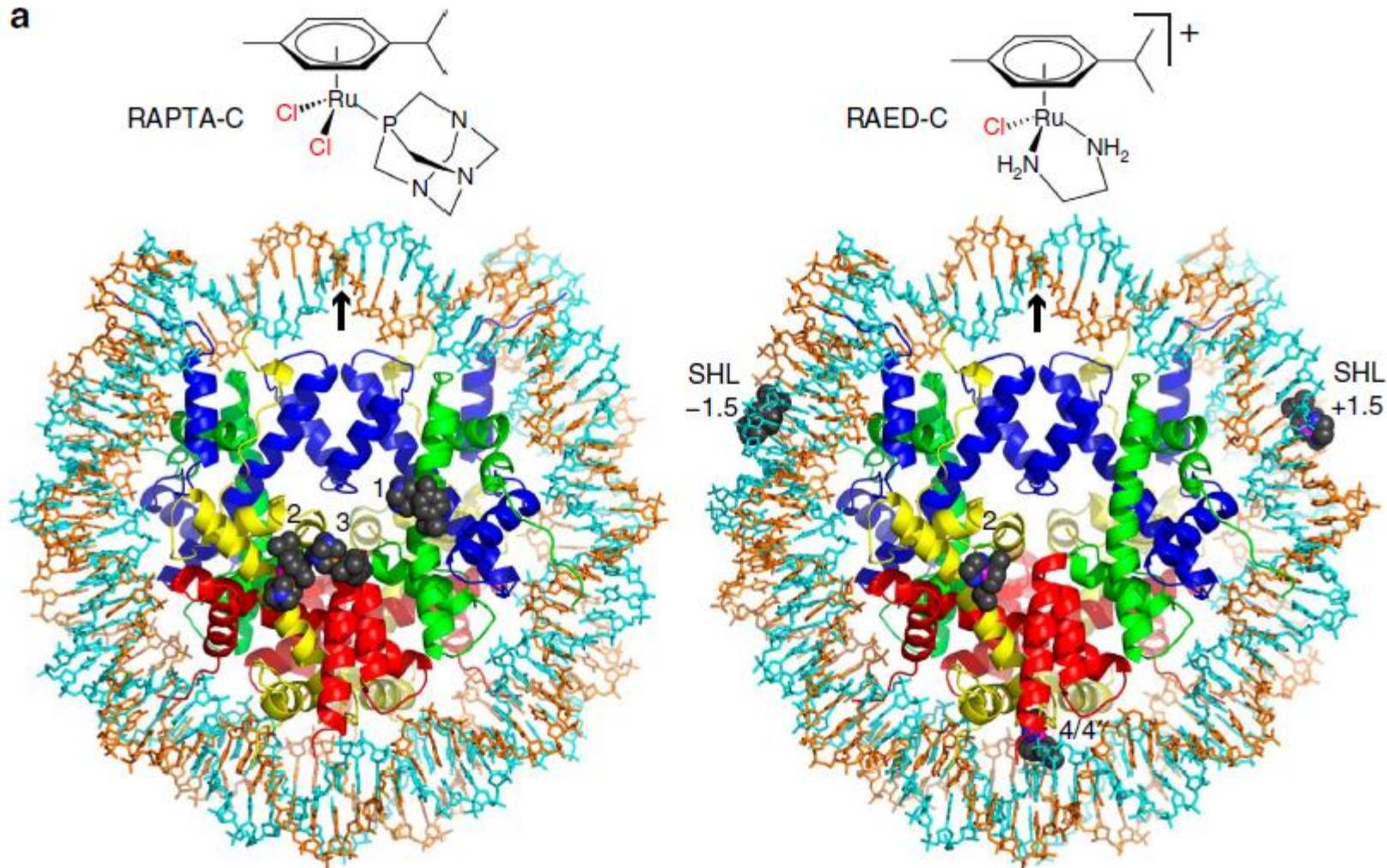
Anticancer organometallic Ru(II) compounds



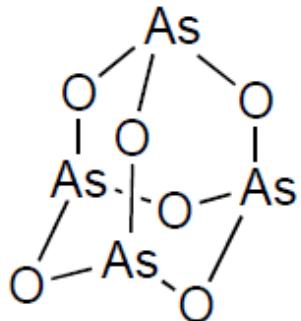
RM175



RAPTA-C

a

Composti antitumorali di arsenico



FDA 2000

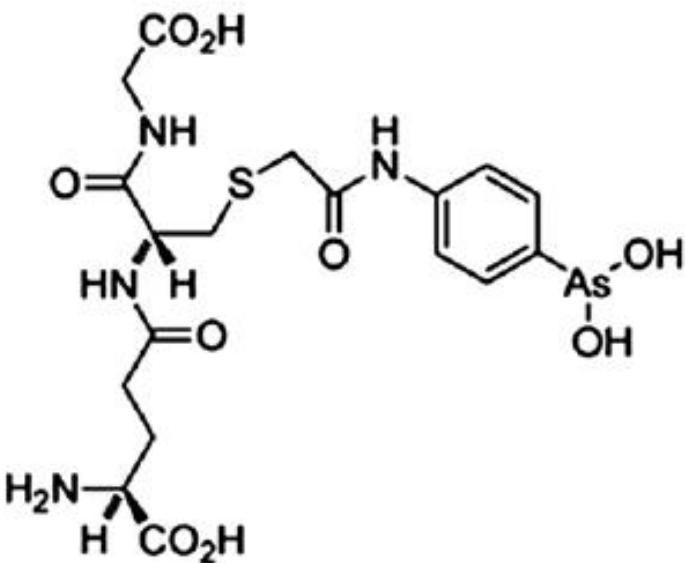
acute promyelocytic leukemia

ATO

0.15 mg/kg

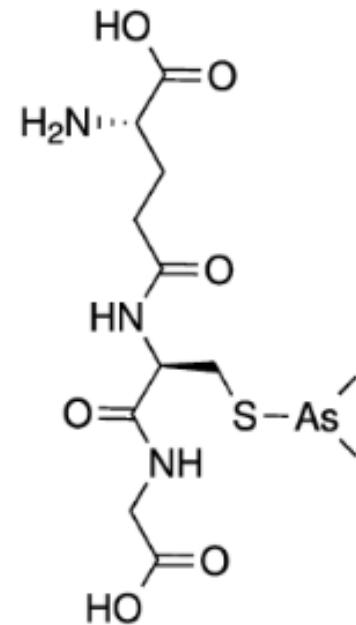
- Degradazione della proteina mutante PML-RAR α che blocca la differenziazione mieloide
- Inibizione di enzimi anti-ROS (glutazione reduttasi, glutatione perossidasi, tioredossina reduttasi e tioredossina perossidasi)

Potenziali composti antitumorali di organo-arsenico



GSAO

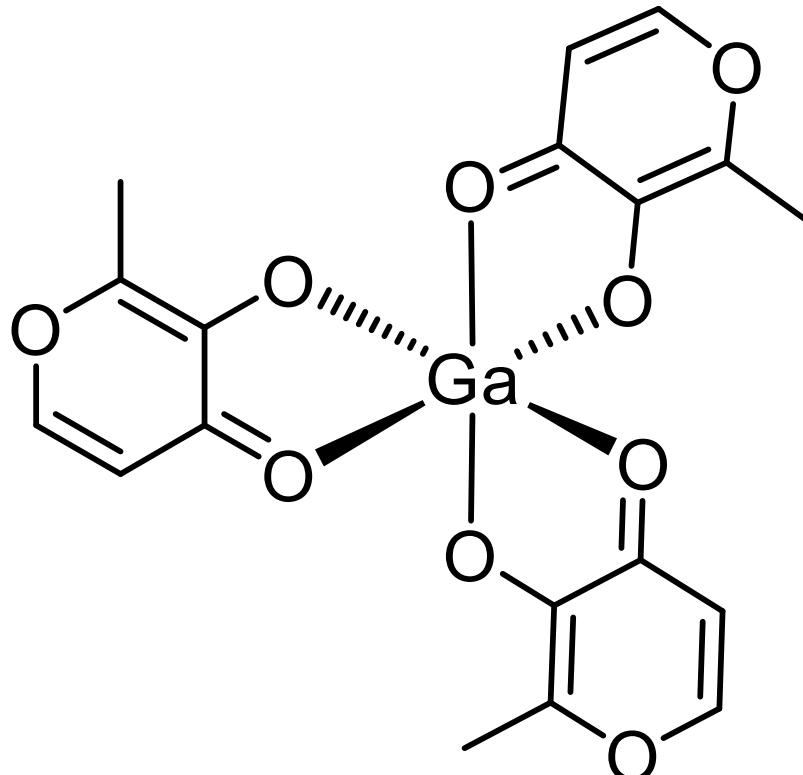
4-(N-(S-glutathionylacetyl)amino)phenylarsonous acid



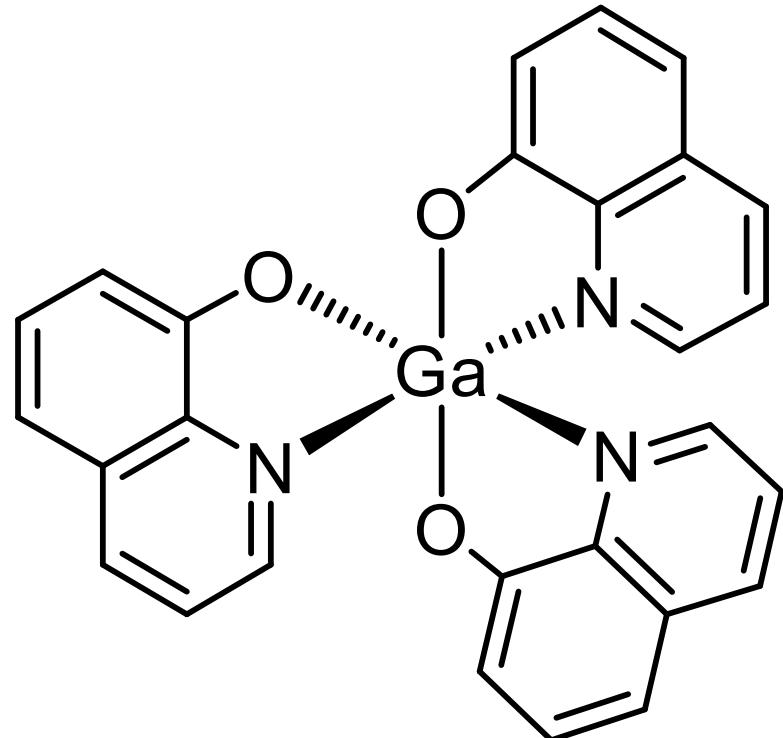
Darinaparsin

S-dimethylarsinoglutathione

Potenziali composti antitumorali di gallio



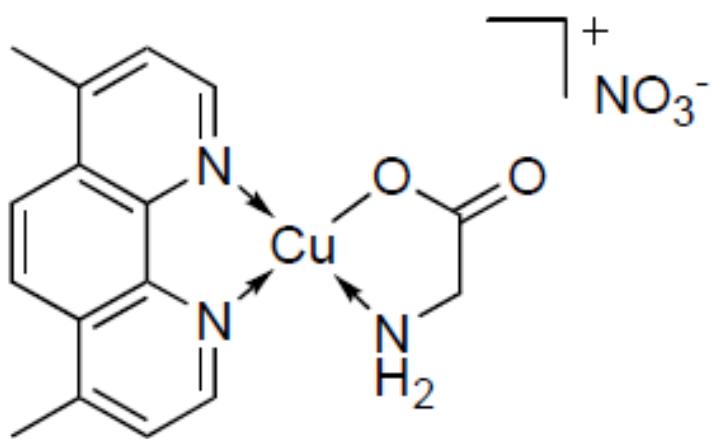
tris-maltolato



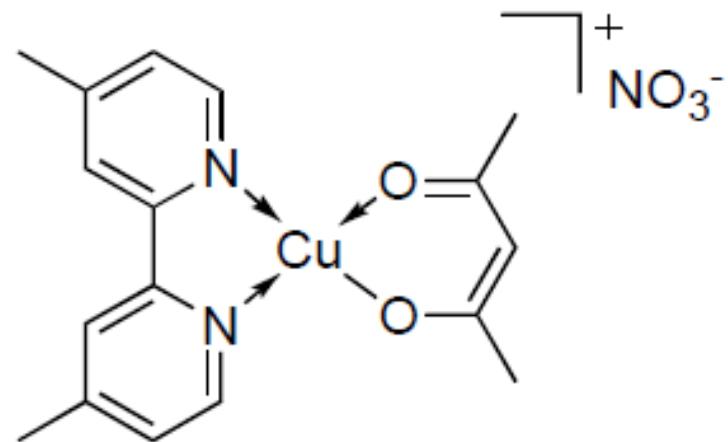
tris(8-idrossichinolinato)

Ga^{3+} è simile a Fe^{3+} ma inibisce la ribonucleotide reduttasi

Potenziali composti antitumorali di rame (Casiopeine)



Cas II-gly



Cas III-ia

Intercalazione nel DNA + generazione di ROS