

Geophysical Fluid Dynamics

Lecture V, VI, VII: Conservation Laws

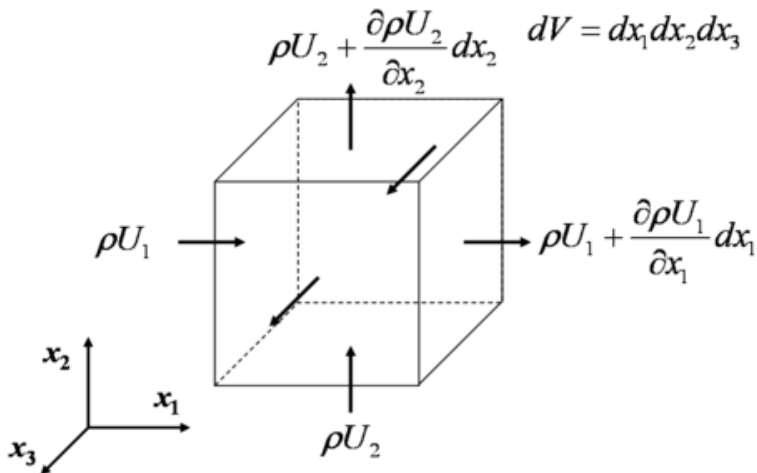
- 1 **Leibniz Theorem for time derivative of volume integrals**
- 2 **Conservation of Mass - Continuity Equation**
 - **Conservation Equation for a tracer**
 - **Advection-Diffusion**
 - **Diffusion**
- 3 **Conservation of Momentum**
 - **Cauchy**
 - **Navier-Stokes Equations**
 - **Euler Equation**
 - **The case of a rotating frame (towards the GFD Eq.)**
- 4 **Conservation of Energy**
 - **Kinetic, Mechanical, Potential and Total Energy**
 - **First and Second law of thermodynamics**
 - **Bernoulli's Equation - Principle**

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- ① **Leibniz Theorem for time derivative of volume integrals**
- ② **Conservation of Mass - Continuity Equation**
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Conservation of Mass



Conservation of Mass

Conservation of Mass

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{u}) = 0 \quad (1)$$

or

$$\frac{1}{\rho} \frac{D\rho}{Dt} + \nabla \cdot \mathbf{u} = 0 \quad (2)$$

For a steady flow, it reduces to

$$\nabla \cdot (\rho \mathbf{u}) = 0 \quad (3)$$

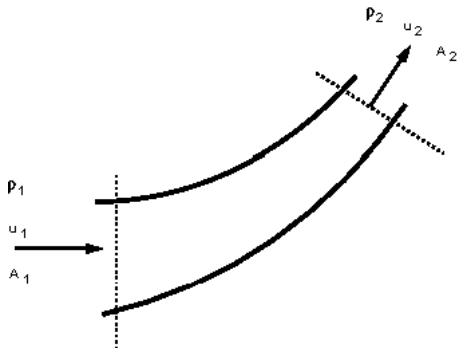
For an incompressible flow, it reduces to

$$\nabla \cdot \mathbf{u} = 0 \quad (4)$$

Continuity

$$\rho_1 u_1 A_1 = \rho_2 u_2 A_2 \quad (5)$$

so that, in this case, $u_2 > u_1$.



Conservation law for a substance/tracer

The molecular coefficient

$$K = U^c \times l, \quad (6)$$

where U^c is the characteristic velocity and l is the free molecular distance. Units are m^2s^{-1} .

If we have advection and diffusion the balance is

$$\frac{\partial C}{\partial t} = (\text{Advection in} - \text{Advection out}) + (\text{Diffusion in} - \text{Diffusion out}) \quad (7)$$

Conservation law for a substance/tracer

$$\boxed{\frac{DC}{Dt} = -C\nabla\mathbf{u} - K\nabla^2 C} \quad (8)$$

For incompressible / steady flows, since

$$\frac{\partial\rho}{\partial t} = 0 \rightarrow \nabla\cdot\mathbf{u} = 0$$

so we have the *Advection-Diffusion Equation*

$$\boxed{\frac{DC}{Dt} = -K\nabla^2 C} \quad (9)$$

If we assume no advection ($\mathbf{u} = 0$), then we have the *Diffusion Equation*

$$\boxed{\frac{\partial C}{\partial t} = -K\nabla^2 C} \quad (10)$$