

Esercizi

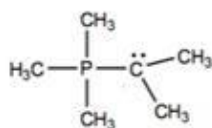
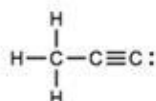
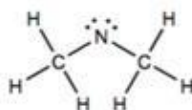
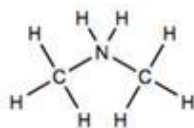
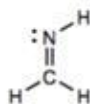
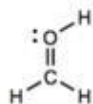
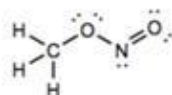
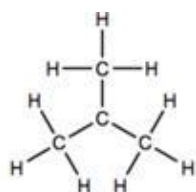
- 1) Disegnare le strutture di Lewis indicando tutti i legami tra gli atomi e i doppietti non condivisi, delle seguenti molecole:

Esempio



CH₃O⁻, CH₃CH₂OH, H₂CO, NH₄⁺, HCN, H₃O⁺, C₂H₂, C₂H₄, AlCl₃, F₃CCO₂H

- 2) Per le seguenti strutture di Lewis, indicare le cariche formali



Il P e' nella stessa colonna del N nella tabella periodica, quindi potete tranquillamente assumere che i comportamenti sono analoghi.

