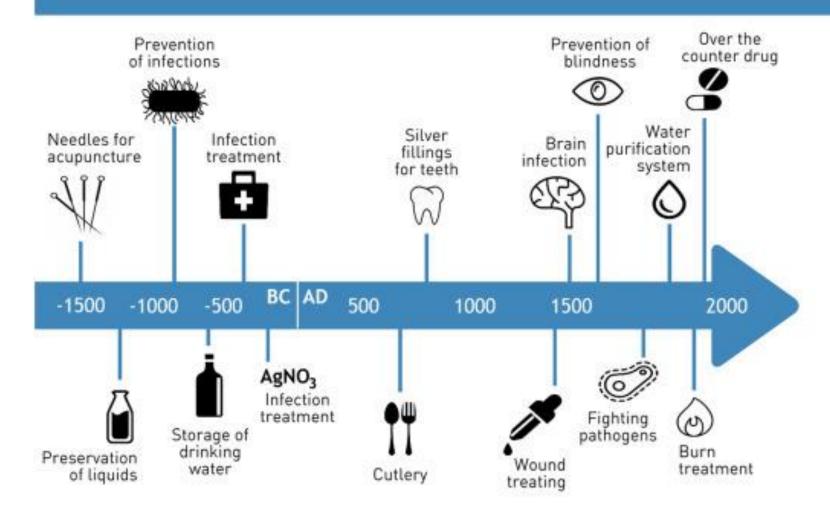


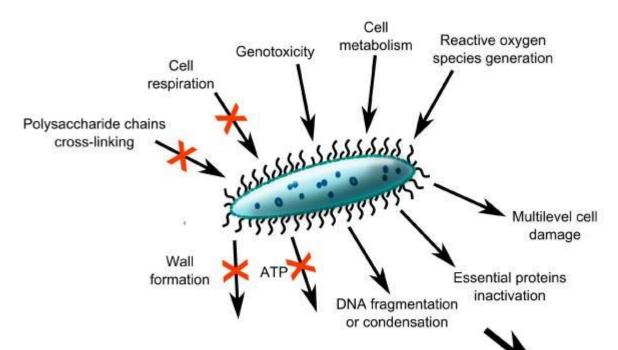
Si stima che nel 2050 a livello mondiale le morti da infezione batterica – dovute soltanto a ceppi di batteri resistenti – arriveranno a 50 milioni all'anno (vs 8.8 milioni dovute al cancro).

Timeline: a brief history of the use of Silver



Metallo massivo – Nanoparticelle – Sali (Ag+)

Multiple mechanism of action of Ag+ ions



Cell death

MIC (minimal inhibitory concentration) of AgNO₃ to E. coli = 3μ M (i.e. $323 \mu g L^{-1}$) LD₅₀ in humans: 140 mg kg⁻¹



Silver salts

$$\begin{array}{c|c}
O & O & N \\
N & N & N \\
N & N & N \\
Ag^{+} & N & SSD
\end{array}$$

silver sulfadiazine



Silver nanoparticles (AgNPs)



Estimated 2014 production of commercial AgNPs: 320 t







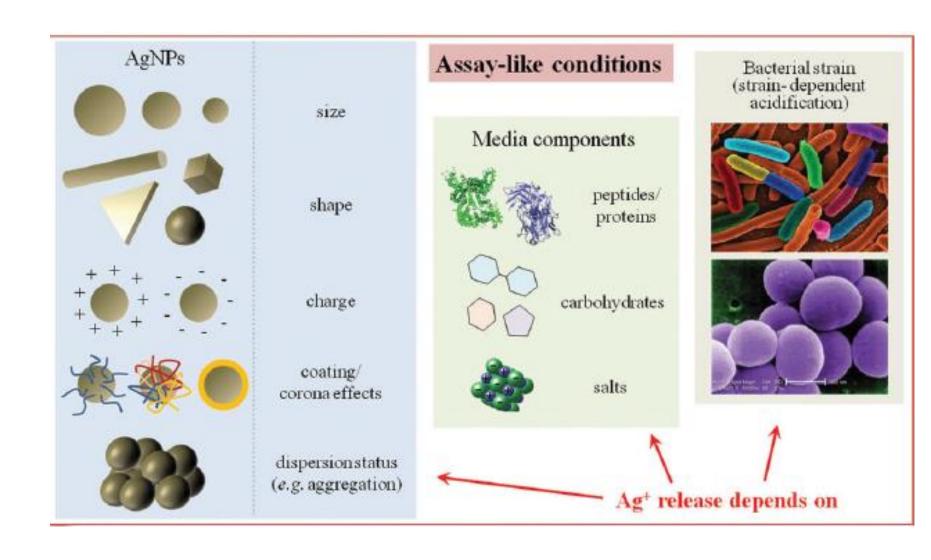




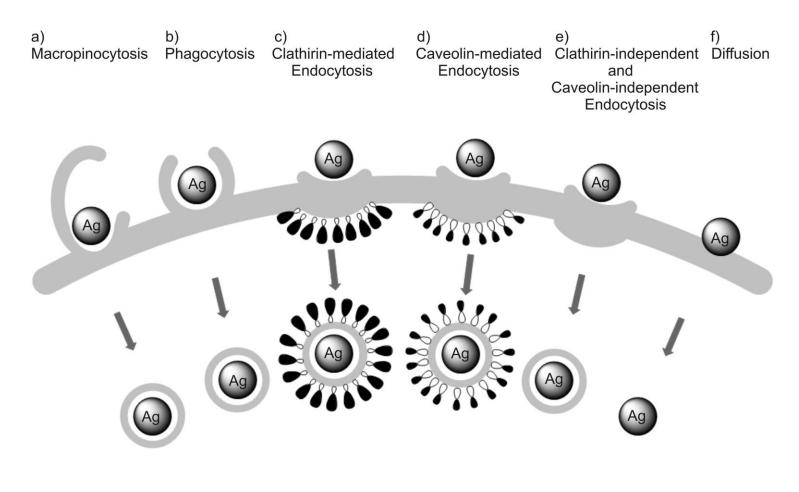




Ag+ release from AgNPs



Uptake of AgNPs



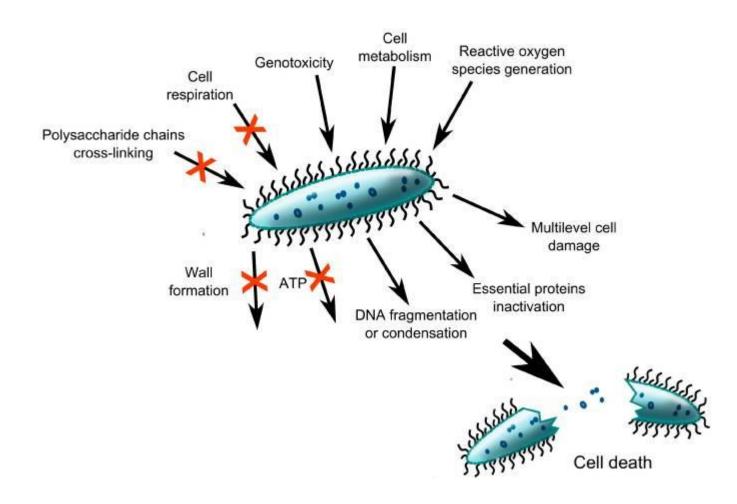
Legend:

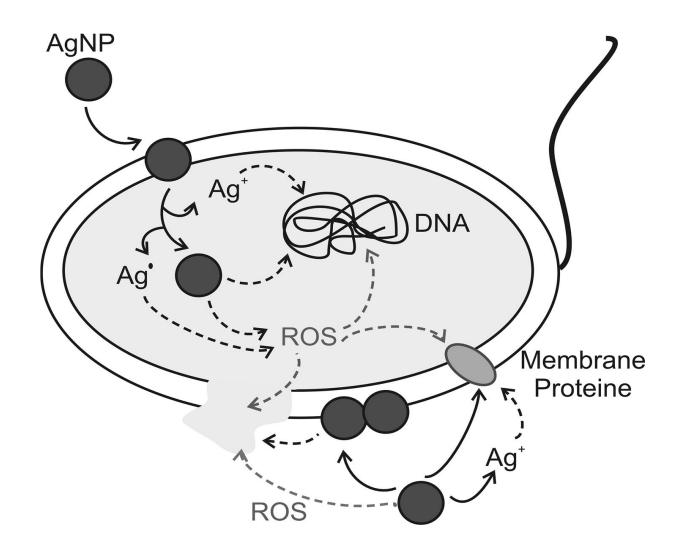






Multiple mechanism of action of Ag+ ions





Antibacterial Ag-NHC compounds

$$\begin{array}{c|c}
O \\
N \\
N \\
N \\
N
\end{array}$$

$$\begin{array}{c}
Ag-O \\
N \\
Silvamist^{\mathbb{R}}
\end{array}$$

Drug candidate: high activity against tobramycin-resistant pathogenic bacteria *in vitro* as well as *in vivo*.

$$CI$$
 N
 R'
 R'
 R'

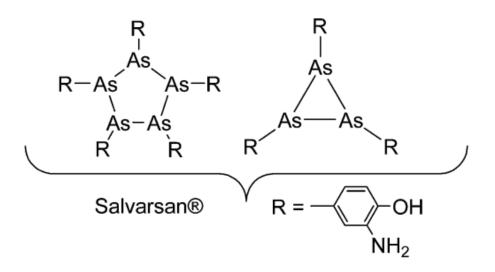
Improved stability to hydrolysis due to the electron-withdrawing CI substituents that pull electron density from the carbene

Other strategies: metal modification of known antibiotics

$$\begin{array}{c} \text{COOH} \\ \text{OH} \\ \text{OH} \end{array} \longrightarrow \begin{array}{c} \text{COOH} \\ \text{OH} \\ \text{OH} \end{array} \longrightarrow \begin{array}{c} \text{Organometallic} \\ \text{moiety} \end{array}$$

Platensimycin mimics

Other metals: As, Sb, Bi, Hg



Antimicrobial agent introduced in early 1900 for the treatment of the deadly bacterial infection *Syphilis*. Later replaced by modern antibiotics

Antiparasitic compounds

melarsoprol

treatment of sleeping sickness (African trypanosomiasis)

Anti-leishmaniasis compounds

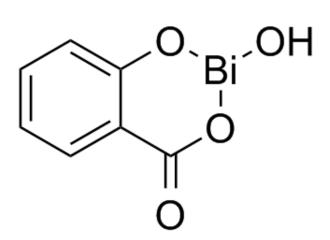
meglumine antimoniate

sodium stibugluconate

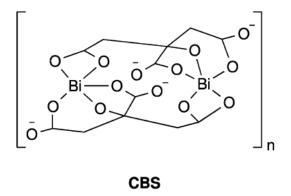
Veterinary use

Sb, reduced to Sb(III), inhibits *trypanothione reductase*, an essential enzyme of the parasite

Infezioni da Helicobacter pylori

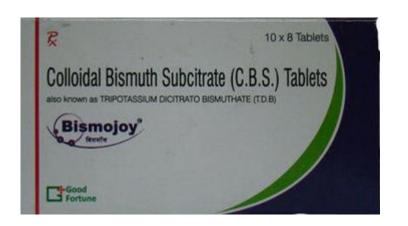


bismuth subsalicylate





The pink stuff (introduced 1901)



bismuto subcitrato colloidale

Infezioni da Helicobacter pylori

$$H_3C$$
 O
 S
 N
 N
 CH_3
 CH_3
 CH_3

ranitidine bismuth citrate



Antibacterial mercury compounds

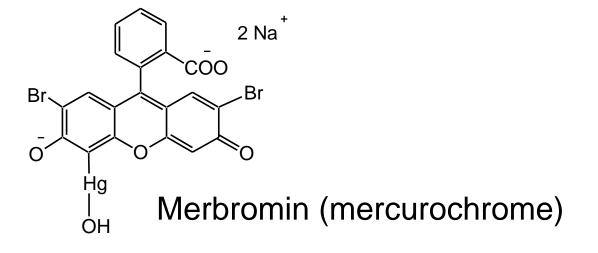
Na
$$O$$

Na O

Thiomersal

vaccine adjuvant

ethyl mercury





FosrenolTM: La₂(CO₃)₃·4H₂O a success story



FOSRENOL Chewable Tablets FOSRENOL Oral Powder Phosphorus Burden in ESRD Patient Support Conferences and Resources

To reduce serum phosphate in patients with end-stage renal disease (ESRD)

HELP IT FALL WITH FOSRENOL*

(lanthanum carbonate)

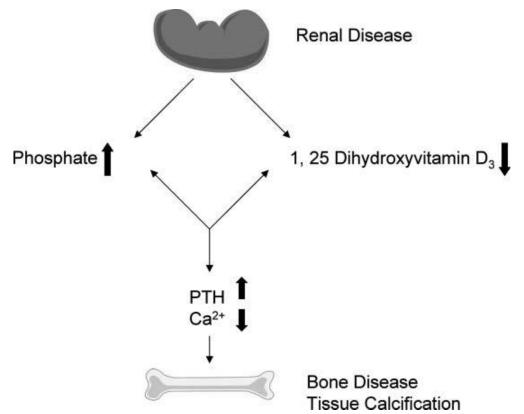
*Phosphorus reductions maintained for up to 3 years in patients remaining on therapy (n=46)¹⁻³

- FOSRENOL Chewable Tablets: Approved in 2004 and used in US clinical practices for more than a decade^{4,5} LEARN MORE
- FOSRENOL Oral Powder: Available since May 2015, offering you another approved administration option⁶ LEARN MORE



Approved by FDA in 2004 for the treatment of hyperphosphatemia (increased phosphate levels in serum) in patients with end stage renal disease.

Phosphate metabolism is intimately linked with calcium metabolism, and is regulated by parathyroid hormone (PTH) and vitamin D



Pathological consequences of hyperphosphatemia: cardiac and vascular tissue calcification, bone malformations in the joints

The ideal phosphate binder should:

- have a high affinity for phosphate and
- be able to bind dietary phosphate rapidly in the guts
- have low solubility and
- little or no systemic absorption.
- be non-toxic,
- be available as a palatable oral dosage form, with a low pill burden.

Calcium phosphate binders are effective....however, calcium can be absorbed, resulting in hypercalcemia and increased risk of cardiovascular calcification.

FosrenolTM: a success story

Among the many lanthanide salts screened, La₂(CO₃)₃·4H₂O possessed the best phosphate binding properties:

- Optimal binding at pH 3–5, but retains binding activity in the full pH range of 1–7
- It is very insoluble and the La³⁺ cation does not cross biological membranes (when given by the oral route, >90% excreted in the feces, and <0.001% absorbed)
- No toxicity observed in animal studies, in particular no direct effects on calcium, vitamin D, or PTH metabolism

FosrenolTM represents a significant improvement in treatment options for patients with end-stage renal disease.

- in the acidic environment of the stomach lanthanum carbonate dissociates sufficiently to allow formation of a highly insoluble phosphate.
- It has the required pharmacokinetic properties, it is poorly absorbed, with both the parent molecule and the phosphate product being eliminated in the feces.
- Because of the lack of absorption it has no systemic toxicity, it has no detrimental effect on calcium, vitamin D or PTH metabolism, and is safe and well tolerated.
- Its effectiveness as a phosphate binder results in a lower pill burden for patients, an advantage over competing medications.