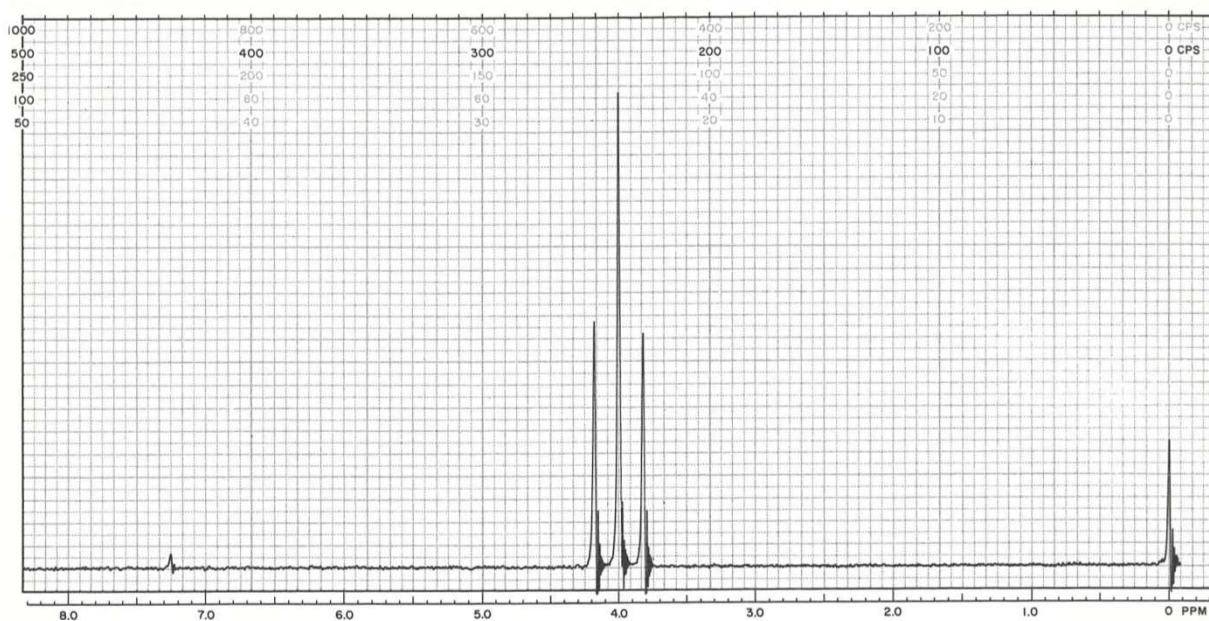


## Esercizio 1

Un composto di formula  $C_2H_2Cl_2F_2$  dà il seguente spettro  $^1H$ -NMR, proporre una struttura per il composto e giustificare la molteplicità del segnale.

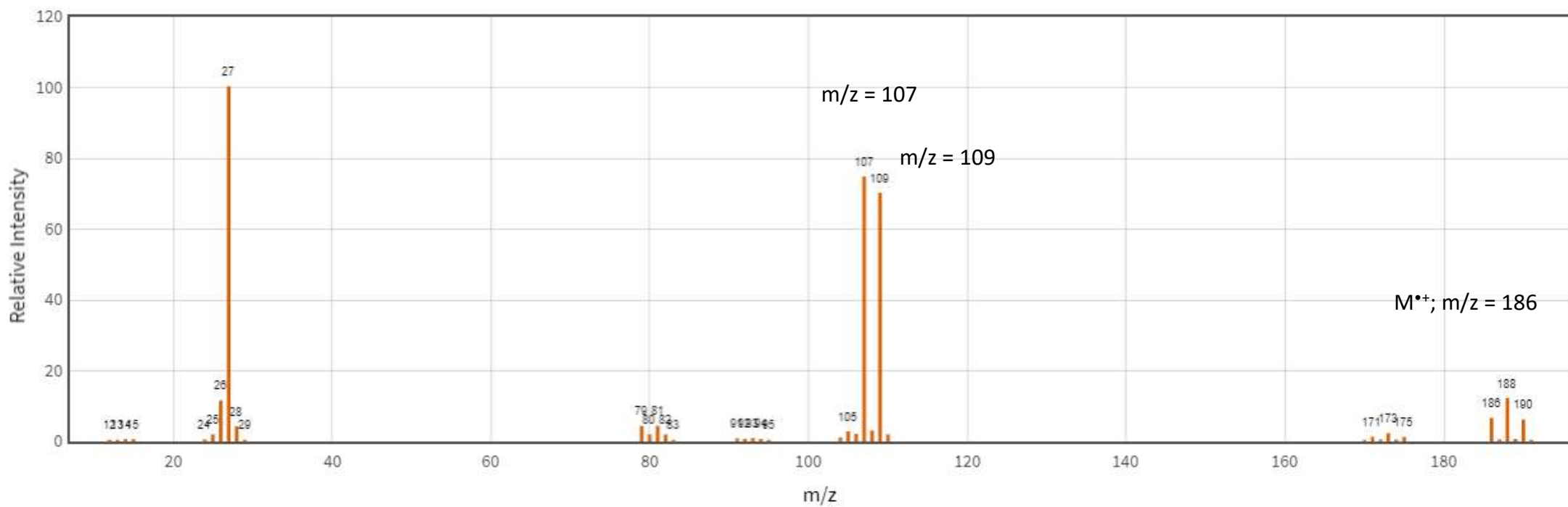


$^1H$ -NMR: 60 MHz,  $CDCl_3$

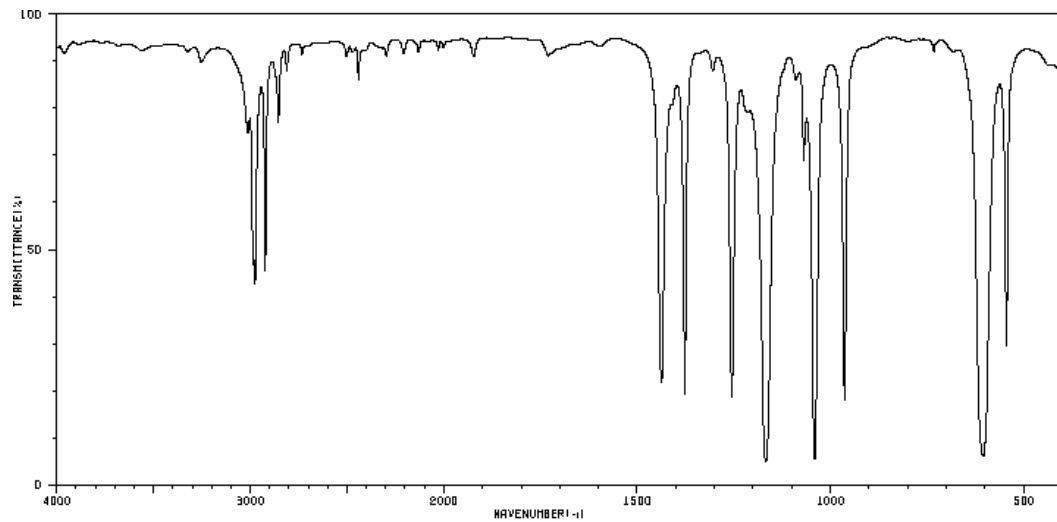
4.0 ppm: tripletto

## Esercizio 2

I seguenti dati spettrali si riferiscono ad un composto con due atomi di carbonio. Determinarne la struttura



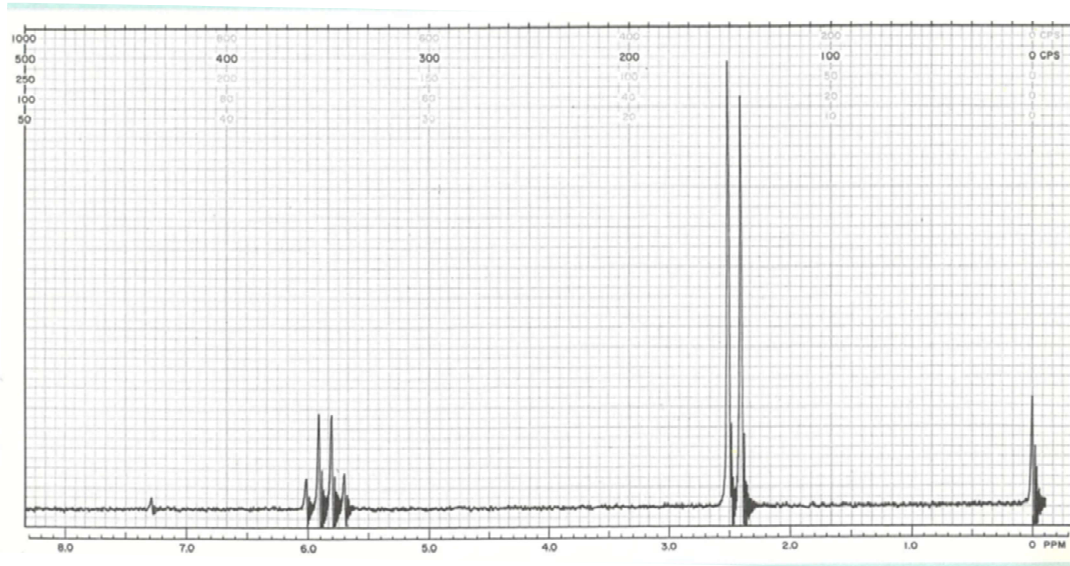
## Esercizio 2



**IR: film liquido**

Assorbimenti rilevanti

2976, 2923, 1438, 1377, 1168, 606 cm<sup>-1</sup>



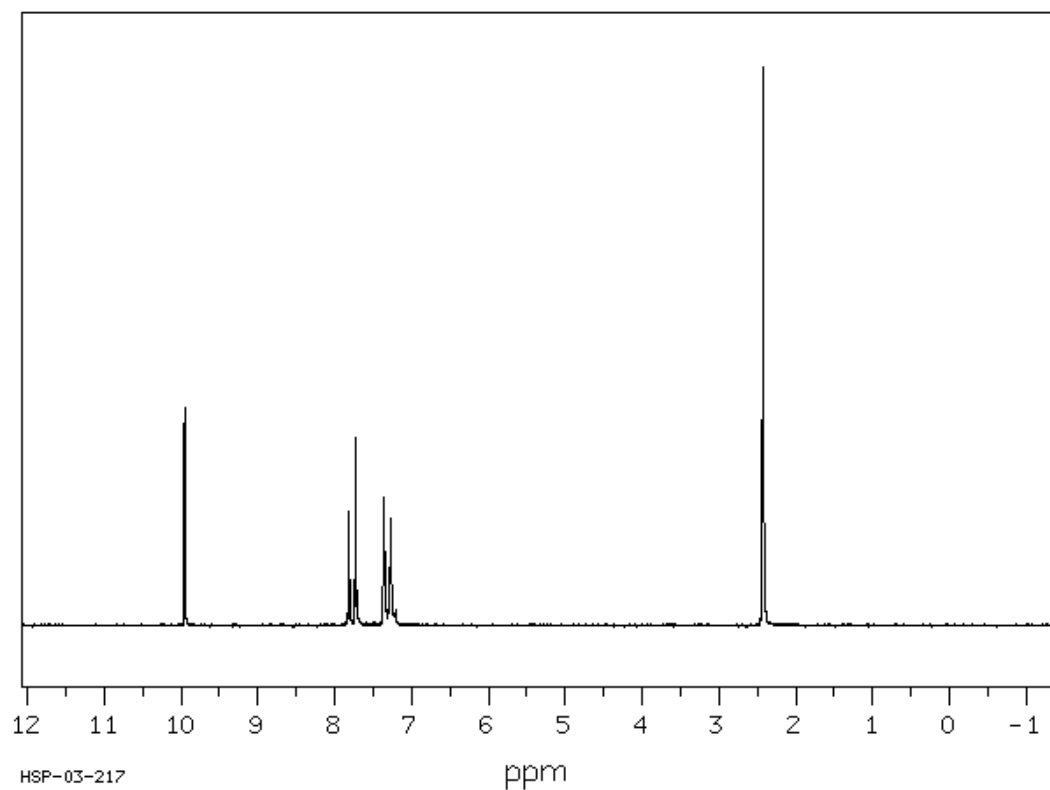
**<sup>1</sup>H-NMR: 60 MHz, CDCl<sub>3</sub>**

5.86 ppm: quartetto di area 1

2.47 ppm: doppietto di area 3

## Esercizio 3

I seguenti dati spettrali si riferiscono ad un composto di formula  $C_8H_8O$ . Determinarne la struttura.



**$^1H$ -NMR: 90 MHz,  $CDCl_3$**

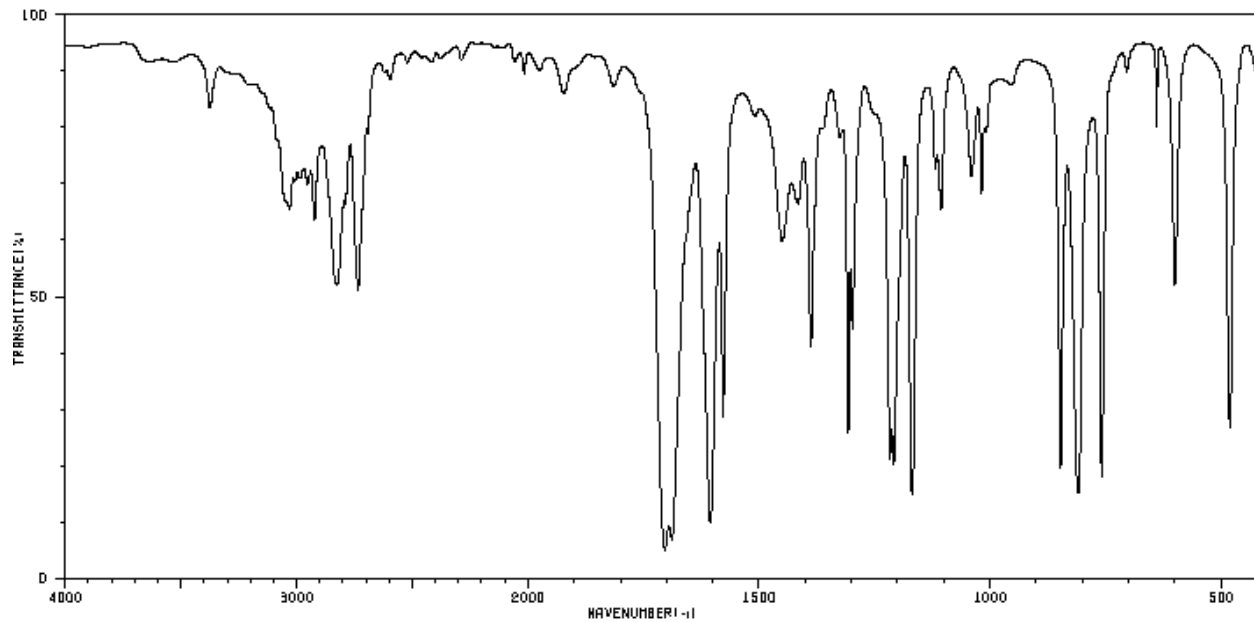
9.95 ppm: s, area 1

7.76 ppm: d, area 2

7.32 ppm: d, area 2

2.43 ppm: s, area 3

## Esercizio 3

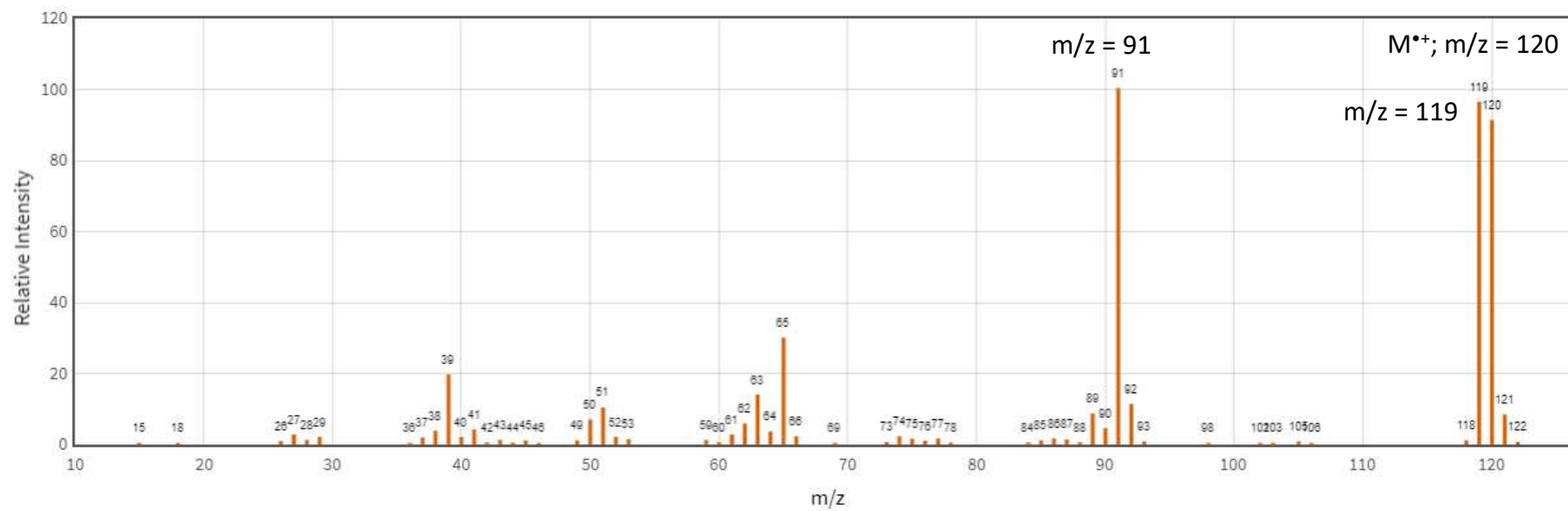


**IR: film liquido**

Assorbimenti rilevanti

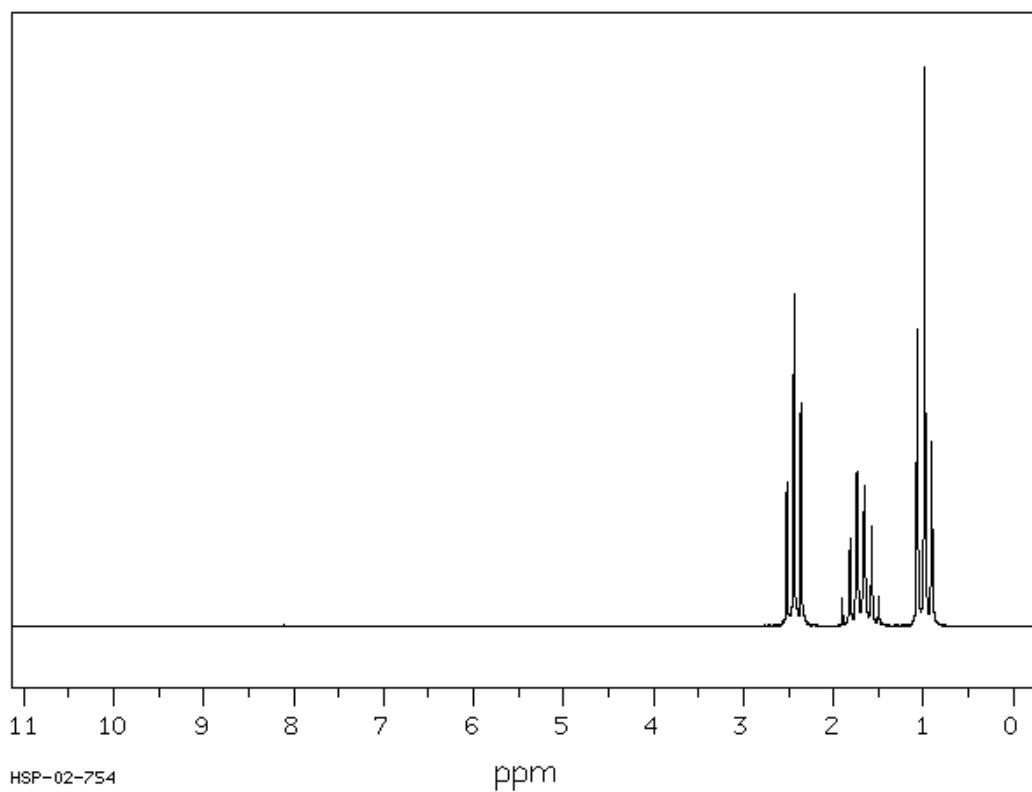
3044, 3031, 2827, 2734,  
1704  $\text{cm}^{-1}$  non 1740  $\text{cm}^{-1}$   
come mostrato a lezione

# Esercizio 3



## Esercizio 4

I seguenti dati spettrali si riferiscono ad un composto di formula  $C_8H_{14}O_3$ . Determinarne la struttura.



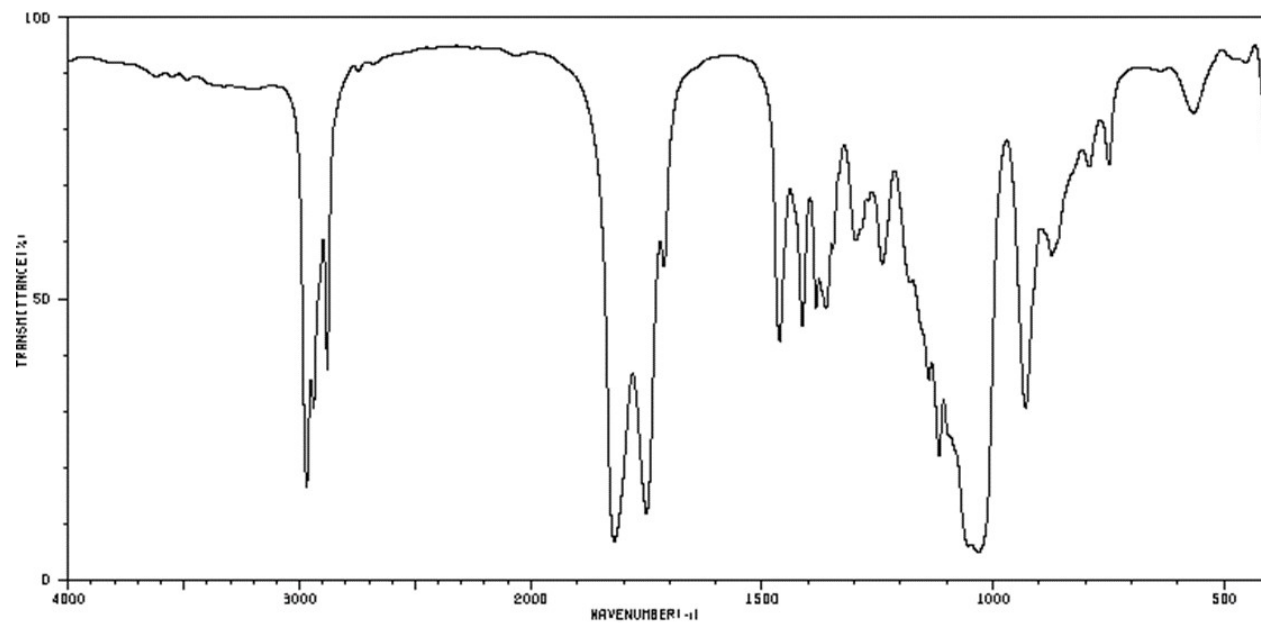
**$^1H$ -NMR: 90 MHz,  $CDCl_3$**

2.43 ppm: t, area 2

1.69 ppm: sest, area 2

1.00 ppm: t, area 3

## Esercizio 4



**IR: film liquido**

2970	16	1413	49	1116	21	666	79
2940	29	1383	46	1039	5		
2880	36	1362	46	1031	4		
1819	6	1297	68	929	29		
1750	11	1287	60	874	55		
1712	53	1240	59	792	70		
1461	41	1140	34	749	70		