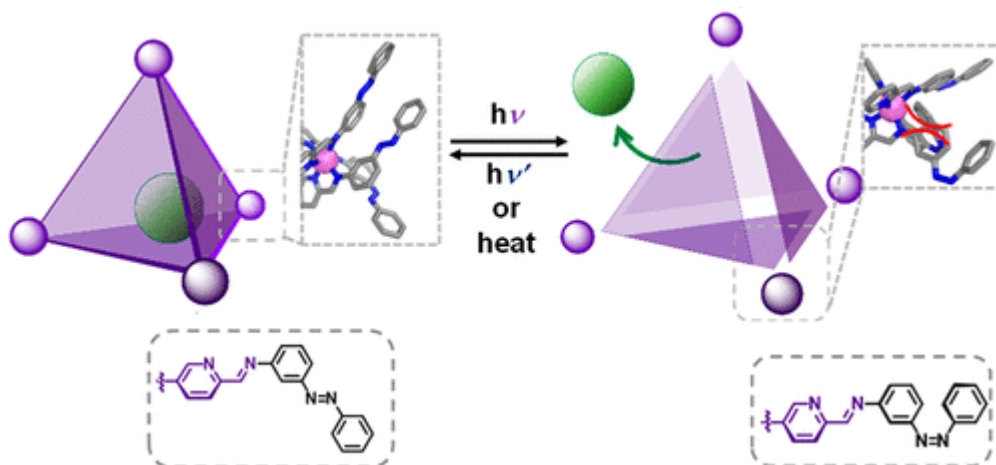


## Light-Powered Reversible Guest Release and Uptake from $Zn_4L_4$ Capsules

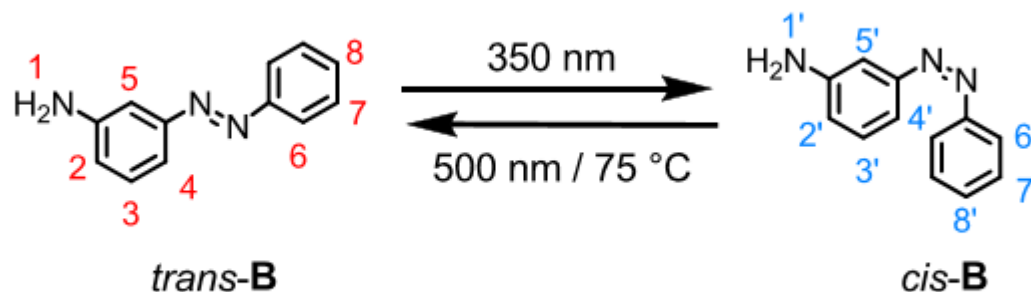
Amit Ghosh, Laura Slappendel, Bao-Nguyen T. Nguyen, Larissa K. S. von Krbek, Tanya K. Ronson, Ana M. Castilla, and Jonathan R. Nitschke\*

 Cite This: *J. Am. Chem. Soc.* 2023, 145, 3828–3832

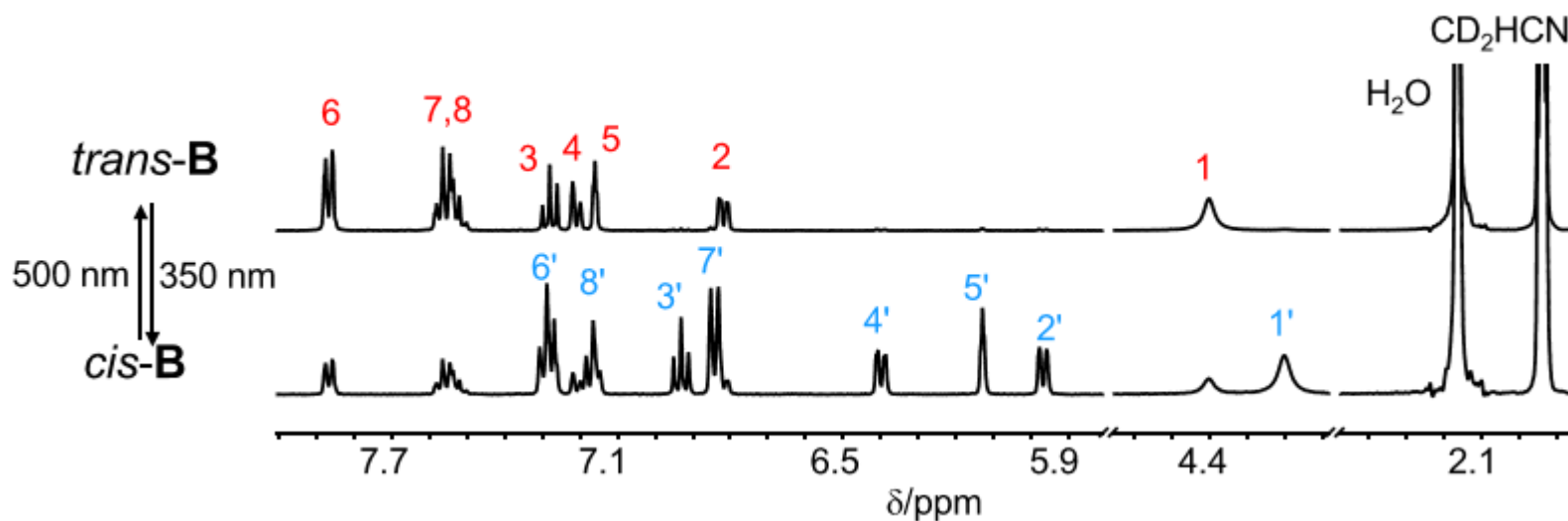
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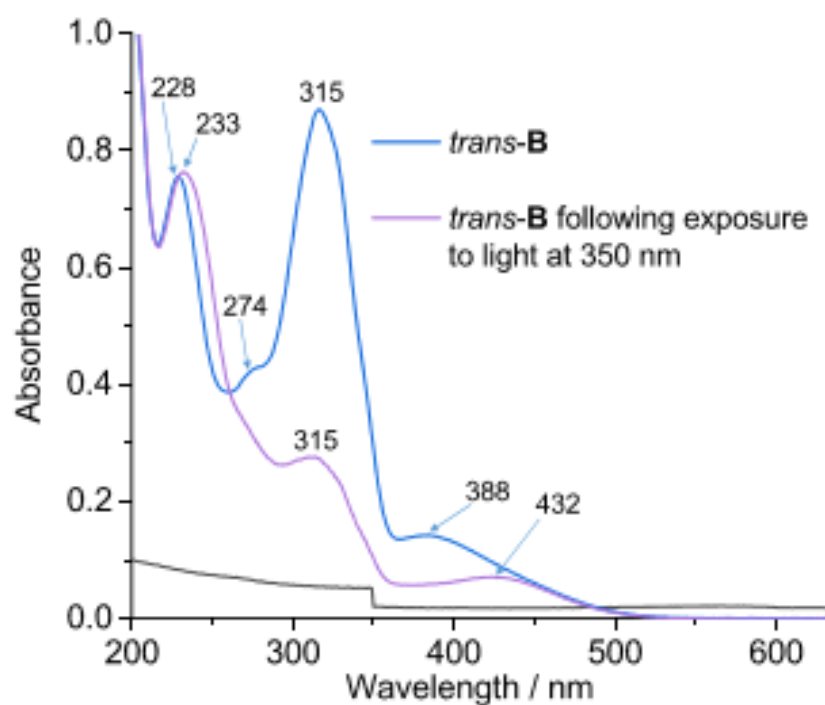


## Subcomponent B

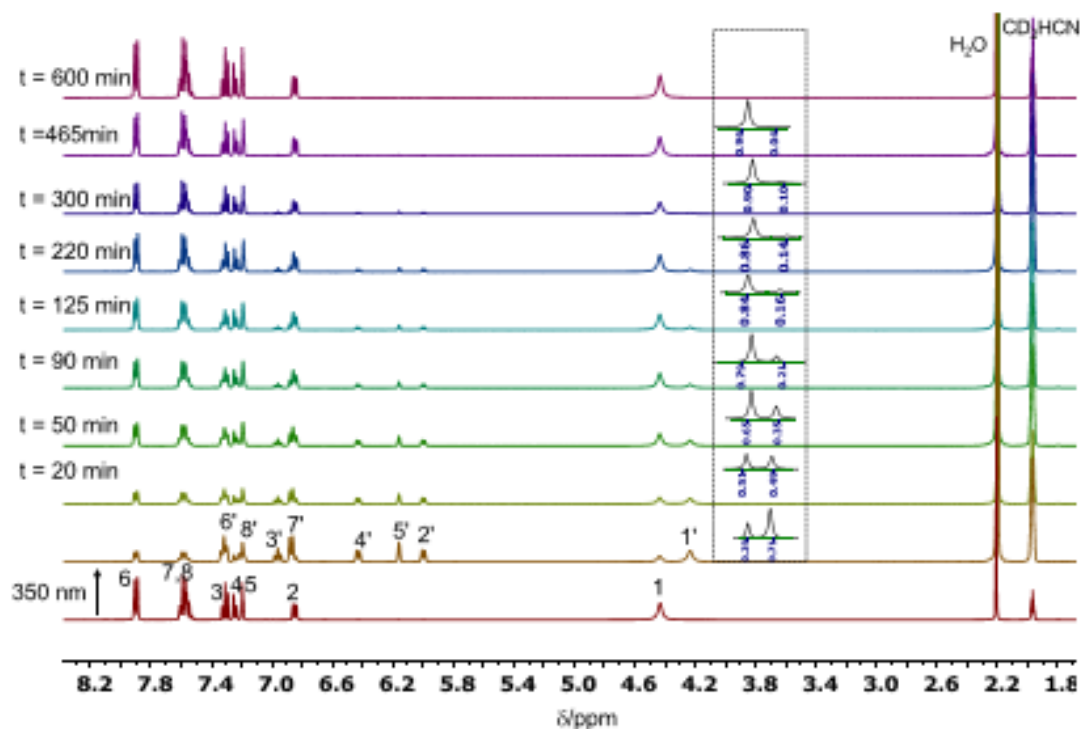


**Figure S20.** Reversible photoswitching of azobenzene-containing subcomponent **B**.

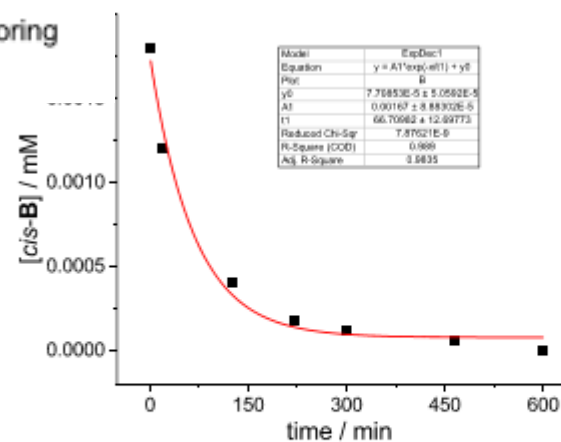




**Figure S24.** The comparison of the UV-vis spectra of *trans-B* ( $7.2 \times 10^{-6}$  M,  $\text{CH}_3\text{CN}$ ,  $25^\circ\text{C}$ ) and then following exposure to light at 350 nm for 10 min.



**Figure S22.** Time-dependent  $^1\text{H}$  NMR ( $\text{CD}_3\text{CN}$ , 400 MHz, 25 °C) spectrum monitoring interconversion between *cis*-**B** to *trans*-**B** by heating the solution at 75 °C.



**Figure S23.** Kinetics for the *cis* to *trans* isomerisation of ligand **B** at 75 °C, monitored by the relative integration of protons 1 and 1' in the  $^1\text{H}$  NMR (Figure S17). The fitting following a 1<sup>st</sup> order kinetics (red line), revealing a half-life of  $t_{1/2} = 46$  min and rate constants of  $k = 2.5 \times 10^{-4} \text{ s}^{-1}$ .

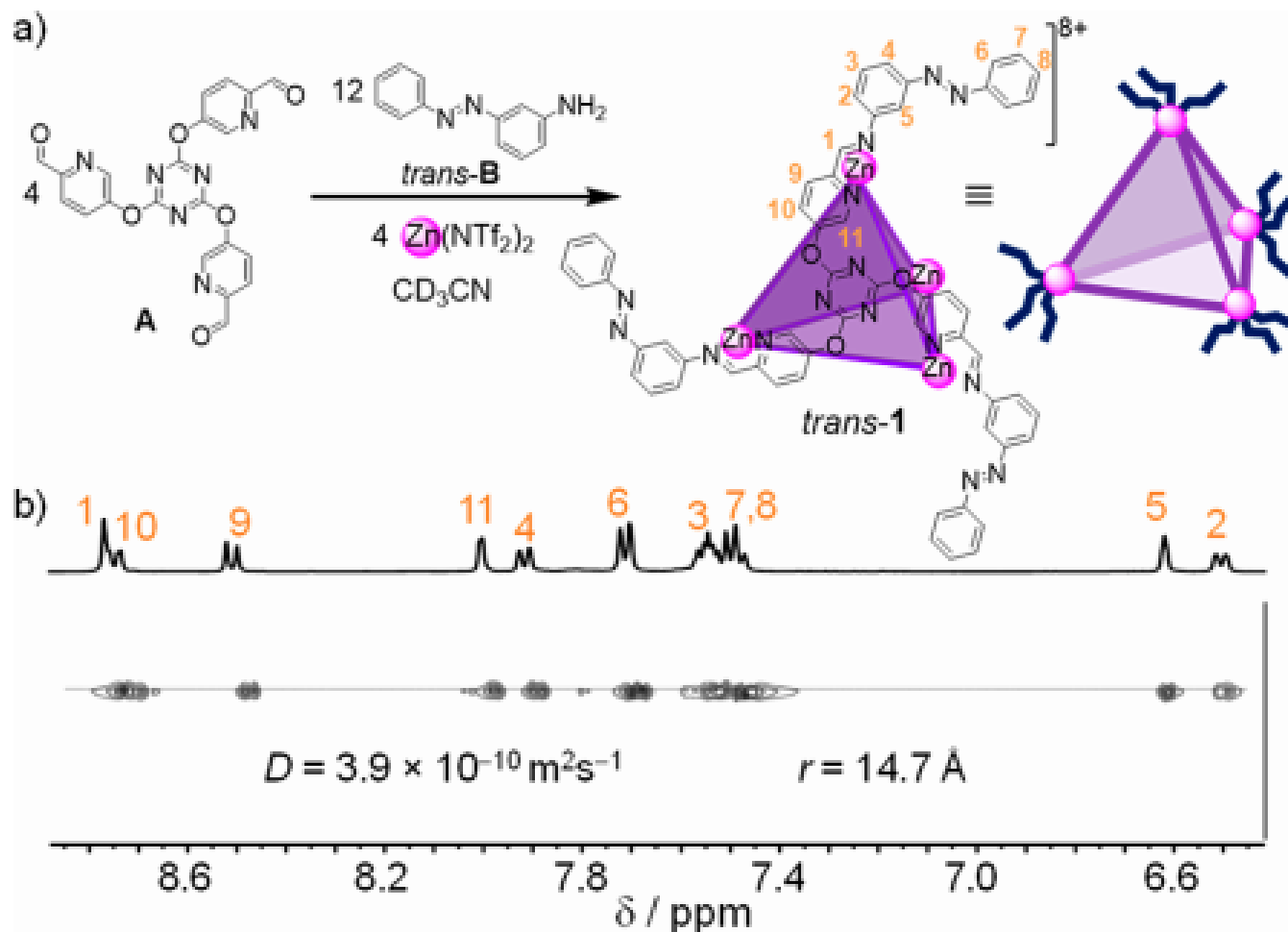
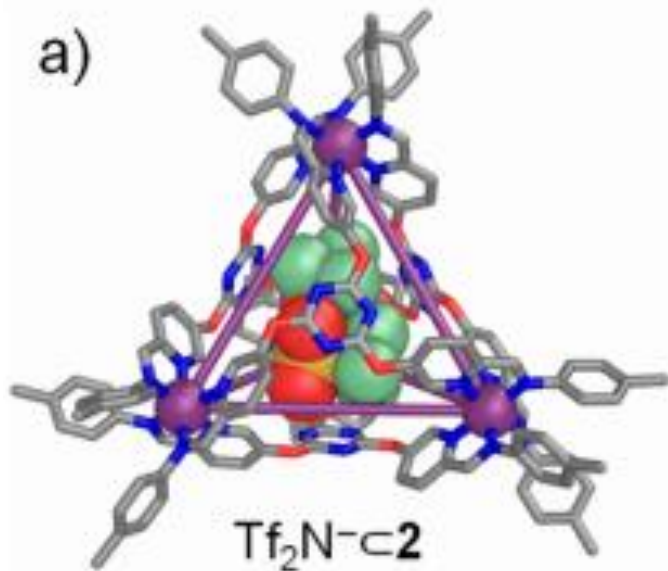


Figure 1. a) Assembly of subcomponents A and *trans-B* with  $\text{Zn}^{\text{II}}$  produced cage *trans-1*. b)  $^1\text{H}$  and DOSY NMR (400 MHz,  $\text{CD}_3\text{CN}$ , 298 K) spectra of *trans-1*.



(a) Crystal structure of  $\text{Tf}_2\text{N}^- \text{c}2$ .  $\text{Tf}_2\text{N}^-$  shown in space-filling mode: N, blue; S, yellow; O, red; C, gray; F, pale green. Disorder, unbound counterions, and solvent of crystallization are omitted for clarity. Attempts to obtain single crystals suitable for X-ray diffraction of an analog of cage 2 with Zn(II) instead of Fe(II) were unsuccessful.

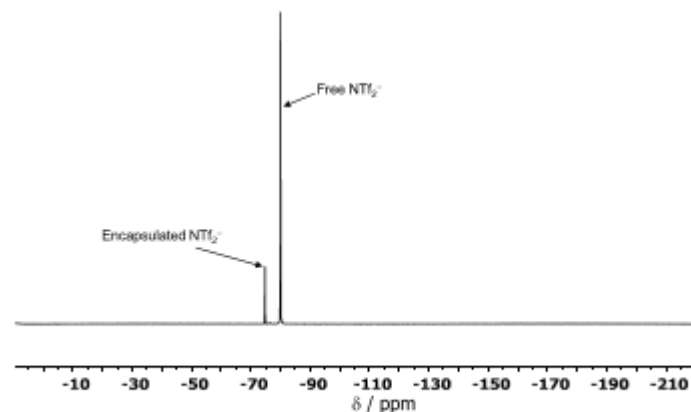
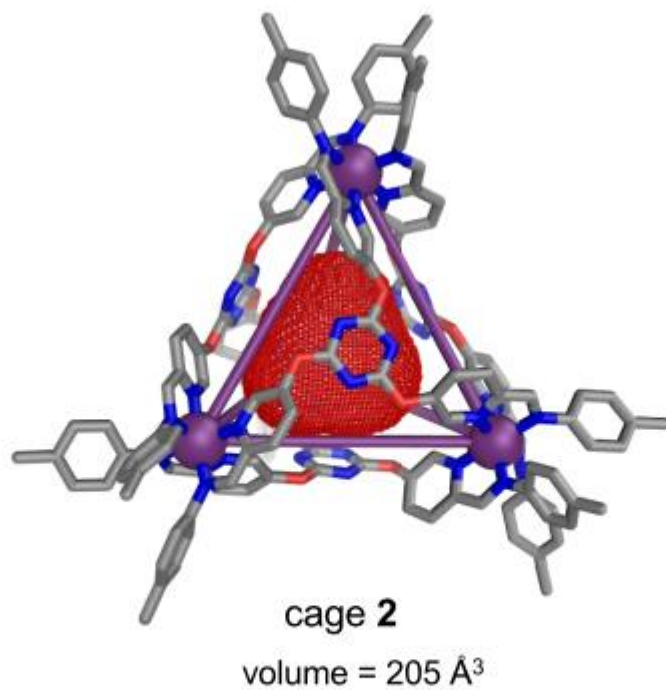
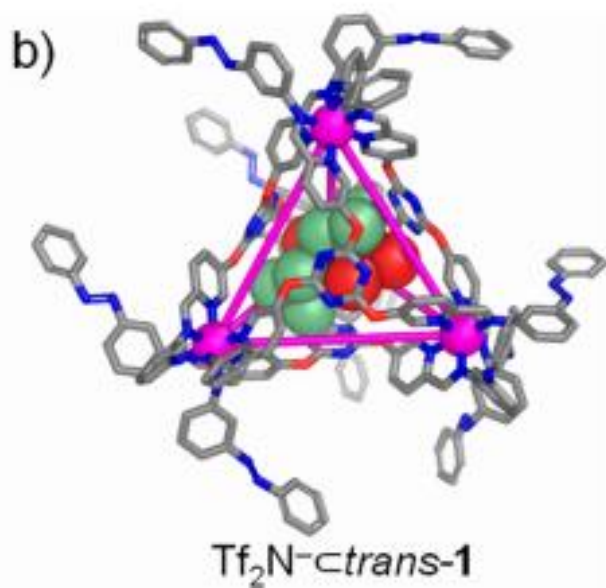


Figure S16.  $^{19}\text{F}$  NMR spectrum of  $\text{Tf}_2\text{N}^- \text{c}2$  (376 MHz,  $\text{CD}_3\text{CN}$ , 25 °C).



**Figure S70.** MoloVol-calculated void space (red mesh) within the crystal structures of cage 2.



b) MM2-optimized molecular model of *trans*-**1**, based on the structure of **2**.

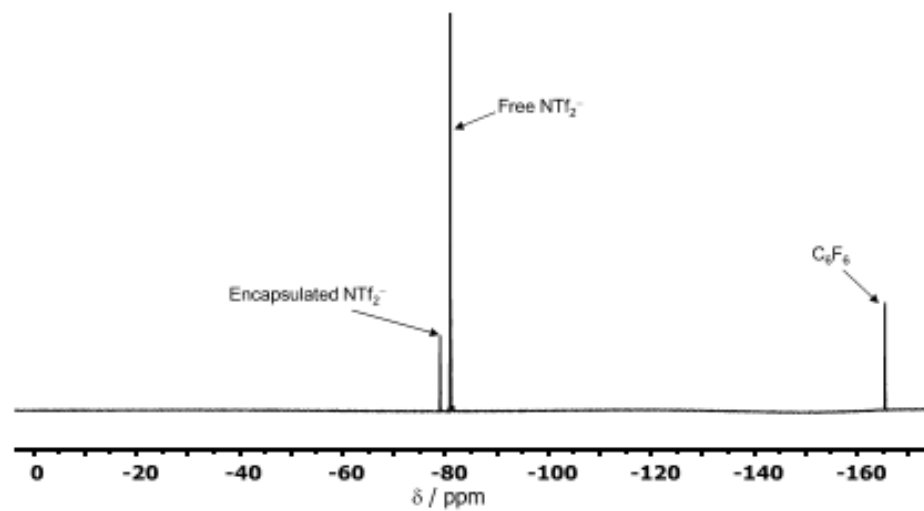
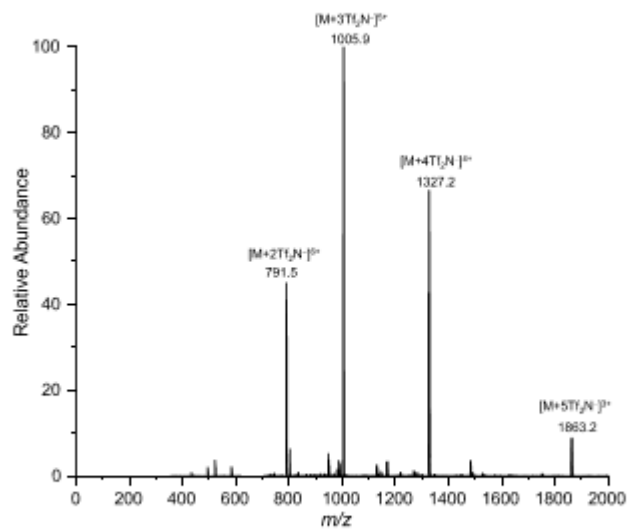
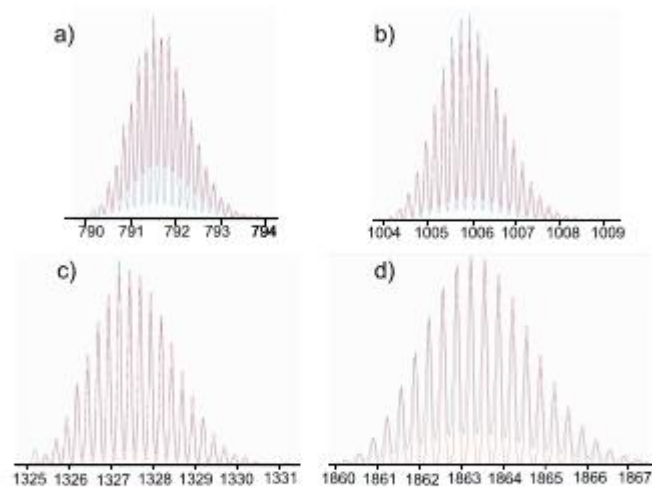


Figure S8.  $^{19}\text{F}$  NMR spectrum of  $\text{Tf}_2\text{N}^- \text{c} \textit{trans}\text{-1}$  (376 MHz,  $\text{CD}_3\text{CN}$ , 25 °C).

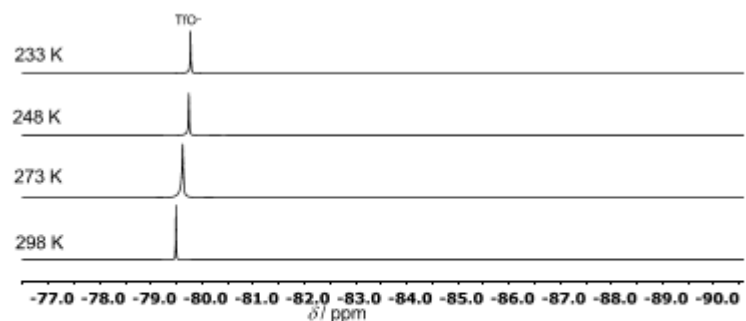
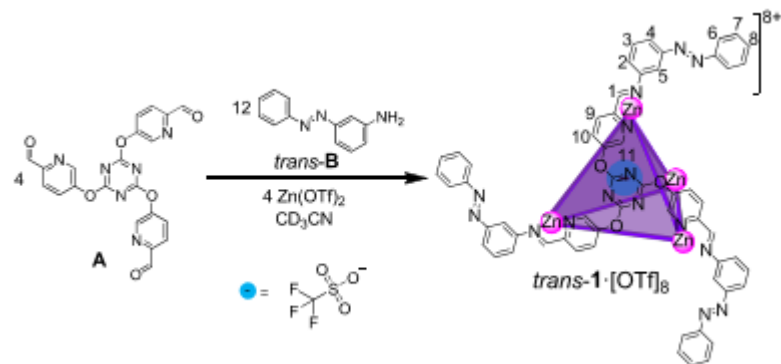




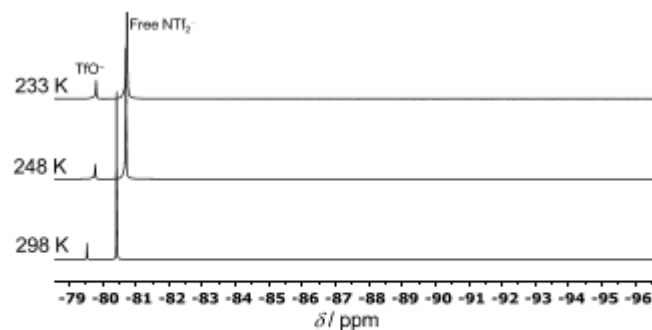
**Figure S9.** Low-resolution ESI-mass spectrum of  $\text{Tf}_2\text{N-cis-1}$  in  $\text{CH}_3\text{CN}$ .



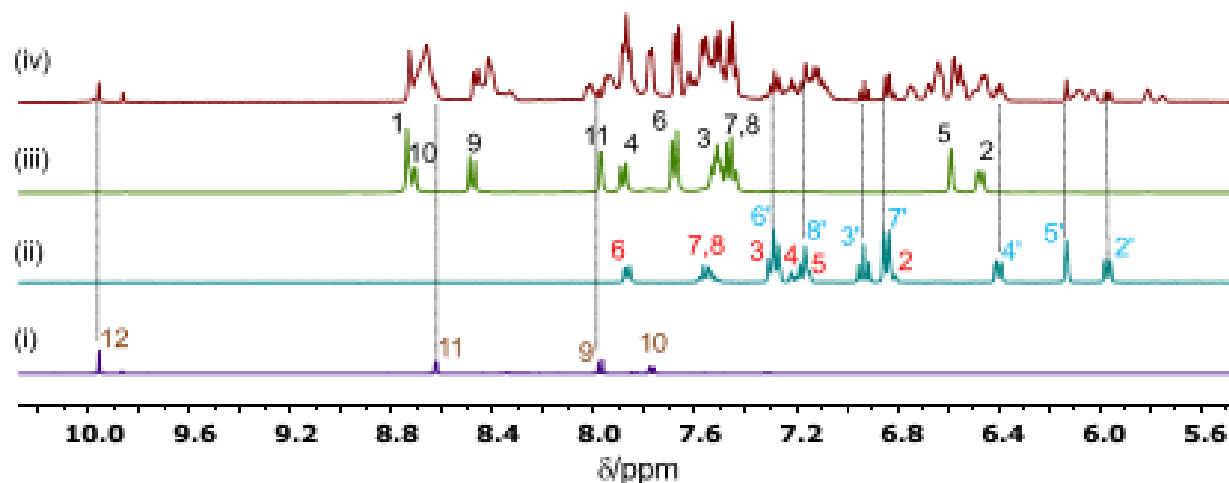
**Figure S10.** High-resolution ESI-MS spectra of  $\text{Tf}_2\text{N-cis-1}$  in  $\text{CH}_3\text{CN}$ . Experimental (blue) and calculated (red) peaks for a)  $[\text{1-2}(\text{NTf}_2)]^{5+}$ :  $m/z$  = 791.4812 (calculated  $m/z$  = 791.4824), b)  $[\text{1-3}(\text{NTf}_2)]^{6+}$ :  $m/z$  = 1005.7605 (calculated  $m/z$  = 1005.7624), c)  $[\text{1-4}(\text{NTf}_2)]^{7+}$ :  $m/z$  = 1327.1812 (calculated  $m/z$  = 1327.1823), d)  $[\text{1-5}(\text{NTf}_2)]^{8+}$ :  $m/z$  = 1862.8807 (calculated  $m/z$  = 1862.8822).



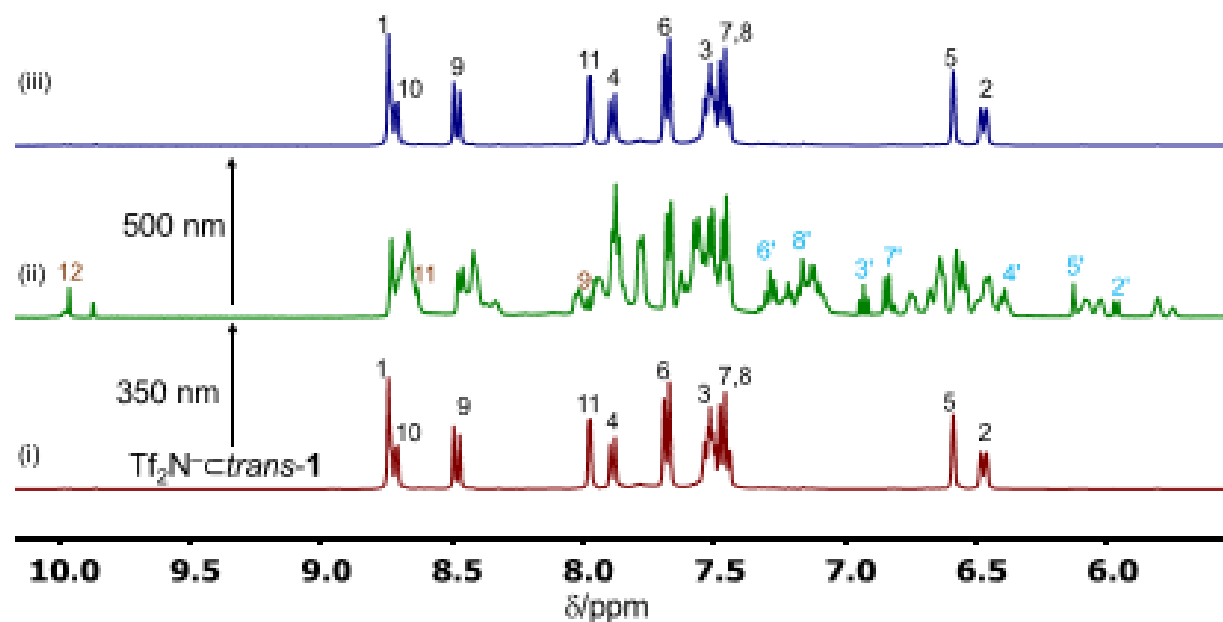
**Figure S54.** <sup>19</sup>F VT-NMR (CD<sub>3</sub>CN, 376 MHz) of *trans-1*-[OTf]<sub>8</sub>. With lowering temperature the signals broadened and shifted, suggesting fast exchange of TfO<sup>-</sup> within cage *trans-1* on the NMR time scale.



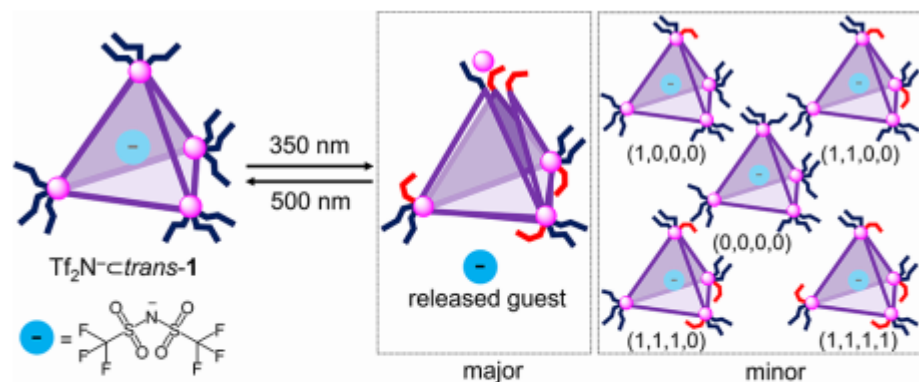
**Figure S52.** <sup>19</sup>F VT-NMR (CD<sub>3</sub>CN, 376 MHz) of TfO-*trans-1*. With lowering temperature the signals broadened and shifted, suggesting fast exchange of TfO<sup>-</sup> within cage *trans-1* on the NMR time scale.



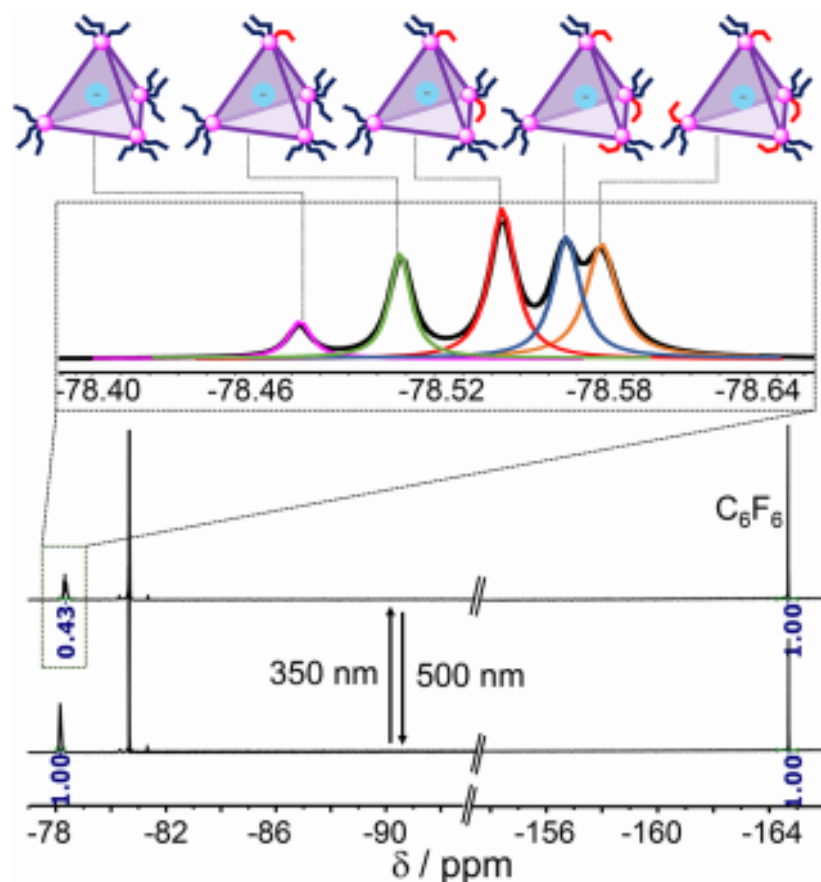
**Figure S27.** Comparison of  $^1\text{H}$  NMR spectra (400 MHz,  $\text{CD}_3\text{CN}$ , 25  $^\circ\text{C}$ ) of (i) subcomponent **A**; (ii) subcomponent **B** after 350 nm irradiation; (iii) pure cage  $\text{Tf}_2\text{N}^-$ -*trans*-**1** and (iv) cage  $\text{Tf}_2\text{N}^-$ -*trans*-**1** after 350 nm irradiation. The signals of subcomponent **A**, *trans*-**B**, *cis*-**B** and  $\text{Tf}_2\text{N}^-$ -*trans*-**1** have been assigned and are indicated by brown, red, cyan and black labels, respectively.



**Figure S28.** Reversible photoswitching of cage  $\text{Tf}_2\text{N-c-trans-1}$  ( $c = 39.6$  mM).  $^1\text{H}$  NMR spectra (500 MHz,  $\text{CD}_3\text{CN}$ ,  $25^\circ\text{C}$ ) of (i) cage  $\text{Tf}_2\text{N-c-trans-1}$ ; (ii) after 350 nm irradiation for 10 min and (iii) after irradiation at 500 nm for 30 min.



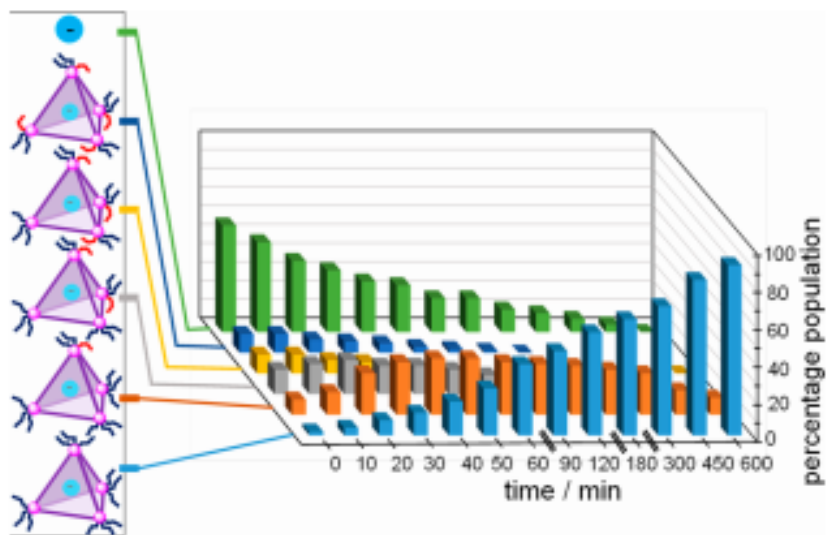
**Figure 3.** Cartoon of the photoswitching of cage  $\text{Tf}_2\text{N}^-$ -*c*trans-1, illustrating the opening of the cage after a fifth azobenzene switches, along with the five observed guest-binding states of the cage, in which 0–4 azobenzene residues have switched to *cis*. B residues with the *trans* and *cis* configurations are colored blue and red, respectively.



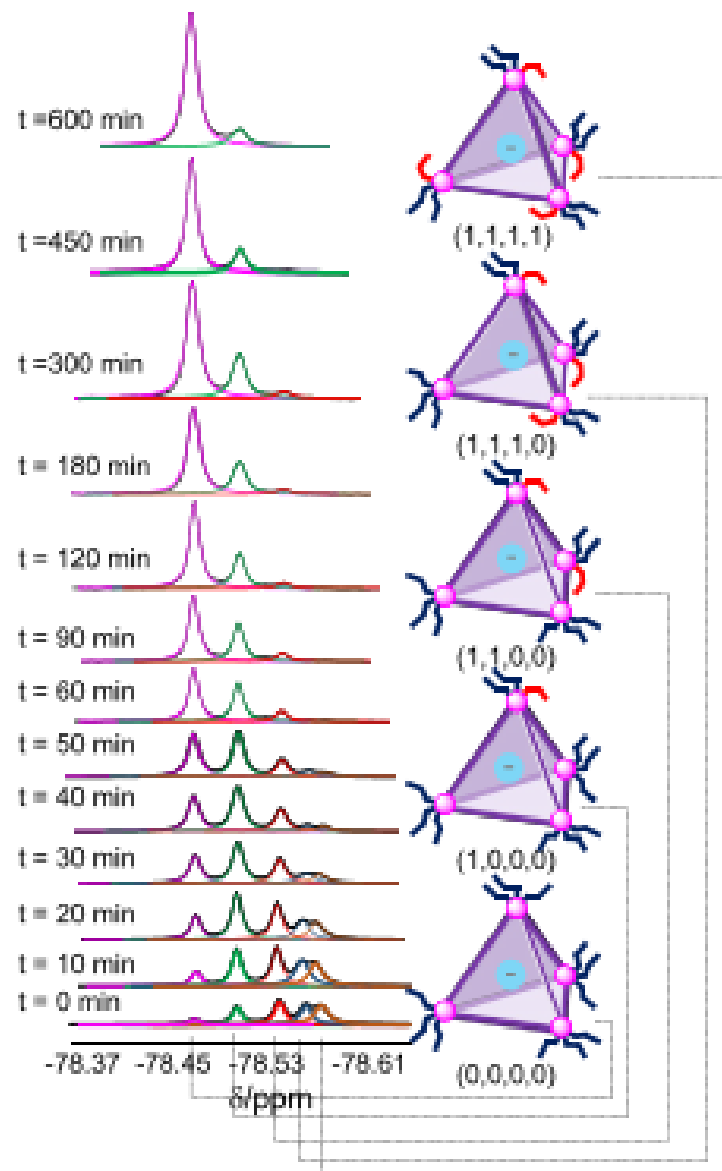
**Figure 4.**  $^{19}\text{F}$  NMR spectra ( $\text{CD}_3\text{CN}$ , 376 MHz, 298 K) show 43% encapsulated/57% released  $\text{Tf}_2\text{N}^-$  guest after UV irradiation for 10 min; the encapsulated  $\text{Tf}_2\text{N}^-$  is quantified by NMR integration with respect to hexafluorobenzene as an internal standard. The five distinct guest-encapsulating cages, incorporating 0–4 *cis*-azobenzene residues, gave rise to five  $^{19}\text{F}$  NMR signals, which were deconvoluted as shown.

						Total (%)
Population of <i>trans</i> diazo moieties (%)	2	7.3	10.8	7.5	6.7	34.3
Population of <i>cis</i> diazo moieties (%)	0%	0.7	2.2	2.5	3.3	8.7

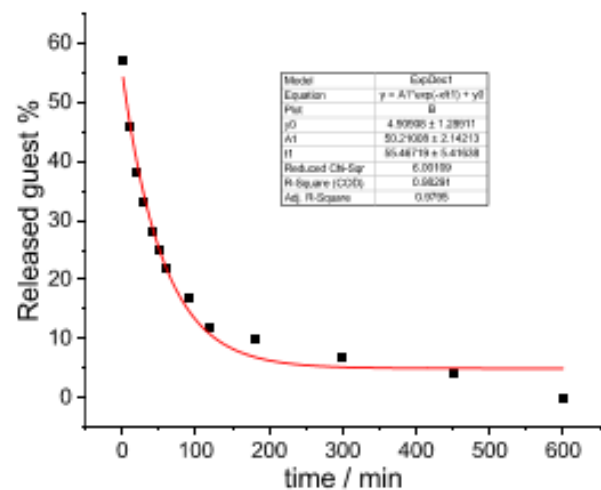
**Figure S34.** Population of *trans* (34.3%) and *cis* (8.7%) diazo moieties in the 43% of intact, triflimide-binding cages following UV irradiation. After UV irradiation, a total of six peaks appear in the  $^{19}\text{F}$  NMR, with five corresponding to bound  $\text{Tf}_2\text{N}^-$  (43%) and one corresponding to free  $\text{Tf}_2\text{N}^-$  (57% released from a cage, along with the unbound counter anions). This 43% / 57% split thus corresponds to the proportion of ligand bound up into cages, vs. free ligand corresponding to cages that have opened. From the 43% of ligand incorporated into cages, we can calculate the amount of isomerization (34.3% *trans* and 8.7% *cis*), but we were unable to quantify the degree of isomerization among the other 57%.



**Figure 5.** Re-formation of cages incorporating progressively more residues of *trans*-B during thermal relaxation while heating to 75 °C, derived from integration of  $^{19}\text{F}$  NMR spectra of the encapsulated triflimide guest (Supporting Information, Figure S30). The percentage population value shown is normalized to the maximum amount of anion that can be encapsulated.

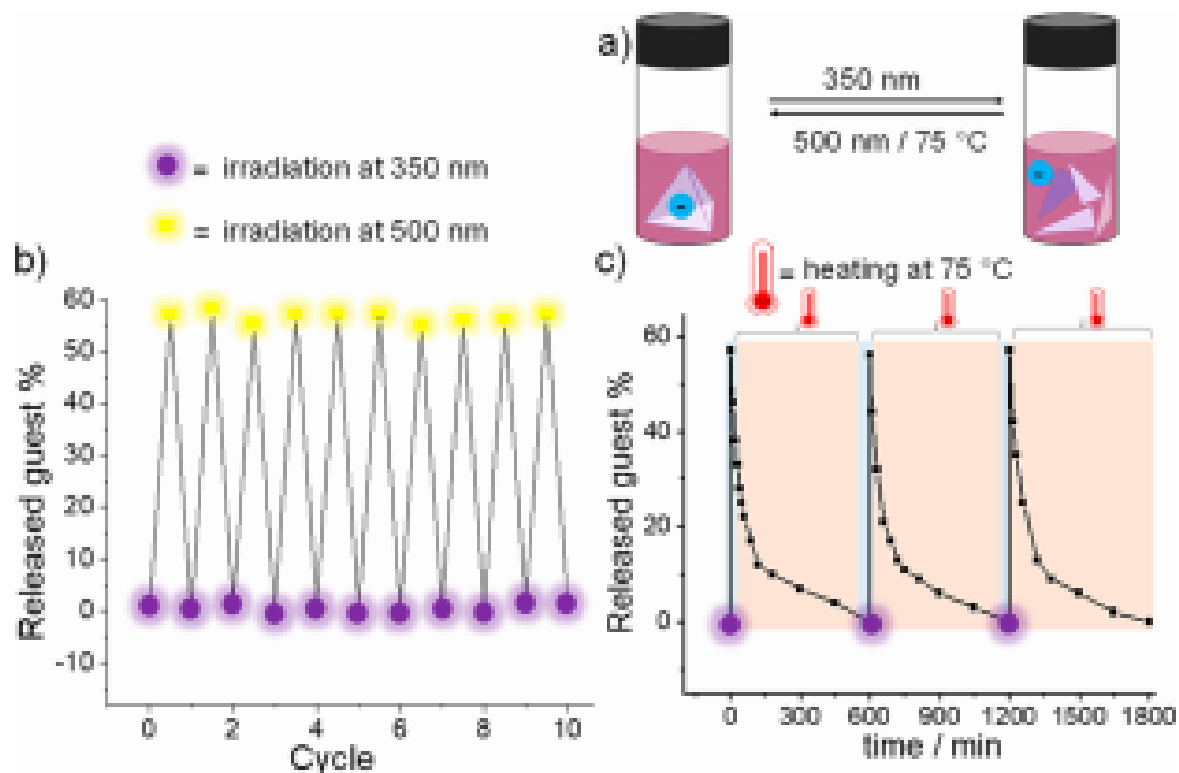


**Figure S30.**  $^{19}\text{F}$  NMR spectra (376 MHz,  $\text{CD}_3\text{CN}$ , 25 °C) of  $\text{Ti}_2\text{N-cis-trans-1}$  (39.6 mM) at the indicated times during 75 °C heating after 350 nm UV irradiation. Deconvolution of peak clusters with calculated Lorentzian peaks are color coded. The peak at -78.47 (magenta), -78.51 (green), -78.54 (red), -78.57 (blue), and -78.58 (brown) represent  $\text{Ti}_2\text{N}^-$  encapsulated within cages incorporating 0, 1, 2, 3, and 4 *cis* diazo moieties, respectively. The speciation profile in Figure 5 was extracted from this experiment.

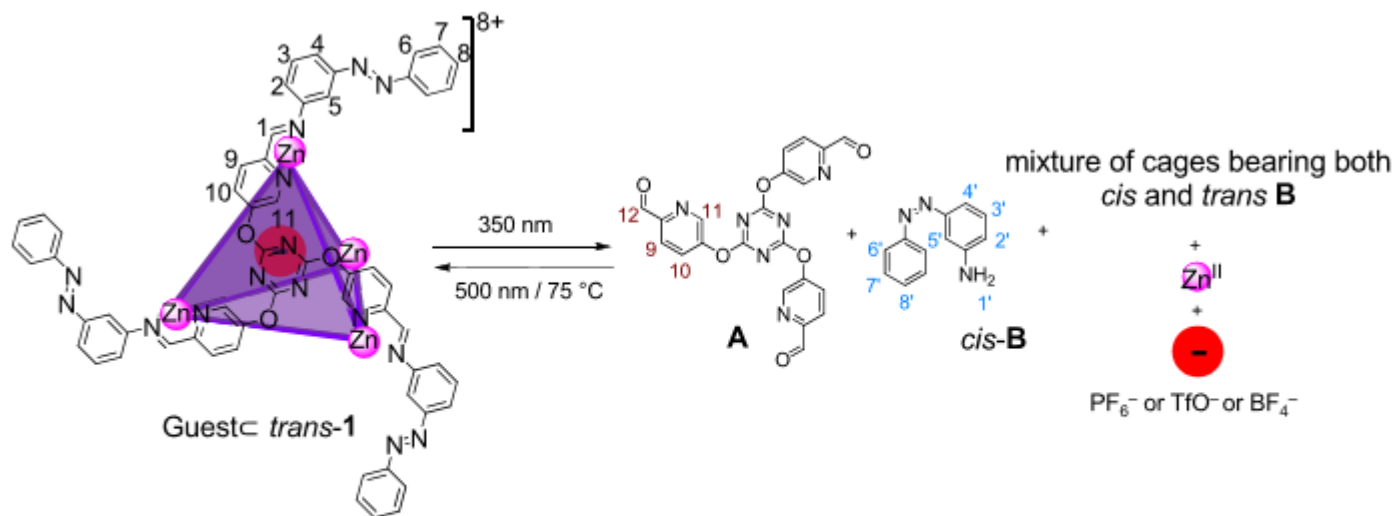


**Figure S32.** Kinetics for the guest uptake process during reverse photoisomerization towards *trans*-1 at 75 °C, monitored by integration of encapsulated  $Tf_2N^-$  with respect to hexafluorobenzene (Figure S27). The fitting following a 1<sup>st</sup> order kinetics (red line), revealing a half-life of  $t_{1/2} = 38$  min and rate constants of  $k = 3.50 \times 10^{-4} \text{ s}^{-1}$ .

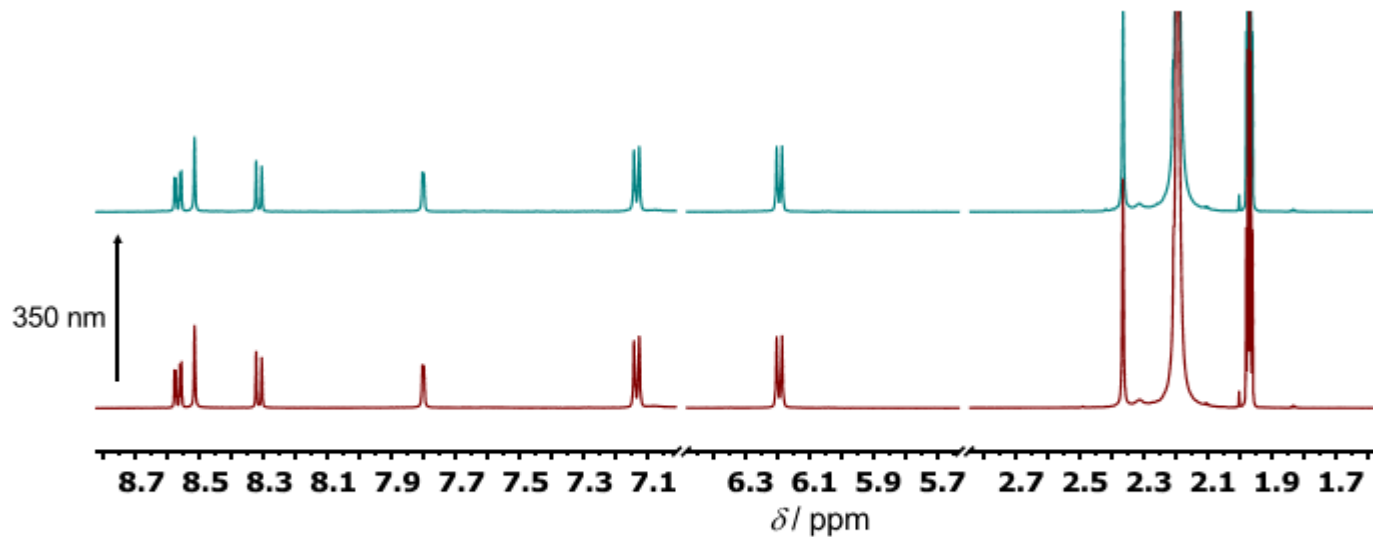




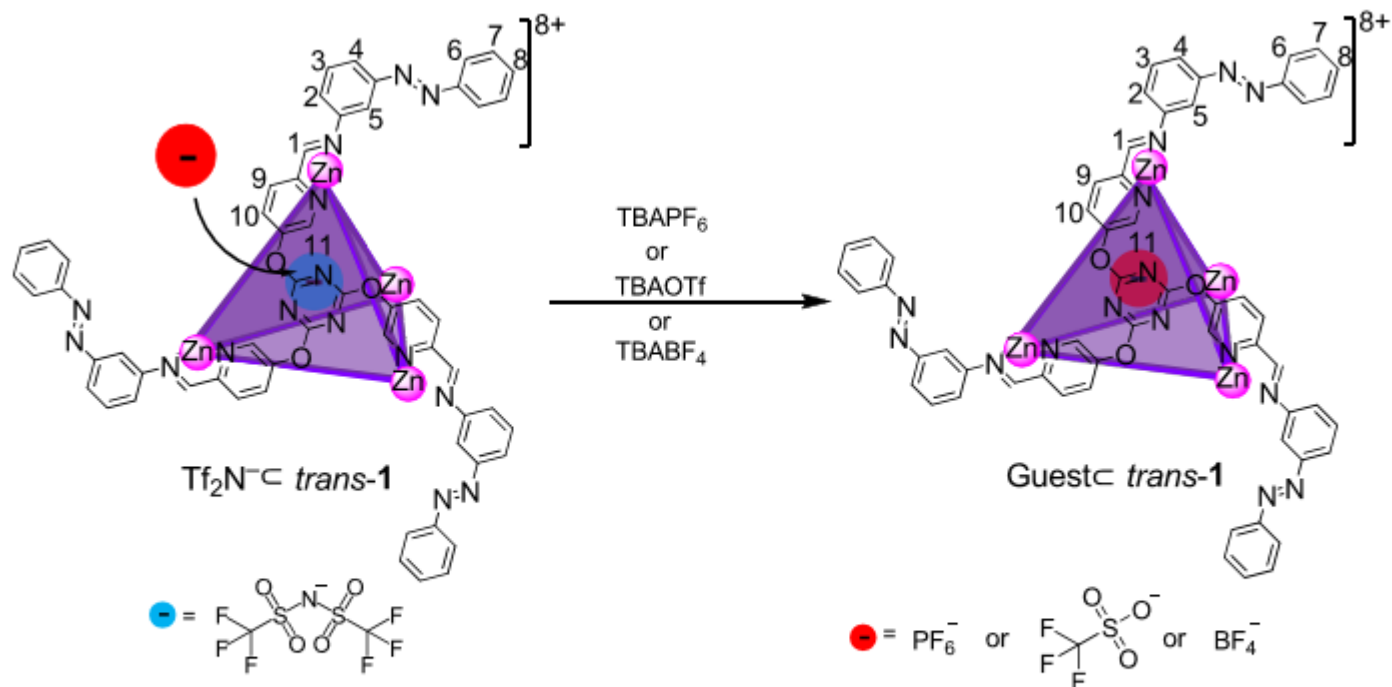
**Figure 6.** a) Cartoon representation of reversible release and uptake of  $Tf_2N^-$  guest by UV light and visible light or heat. b) Ten cycles of guest release driven by irradiation at 350 and 500 nm in  $CD_3CN$  in an alternating sequence. c) Three such cycles using light (350 nm) and temperature (75 °C) as stimuli; in both cases no evidence of fatigue was observed.



**Figure S59.** Disassembly and assembly of cage  $\text{Guest} \subset \text{trans-1}$  and subsequent guest ( $\text{PF}_6^-/\text{TfO}^-/\text{BF}_4^-$ ) release and uptake.



**Figure S76.** <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 25 °C) of a model cage (cage **2** with Zn<sup>II</sup> instead of Fe<sup>II</sup>) before and after UV light irradiation. No disassembly was observed, in contrast with cage *trans-1*.



**Figure S36.** Encapsulation of other anionic guests (PF<sub>6</sub><sup>-</sup>, TfO<sup>-</sup> and BF<sub>4</sub><sup>-</sup>) with subsequent release of Tf<sub>2</sub>N<sup>-</sup>.

Anion	Volume (Å <sup>3</sup> )	$K_{rel}$ ( $K_a^2/K_{\text{Tf}_2\text{N}^-}$ )
Tf <sub>2</sub> N <sup>-</sup>	131	-
PF <sub>6</sub> <sup>-</sup>	90	0.15
TfO <sup>-</sup>	86	0.11
BF <sub>4</sub> <sup>-</sup>	50	0.04

**Table S2.** Reported volumes<sup>2,3</sup> and relative binding affinities of the anions.