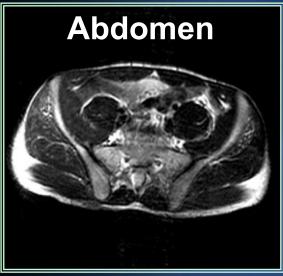
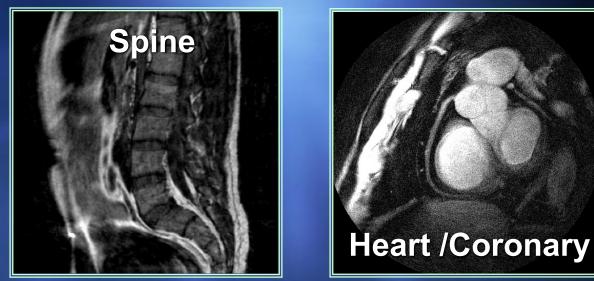
Magnetic Resonance Imaging MRI





Key resources:

- "Introduction to Medical Physics" S. Keevil et al CRC Press (2022)
 - Chapter 7
- ✓ Some papers and link in Moodle
- ✓ the following web site:

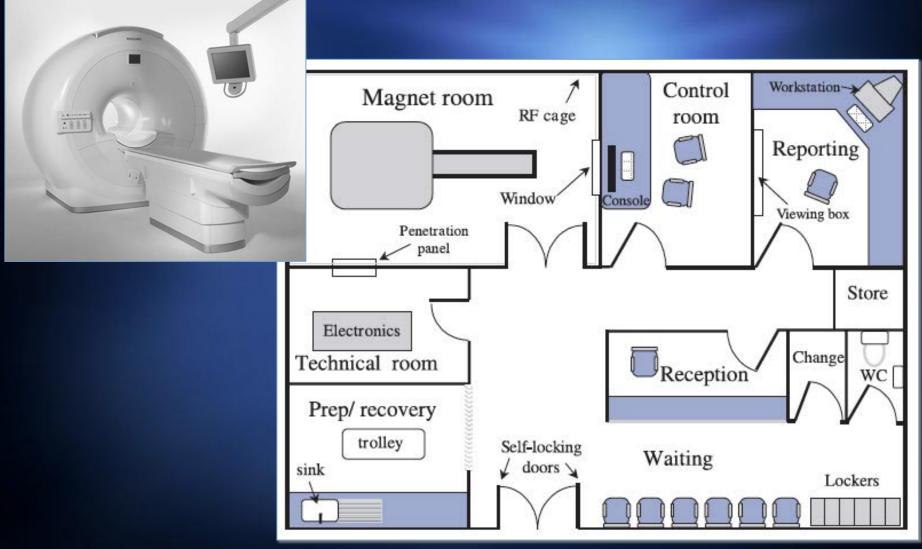
http://www.imaios.com/en/e-Courses/e-MRI



- Understanding the principles of nuclear magnetic resonance (NMR)
- The knowledge of the technology of the NMR imager
- ✓ To learn how to optimize a MRI exam



Welcome to the MR unit



https://www.youtube.com/watch?v=8m_RUVrxhew

Historical Introduction

 1930s and 1940s: The discovery of nuclear magnetic resonance (NMR) confirms predictions of the quantum theory

 Nobel Prizes: Isidor Rabi, Felix Bloch, Edward Purcell

 1971: Raymond Damadian discover differences between NMR signals obtained from normal tissue and from tumours

1973: Imaging methods by Paul Lauterbur and further developed by Peter Mansfield

MRI in medicine

 High spatial resolution imaging of both structure and function

Growing role in planning and guiding treatment and in the evaluation of patients' response to therapy

A tool for understanding of basic physiology and neuroscience

Much of this flexibility can be exploited by programming existing scanners to obtain images in new ways, rather than requiring expensive hardware upgrades



✓ Nuclear

- a phenomenon involving atomic nuclei
 *not all nuclei can be studied using NMR
- almost all medical MRI involves only the nucleus of the hydrogen atom
- MagneticResonance



MAGNETIC PISCUSSION

brun Towneh.



✓ Nuclear

- ✓ Magnetic
 - NMR involves the magnetic properties of nuclei
 - it occurs when nuclei are placed in a magnetic field





MAGNETIC PISCUSSION

brun Tourhel.



- Nuclear
 Magnetic
 Resonance
 - NMR occurs when energy is applied to the nuclei at a specific resonance frequency *nuclei absorb and then re-emit this energy *analysis of the resulting signal can reveal various properties of the material under study



MAGNETIC PISCUSSION

brun Toushel.



- NMR is fundamentally a quantum mechanical phenomenon
- For most purposes in medical MRI, it is sufficient to adopt a 'semi-classical' approach
 - in some situations we will have to refer back to the quantum mechanical model
 - in others we will have to accept that our semi-classical approach is an approximation that doesn't bear too much scrutiny!

Resonance

Resonance occurs when a system can efficiently transfer energy between two different storage modes

 such as kinetic energy and potential energy in the case of a pendulum

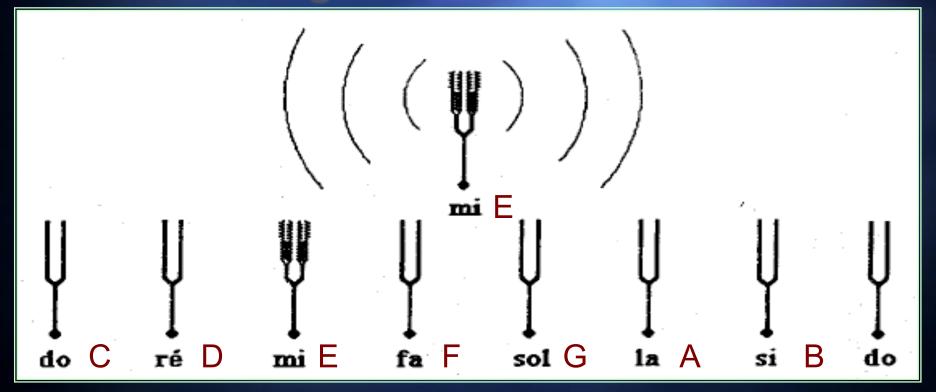


Swinging a child in a playground swing is an easy job because you are helped by its natural frequency.

But can you swing it at another frequency?

Resonance

Resonance occurs when a system can easily transfer energy between two or more different storage



https://www.youtube.com/watch?v=8xE_nT3QySo

Quantum mechanics

 many physical variables that appear to be continuous on the larger scale of everyday life are actually quantised when very small objects are considered
 atoms and nuclei

 quantised variables include energy, electric charge and <u>angular momentum</u>

Quantum mechanics

Angular momentum of atomic nucleus

$$\left| {{f J}}
ight| = \hbar \sqrt{I(I+1)}$$

✓ ħ = h/2π

h Planck's constant 6.63 × 10⁻³⁴ J s

- Ispin quantum number
 - a property of the nucleus
 - I can only take integer or half integer values (1/2, 1, 3/2, etc.)

which in turn restricts the angular momentum to certain values too

Quantum mechanics

The value of *I* depends on the number of protons and neutrons

Protons and neutrons have spin $I = \frac{1}{2}$

- protons pair off with each other in such a way that the spin of each pair of protons cancels out
- If there is an odd number of protons, the unpaired proton contributes *I* = 1/2

Neutrons pair off in a similar way

nuclei containing an even number of protons and an even number of neutrons have an overall spin I = 0

<u>Magnetic Nuclei</u>

A nucleus has magnetic properties if it has an odd number of protons and/or neutrons

- ¹H₁ (proton), ²H₁ (Deuterium; odd-odd)
- ⁷Li₃ (Litium; even-odd)
- ¹³C₆ (Carbonium; even-odd)
- ¹⁴N₇ (Nitrogen; odd-odd)
- ¹⁹F₉ (Fluorine)
- ²³Na₁₁ (Sodium)
- ³¹P₁₅ (Phosphor)
- ³⁵Cl₁₇ (Chlorine)
- ⁶³Cu₂₉ (Copper)
- ¹²⁷I₁₅₃ (lodine)

Nuclei with Nonzero Spin of Potential Biomedical Interest

Nucleus	Spin	Relative NMR Sensitivity	Natural Abundance of Isotope (%)	Abundance of Element <i>in vivo</i> (atomic %)
ΙΗ	1/2	1.00	99.98	62
¹³ C	1/2	0.0159	1.11	12
¹⁴ N	1	0.00101	99.63	1.1
¹⁵ N	1/2	0.00104	0.37	1.1
¹⁷ O	5/2	0.02291	0.04	24
¹⁹ F	1/2	0.83	100.00	0.0012
²³ Na	3/2	0.0925	100.00	0.037
³¹ P	1/2	0.0663	100.00	0.22

The Relative NMR Sensitivity indicates the amount of signal that would be obtained from numbers of each nucleus, relative to that of the ¹H nucleus



More complex aspects of nuclear structure mean that some nuclei have higher spin quantum numbers
 specifically "orbital angular momentum"

The hydrogen nucleus, and most of the other nuclei that are of interest in biomedical applications of NMR, have I = 1/2

¹H images

¹H nucleus combines the highest NMR sensitivity with natural abundance close to 100% and very high abundance in the body

¹H nuclei are predominantly found in water molecules and in lipids

other NMR-visible nuclei

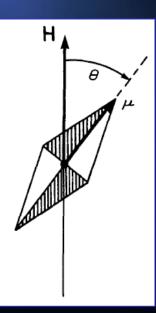
specific applications to investigate other NMR-visible nuclei

- spectroscopic studies of energy metabolism using ³¹P
- molecular imaging using ¹⁹F-containing compounds
 - introduced into the body artificially

Basic Concepts Hydrogen Protons

- Magnetic Resonance Imaging (MRI) is based on ¹H protons
- Very qualitative introduction
 - a solid ball spinning on an axis, with the rate at which it is spinning determined by the quantum number I
 - a positively charged spinning proton acts like a tiny magnet

a crude representation of a quantum mechanical object that is at once a wave and a particle !!!



The physics of proton NMR

μ=γJ

μ magnetic moment
 γ gyromagnetic ratio of ¹H
 γ = 42.57 MHz T⁻¹
 γ = 2.68 × 10⁸ rad s⁻¹ T⁻¹
 J angular momentum

Angular momentum of an atomic nucleus is due to internal structures of p and n (quarks)

https://www.imaios.com/en/e-mri/nmr/nuclear-spin

 $\left| {f J}
ight| = \hbar \sqrt{I(I+1)}$

Effect of the Static Magnetic Field

н

When a magnetic field B₀ is applied to a sample containing nuclear dipoles the dipoles align themselves in specific orientations
 with respect to the applied field

- The orientations depend on m_l
 - *m_l* the magnetic quantum number
 - *m_l* can take a range of values in integer steps from –*l* to *l*

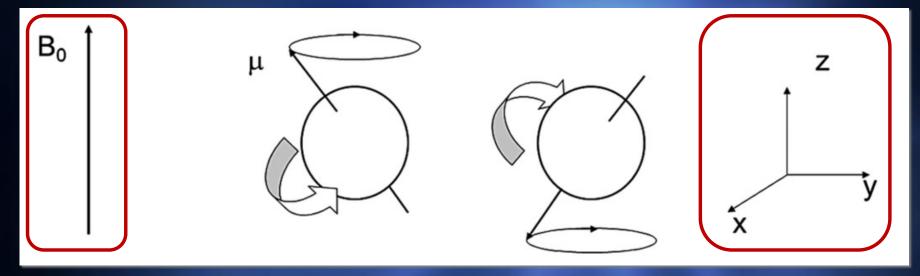
I spin quantum number

The physics of proton NMR

✓ ¹*H* spin quantum number: *I* = 1/2
 ✓ ¹*H* magnetic quantum number: *m_l* = ±1/2
 ✓ *μ* can adopt 2 possible orientations
 ✓ *μ_z* is the component lying along the direction of the static magnetic field B₀

$$\mu_z=\gamma \hbar m_I=\pm rac{1}{2}\gamma \hbar$$

Spin-up and spin-down

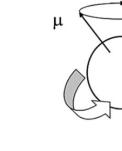


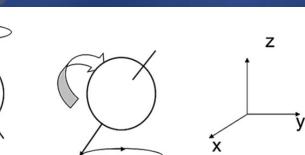
 μ_z the component lying along the direction of the static magnetic field B_0

- ✓ 2 orientations corresponding to $m_l = \pm 1/2$
- $\checkmark \mu$ lies at an angle of 54.7° to B₀
 - the xy-plane is known as the transverse plane

Larmor frequency According to classical electromagnetism a magnetic moment lying at an angle to a magnetic field experiences a torque • turning force This causes the magnetic moment to rotate, or precess, around the direction of the B_0 at the Larmor frequency ω_0

B₀

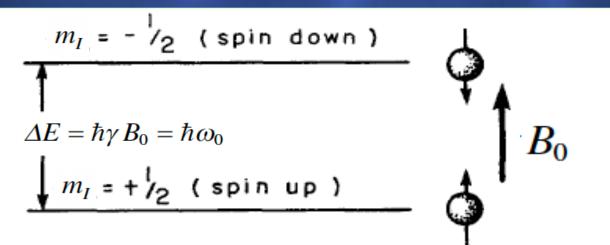




$$\omega_0 = \gamma B_0$$

https://www.imaios.com/en/e-mri/nmr/precession-and-larmor-frequen

The energy of a magnetic moment in a magnetic field **Converse of the nuclei** $E = \pm \frac{1}{2} \hbar \gamma B_0$ I=1/2 • Larmor frequency $\omega_0 = \gamma B_0$ Energy difference between the energy of nuclei in the 2 orientations



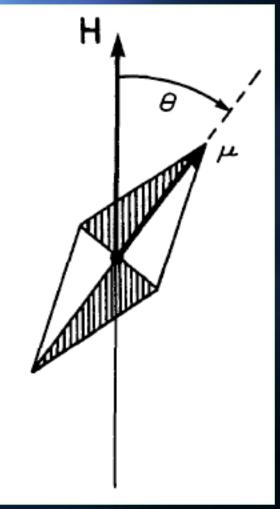
The physics of the compass needle

The torque in the magnetic needle = $\mu \times H$

Vector product

potential energy -μ·H= -μ H cos θ

- Scalar product
- Minimum: $\theta = 0$, cos $\theta = 1$
- Maximum: θ=180, cos θ=-1



Effect of the Static Magnetic Field Nature prefers the lowest energy state more nuclei in the lower energy level than the higher one It the difference between the populations of the 2 energy levels

$$rac{N \ \uparrow -N \ \downarrow}{N} pprox rac{\hbar \gamma B_0}{2kT}$$

k is Boltzmann's constant
 \$1.38 × 10-23 J K⁻¹

Effect of the Static Magnetic Field the difference in population amounts to only a few nuclei per million A few ppm It the small excess of nuclei oriented along the B₀ direction gives rise to the B. NMR signal • B₀ ~ 1-3 T ħγſ

https://www.imaios.com/en/e-mri/nmr/net-magnetization

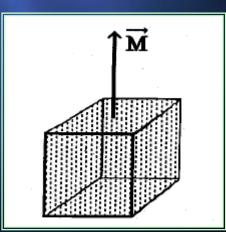
The Bulk Magnetization

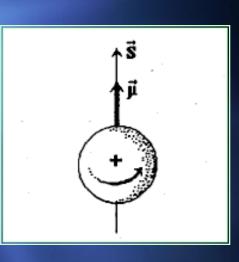
$$\vec{M} = \sum \vec{\mu}$$

- M is the net magnetization (magnetic moment) of a sample placed in a magnetic field
 - typically >> 10¹⁸ nuclei

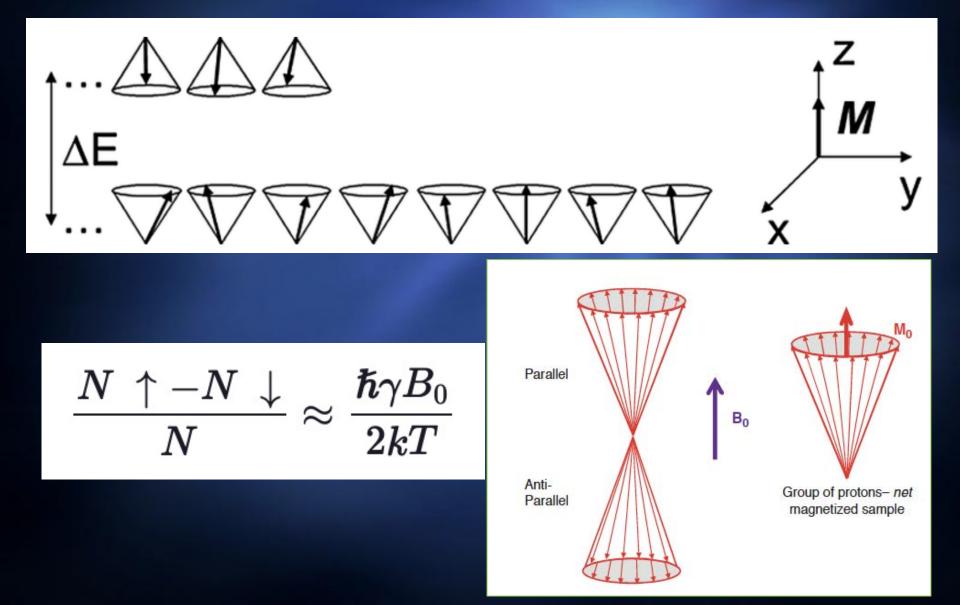
\checkmark M lies along the B₀ direction







The Bulk Magnetization



Effect of the **Radiofrequency Field** The resonance of NMR Additional magnetic fields, which vary in time are used to manipulate the direction and size of M • M bulk magnetisation vector • to collect a signal from the spin system classical electromagnetism is used quantum-mechanical description is briefly considered

Classical description of a spin in a magnetic field R.L. Dixon and K.E. Ekstrand Medical Physics vol 9 pp 807-818 1982

When a force is exerted on a spinning object (S), it tends to move at right angles to the force

$$\frac{d\vec{S}}{dt} = \vec{\mu} \times \vec{H}$$
$$\vec{\mu} = \gamma \vec{S}$$
$$\frac{d\vec{\mu}}{dt} = \vec{\mu} \times \gamma \vec{H}$$

b the magnitude of µ remains constant

since dµ is perpendicular to both µ e H (or B)

The motion is a precession of µ about H (or B)

> Angular (Larmor) frequency $\omega = \gamma H = \gamma B$

http://www.imaios.com/en/e-Courses/e-MRI/NMR/Precession-and-Larmor-frequency

Classical description of a spin in a magnetic field

When a force is exerted on a spinning object, it tends to move at right angles to the force

$$\frac{d\vec{S}}{dt} = \vec{\mu} \times \vec{H}$$
$$\vec{\mu} = \gamma \vec{S}$$
$$\frac{d\vec{\mu}}{dt} = \vec{\mu} \times \gamma \vec{H}$$

A group of isolated classical spin precesses ad infinitum

B and **H** fields

Both B and H are called magnetic field B is measured in Tesla (T) or Gauss (G) • 1T=10000G In vacuum H is proportional to B $H \equiv -M$ ✓In medium • M magnetization vector field

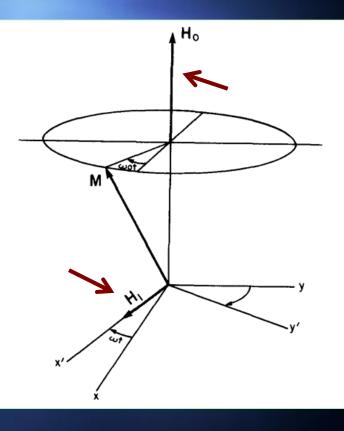
μ₀ permeability of free space

Effect of the **Radiofrequency Field** $\checkmark B_1$ is orthogonal to B_0 • it lies in the transverse plane \checkmark It is much weaker than B_0 of the order of μT It is applied for a short period of time typically hundreds of μs

 magnetic moments experience a torque in the presence of B₀ which causes them to precess

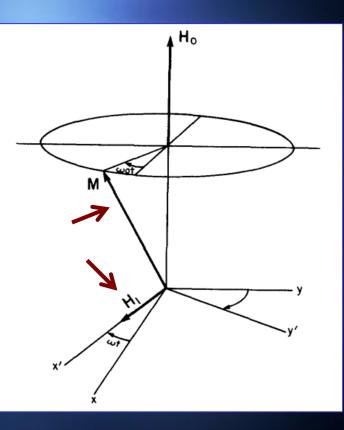
In the same way, there is precession of M around the direction of B₁

 causing the magnetisation vector M to tip (or nutate) away from the z-axis



In order for nutation of M to happen, the B₁ field must be kept 'in step' with the precessing nuclei

 the B₁ field must itself be rotating in the transverse plane at the Larmor frequency ω₀



Larmor frequency in MRI

 Larmor frequency lies in the same frequency range as the RF of the electromagnetic spectrum

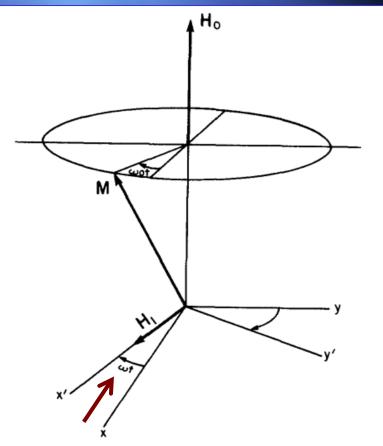
- 10–100 MHz
- B₁ is often termed the RF magnetic field
- B₁ can be generated by using 2 RF coils aligned along the x- and y-axes
 - generating sinusoidal magnetic fields 90° out of phase and so add up to produce a circularly polarised magnetic field

Motion of M in an RF field
This is the resonance part of NMR
Inutation only occurs if B₁ is applied at the Larmor frequency
This fact allows to choose which nuclear species to interrogate

- B₀ generates bulk magnetisation from all nuclei that have nonzero spin
 ¹H, ¹³C, ³¹P
- The choice of the frequency of B₁ allows to select which of these to study

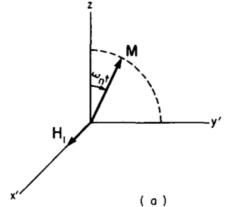
 $\sqrt{if \omega} = \omega_0 B_1$ stays in constant position relative to M ω frequency of the RF pulse Thus the small torque of B₁ on M over many precessions has significant effect $\omega = \omega_0$ is the resonance condition

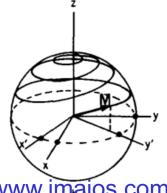
$$\frac{d\vec{M}}{dt} = \vec{M} \times \gamma(\vec{H}_0 + \vec{H}_1(t))$$



http://www.imaios.com/en/e-Courses/e-MRI/NMR/Excitation

Motion of M in an RF field **Once** B_1 is applied: M tips away from the z-axis because of precession around B₁ \checkmark it will also precess around B_0 The overall motion is quite complicated: the magnetisation vector following an expanding helical trajectory



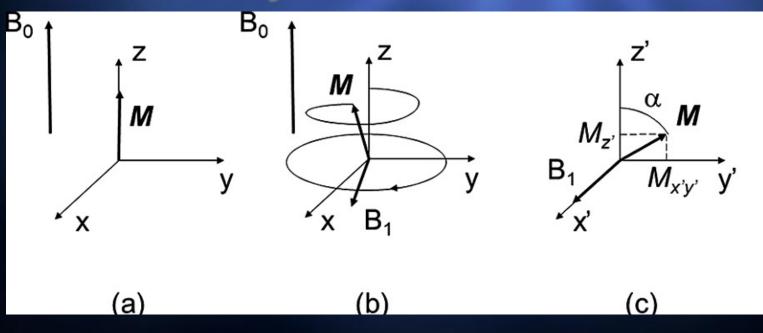


www.imaios.com/en/e-Courses/e-MRI/NMR/Excitation

a) Static position

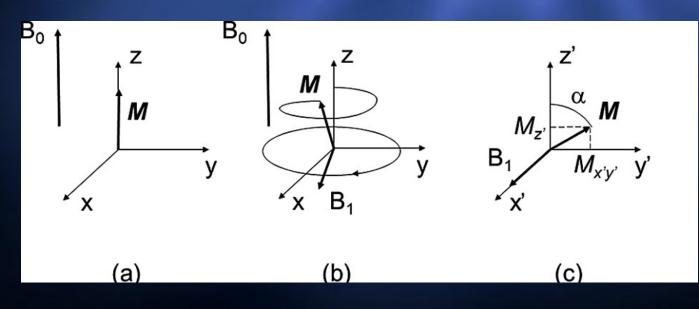
 b) the magnetisation vector M following an expanding helical trajectory

laboratory frame of reference



The rotating frame

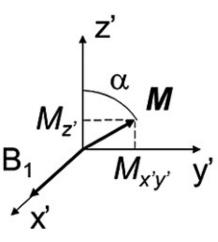
- c) rotating frame of reference moving at the Larmor frequency (ω₀)
 - the motion of M is simplified to nutation from the z'-axis towards the x'y' plane
 - x'y'z' the rotating frame

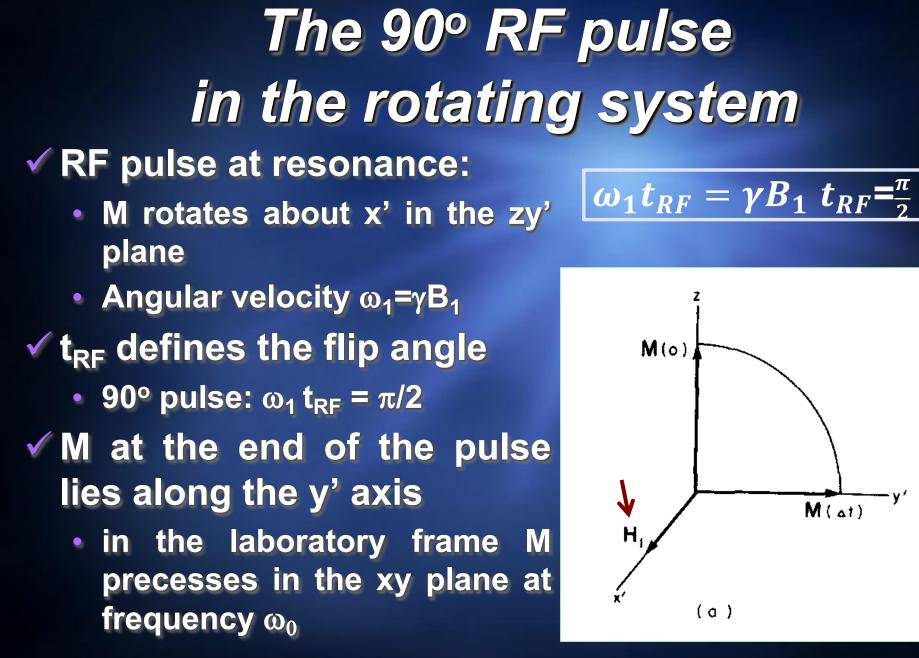


The flip angle α

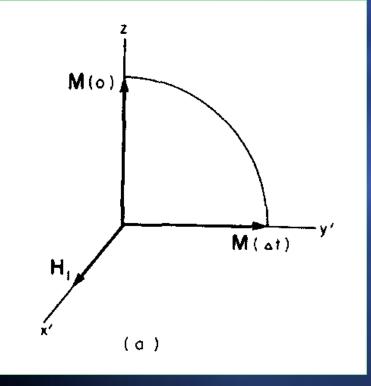
In the rotating frame of reference It the motion of M is simplified to nutation from the z'-axis towards the x'y' plane \checkmark nutation continues for as long as the B₁ field is present (t_{RF}) \checkmark The nutation angle α Or flip angle

$$\alpha = \gamma B_1 t_{RF}$$





The 90° RF pulse: how long is t_{RF} ?



H₁ =10⁻³ T ; protons t_{RF} ?

v' =42.5 kHz t_{RF} = 1/(4ν') ~ 6 μsec

How long a 180° pulse ?

v =1/T T period of the motion

Larmor frequency

Nucleus	γ_n (10 ⁶ rad s ⁻¹ T ⁻¹)	$\gamma_n/(2\pi)$ (MHz T ⁻¹)
¹ H	267.513	42.576
² H	41.065	6.536
³ He	203.789	32.434
⁷ Li	103.962	16.546
¹³ C	67.262	10.705
¹⁴ N	19.331	3.077
¹⁵ N	-27.116	-4.316
¹⁷ O	36.264	5.772
¹⁹ F	251.662	40.052
²³ Na	70.761	11.262
²⁷ AI	69.763	11.103
²⁹ Si	-53.190	-8.465
³¹ P	108.291	17.235
⁵⁷ Fe	8.681	1.382
⁶³ Cu	71.118	11.319
⁶⁷ Zn	16.767	2.669
¹²⁹ Xe	73.997	11.777

 ✓ Angular Larmor frequency (rad/s)
 ω=γΗ

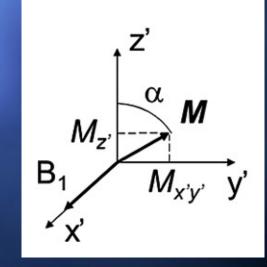
 Larmor frequency (MHz)
 ν=γH/2π

Motion of M in an RF field The net effect of the RF pulse is to reduce the longitudinal magnetisation (M_z) while generating transverse magnetisation (M_{xy})

$$M_{z'} = |\mathbf{M}| \cos\alpha = |\mathbf{M}| \cos(\gamma B_1 t_{RF})$$

$$M_{y'} = |\mathbf{M}| \sin \alpha = |\mathbf{M}| \sin \left(\gamma B_1 t_{RF}\right)$$

$$M_{x'}=0$$



assuming B₁ along the x'-axis in the rotating frame, so that the transverse magnetisation lies along the y'-axis <u>https://www.imaios.com/en/e-mri/nmr/excitation</u>

Quantum mechanics ?

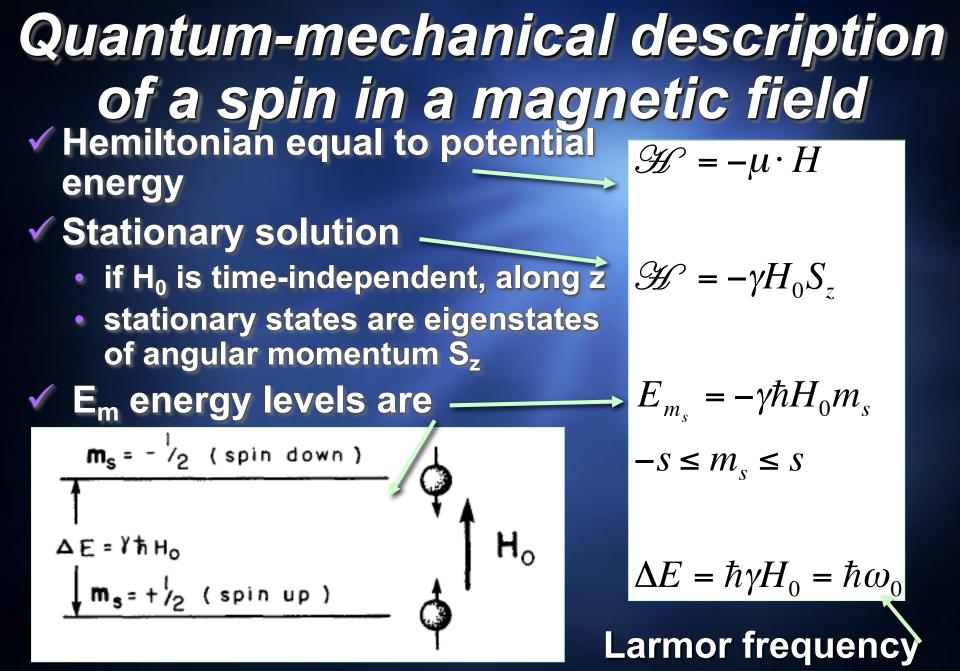
- ✓ an electromagnetic field of frequency ω_0 carries energy E = $\hbar \omega_0$
- ✓ this is the energy gap between up and down energy levels
- μ in the lower energy level are able to absorb energy from the RF field
 - If the RF frequency matches the Larmor frequency

The effect is to reduce the difference between the populations of the two energy levels, and hence the size of the longitudinal magnetisation

Quantum mechanics ?

The effect of the RF pulse is to reduce the difference between the populations of the two energy levels, and hence the size of the longitudinal magnetisation

Less obviously, the excited nuclei are made to precess in phase with each other, so that net magnetisation M_{xy} is generated in the transverse plane



http://www.imaios.com/en/e-Courses/e-MRI/NMR/nuclear-spin

Quantum-mechanical descript		
of a spin in a magnetic field		
✓ The transition from lower to upper state	is	
made by energy absorption		
Radio frequency energy		
E=(hω ₀)/2π=hν ₀	ics of proton kstrans hysics 1982	
• proton v_0 (MHz)=42.58 H ₀ (T) • $v_0 = \omega_0/2\pi$		
The magnetic dipole transitions a	proton NMR ns s 1982 e	
influenced only by the magnetic compone	nt	
H' of the RF field, and <u>only by x and</u>	y	
<u>components</u> $\langle m'_{s} \vec{\mu} \cdot H' m_{s} \rangle$		
 Since transition matrix element is no zero only f 	for	
μ_x and μ_y		

Quantum-mechanical description of a spin in a magnetic field \checkmark RF irradiation at v₀ can cause de-excitation Stimulated emission of a photon ω_0 obtained from the Hamiltonian is the same as classical Larmor frequency $\langle \mu_x \rangle = \langle \mu_y \rangle = 0$ but the state of the quantum- $\langle \mu_z \rangle = \pm \frac{1}{2} \gamma \hbar$ mechanical spin time-independent • No precession ! A wave function which is an equal 1 mixture of spin-up and down states the expectation values are

• A precession of <µ> in the xy plane !

$$\langle \mu_x \rangle = \frac{1}{2} \gamma \hbar \cos(\omega_0 t)$$
$$\langle \mu_y \rangle = -\frac{1}{2} \gamma \hbar \sin(\omega_0 t)$$
$$\langle \mu_z \rangle = 0$$

http://www.imaios.com/en/e-Courses/e-MRI/NMR/Net-magnetization

Quantum-mechanical description of a spin in a magnetic field

✓ The expectation value of the magnetic moment of an ensemble of spins _____d

- [ℋ,μ] is the commutator of the Hemiltonian
- It may be shown
 - Commutation properties of angular momentum operators
- For an ensemble of N noninteracting spin with an associated magnetic moment

$$M = N \langle \mu \rangle$$

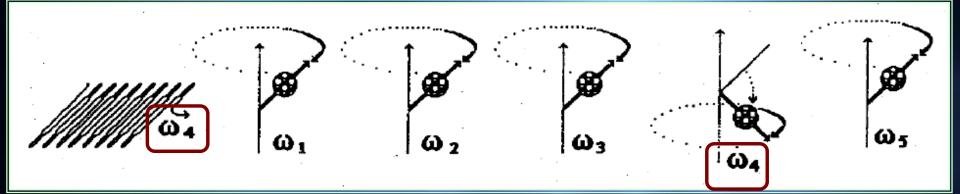
- M the observable quantity
- M follows the classical equation!
 - Individual spins undergo complicated behavior

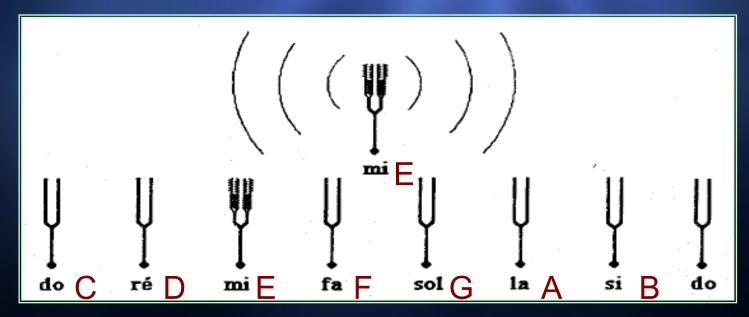
$$\frac{d}{dt}\langle\mu\rangle = \frac{i}{\hbar}\langle\left[\mathscr{H},\mu\right]\rangle$$

$$\frac{d}{dt} \langle \mu \rangle = \langle \mu \rangle \times \gamma H$$

$$\frac{d}{dt}M = M \times \gamma H$$

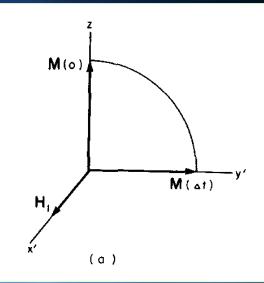
Effect of the Radiofrequency Field





homework

A hydrogen sample is at equilibrium in a 1.5 Tesla magnetic field.



A constant B₁ field of 2.34x10⁻⁴ T is applied along the +x'-axis for 25 μs

What is the direction of the net magnetization vector after the B₁ field is turned off?

MRI: Basic Physics & a Brief History a video

https://www.youtube.com/watch?v=djAxjtN_7VE