





Università di Trieste LAUREA MAGISTRALE IN GEOSCIENZE SM62 Percorso Esplorazione Geologica

Anno accademico 2023 - 2024

Geologia Marina 953SM

Parte IV

Modulo 4.4/5
Identificatori di movimento di fluidi: Idrati del metano

Docente **A. Camerlenghi**







Outline

Review of main mechanisms of fluid flow:

- Mud diapirs and mud volcanoes
- Gas chimneys
- Pockmarks
- Seafloor vents in general
- Polygonal fault systems
- Diagenetic fronts
- Gas hydrates







NATURAL GAS HYDRATES ARE CONSIDERED AS:

- Natural controllers of climatic change
- Geo-hazards
- Potential energy source
- Controllers of deep sea eco-systems
- Potential safe and effective way of methane transport
- Problem in pipelines
- Possible cause of ship and airplane disappearance in the Bermuda triangle







METHANE

Is the simplest saturated hydrocarbon gas (alkane):

CH₄

Other common hydrocarbon gases are:

ethane (C_2H_6) , propane (C_3H_8) butane (C_4H_{10})

When the number of carbon atoms exceeds 5, the name of gases is from Greek numbers: **pentane**, **hexane**, **heptane**...

The general formula of hydrocarbon gases is:







Human-related sources

- Landfills (waste disposal)
- Natural gas and petroleum systems (natural gas production)
- Coal mining
- Livestock enteric fermentation
- Livestock manure management
- Wastewater treatment
- Rice cultivation

Natural sources

- Wetlands
- Termites
- Oceans (anaerobic digestion in marine zooplankton and fish, and also from methanogenisis in sediment)
- Hydrates
- Juvenile





HYDRATES OF NATURAL GAS

(discovered in 1810)

- Gas hydrates are a solid phase composed of water and low-molecular weight gases (predominantly methane)
- Water molecules form an ordered crystalline solid matrix which includes gas molecules with weak electro-static bounds
- The state of saturation of hydrates is a function of pressure and temperature. They form under conditions of low temperature, high pressure, and adeguate gas concentration
- The ion content of water and the type of gas (methane, propane, pentane...) affect the phase diagram of the gas hydrate



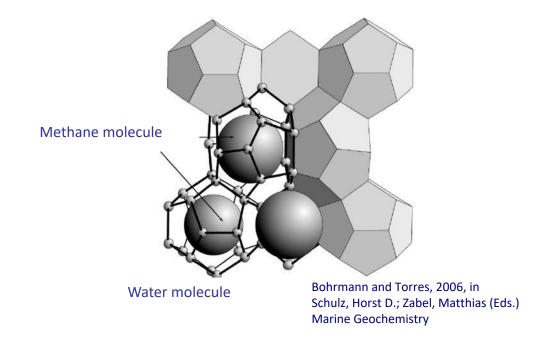


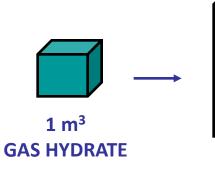


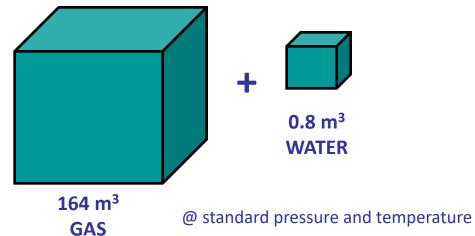
Water molecule



Hydrate crystal



















RCOM - University of Bremen



Courtesy:

Carlo Giavarini, Filippo Maccioni and Maria Laura Santarelli

Università di Roma "La Sapienza" Dipartimento di Ingegneria Chimica.

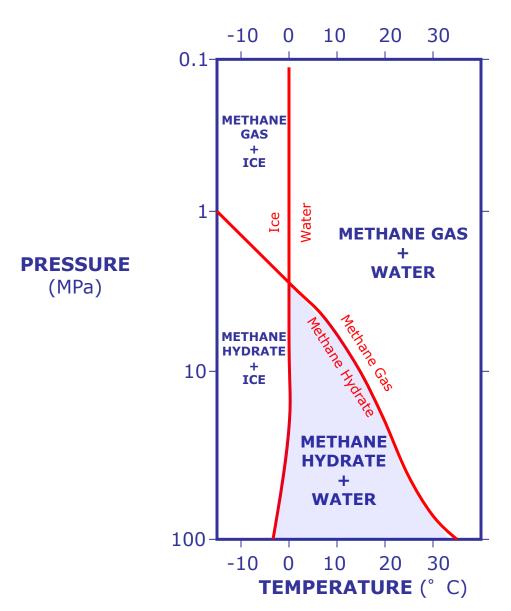








PHASE DIAGRAM OF METHANE HYDRATES











Hydrate formation in arctic gas pipelines reduces or blocks methane flow. Chemical additives are

necessary to prevent hydrate formation.





Alaska Highway Pipeline TransCanada







LIQUID NATURAL GAS TRANSPORT IS EXPENSIVE AND DANGEROUS



Methane is condensed into a liquid at almost atmospheric pressure (Maximum Transport Pressure set around 25 kPa) by cooling it to approximately -163° C.

LNG is transported by specially designed **cryogenic sea vessels** and cryogenic road tankers, and stored in specially designed tanks.



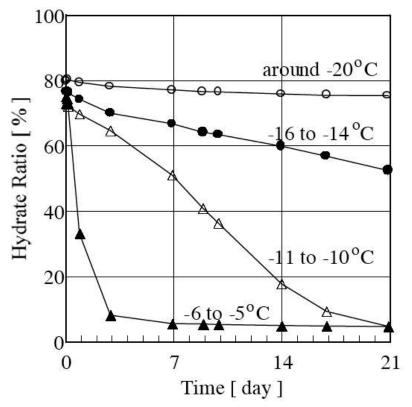






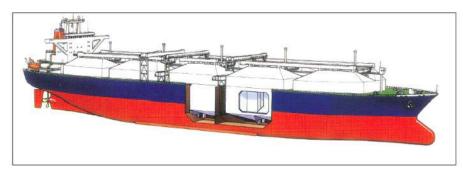
self-preservation

At 1 Atm, CH₄ Hydrates are stable below -20° C.



Self-preservation is thought to be caused by a layer of ice from dissociated hydrate; this layer coats the hydrate and seals it from further dissociation.





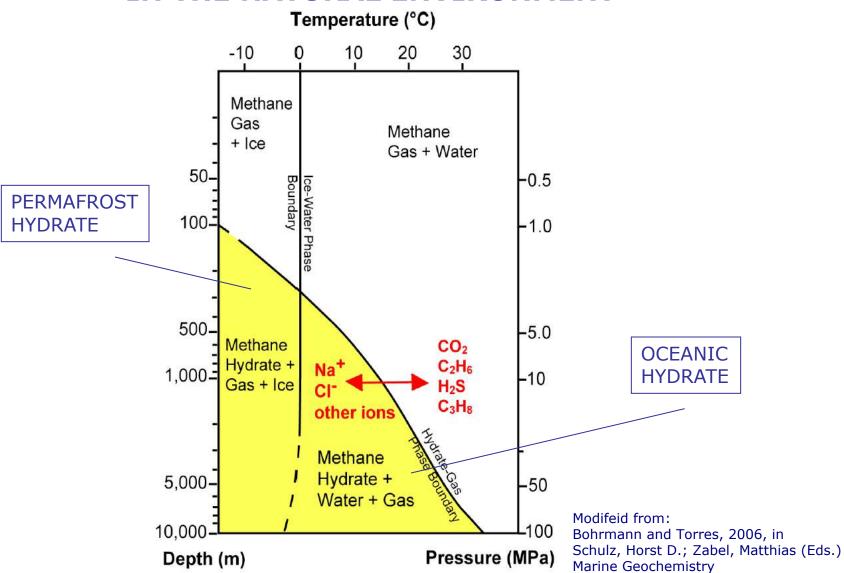
Possible ship specifically designed to carry natural gas **hydrate pellets**@ **-20° C and atmospheric pressure**(National Maritime Research Institute, Mitsui Shipbuilding & Engineering, and Osaka University)







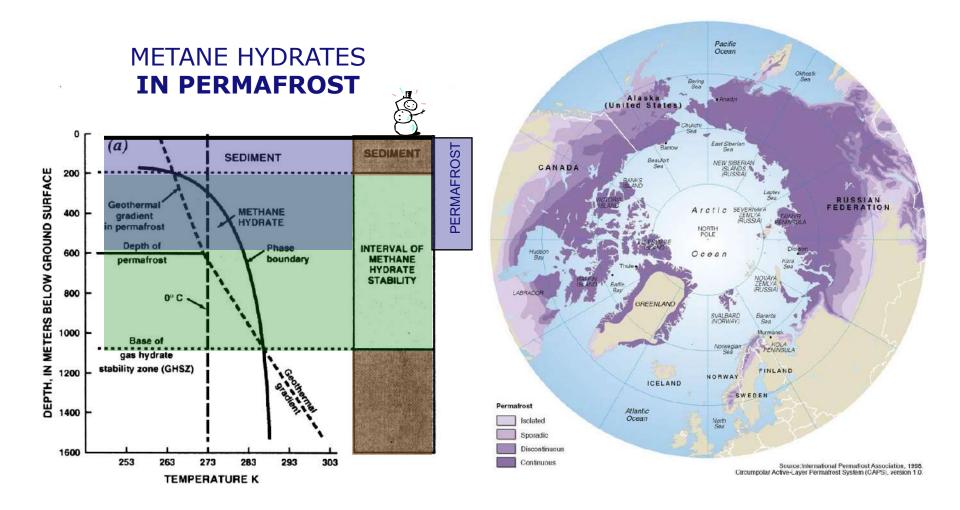
PHASE DIAGRAM OF METHANE HYDRATES IN THE NATURAL ENVIRONMENT











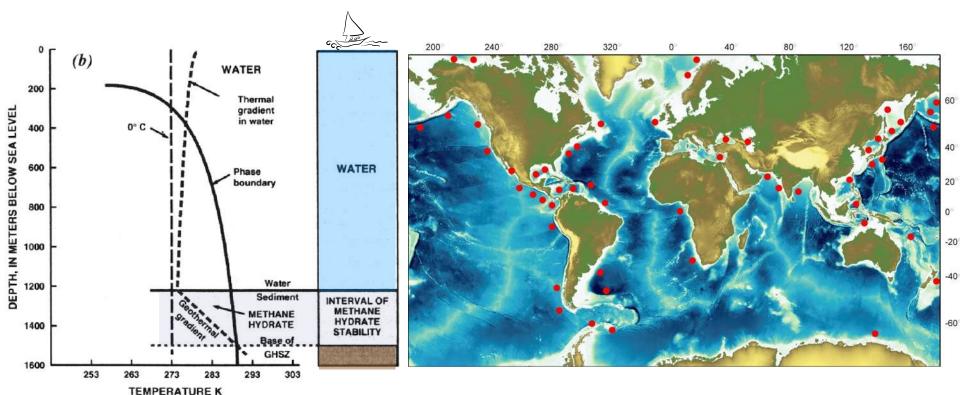
Permafrost is defined as soil, sediment or rock whose temperature is continuosly below 0° C for at least two years







METHANE HYDRATES IN OCEANIC SEDIMENTS

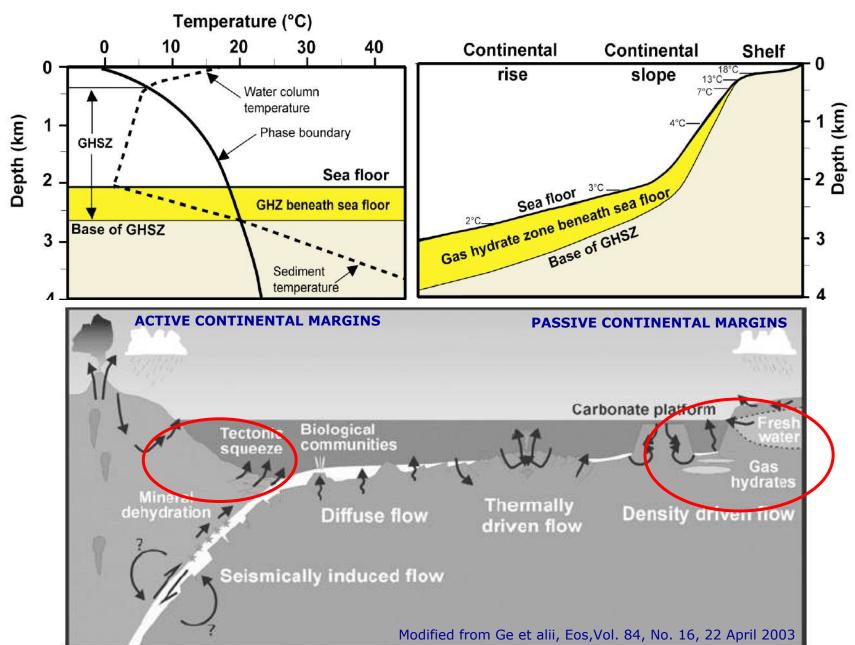


Natural gas hydrates are distributed along the ocean margins world wide within the Gas Hydrate Stability Zone (GHSZ), that ranges from the seafloor down to a maximum of 700-800 m



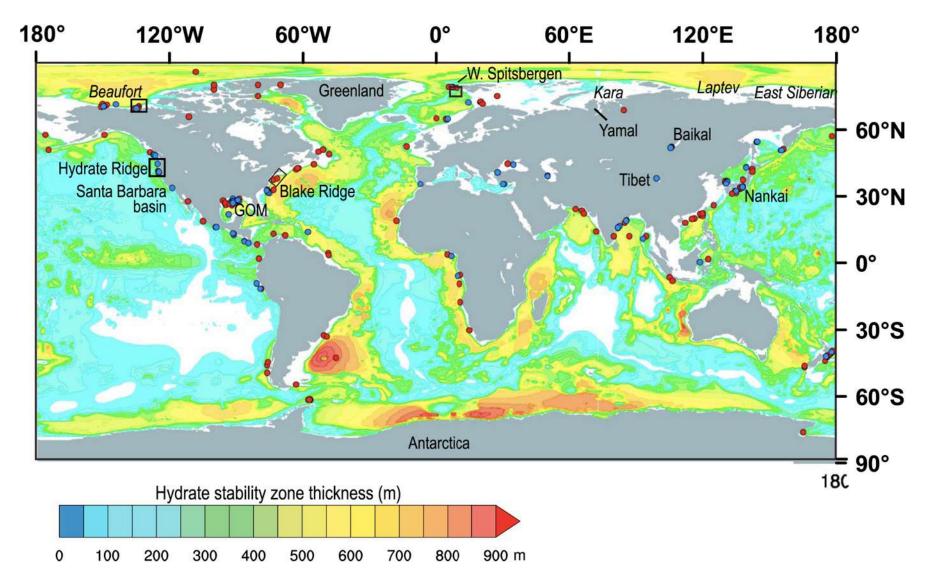












Ruppel, C. D., and J. D. Kessler (2017), The interaction of climate change and methane hydrates, Rev. Geophys., 55, 126-168, doi:10.1002/2016RG000534.

































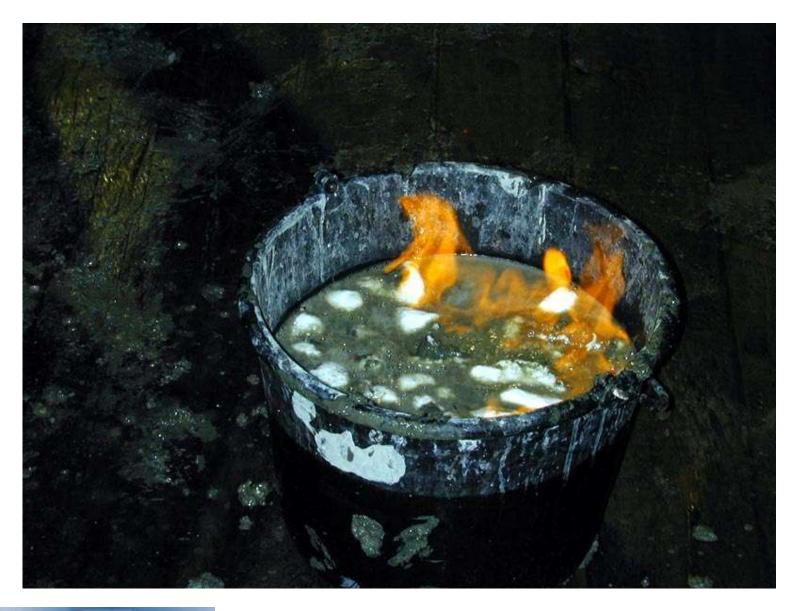






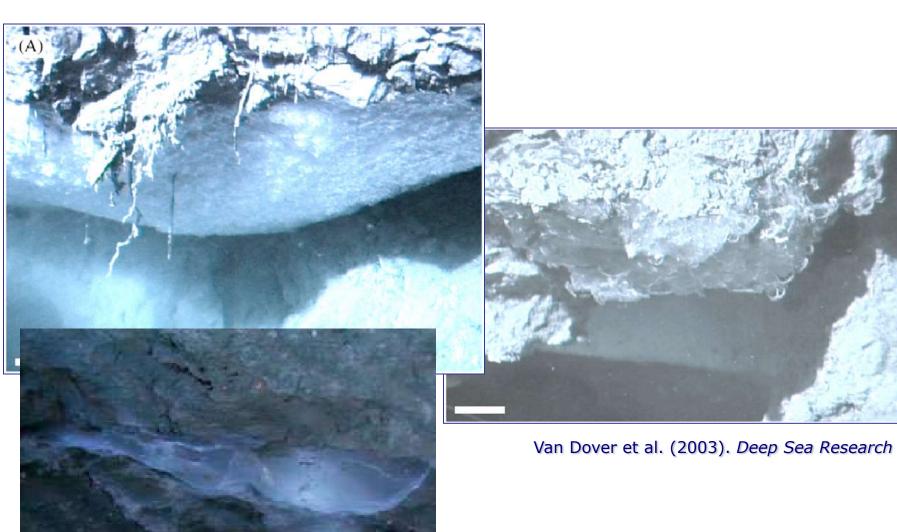


















METHODS OF DETECTION OF NATURAL GAS HYDRATES

• SEISMICS

(reflection, multichannel, OBS)

DRILLING

(essentially through scientific drilling)

GEOCHEMICAL

(inorganic- organic geochemistry)

MODELLING

(laboratory models - data inversion)

BASIN-WIDE ANALYSES

(modelling + data)













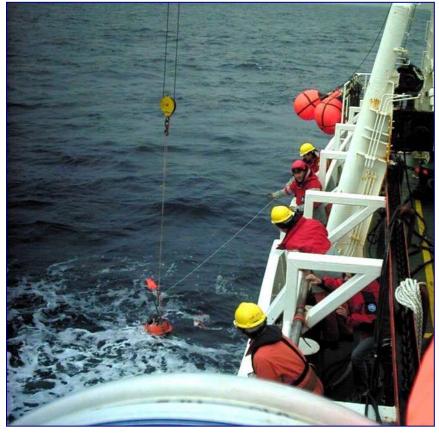






OCEAN BOTTOM SEISMOMETERS (OBS)







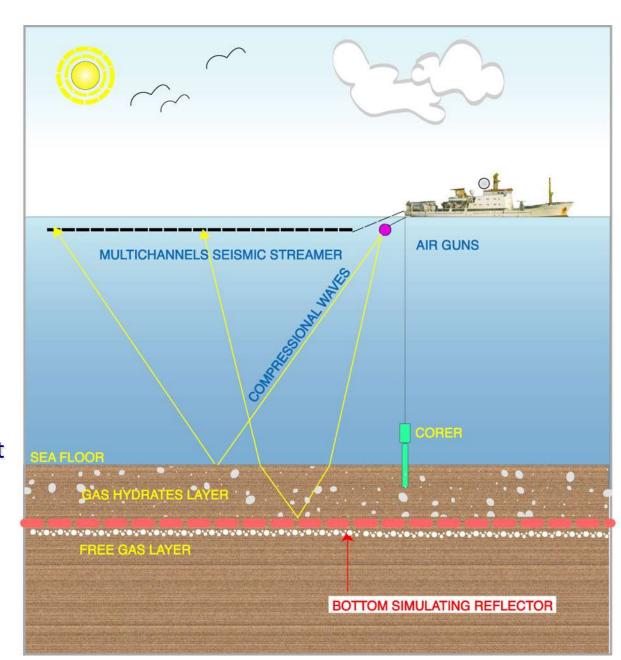




SEISMICS REFLECTION

The concept of Bottom
Simulating
Reflector
(BSR)

The BSR is a seismic reflector that simulates the sea bottom reflector. It is produced by the acoustic impedance contrast between sediments with hydrates (above) and sediments with free gas (below).



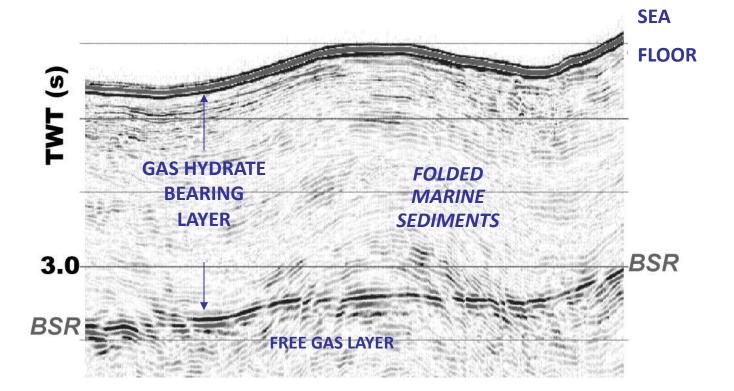






SEISMC PROFILE

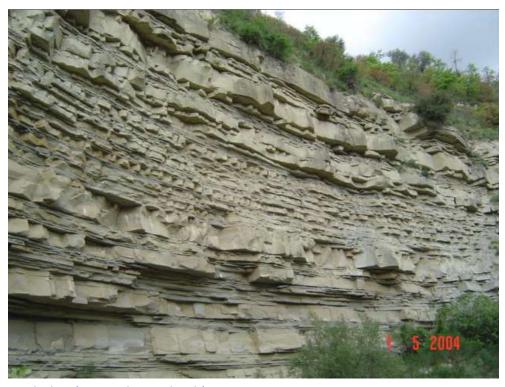








Stratification in Marine Sediments



Turbidite (Gorgoglione Flysch), Miocene, South Italy

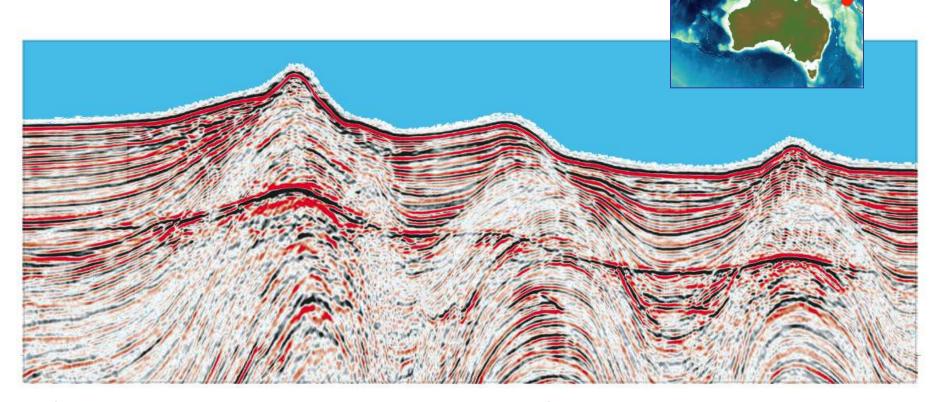


Marne a Fucoidi (Sezione del Bosso) Cretaceous (Albian), Central Apennines, Italy FOTO MASSIMO GALLI





Makassar Straits, Indonesia



Data from TGS-NOPEC Geophysical Company, published in Max et al., 2006. Economic Geology of Natural Gas Hydrates, Springer.





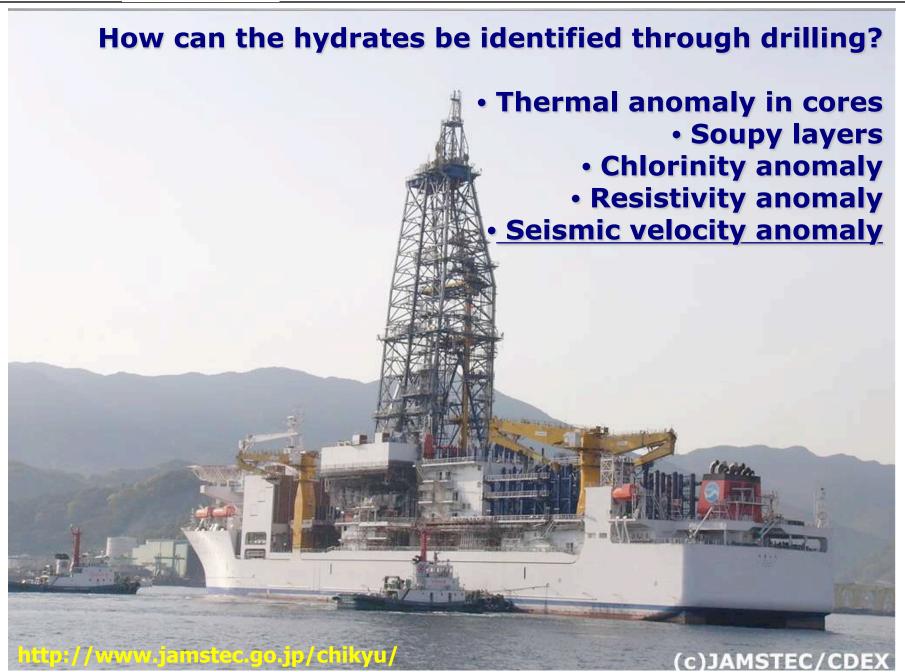
Characteristics of the BSR

- High reflection coefficient
- Reverse polarity with respect to the seafloor reflector
- Cuts through lithologic reflectors
- Its depth approximately coincides with the theoretical depth of the base of the hydrates stability zone.
- It may be underlined by another reflector produced by the base of the free gas layer (BGR)







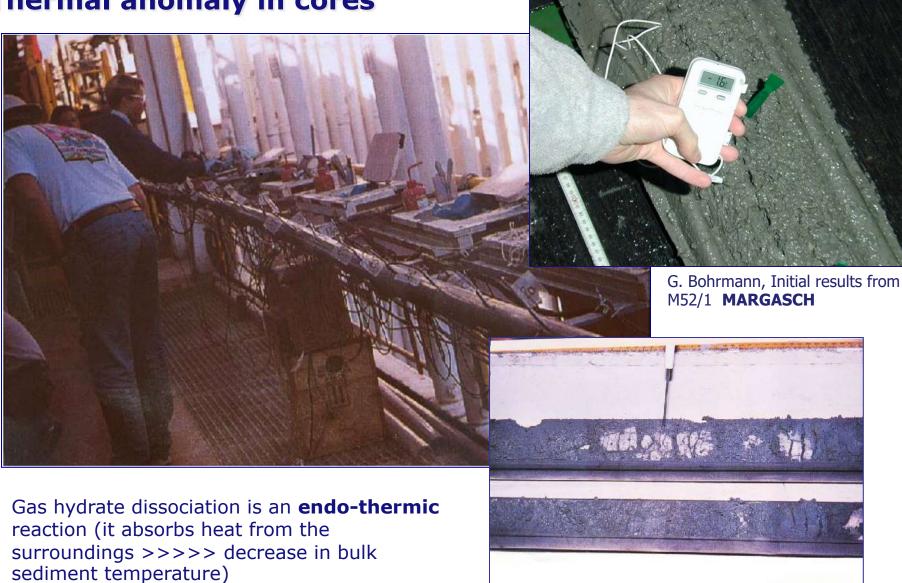






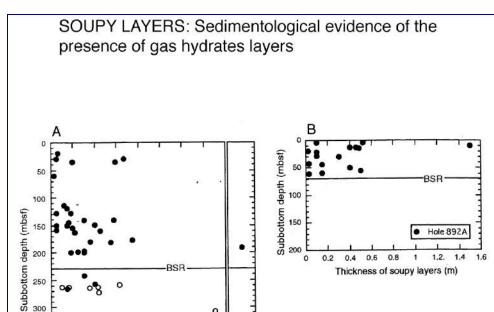






350





Hole 889 A

Thickness of soupy layers (m)

1.4 1.66

Soupy layers

Gas hydrate dissociation during core recovery destroys the sediment fabric, producing disturbed, liquified 'soupy' sediment layers in cores





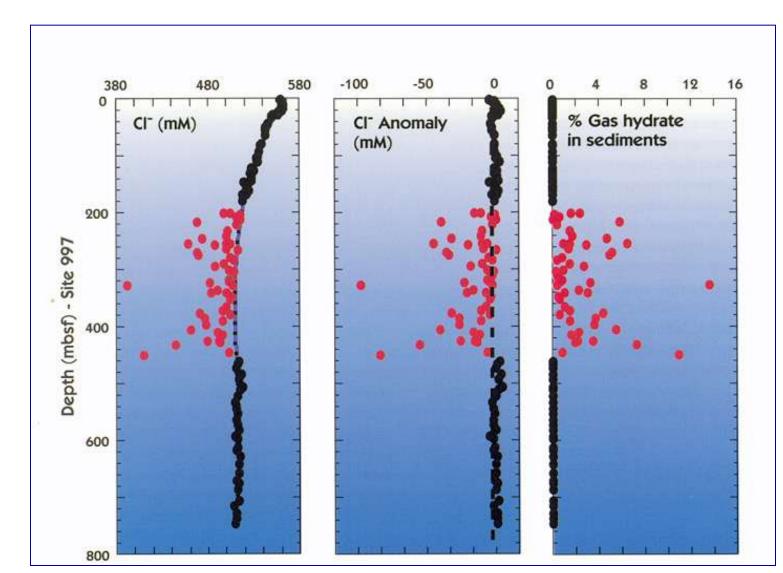






Chlorinity anomaly

Gas hydrate dissociation during core recovery produces **residual fresh water** that dilutes the ionic concentration of the pore water





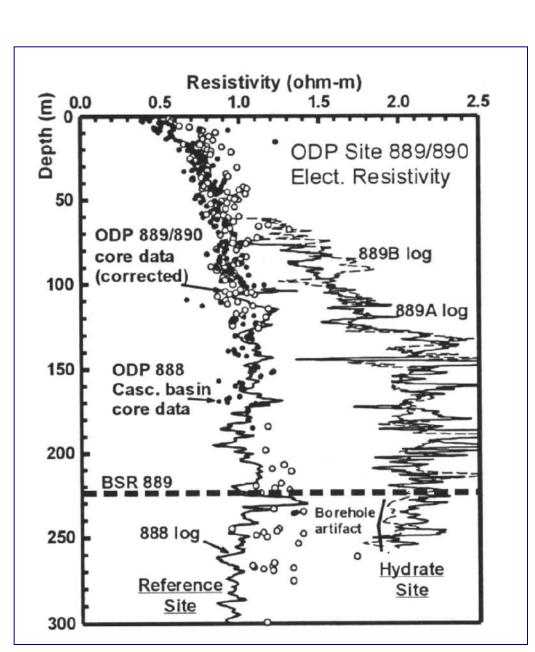




In situ Electrical Resistivity anomaly

The electrical resistivity of gas hydrates is **higher** than that of seawater.

The anomaly is identified with respect to a resistivity profile in a nearby borehole with similar sediment NOT containing hydrates





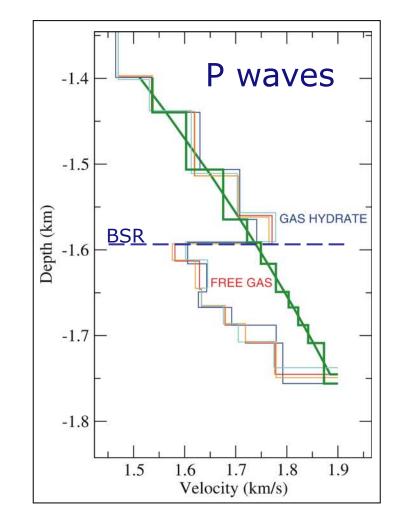


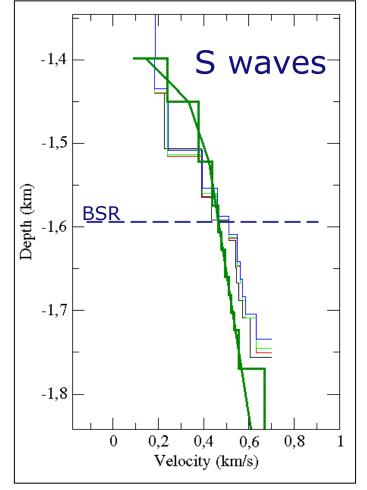


In situ seismic velocity anomaly

The compressional wave velocity of gas hydrates is **higher** than that of seawater.

The anomaly is identified with respect to a resistivity profile in a nearby borehole with similar sediment NOT containing hydrates









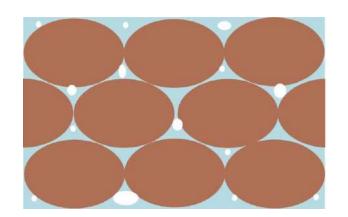




VELOCITY MODEL

Biot-Geerstma-Smit equations

$$V_{p} = \left\{ \left[\left(\frac{1}{C_{m}} + \frac{4}{3} \mu \right) + \frac{\frac{\phi_{eff}}{k} \frac{\rho_{m}}{\rho_{f}} + \left(1 - \beta - 2 \cdot \frac{\phi_{eff}}{k} \right) \cdot (1 - \beta)}{\left(1 - \phi_{eff} - \beta \right) C_{b} + \phi_{eff} C_{f}} \right] \cdot \frac{1}{\rho_{m} \left(1 - \frac{\phi_{eff}}{k} \frac{\rho_{f}}{\rho_{m}} \right)} \right\}^{1/2}$$
(1)



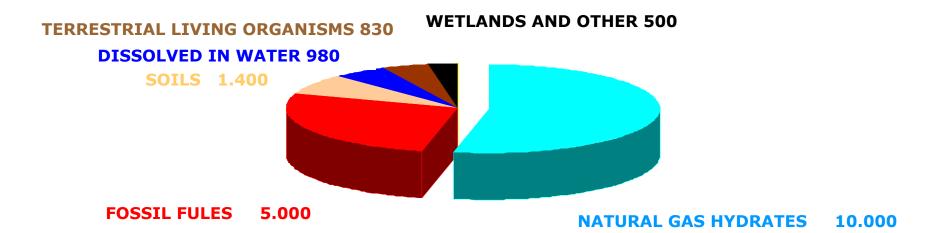






GLOBAL ESTIMATES

Estimation as amount of Carbon atoms · 10¹⁵ g



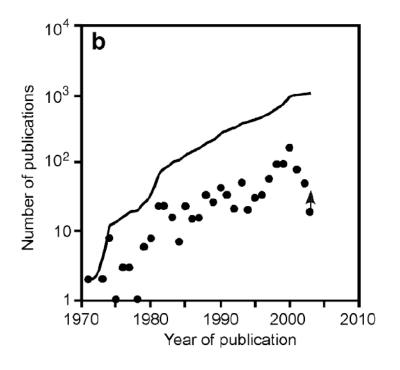


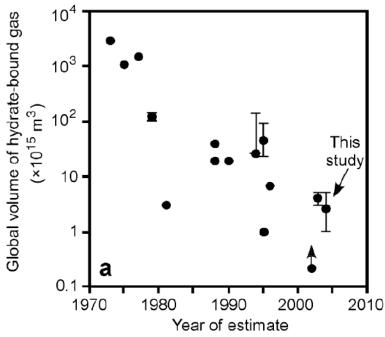




HOWEVER.....

Global estimates of the volume of hydrate-bound gas in marine sediments versus the year in which the estimate was made.











Detrital OM

Atmosphere

Marine biota

66.6 (0.6%)

Gas

hydrates

2,500 (22%)

Peat

500 (4%)

Land biota

830 (7%)

Dissolved OM

in water

980 (9%)

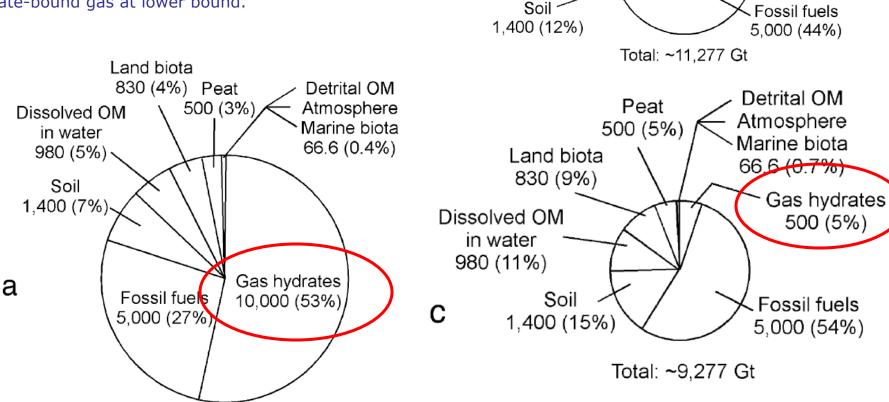
b

Distribution of organic carbon in the Earth

(excluding dispersed organic carbon such as kerogen and bitumen)

- a) The distribution based on the estimate of 10000 Gt of methane carbon in gas hydrates (Kvenvolden, 1993).
- b) The distribution based on the revised estimate of the global gas hydrate inventory assuming the global volume of hydrate-bound gas at upper bound.
- c) The distribution based on the revised estimate of the global gas hydrate inventory assuming the global volume of hydrate-bound gas at lower bound.

Total: ~18,777 Gt









In spite of changing global hydrates estimates......

Methane is the cleanest fossil fuel:

$$CH_4 + 2O_2 \longrightarrow 2H_2O + CO_2 + heat$$

- Methane is presently the preferred fossil fuel for power generation in OECD countries,
- Methane demand is increasing also in other regions such as the Middle East, China and India
- Despite current high prices, the vast majority of new power generation on line in the review period will be gas-fired.







Methane Hydrates are considered as NON CONVENTIONAL HYDROCARBONS by most oil companies

- There is not an established extraction technology.
- The hydrates must be dissociated in situ by:
 - Heat injection
 - Injection of chemical inhibitors
 - depressurization

•

- There are severe environmental implications and the methods will not be considered as economically feasible until the gas price will rise considerably.
- USA, Japan, India, Korea and other countries of the Pacific Area have started National Plans for the gas hydrates assessment in their Economic Exclusive Zones.







What Happens in Europe?

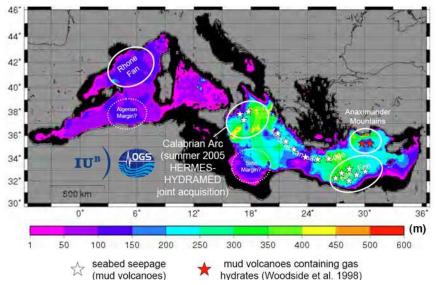
- No policy has been established so far
- Several EC projects have been dedicated to the matter from a purely scientific point of view (see next part of this talk)
- Oil companies and academia have focussed don this issue regarding the implications for submarine slope stability (see next part of this talk)
- Regional evaluations of gas hydrate potential of European Margins are in due course:
 - Atlantic European Margins
 - Mediterranean
 - Black Sea







HYDRATES PROSPECTIVIY in the Mediterranean Sea



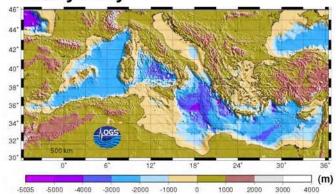
D. Praeg, V. Unnithan, and A. Camerlenghi, American Geohpysical **Unions Fall** Meeting, December 2006

Due to:

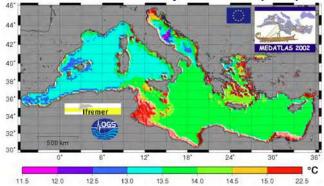
- High temperature of bottom water
- High salinity of pore water
- Presence of a salt seal (Messinian) preventing deep fluid migration

The thickness of the Gas Hydrates Stability Zone is reduced in the **Mediterranean Sea with respect to** world Oceans.

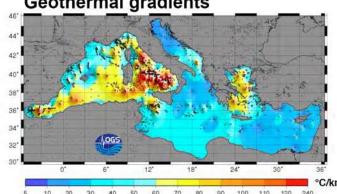
Bathymetry



Bottom water temperatures (bwt)



Geothermal gradients









METHANE HYDRATES AND GLOBAL CLIMATE



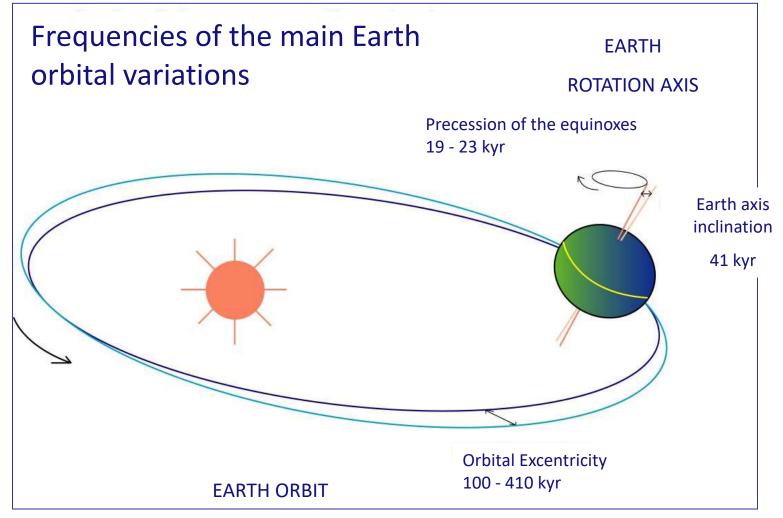






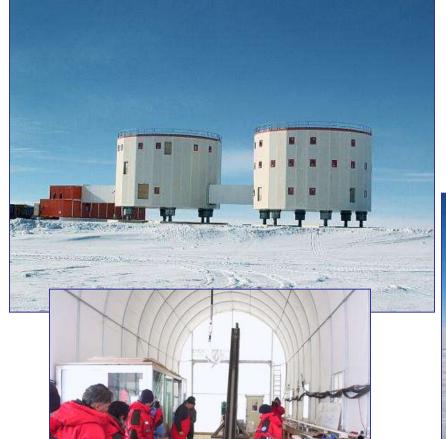


MILANKOVITCH ORBITAL CYCLES

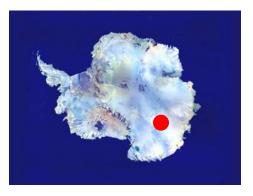


The mean Solar radiation received by the earth changes through time. Such changes are the primary cause of glacial interglacial epochs and other climatic stages.





Programma Nazionale di Ricerche in Antartide (PNRA)

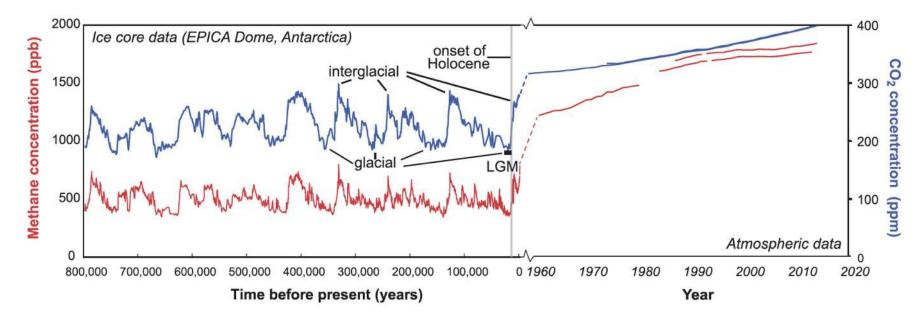


Dome-C Station, Antarctica



European Project for Ice Coring in Antarctica (EPICA)





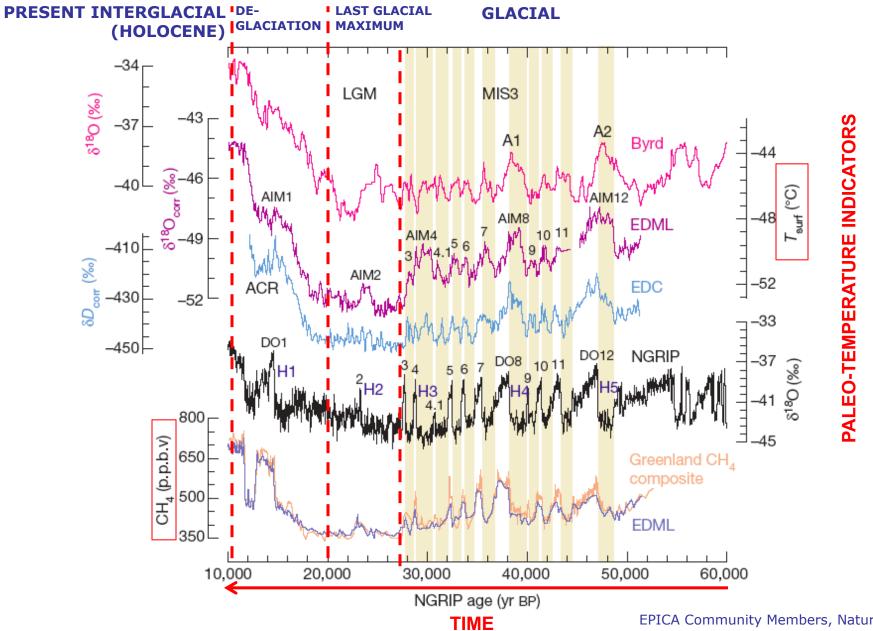
Methane and CO2 concentrations in the atmosphere since 800 ka from Environmental Protection Agency [2015]. Deep time data are from analysis of the European Project for Ice Coring in Antarctica (EPICA) Dome ice core and thus reflect Southern Hemisphere methane. Note the difference in time scales between the panels.

Ruppel, C. D., and J. D. Kessler (2017), The interaction of climate change and methane hydrates, Rev. Geophys., 55, 126-168, doi:10.1002/2016RG000534.















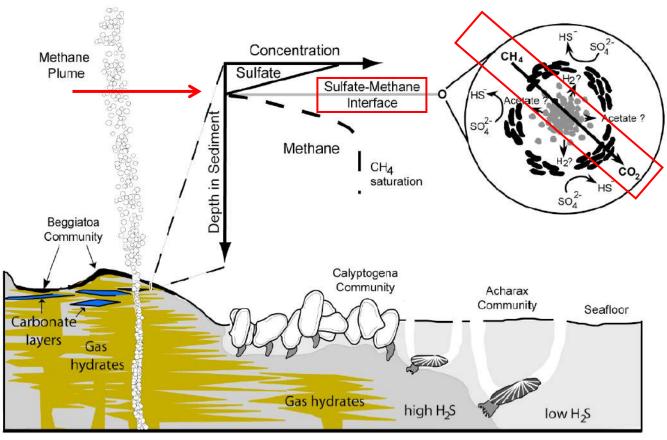
METHANE HYDRATES STORED IN PERMAFROST AND MARINE SEDIMENTS ARE HIGHLY SENSITIVE TO AMBIENT TEMPERATURE AD PRESSURE CHANGES

IN MARINE SEDIMENTS:

- WHEN OCEAN BOTTOM WATER TEMPERATURE RISES (DE-GLACIACION) SOME METHANE CHANGES FROM HYDRATE FORM TO GASE FORM AND ESCAPES FROM TEH SEAFLOOR INTO THE WATER COLUMN, AND FROM THERE TO THE ATMOSPHERE.
- HOWEVER, BECAUSE SEALEVEL CHANGES ACCORDING TO CLIMATIC CHANGES, THE INFLUENCE OF PRESSURE HAS TO BE TEKEN INTO ACCOUNT.











Bathymodiolus heckerae mussel beds. (A) Juvenile and adult mussels at Marker 'E'. (B) Dead mussels and octopus. (C) Extensive bed of live mussels of relatively uniform size, partially covered by bacterial mats, at Marker 'B'. (D) Dead mussels at the eastward periphery of Marker 'B'. (E) Mussels with a chiridotid holothurian and Alvinocaris sp. (F) Mussels with Alvinocaris sp. And ophiuroids. Scale bars: A-D: 10 cm, E; F: 5 cm.

Van Dover et al. (2003). Deep Sea Research





FORAMINIFERA (Marine zooplankton)

- Protists: Unicellular marine organisms
- They live in all marine environments
- They have existed since the Cambrian
- Size: from 0.01 to 5 mm (> 50 mm)

Shell composition:

- Agglutinated
- Tectine
- Calcium Carbonate (CaCO₃)









- Foraminifera form their skeleton in **isotopic equilibrium** with ambient seawater
- The isotopic composition of the skeleton of benthic foraminifera reflects the **micro-habitat** where they live

The $\delta^{13}C$ of foraminifera depends on the $\delta^{13}C$ of the Dissolved Inorganic Carbon (DIC) of the water in which they live:

Normal marine environment: $\delta^{13}C_{DIC} \cong 0\%$

(Peterson e Fry, 1987)

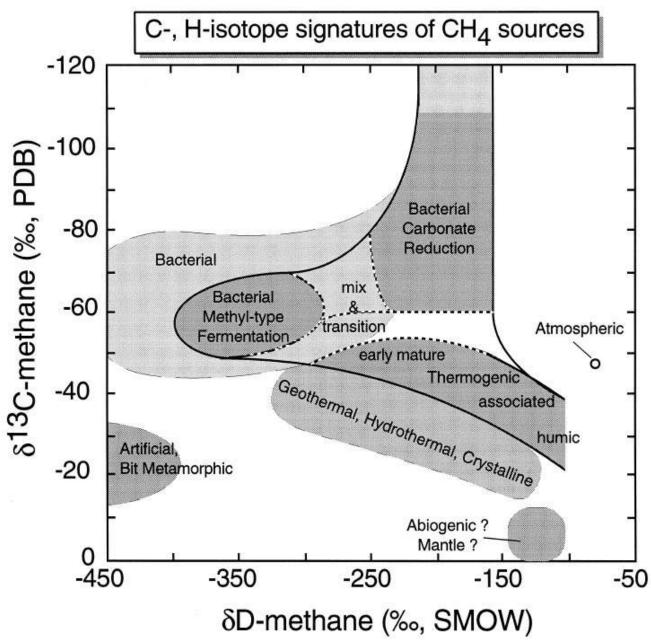
Methane seeps environment: $\delta^{13}C_{DIC} \cong -30/-60\%$

(Aharon et al., 1992)





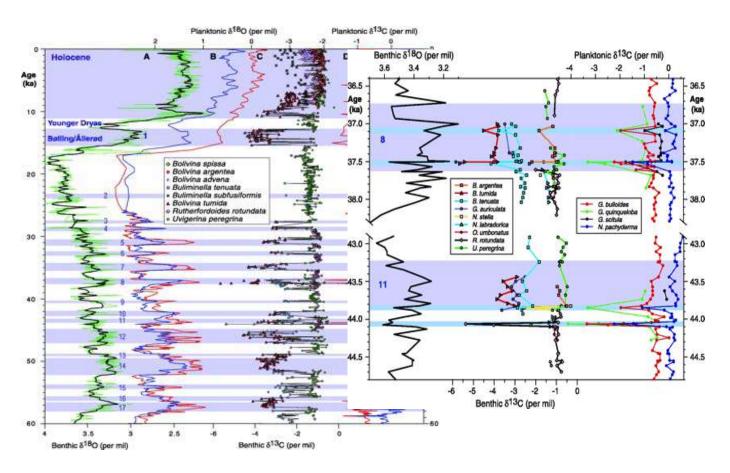








Negative excursions of $\delta^{13}C$ in marine sediments during warm periods can be explained ONLY with massive emissions of Methane from the seafloor

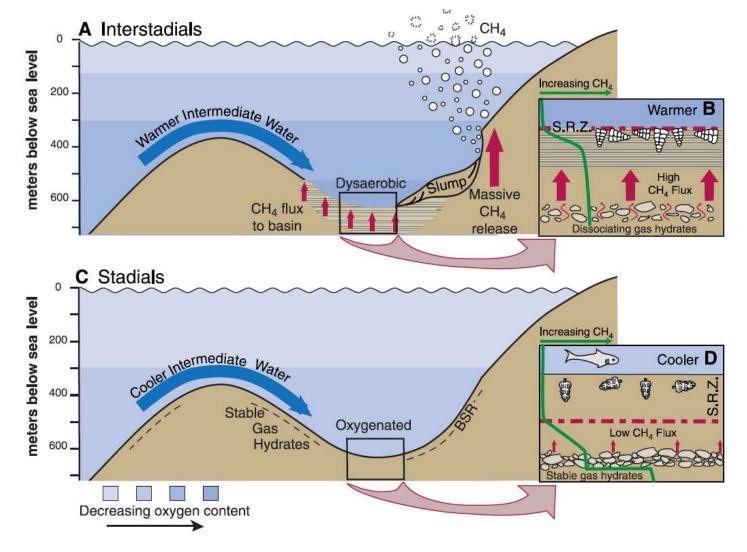








The Clathrate Gun Hypothesis

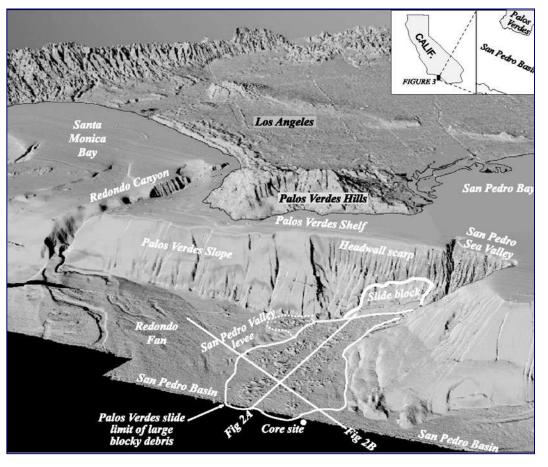








Methane hydrate dissociation as source of submarine slope instability

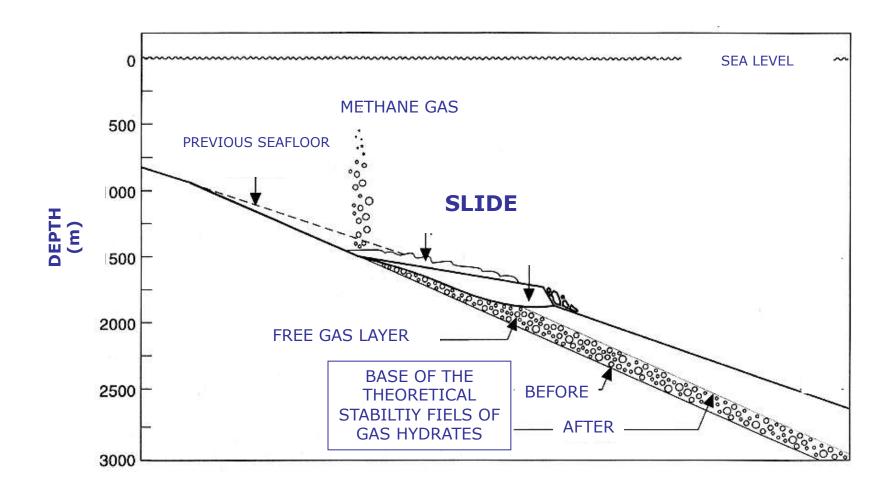








GLOBAL WARMING

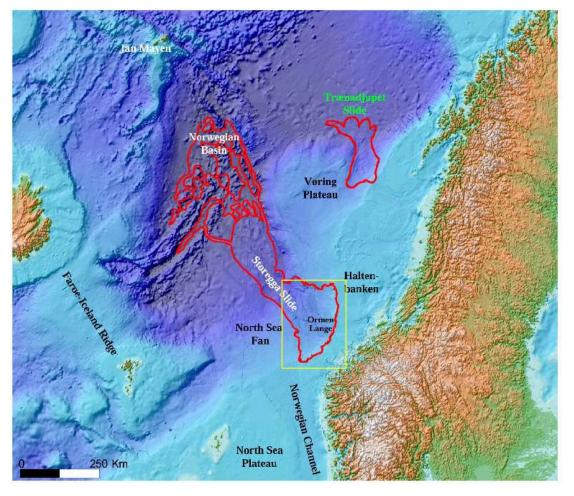








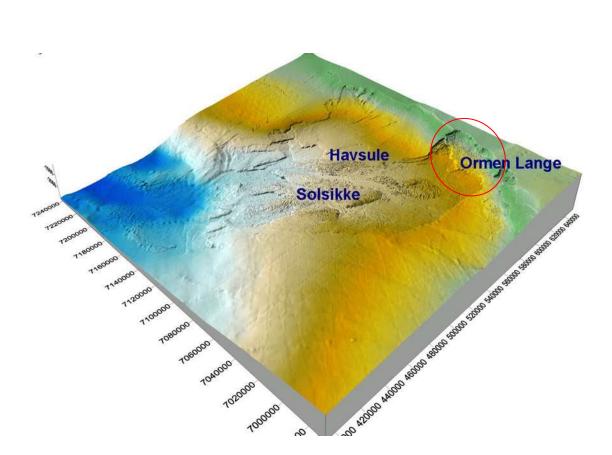
THE STOREGGA SLIDE ON THE NORWEGIAN CONTINENTAL MARGIN(7000 yr)











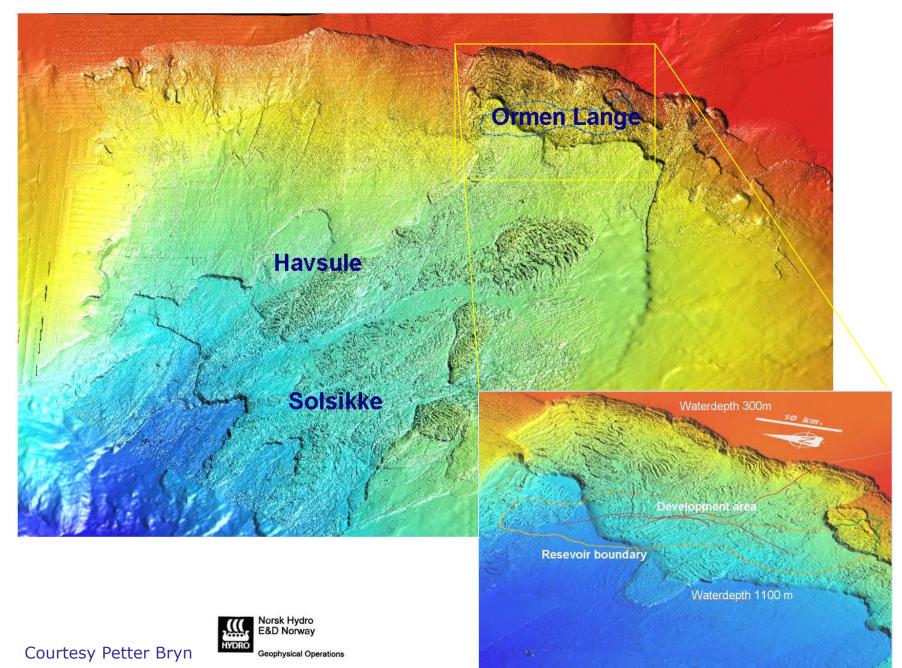


















THE STOREGGA SLIDE CAUSED A HUGE TSUNAMI

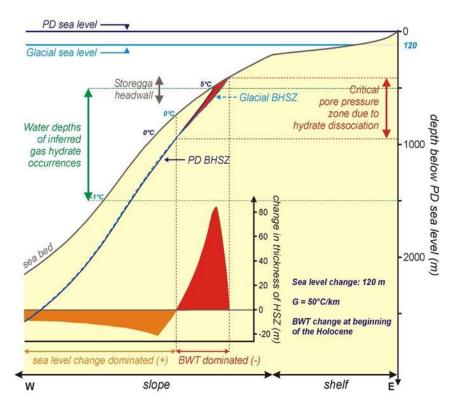


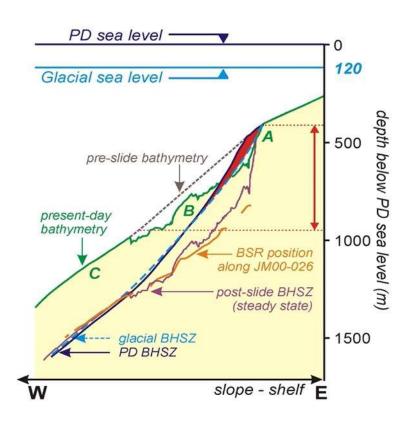






The effect of gas hydrates



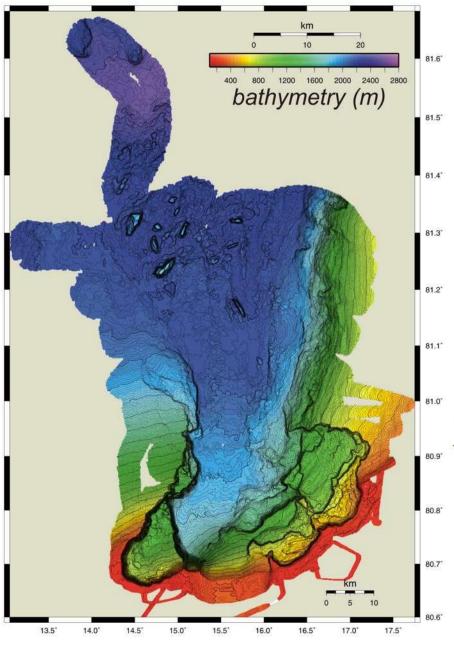


Mienert et al., 2005

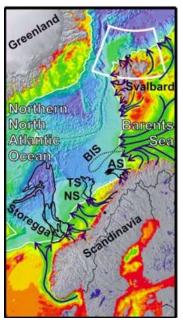
Change in theoretical thickness of hydrate stability zone due to sea level change (small, negative) and bottom water temperature (locally large, positive). Biggest effect near upper headwall in Storegga.

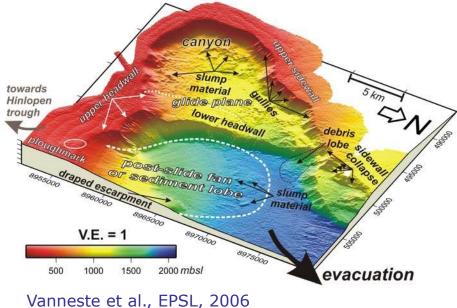






THE NORTH SVALBARD SLIDE



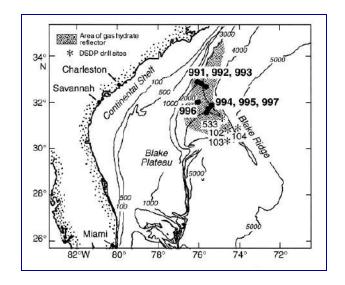


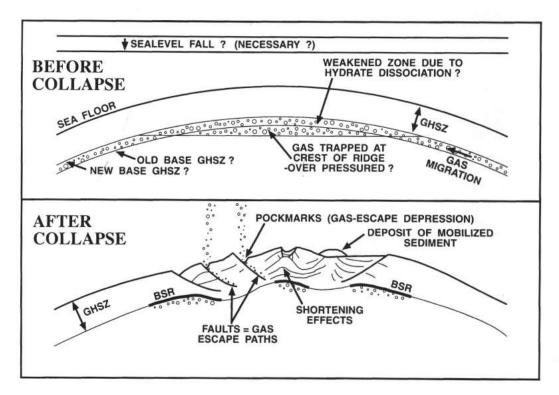


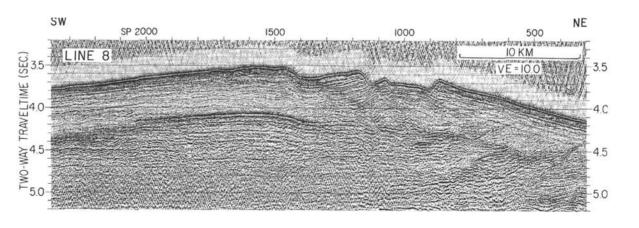




BLAKE RIDGE







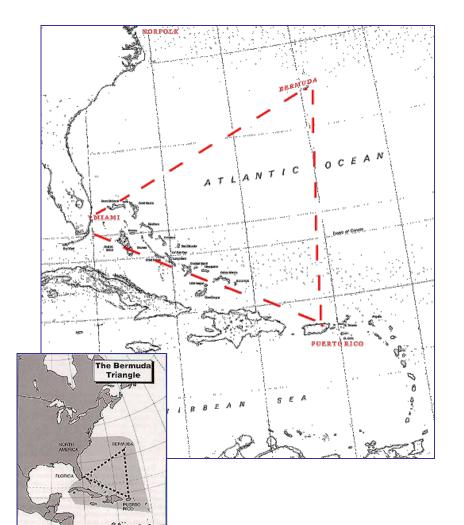
(Dillon et al., 1998)

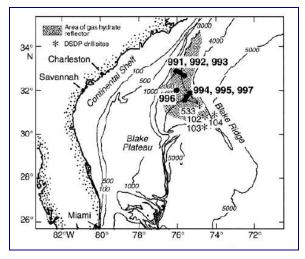






AT THE EDGE OF SCIENCE





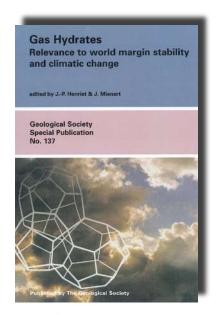


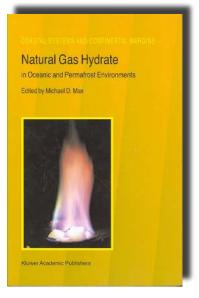


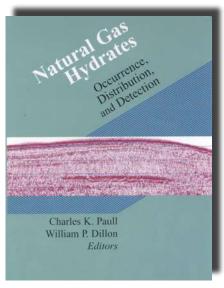


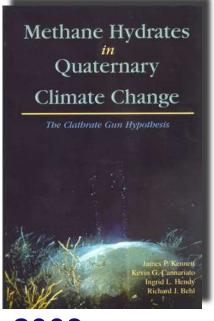


4 scientific books







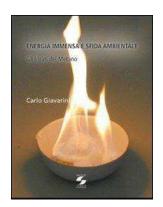


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