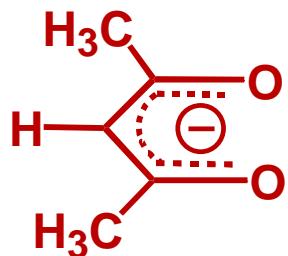
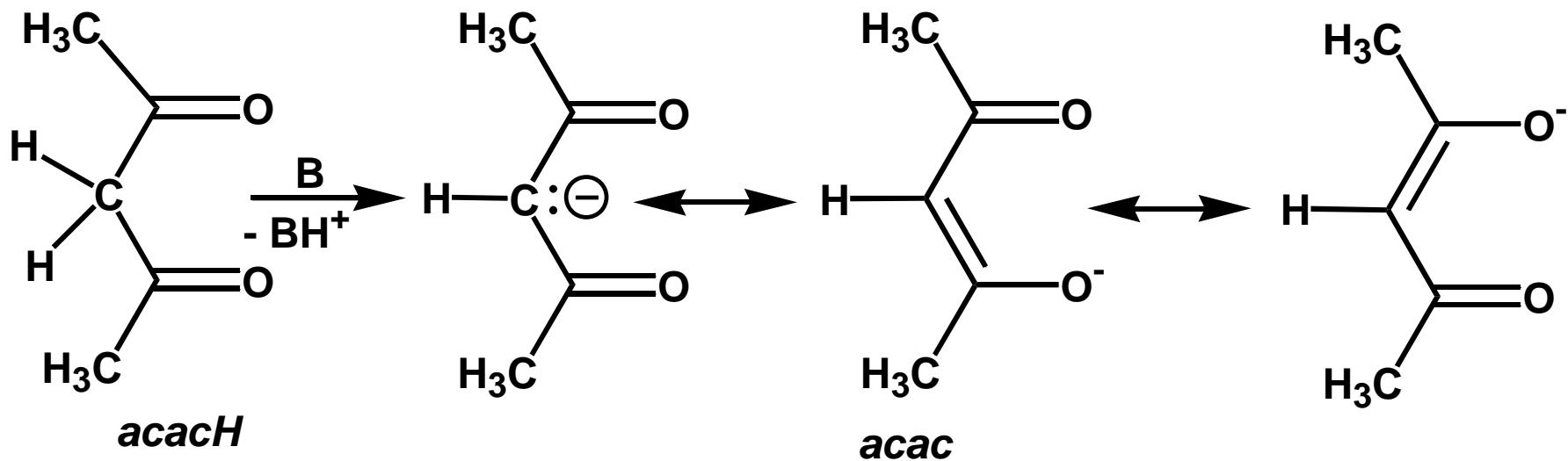


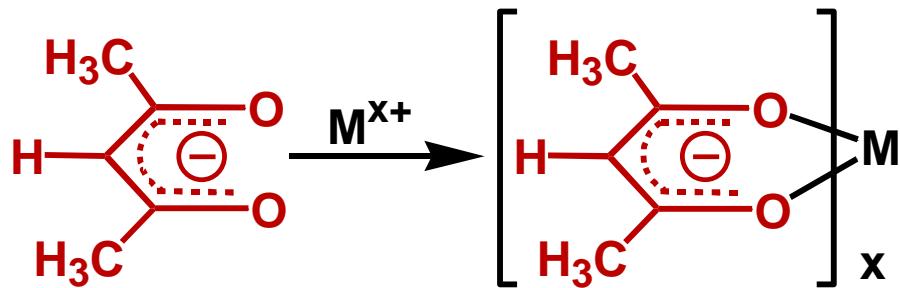
ESPERIENZA 1

Sintesi di acetilacetonato complessi di metalli di transizione della prima serie

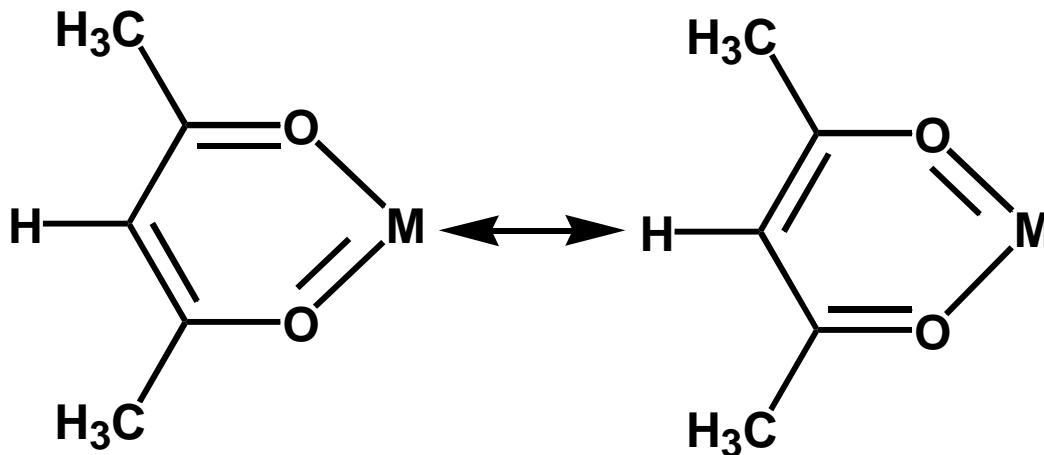
Il comportamento acido/base del 2,4-pentandione



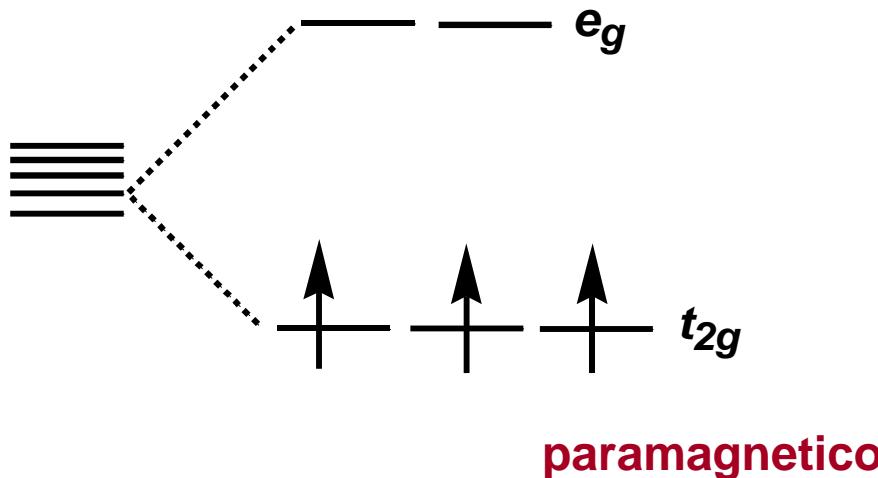
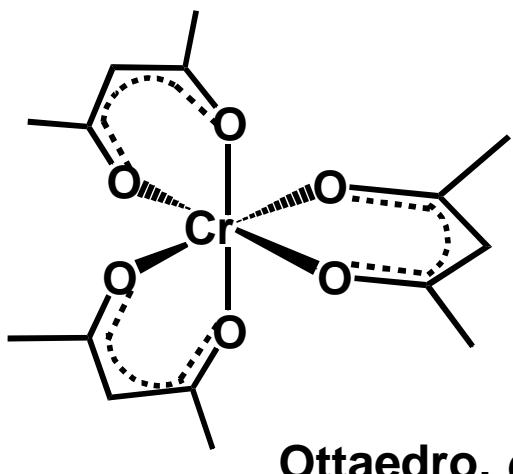
La coordinazione dello ione acetilacetonato ai metalli di transizione



Altre forme di risonanza dovute alla coordinazione



Distanze di legame a due a due uguali: M-O, C-O, C-C

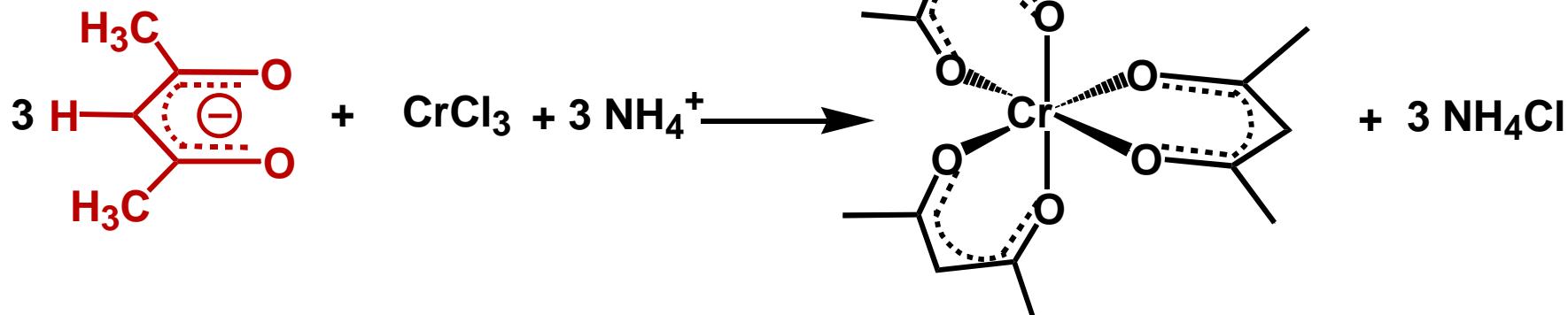


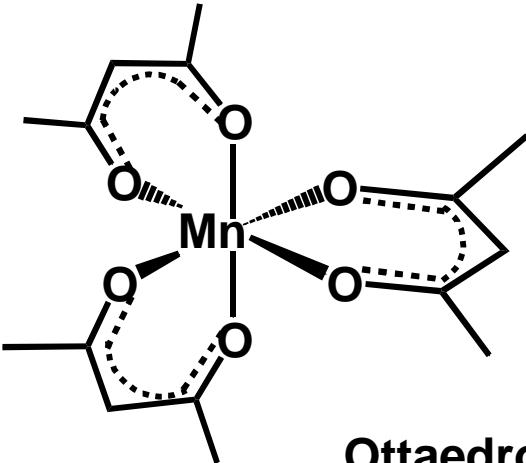
Sintesi di $[\text{Cr}(\text{acac})_3]$

Idrolisi dell'urea



Sintesi del complesso



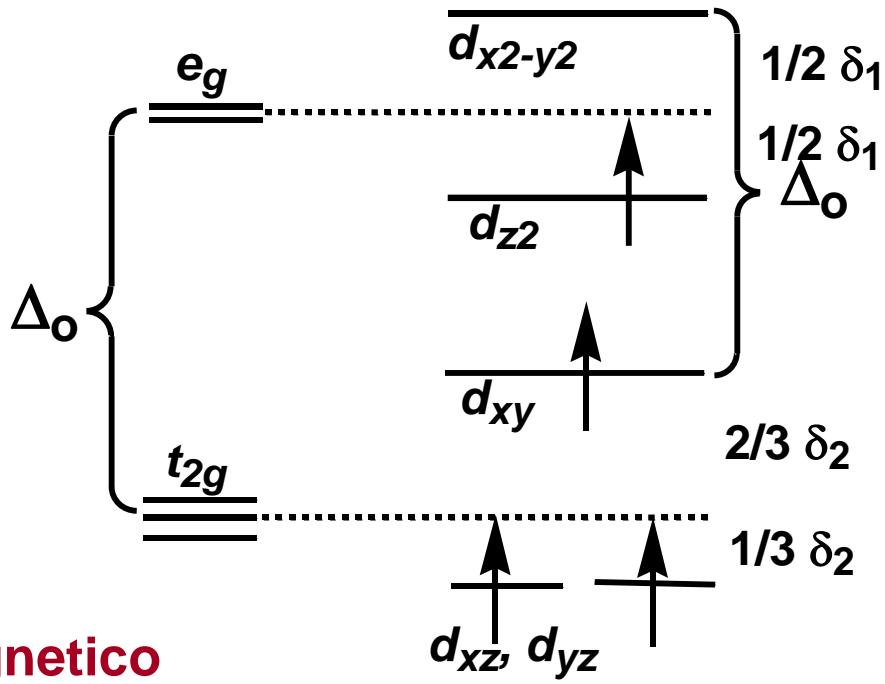
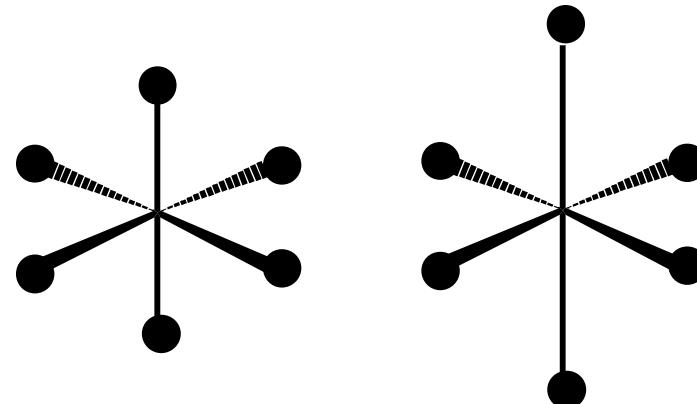


Ottaedro, d^4

Distorsioni per effetto Jahn-Teller:

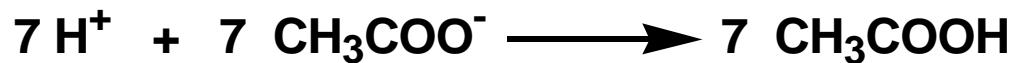
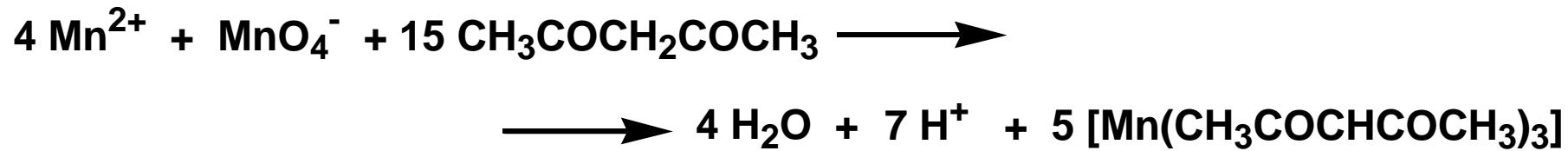
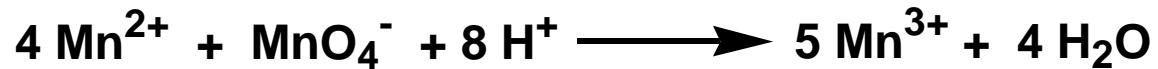
Allungamento: 2 Mn-O 2.12 Å;
4 Mn-O 1.93 Å;

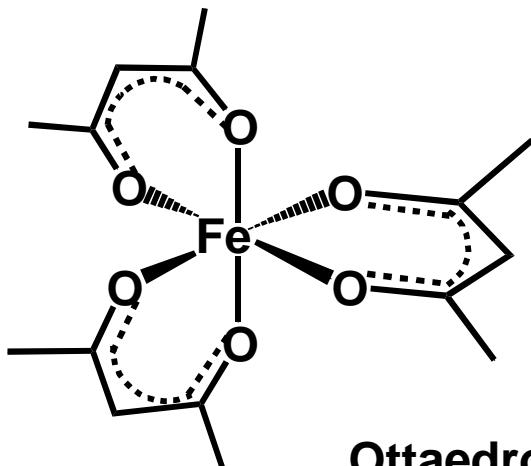
Compressione: 2 Mn-O 1.95 Å;
4 Mn-O 2.00 Å.



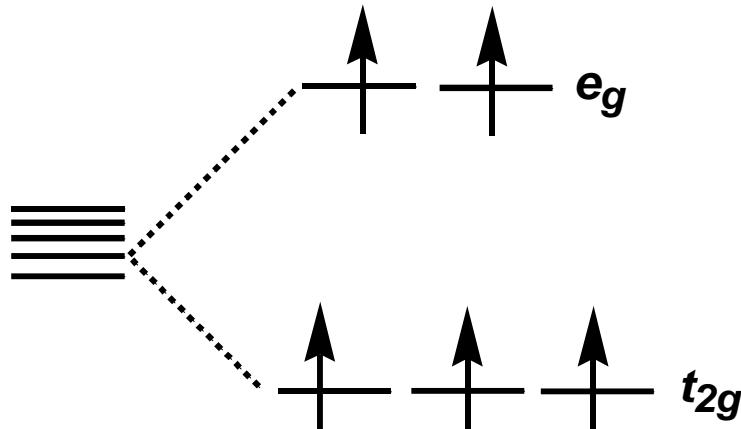
Sintesi di $[\text{Mn}(\text{acac})_3]$

Sintesi del complesso





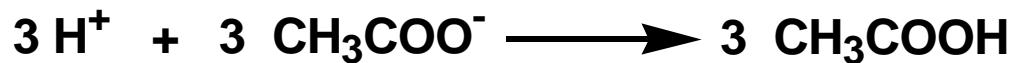
Ottaedro, d^5

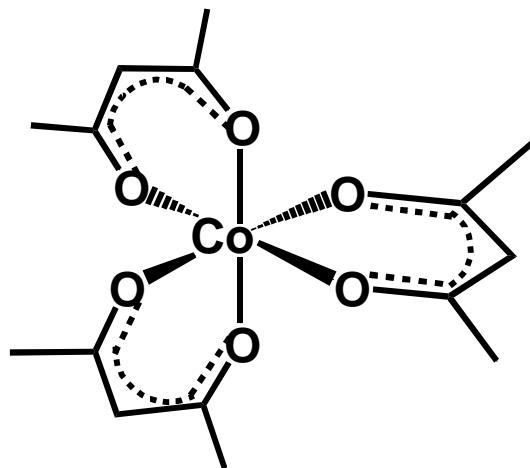


paramagnetico

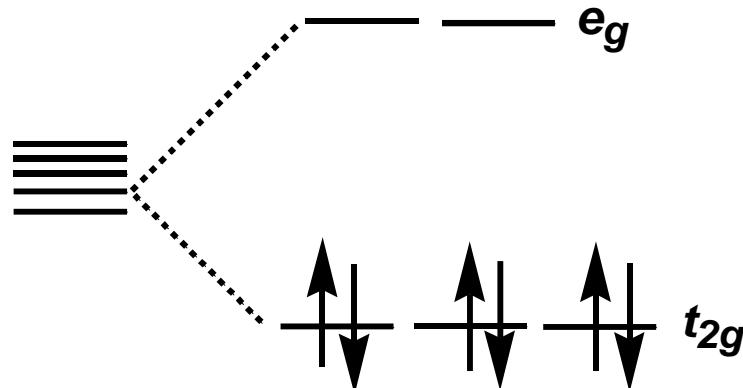
Sintesi di $[\text{Fe}(\text{acac})_3]$

Sintesi del complesso





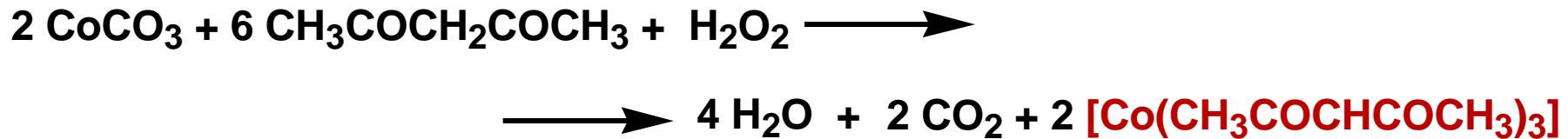
Ottaedro, d^6



diamagnetico

Sintesi di $[\text{Co}(\text{acac})_3]$

Sintesi del complesso



Da bilanciare



26.1

Average residence time for H₂O molecule in first hydration shell / s

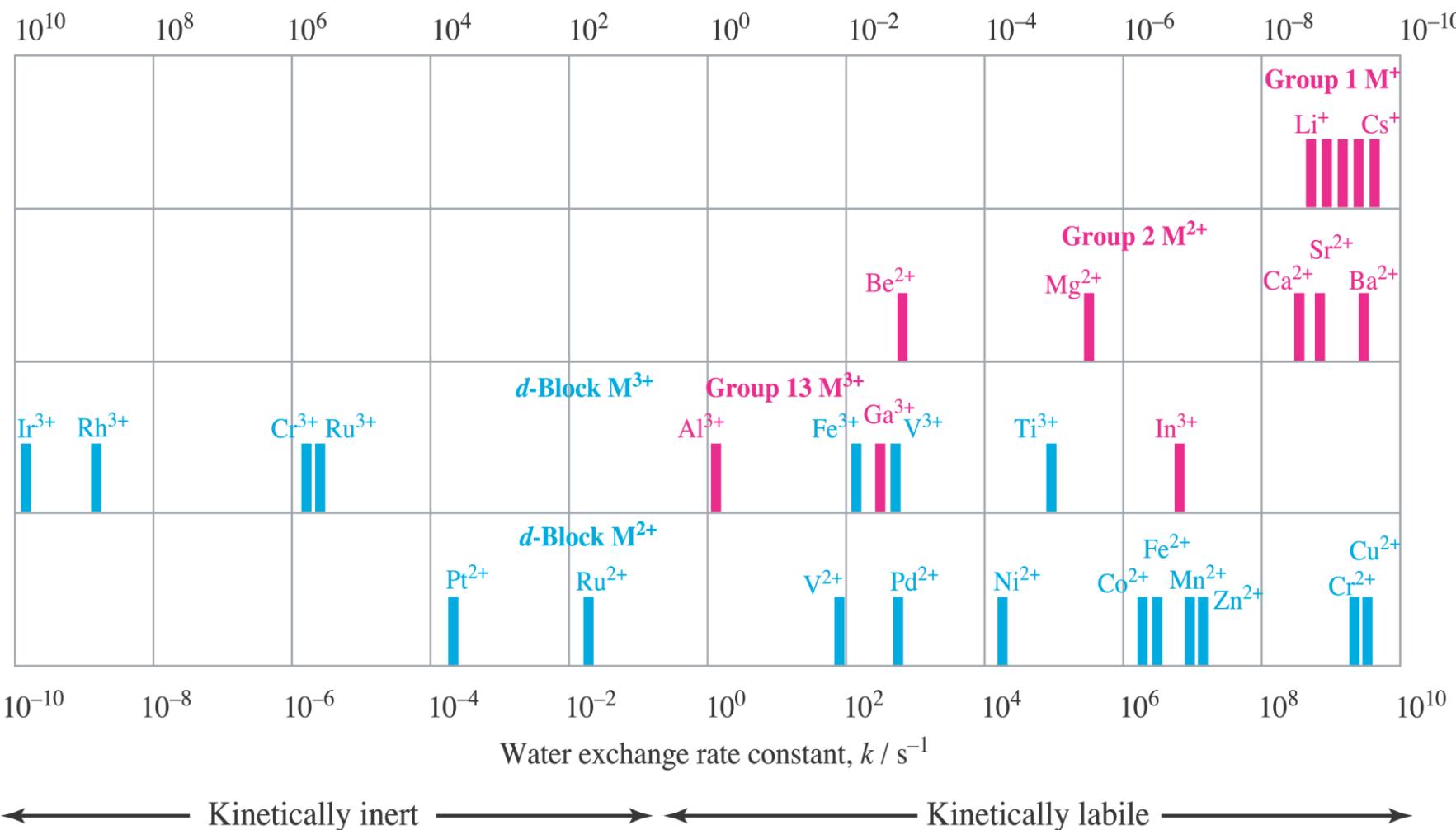
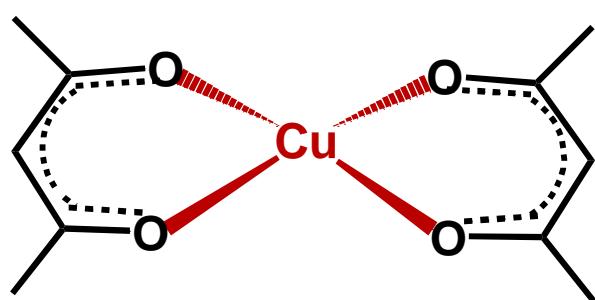
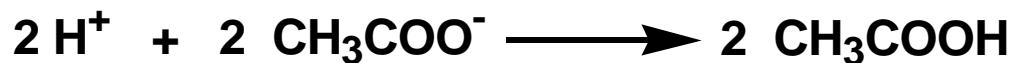
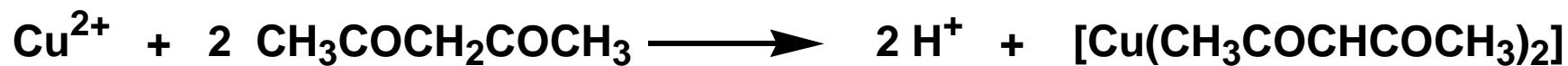


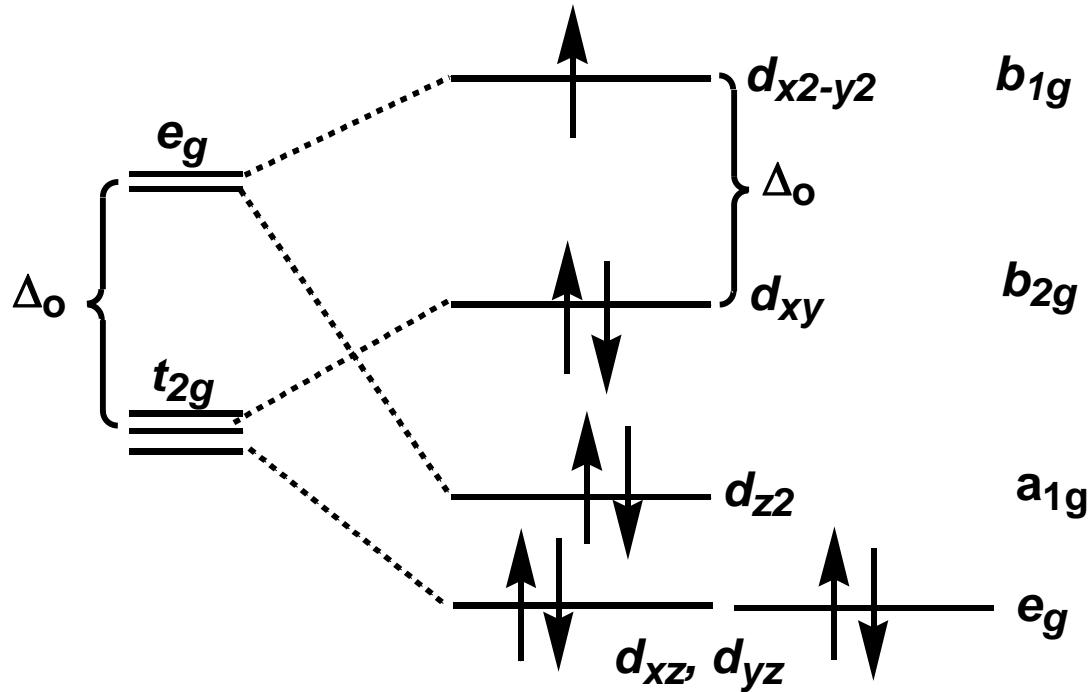
Fig. 26.1 Water exchange rate constants and average residence times for water molecules in the first coordination sphere of aquated metal ions at 298 K. Group 1, 2 and 13 metal ions are shown in pink, and *d*-block metal ions in blue. [Based on S.F. Lincoln (2005) *Helv. Chim. Acta*, vol. 88, p. 523 (Figure 1).]

Sintesi di $[\text{Cu}(\text{acac})_2]$

Sintesi del complesso

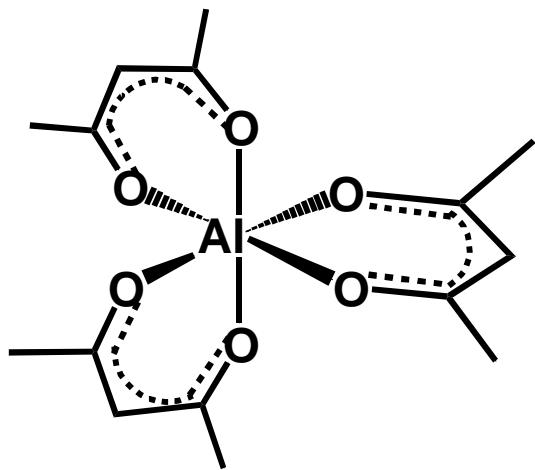


planare quadrato, d^9



Sintesi di $[\text{Al}(\text{acac})_3]$

Sintesi del complesso



ottaedro, **no elettroni d**

diamagnetico