

Geophysical Fluid Dynamics

Lecture II: Statics

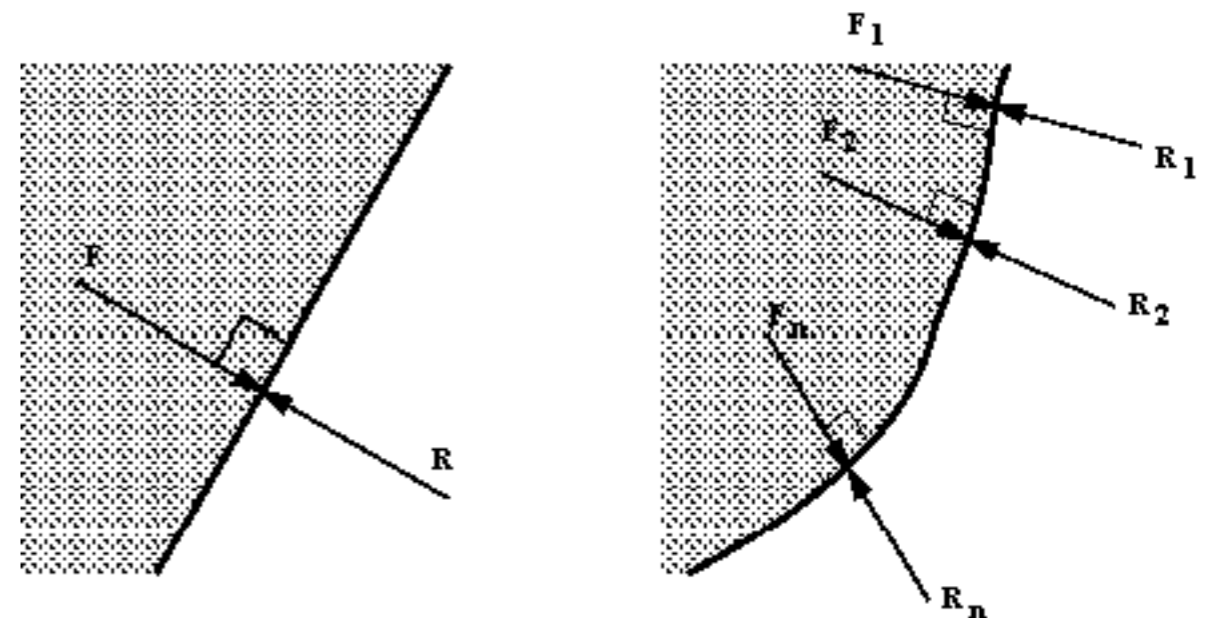


Summary of previous lecture

- What is a fluid? definitions
- Properties (mass, weight, density)
- Ideal fluids vs Real fluids
- Viscosity is a very important fluid property
- Newton's law of viscosity: τ is proportional to fluid μ and the velocity gradient
- Newtonian fluid vs Non-Newtonian fluid
- liquid (gases) have high (low) viscosity

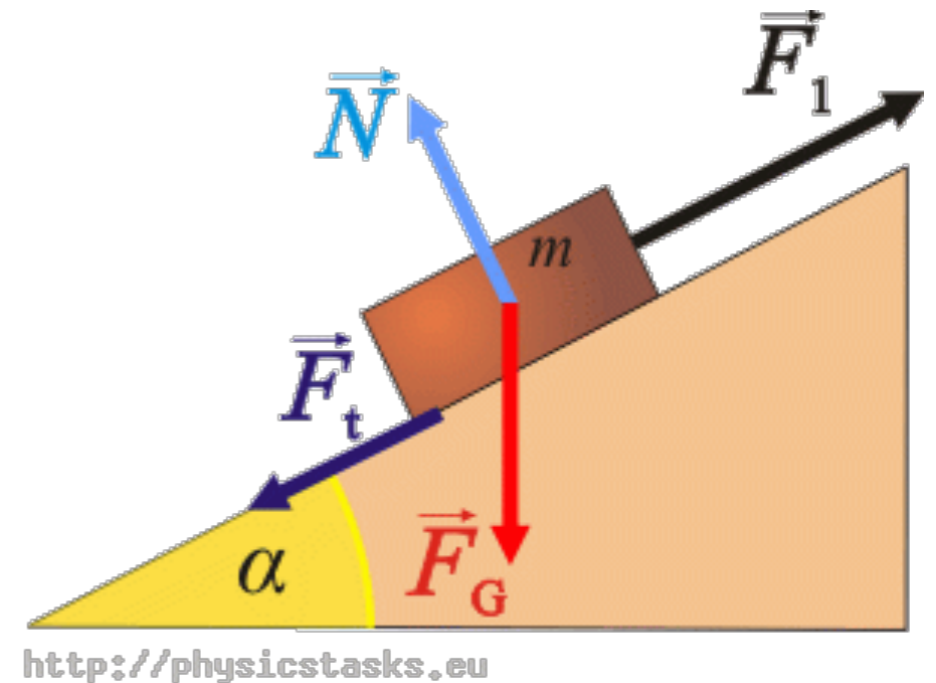
Fluid Statics

- Fluid is at rest
- A static fluid can have no shearing force acting on it.
- The only forces are due to pressure.
- Any force between fluid and boundary must be acting at right angles (normal to).
- Fluid at rest is in equilibrium: sum of components of forces in any direction must be zero.



does pressure have a direction?

- FORCE is a vector (forces on box have different directions and magnitude)
- is PRESSURE a vector too?



Isotropy of Pressure

- In a fluid at rest, the tangential viscous stresses are absent and the only force is normal to the surface.
- The surface force per unit area (PRESSURE) is equal in all directions.
- Pressure at any point in a fluid at rest has a single value (is a scalar). This is known as **Pascal's Law**.

Gauss theorem (or the divergence theorem)

- relates the flow flux of a vector field through a surface to the behavior of the vector field inside the surface
- The outward flux of a vector field through a closed surface is equal to the volume integral of the divergence over the region inside the surface
- The sum of sources and sinks (divergence) will give you the outward flux

$$\iiint_v (\nabla \cdot F) dV = \iint_A F dA$$

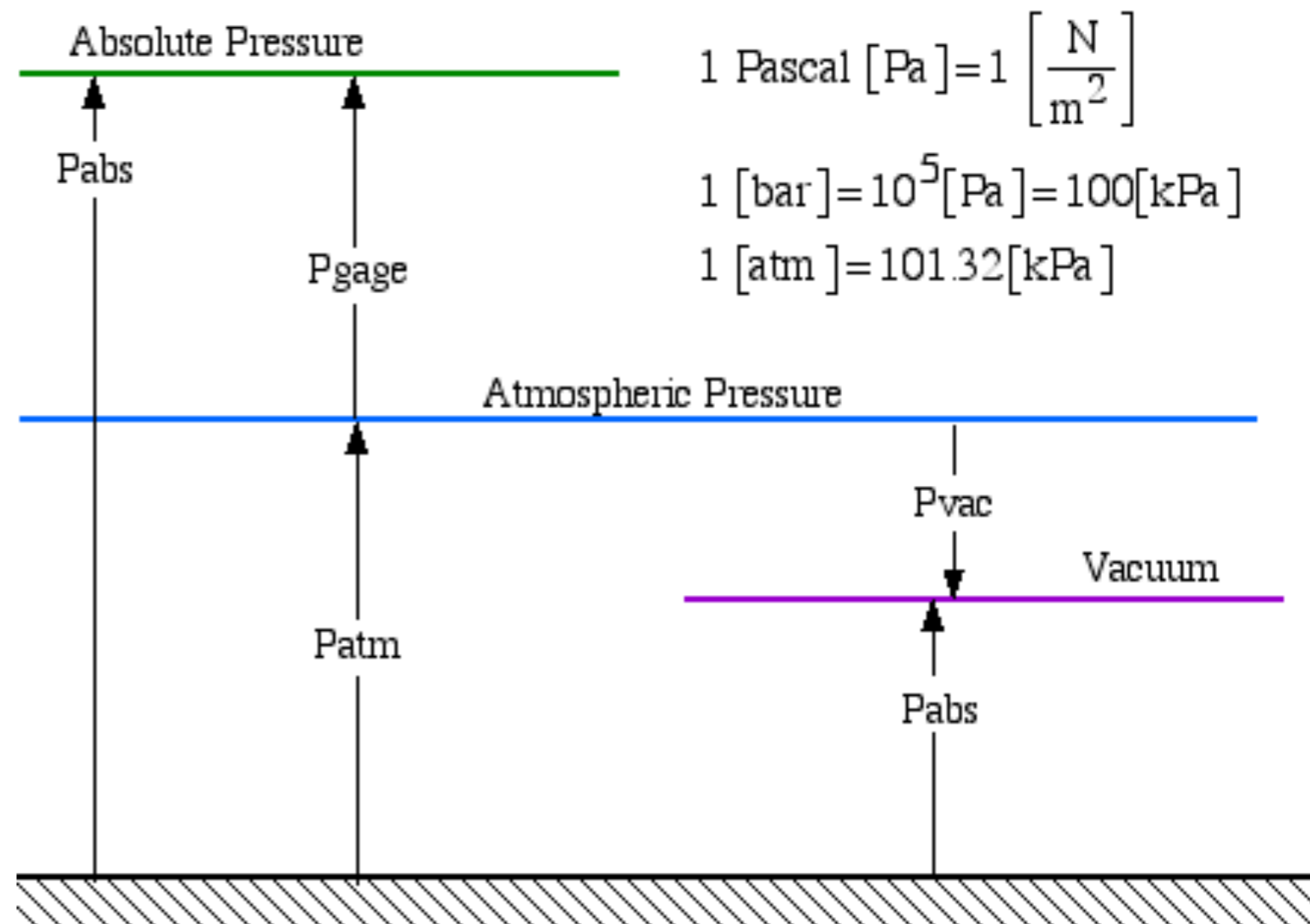
Pressure variations for incompressible fluids

- $P - P_0 = -\rho g (z - z_0)$
- Applies to liquids (no need to consider compressibility unless dealing with large changes in z ... deep in the ocean)
- Applies to gases for small changes in z only
- **$P = -\rho g h$** Pressure related to the height h of a fluid column: *Pressure head*

Absolute and Gage Pressure

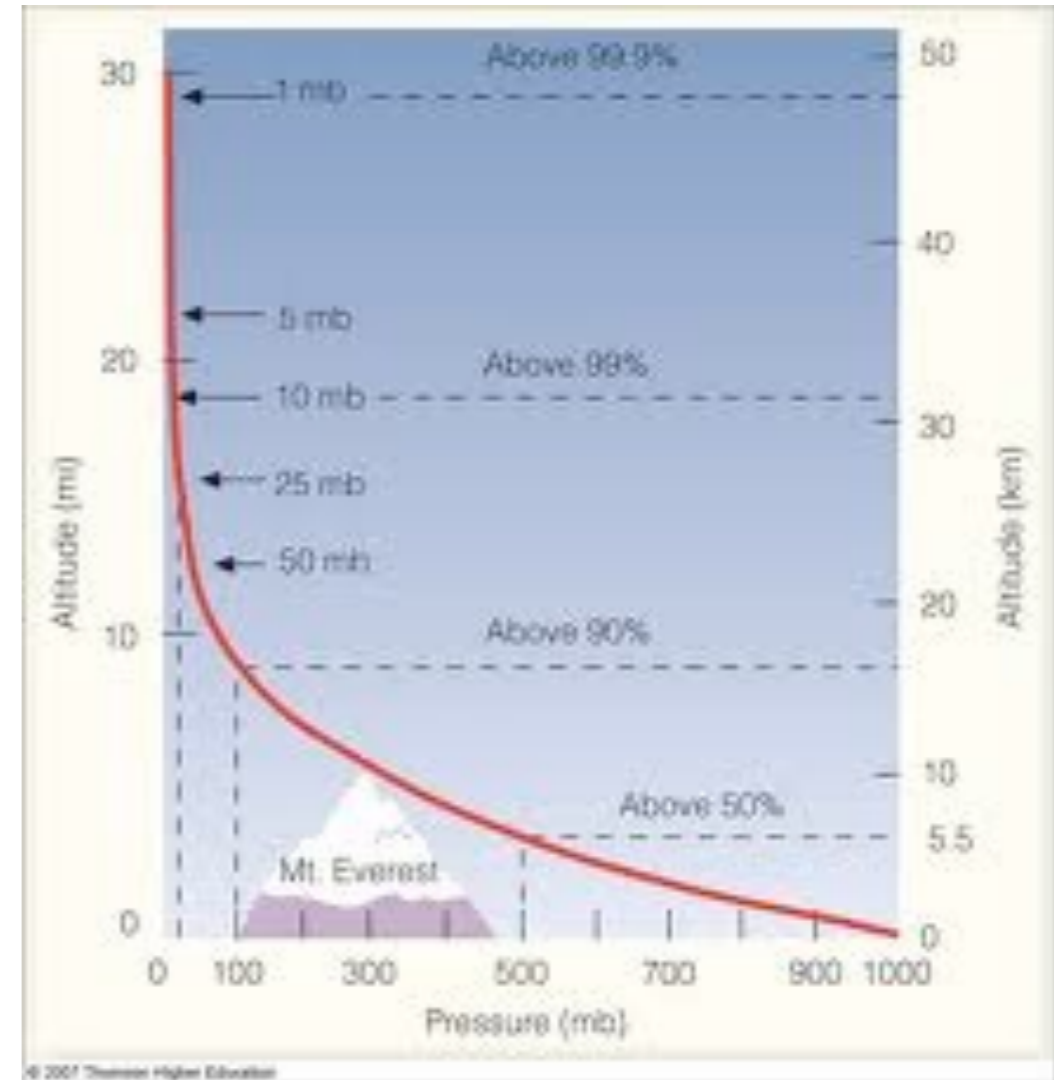
- **Absolute** relative to absolute zero (perfect vacuum)
- **Gage** relative to atmospheric pressure (>0 if $>P_{atm}$; <0 if $<P_{atm}$)
- if $P < P_{atm}$ we call it a vacuum

$$P_{abs} = P_{atm} + P_{gage}$$



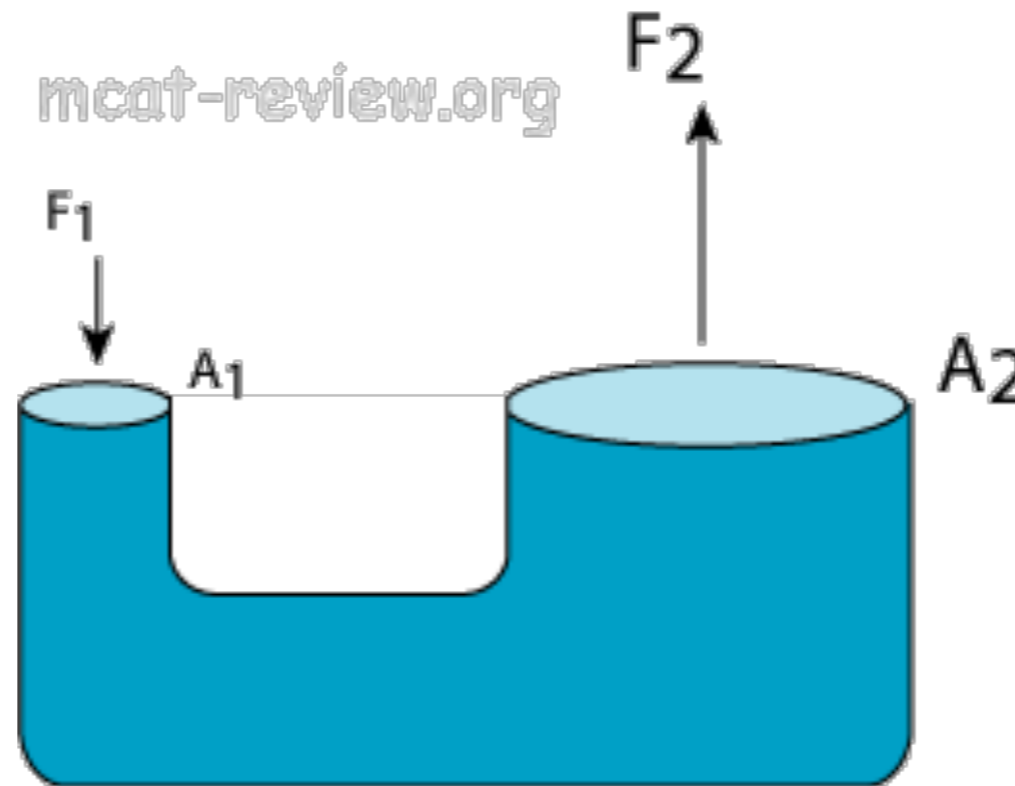
Pressure

- Atmospheric pressure is also called *barometric pressure* (1 bar = 10^5 Pa). It varies with elevation and changes in meteorological conditions
- Absolute pressure used for most problems related to gases/vapor
- Gage pressure related to liquids



Pascal's Law

- *All points in a connected body of constant-density fluid at rest are under the same pressure if they are at the same depth below the liquid surface.*



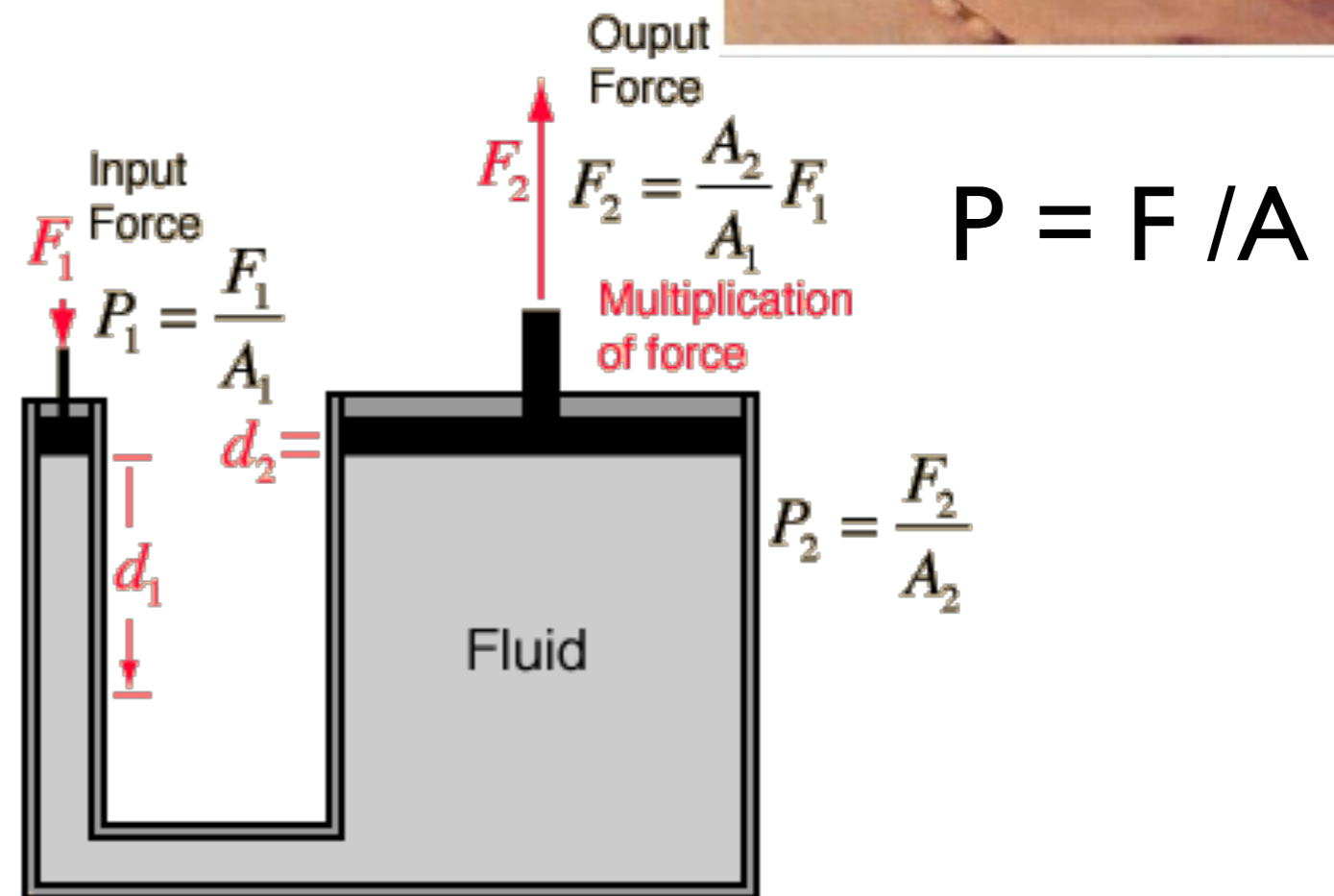
$$P_1 = P_2$$

$$\frac{F_1}{A_1} = \frac{F_2}{A_2}$$

Pascal's Law



- if you apply pressure on a liquid, the pressure is transmitted equally and unchanged to all parts of the liquid.

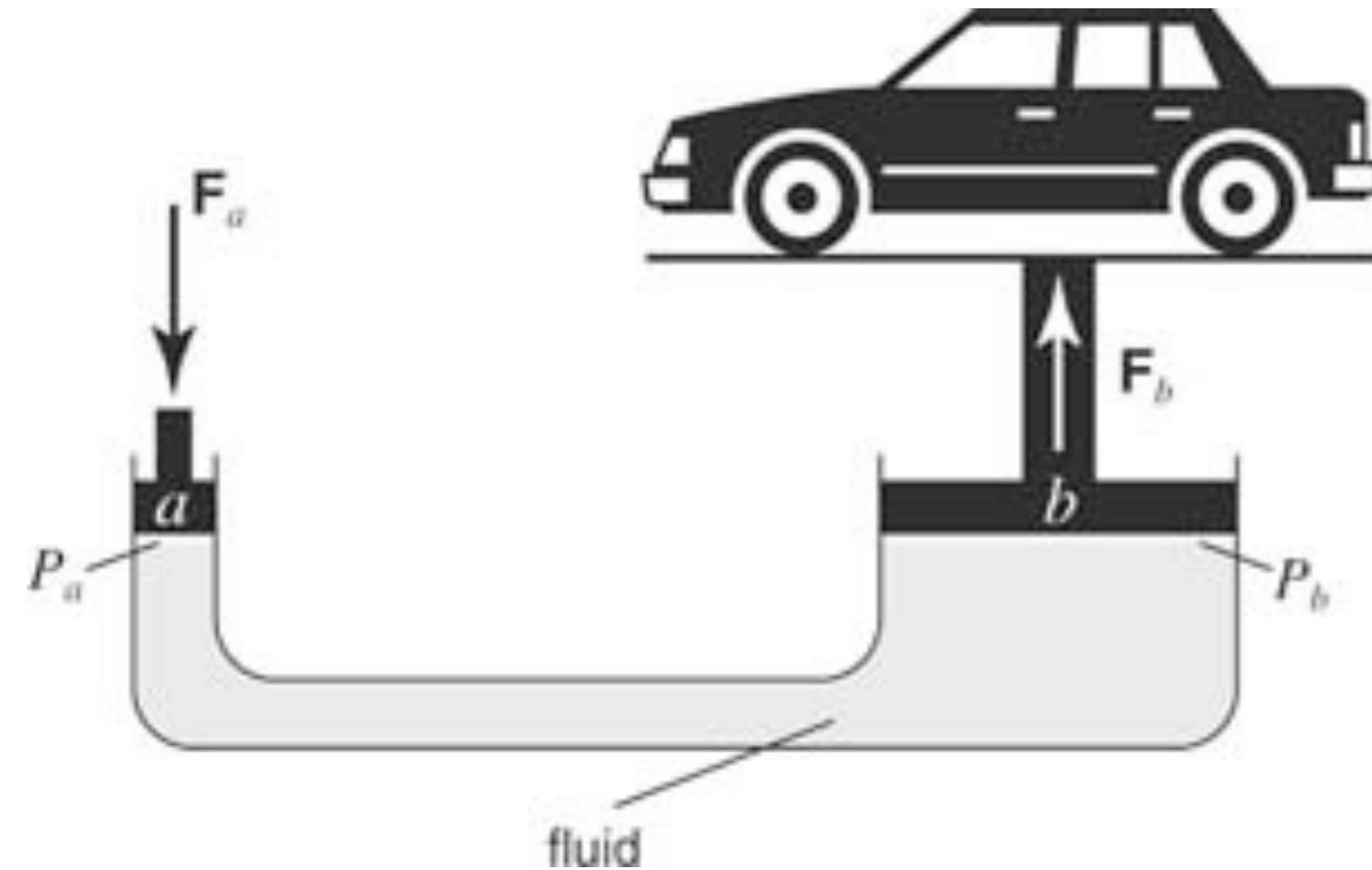


$$F_1 d_1 = F_2 d_2$$

$$d_1 = \frac{F_2}{F_1} d_2 = \frac{A_2}{A_1} d_2$$

You have to pay for the multiplied output force by exerting the smaller input force through a larger distance.

Automobile Hydraulic Lift



diameter $d_1 = 1.25\text{cm}$
diameter $d_2 = 25\text{ cm}$

Areas: $A_1 = 1.22$; $A_2 = 490$

--> $A_2/A_1 = 400$

--> $F_2 = 400F_1$

If car is 6000N ---> $F_1 = 6000\text{N}/400 = 15\text{N}$
to lift it 10 cm ---> $400 \times 10 = 40\text{ m}!!$

Static Equilibrium

In an **incompressible** fluid, where density is not a function of pressure, it is simple to determine the stability of the medium in static state.

1. **Stable:** if density decreases upward. A particle displaced upward would be at a level where density of the surrounding fluid is lower and the particle is forced to move back to its original level
2. **Unstable:** if density increases upward. A displaced particle would continue to move away from its original position.
3. **Neutral:** if the density is uniform.

Static Equilibrium

In a **compressible** medium the previous arguments do not hold. In a neutral state it is not density to be constant but rather entropy.

A particle displaced upward would expand adiabatically because of the decrease in pressure with height.

Displacing the particle upward, the original density and temperature would decrease to a new density and temperature according to their isentropic relations.

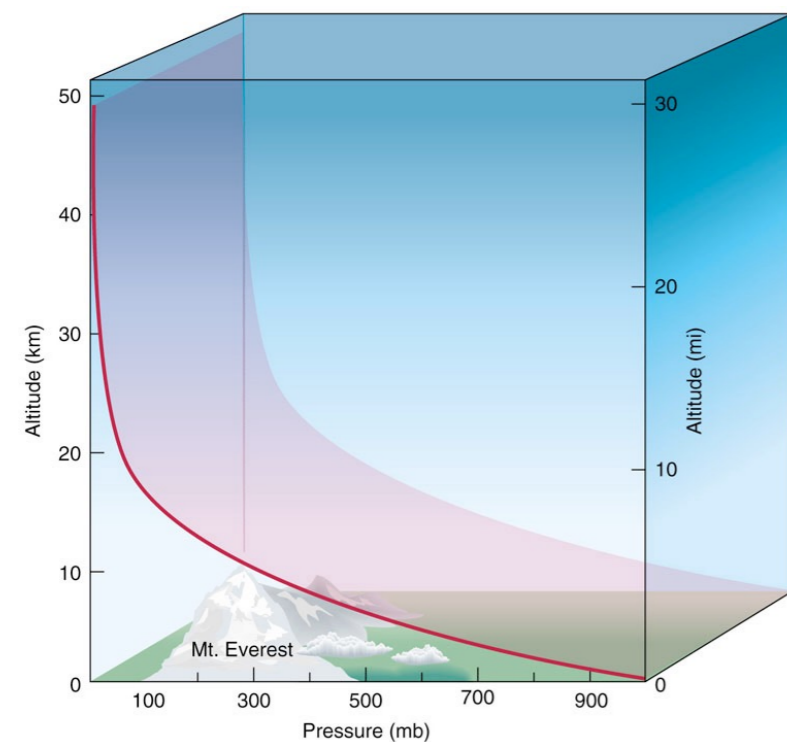
The particle would move back to its original position if the new density is lower than that of the surrounding level.

But if the properties of the surrounding air also vary with height so that entropy is uniform with height, the displaced particle would always find itself in a region where density is the same as its own density.

A neutral atmosphere (isentropic atmosphere) is thus one in which pressure, density and temperature decrease so that entropy is constant with height.

Buoyancy force

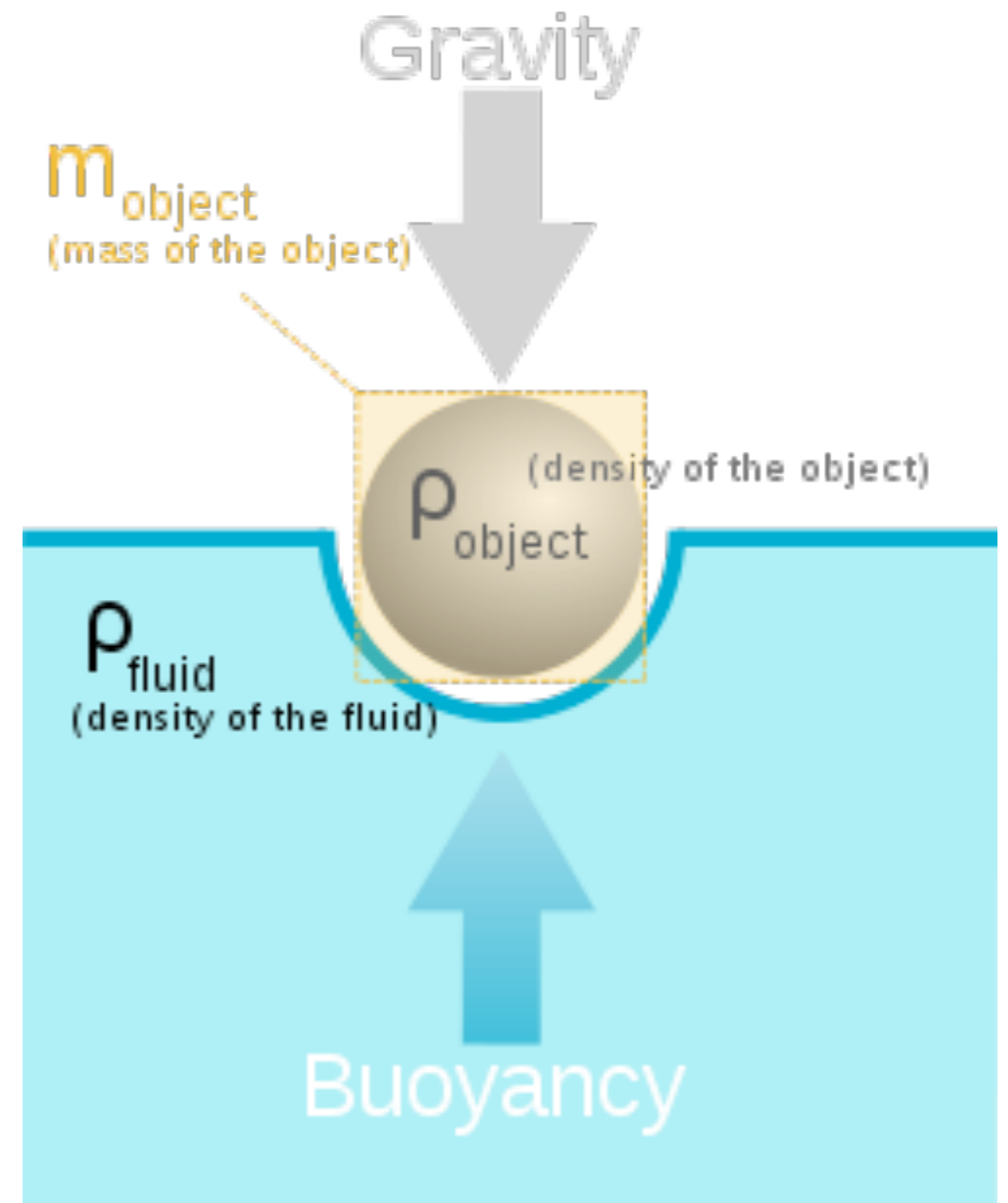
- Pressure in the atmosphere decreases with height (hydrostatics)
- Pressure force on balloon: bottom greater than at top



- Buoyancy force is the difference
- There is always a buoyancy force in a fluid, and it is always positive.

- A force exerted by a fluid that opposes an object's weight
- force is equal to weight of fluid displaced by the object
- $F_b = \rho(\text{fluid}) \times g \times V_{\text{disp}}$
- An object whose density (specific weight) is greater than that of the fluid in which it is submerged tends to sink ...

Buoyancy



is it easier to float in a pool or at sea?

- In equilibrium, the net Force must be zero, so that:

$$m g = \rho V_{disp} g = 0$$

*If the buoyancy of an object exceeds its weight, it tends to rise.
An object whose weight exceeds its buoyancy tends to sink.*

Archimedes' principle indicates that the upward buoyant force that is exerted on a body immersed in a fluid, whether fully or partially submerged, is equal to the weight of the fluid that the body displaces.

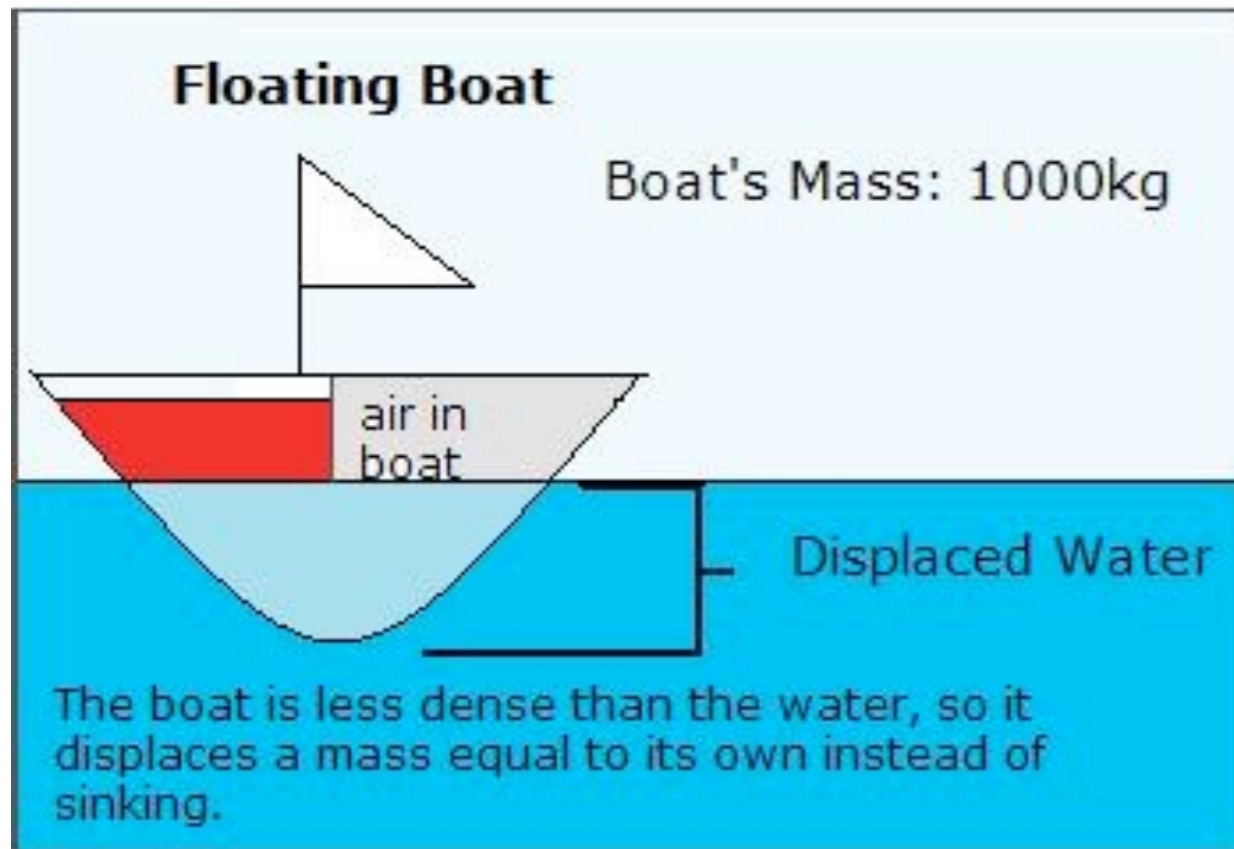
Materials of human body (density Kg/l):

muscle = 1.1; bone = 1.5; air = 0.0012

In fresh water (with air out): MEN all sink - WOMEN some float

In fresh water (with air in): MEN some sink - WOMEN all float

Buoyancy and floating



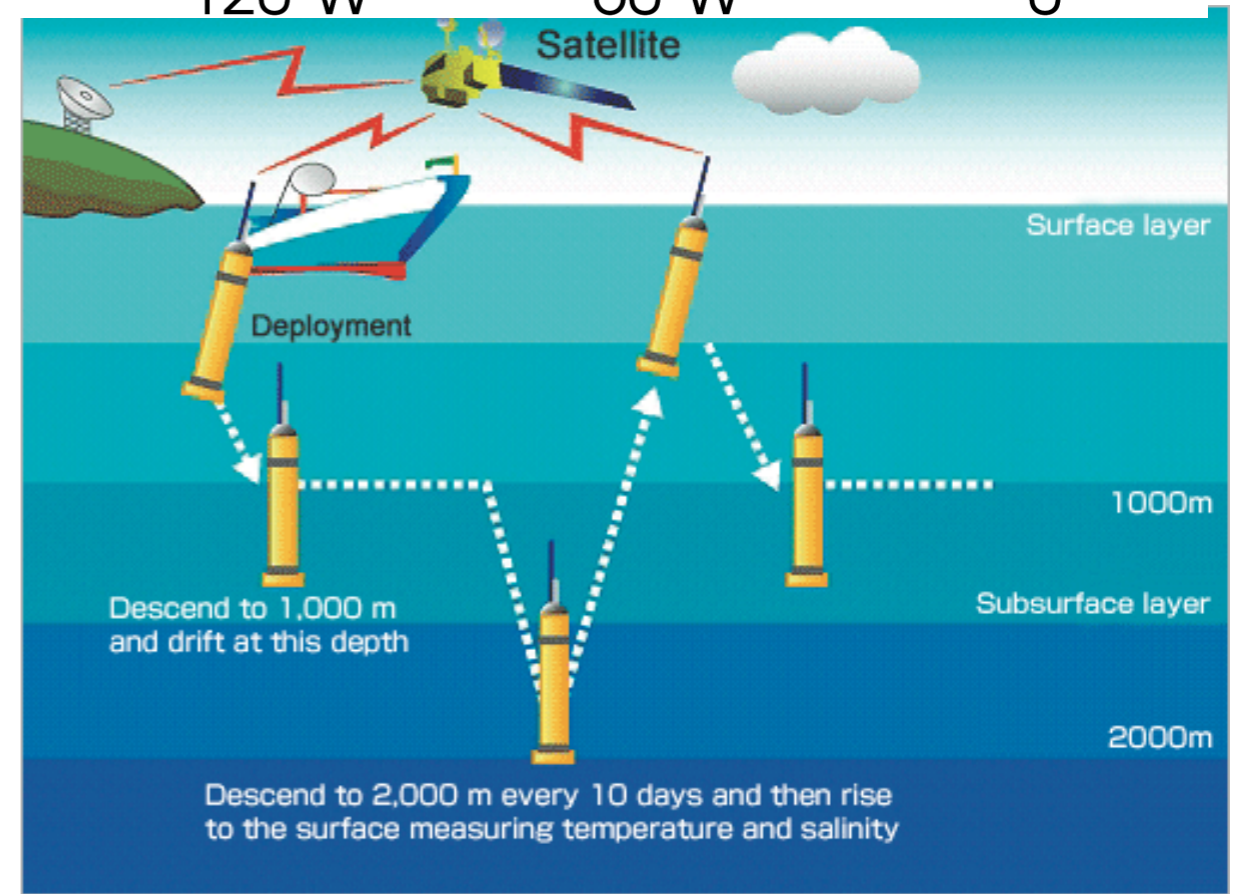
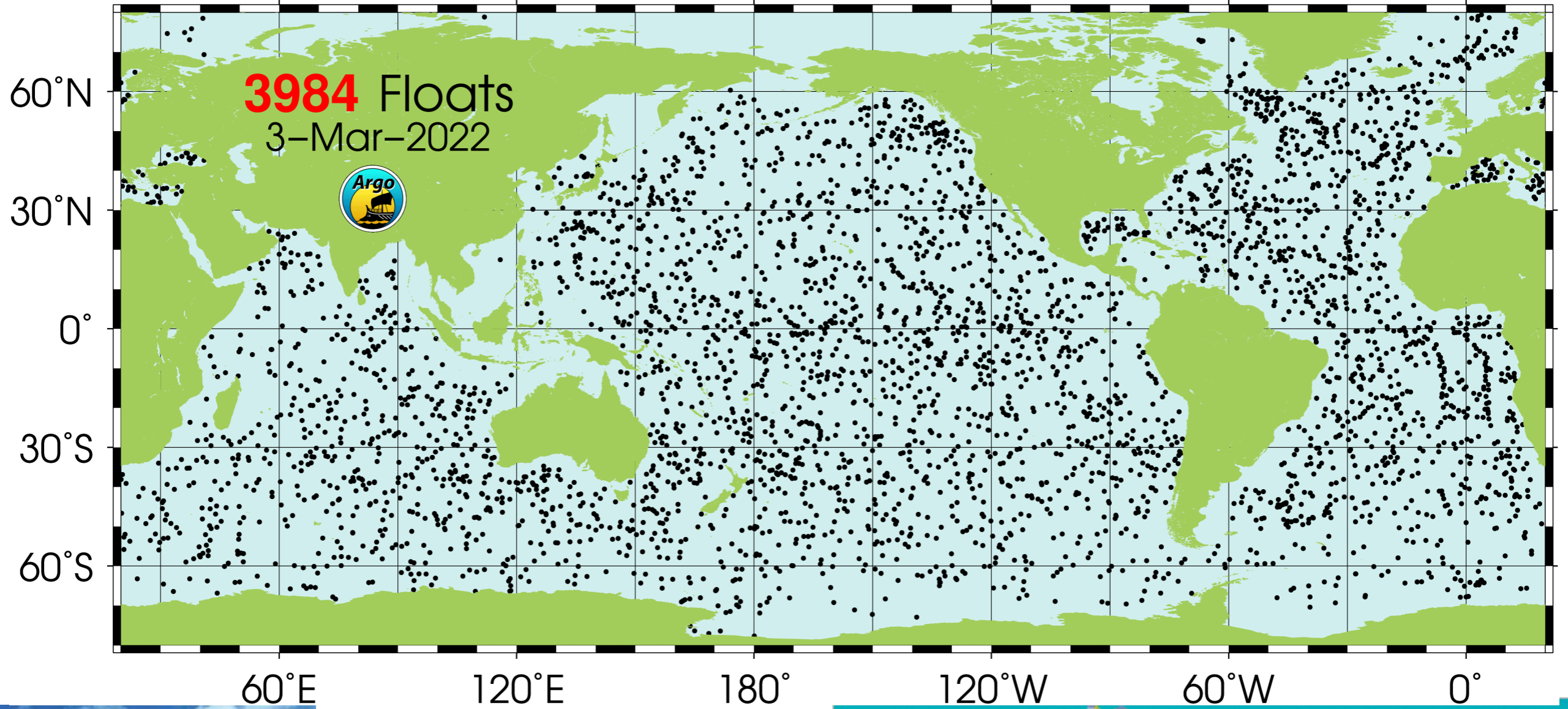
A block of iron dipped in water will sink, while the same metal block shaped like a boat will float.

Buoyancy is thus related to the **density**, **volume** and **shape** of the immersed body.

If $F_b = \rho \times g \times V_{disp}$, what is volume of the displaced water?

Stationarity $\rightarrow W_{boat} = F_b = \rho \times g \times V_{disp}$

$V_{underwater} = W_{boat} / (\rho \times g)$





A curiosity ... (the Iceberg)

- Roughly: $\rho_{\text{ice}} = 92\% \rho_{\text{water}}$

(another curiosity in itself ...)

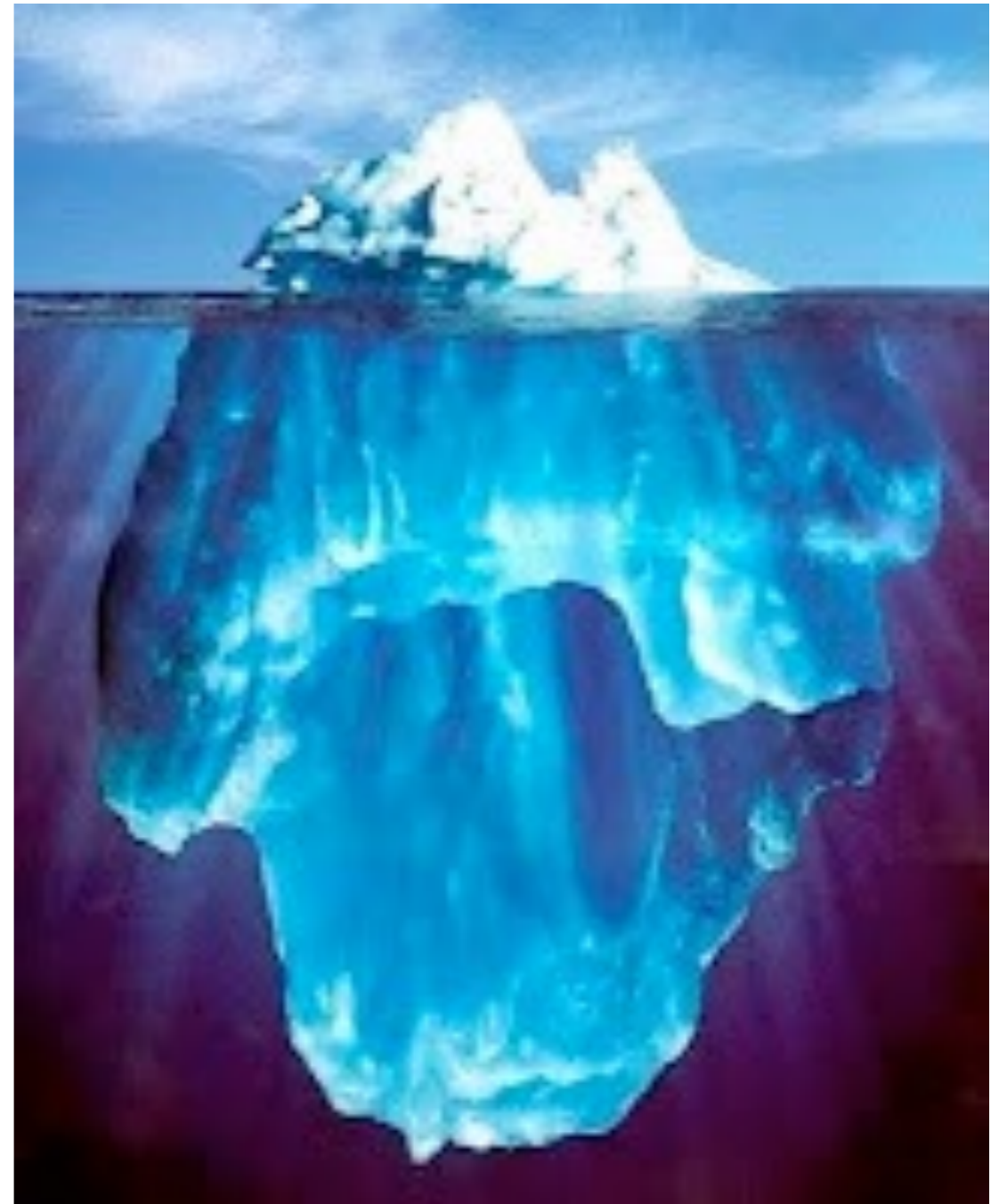
- It is in equilibrium,
so that $mg = F_b$
- how much of the
iceberg is
submerged?

A curiosity ... (the Iceberg)

- Roughly: $\rho_{\text{ice}} = 92\% \rho_{\text{water}}$

(another curiosity in itself ...)

- It is in equilibrium, so that $mg = F_b$
- how much of the iceberg is submerged?



- 92% (... “you only see the tip of the iceberg ...”)

Summary

- Fluid Statics
- Pascal's Law
- Absolute and Gage Pressure
- Buoyancy and Archimedes' Principle