

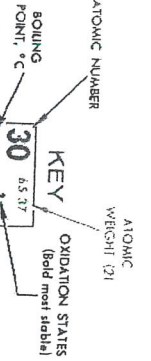
PERIODIC TABLE OF THE ELEMENTS

Table of Radioactive Isotopes

1	1.00797 H Hydrogen	2	4.0026 He Helium	3	6.939 Li Lithium	4	9.0122 Be Beryllium	5	10.811 B Boron	6	12.01115 C Carbon	7	14.0067 N Nitrogen	8	15.9994 O Oxygen	9	18.9984 F Fluorine	10	20.183 Ne Neon																
11	22.9898 Na Sodium	12	24.312 Mg Magnesium	13	26.9815 Al Aluminum	14	28.086 Si Silicon	15	30.9738 P Phosphorus	16	32.064 S Sulfur	17	35.453 Cl Chlorine	18	39.948 Ar Argon	19	39.102 K Potassium	20	40.08 Ca Calcium																
21	44.956 Sc Scandium	22	47.90 Ti Titanium	23	50.942 V Vanadium	24	51.996 Cr Chromium	25	54.938 Mn Manganese	26	55.847 Fe Iron	27	58.933 Co Cobalt	28	58.71 Ni Nickel	29	63.54 Cu Copper	30	65.37 Zn Zinc	31	69.72 Ga Gallium	32	72.59 Ge Germanium												
37	85.47 Rb Rubidium	38	87.62 Sr Strontium	39	88.905 Y Yttrium	40	91.224 Zr Zirconium	41	92.906 Nb Niobium	42	95.94 Mo Molybdenum	43	98.906 Tc Technetium	44	101.07 Ru Ruthenium	45	102.905 Rh Rhodium	46	106.4 Pd Palladium	47	107.87 Ag Silver	48	112.40 Cd Cadmium	49	114.82 In Indium	50	118.69 Sn Tin								
55	132.905 Cs Cesium	56	137.34 Ba Barium	57	138.91 La Lanthanum	72	178.49 Hf Hafnium	73	180.948 Ta Tantalum	74	183.85 W Tungsten	75	186.2 Re Rhenium	76	186.94 Os Osmium	77	192.2 Ir Iridium	78	195.09 Pt Platinum	79	196.967 Au Gold	80	200.59 Hg Mercury	81	204.37 Tl Thallium	82	207.19 Pb Lead	83	208.980 Bi Bismuth	84	210 Po Polonium	85	210 At Astatine	86	222 Rn Radon
87	223 Fr Francium	88	226 Ra Radium	89	227 Ac Actinium	104		105		106		107		108		109		110		111		112		113		114		115		116		117		118	

Naturally occurring radioactive isotopes are indicated by a blue mass number. Half lives are in parentheses where s, m, h, d and y stand for seconds, minutes, hours, days and years respectively. The symbol denoting the mode of decay and resulting radiation are defined as follows:

- α - alpha particle
- β^- - beta particle
- β^+ - positron
- K - K-capture
- L - L-capture
- SF - spontaneous fission
- γ - gamma ray
- IC - internal electron conversion



KEY: Oxidation States (bold most stable)

BOILING POINT, °C

MELTING POINT, °C

DENSITY (g/ml)

SYMBOL (I)

ELECTRON STRUCTURE

58	140.12 Ce Cerium	59	140.907 Pr Praseodymium	60	144.24 Nd Neodymium	61	147.9 Pm Promethium	62	150.35 Sm Samarium	63	151.96 Eu Europium	64	157.25 Gd Gadolinium	65	158.934 Tb Terbium	66	162.50 Dy Dysprosium	67	164.930 Ho Holmium	68	167.26 Er Erbium	69	168.934 Tm Thulium	70	173.04 Yb Ytterbium	71	174.97 Lu Lutetium
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NOTES:

(1) Black - solid, Red - gas, Blue - liquid, Outline - synthetically prepared.

(2) Based upon carbon - 12. (I) indicates most stable or best known isotope.

(3) Values for gaseous elements are for liquids at the boiling point.

SARGENT-WELCH
SARGENT-WELCH SCIENTIFIC COMPANY
7300 LINDER AVENUE, SKOKIE, ILLINOIS 60076