

- Sensibilità paragonabile a quella di SPECT e PET
- Consente imaging molecolare (in vitro, ex-vivo)
- Applicazioni in chirurgia guidata ed endoscopia
- Possibilità di agenti switchable (responsive)
- Possibilità di *time-resolved detection* (autofluorescenza di fondo)
- No quantificazione

- Window
- Stokes shift
- Brightness
- Stability



Esempio di *switchable fluorescent probe* sensore di ¹O₂

(a) Singlet Oxygen Probes

Key reaction: Endoperoxide formation



Reaction scheme for detection of singlet oxygen



FRET *fluorescence* – *resonance energy transfer*



II FRET è attivo solo quando i due cromofori distano fra loro pochi nanometri (<10 nm) e hanno la giusta orientazione reciproca.



quantum dots (QD) nano-cristalli di semiconduttori (e.g. CdSe)



Ampio intervallo di frequenze di eccitazione, banda di emissione stretta, molto intensa e modulabile con le dimensioni del QD

Complessi dei lantanidi



Upconverting QDs e LnNPs



Dark-field fluorescence imaging con AuNP



EGFR = *epidermal growth factor receptor*, marcatore tumorale

Complessi polipiridilici di Ru(II) come DNA light switch



G-quadruplex sensing



Sviluppi futuri

Multimodal imaging agents and theranostics









Si stima che nel 2050 a livello mondiale le morti da infezione batterica – dovute soltanto a ceppi di batteri resistenti – arriveranno a 50 milioni all'anno (vs 8.8 milioni dovute al cancro).

Timeline: a brief history of the use of Silver



Metallo massivo – Nanoparticelle – Sali (Ag+)

Multiple mechanism of action of Ag⁺ ions



WHY CHOOSE SILVER PLATING FOR MEDICAL DEVICES?

Silver salts



silver sulfadiazine





Silver nanoparticles (AgNPs)



Estimated 2014 production of commercial AgNPs: 320 t





Ag⁺ release from AgNPs



Uptake of AgNPs



Multiple mechanism of action of Ag⁺ ions



Multiple mechanism of action of AgNP's



Antibacterial Ag-NHC compounds





Drug candidate: high activity against tobramycin-resistant pathogenic bacteria *in vitro* as well as *in vivo*. Improved stability to hydrolysis due to the electron-withdrawing CI substituents that pull electron density from the carbene

Other strategies: metal modification of known antibiotics



Platensimycin mimics







Other metals: As, Sb, Bi, Hg



Antimicrobial agent introduced in early 1900 for the treatment of the deadly bacterial infection *Syphilis*. Later replaced by modern antibiotics

Antiparasitic compounds



melarsoprol

treatment of sleeping sickness (African trypanosomiasis)

Anti-leishmaniasis compounds



Sb, reduced to Sb(III), inhibits *trypanothione reductase*, an essential enzyme of the parasite

Infezioni da Helicobacter pylori





bismuth subsalicylate

The pink stuff (introduced 1901)



CBS



bismuto subcitrato colloidale

Infezioni da Helicobacter pylori



ranitidine bismuth citrate



Antibacterial mercury compounds



FosrenolTM: $La_2(CO_3)_3 \cdot 4H_2O$ a success story



FOSRENOL Chewable Tablets FOSRENOL Oral Powder Phosphorus Burden in ESRD Conferences and Resources

Patient

Support

To reduce serum phosphate in patients with end-stage renal disease (ESRD)

HELP IT FALL WITH FOSRENOL* (lanthanum carbonate)

*Phosphorus reductions maintained for up to 3 years in patients remaining on therapy (n=46)¹⁻³

- FOSRENOL Chewable Tablets: Approved in 2004 and used in US clinical practices for more than a decade^{4,5} LEARN MORE
- FOSRENOL Oral Powder: Available since May 2015, offering you another approved administration option⁶ LEARN MORE

Approved by FDA in 2004 for the treatment of hyperphosphatemia (increased phosphate levels in serum) in patients with end stage renal disease.

Phosphate metabolism is intimately linked with calcium metabolism, and is regulated by parathyroid hormone (PTH) and vitamin D



Pathological consequences of hyperphosphatemia: cardiac and vascular tissue calcification, bone malformations in the joints

The ideal phosphate binder should:

- have a high affinity for phosphate
- be able to bind dietary phosphate rapidly in the guts
- have low solubility
- little or no systemic absorption
- be non-toxic
- be available as a palatable oral dosage form, with a low pill burden

Calcium phosphate binders (e.g. calcium carbonate or calcium acetate) are effective....however, calcium can be absorbed, resulting in hypercalcemia and increased risk of cardiovascular calcification.

Fosrenol[™]: a success story

Among the many lanthanide salts screened, $La_2(CO_3)_3 \cdot 4H_2O$ possessed the best phosphate binding properties:

- Optimal binding at pH 3–5, but retains binding activity in the full pH range of 1–7
- It is very insoluble and the La³⁺ cation does not cross biological membranes (when given by the oral route, >90% excreted in the feces, and <0.001% absorbed)
- No toxicity observed in animal studies, in particular no direct effects on calcium, vitamin D, or PTH metabolism

FosrenolTM represents a significant improvement in treatment options for patients with end-stage renal disease.

- in the acidic environment of the stomach lanthanum carbonate dissociates sufficiently to allow formation of a highly insoluble phosphate.
- It has the required pharmacokinetic properties, it is poorly absorbed, with both the parent salt and the phosphate product being eliminated in the feces.
- Because of the lack of absorption it has no systemic toxicity, it has no detrimental effect on calcium, vitamin D or PTH metabolism, and is safe and well tolerated.
- Its effectiveness as a phosphate binder results in a lower pill burden for patients, an advantage over competing medications.