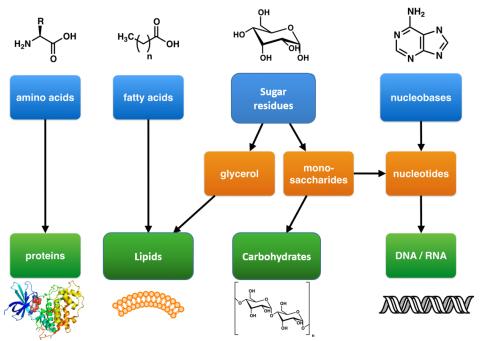
Prof. Sabrina Pricl

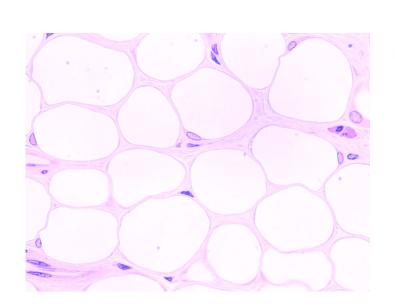
A.Y. 2023-2024

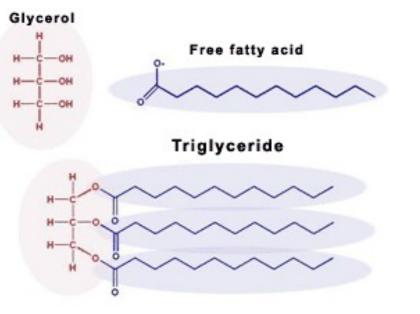
# Lesson 2 Recognizing Macromolecules

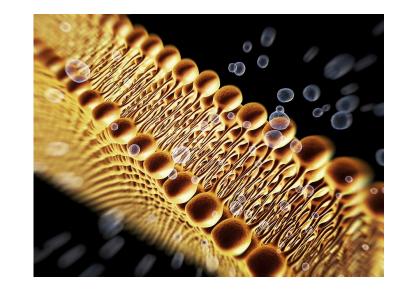


- Macro = big  $\rightarrow$  Macromolecules = BIG molecules
- 4 major classes of macromolecules in cells:

• Lipids

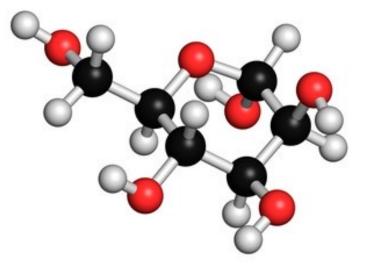








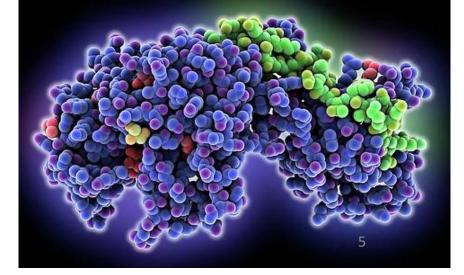
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  - Carbohydrates



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  - Carbohydrates
  - Nucleic acids



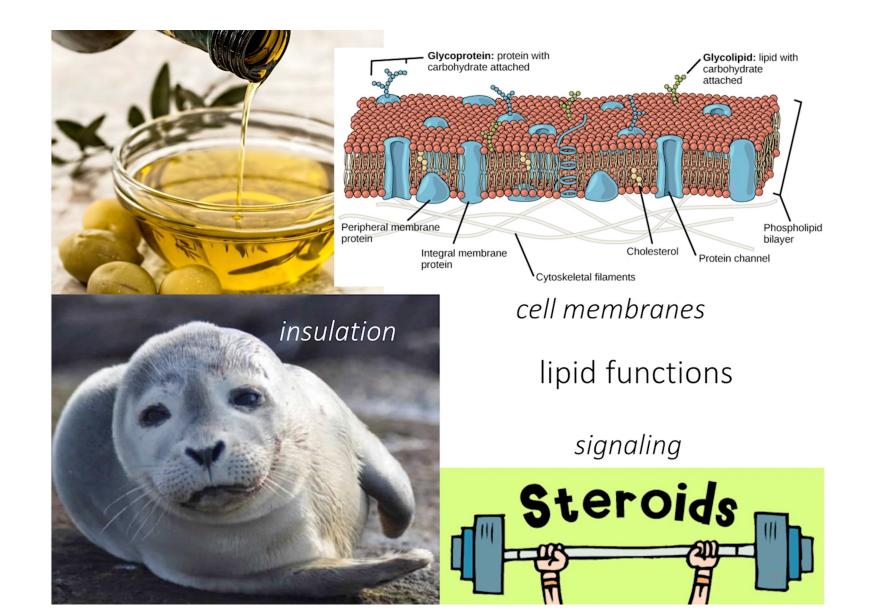
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  - Nucleic acids
  - Proteins

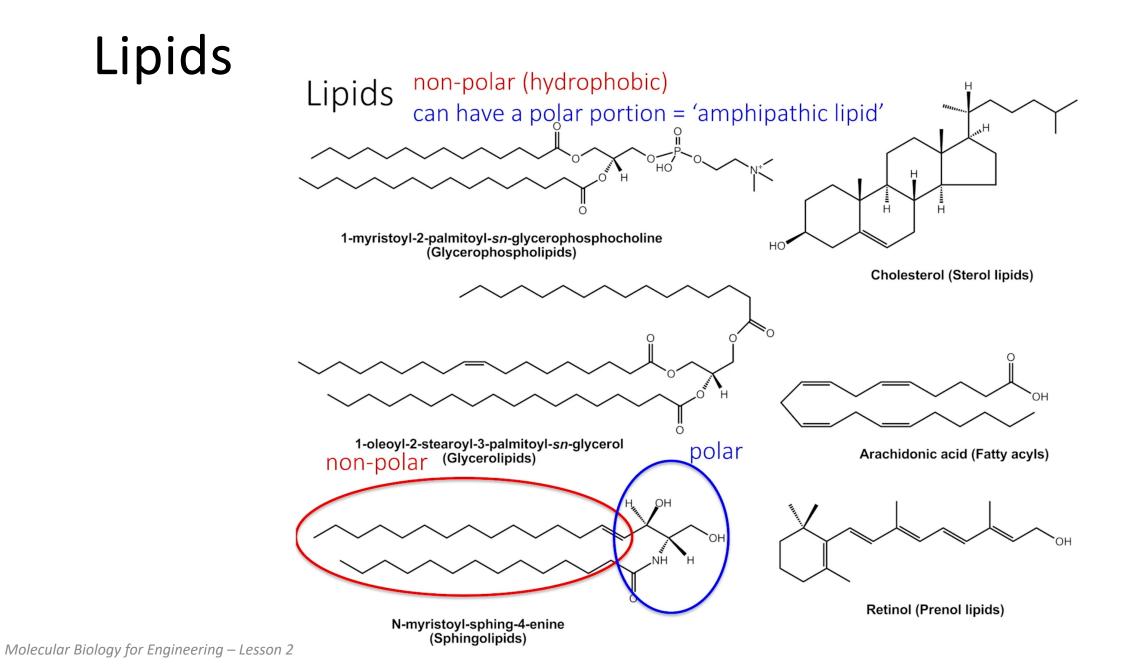


- Macro = big  $\rightarrow$  Macromolecules = BIG molecules
- 4 major classes of macromolecules in cells:
  - Lipids
  - Carbohydrates
  - Nucleic acids
  - Proteins
- They are often polymers -> (monomer = M, polymer = M<sub>n</sub>)

# Lipids

- Membranes, signals, energy storage, protection
- Non-polar (hydrophobic) (key attribute)
- Or amphipathic (partly polar)
- Long chain or small



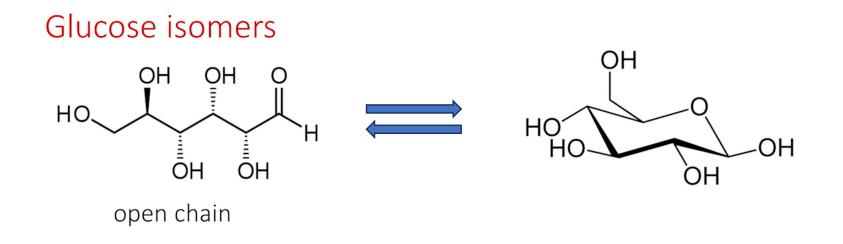


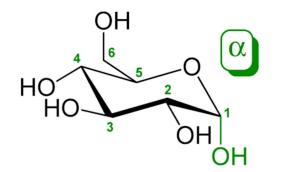
## Carbohydrates

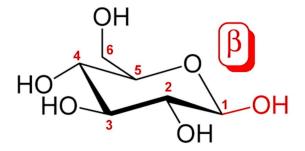
- Energy, information, structure
- Basic chemical formula CH<sub>2</sub>O (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>)
- Monomers (Ms) = monosaccharides (sugars)
- Polymers = polysaccharides (glycogen, starch, cellulose...)
- Ms joined by glycosidic bonds C-O-C

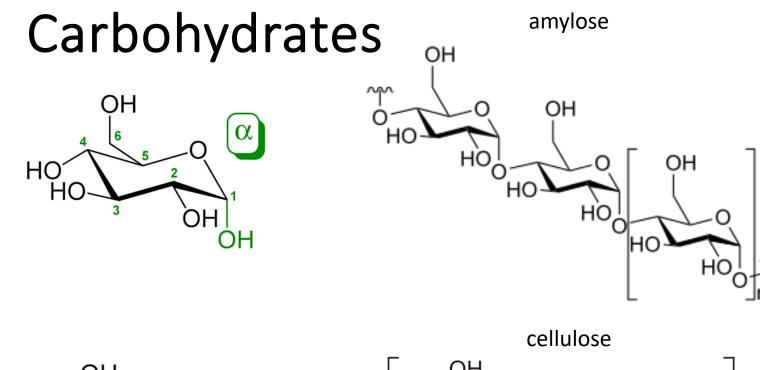


## Carbohydrates

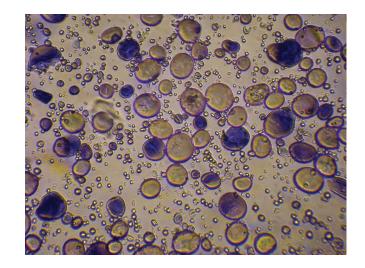


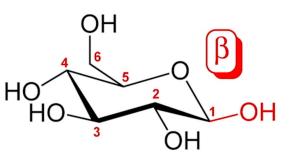






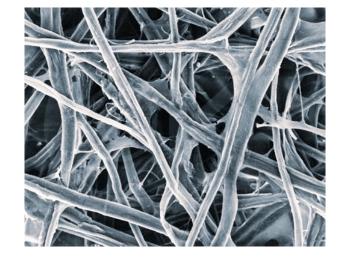
glycogen





OH HO OH OH OH

 $\neg$ n



cellulose fibers

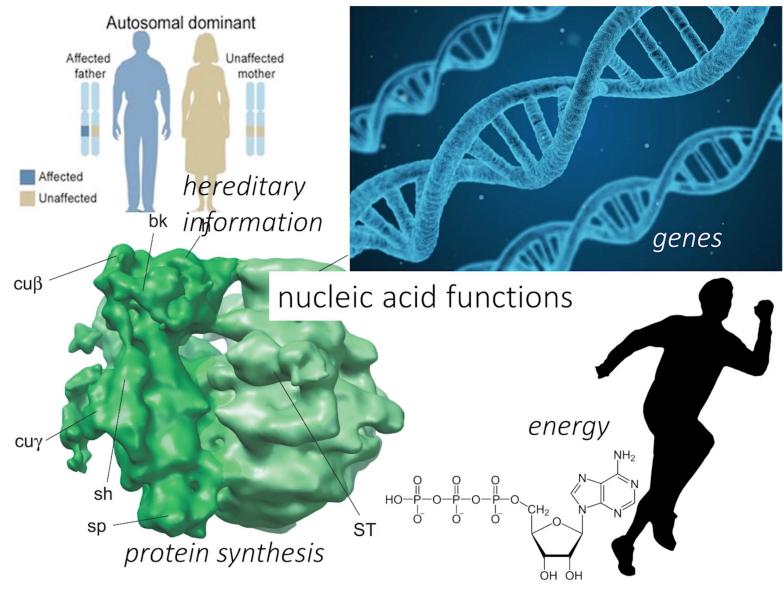
In starch  $\rightarrow \alpha$ -1,4 in cellulose  $\rightarrow \beta$ -1,4. WHAT CAN WE DIGEST AND WHY?

## Nucleic acids

- Hereditary information (genes), energy
- Monomer (Ms) = nucleotides
- Polymers = DNA or RNA
- The nucleotide structure is stereotypical:

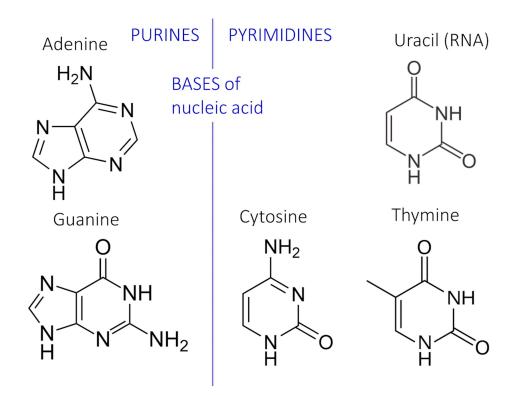
#### Phosphate-sugar-base (P-S-B)

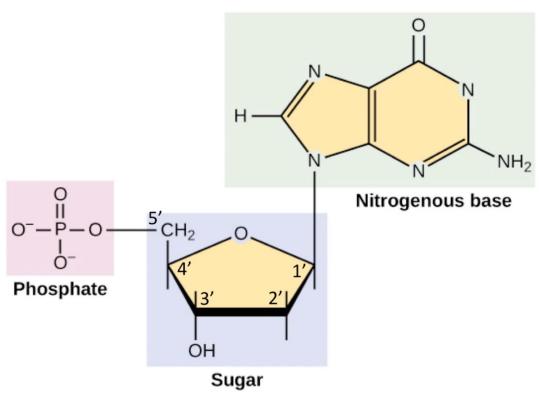
- Sugar = (5C) ribose (RNA) or deoxyribose (DNA)
- 5 bases:
  - Adenine (A), Guanine (G) =
    PURINES
  - Cytosine (C), Thymine (T), Uracil (U) = PYRIMIDINES
- A,G,C,**T** = DNA
- A,G,C,**U** = RNA



## Nucleic acids

#### Nucleotide





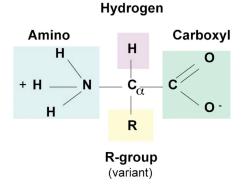
Note that the carbon atoms of the sugar are called C1', C2' etc C1' is where the base attaches

C2' can have attached a H (as here, for DNA) or a OH group (for RNA) C3' is very important and where the next nucleotide will join C5' is where the phosphate group attaches

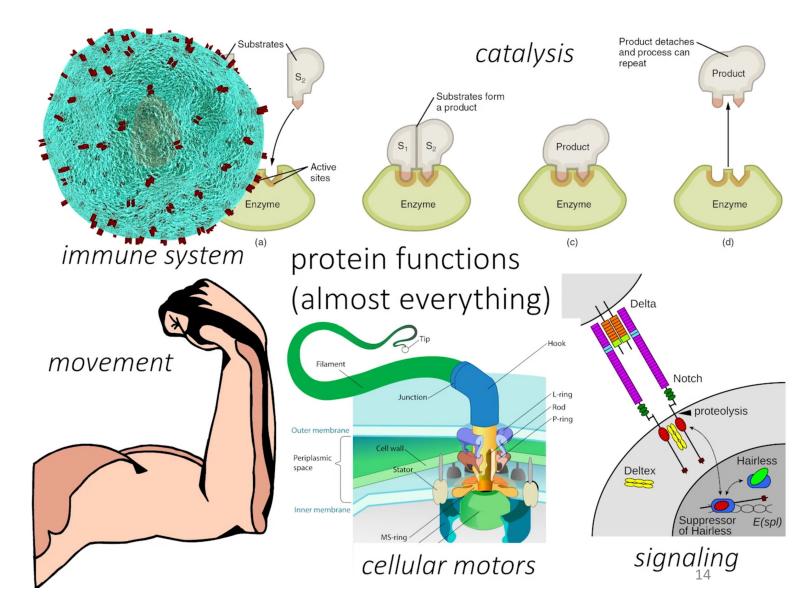
## Proteins

- Gene expression products
- Monomers (Ms) = amino acids
- 20 natural common amino acids
- Polymer = protein
- Amino acid structure:

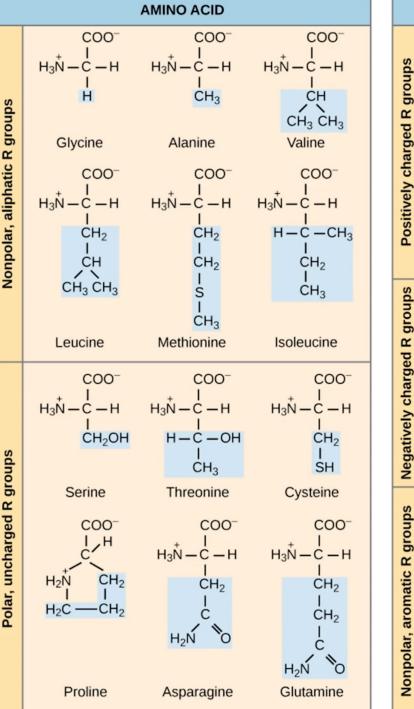
#### Alpha C + NH<sub>2</sub> + COOH + R (side group)

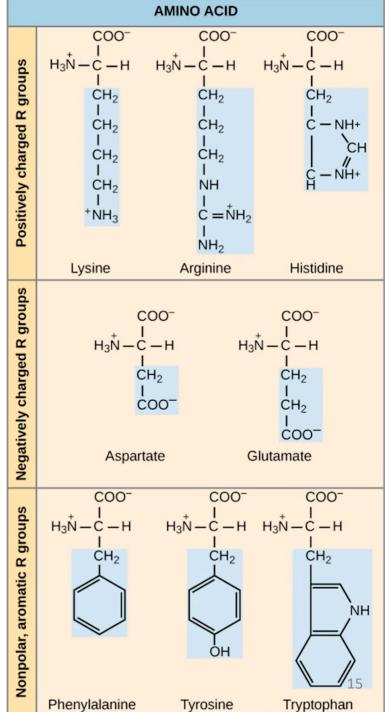


- R can be polar, non-polar, charged, uncharged
- Three or one letter code:
  - Valine = VAL
  - Valine = V



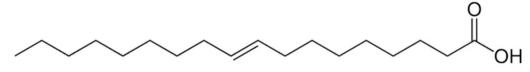
### Proteins



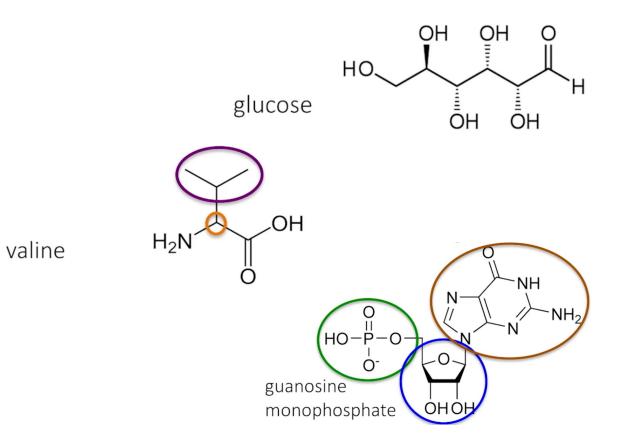


Molecular Biology for Engineering – Lesson 2

### Recognizing macromolecules - review



trans-oleic acid



Lipids: nonpolar/hydrophobic

Carbohydrates: CH<sub>2</sub>O formula/polar

Proteins:  $M = amino acid = \alpha C bonded to NH_2, COOH and R, R = polar/nonpolar$ 

Nucleic acids: M = nucleotide = phosphate + sugar + base, polar