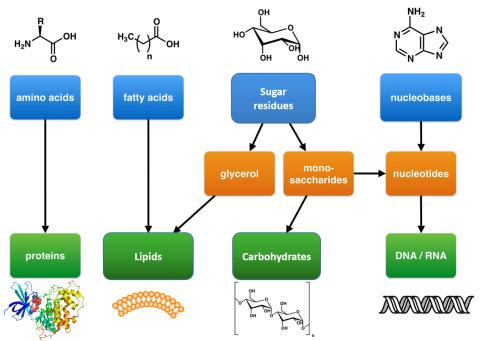
Prof. Sabrina Pricl

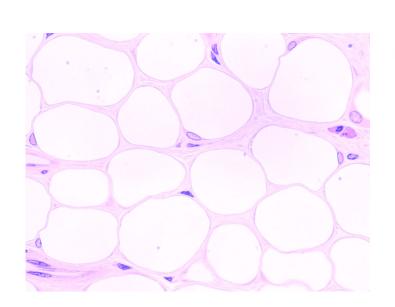
A.Y. 2023-2024

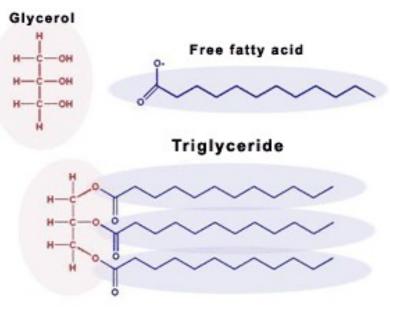
Lesson 2 Recognizing Macromolecules

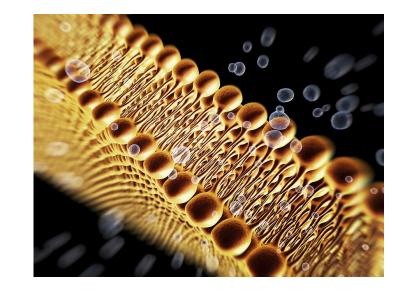


- Macro = big \rightarrow Macromolecules = BIG molecules
- 4 major classes of macromolecules in cells:

• Lipids

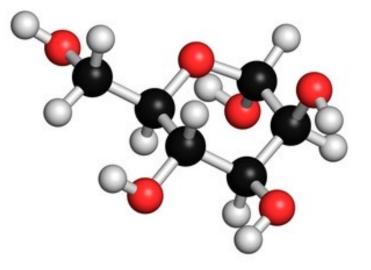








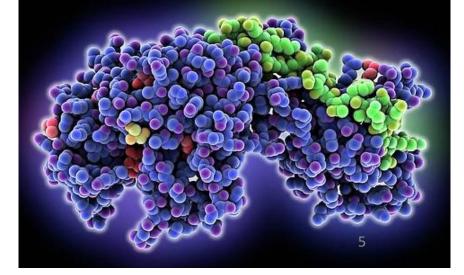
- Macro = big \rightarrow Macromolecules = BIG molecules
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 - Lipids
 - Carbohydrates



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- 4 major classes of macromolecules in cells:
 - Lipids
 - Carbohydrates
 - Nucleic acids



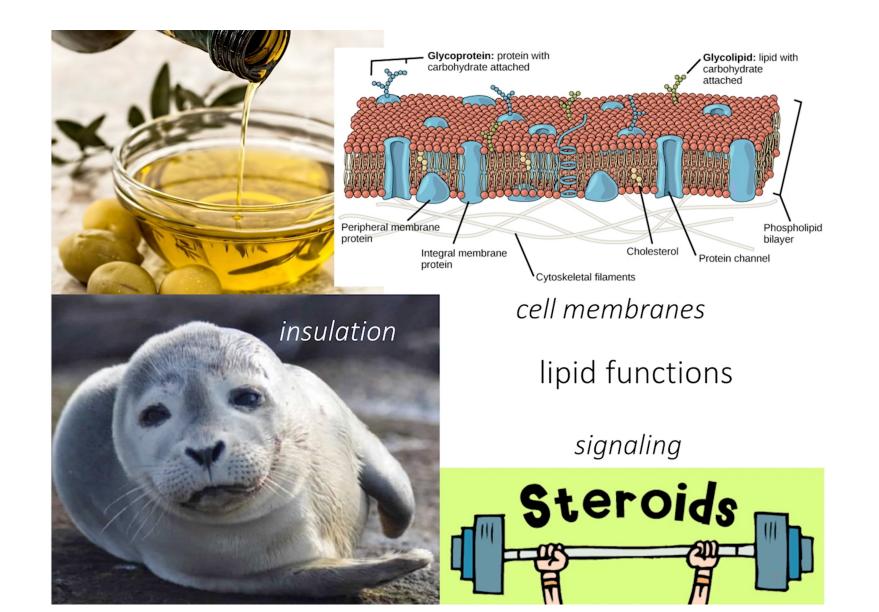
- Macro = big \rightarrow Macromolecules = BIG molecules
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 - Lipids
 - Carbohydrates
 - Nucleic acids
 - Proteins

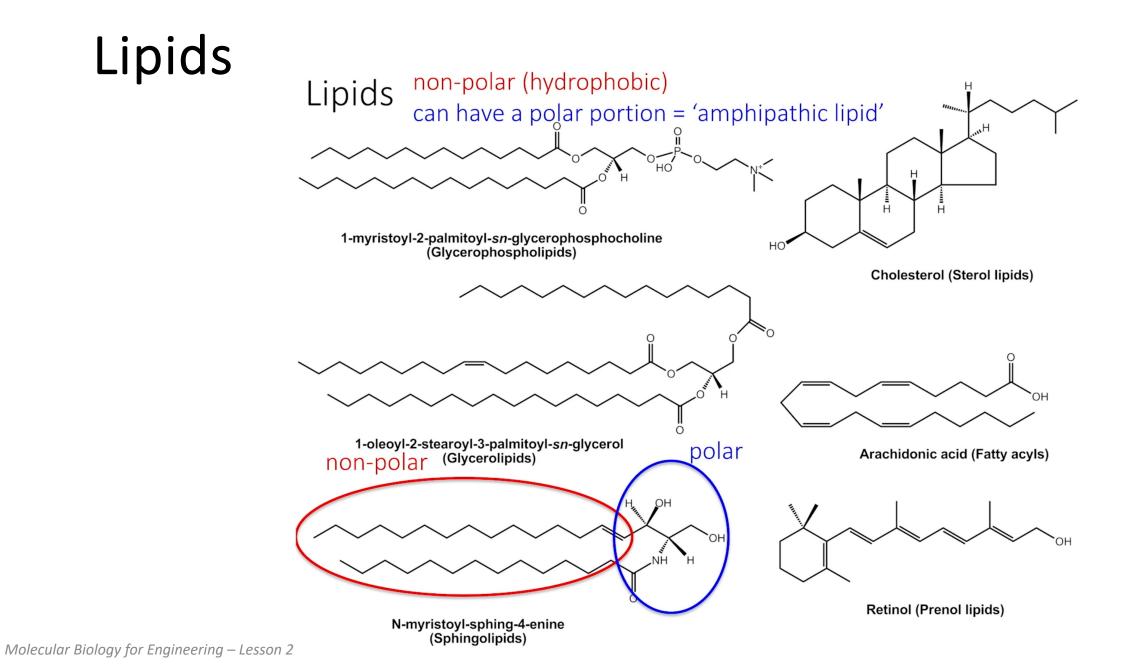


- Macro = big \rightarrow Macromolecules = BIG molecules
- 4 major classes of macromolecules in cells:
 - Lipids
 - Carbohydrates
 - Nucleic acids
 - Proteins
- They are often polymers -> (monomer = M, polymer = M_n)

Lipids

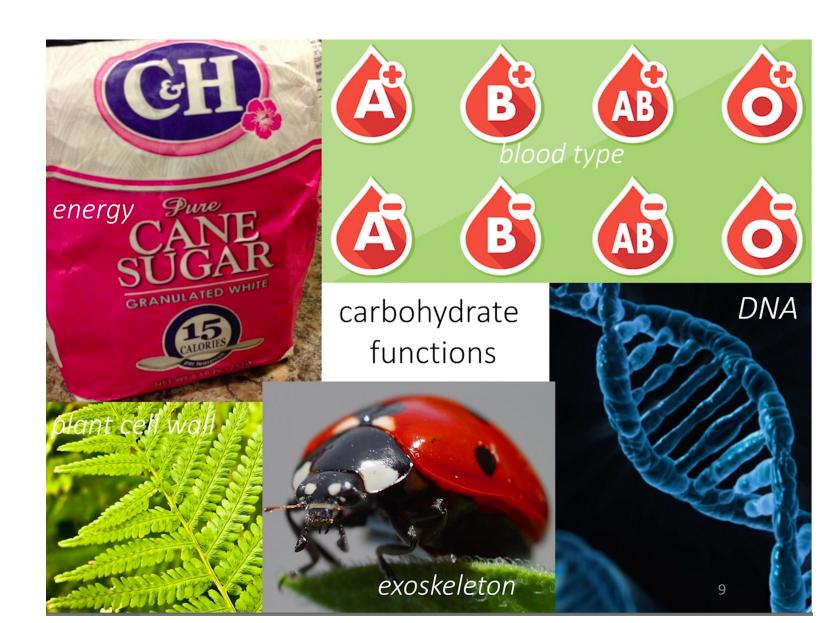
- Membranes, signals, energy storage, protection
- Non-polar (hydrophobic) (key attribute)
- Or amphipathic (partly polar)
- Long chain or small



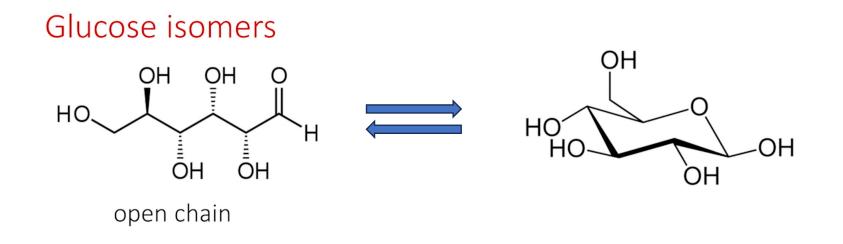


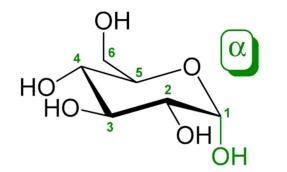
Carbohydrates

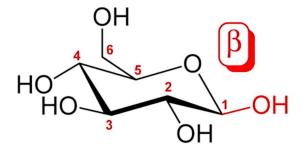
- Energy, information, structure
- Basic chemical formula CH₂O (C₆H₁₂O₆)
- Monomers (Ms) = monosaccharides (sugars)
- Polymers = polysaccharides (glycogen, starch, cellulose...)
- Ms joined by glycosidic bonds C-O-C

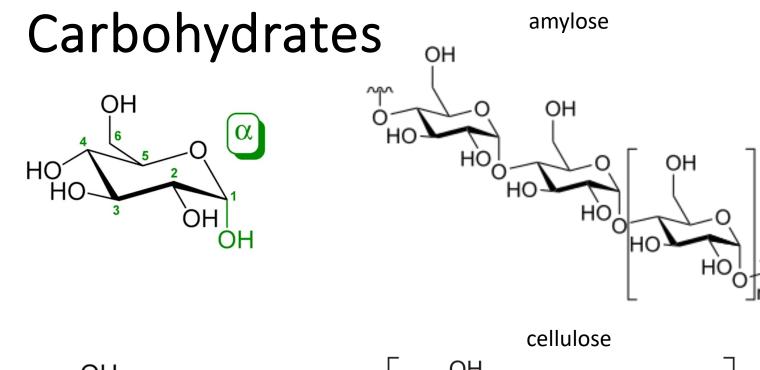


Carbohydrates

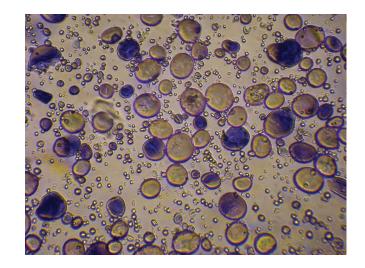


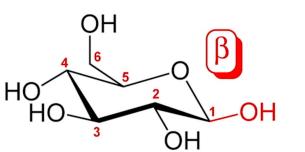






glycogen





OH HO OH OH OH

 \neg n



cellulose fibers

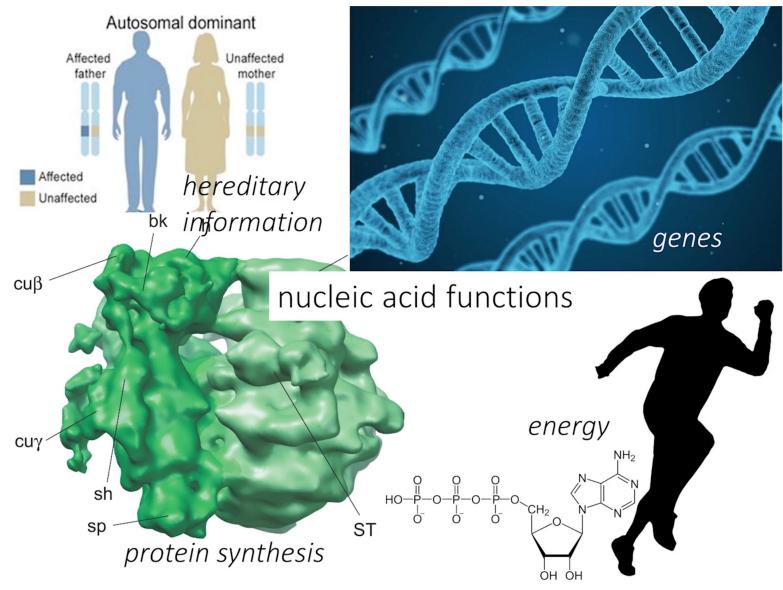
In starch $\rightarrow \alpha$ -1,4 in cellulose $\rightarrow \beta$ -1,4. WHAT CAN WE DIGEST AND WHY?

Nucleic acids

- Hereditary information (genes), energy
- Monomer (Ms) = nucleotides
- Polymers = DNA or RNA
- The nucleotide structure is stereotypical:

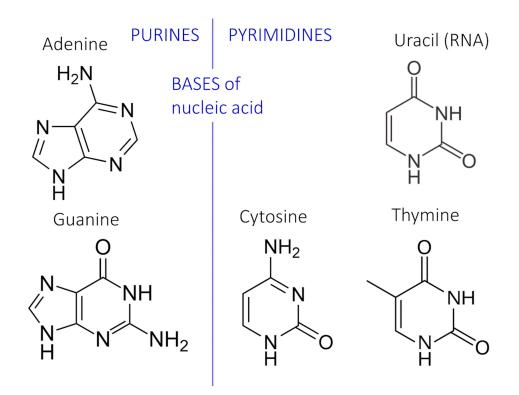
Phosphate-sugar-base (P-S-B)

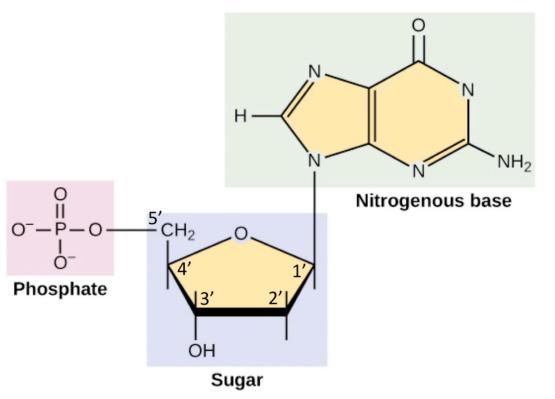
- Sugar = (5C) ribose (RNA) or deoxyribose (DNA)
- 5 bases:
 - Adenine (A), Guanine (G) =
 PURINES
 - Cytosine (C), Thymine (T), Uracil (U) = PYRIMIDINES
- A,G,C,**T** = DNA
- A,G,C,**U** = RNA



Nucleic acids

Nucleotide





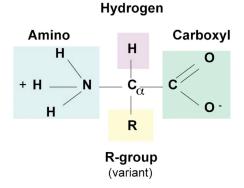
Note that the carbon atoms of the sugar are called C1', C2' etc C1' is where the base attaches

C2' can have attached a H (as here, for DNA) or a OH group (for RNA) C3' is very important and where the next nucleotide will join C5' is where the phosphate group attaches

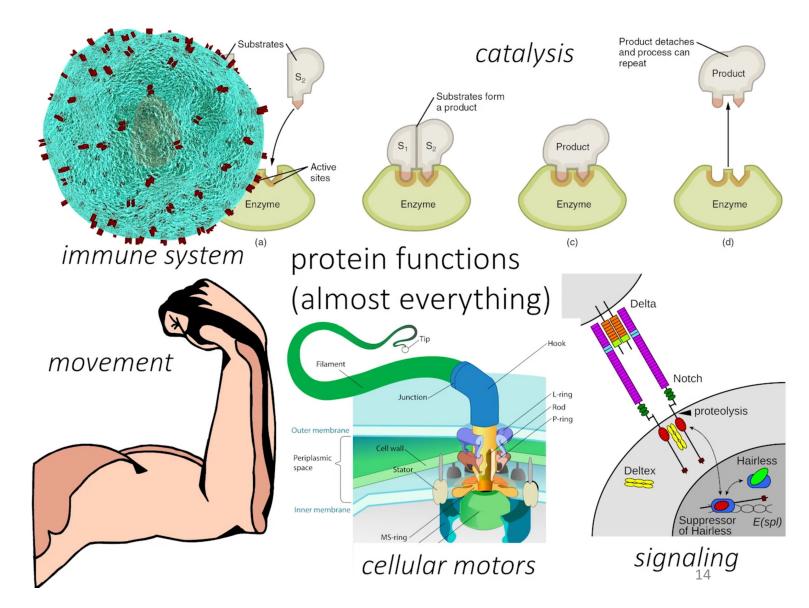
Proteins

- Gene expression products
- Monomers (Ms) = amino acids
- 20 natural common amino acids
- Polymer = protein
- Amino acid structure:

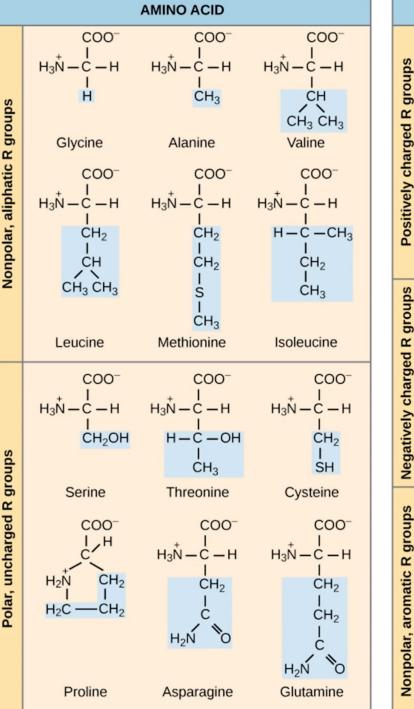
Alpha C + NH₂ + COOH + R (side group)

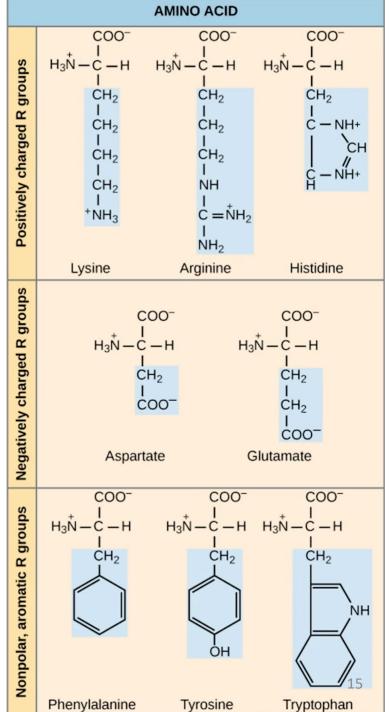


- R can be polar, non-polar, charged, uncharged
- Three or one letter code:
 - Valine = VAL
 - Valine = V



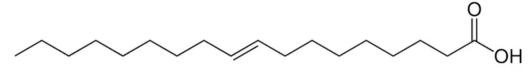
Proteins



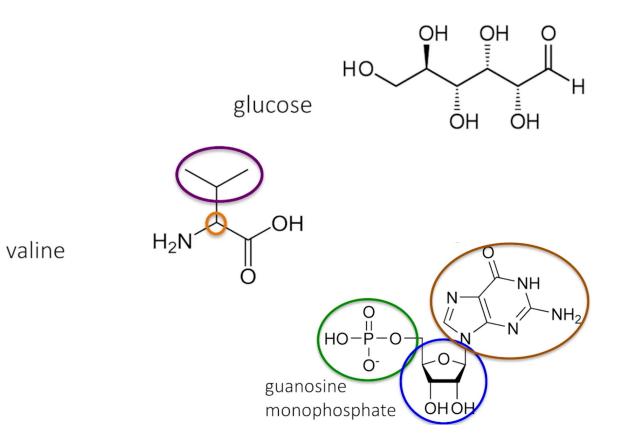


Molecular Biology for Engineering – Lesson 2

Recognizing macromolecules - review



trans-oleic acid



Lipids: nonpolar/hydrophobic

Carbohydrates: CH₂O formula/polar

Proteins: $M = amino acid = \alpha C bonded to NH_2, COOH and R, R = polar/nonpolar$

Nucleic acids: M = nucleotide = phosphate + sugar + base, polar