

Chimica Computazionale

Introduction to computational chemistry

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Definition of computational chemistry

Computational chemistry

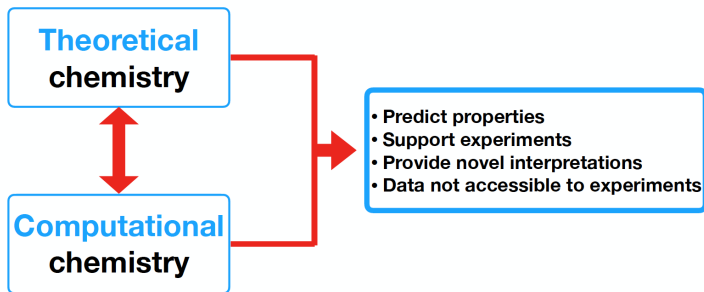
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 - Transformation of matter
 - Properties of atoms, molecules and materials

Computational chemistry

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 - Properties of atoms, molecules and materials
- Theoretical chemistry:
 - Mathematical methods
 - Fundamental laws of physics
 - Processes of chemical relevance

Computational chemistry

- **Chemistry:**
 - Transformation of matter
 - Properties of atoms, molecules and materials
- **Theoretical** chemistry:
 - **Mathematical** methods
 - Fundamental laws of **physics**
 - Processes of chemical relevance
- **Numerical approaches** needed to find approximate solutions to theory
- **Computational** chemistry: computer as a **research tool**



Theory, models and computation

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 - Rules to describe the behavior of physical systems
 - Quantitative relations as mathematical equations
 - General

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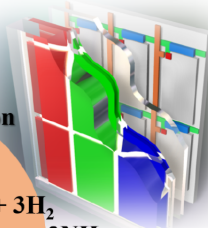
- Theory:
 - Rules to describe the behavior of physical systems
 - Quantitative relations as mathematical equations
 - General
- Models:
 - Simplifying approximations
 - Practical utility
 - Empirical constants
- Computation:
 - Use of digital technology to solve equations for a theory or model
 - Focus (also) on efficiency in simulations
 - Apply theory/models to increasingly complex systems

Computational chemistry

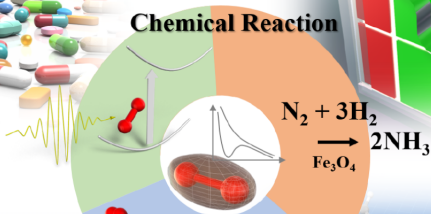
Drug discovery



Material science

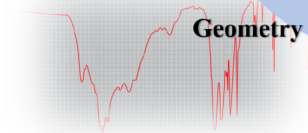


Chemical Reaction

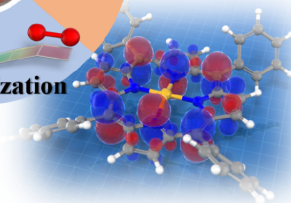


Geometry Optimization

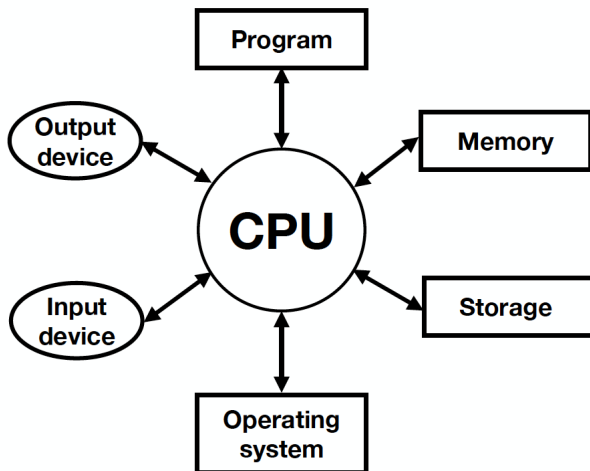
Molecular spectroscopy



Complex molecular dynamics

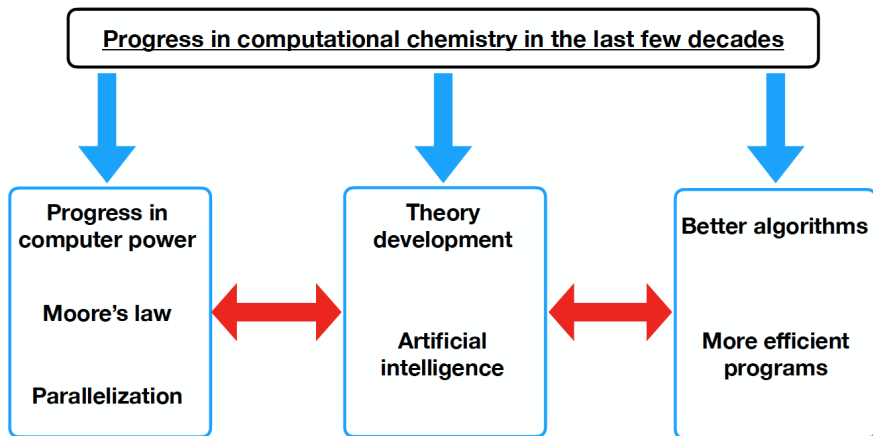


How computers work



- Programs of computational chemistry written with: [Fortran](#), [C](#), [C++](#), [Python](#) etc.

How computers work



Designing a computational project

1 Defining an aim

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- 1 Defining an **aim**
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- 4 Being **critical** of the computational results
- 5 **Testing** the calculations

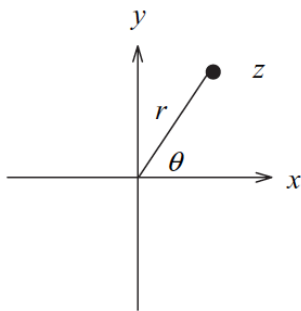
Atomic units

Symbol	Quantity	Value in au	Value in SI units
m_e	Electron mass	1	$9.110 \times 10^{-31} \text{ kg}$
e	Electron charge	1	$1.602 \times 10^{-19} \text{ C}$
t	Time	1	$2.419 \times 10^{-17} \text{ s}$
$\hbar$	$\hbar/2\pi$ (atomic momentum unit)	1	$1.055 \times 10^{-34} \text{ J s}$
h	Planck's constant	2π	$6.626 \times 10^{-34} \text{ J s}$
a_0	Bohr radius (atomic distance unit)	1	$5.292 \times 10^{-11} \text{ m}$
E_H	Hartree (atomic energy unit)	1	$4.360 \times 10^{-18} \text{ J}$
c	Speed of light	137.036	$2.998 \times 10^8 \text{ m/s}$
α	Fine structure constant ($= e^2/\hbar c 4\pi\epsilon_0 = 1/c$)	0.00729735	0.00729735
μ_B	Bohr magneton ($= e\hbar/2m_e$)	$1/2$	$9.274 \times 10^{-24} \text{ J/T}$
μ_N	Nuclear magneton	2.723×10^{-4}	$5.051 \times 10^{-27} \text{ J/T}$
$4\pi\epsilon_0$	Vacuum permittivity	1	$1.113 \times 10^{-10} \text{ C}^2/\text{J m}$
μ_0	Vacuum permeability ($4\pi/c^2$)	6.692×10^{-4}	$1.257 \times 10^{-6} \text{ N s}^2/\text{C}^2$

Math overview

Math overview

- Complex number $z = x + iy$, x and y are real numbers, $i^2 = -1$



$$z = x + iy$$

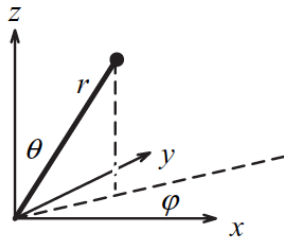
$$z = re^{i\theta} = r\cos(\theta) + ir\sin(\theta)$$

- Complex conjugate $z^* = x - iy$ or $z^* = re^{-i\theta}$

- **scalar**: single number (mass, charge etc.)

Math overview

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- **vector**: set of scalars (coordinates, velocity etc.)

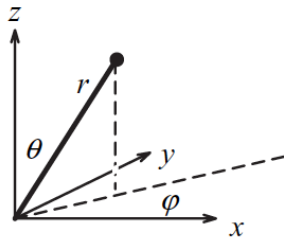


$$\begin{aligned}x &= r \sin\theta \cos\varphi \\y &= r \sin\theta \sin\varphi \\z &= r \cos\theta\end{aligned}$$

- Cartesian and polar systems

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- Cartesian and polar systems
- velocity $\mathbf{v} = \{v_x, v_y, v_z\} = \left\{\frac{\partial x}{\partial t}, \frac{\partial y}{\partial t}, \frac{\partial z}{\partial t}\right\}$

- **matrix**: two-dimensional set of numbers (**Hessian** etc.)

$$\begin{pmatrix} A_{11} & A_{12} & \cdots \\ A_{21} & A_{22} & \cdots \\ \vdots & \vdots & \ddots \end{pmatrix} = \begin{pmatrix} \frac{\partial^2 E}{\partial x_1^2} & \frac{\partial^2 E}{\partial x_1 \partial x_2} & \cdots \\ \frac{\partial^2 E}{\partial x_2 \partial x_1} & \frac{\partial^2 E}{\partial x_2^2} & \cdots \\ \vdots & \vdots & \ddots \end{pmatrix}$$

- adjoint matrix: complex conjugate of the transpose
- **Hermitian** matrices in quantum chemistry are **self-adjoint**, e.g. $\mathbf{H} = \mathbf{H}^\dagger$
- with real elements, the matrix is symmetric, $\mathbf{H} = \mathbf{H}^T$

- Row-by-column matrix-matrix multiplication **AB**

$$\mathbf{AB} = \begin{pmatrix} a_{11} & a_{12} \\ a_{21} & a_{22} \end{pmatrix} \begin{pmatrix} b_{11} & b_{12} \\ b_{21} & b_{22} \end{pmatrix} = \begin{pmatrix} a_{11}b_{11} + a_{12}b_{21} & a_{11}b_{12} + a_{12}b_{22} \\ a_{21}b_{11} + a_{22}b_{21} & a_{21}b_{12} + a_{22}b_{22} \end{pmatrix}$$

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- Determinant $|\mathbf{A}|$ for a 2×2 matrix (only square matrices!)

$$|\mathbf{A}| = \begin{vmatrix} a_{11} & a_{12} \\ a_{21} & a_{22} \end{vmatrix} = a_{11}a_{22} - a_{12}a_{21}$$

$$\begin{aligned} |\mathbf{A}| &= \begin{vmatrix} a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33} \end{vmatrix} \\ &= a_{11} \begin{vmatrix} a_{22} & a_{23} \\ a_{32} & a_{33} \end{vmatrix} - a_{12} \begin{vmatrix} a_{21} & a_{23} \\ a_{31} & a_{33} \end{vmatrix} + a_{13} \begin{vmatrix} a_{21} & a_{22} \\ a_{31} & a_{32} \end{vmatrix} \\ &= a_{11}a_{22}a_{33} - a_{11}a_{23}a_{32} - a_{12}a_{21}a_{33} \\ &\quad + a_{12}a_{23}a_{31} + a_{13}a_{21}a_{32} - a_{13}a_{22}a_{31} \end{aligned}$$

- Interchanging two rows or columns in a matrix changes the **sign** of the determinant ([Slater determinants](#))

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- If two rows or columns are **identical** except for a multiplicative constant, the determinant is **zero**

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- If two rows or columns are **identical** except for a multiplicative constant, the determinant is **zero**
- System of **linear** equations **$\mathbf{Ax} = \mathbf{b}$**

$$\begin{aligned}a_{11}x_1 + a_{12}x_2 + \cdots + a_{1n}x_n &= b_1 \\a_{21}x_1 + a_{22}x_2 + \cdots + a_{2n}x_n &= b_2 \\ \vdots \quad \quad \quad \vdots \quad \quad \quad \vdots \quad \quad \quad \vdots &= \quad \quad \quad \vdots \\a_{n1}x_1 + a_{n2}x_2 + \cdots + a_{nn}x_n &= b_n\end{aligned}$$

- **$\mathbf{x} = \mathbf{A}^{-1}\mathbf{b}$**

Math overview

- Function $E(\mathbf{R}): \mathbf{R} \rightarrow E$
- Functional $E_{xc}[\rho(\mathbf{x})]: \mathbf{x} \rightarrow \rho \rightarrow E_{xc}$
- Operator $\hat{O}$: a function from another function
- Hilbert space: functions or operators as vectors
- Bra-ket notation

$$\text{bra} : \langle \mathbf{f} | \mathbf{x} \rangle = \mathbf{f}^*(\mathbf{x})$$

$$\text{ket} : \langle \mathbf{x} | \mathbf{f} \rangle = \mathbf{f}(\mathbf{x})$$

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$$\langle \mathbf{f} | \mathbf{g} \rangle = \int \mathbf{f}^*(\mathbf{x}) \mathbf{g}(\mathbf{x}) d\mathbf{x}$$

$$|\mathbf{f}(\mathbf{x})| = \sqrt{\langle \mathbf{f} | \mathbf{f} \rangle} = \sqrt{\int \mathbf{f}^*(\mathbf{x}) \mathbf{f}(\mathbf{x}) d\mathbf{x}}$$

Math overview

- Functions and operators with bra-ket notation

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Linear operators

$$\hat{O}(\mathbf{f} + \mathbf{g}) = \hat{O}\mathbf{f} + \hat{O}\mathbf{g}$$

$$\hat{O}(c\mathbf{f}) = c\hat{O}\mathbf{f}$$

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- Normalization

$$\langle \mathbf{f} | \mathbf{f} \rangle = N^2$$

$$\mathbf{f}' = N^{-1} \mathbf{f}$$

$$\langle \mathbf{f}' | \mathbf{f}' \rangle = 1$$

- Orthogonalization (orthonormalization)

$$\langle \mathbf{f}_i | \mathbf{f}_j \rangle = \delta_{ij}$$