# Fermentations for the production of amino acids &, antibiotics



### Fermentations



Definition(s)

 sensu stricto: a metabolic process where O<sub>2</sub> is replaced by an organic molecule as final acceptor of electrons

#### **KEY CONCEPT**

Fermentation allows the production of a small amount of ATP without O<sub>2</sub>

- Fermentation allows Glycolysis to continue producing ATP when oxygen is unavailable.
  - Anaerobic process



#### FERMENTATION

energy  
$$C_6H_{12}O_6$$
 + low  $O_2 \rightarrow CO_2$  + waste + energy



# **Two examples of Fermentation**

- Lactic Acid
- Alcoholic

(b) Alcohol fermentation occurs in yeast.



(a) Lactic acid fermentation occurs in humans.





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# Uses of fermentation in industry

- Sewage treatment
- Biofuels
- Hydrogen gas



TABLE 5.4	Some Industrial Uses for Different Types of Fermentations			
Fermentation End-Product(s)	Industrial or Commercial Use	Starting Material	Microorganism	
Ethanol	Beer	Malt extract	Saccharomyces cerevisiae (yeast, a fungus)	
	Wine	Grape or other fruit juices	Saccharomyces cerevisiae var. ellipsoideus	
	Fuel	Agricultural wastes	Saccharomyces cerevisiae	
Acetic acid	Vinegar	Ethanol	Acetobacter (bacterium)	
Lactic acid	Cheese, yogurt	Milk	Lactobacillus, Streptococcus (bacteria)	
	Rye bread	Grain, sugar	Lactobacillus bulgaricus (bacterium)	
	Sauerkraut	Cabbage	Lactobacillus plantarum (bacterium)	
	Summer sausage	Meat	Pediococcus (bacterium)	
Propionic acid an carbon dioxide		Lactic acid	Propionibacterium freudenreichii (bacterium)	
Acetone and butanol	Pharmaceutical, industrial uses	Molasses	Clostridium acetobutylicum (bacterium)	
Glycerol	Pharmaceutical, industrial uses	Molasses	Saccharomyces cerevisiae	
Citric acid	Flavoring	Molasses	Aspergillus (fungus)	
Methane	Fuel	Acetic acid	Methanosarcina (bacterium)	
Sorbose 12/2/16	Vitamin C (ascorbic acid)	Sorbitol Cellular Energy: Fermentation	Acetobacter 17	

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# Fermentations in wide sense: the products



- Microbial biomasses
- Fermentation products (sensu stricto)
- Primary metabolites
- Products from incomplete oxidation
- Secondary metabolites
- Recombinant proteins
- Polisaccharides

starters for the food industry, probiotics etc. ethanol, butanol, lactic acid etc. aminoacids, vitamins etc. organic acids **antibiotics, anticancers, statins etc. insulin, growth hormone, industrial enzymes** 

# Produzione di biomassa (per es. produzione di enzimi)

Curva di crescita microbica



## Fase Lag

## Stasi prima di una rapida crescita, può dipendere da:

- ·Le cellule potrebbero essere danneggiate
- ·Le cellule si debbono adattare al terreno
- ·Le cellule possono essere vecchie o fredde
- ·Le cellule producono nuovi ribosomi
- ·Le cellule sintetizzano nuovi enzimi
- ·Le cellule iniziano a fare cellule



- •Le cellule si riproducono velocemente
- ·La biomassa aumenta con rapidità

Fase Log

- ·Le sostanze nutrienti sono consumate in fretta
- ·L'Ossigeno (se usato) è consumato rapidamente
- ·Alcune colture producono calore
- ·Variazioni di pH dovute ai microrganismi
- ·Le proteine nel brodo possono formare schiuma
- ·La coltura può mutare reologia (mixing)

# >Fase Stazionaria

>Esaurimento dei Nutrienti

>L'Ossigeno può essere limitato

>Rilascio di sostanze cellulari: per es. tossine

>Cellule che crescono ~= cellule che muoiono

>La divisione cellulare non è più logaritmica

>Possono essere prodotti metaboliti secondari



# Fase letale



- ·Le cellule diminuiscono esponenzialmente
- Può verificarsi autolisi cellulare
- •Le cellule sopravvissute non si duplicano

# **Fermentazione Industriale**



- Primary metabolite
  - Produced during exponential growth
  - Example: alcohol
- <u>Secondary metabolite</u>

- Produced during stationary phase

- <u>Secondary metabolites</u>
  - Not essential for growth
  - Formation depends on growth conditions
  - Produced as a group of related compounds
  - Often significantly overproduced
  - Often produced by spore-forming microbes during sporulation



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- Secondary metabolites are often large organic molecules that require a large number of specific enzymatic steps for production
  - Synthesis of tetracycline requires at least
    72 separate enzymatic steps
  - Starting materials arise from major biosynthetic pathways

# Fermentors

- *Fermentor* is where the microbiology process takes place
- Fermentors vary in size from 5 to 500,000 liters
  - Aerobic and anaerobic fermentors
- Large-scale fermentors are almost always stainless steel
  - Impellers and spargers supply oxygen





### Secondary metabolites: a wide array of chemical structures





### Secondary metabolites: pharmaceutical properties



Data kindly furnished by prof. Flavia Marinelli (VICURON ANTIBIOTIC LITERATURE DATABASE)

#### Secondary metabolites : pharmaceutical relevance

Table 1 The 50 most important microbial secondary metabolites or derivatives



<sup>a</sup>Plant product, also made by microorganisms



## Penicillin G biosynthesis



1) Polymerization: A tripeptide is formed by sequential addition of 3 aminoacids



2) Cyclization: the peculiar structure of penicillins (2 fused rings) is formed



3) Decoration: transacylation with introduction of the phenylacetic moiety



#### **Production of penicillins**

Year	Production (kg)	Cost (\$/kg)
1945	2,300	11,000
1963	3,000,000	150
1978	15,000,000	18.50
1992	22,000,000	_
1995	31,000,000	4.5

→ prezzo (\$/kg) → titolo medio (g/L)



#### **Amino acids: industrial impact**

Of the 20 standard protein amino acids, the **9 essential amino acids** -valine, -leucine, isoleucine, -lysine, -threonine, -methionine, -histidine, -phenylalanine, and tryptophan occupy a key position in that they are not synthesized in animals and humans but must be **ingested with feed or food**.

In terms of market volume, development over the last 20 years has been tremendously bullish in the so-called **feed amino acids** L-lysine, DL-methionine, L-threonine, and Ltryptophan, which constitute the largest share (56%) of the total amino acid market, estimated in 2004 at approximately US \$4.5 billion.

Also substantial is the share of the **food sector**, which is determined essentially by three amino acids: **\_-glutamic acid** in the form of the flavorenhancer monosodium glutamate (MSG) and the amino acids **\_-aspartic acid and \_-phenylalanine**, both of which are starting materials for the peptide sweetener **\_-**aspartyl **\_phenylalanyl methyl** ester (Aspartame), used, for example, in "lite" colas.

The amino acid market for synthesis applications is growing at an annual rate of 7% (US \$1 billion in the year 2009), of which the share of amino acids for **peptide sweeteners** alone is expected to be more than US \$400 million.

#### **Amino acids: industrial impact**

The use of enzymes and whole cell biocatalysts has proven particularly valuable in production of both **proteinogenic** and **nonproteinogenic** L-amino acids D-amino acids, and enantiomerically pure amino acid derivatives, which are of great interest as **building blocks** for active ingredients that are applied as **pharmaceuticals, cosmetics, and agricultural products**.





# <u>Chiral pool</u>

HO HO O	OH O				
но он	Compound	Approx. price (US dollars kg <sup>-1</sup> )	H <sub>3</sub> C NH <sub>2</sub> OH		
	Ascorbic acid (+)-Calcium pantothenate (-)-Carvone Anhydrous dextrose Ephedrine hydrochloride (+)-Limonene L-Lysine Mannitol Monosodium glutamate Norephedrine hydrochloride Quinidine sulphate Quinine sulphate Sorbitol L-Threonine	$(US dollars kg^{-1})$ 13 16 23 1.2 62 3 3.2 7.5 2 24 130 75 1.7 12-50, depending on grade	$\dot{N}H_2$ $HO \longrightarrow OH $		
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#### **PRODUCTION OF AMINOACIDS**

**Extraction of amino acids from protein hydrolysate** as a method of obtaining *L*-amino acids is now of only limited importance; although still relevant for production of *L*-**serine**, *L*-**proline**, *L*-**hydroxy-proline**, and *L*-**tyrosine**, for example, it is not suitable for large-scale production of amino acids.

The extraction method for obtaining **-glutamate** was superseded nearly 50 years ago by fermentation, following a sharp increase in demand.

### **MICROBIAL PRODUCTION OF AMINOACIDS**

The discovery of the soil bacterium, *Corynebacterium glutamicum*, which is capable of producing L-glutamic acid with high productivity from sugar, paved the way for the success of the fermentation technique in amino acid production (Kinoshita et al. 1957). It was advantageous that the wild strain could be used on an industrial scale under optimized fermentation conditions for **mass production of glutamate**.

#### **Biotechnological production of lysine:** Corynebacterium glutamicum

Lysine is a preferred additive to animal feeds for pig breeding (as the first limiting amino acid) and poultry (second limiting amino acid, after methionine).



Fig. 3 Global market for L-lysine (1970-2005). The picture shows the lysine-producing mutant of C. glutamicum—after cell division

#### **Selected amino acid producing strains**

The amino acids **-phenylalanine and cysteine**, both of which were previously produced mainly with the help of enzymes, can now be obtained more cost effectively by fermentation with *E. coli* strains and are thus available to a larger and growing market. Almost all proteinogenic amino acids, with a few exceptions, can be produced industrially by specially developed mutants of *C. glutamicum* or *E. coli*.

Amino acid	Strain/mutant	Titer (g/l)	Estimated yield (g/100 g sucrose)
L-Lysine HCl	C. glutamicum B-6	100	40-50
L-Threonine	E. coli KY 10935	100	40-50
L-Tryptophan	C. glutamicum KY9218/pIK9960	58	20-25
L-Tryptophan	E. coli	45	20-25
L-Phenylalanine	E. coli MWPWJ304/pMW16	51	20-25
L-Arginine	Brevibacterium flavum AJ12429	36	30-40
L-Histidine	C. glutamicum F81/pCH99	23	15-20
L-Isoleucine	E. coli H-8461	30	20-30
L-Serine	Methylobacterium sp. MN43	65	30-35
L-Valine	C. glutamicum VR 3	99	30-40

#### **Enzymatic production of enantiomerically pure aminoacids**

- For other amino acids, there is no comparable enzyme system for conversion of the D-form, and there is no fermentation process with adequate yield.
- For these amino acids, it is necessary to produce the enantiomerically pure form using enzymatic procedures.
- The racemates are generally produced by chemical synthesis.

# Synthesis of intermediates of D,L-amino acids:





# Synthetic methods for the production of pure enantiomers



Kinetic resolution of racemates:

# typical application of hydrolases

When an enzymatic catalytic reaction is followed in time, *ideally* only one enantiomer reacts and the reaction stops at 50% conversion.

- Resolutions have a maximum theoretical yield of 50%
- Unwanted enantiomer is wasted or at best recycled





- R, S = substrate enantiomers
- P, Q = product enantiomers

# **DKR: Hydrolysis of N-acylated amminoacids**



- *rac N* -acyl amino acids as substrates.
- use of acylases from porcine kidney or from *Aspergillus* or *Penicillium* sp.
- resolution of *N* -acetyl tryptophan and -phenylalanine on an industrial scale using immobilized enzymes in column reactors.
- the non-reacting D-enantiomer may be recycled *via* **racemization** of the corresponding **mixed anhydride** intermediate in a separate step.

# **DKR: Hydrolysis of amides of aminoacids**



- use of L-selective amidases from *Pseudomonas*, *Aspergillus* or *Rhodococcus* sp.,
- hydrolyze L-amino acid amides from a racemate.

The possibility to **recycle** the unreacted D-configured amide *via* its corresponding Schiff-base with benzaldehyde in a separate step makes this procedure economical.

# Synthetic intermediates of D,L-amino acids: hydantoines



A promising route to enantiomerically pure amino acids, both L- and Denantiomers, is based on conversion of hydantoins via hydantoinases and, additionally, carbamoylase.

# **5-substitued hydantoins**

- 1. hydantoinase
- 2. carbamoylase,
- 3. hydantoin racemases



#### **Dynamic Kinetic Resolution (DKR)**

**Racemization using enzymes** 

The use of an **enzyme**, rather than a transition metal catalyst, represents an attractive option for combined DKR reactions in view of the likely mild conditions associated with enzyme-catalyzed racemization processes.

**Racemases** belong to the group of enzymes EC 5.1.X.X and contain notable members such as mandelate racemase and various **amino acid racemases**.

# Method of the 5-substitued hydantoins for side chain antibiotics production



# **Method of the 5-substitued hydantoins**

D-serine L-methionine

Degussa (D): whole cells coexpressing L-carbamoylase + hydantoin racemase + hydantoinase

# L-Aspartic acid: Asymmetric enzymatic synthesis

Achiral substrate + chiral (enantioselective) biocatalyst

1.addition of ammonia to fumaric acid catalyzed by **ammonia lyase** from *E.coli*, also called aspartase

2.L-aspartate (which is required in large quantities for the sweetener Aspartame).

3.aspartate β-decarboxylase from *Pseudomona dacunhae* trasforms aspartic acid into L-alanine



• Lyases catalyze the addition or removal of a chemical group without passing through hydrolysis, oxidation, transfer

Lyases can act on

- C C bonds (decarboxylases; aldolases)
- C O (hydratases or dehydratases)
- C N
- C S
- C X

# **Asymmetric enzymatic synthesis**

Synthesis of L-*tert*-leucine via <u>reductive amination</u> of trimethyl pyruvate



Use of recombinant *E. coli* coexpressing **Leucine dehydrogenase** and NAD+dependent **Formate dehydrogenase**.