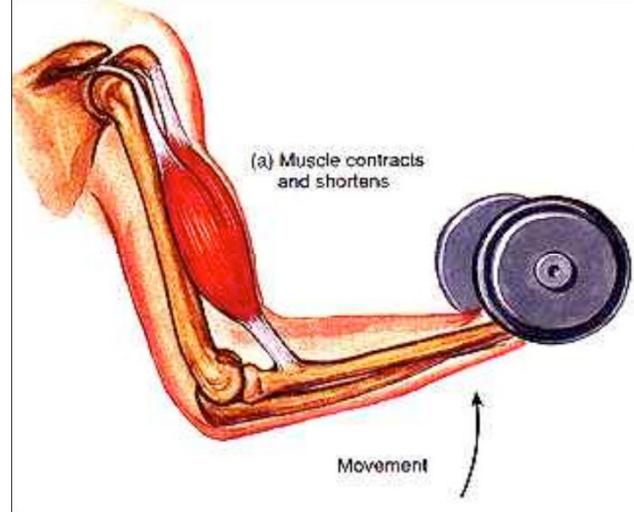
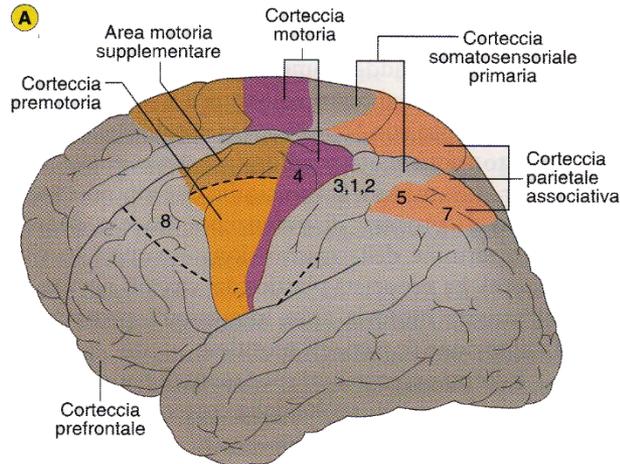
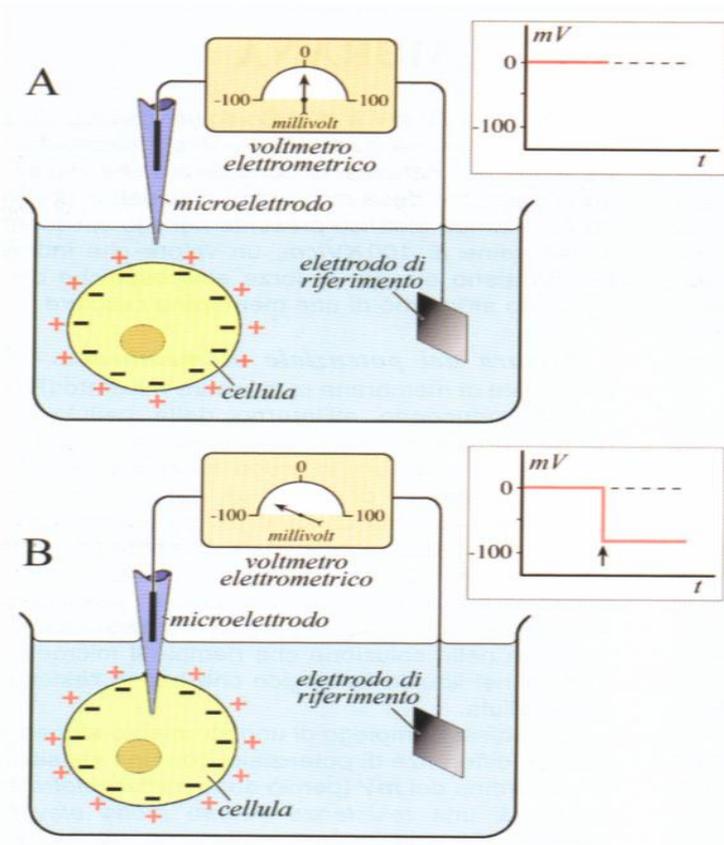


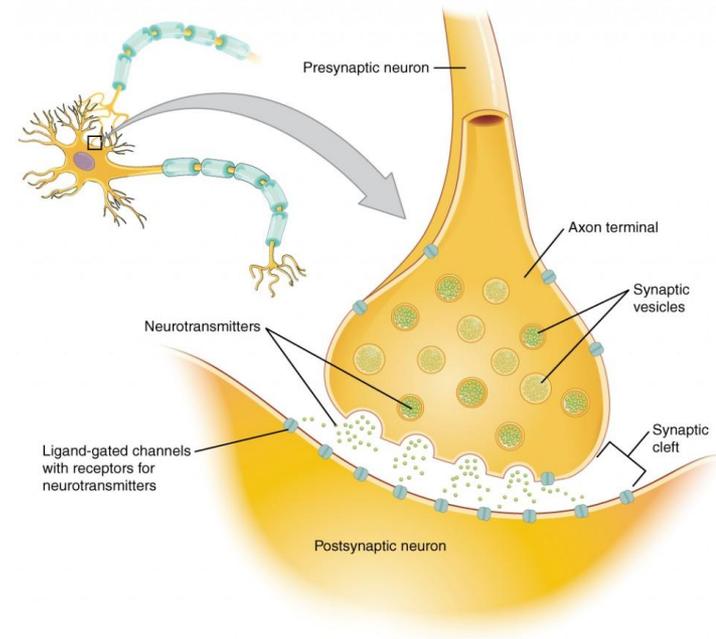
# Neurons and muscle cells are electrically excitable they generate and conduct electrical signals



# Neuronal Resting Membrane Potential ~ -65 mV



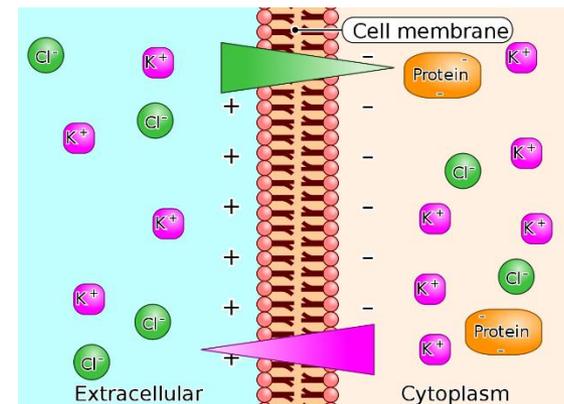
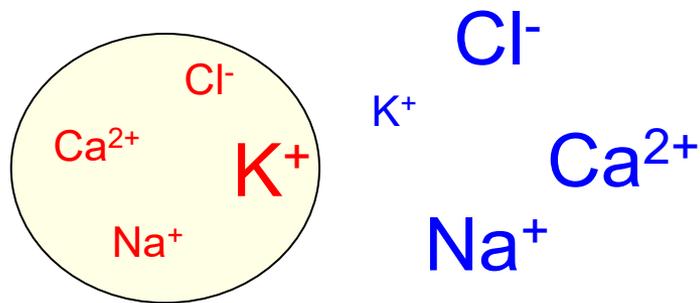
Neurons communicate through changes in membrane potential



Negatively charged comparing with the outside !!

# Ion distribution (in mM) at resting membrane potential

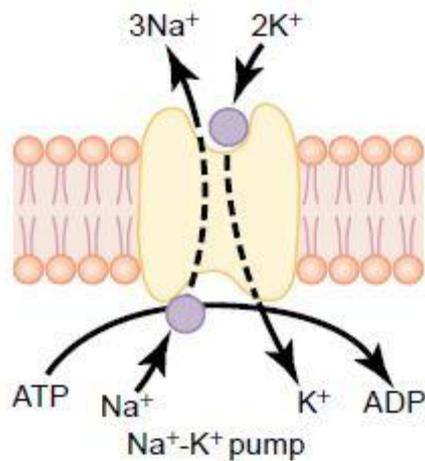
| <i>Intracellular</i> |        | <i>Extracellular</i> |
|----------------------|--------|----------------------|
| $\text{Na}^+$        | 12     | 145                  |
| $\text{K}^+$         | 150    | 5.5                  |
| $\text{Cl}^-$        | 9      | 125                  |
| $\text{Ca}^{2+}$     | 0.0001 | 1.8                  |



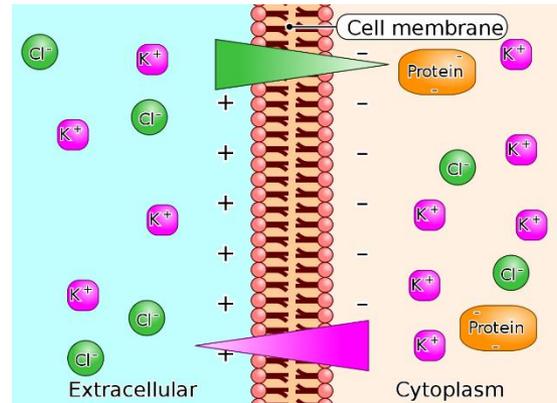
# Resting membrane potential

How it is generated and maintained

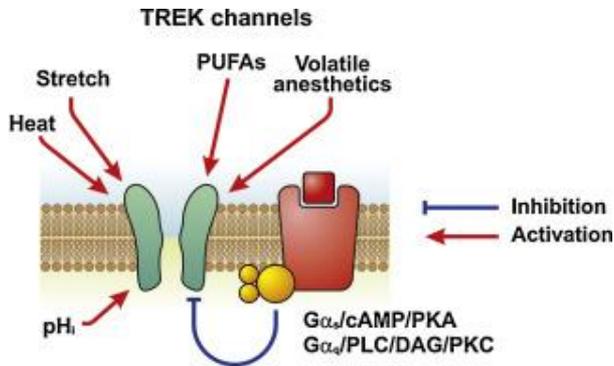
- $\text{Na}^+$ -  $\text{K}^+$  pumps
- Non-diffusible anions
- Membrane-selective permeability / leak channel



Primarily active transporter



# Leak channels



**Participate in:** excitability, neuronal integration, volume regulation.

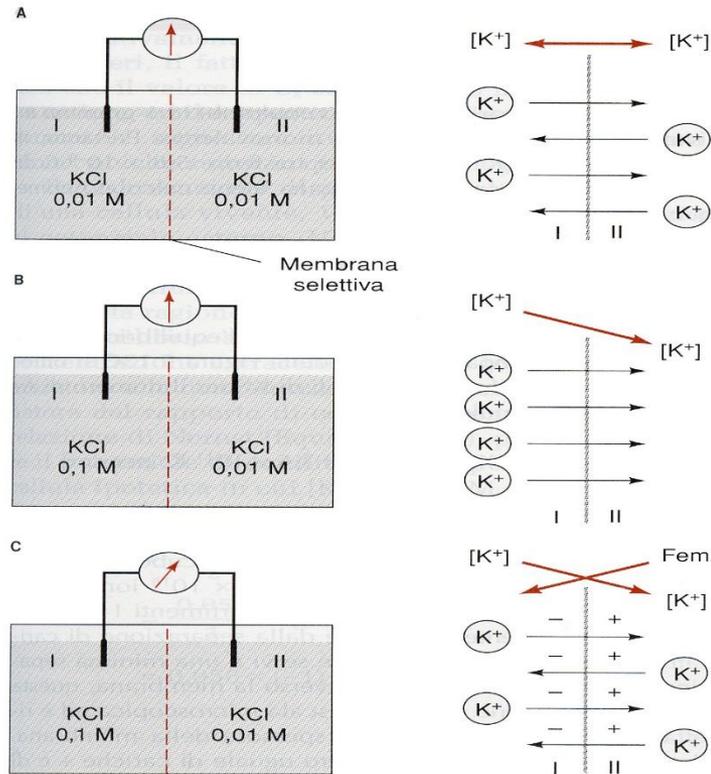
**Modulated by:** temperature, mechanical stretch, pH, cyclic nucleotides, kinase, phosphatases, molecular oxygen

**Involved in** depression, pain perception, anesthesia

High selective flux of K<sup>+</sup> down the electrochemical K<sup>+</sup> gradient

$$P_K : P_{Na} : P_{Cl^-} = 1 : 0.05 : 0.45$$

# «Equilibrium potential» of diffusible ions



The Equilibrium Potential is reached when  
**WORK OF CHEMICAL FORCES = WORK OF ELECTRICAL FORCES**

**net ion flow = 0**

Equilibrium potential for each diffisible ion is calculated from...

WORK OF CHEMICAL FORCES = WORK OF ELECTRICAL FORCES

$$-RT \ln (X_i/X_e) = z_x F V_m$$

$$V_m = - RT/zF \ln (X_i/X_e)$$

Nernst equation

$$V_m = \frac{RT}{z_x F} \ln \frac{X_e}{X_i}$$

At room temperature

$$E_m = 58 \log \frac{X_e}{X_i}$$

The equilibrium potential for each diffusible ion

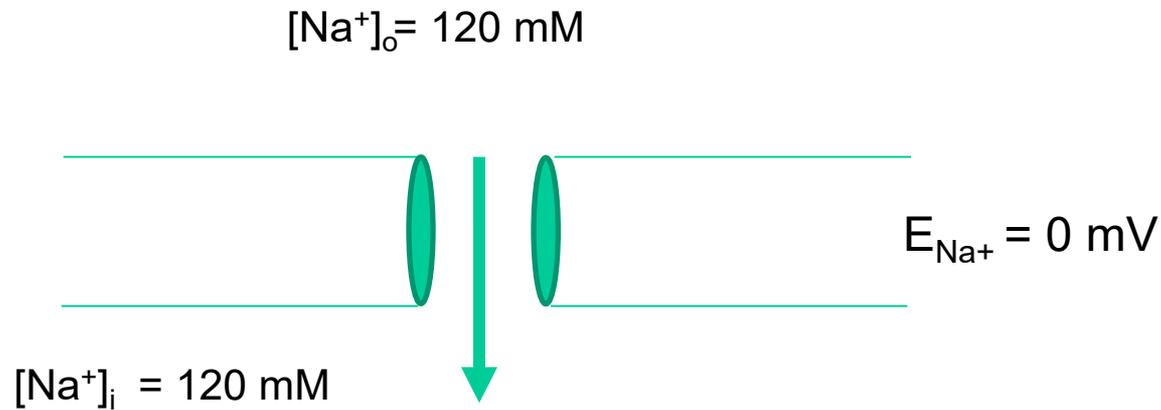
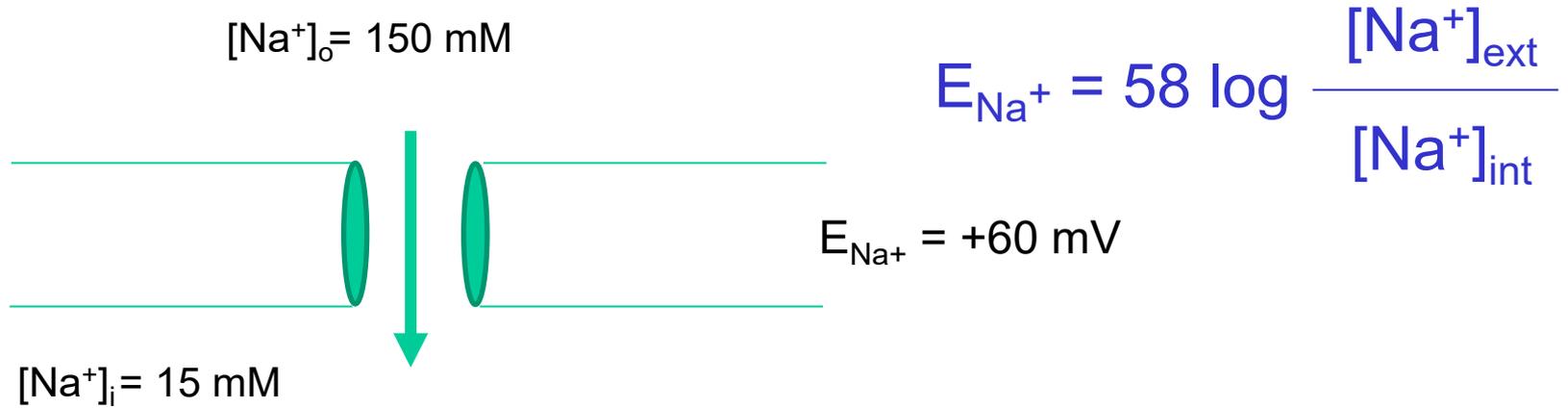
$$E_{\text{Na}^+} \sim +65 \text{ mV}$$

$$E_{\text{Ca}^{2+}} \sim +120 \text{ mV}$$

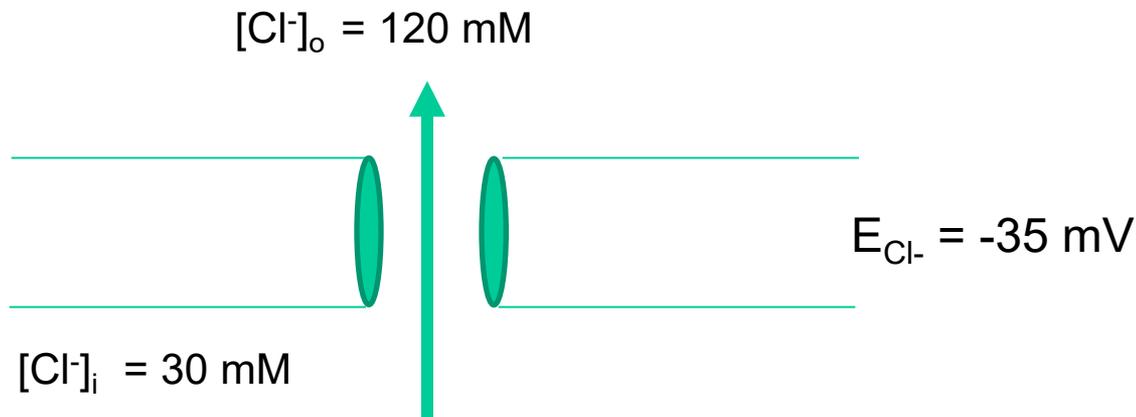
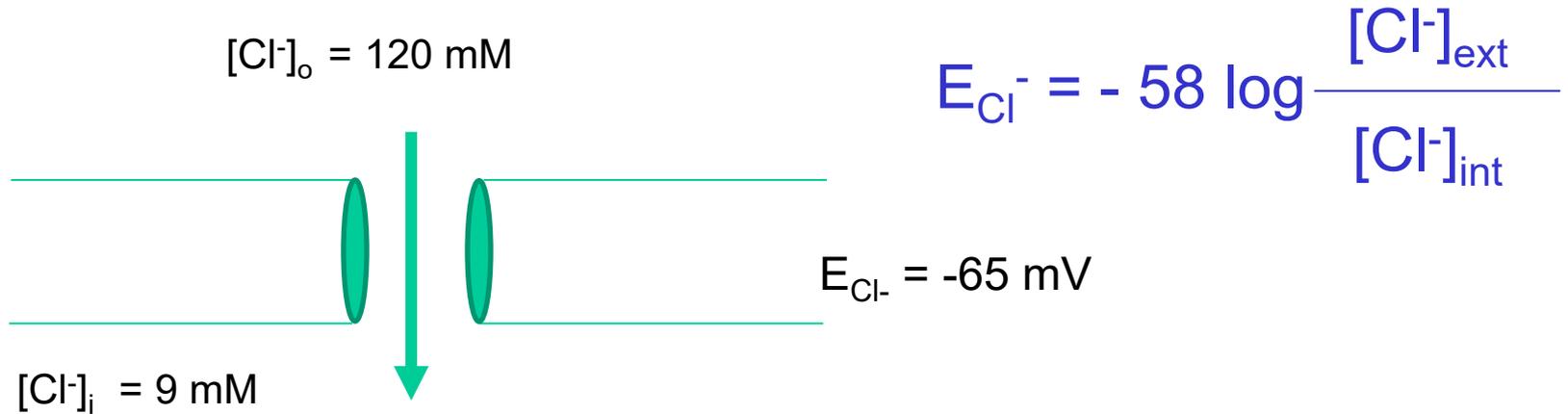
$$E_{\text{K}^+} \sim -85 \text{ mV}$$

$$E_{\text{Cl}^-} \sim -66 \text{ mV}$$

$$V_m = -60 \text{ mV}$$



$$V_m = -60 \text{ mV}$$



The electrochemical driving force =  $(V_m - V_{eq})$

Impaired Cl<sup>-</sup> homeostasis:

Brain injury:

- Hypoxia
- Encephalopathy
- Brain edema

Neurodevelopmental, neuropsychiatric and neurological disorders

Because a given set of ions is present,  
membrane potential is given by

## Goldman-Hodgkin-Katz Equation

$$V_m = \frac{RT}{F} \ln \frac{[K^+]_e P_K + [Na^+]_e P_{Na} + [Cl^-]_i P_{Cl}}{[K^+]_i P_K + [Na^+]_i P_{Na} + [Cl^-]_e P_{Cl}}$$

*Each ion will move down its electrochemical gradient*

The contribution of each ion is determined by:

- *its concentration difference across the membrane*
- *its relative permeability.*

# Equilibri di membrana

THE  
**NERNST/GOLDMAN**  
equation simulator

## ⊕ Potassium [K<sup>+</sup>] (mM)

PK<sup>+</sup>

[K<sup>+</sup>]<sub>o</sub>

[K<sup>+</sup>]<sub>i</sub>

## ⊕ Sodium [Na<sup>+</sup>] (mM)

PNa<sup>+</sup>

[Na<sup>+</sup>]<sub>o</sub>

[Na<sup>+</sup>]<sub>i</sub>

## ⊖ Chloride [Cl<sup>-</sup>] (mM)

PCl<sup>-</sup>

[Cl<sup>-</sup>]<sub>o</sub>

[Cl<sup>-</sup>]<sub>i</sub>

## 🌡 Temperature

°C

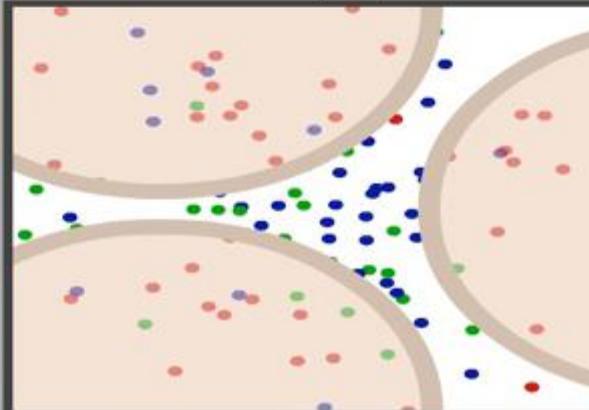
Nernst

Nernst @37°C

**Goldman**

Goldman @37°C

$$E_m = \frac{-RT}{F} \ln \frac{P_{K^+}[K^+]_i + P_{Na^+}[Na^+]_i + P_{Cl^-}[Cl^-]_o}{P_{K^+}[K^+]_o + P_{Na^+}[Na^+]_o + P_{Cl^-}[Cl^-]_i} = \mathbf{-58.6 \text{ (mV)}}$$



**E<sub>K</sub>** -61.5 (mV)

**E<sub>Na</sub>** 58.1 (mV)

**E<sub>Cl</sub>** -69.6 (mV)

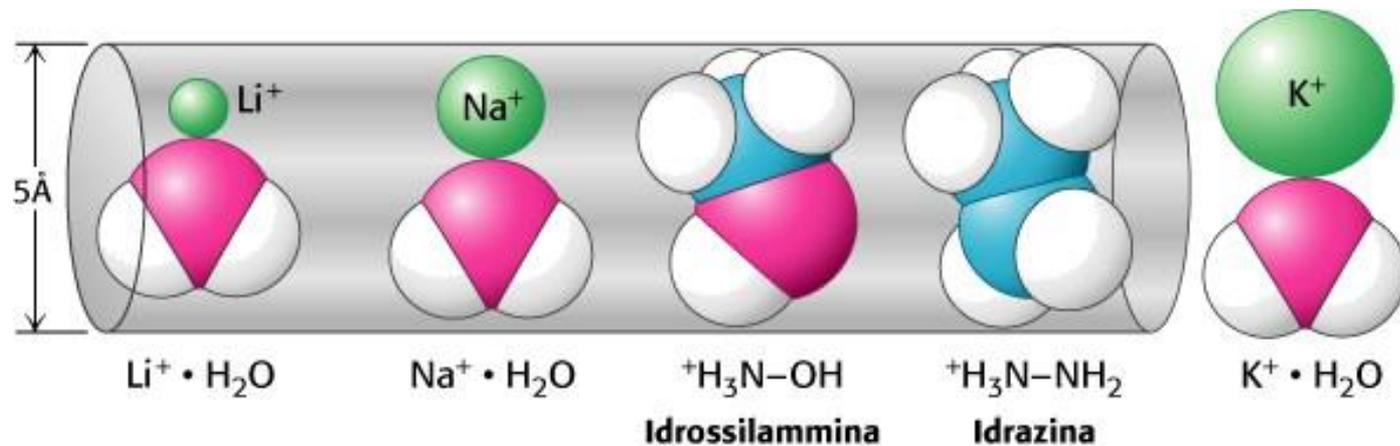
**Ions cross membranes**  
**due to the presence of channel proteins**  
forming hydrophilic pores across membranes

For transport efficiency, channels have an advantage over carriers in that **up to 100 million ions can pass through one open channel each second**—a rate  $10^5$  times greater than the fastest rate of transport mediated by any known carrier protein

# Ion channel selectivity

1) Steric factors (**pore size**)

1) Rings of acidic residues (Glu, Asp) at the edges of the pore (**selectivity filter**)



The main stimuli that are known to cause ion channels to open are:

- a change in the voltage across the membrane (*voltage-gated channels*)
- the binding of a ligand (*ligand-gated channels*).
- a mechanical stress (*mechanically gated channels*)

The activity of many ion channels is regulated, in addition, by protein phosphorylation and dephosphorylation.