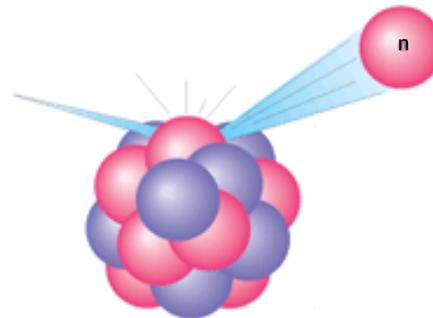


Neutron scattering from biomembranes

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*Dept. of Medical Biotechnologies and Translational Medicine,
Università degli Studi di Milano, Italy*



Why radiation scattering?

How can we determine the relative positions and motions of atoms in a bulk sample of solid or liquid?

We need to 'see' inside the material with a suitable tool
?

By **EYE** we can see only through transparent materials
but
not the atoms

limit for visible light: 10^{-6}m
Interatomic distance in a solid $\sim 10^{-10}\text{m}$

→ Radiation scattering

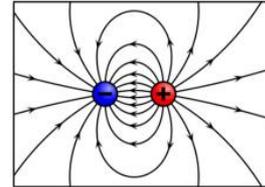
What is a neutron?

Quantum mechanics tells us that,
whilst it is certainly **particulate**,
the neutron also has a **wave** nature
and as such can display the gamut of wave behaviors
including reflection, refraction and diffraction.

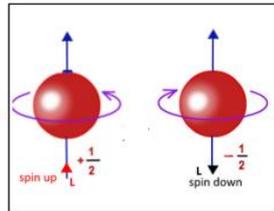
What is a neutron?

Neutron: a subatomic particle, symbol n (discovered by James Chadwick in 1932).

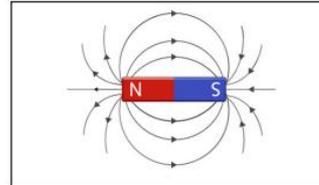
- No electric charge: $q_n=0$
- No electric dipole: $d_n=(0.0\pm 1.1\pm 0.2) \cdot 10^{-26} \text{ e}\cdot\text{cm}$



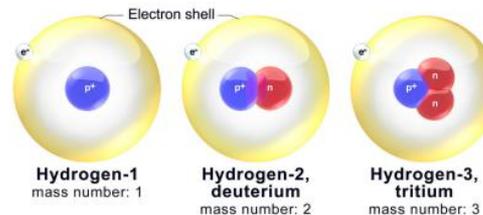
- Spin (*i.e.*, polarization): $s_n=1/2$.



- Large magnetic moment: $\mu_n=-1.913043 \mu_N = -9.662364 \cdot 10^{-27} \text{ JT}^{-1}$.



- Mass slightly greater than that of a proton (p): $m_n=1.674927 \cdot 10^{-27} \text{ kg}$ ($m_p=1.672622 \cdot 10^{-27} \text{ kg}$).
- Neutrons and protons form atomic nuclei: neutrons are in all nuclei except in ^1H . Number of neutrons, N : $N = A - Z$ (*i.e.*, atomic mass number minus proton number).



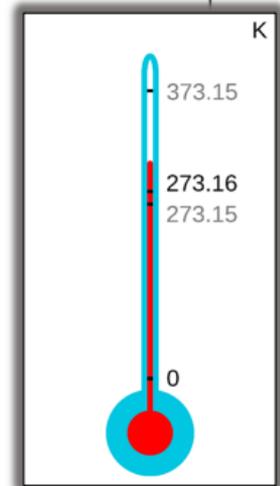
- Lifetime: virtually infinite inside **stable** atomic nuclei. “Half-life”, $t_{1/2}=611 \text{ s}$, when free.

What is a neutron?

First disclaimer: in this lesson we will not study **bound neutrons** (in either stable or instable nuclei), leaving them to nuclear physics.

We will focus on **free neutrons** only, which are usually divided according to their “temperature” T or, more properly, by their (kinetic) energy $E = k_B T$ as:

- **Ultracold:** $E < 0.1 \text{ meV}$ $\rightarrow T < 1.2 \text{ K}$
- **Very cold:** $0.1 \text{ meV} < E < 0.5 \text{ meV}$ $\rightarrow 1.2 \text{ K} < T < 5.8 \text{ K}$
- **Cold:** $0.5 \text{ meV} < E < 5.0 \text{ meV}$ $\rightarrow 5.8 \text{ K} < T < 58 \text{ K}$
- **Thermal:** $5 \text{ meV} < E < 100 \text{ meV}$ $\rightarrow 58 \text{ K} < T < 1,200 \text{ K}$
- **Epithermal:** $0.1 \text{ eV} < E < 1 \text{ eV}$ $\rightarrow 1,200 \text{ K} < T < 12,000 \text{ K}$
- **Resonant:** $1 \text{ eV} < E < 100 \text{ eV}$
-



What is a neutron?

Second Disclaimer: in this lesson we will deal only with the so-called “**slow neutrons**”, *i.e.*, those with $E < 1 \text{ keV}$. Among the other things (see later), slow neutrons are totally **non-relativistic objects** since $m_n c^2 = 939.6 \text{ MeV} \gg 1 \text{ keV}$, and so one can easily work out their speed from $E = 1/2 m_n v^2$.

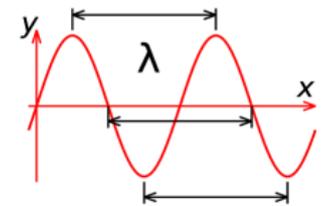
- | | | | |
|----------------------|---|---|---|
| • Ultracold: | $E < 0.1 \text{ meV}$ | → | $v < 138 \text{ m/s}$ |
| • Very cold: | $0.1 \text{ meV} < E < 0.5 \text{ meV}$ | → | $138 \text{ m/s} < v < 309 \text{ m/s}$ |
| • Cold: | $0.5 \text{ meV} < E < 5.0 \text{ meV}$ | → | $309 \text{ m/s} < v < 978 \text{ m/s}$ |
| • Thermal: | $5 \text{ meV} < E < 100 \text{ meV}$ | → | $978 \text{ m/s} < v < 4,374 \text{ m/s}$ |
| • Epithermal: | $0.1 \text{ eV} < E < 1 \text{ eV}$ | → | $4,374 \text{ m/s} < v < 13,830 \text{ m/s}$ |
| • Resonant: | $1 \text{ eV} < E < 100 \text{ eV}$ | → | $1.383 \times 10^4 \text{ m/s} < v < 1.383 \times 10^5 \text{ m/s}$ |
| • | | | |



What is a neutron?

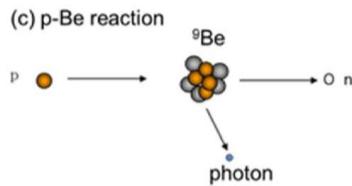
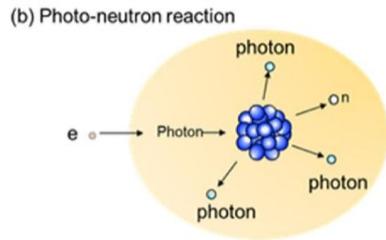
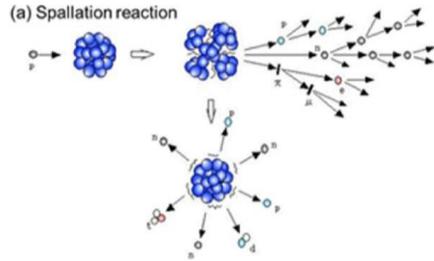
Neutrons are microscopic objects, so they are better described by **quantum mechanics (QM)** rather than by classical Newtonian mechanics. **QM** implies the famous **particle-wave dualism**: every particle with mass m and speed v is associated to a wave, showing a wavelength λ which is expressed by the famous **de Broglie equation**, where h is the **Planck constant**. In the neutron case: $\lambda = h/(m_n v) = h/(2m_n E)^{0.5}$:

- **Ultracold:** $E < 0.1$ meV \rightarrow $\lambda > 28.6$ Å [1 Å = 10^{-10} m]
- **Very cold:** 0.1 meV $< E < 0.5$ meV \rightarrow 12.8 Å $< \lambda < 26.8$ Å
- **Cold:** 0.5 meV $< E < 5.0$ meV \rightarrow 4.04 Å $< \lambda < 12.8$ Å
- **Thermal:** 5 meV $< E < 100$ meV \rightarrow 0.904 Å $< \lambda < 4.04$ Å
- **Epithermal:** 0.1 eV $< E < 1$ eV \rightarrow 0.286 Å $< \lambda < 0.904$ Å
- **Resonant:** 1 eV $< E < 100$ eV \rightarrow 0.029 Å $< \lambda < 0.286$ Å
- ...

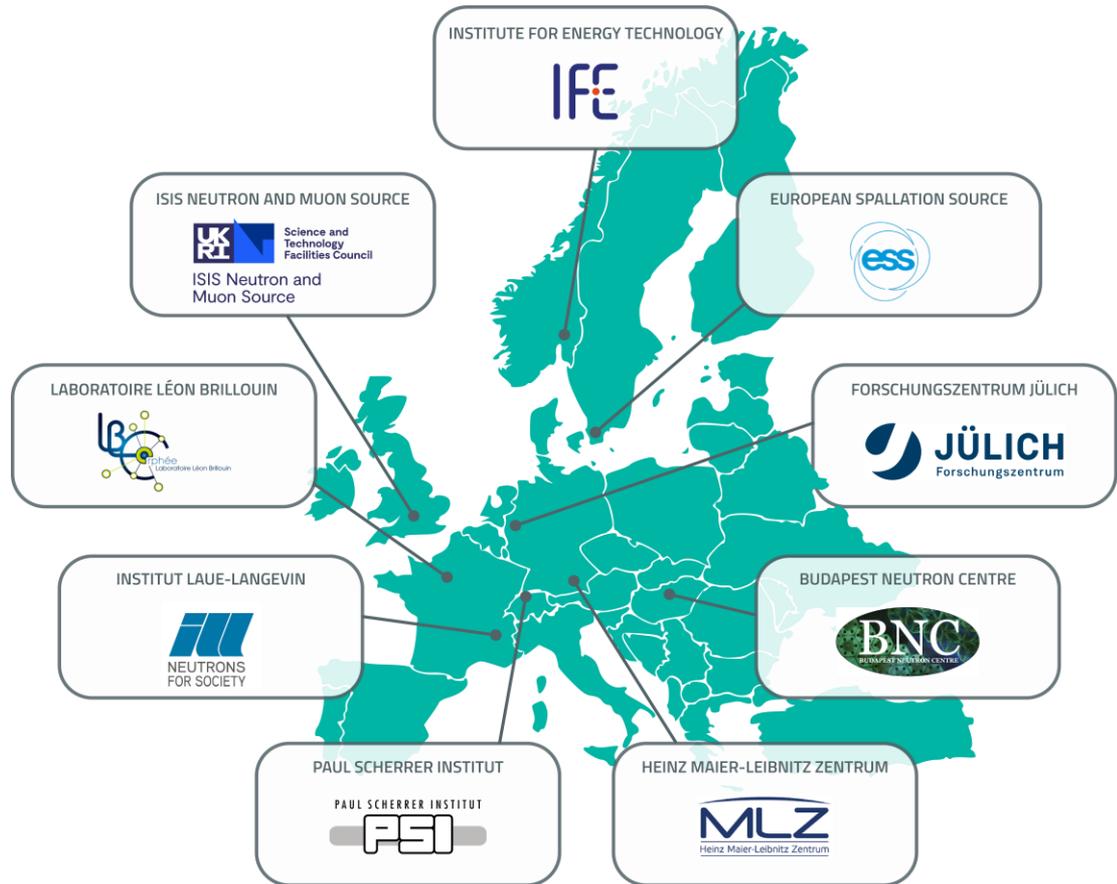


All this is very interesting for **chemico-physical applications** since the typical atomic/molecular size is actually ~ 1 Å !

How and where?

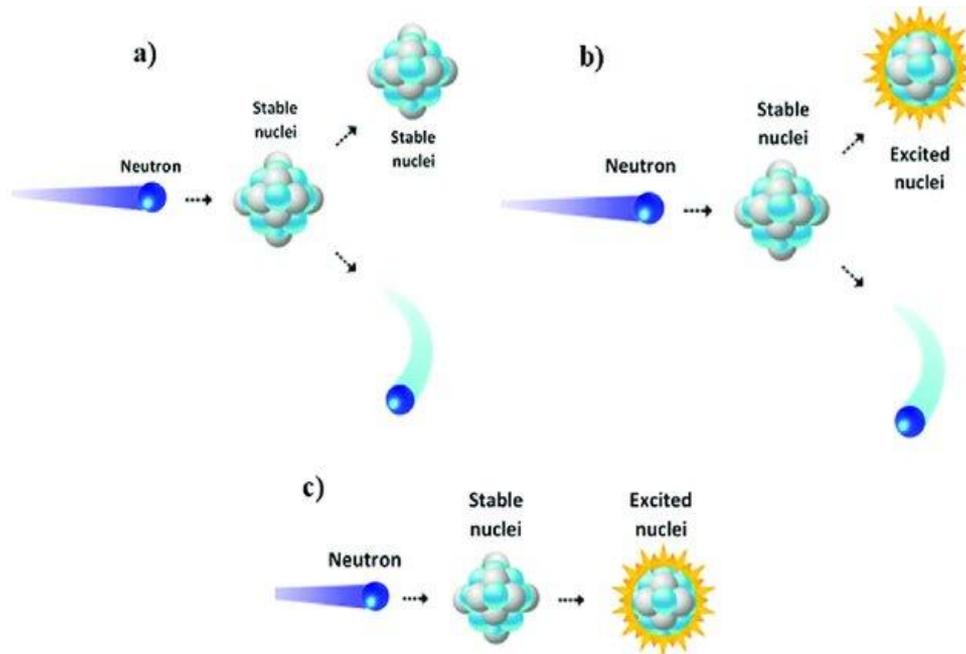


Kiyanagi, Y. AAPS Bull. (2021)



What does 'scattering' mean?

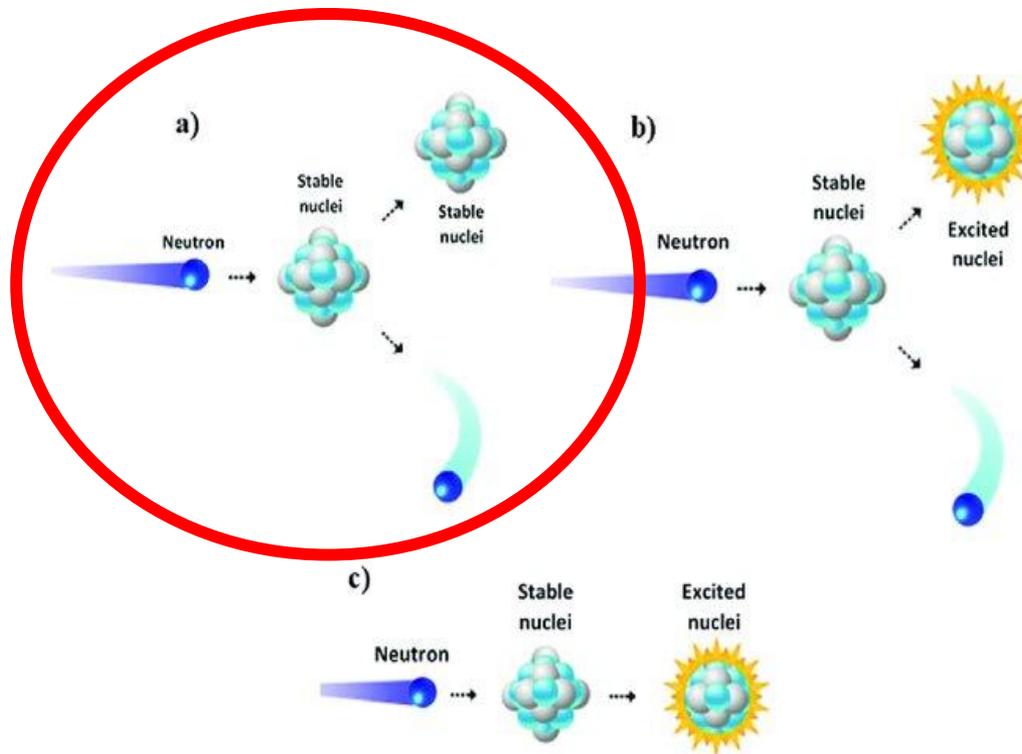
In physics, **scattering** is a process where **moving particles or waves**, such as light or sound, **are forced to deviate** from a straight trajectory **by interacting** with localized non-uniformities in a medium or with other particles. This deviation can result in a change in direction, energy, frequency, or phase.



10.1016/B978-0-12-820549-5.00010-3

Neutron scattering

We will focus on what happens to the neutrons neither transmitted nor adsorbed by nuclei, but **just scattered** by them



10.1016/B978-0-12-820549-5.00010-3

Double-differential cross section

Since the neutron-nucleus interaction is probabilistic, we need a way to quantify
WHAT PRECISELY IS MEASURED IN A NEUTRON SCATTERING EXPERIMENT

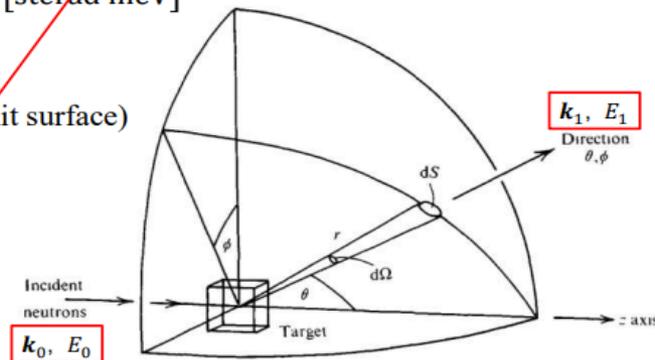
$$\left(\frac{d^2\sigma}{d\Omega dE_1} \right) \doteq \frac{\text{No. of neutrons scattered per second in } d\Omega(\theta, \phi) \text{ and in } [E_1, E_1 + dE_1]}{\Phi d\Omega dE_1} =$$

$$= \frac{1}{\Phi} \frac{d^2N(\theta, \phi, E_1)}{d\Omega dE_1} \quad \left[\frac{\text{m}^2}{\text{sterad meV}} \right]$$

with

$\Phi \doteq$ flux of incident neutrons
 = (neutrons per second through unit surface)

$$1 \text{ barn (bn)} \doteq 10^{-24} \text{ cm}^2$$



Is a measure of the **probability that a specific nuclear interaction will occur** between an incident particle and a target nucleus.

It represents **the effective target area of a nucleus for a collision with a neutron**.

The cross-section **is not the physical size of the nucleus but a quantum mechanical measure of a specific interaction's likelihood**. For example, the scattering cross-section of a nucleus for slow neutrons can be much larger than its actual geometric area.

There are total, absorption, and **scattering cross-sections**.

Double-differential cross section

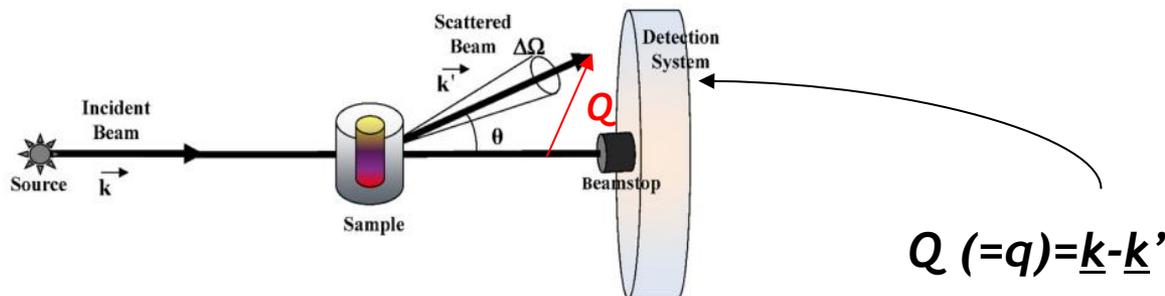
$$\left(\frac{d^2\sigma}{d\Omega dE_1} \right) = \frac{k_1}{k_0} \frac{1}{2\pi\hbar} \sum_{ij} b_i b_j \int_{-\infty}^{\infty} \langle e^{-i\mathbf{Q}\cdot\mathbf{R}_i(0)} e^{i\mathbf{Q}\cdot\mathbf{R}_j(t)} \rangle e^{-i\omega t} dt$$

Neutron flux
and kinematics

Coupling

Sample Properties!
 $R_i(0), R_j(t)$
where atoms ARE
what atoms DO

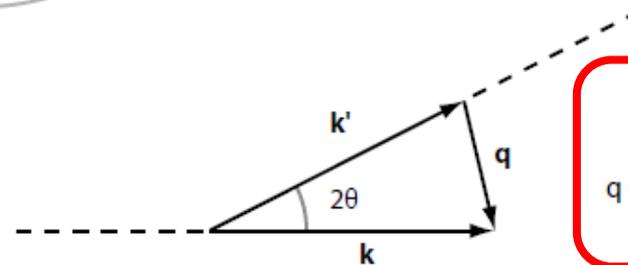
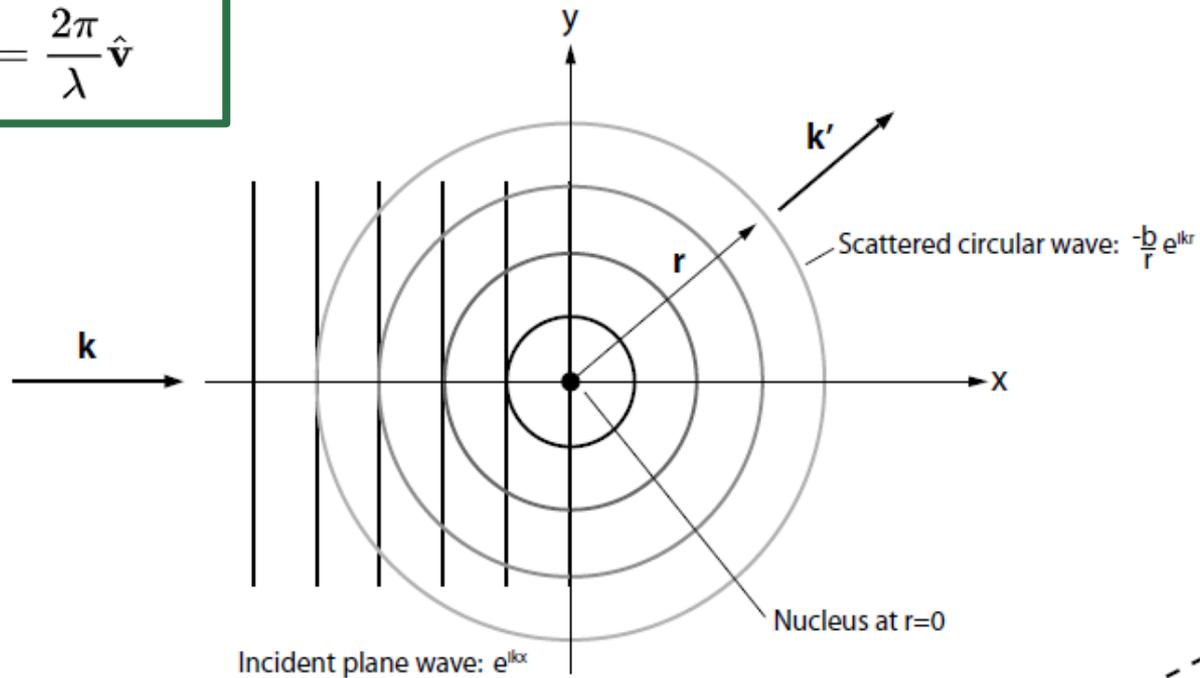
STRUCTURE
and
DYNAMICS



Elastic scattering from a nucleus

Wave vector

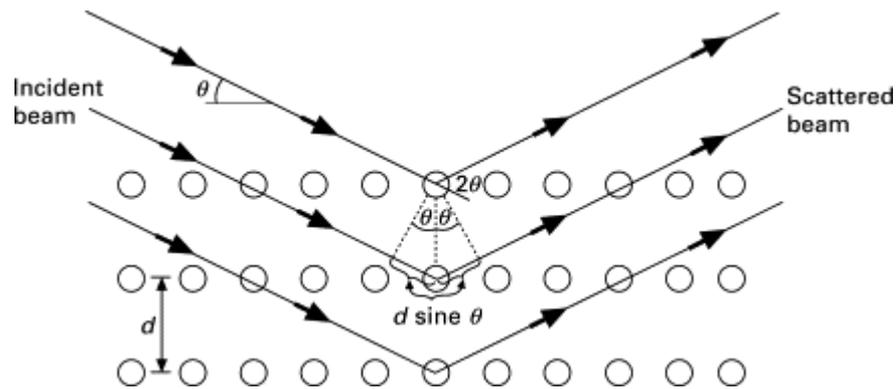
$$\vec{k} = \frac{2\pi}{\lambda} \hat{v}$$



$$\sin\theta = \frac{|q|}{2|\mathbf{k}|}$$
$$q = 2k \sin\theta = \frac{4\pi}{\lambda} \sin\theta$$

What about an ensemble of nuclei?

Interference of scattered waves



A.R. Bunsell, 2009

What about an ensemble of nuclei?

The first one who applied the properties of the Fourier transform to the experiments of X-ray diffraction in the crystal was W.H. Bragg

IX. BAKERIAN LECTURE.—*X-rays and Crystal Structure.*

By W. H. BRAGG, D.Sc., F.R.S., *Cavendish Professor of Physics in the University of Leeds.*

Lecture delivered March 18,—MS. received April 7, 1915.

THE method of investigating crystalline structure by the use of X-rays has already been explained in papers read before this Society. It will be convenient nevertheless to re-state its principle very briefly in order to introduce some further considerations which I propose to lay before you.

The statement of the principle may be made in the following way. Let a train of waves of length λ be passing through a medium in which are particles having the power of scattering the radiation. Suppose, further, that the scattering power is not distributed evenly through the medium, but that directions can be found along each of which there is a periodic variation of the scattering power of the material contained in strata perpendicular to the given direction, strata being, of course, taken of equal thickness for comparison. Let the distance of recurrence or spacing be called d . Let θ be the angle between the rays and the strata. Then there will be a "reflection" of the radiation by the medium of $n\lambda = 2d \sin \theta$, where n is any integer.

For instance, the Lippmann process of colour photography produces such a distribution of scattering power in the sensitive film through the agency of stationary waves.* If light is incident on the film it is strongly reflected when $\lambda = 2d \sin \theta$; if the light is white the film selects the appropriate wave for reflection, and this is the origin of the colour manifestation.

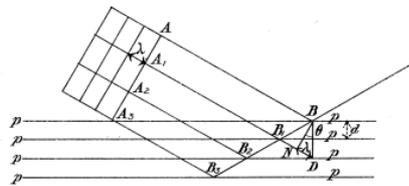


Fig. 1.

The formula is readily explained by aid of the figure, which shows a set of regularly spaced planes p, p, p, \dots , each reflecting a minute fraction of the incident

* 'Physical Optics,' WOOD, p. 149.

Double differential cross section and scattering length

It is possible to derive that -for elastic scattering in a spherical scattering symmetry- the cross section can be written as:

$$\sigma = 4\pi b^2$$

Where **b** is the (bound) **SCATTERING LENGTH**

b is related to the interaction with each nucleus due to

- i) Nuclide composition
- ii) Neutron-nucleus combined spin

If we have many nuclei?

→ We calculate the coherent scattering length of all nuclei in a volume, divided by that volume, also called **SCATTERING LENGTH DENSITY**

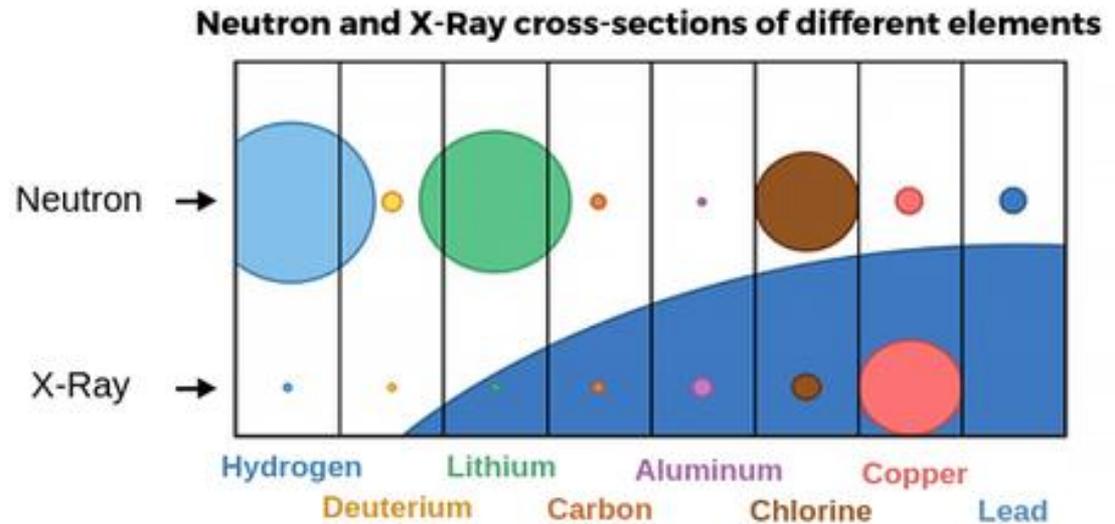
$$SLD = \frac{\sum b_i}{V}$$

Double differential cross section and scattering length

The processes by which neutrons interact with matter are fundamentally different from those of photons.

While photons primarily interact with atomic electrons, neutrons interact mainly with the atomic nucleus.

Describing the interaction of a neutron with a nucleus involves complex interactions among all the nucleons within the nucleus and the incoming neutron. As a result, **there is currently a lack of fundamental theories that can accurately predict the variations in neutron cross-sections.**



To sum up

- Neutrons interact with nuclei, not the electron cloud
- We use the concept of cross-section to quantify this probabilistic interaction
- The specific measurement is the double-differential cross section, which gives detailed information about the direction and energy of scattered neutrons
- The experimental data is ultimately related to the scattering length density (SLD), key parameter for describing the scattering power of a material at the nanoscale.

To sum up...consequences

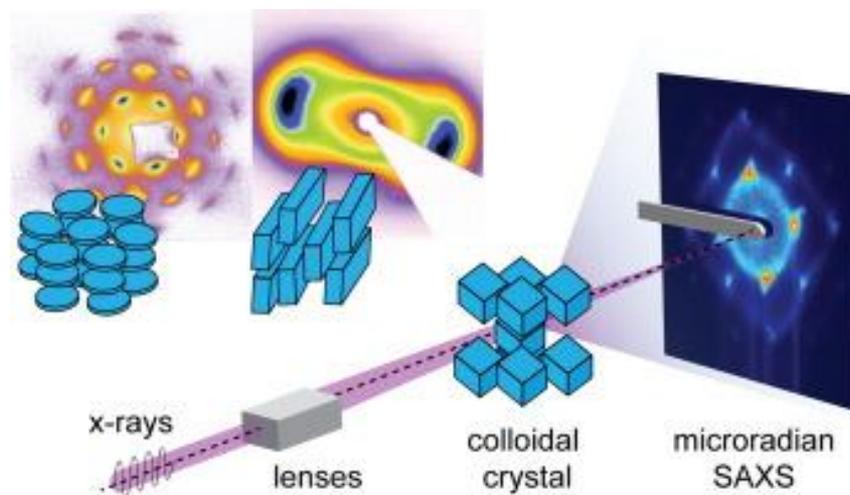
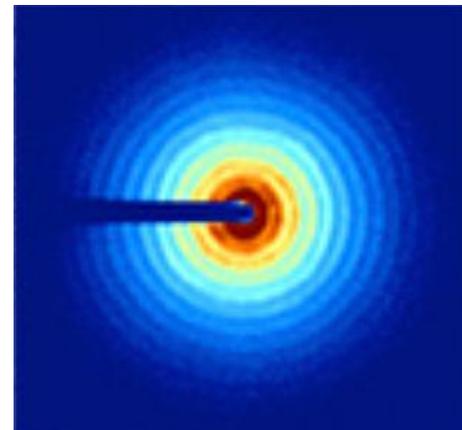
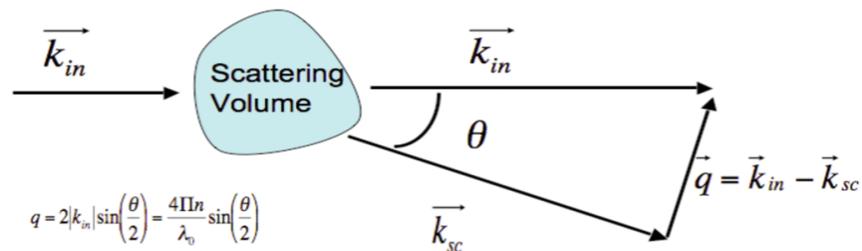
- Neutrons penetrate **bulk samples**

As neutral particles, neutrons are not repelled by the electron clouds of atoms, so they can travel deep into materials and interact directly with atomic nuclei.

- Sensitive to **isotopes**

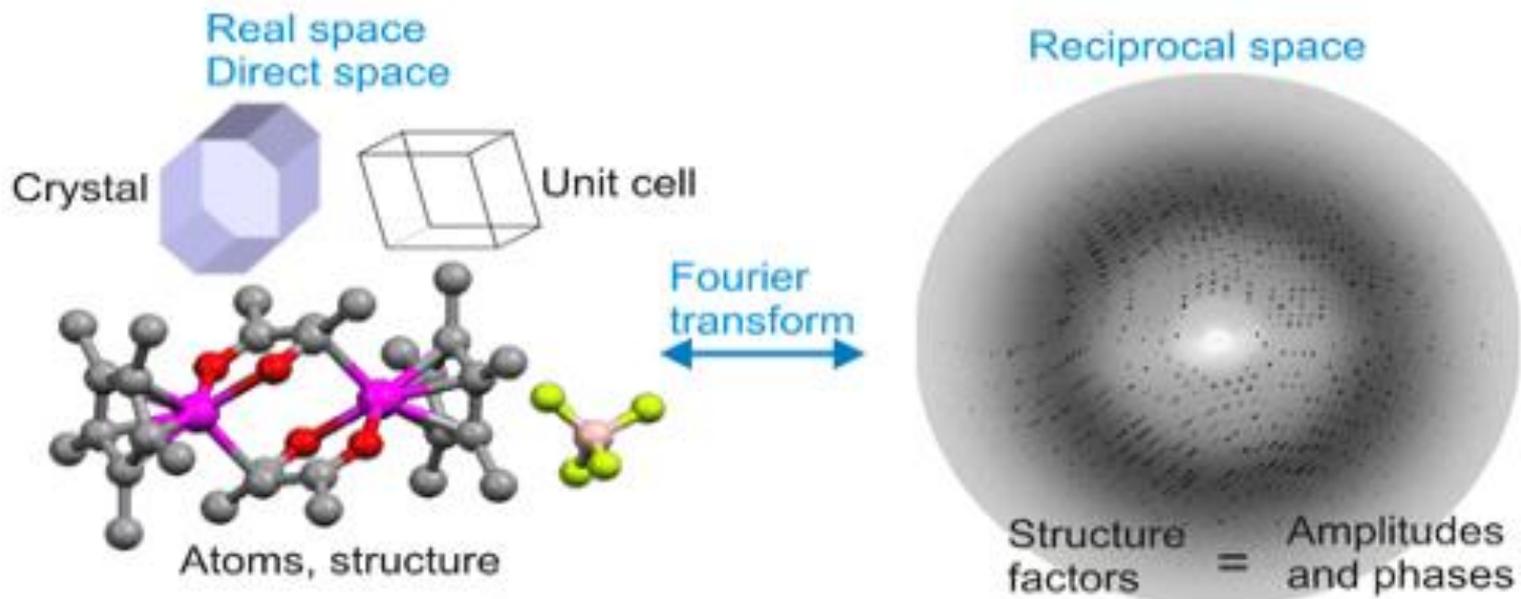
Neutron interaction cross-sections vary significantly between different isotopes of the same element, enabling neutron scattering techniques to distinguish between and analyze different isotopes within a material.

What do we see on the detector?



V. Petukhov, Janne-Mieke Meijer, Gert Jan Vroege, Current Opinion in Colloid & Interface Science
 Volume 20, Issue 4, August 2015, Pages 272-281

What space?



$$\rho(xyz) = \frac{1}{V} \sum_{hkl} \underset{\text{Amplitudes}}{|F(hkl)|} \cdot e^{-2\pi i [hx+ky+lz - \underset{\text{Phases}}{\phi(hkl)}]}$$

Real space
Direct space

Reciprocal space

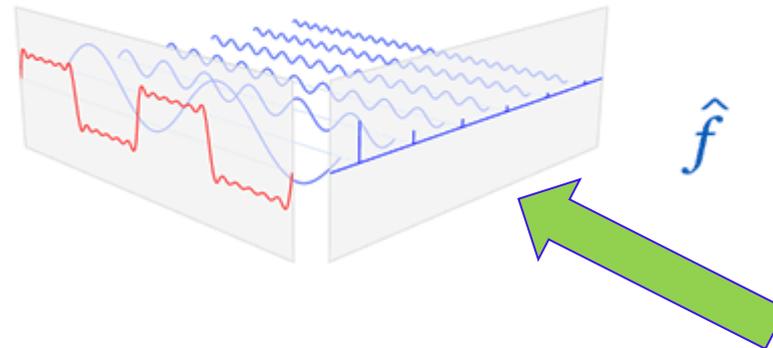


What space?

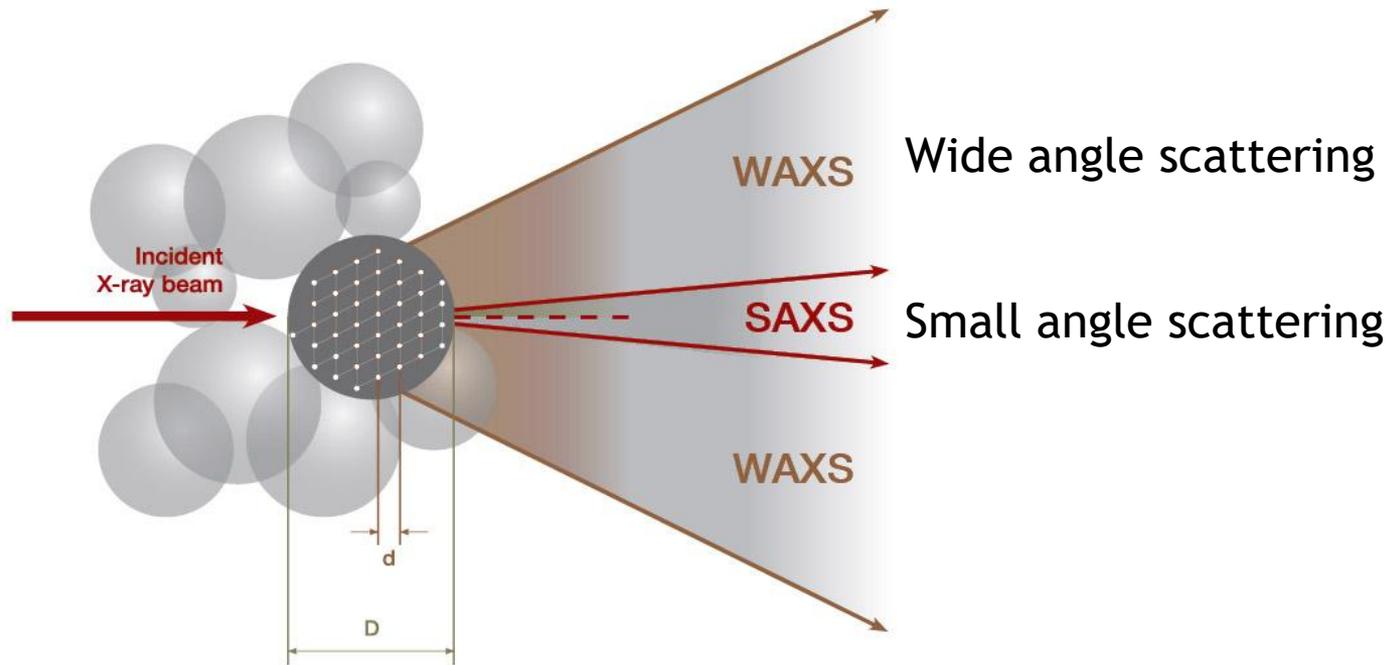
Fourier transform is a mathematical tool that decomposes any periodic function into a sum of sinusoidal functions (frequency dependent)



$$a_n \cos(nx) + b_n \sin(nx)$$



What space?



<https://wiki.anton-paar.com/en/saxs-nanostructure-analysis/>

Neutron scattering

Collision can be either *elastic* or *inelastic*



Structure, shape, size

Elastic scattering:

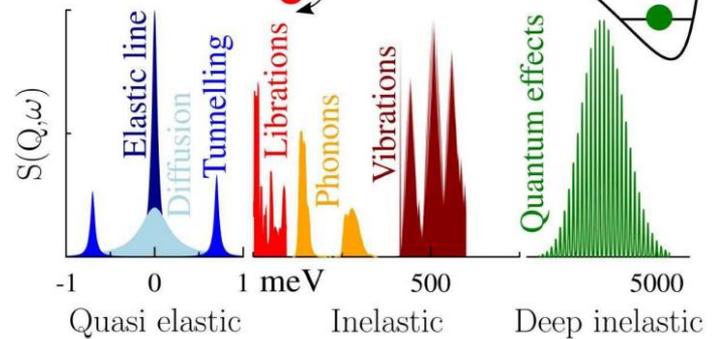
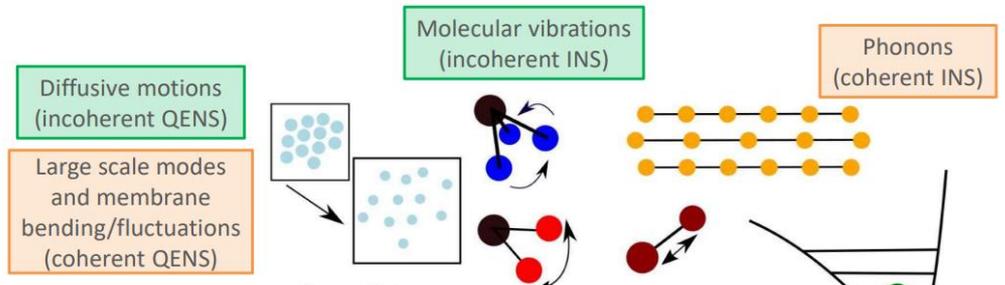
$$|k_i| = |k_f| = \frac{2\pi}{\lambda}; \quad Q = 2|k|\sin\theta = \frac{4\pi}{\lambda}\sin\theta$$

$$\Delta E = 0$$

Inelastic scattering:

$$|k_i| \neq |k_f|;$$

$$\Delta E = \hbar\omega = E_i - E_f = \frac{\hbar}{2m}(k_i - k_f)$$



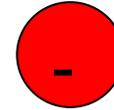
Vibrations, rotations, diffusion, dynamics



Neutrons vs X-rays



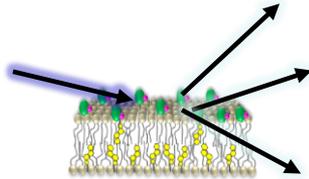
- ✓ Interaction with nucleus, not with the electron cloud
- ✓ Sensitivity to isotopes
- ✓ Can measure magnetic structures
- ✓ Can measure lattice vibrations
- ✓ Large penetration depth (depending on isotope)
- ✓ No radiation damage
- ✓ Contrast variation techniques



- ✓ Hard to get high neutron flux
 - need for big facilities
 - need for large samples
- ✓ Samples activation
- ✓ Incoherent background (especially from ^1H)

COMPLEMENTARY

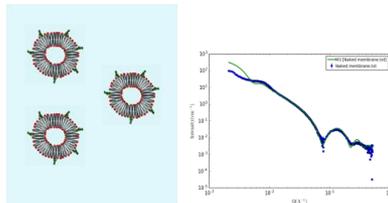
Why elastic neutron scattering on model membranes?



- ✓ Probe relevant length (\AA to μm) and time (ps to hr) scales
- ✓ Non-destructive
- ✓ Possibility of selective deuteration to play with contrast
- ✓ Deep material penetration (buried systems)

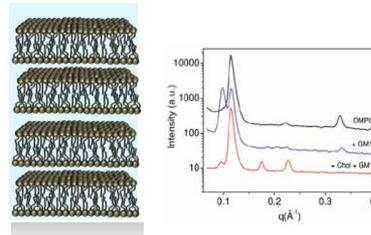
COMPLEMENTARY TECHNIQUES FOR MEMBRANE STRUCTURE

SANS from LUVs



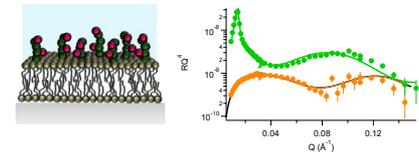
Membrane form factor
Aggregate shape
Structuring in solution

ND from membrane stacks



Membrane thickness/thicknesses
(lateral domains)

NR from SLB



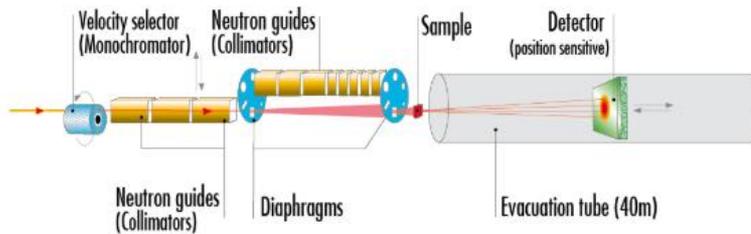
Membrane transverse structure

Techniques for structural characterization

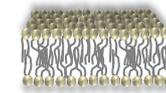


In bulk:

Small Angle Neutron Scattering
-SANS-

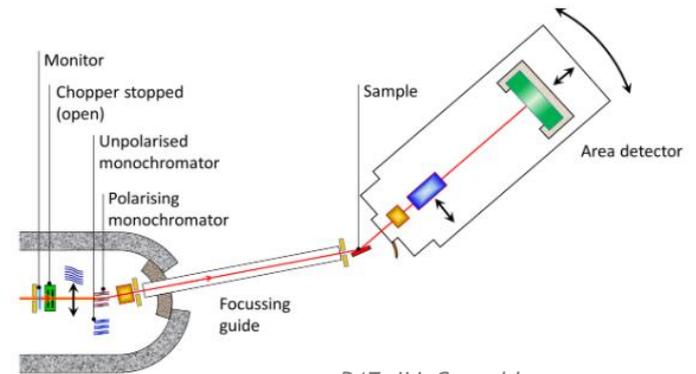


D11, ILL Grenoble



At interfaces:

Neutron Reflectometry
-NR-



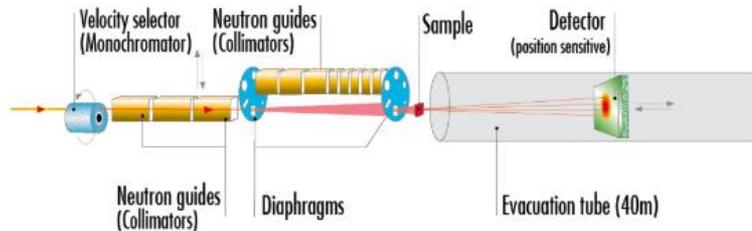
D17, ILL Grenoble

Techniques for structural characterization



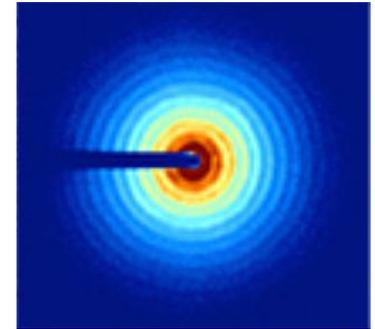
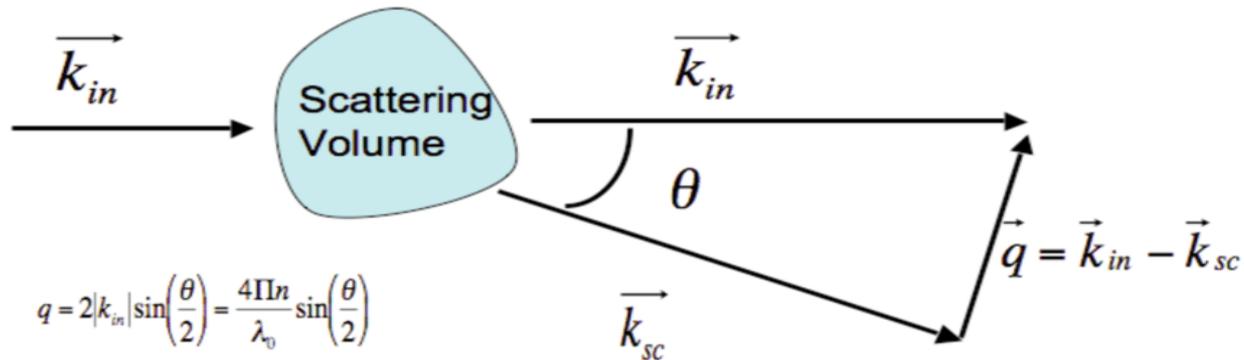
In bulk:

Small Angle Neutron Scattering -SANS-



D11, ILL Grenoble

Scattering from powders



$$I(q) \div c \overset{\text{mass}}{\underset{\text{concentration}}{\uparrow}} M \overset{\text{structure}}{\underset{\text{form}}{\uparrow}} P(q) S(q) \overset{\text{'visibility'}}{\underset{\text{contrast}^2}{\uparrow}}$$

P(q), the FORM FACTOR

$$I(q) \div c M \text{ P}(q) S(q) \text{ contrast}^2$$

Interference of X-rays/neutrons scattered from different parts of the same object

For a sphere of radius r

$$P(q) = \left[\frac{3(\sin(qr) - qr \cos(qr))}{(qr)^3} \right]^2$$

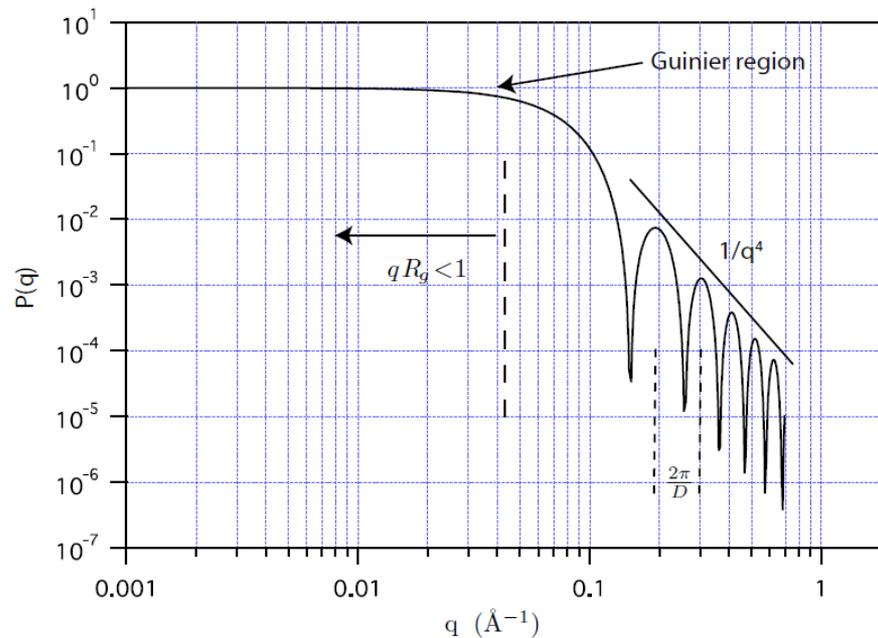
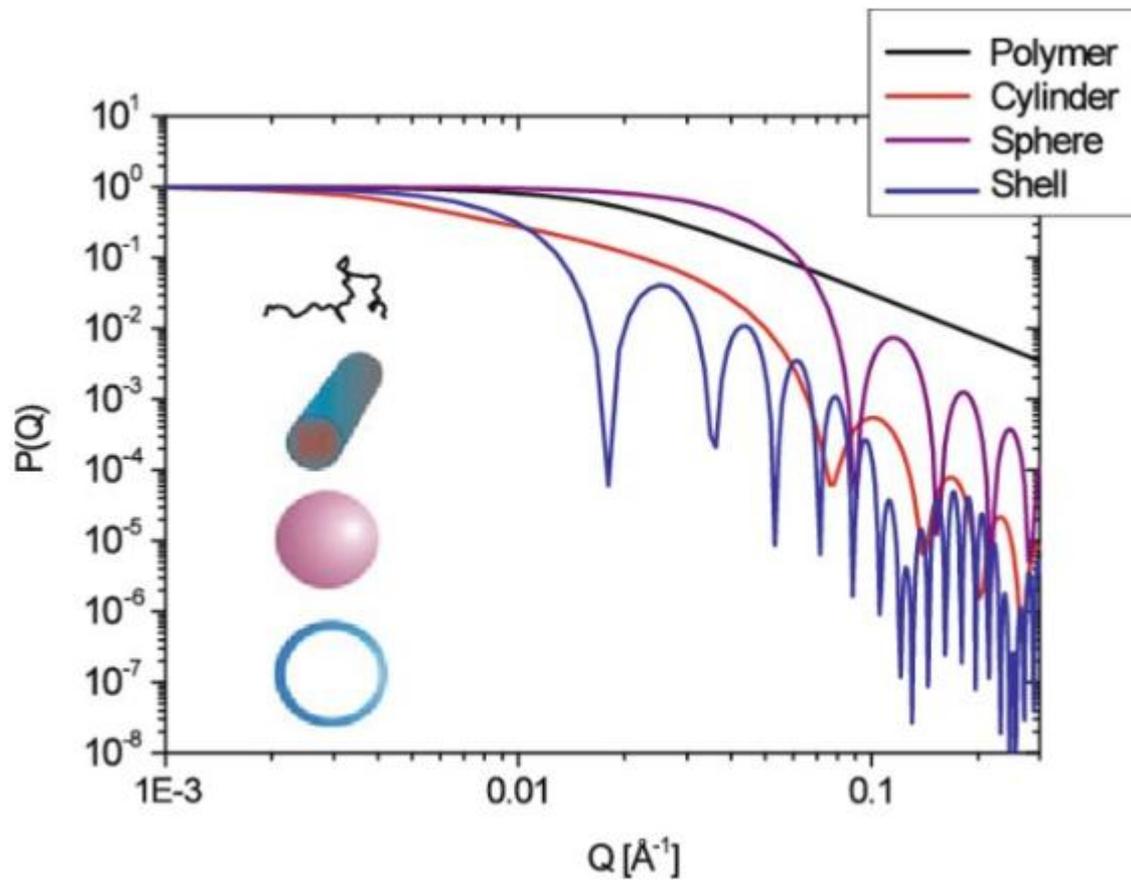


Figure 7: Form Factor spheres of radius 3\AA . $R_g = 23\text{\AA}$

P(q), the FORM FACTOR

$$I(q) \div c M P(q) S(q) \text{ contrast}^2$$



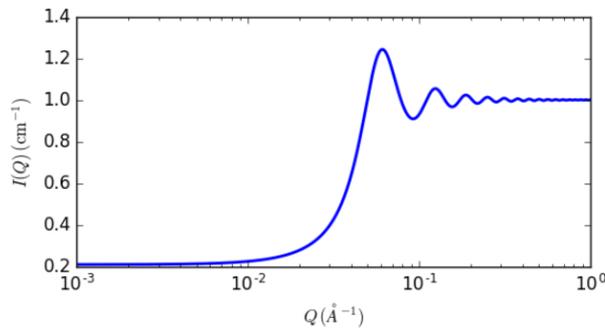
S(q), the STRUCTURE FACTOR (in liquids)

Interference of X-rays/neutrons scattered from different objects in the sample (interparticle correlation)

$$I(q) \div c M P(q) S(q) \text{contrast}^2$$

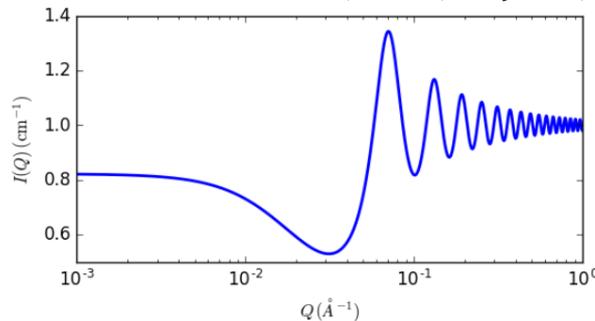
No long range order as crystals

Dilute solutions: interparticle correlation = 0
 $\rightarrow S(q)=1$



J K Percus, J Yevick, *J. Phys. Rev.*, 110, (1958) 1

Hard spheres, radius 50 Å and volume fraction 0.2

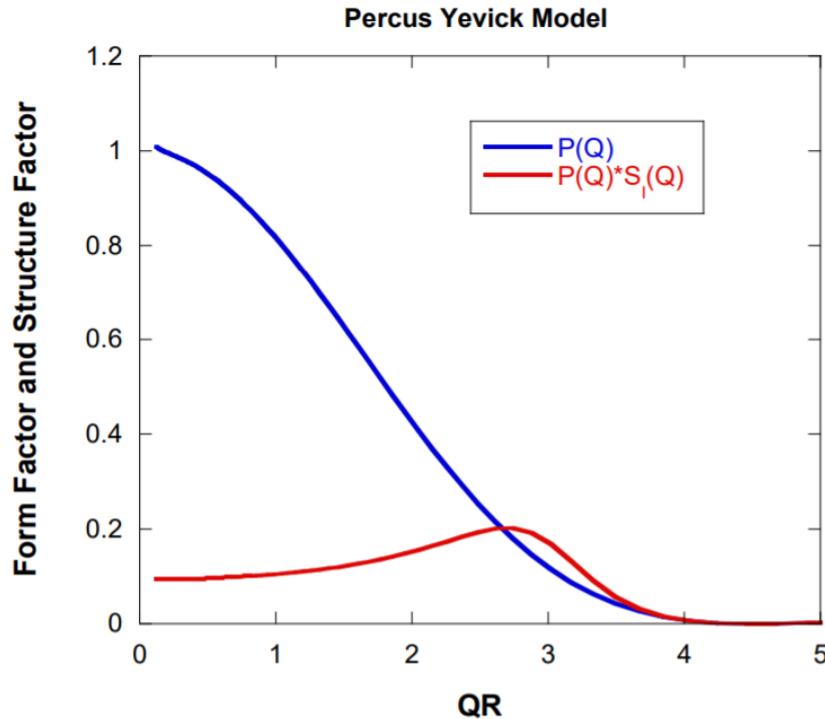


S V G Menon et al., *Chem. Phys.*, 95(12) (1991) 9186-9190

Sticky spheres, radius 50 Å, volume fraction 0.2, stickiness 0.2

<http://www.sasview.org/>

$$I(q) \div c M P(q) S(q) \text{ contrast}^2$$



Form factor $P(Q)$ for isolated spheres (infinite dilution limit), and product $P(Q)S(Q)$ for a solution of spheres with a volume fraction of $\varphi = 0.30$.

https://www.ncnr.nist.gov/staff/hammouda/distance_learning/chapter_32.pdf

Small Angle Scattering

$$I(q) \div c M P(q) S(q) \text{contrast}^2$$

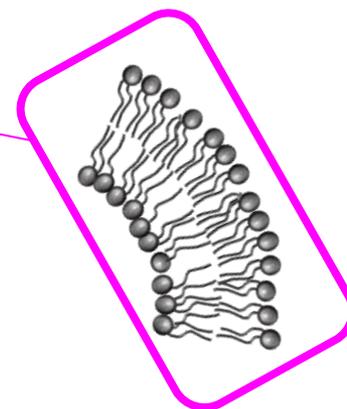
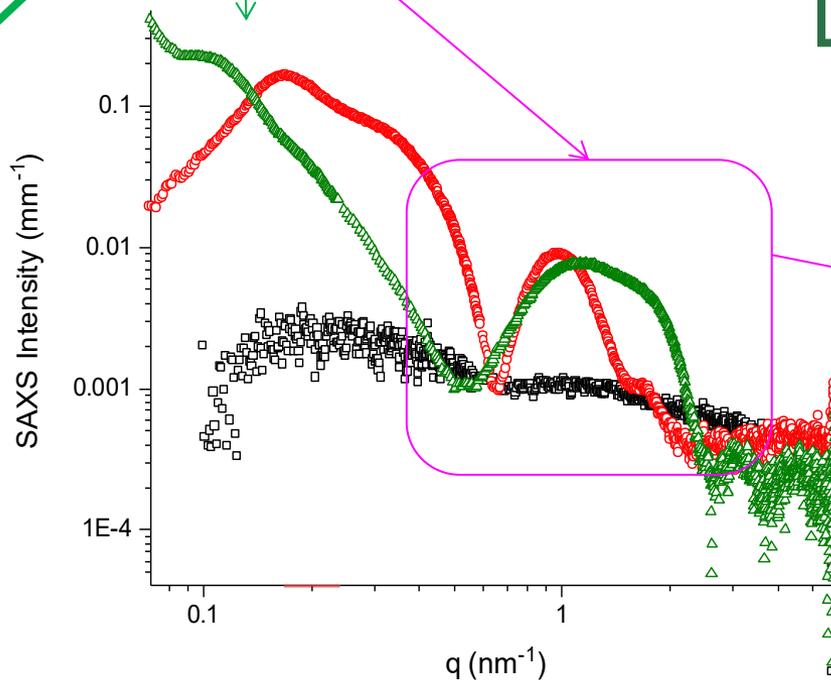
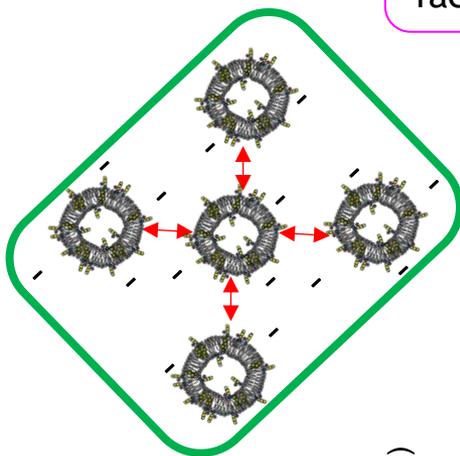
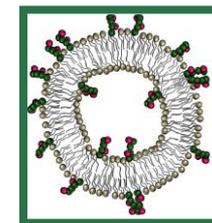
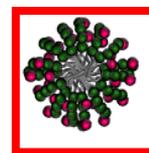
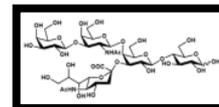
Form factor

Structure factor

Oligo GM1

GM1 micelle

Phospholipid+GM1 LUV

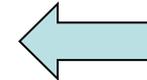


The scattering length density (sld)

$$\rho = \frac{\sum_i^n b_i}{V}$$

ρ scattering length density (SLD)
 b_i scattering length of the relevant atom
 V volume occupied by the n atoms

X-rays: electron density
 ρ_e

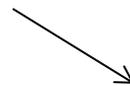


Probability of interaction

Contrast: describes the differences in SLDs from different phases of the sample

$$I(q) \div c M P(q) S(q) \text{ contrast}^2$$

$$\frac{d\Sigma}{d\Omega}(q) = \frac{N}{V} (\rho_1 - \rho_2)^2 V_p^2 P(q) S(q)$$



What counts is the contrast DIFFERENCE

Contrast variation

Isotope	conc	Coh b	Inc b
1H	99.985	-3.7406	25.274
2H	0.015	6.671	4.04

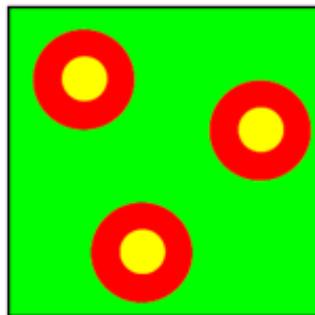
<https://ncnr.nist.gov/resources/n-lengths/element/h.html>

$$\rho = \frac{\sum_i^n b_i}{V}$$

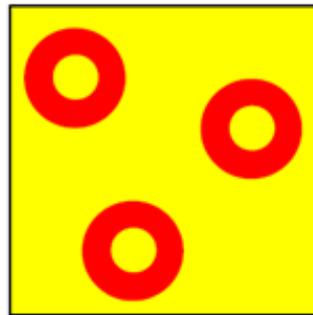
$$\frac{d\Sigma}{d\Omega}(q) = \frac{N}{V} (\rho_1 - \rho_2)^2 V_p^2 P(q) S(q)$$

H-D selective substitution

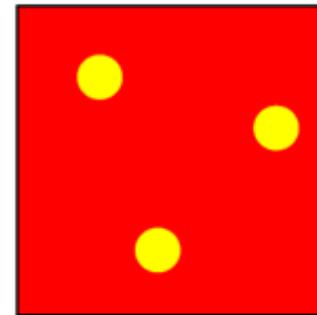
Core-shell particles



Natural contrast



r solvent = r core
(shell visible)



r solvent = r shell
(core visible)

A.J. Jackson, NIST Center for Neutron Research (2008)

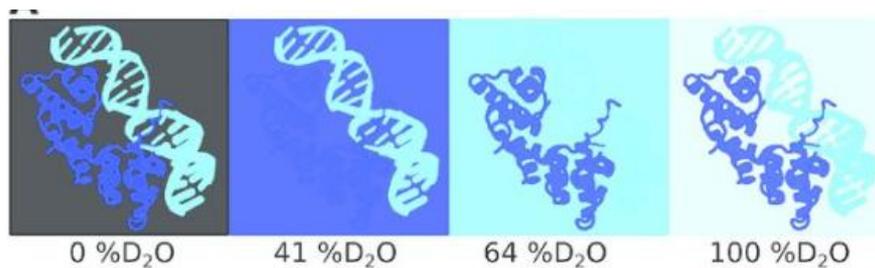
We can perform SANS from the different solutions and fit simultaneously to the same model varying only the SLDs

Contrast variation



Z. Bu et al., *Adv Protein Chem Struct Biol.* 2011;83:163-221

Protein-DNA complex



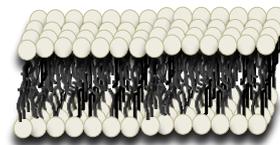
M.M. Castellanos et al., *Computational and Structural Biotechnology Journal* (2016)

Why different radiations?

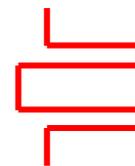
$$I(q) \div c M P(q) S(q) \text{contrast}^2$$

X-ray

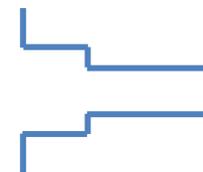
Good contrast for SUGARS (high electron density)



X-rays



Neutrons

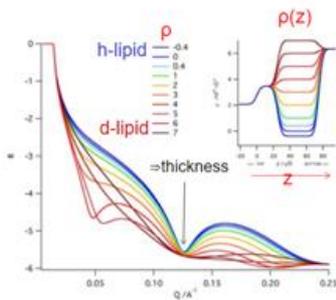


Neutron

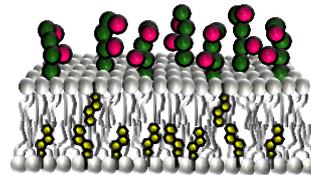
Hydrogen coherent scattering length : -3.74 fm
Deuterium coherent scattering length : $+6.67$ fm



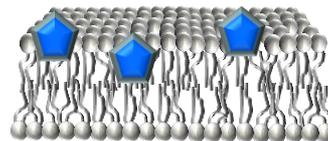
Playing with selective deuteration protiated molecules can be evidenced in the deuterated phospholipid matrix



H. Wacklin, ESS



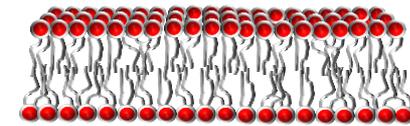
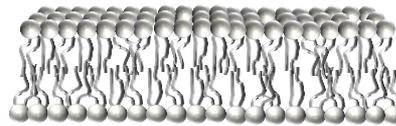
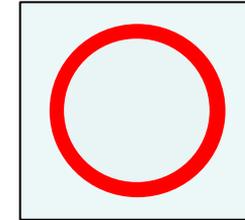
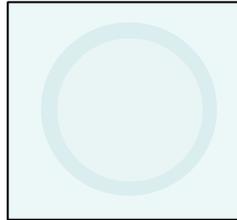
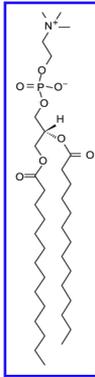
Membrane components distribution



External interacting molecules distribution

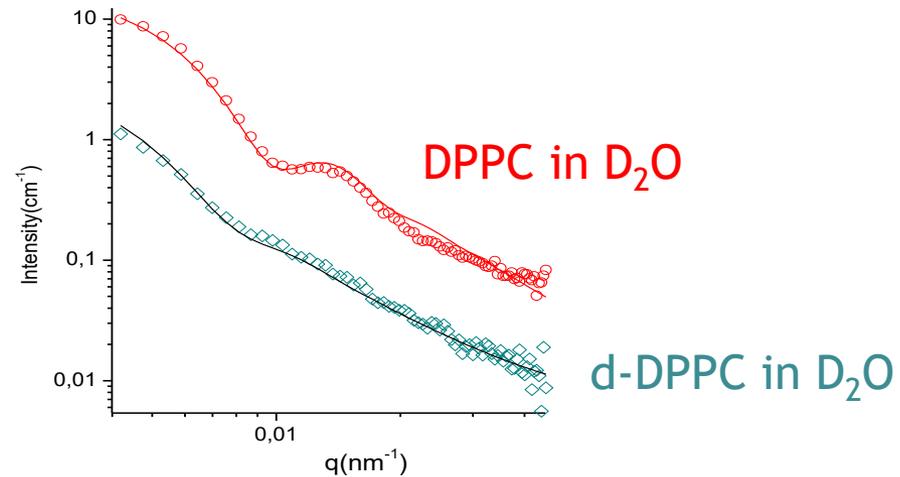
Small Angle Neutron Scattering SANS

DPPC vesicles

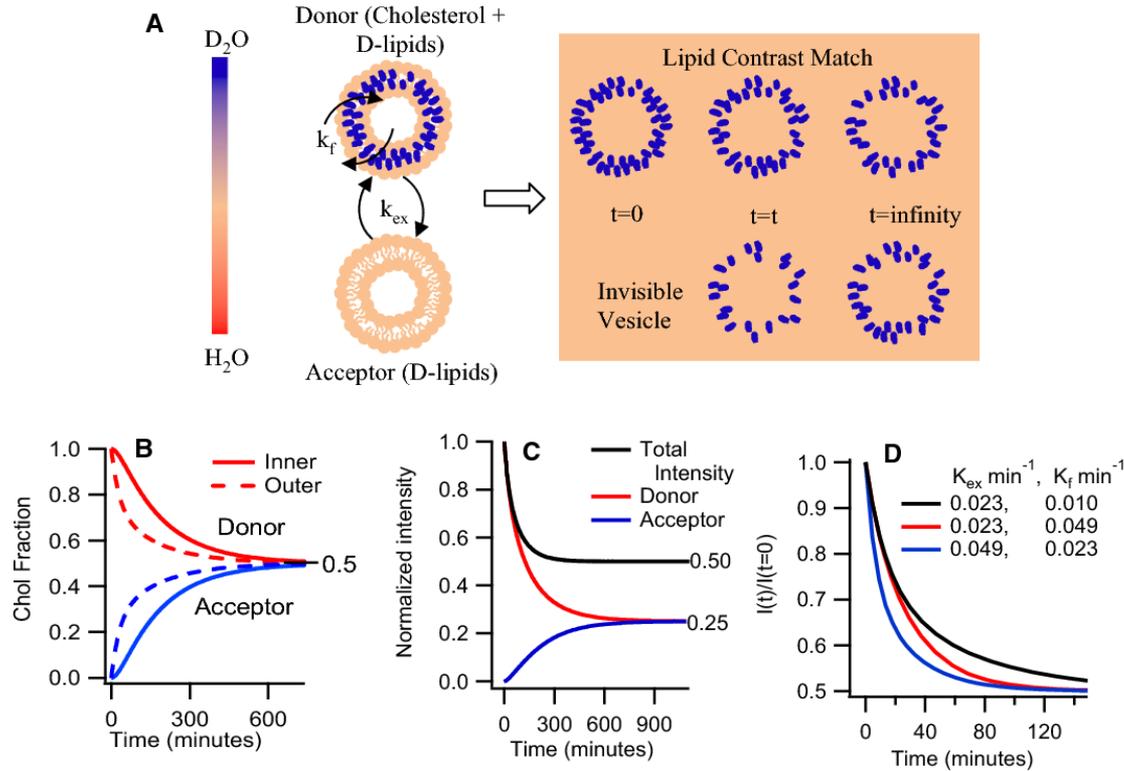


Play with selective deuteration

→ Play with contrast, visibility!

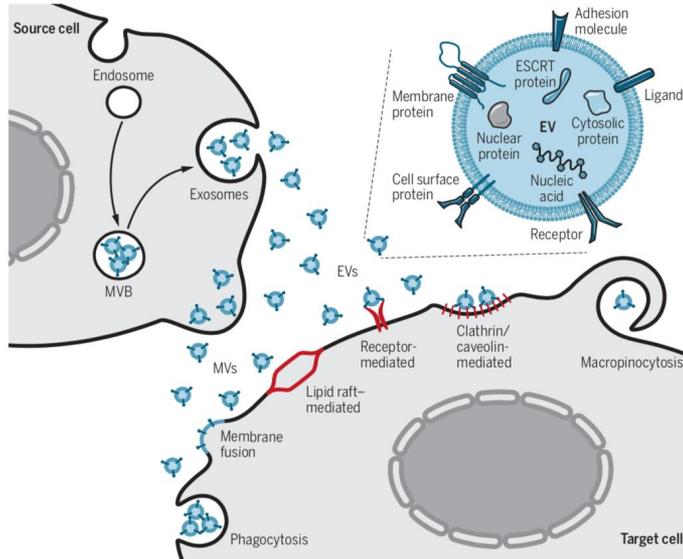


SANS reveals cholesterol transport in lipid membranes



S. Garg *et al.*, Noninvasive neutron scattering measurements reveal slower cholesterol transport in model lipid membranes, *Biophysical Journal* (2011)

Extracellular vesicles - EVs

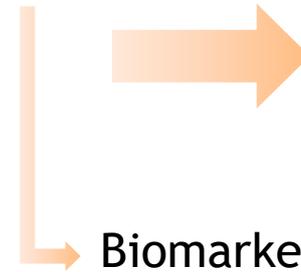


Wiklander et al., *Science Trans. Medicine* (2019)

Definition

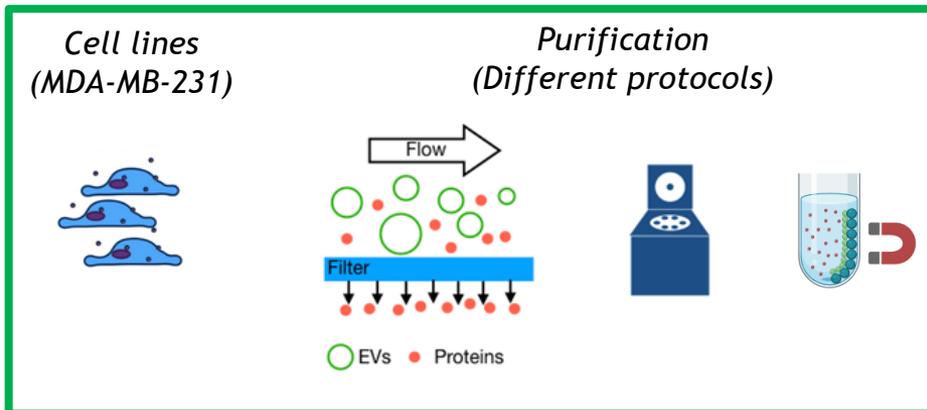
« Particles that are released from the cells, delimited by a lipid bilayer and cannot replicate on their own »

Specific signatures from originating cells

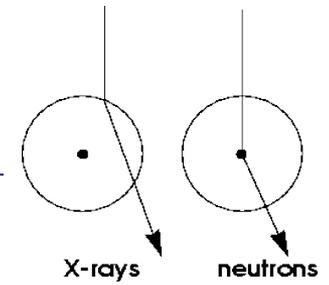


Influence on the fate of recipient cells

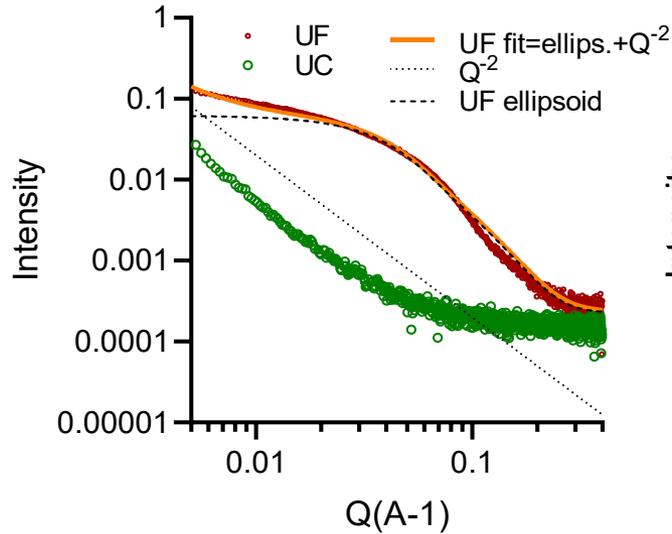
Biomarkers



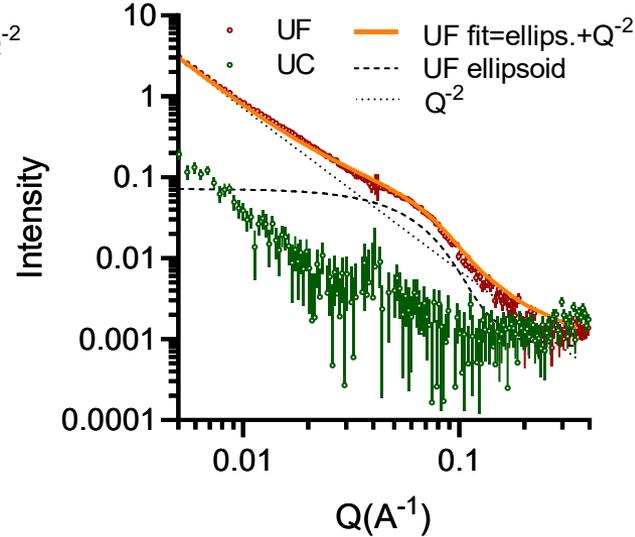
EVs characterization



X-ray scattering

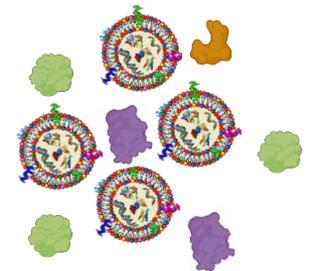


Neutron scattering

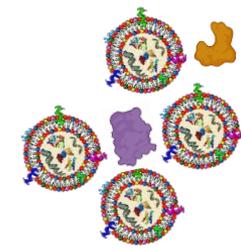


Complementary!

Ultrafiltered: 2D objects + ellipsoidal objects (proteins)



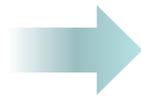
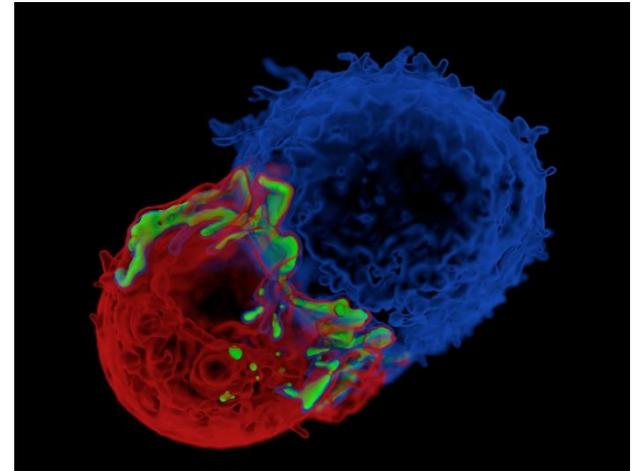
Ultracentrifuged: mainly 2D objects (membranes)



Cell (plasma membrane) interaction with approaching bodies

Biology question is far...

what can we face?



simplified systems

keeping the *main 'bio'-features*



ARTISTIC RENDERING OF THE ZIKA VIRUS PREPARING TO ENTER A CELL (BLUE) BY BINDING TO ITS PROTEIN RECEPTORS (GREEN)

David Goodsell, The Scripps Research Institute

Challenging aspects

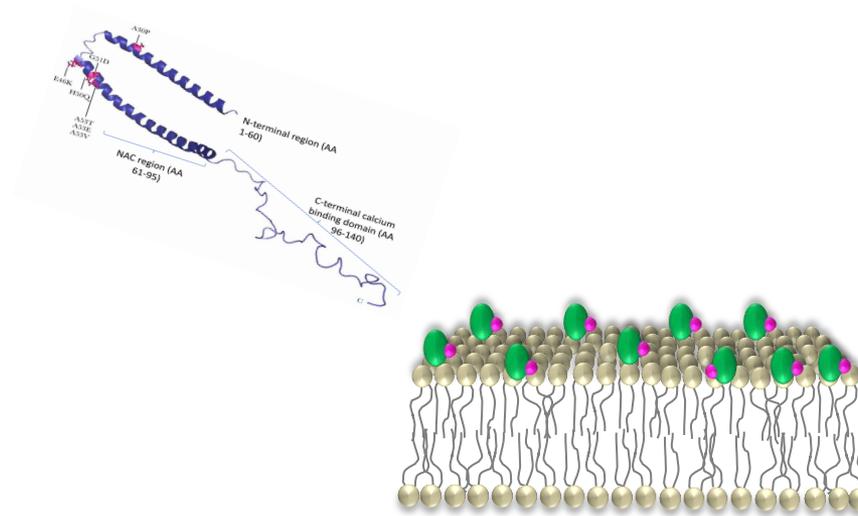


The model



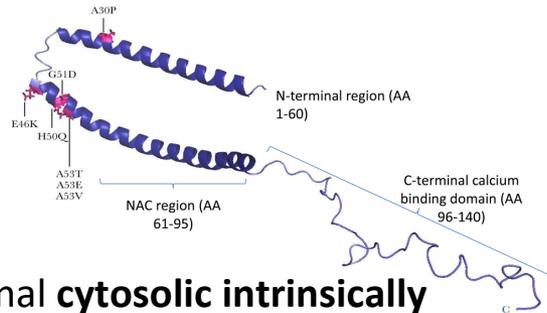
The accessibility (the techniques)

Protein-membrane interaction

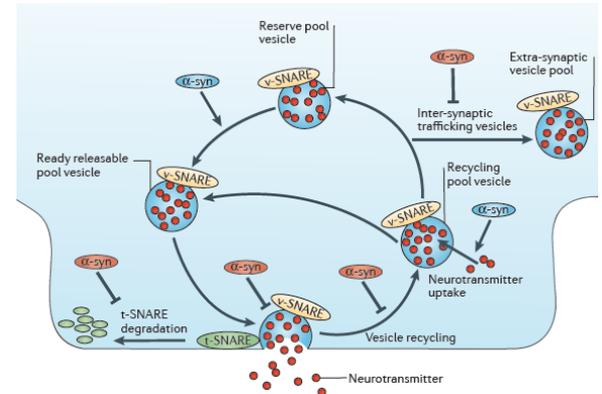


Interaction of **Alpha synuclein**,
the main protein involved in Parkinson's disease,
with artificial lipid bilayers

The protein: Alpha synuclein

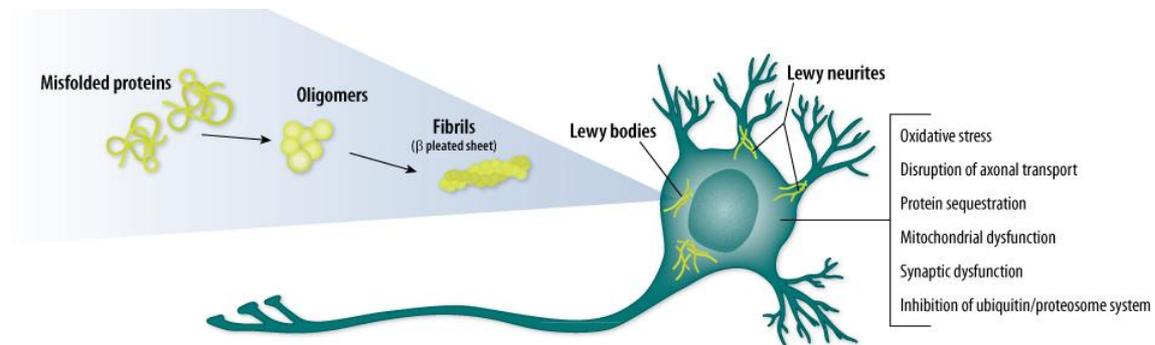


- 140 aa neuronal **cytosolic intrinsically disordered protein**
- **Presynaptic vesicle homeostasis** (*in-vivo* lipid raft binding)



Lashuel et al. Nat. Rev. Neurosci. (2012). 14, 38-48.

In Parkinson's disease: misfolding and aberrant aggregation



© R&D Systems, Inc.

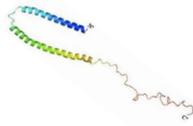
Aim

Investigate the interaction of Alpha synuclein and its aggregates
with different model membranes

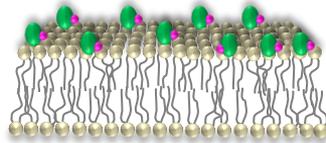
Why?

Type and extent of
interaction may depend on
membrane composition
(chemistry and lipid phase)

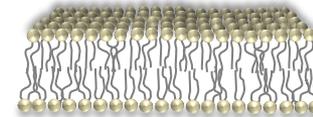
Monomer



Oligomer
(Fe mediated formation)



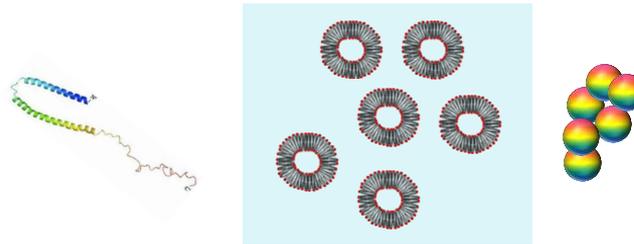
Raft-like
(Ceramide-lipid containing)



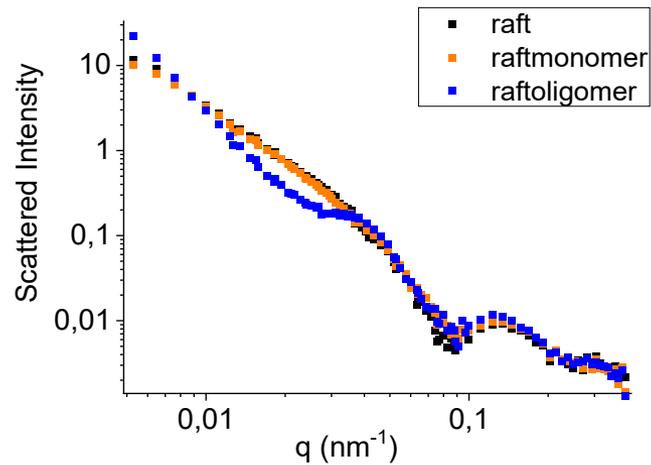
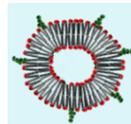
Non raft
(phospholipid only)

SANS for structural investigation

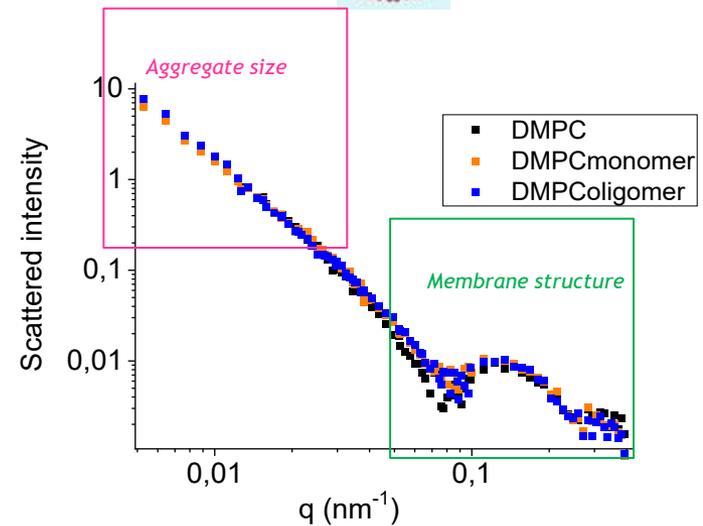
The system: vesicles in solution



DMPC-GM1 ganglioside (Raft like)

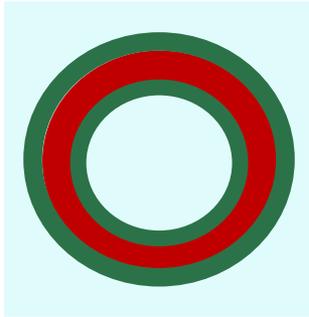


DMPC (non raft)



SANS for structural investigation

Core Multi-Shell Model

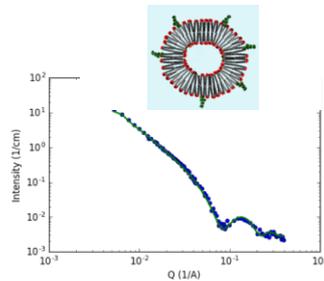


$$P(q) = \frac{\text{scale}}{V} F^2(q) + \text{background}$$

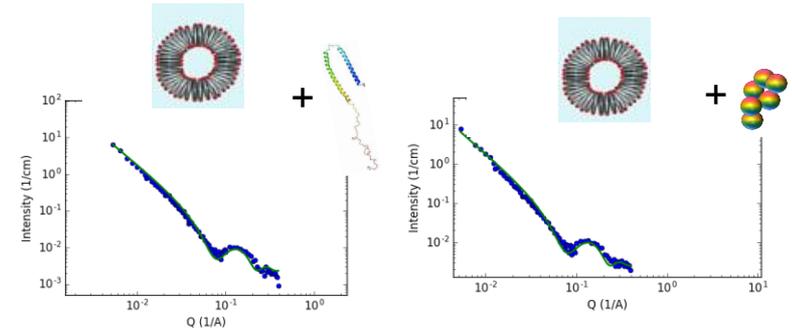
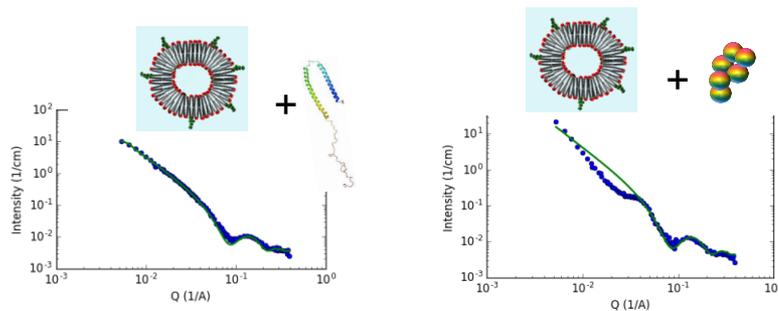
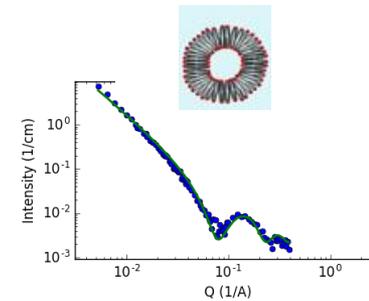
$$F(q) = \frac{3}{V_s} \left[V_c(\rho_c - \rho_s) \frac{\sin(qr_c) - qr_c \cos(qr_c)}{(qr_c)^3} + V_s(\rho_s - \rho_{\text{solv}}) \frac{\sin(qr_s) - qr_s \cos(qr_s)}{(qr_s)^3} \right]$$

(Guinier, 1955)

Raft like

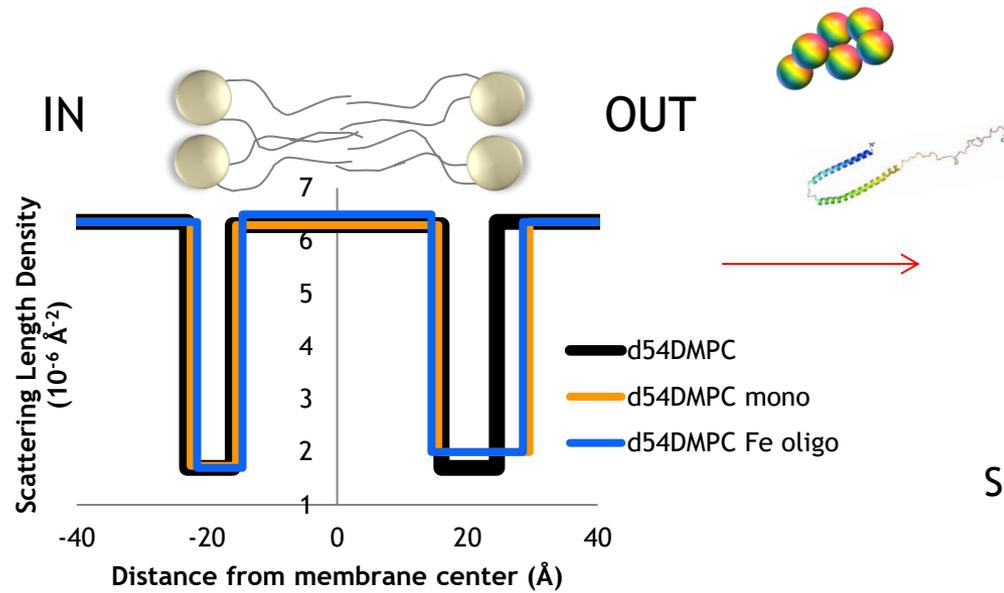
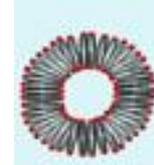
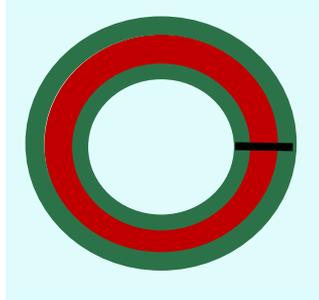


Non raft



SANS for structural investigation

Bare phospholipid target membrane



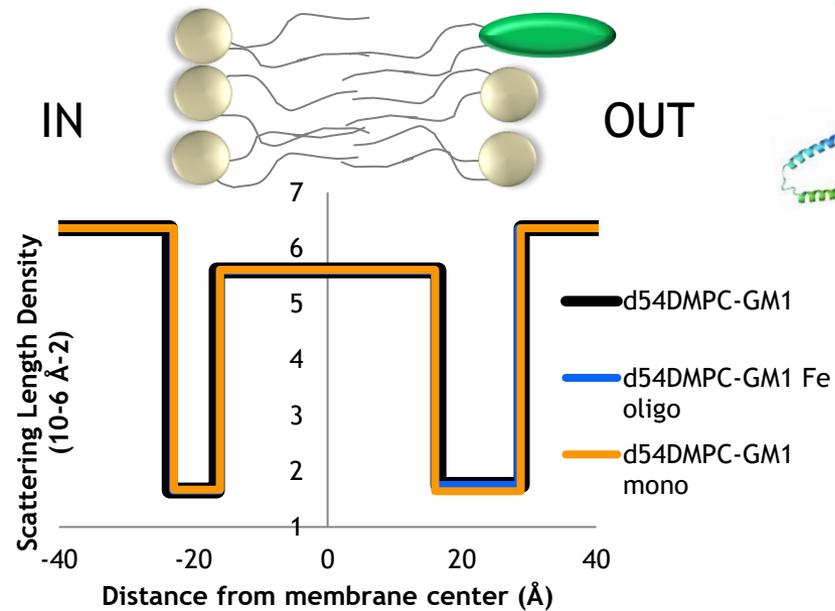
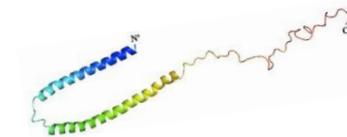
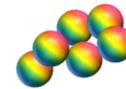
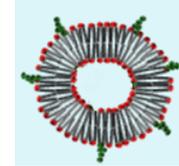
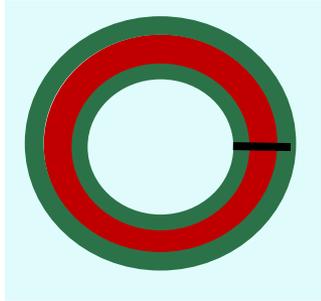
Increased heads thickness

Something on the surface?



SANS for structural investigation

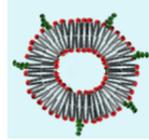
RAFT target membrane



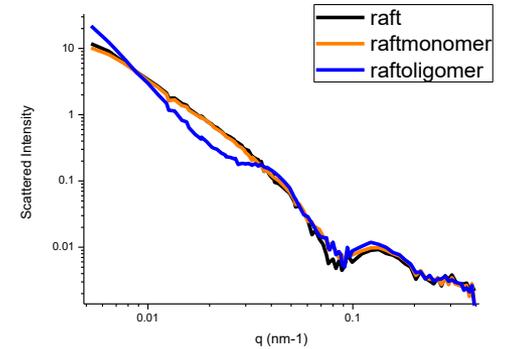
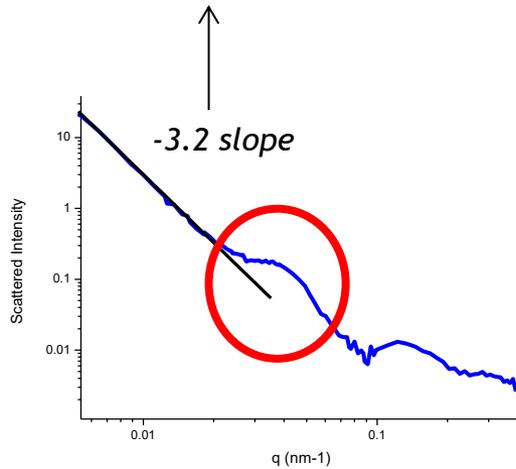
SANS for structural investigation

RAFT target membrane

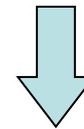
RAFT membrane + oligomers



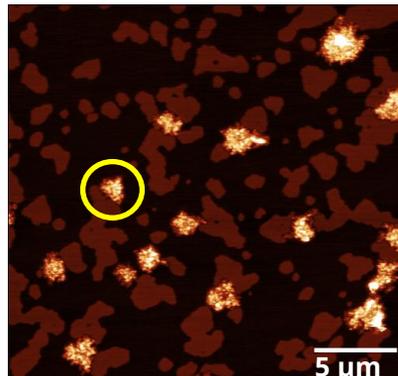
Surface fractal large objects (*Porod regime*)



20 nm sized structures



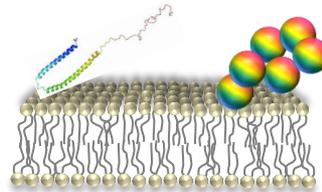
Oligomer cluster on membrane surface?



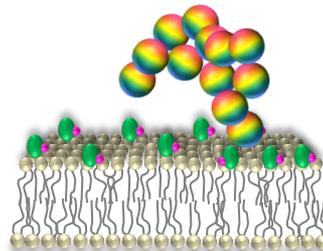
AFM

SANS for structural investigation

- ✓ Both monomers and oligomers interactions is limited to the external membrane leaflet in non-raft domains and results in membrane thickening → role in non raft membrane destabilization?



- ✓ Oligomers accumulate on raft-like domains → strengthen the hypothesis of a critical role of lipid rafts not only for the biological function of the protein, but even for the development and progression of the disease



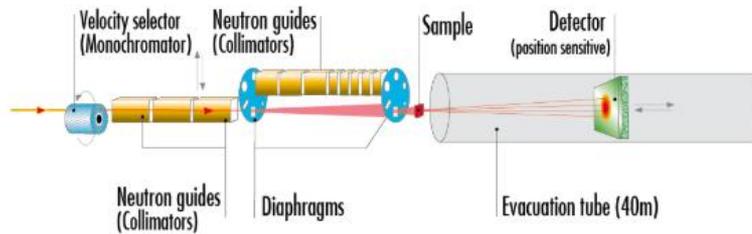
- ✓ MULTITECHNIQUE APPROACH

Techniques for structural characterization

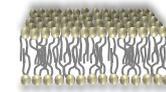


In bulk:

Small Angle Neutron Scattering
-SANS-

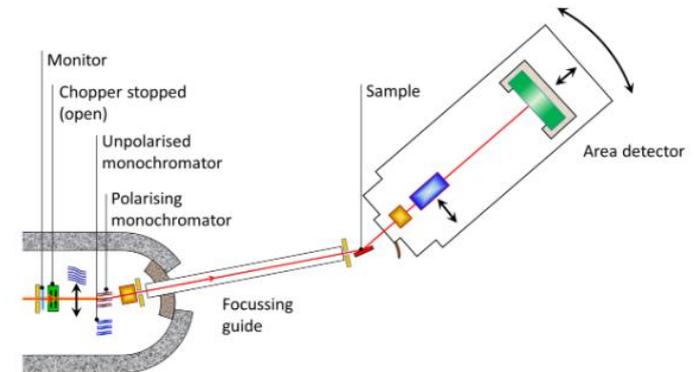


D11, ILL Grenoble



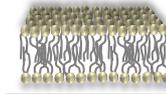
At interfaces:

Neutron Reflectometry
-NR-



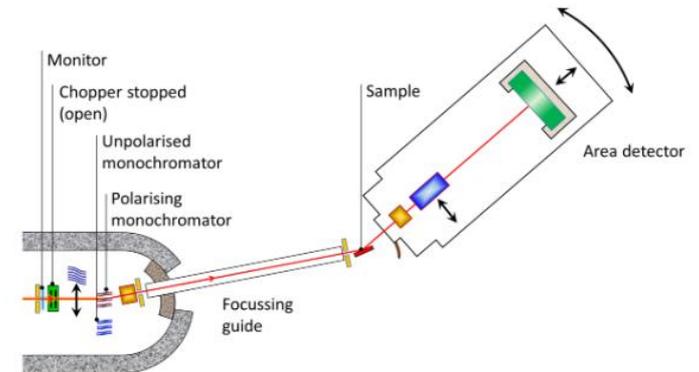
D17, ILL Grenoble

Techniques for structural characterization



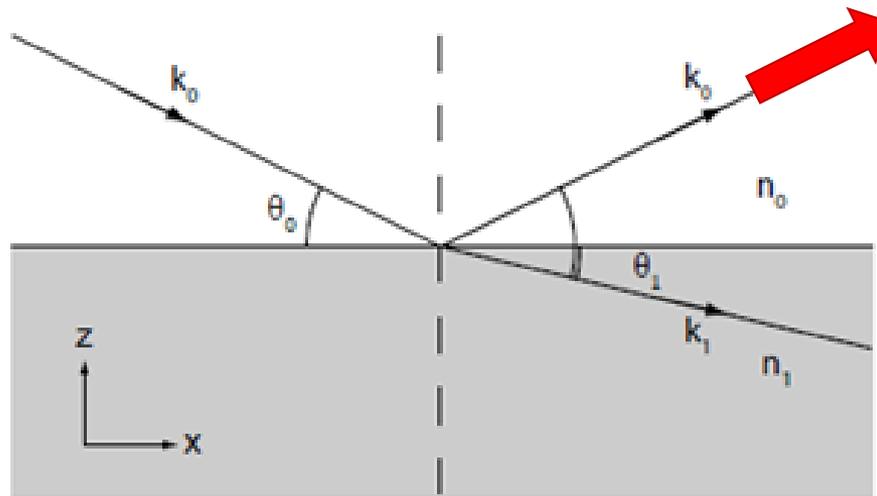
At interfaces:

Neutron Reflectometry -NR-



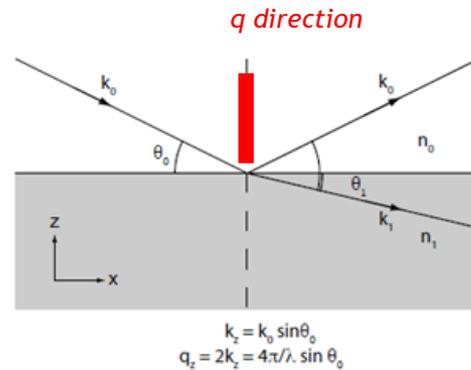
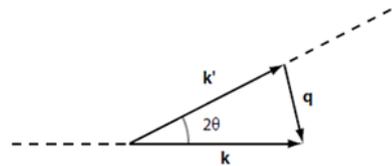
D17, ILL Grenoble

Reflectivity



$$k_z = k_0 \sin \theta_0$$
$$q_z = 2k_z = 4\pi/\lambda \cdot \sin \theta_0$$

Reflectivity



As for small angle scattering, the differential cross section is given by the Fourier transform of the scattering length density distribution over the whole sample. Taking into account the specular geometry such that only q_z varies we can obtain an expression for the reflectivity

$$R(q_z) = \frac{16\pi^2}{q_z^2} |\hat{\rho}(q_z)|^2$$

where $\hat{\rho}(q_z)$ is the *one-dimensional* Fourier transform of the scattering length density profile normal to the interface.

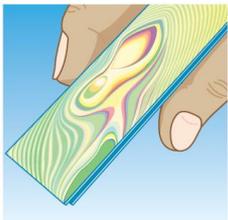
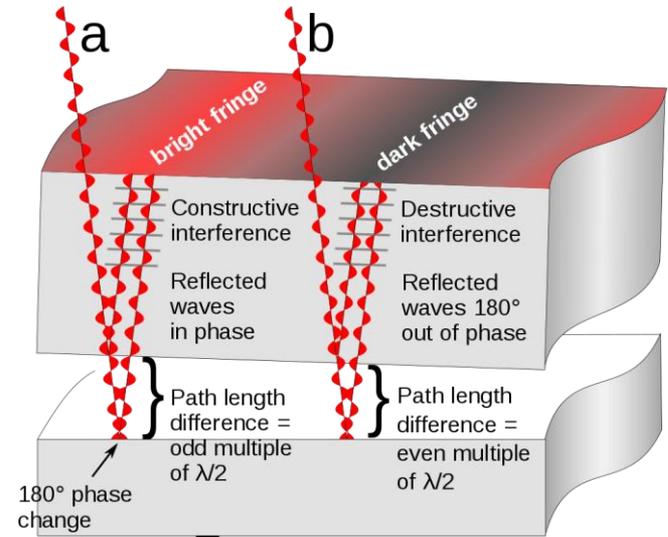
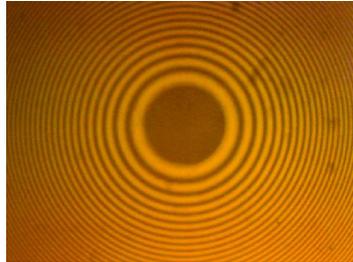
$$R(q) = \frac{16\pi^2}{q^4} |N'_b(q)|^2$$

$$N_b = \frac{\sum_i n_i b_i}{V}$$

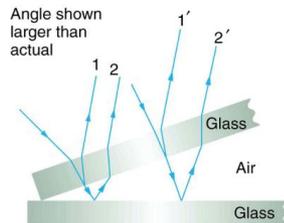
Interference of reflected beams

NEWTON'S RINGS

monochromatic beam

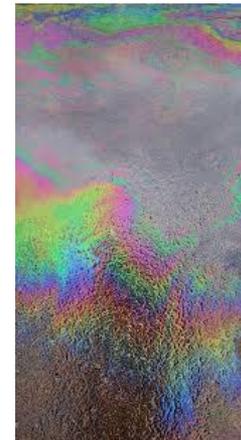


(a)



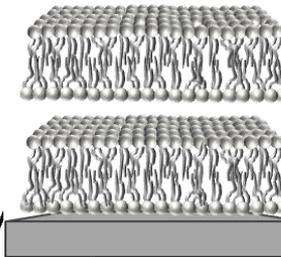
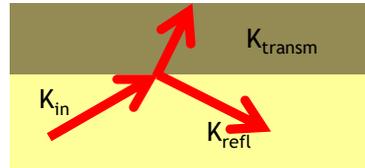
(b)

Light striking a thin film is partially reflected (ray 1) and partially refracted at the top surface. The refracted ray is partially reflected at the bottom surface and emerges as ray 2. These rays will interfere in a way that depends on the thickness of the film and the indices of refraction of the various media.

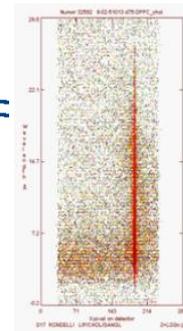
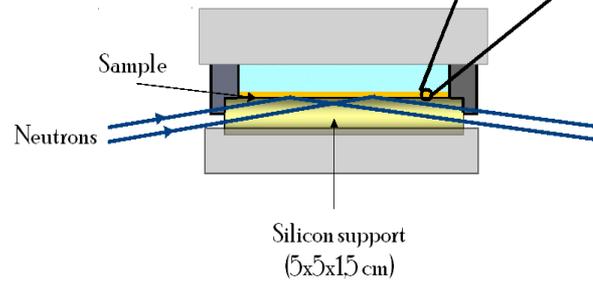


When we have more than one interface: A Neutron Reflectivity experiment

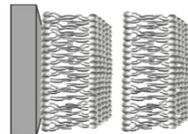
For the incident beam (Neutrons, X-Rays) is as if the sample is divided in layers, depending on their composition (water, polar heads, chains)



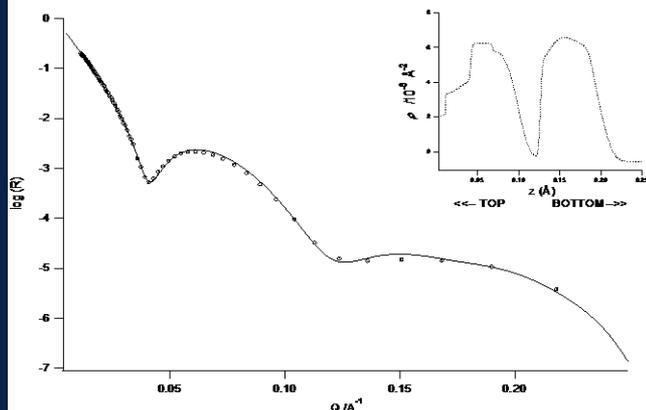
- M water
- L. heads
- l. chains
- H. chains
- G. heads
- F. water
- E. heads
- D. chains
- C. chains
- B. heads
- A. water



The Reflectivity spectrum obtained is given by the **interference** of the waves **reflected** from the top and bottom of each layer



SAMPLE CONTRAST PROFILE



Information about the transverse structure of the sample, layer by layer:
thickness, composition, compactness, roughness

LIPID «RAFTS» - GEMs

functional and structural domains

recognition

cell adhesion

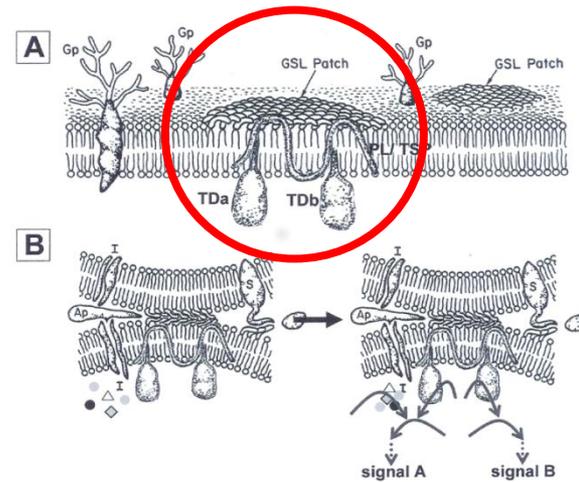
transfer of information

cell growth

proliferation

apoptosis

...



Hakomori (1998)

- lipid driven

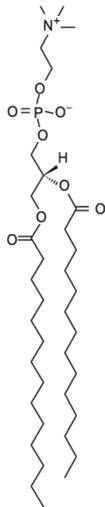
- glycolipids only on the outer side of the membrane, cholesterol mainly in the inner

Valid model for lipid rafts - GEMs

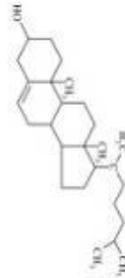
→ lipid composition

→ structural asymmetry

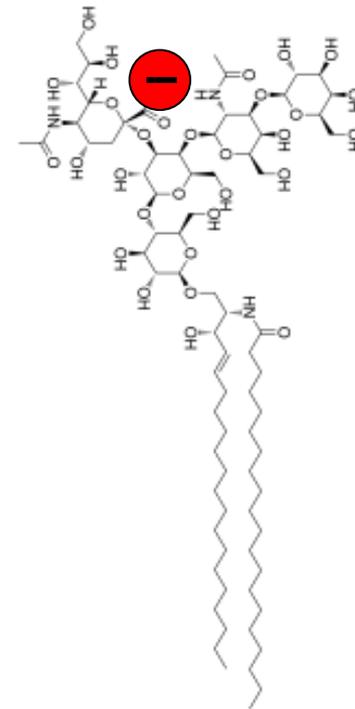
Glycerophospholipid



Cholesterol

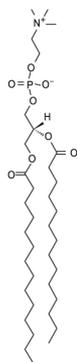


Glycosphingolipid



The contrast

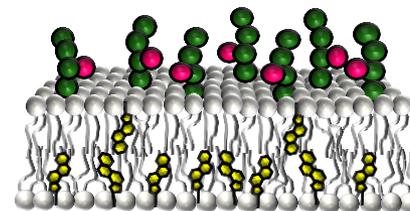
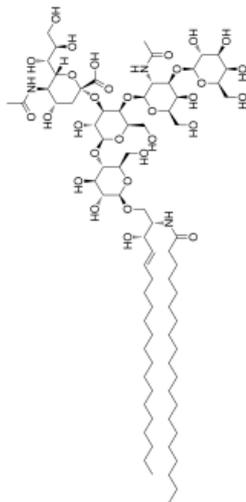
DPPC



Cholesterol

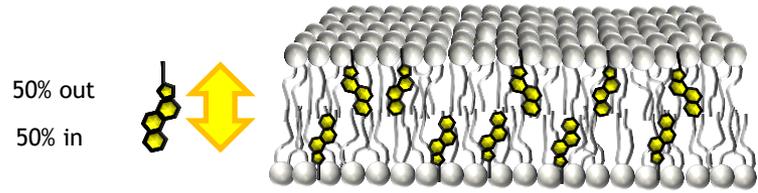


GM1



Material	SLD (10^{-6}Å^{-2}) ^a
H ₂ O	-0.56
D ₂ O	6.36
Cholesterol	0.22
GM1 chains (gel phase)	-0.41
GM1 heads	1.88
Lipid D-heads	5.70
Lipid D-chains (gel phase)	7.66
Lipid D-chains (fluid phase)	6.13

Is asymmetry possible? ...conditions to be kept?



V. Rondelli et al., J. Phys. (2012)

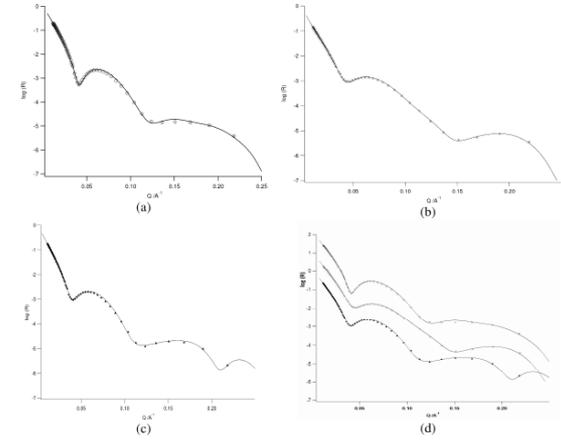


Figure 3. Reflectivity spectra of the cholesterol-containing sample in H₂O (a) at 22°C before annealing, (b) at 51°C, (c) at 22°C after annealing.
In insert (d) the curves have been arbitrarily shifted by 5 (51°C) and 10 (22 °C) for better clarity in comparing them: upper curve: 22°C before annealing. Middle curve: 51°C. Lower curve: 22°C after annealing.

NON-INVASIVE INSIGHT IN THE INTERNAL STRUCTURAL MECHANISMS

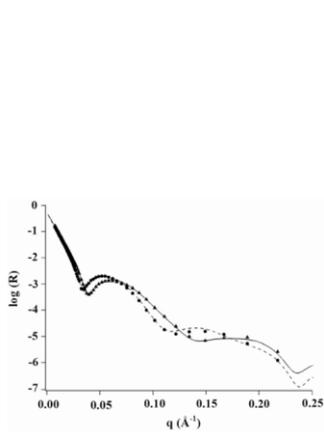


Fig. 3. Modification of the neutron reflectivity spectrum induced by the incubation of GM1 (before: triangles, after: circles) in a d₅₅-DPPC + cholesterol bilayer. Straight and dashed lines represent the curves obtained from the parameters used to fit the data (see Tables 2 and 3). The solvent is H₂O and T=49.5 °C. Error bars are smaller than graphical symbols.

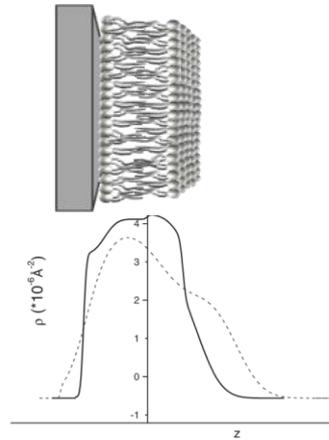
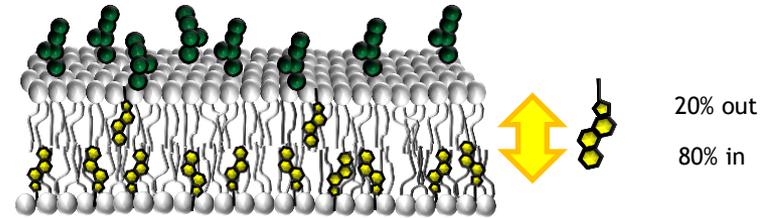
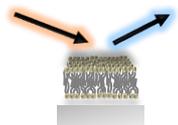
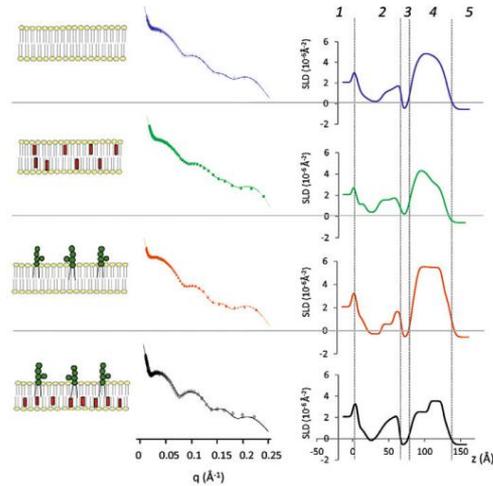


Fig. 4. Contrast profiles, $\rho(z)$, of the floating bilayer of Sample A at 49.5 °C in H₂O. Full line: before GM1 incubation. Dashed line: after GM1 incubation. In the figure the vertical axis is placed roughly at the center of the floating bilayer, to guide the eye. The contrast profile account for lipid components, included solvent and roughness.

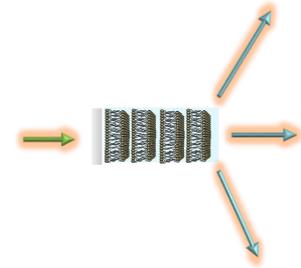
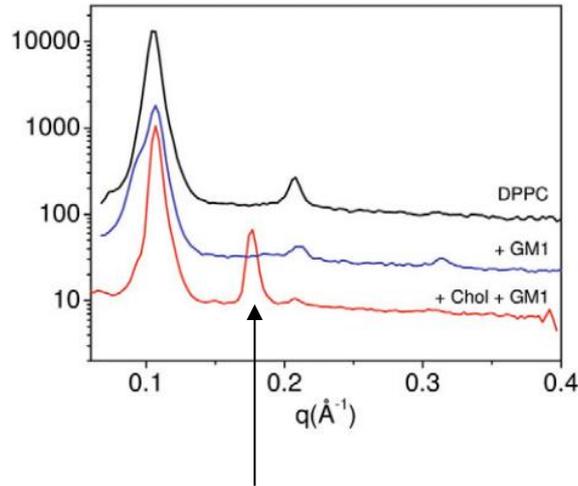


V. Rondelli et al., BBA - Biomembranes (2012)

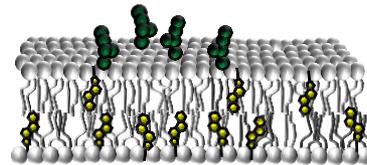
What about lateral distribution?



NR

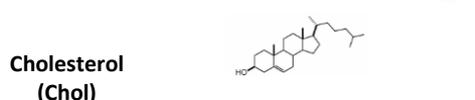
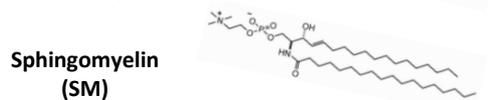
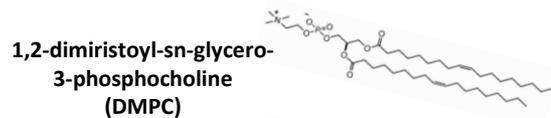


Neutron Diffraction

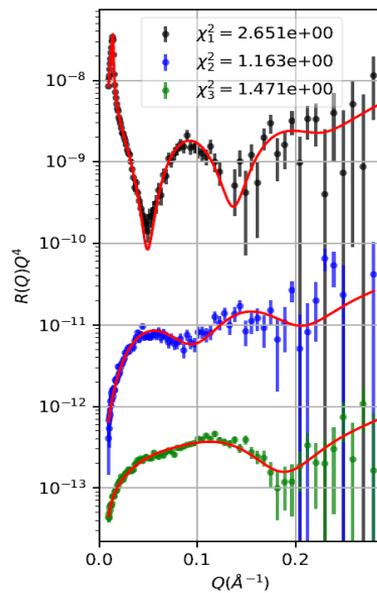
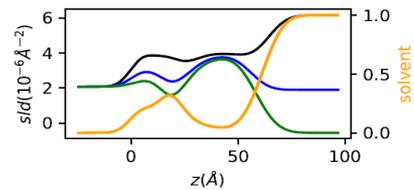
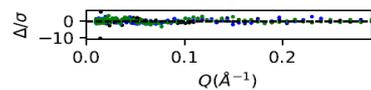
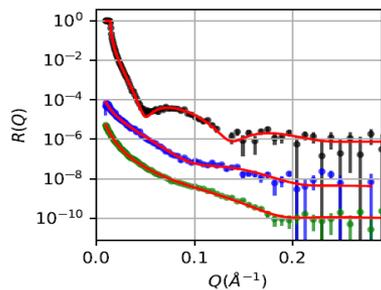


V. Rondelli et al., *Eur. Phys. J. E* (2013)

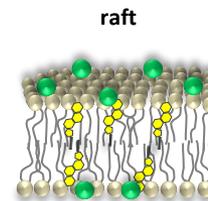
Sugar-free raft model



SM and Chol co-localize symmetrically

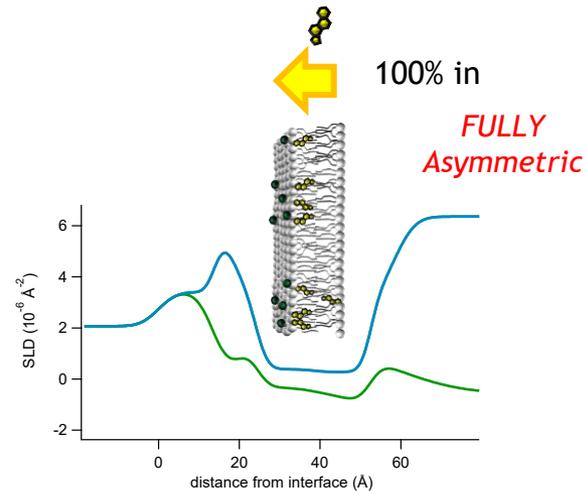
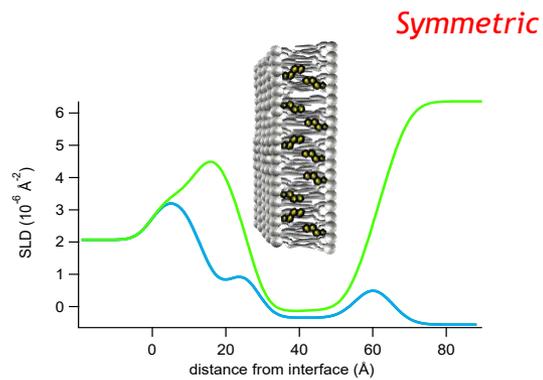
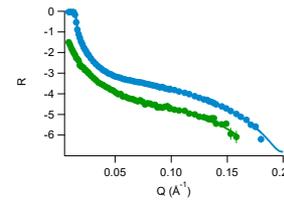
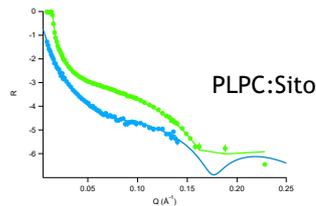
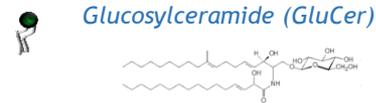
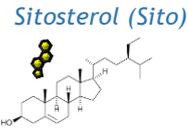
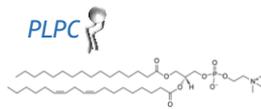


PC:SM:Chol 2:1:0.15 mol



S. Abdalla et al., JGIS (2025)

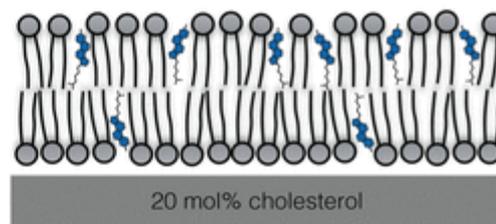
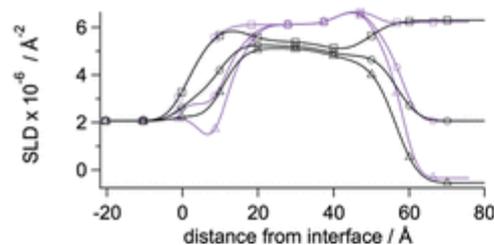
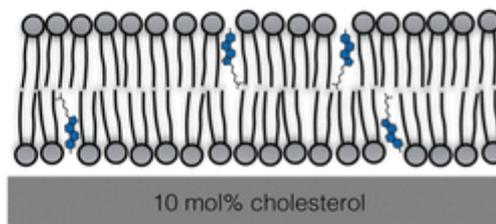
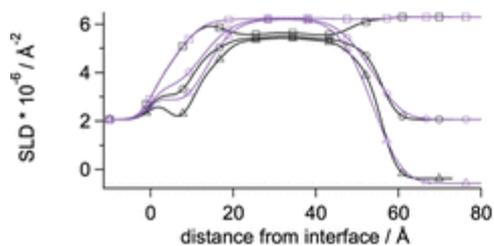
What about plant membranes?



→ Uneven sitosterol distribution in leaflets ONLY in presence of Glucosylceramide

Deuterated lipid mixtures from Escherichia Coli

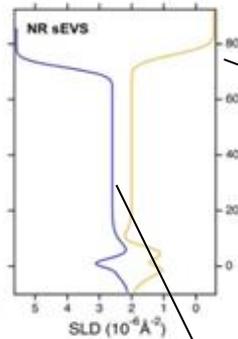
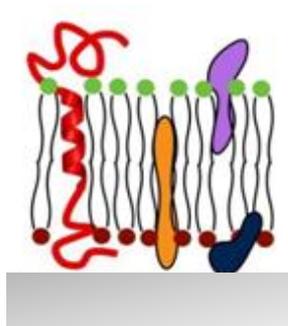
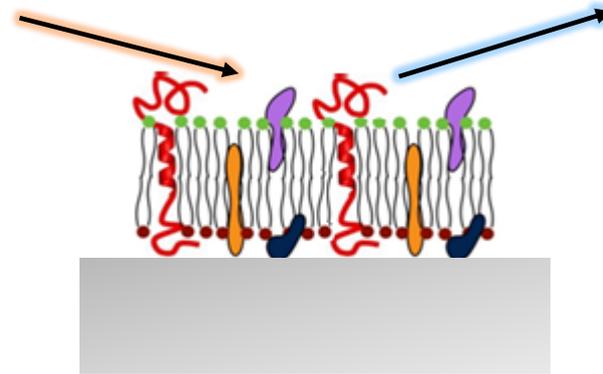
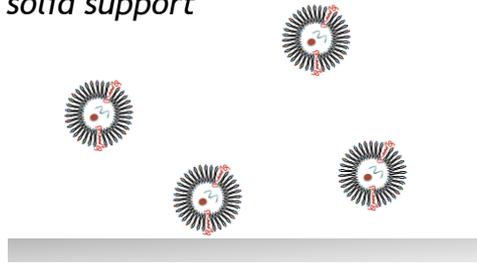
	SLD * 10 ⁻⁶ / Å ⁻²	Volume / Å ³
Si	2.07	
SiO ₂	3.47	
dPC tail	6.1	905 ¹
dPC head	6.1	322 ²
h-cholesterol	0.2	622 ³
d-cholesterol	7.2	622 ³



Cholesterol is located closer to the lipid head group-tail interface in natural PC extract rather than in the center of the core of the bilayer as seen for very thin or polyunsaturated membranes.

Neutron reflectometry on EVs-derived supported bilayers

Extracellular Vesicles fusion on solid support



Thickness:
 $6.9 \pm 0.2 \text{ nm}$

single bilayer
containing molecules
other than lipids, as
large proteins

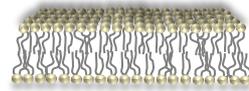
SLD $2 \pm 0.2 \times 10^{-6} \text{ \AA}^{-2}$

lipid : protein
22 : 78
(by volume)

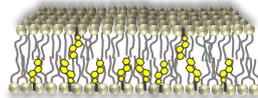
$$R \approx \left(\frac{16\pi^2}{q^4} N_b^2 \right) e^{-q_z^2 \sigma^2}$$

Membrane models

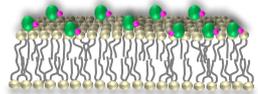
phospholipids



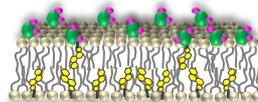
+
cholesterol



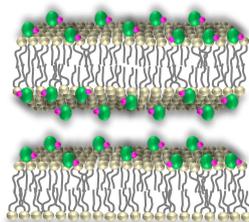
+
glycolipids



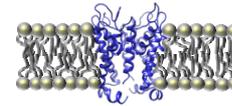
raft models



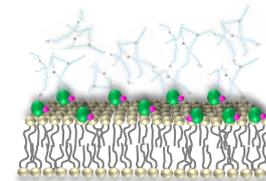
*glycolipid-containing
facing double-membranes*



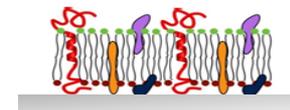
+
transmembrane protein (K⁺ channel)



+
gel forming proteins (mucin)



from native extracts

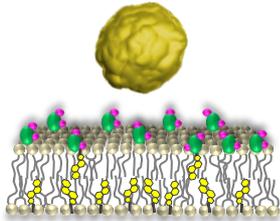


IT COULD WORK

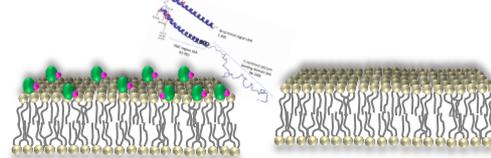


We can model interactions at membrane surface

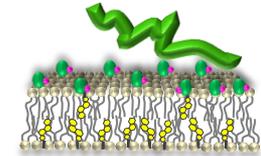
Enzymatic digestion on membrane surface



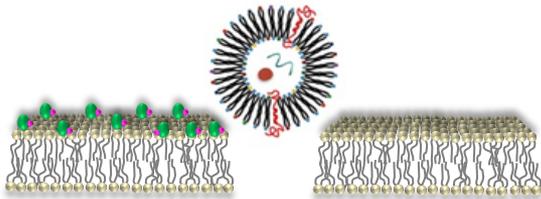
α -synuclein interaction



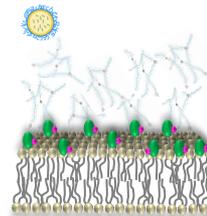
A β Peptide interaction



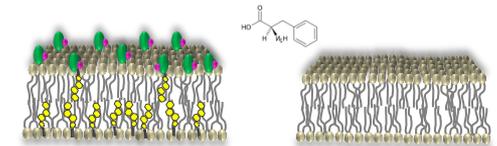
Extracellular vesicles interaction



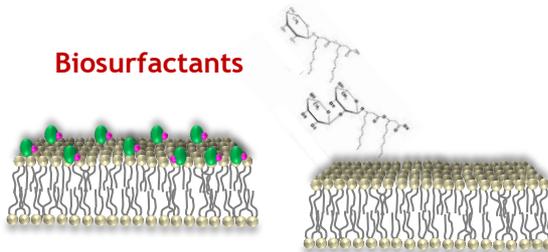
Transmucosal delivery



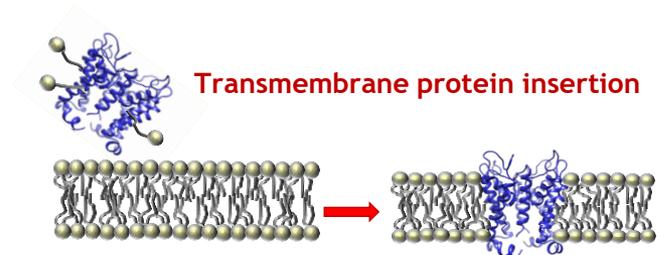
Amino acid interaction



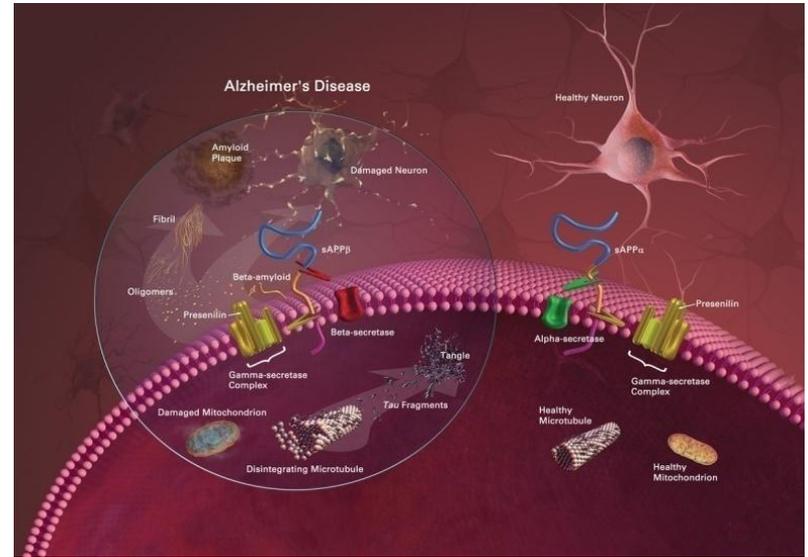
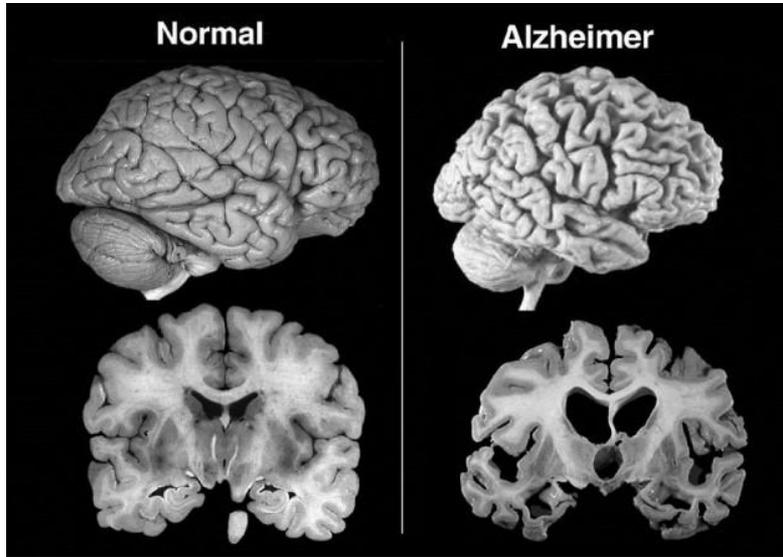
Biosurfactants



Transmembrane protein insertion

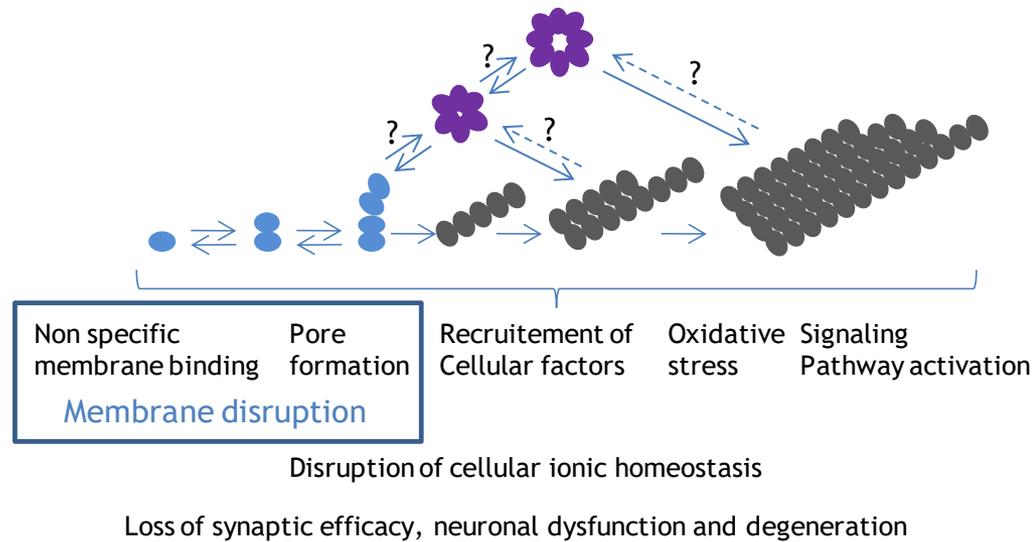


A β evidence and hypothesis

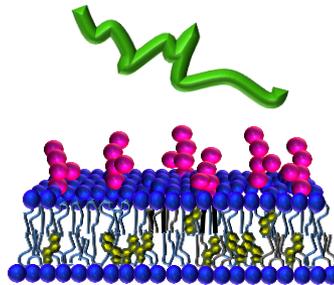


Complex landscape

Simplified experimental models



Membrane-active species

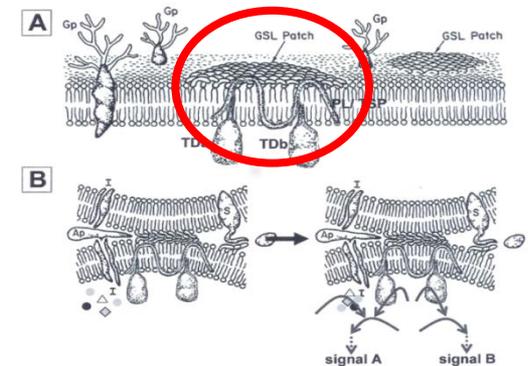


Structural modifications induced by membrane-peptide interaction
integrated approach with complementary techniques

→ complementary information on different properties

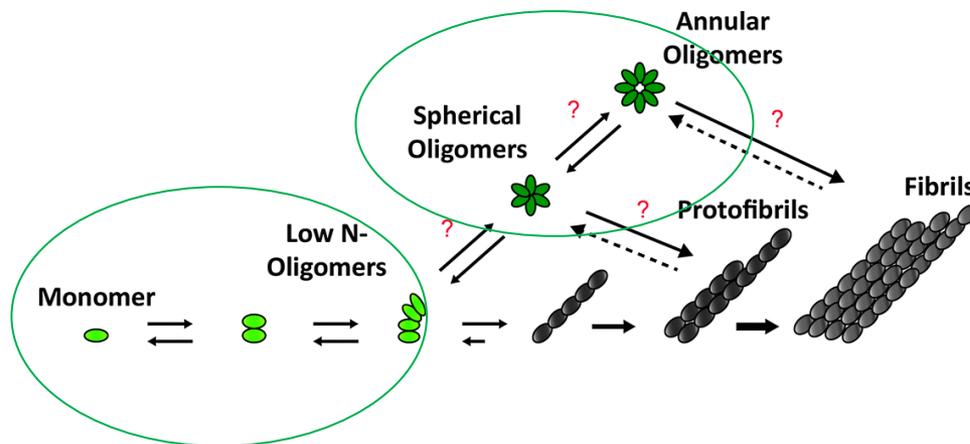
GEMs are claimed to be involved in
“membrane-mediated” amyloidogenesis

(K. Matsuzaki)

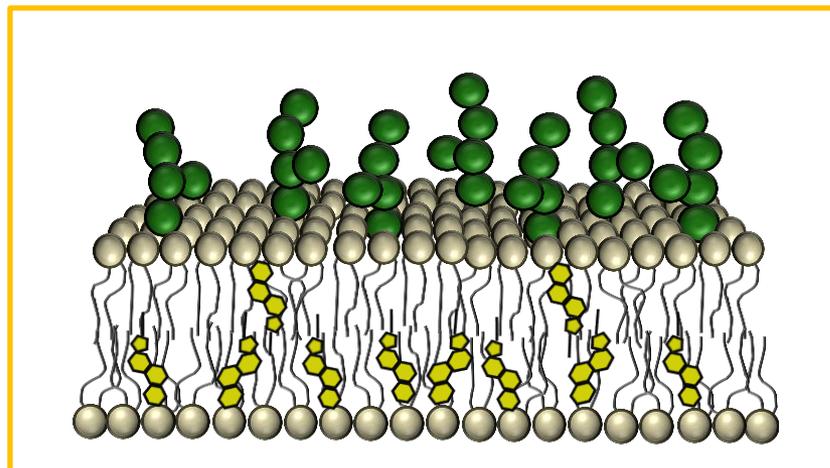


Hakomori (1998)

A β ₁₋₄₂ interaction with a single raft mime



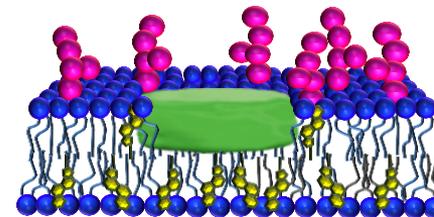
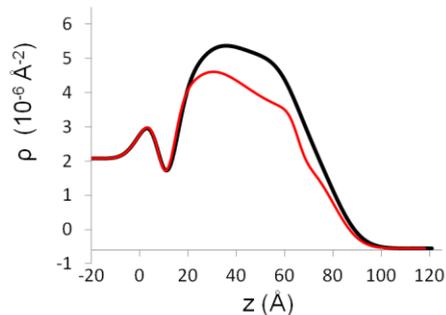
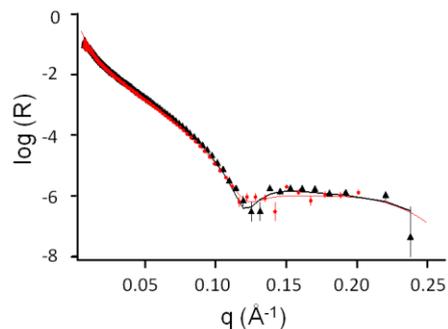
dDSPC:chol:GM1
10:1.25:0.5 mol



Protiated peptide in deuterated membrane

RAFT + structured oligomers

50 mM
Phosphate buffer
22° C
5h incubation



Embedded as such in the external leaflet of the membrane

...seed?

Naked Membrane A

Membrane A after interaction with Aβ1-42 structured oligomers

Aβ1-42 penetration hypothesis

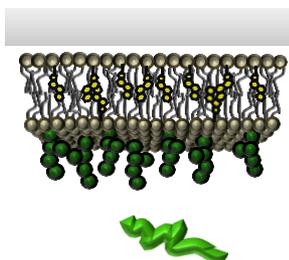
lipid loss hypothesis

T	r	ρ _{lip} (z)	W
6	5	4.87	30
21	8	7.01	18
23	9	6.89	25
16	3	4.21	31

T	r
6	5
21	8
23	15
17	4

ρ _{lip} (z)	W	Aβ penetration (%vol)
4.87	30	
6.99	18	
5.9	25	19%
3.2	31	37%

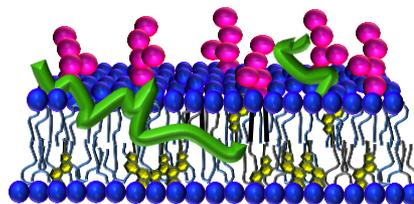
ρ _{lip} (z)	W	H ₂ O penetration (%vol)
4.87	30	
7.01	18	
6.89	39	14%
4.21	52	21%



hydrophilic in (3)
hydrophobic in (4)
hydrophobic out (5)
hydrophilic out (6)

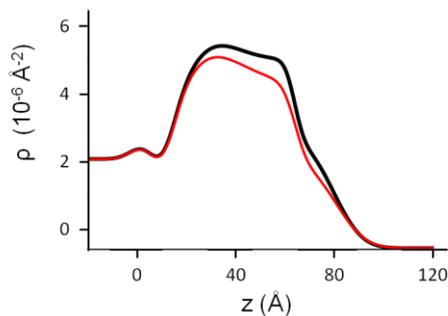
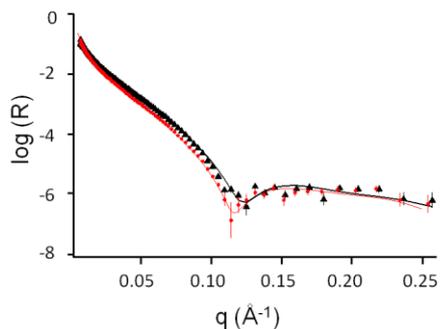
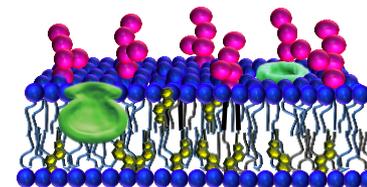
RAFT + early oligomers

DEPSIPEPTIDE
1 μM



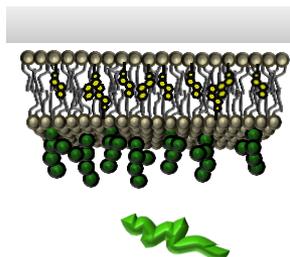
Aggregation templated
by the membrane

...pore formation?



Naked Membrane B

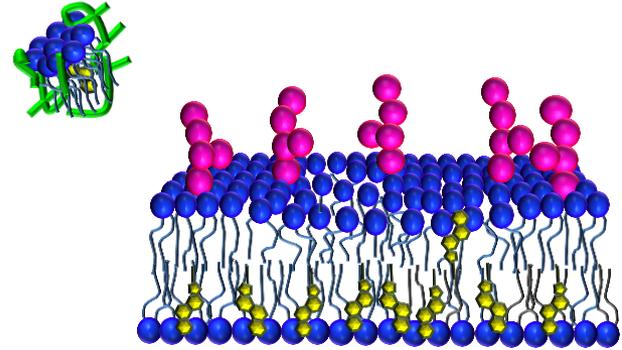
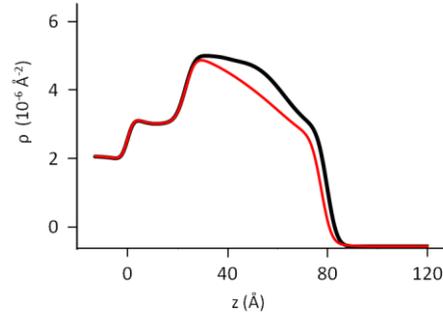
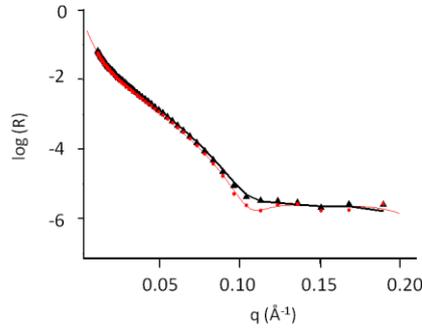
Membrane B after interaction with Aβ1-42 early oligomers



hydrophilic in (3)
hydrophobic in (4)
hydrophobic out (5)
hydrophilic out (6)

				Aβ1-42 penetration hypothesis					lipid loss hypothesis		
T	r	$\rho_{lip}(z)$	W	T	r	$\rho_{lip}(z)$	W	Aβ penetration (%vol)	$\rho_{lip}(z)$	W	H ₂ O penetration (%vol)
6	4	4.87	30	6	4	4.87	30	6% 42% 32%	4.87	30	
21	9	7.01	18	21	12	6.7	18		7.01	22	4%
23	9	6.89	25	25	15	4.6	25		6.89	55	30%
16	7	4.21	31	14	3	3.3	31		4.21	49	18%

RAFT + N-terminus

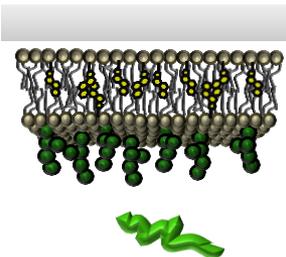


Role in membrane destabilization?

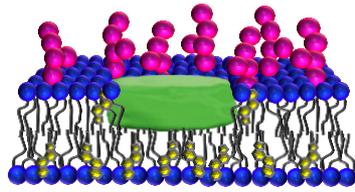
Naked Membrane C

Membrane C after interaction with Aβ1-6

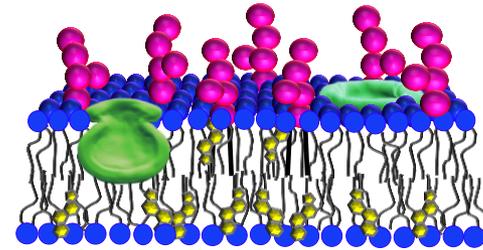
				Aβ1-6 penetration hypothesis			lipid loss hypothesis					
T	r	$\rho_{lip}(z)$	W	T	r	$\rho_{lip}(z)$	W	Aβ penetration (%vol)	$\rho_{lip}(z)$	W	H ₂ O penetration (%vol)	
hydrophilic in (3)	9	8	4.87	34	9	8	4.87	34		4.87	34	
hydrophobic in (4)	17	3	7.01	27	17	3	6.99	27		6.99	27	
hydrophobic out (5)	23	3	6.89	27	22	10	5.9	27	20%	6.89	40	13%
hydrophilic out (6)	17	9	4.21	30	15	12	3.6	30	26%	4.21	43	13%



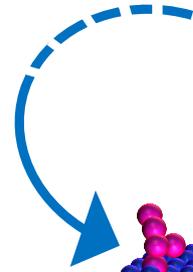
Model raft + A β ₁₋₄₂



Structured oligomers

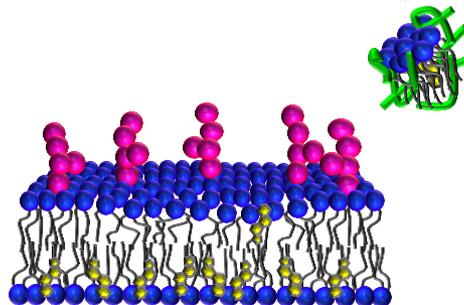


Early oligomers



Therapeutic hint?

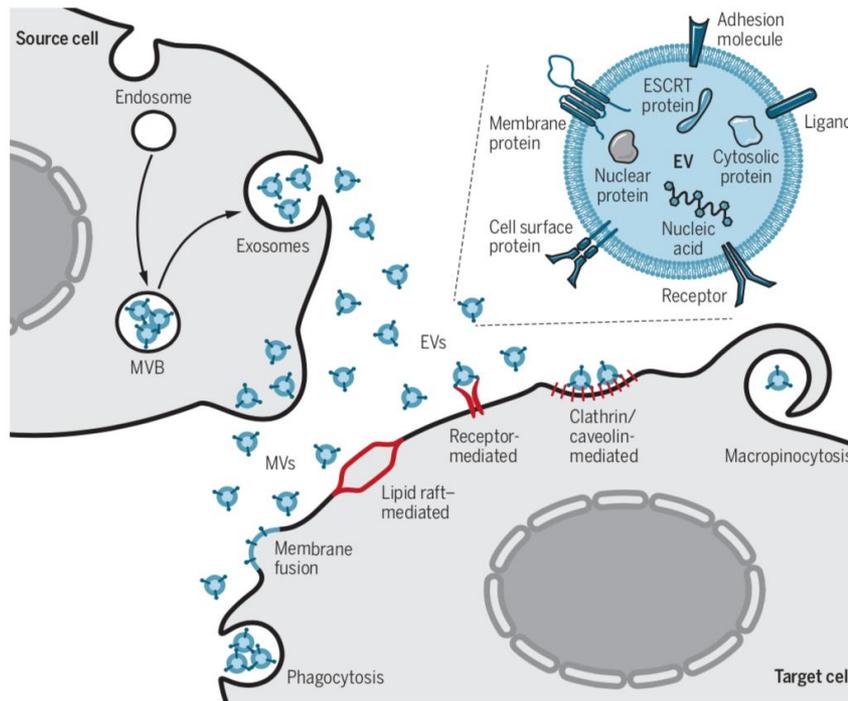
Model raft + A β ₁₋₆



Back to extracellular vesicles - EVs

Definition

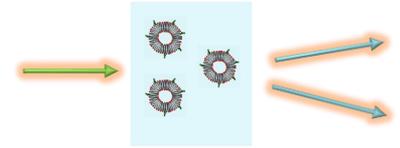
« Particles that are released from the cells, delimited by a lipid bilayer and cannot replicate on their own »



Wiklander et al., Science Trans. Medicine (2019)

Back to extracellular vesicles - EVs

SANS

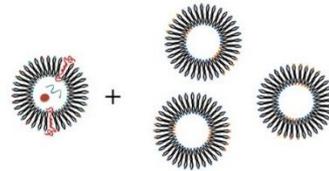


sEVs : dDMPC Vs

1: 15000

1:3000

1:2700

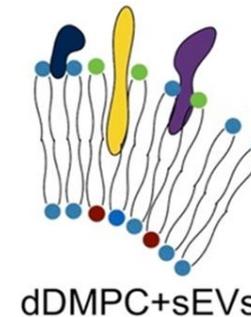
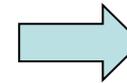
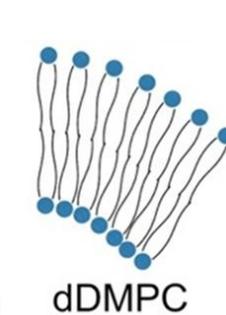
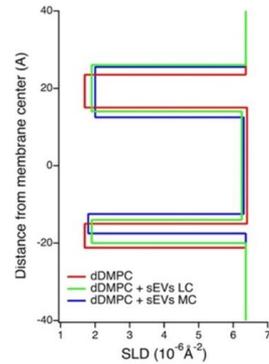
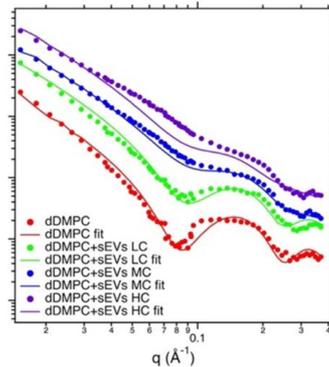


FITTING MODEL
Multilayered spherical form factor

$$P(q) = \frac{\text{scale}}{V} F^2(q) + \text{background}$$

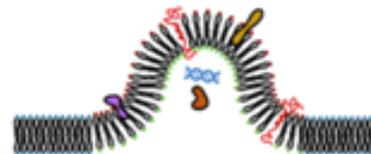
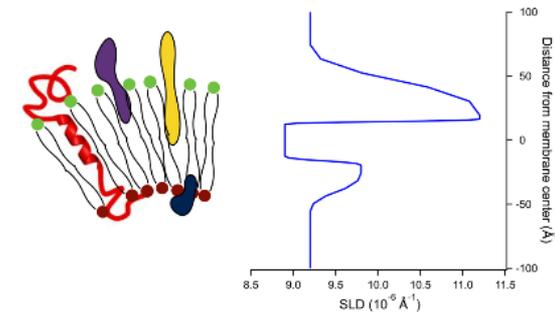
$$F(q) = \frac{3}{V_s} \left[V_c(\rho_c - \rho_s) \frac{\sin(qr_c) - qr_c \cos(qr_c)}{(qr_c)^3} + V_s(\rho_s - \rho_{\text{solv}}) \frac{\sin(qr_s) - qr_s \cos(qr_s)}{(qr_s)^3} \right]$$

(Guinier, 1955)



- Change in contrast spans whole membrane thickness
- Asymmetric

EVs SLD profile



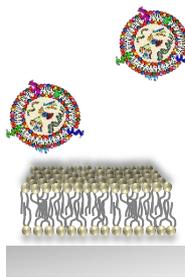
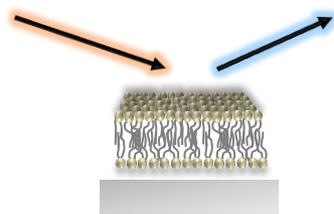
EVs uptake mechanisms - Neutron Reflectometry

DMPC vesicles fusion

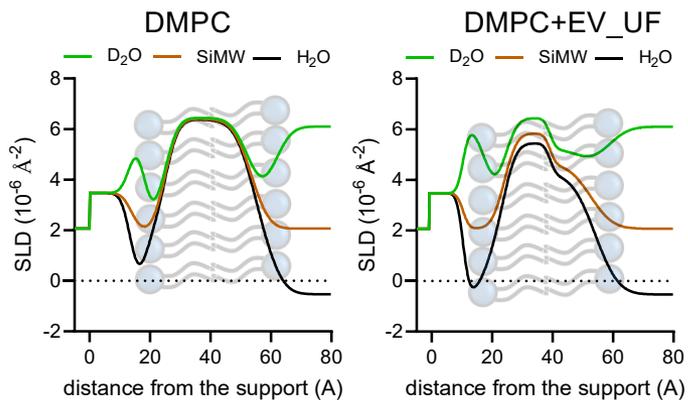
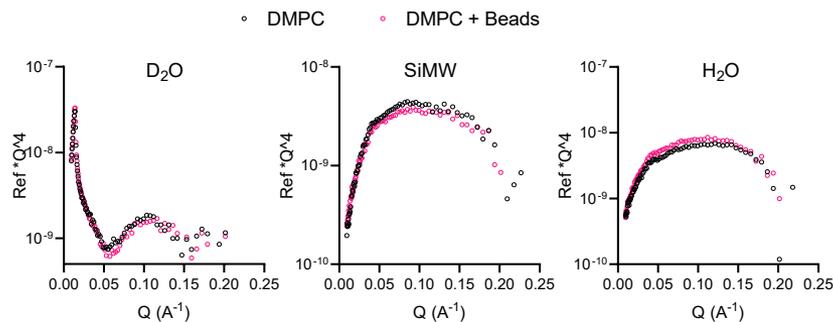
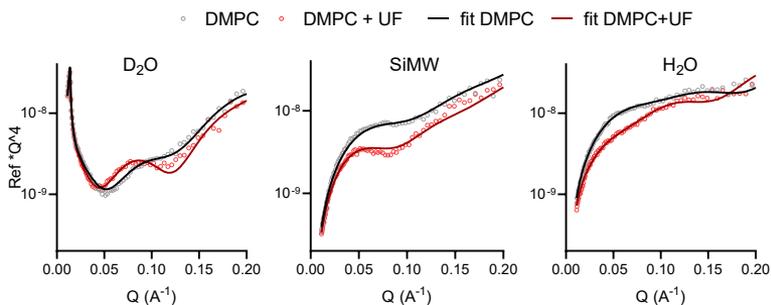
DMPC characterization

EVs injection from bulk water

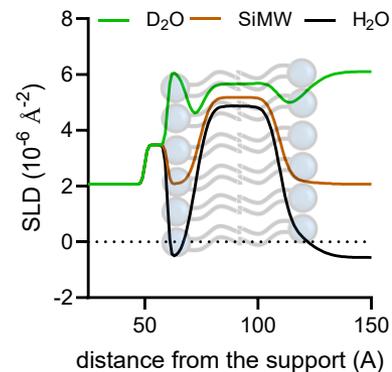
mixed system characterization



Material	SLD (10^{-6} \AA^{-2})
H ₂ O	-0.56
D ₂ O	6.36
Lipid chains	-0.33
Protein	2 - 2.5

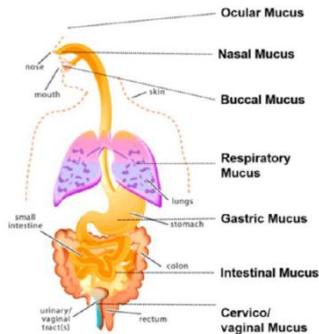


DMPC+EV ultrapure



The mucus selective barrier

MUCUS: highly viscoelastic secretion, covering the epithelia surfaces of the gastrointestinal, pulmonary, oral, nasal and genital tracts.

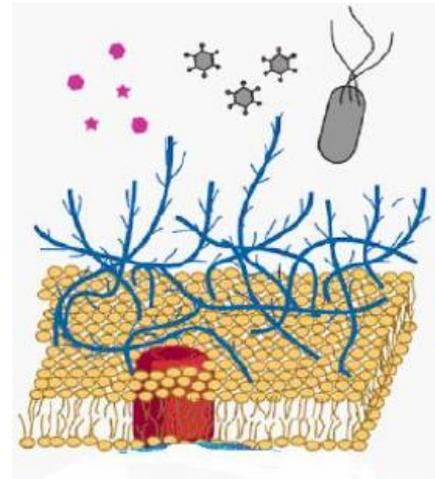


Tasks:

- to protect mucosal tissues from dehydration, mechanical stress
- to act as **SELECTIVE BARRIER** against microorganisms and toxic substances regulating the distribution of drugs, ions and proteins

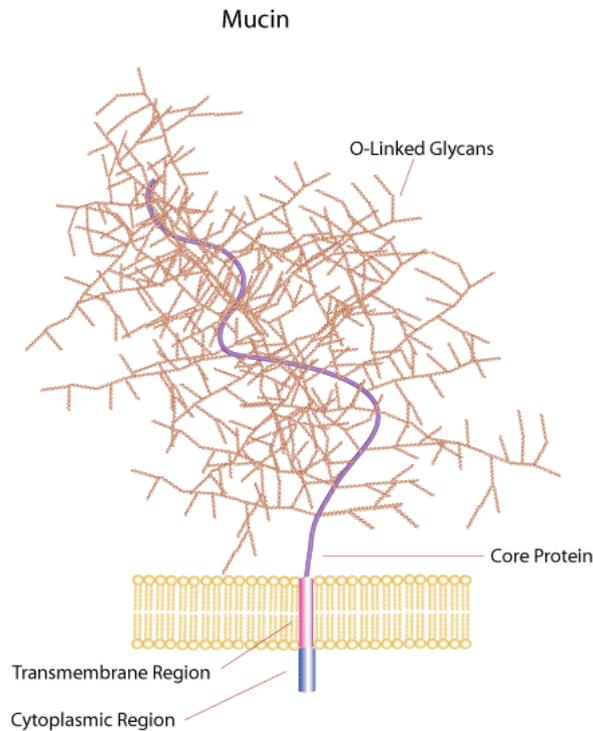
Structure:

- water (up to 98%)
- lipids
- small proteins
- nucleic acids
- high molecular weight glycoproteins: **MUCINS**



Mucins: a mucus model environment

MUCIN:
amino-acidic backbone
+
regular distribution of sugar brushes

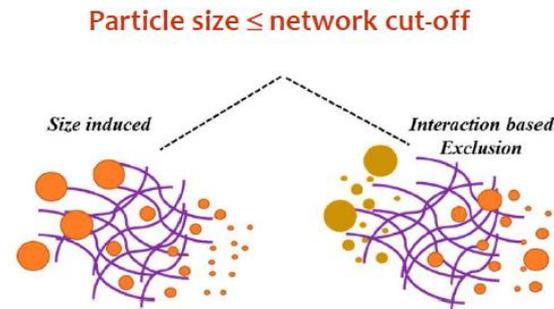


Mechanical and viscoelastic properties of mucus depend on mucin

Delivery in mucus environments

Why transmucosal delivery?

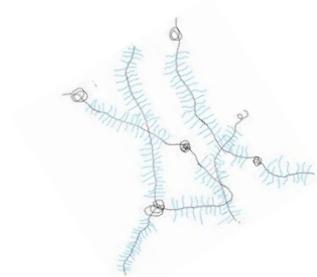
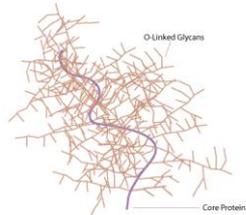
The transmucosal membranes are relatively permeable, have a rich blood flow and hence allow the rapid uptake of a drug into systemic circulation to avoid first pass metabolism



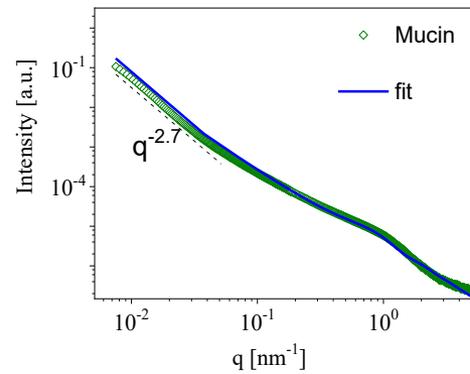
O. Lieleg, K. Ribbeck in *Trends Cell. Biol.* 21(9), 2011

- penetrating particle
- adjuvants for mucus destabilization

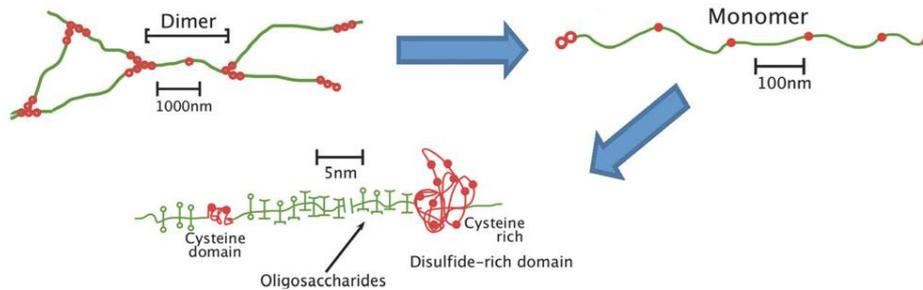
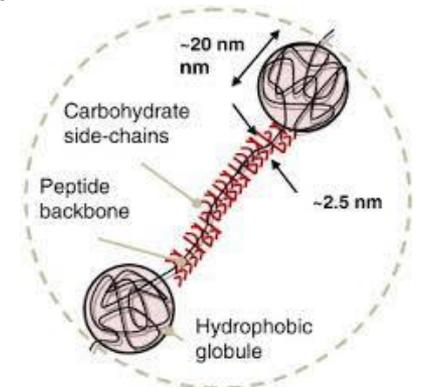
Mucines: a mucus model environment



SAXS from bulk systems

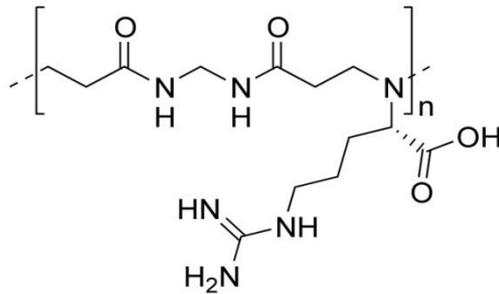


Mucin
dumbbell-like shape
form factor



D,L-ARGO7 polymer

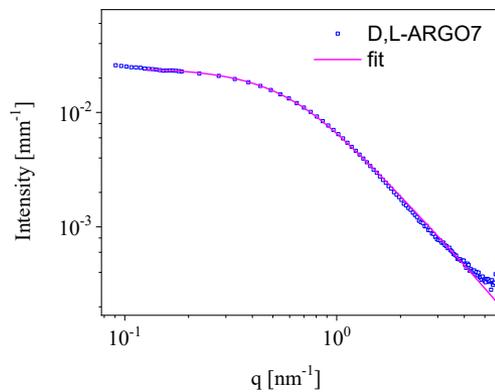
D,L-ARGO7



- Water soluble polyamidoamine
- Amphoteric
- good **biocompatibility** and **biodegradability**
- promising for DNA and siRNA delivery.
- strong **protection** actions against virus infection
- **low cytotoxicity**

MW= 10 kDa

SAXS from bulk systems

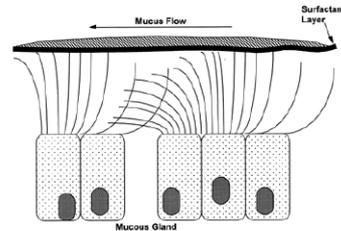


Gaussian polymer
model, radius of
gyration 2.5 nm

Is it enough?

In bulk we observe the mean of systems behaviour and phases

CLEARENCE



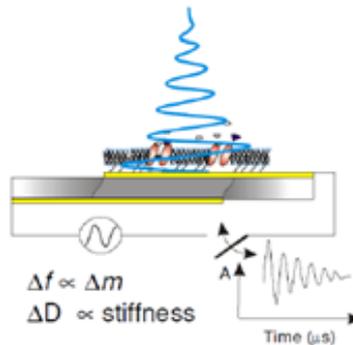
We care about destabilization, phase separation and changes in mobility



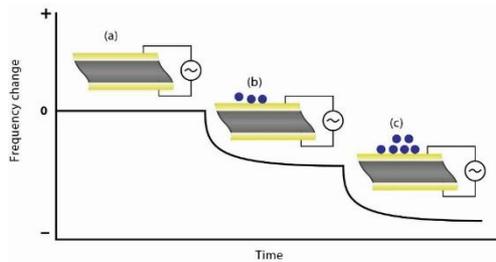
Thin films

QCM-D from thin layers

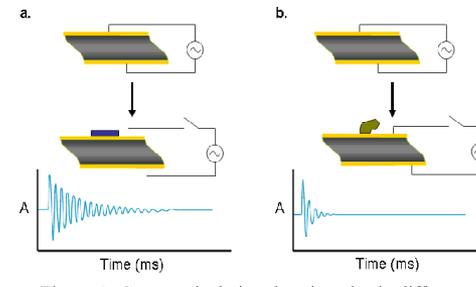
Excitation resonance of a freely oscillating quartz crystal



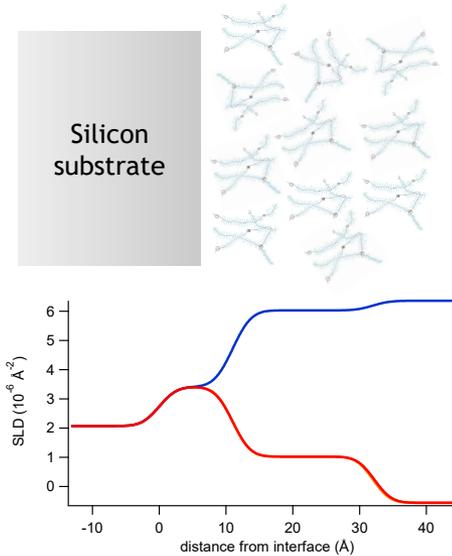
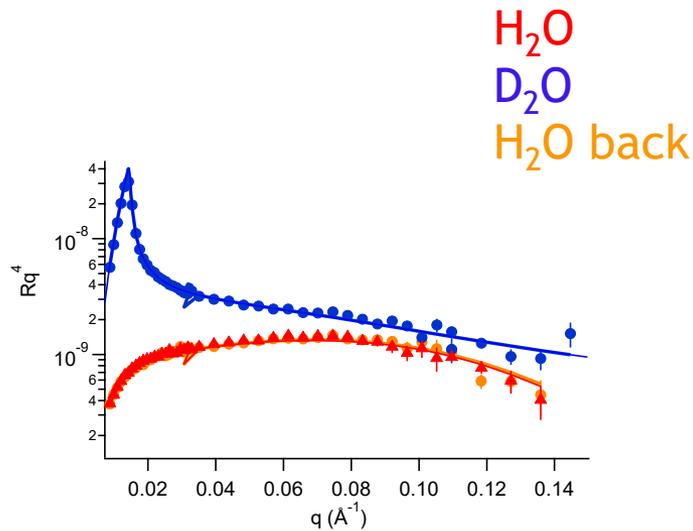
Resonance frequency:
information about the mass (ng to μg) deposited



Dissipation:
information about the viscoelastic properties of the deposited sample



Neutron reflectometry

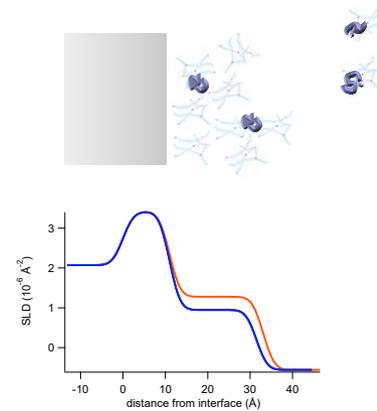
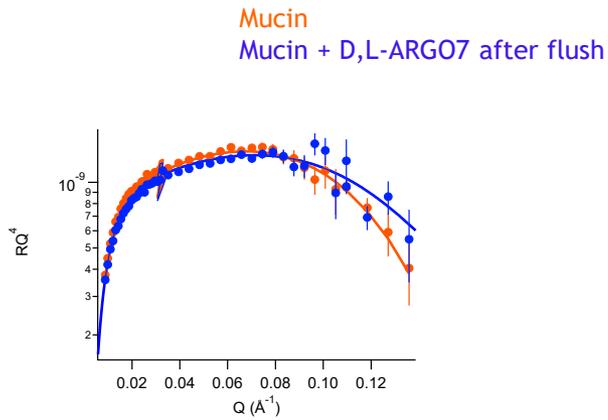
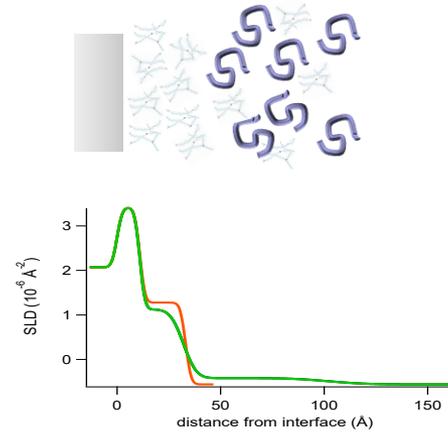
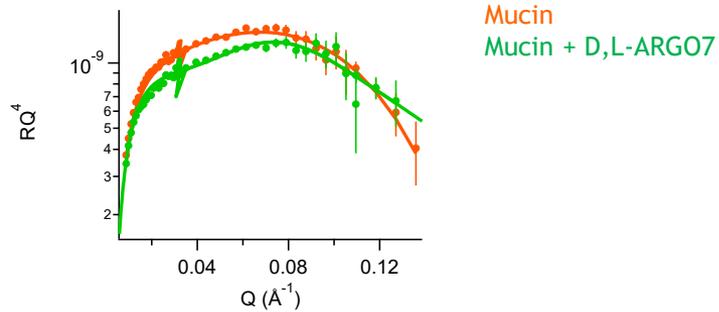


20Å mucin layer
Stable under solvent flushing

Table S1. Fit parameters of the mucin layer deposited on the silicon surface. Parameters correspond to a contemporary fit performed on H₂O and D₂O solutions with 150mM NaCl. Errors have been estimated by changing the parameters up to a variation of two in the χ^2 .

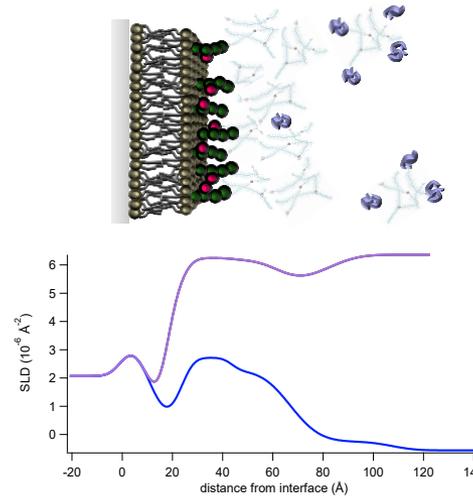
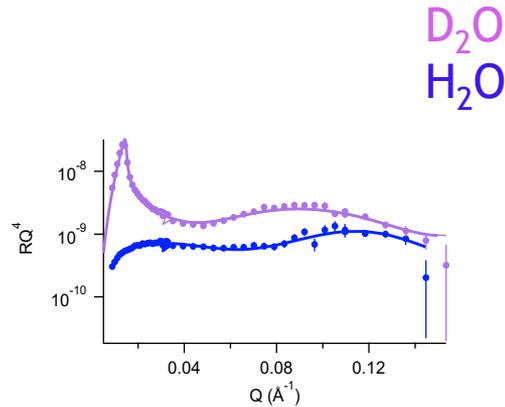
	Thickness ($\pm 1 \text{ \AA}$)	SLD ($\pm 0.05 \cdot 10^{-6} \text{ \AA}^{-2}$)	Solv p ($\pm 5\% \text{ vol}$)	Roughness ($\pm 2 \text{ \AA}$)
Mucin	21	2.5 (H ₂ O) 5.6 (D ₂ O)	40	3

Mucin + D,L-ARGO7: neutron reflectometry



Mucin destabilization and partial removal by flushing

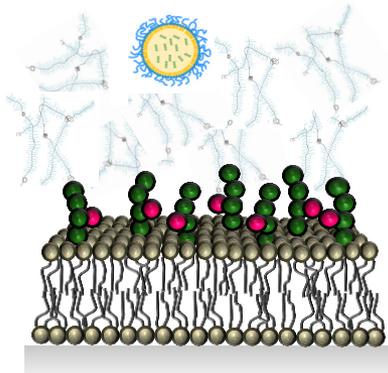
A model for epithelial mucus covered cells



Rondelli V. *et al.*
International Journal of Molecular Sciences (2019)

➡ *Mucin destabilization and partial removal by flushing*

➡ *Complex system stability*



Access not only to the fate of particle and mucus but also to the cross details of model tissue-biomolecules interaction with few Å sensitivity

CONCLUSIONS

Neutron scattering/reflectometry are essential complementary techniques for cross-structural investigations down to the nanoscale of self assembled systems at interfaces

Selective deuteration allows for selective visibility of components

Possibility to work in physiological conditions

Possibility for in-situ interaction studies

Perspectives in biology are numerous

Literature

- R. Pinn '*Neutron Scattering- A Primer*', Lansce, 1990
- G.L. Squires '*Introduction to Thermal Neutron Scattering*', Cambridge university press, 1978
- A. J. Jackson, NIST Center for Neutron Research, '*Introduction to Small-Angle Neutron Scattering and Neutron Reflectometry*', 2008
- <http://www.xtal.iqfr.csic.es/Cristalografia/index-en.html>
- W. Stillwell, '*An Introduction to Biological Membranes - Composition, Structure and Function*', Academic Press 2016