

Riepilogo equazioni

I PTD: $\Delta U = Q - L$

$dU = \delta Q - \delta L$ con $\delta L = p dV$

$$\delta Q = \begin{cases} \left(\frac{\partial U}{\partial p}\right)_V dp + \left[\left(\frac{\partial U}{\partial V}\right)_p + p\right] dV & U = U(p, V) \\ \left[\left(\frac{\partial U}{\partial p}\right)_T + p\left(\frac{\partial V}{\partial p}\right)_T\right] dp + \left[\left(\frac{\partial U}{\partial T}\right)_p + p\left(\frac{\partial V}{\partial T}\right)_p\right] dT & U = U(p, T) \\ \left[\left(\frac{\partial U}{\partial V}\right)_T + p\right] dV + \left(\frac{\partial U}{\partial T}\right)_V dT & U = U(V, T) \end{cases}$$

(per sistemi idrostatici, vedi pag 3 sez. 2.1.2)

c_p e c_v :

$c_v = \frac{1}{n} \left(\frac{\delta Q}{dT}\right)_V \stackrel{\text{S.I.}}{=} \frac{1}{n} \left(\frac{\partial U}{\partial T}\right)_V \xrightarrow{\text{G.I.}} \frac{1}{n} \left(\frac{dU}{dT}\right) \rightarrow dU = n c_v dT$

$c_p = \frac{1}{n} \left(\frac{\delta Q}{dT}\right)_p \stackrel{\text{S.I.}}{=} \frac{1}{n} \left[\left(\frac{\partial U}{\partial T}\right)_p + p\left(\frac{\partial V}{\partial T}\right)_p\right] = \frac{1}{n} \left(\frac{\partial H}{\partial T}\right)_p \xrightarrow{\text{G.I.}} \frac{1}{n} \left(\frac{dH}{dT}\right) \rightarrow dH = n c_p dT$

Da cui, sempre per i G.I.:

$$\delta Q = \begin{cases} n c_v dT + p dV = dU + p dV & \xrightarrow{V \text{ cost}} \delta Q = dU \\ n c_p dT - V dp = dH - V dp & \xrightarrow{p \text{ cost}} \delta Q = dH \end{cases}$$

Equazioni di Clapeyron (per sistemi idrostatici):

$$\delta Q = \begin{cases} n c_p \left(\frac{\partial T}{\partial V}\right)_p dV + n c_v \left(\frac{\partial T}{\partial p}\right)_V dp & U = U(p, V) \\ n c_p dT - T \left(\frac{\partial V}{\partial T}\right)_p dp & U = U(p, T) \\ n c_v dT + T \left(\frac{\partial p}{\partial T}\right)_V dV & U = U(V, T) \end{cases}$$

Veniamo dimostrate più avanti. Si possono facilmente verificare per i gas ideali. Ad esempio la prima (seconda e terza sono banali):

* $T = \frac{pV}{nR}$; $\left(\frac{\partial T}{\partial V}\right)_p = \frac{p}{nR}$; $\left(\frac{\partial T}{\partial p}\right)_V = \frac{V}{nR}$

$$\begin{aligned} \delta Q &= n c_p \frac{p dV}{nR} + n c_v \frac{V dp}{nR} && \text{ma } V dp = nR dT - p dV \\ &= n c_p \frac{p dV}{nR} + \frac{n c_v}{nR} (nR dT - p dV) \\ &= \frac{c_p - c_v}{R} p dV + n c_v dT = n c_v dT + p dV \quad \square \end{aligned}$$