



**Università di Trieste**  
**Corso di Laurea in Geologia**

**Anno accademico 2015 - 2016**

**Geologia Marina**

Parte VI

**Modulo 6.3** Confinamento geologico della CO<sub>2</sub>

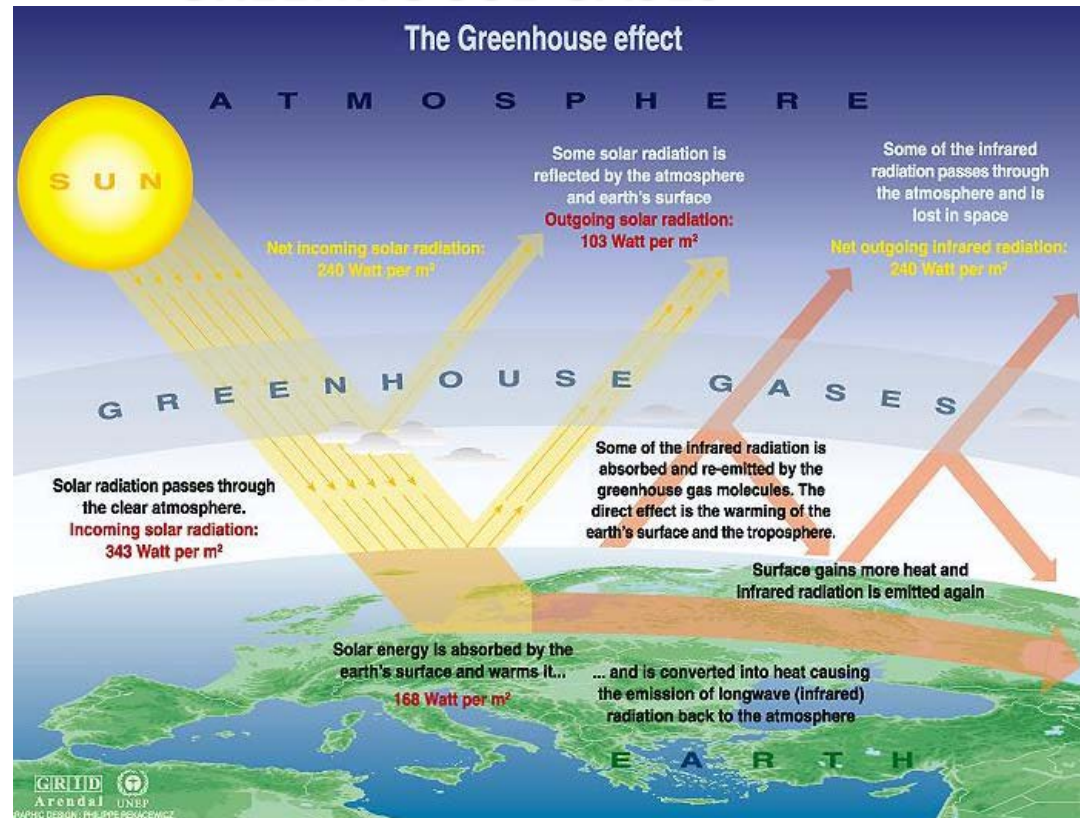
Docente

**Valentina Volpi**

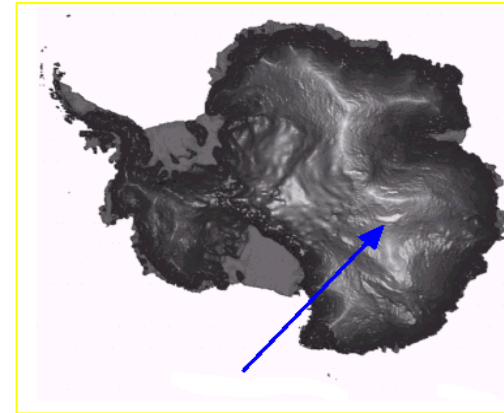
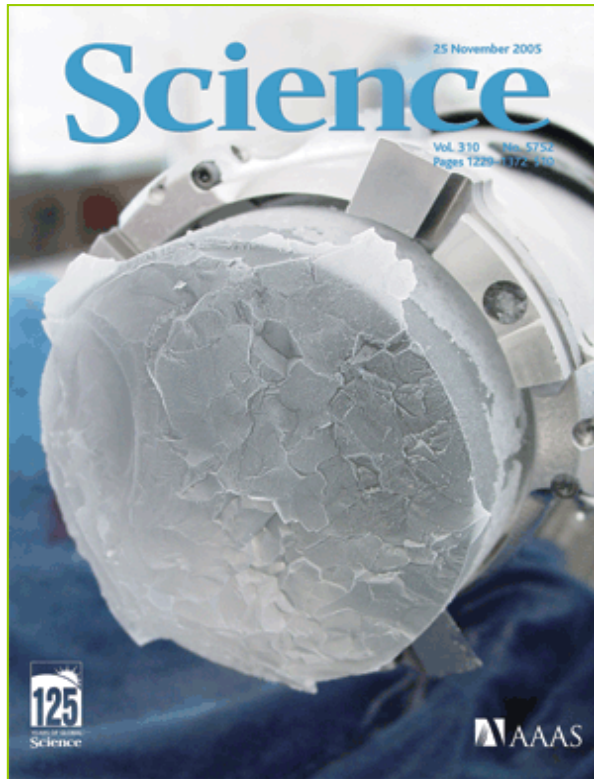
**Global warming** and **climate change** are terms for the observed century-scale rise in the average temperature of the Earth's climate system and its related effects.

This process consists of the global warming due to the emission of gas ( $\text{CO}_2$ , water steam, methane...) in the atmosphere. Greenhouse gases allow sunlight to pass through the atmosphere while obstructing the passage to the space of the infrared radiation from the Earth's surface and lower atmosphere (the heat re-issued); in practice they behave like the glass of a greenhouse and help to regulate and maintain the temperature of the earth with today.

## GREENHOUSE GASES

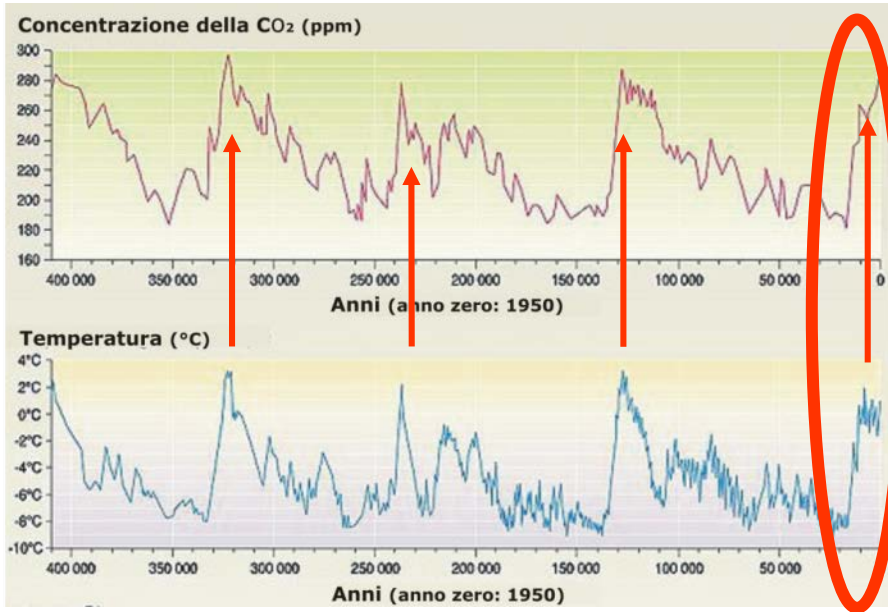


This is a natural process and allows that the temperature of the Earth be  $33^\circ\text{C}$  higher than what it would be without the presence of the gases.



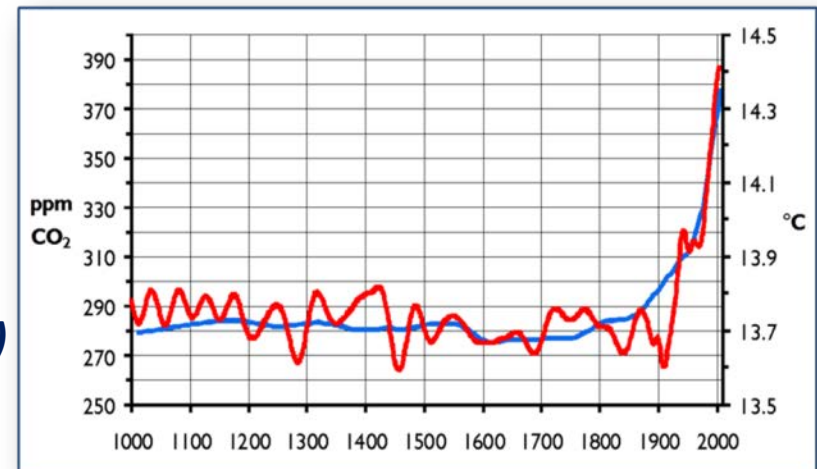
**Ice cores from Antarctica** have allowed to reconstruct the temperature trend and the CO<sub>2</sub> concentration in the atmosphere for the the last 400.000

# GLOBAL WARMING



Correlation between temperature increase and concentration of  $\text{CO}_2$  in the atmosphere over the last 400,000 years (drilling of ice in Antarctica)

$\text{CO}_2$  concentration in the atmosphere is increased by circa **~40%** from 1750 (Rivoluzione Industriale; IPCC, 2014)



Global variation of the temperature (red) and the  $\text{CO}_2$  present in the atmosphere (blue) in the last 1000 years.

**Maximun concentration of  $\text{CO}_2$  (last 400.000 years)**

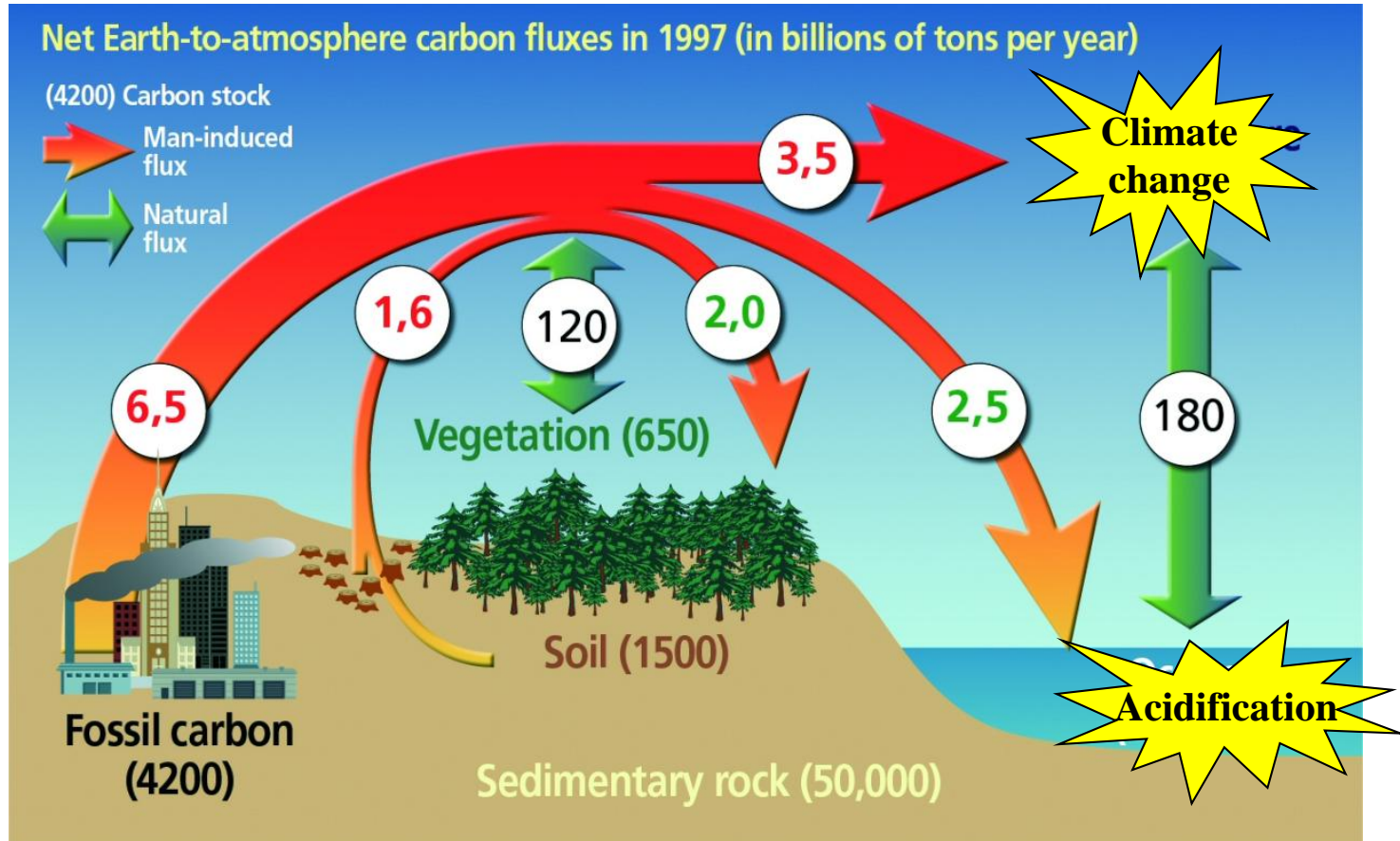
**300 ppm**

**IN 2005:**

**381 ppm**

# CO<sub>2</sub> exchange between Earth and Atmosphere (Billiontons/years of Carbon)

**Total amount of emitted CO<sub>2</sub> : 30 billion tons /year or 8.1 billiontons/years of carbon**



© BRGM [im@gé](mailto:im@gé)

World emissions of CO<sub>2</sub> from the usage of fossil fuels:

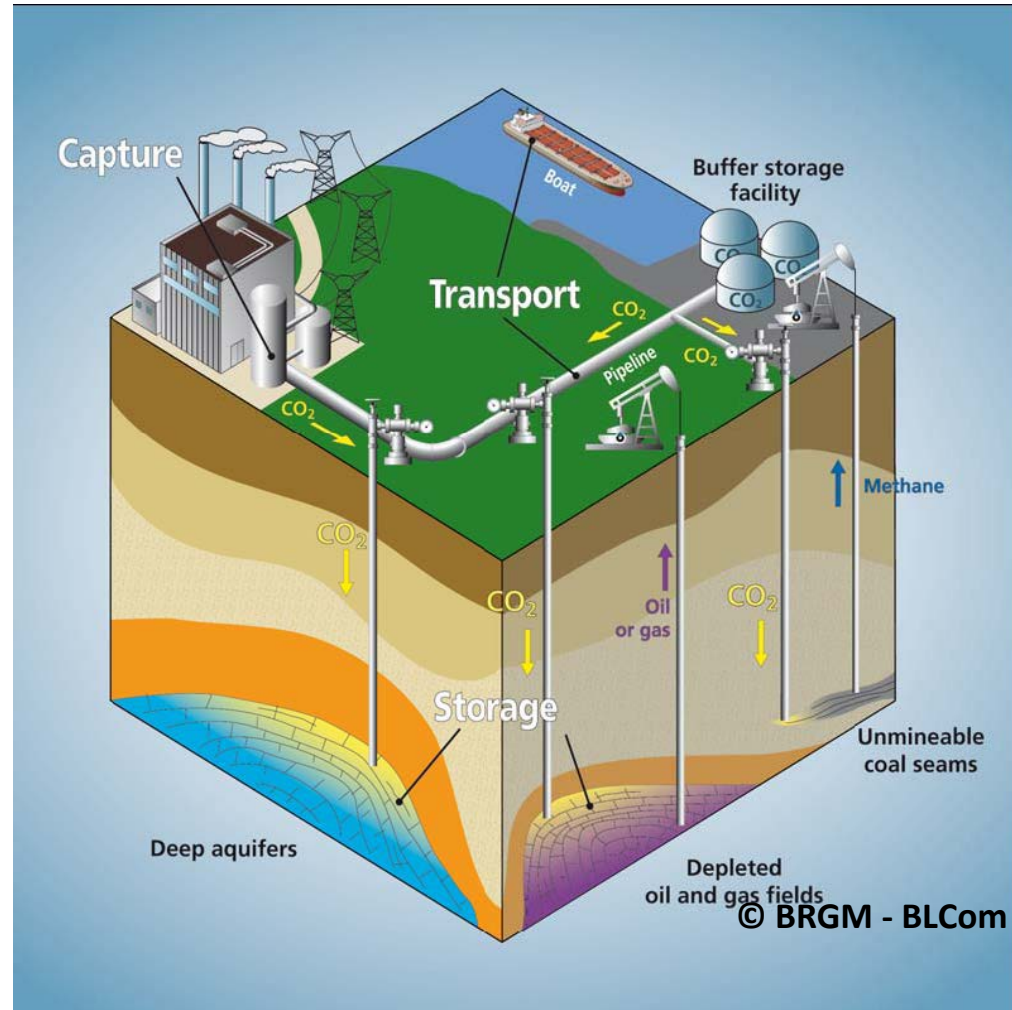
6.5 Gt C/y (o 24 Gt CO<sub>2</sub>/a)

# CO<sub>2</sub> GEOLOGICAL STORAGE CARBON CAPTURE AND STORAGE

.. one of the options to reduce the global CO<sub>2</sub> emissions by 2050

## Three main phases:

1. Capture
2. Transport
3. Storage



## CO<sub>2</sub> CAPTURE: MAIN CO<sub>2</sub> EMISSIONS

The main sources of CO<sub>2</sub> emissions consist of the **BIG STATIONARY SOURCES**:

- fossil fuel power plants
- industrial installations for the production of iron, steel, cement and
- chemicals refineries

### CAPTURE PROCESSES

- **POST-COMBUSTION:** separation of CO<sub>2</sub> from other products (water vapor, N<sub>2</sub> ...) of combustion of primary resources and being absorbed by chemical solvents. The CO<sub>2</sub> is separated from the solvent and compressed to be transported and stored.
- **PRE-COMBUSTION:** treatment of the fuel with oxygen or air and / or steam before combustion to produce a gas mixture consisting of carbon dioxide and hydrogen, separated before emission
- **OXYGEN COMBUSTION:** oxygen is separated from the air before combustion and mixed with fuel at high pressure. This combustion process produces only vapor and concentrated CO<sub>2</sub> that can be treated easily and sent to be stored.

## TRANSPORT OF CO<sub>2</sub>

La CO<sub>2</sub> can be transported, both onland and offshore, in three phases:

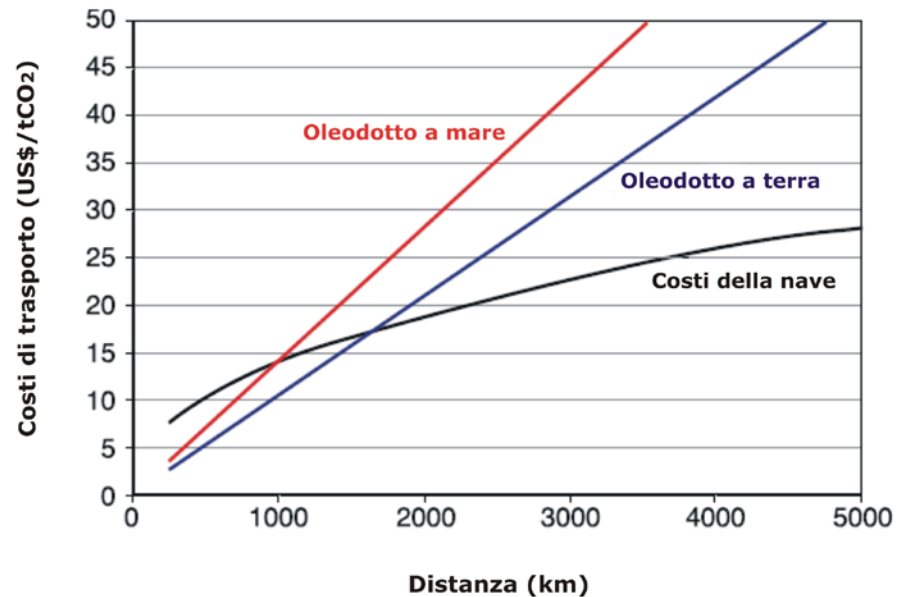
**GAS**

**LIQUID**

**SOLID**

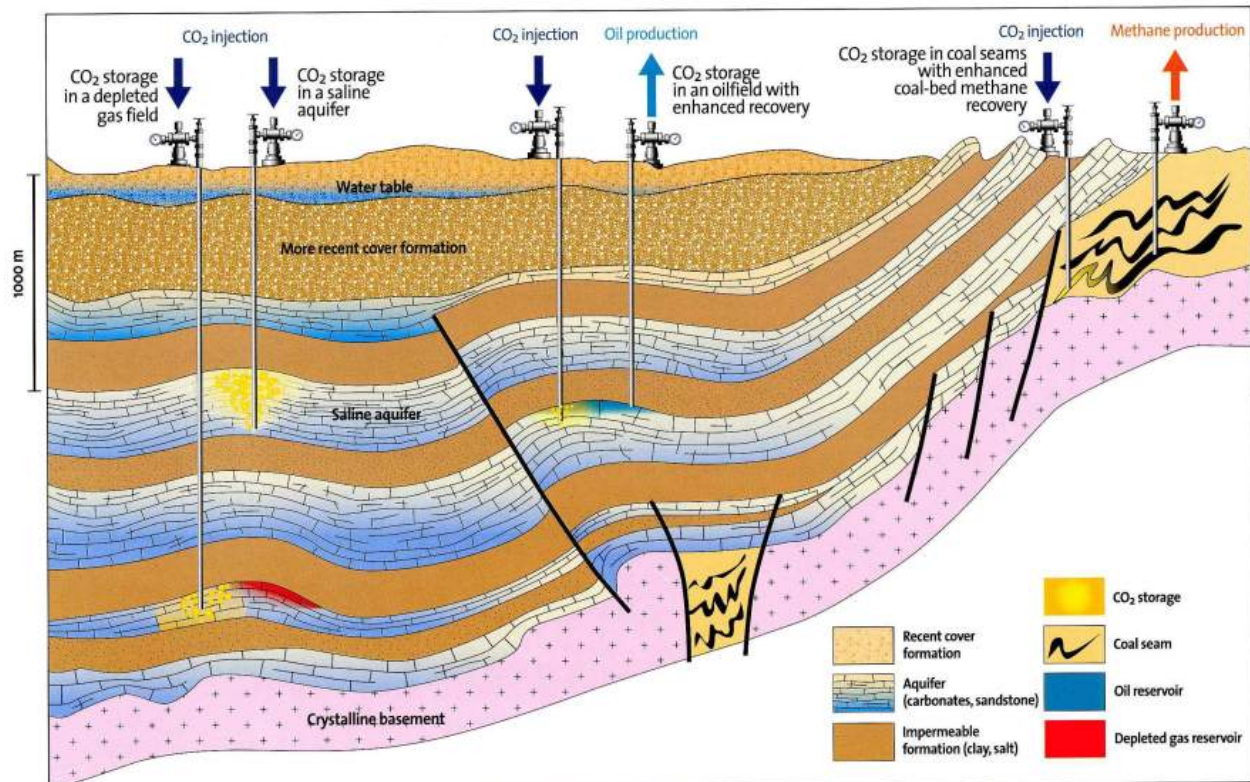
Tanks, pipelines and ships

Not economically convenient





## STORAGE OPTIONS



### Existing Reservoir

- Saline aquifers
- Oil and gas filed depleted
- Coal seams

## CRITERIA FOR IDENTIFICATION OF SUITABLE SITES FOR CO<sub>2</sub> STORAGE

**Depth** : between 800 (to allow the CO<sub>2</sub> supercritical stage) and 2000-3000 m

**Characteristics of the reservoir**: good porosity e permeability

**Caprock**: presence of a sealing geological formation

**Distance**: within a radius of 200 km from the source of emission of CO<sub>2</sub>

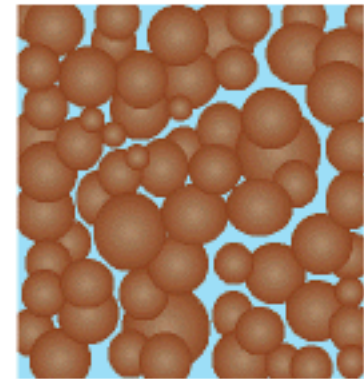
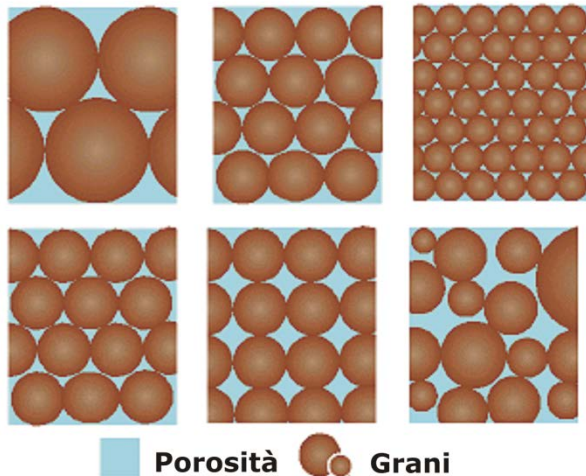
**Heat flow**: the heat flow does not have to be high, in order not to alter the conditions of stability of CO<sub>2</sub>

**Tectonic setting/seismicity**: the area must be stable to ensure the structural conditions for storage

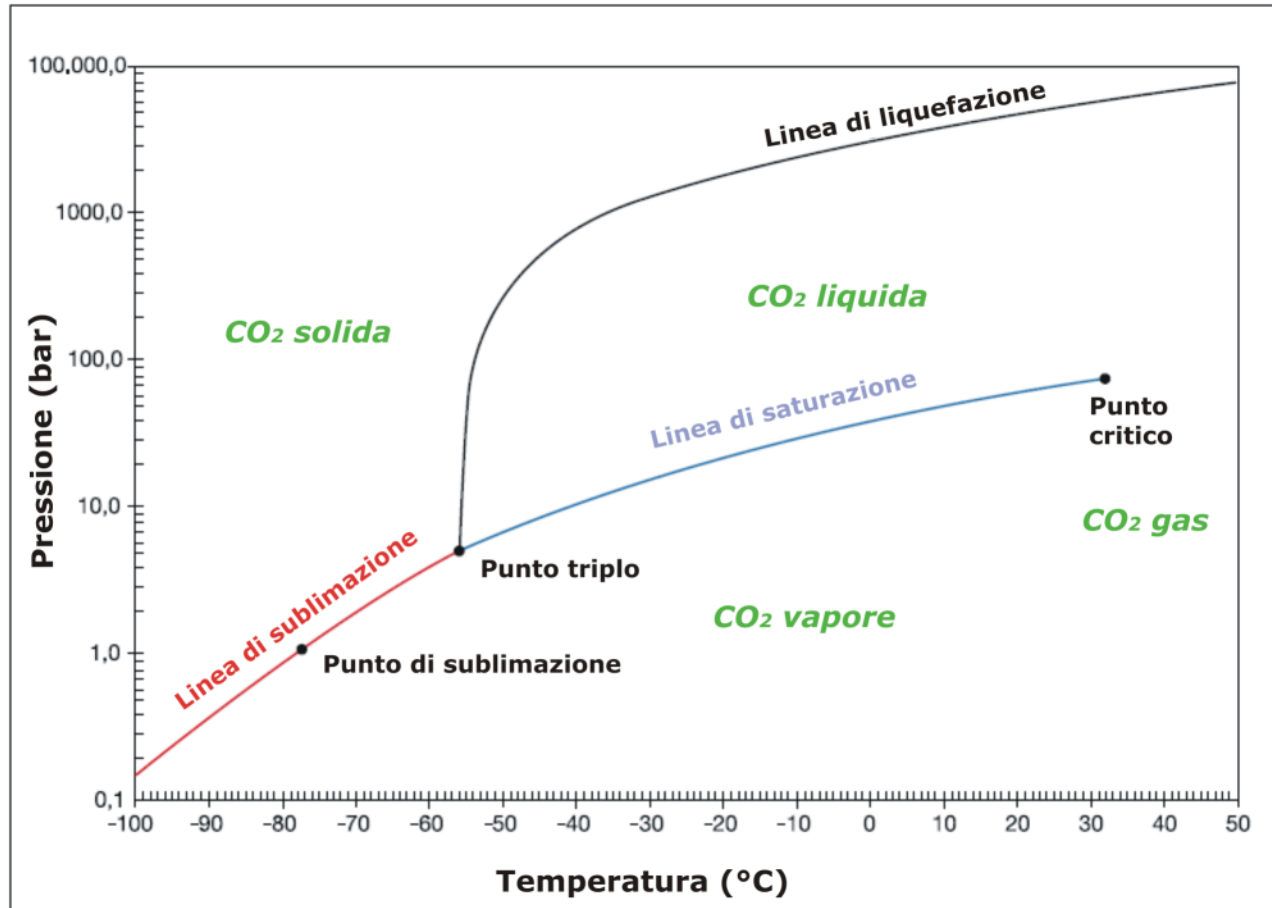
## CO<sub>2</sub> STORAGE

For the purposes of CO<sub>2</sub> storage, the rock that serves as a reservoir must meet the following requirements :

- they must be at a DEPTH between 800 (so that the CO<sub>2</sub> remains in conditions of supercritical state) and 1500 m;
- they must have a certain porosity and permeability;



## CO<sub>2</sub> PHASE: “supercritical state”



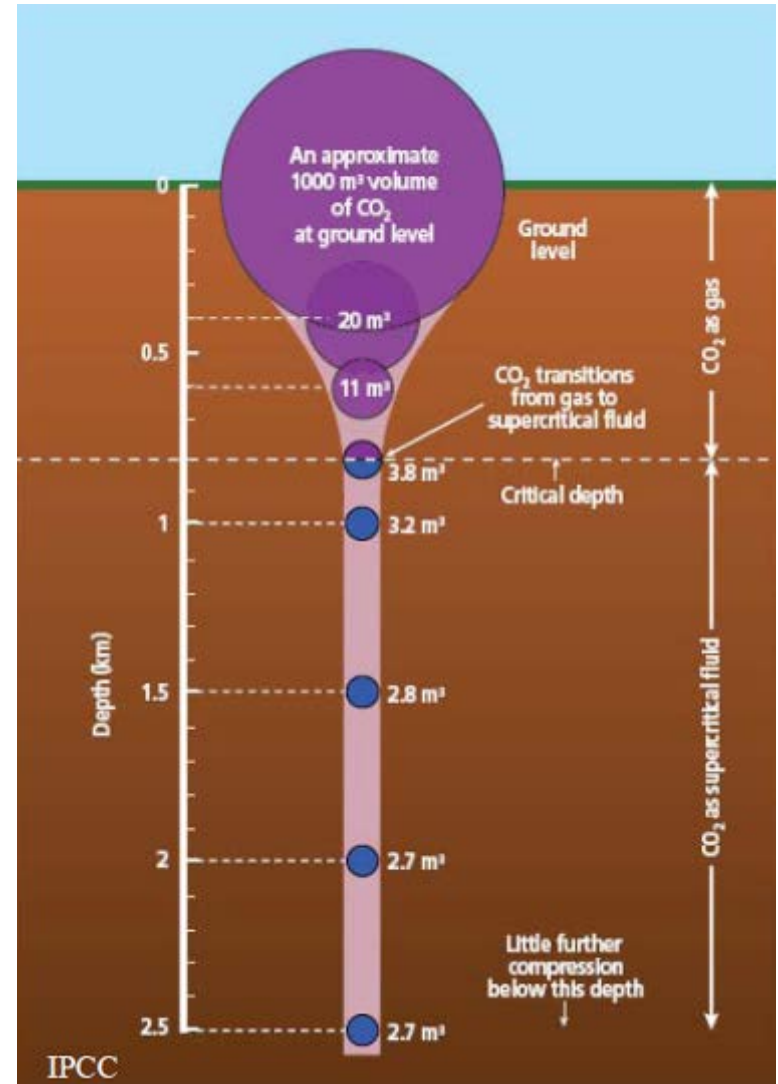
- $T > 31.1^\circ \text{C}$
- $P > 73.9 \text{ bar}$

## ...CO<sub>2</sub> in supercritical state is liquid or gas?

ANSWER:

- density similar to liquid
- viscosity similar to gas

T=100°C, P=280bar (2800m)	density (kg/m <sup>3</sup> )	Viscosity (cP)
CO <sub>2</sub> supercritic	615	0.05
water	804	0.16
gas (methan)	150	0.02

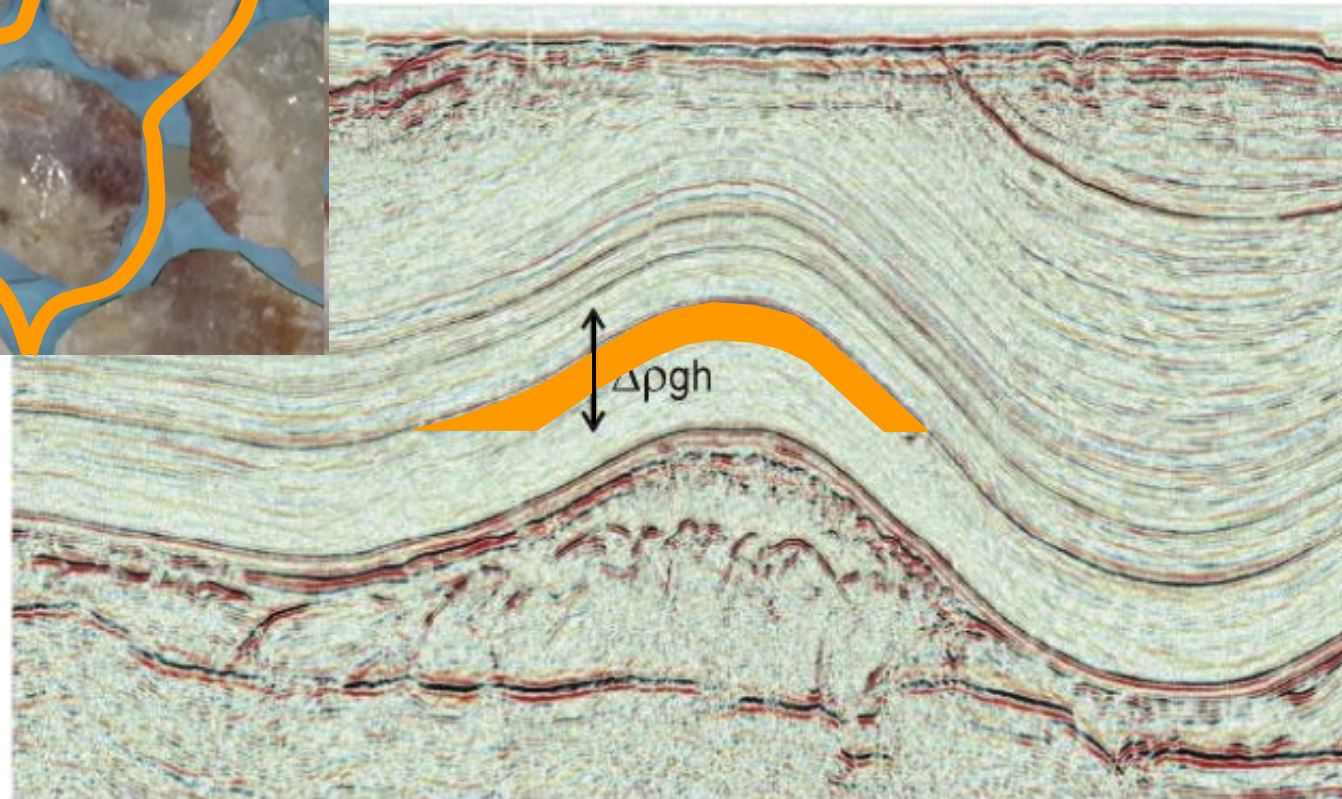


# The CO<sub>2</sub> at supercritical conditions tends to rise ...

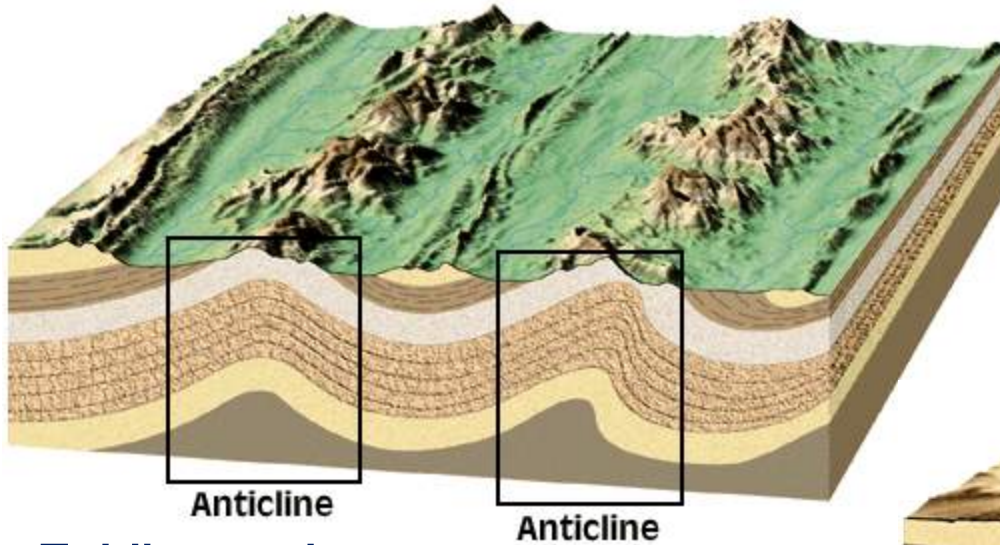
CAP ROCK



ESSENTIAL PRESENCE OF SEALING  
ROCK FORMATIONS (CAPROCK)



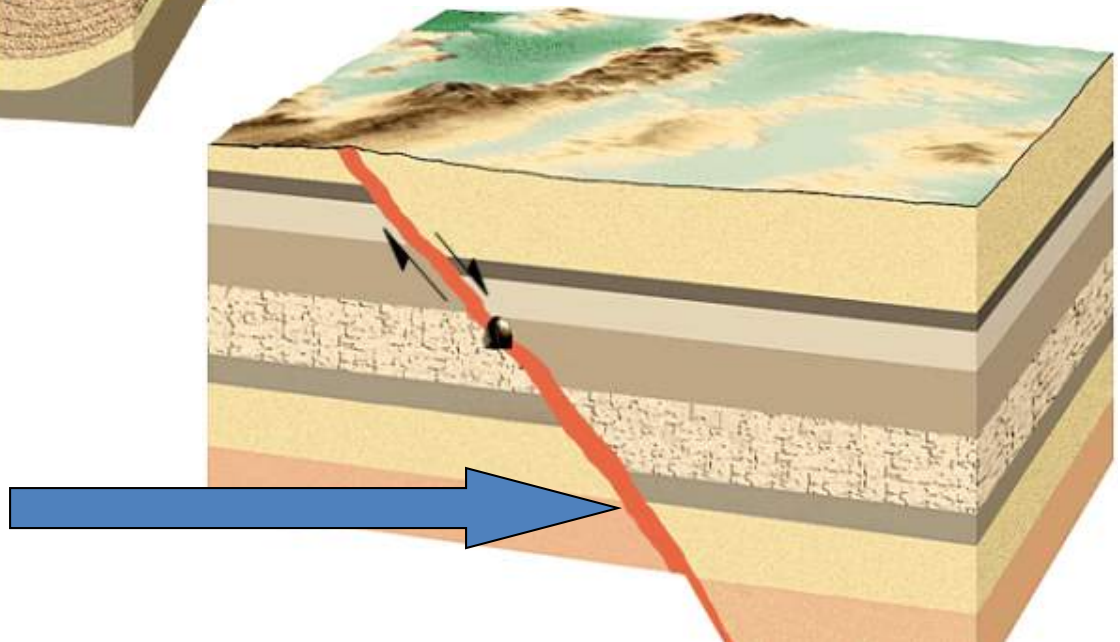
# STRUCTURAL TRAPS



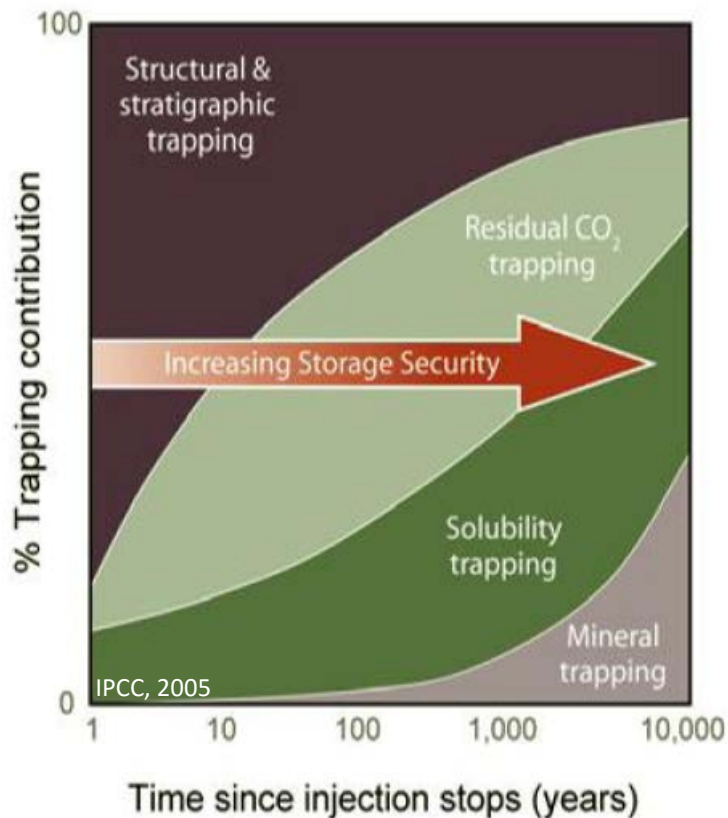
**Folding and anticlines**

**Fault consists of different material**

**Faults and unconformities**



## Trapping mechanisms



- **Structural trapping:** the CO<sub>2</sub> is lighter than the salt water present in the interstices of the rock and it tends to rise upward and trapped by the impermeable rocks (caprock)

- **Hydrodynamic trapping,** where CO<sub>2</sub> is injected into supercritical conditions at depths > 800 m and it moves the present salt water

- **Dissolution trapping:** once injected CO<sub>2</sub> starts to dissolve in salt water. The water now becomes heavier and tends to drop. This mechanisms put in contact water with dissolved CO<sub>2</sub> with fresh water, promoting additional dissolution. After 10 years: 15% of injected CO<sub>2</sub> is dissolved; after 10.000 years 95% of CO<sub>2</sub> is dissolved.

- **Mineral trapping** where CO<sub>2</sub> reacts with some minerals in the aquifer to form crystalline carbonates



## KEY DATA FOR THE CHARACTERIZATION OF A RESERVOIR-CAPROCK SYSTEM

### Wellbore data

- Logs (Sonic, Gamma Ray)
- Porosity e permeability of reservoir e caprock rock formations
- Temperature and pressure at reservoir depth

### Multichannel seismic data

2D - regional scale

3D - site scale

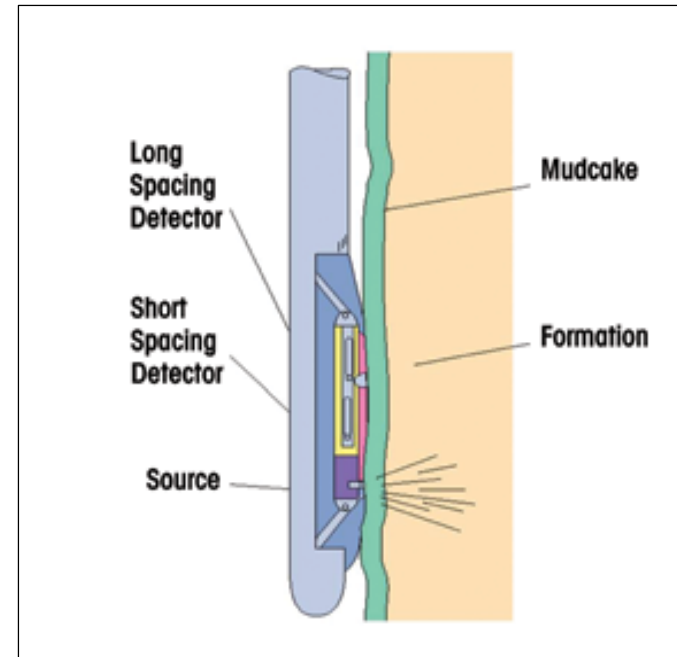


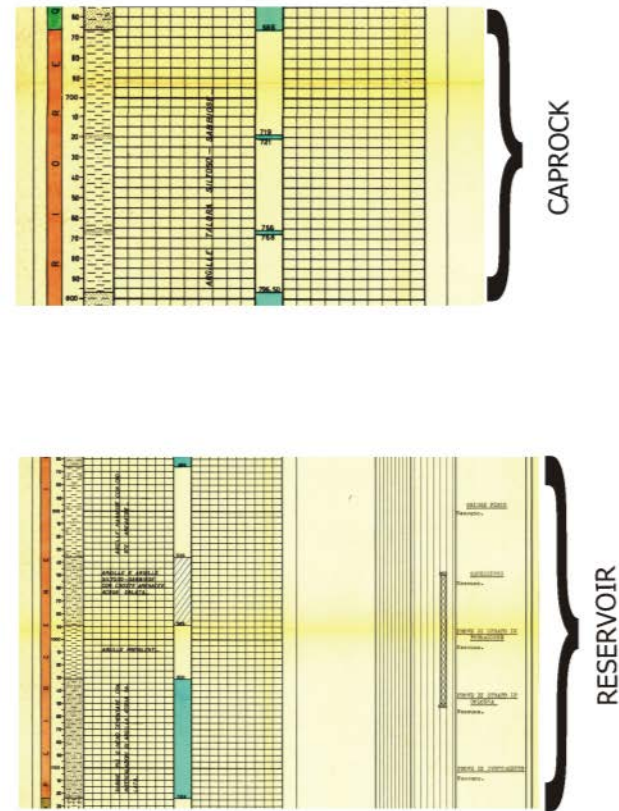
Image of a logging tool in a hole

# CHARACTERIZATION RESERVOIR-CAPROCK: WELL DATA analysis

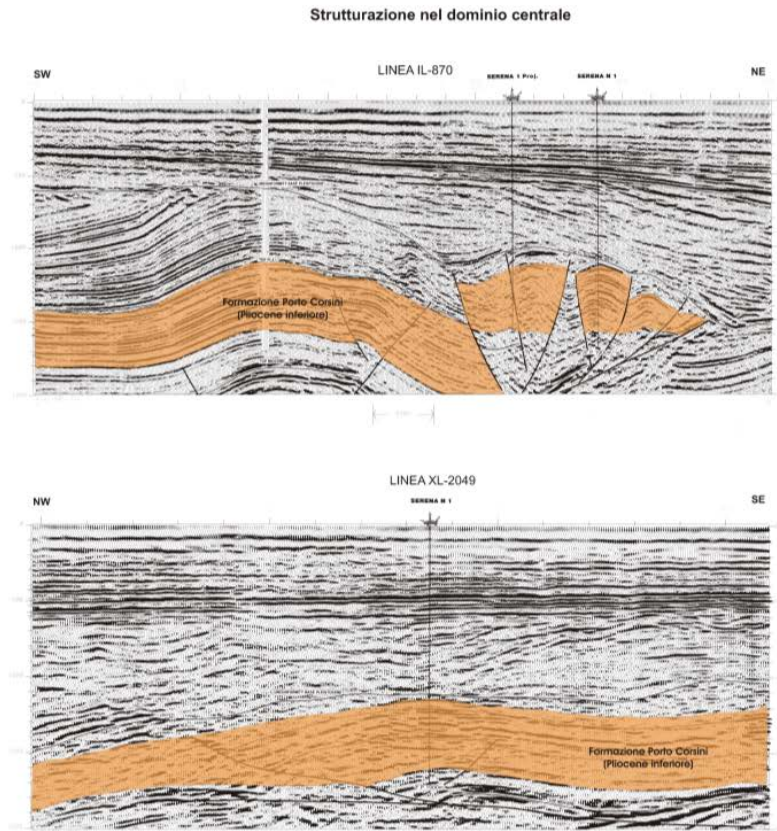
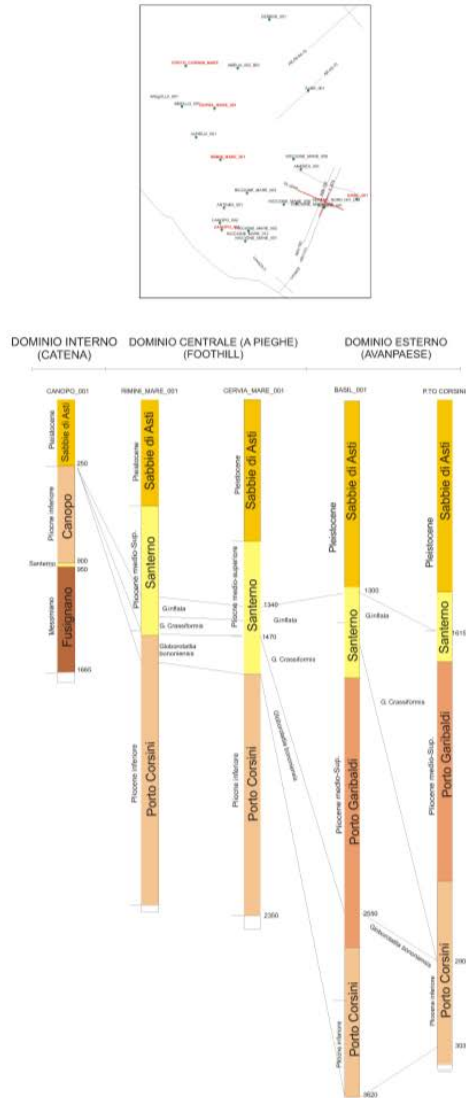
ANTINEA 1



RICCIONE MARE 2



# CHARACTERIZATION RESERVOIR-CAPROCK: SEISMIC DATA ANALYSIS



## *Main characteristics of a potential site for CO<sub>2</sub> storage*

- *Capacity*, to contain the amount of CO<sub>2</sub> to be stored; key parameter: **porosity**
- *Injectivity*, to inject the CO<sub>2</sub> a certain rate of injection; key parameter: **permeability of reservoir**
- *Containment*, to avoid CO<sub>2</sub> leakage; key parameter: **permeability of caprock**

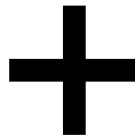
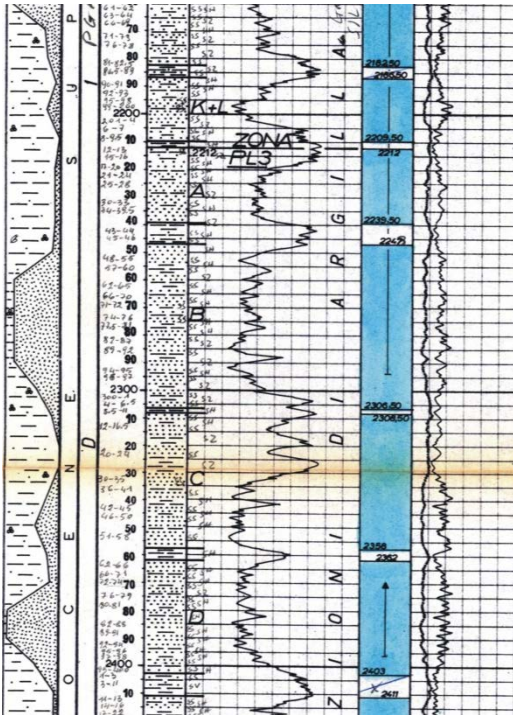
# ***CCS Project***

## ***Main steps***

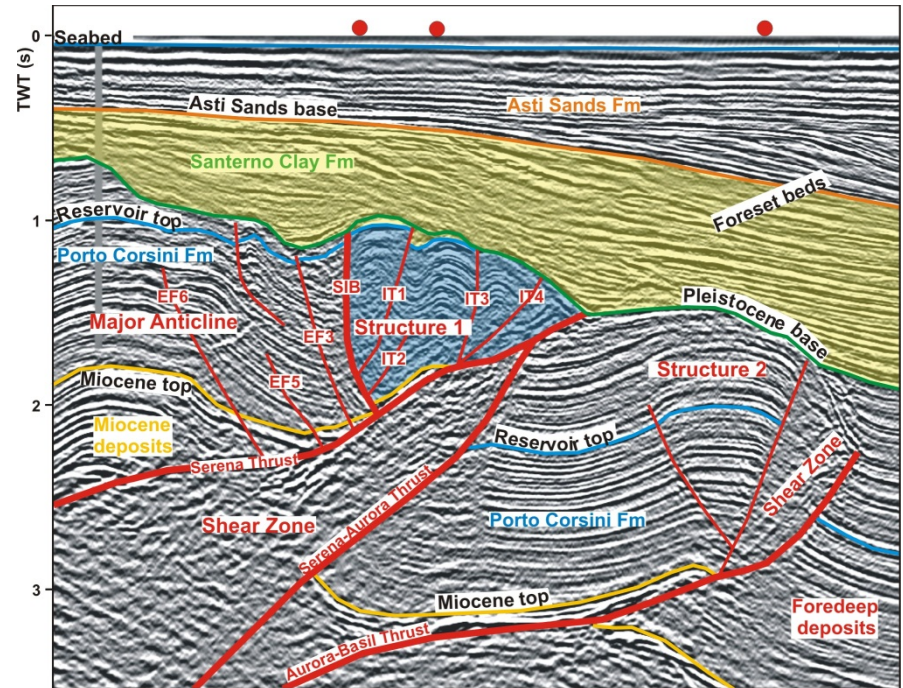
1. Identification of the potential storage site
2. Modelling of CO<sub>2</sub> injection
3. Monitoring (pre-, during and post-injection)
4. Risk evaluation and remediation plan

# Data analysis

## Geophysical log analysis

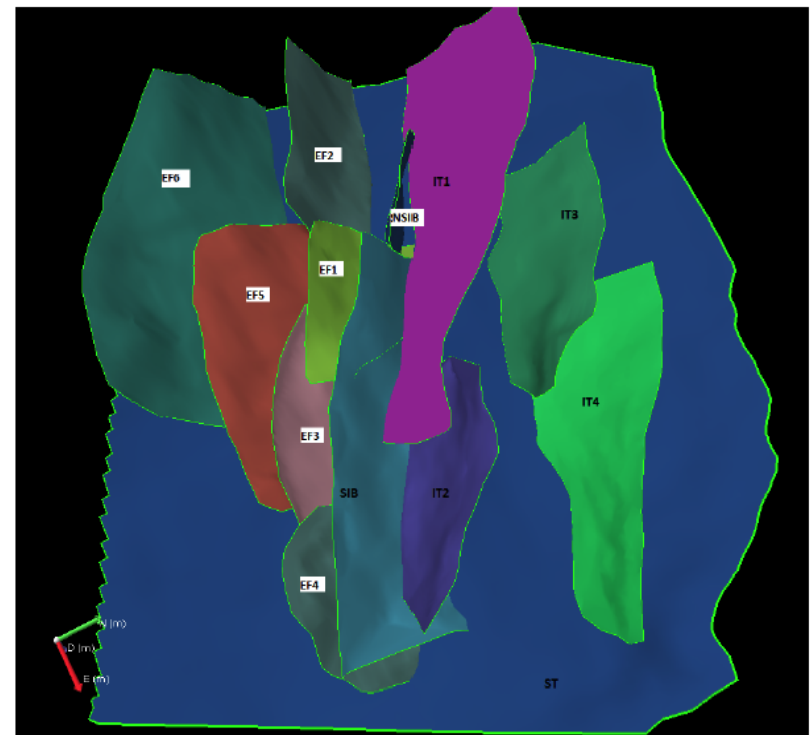
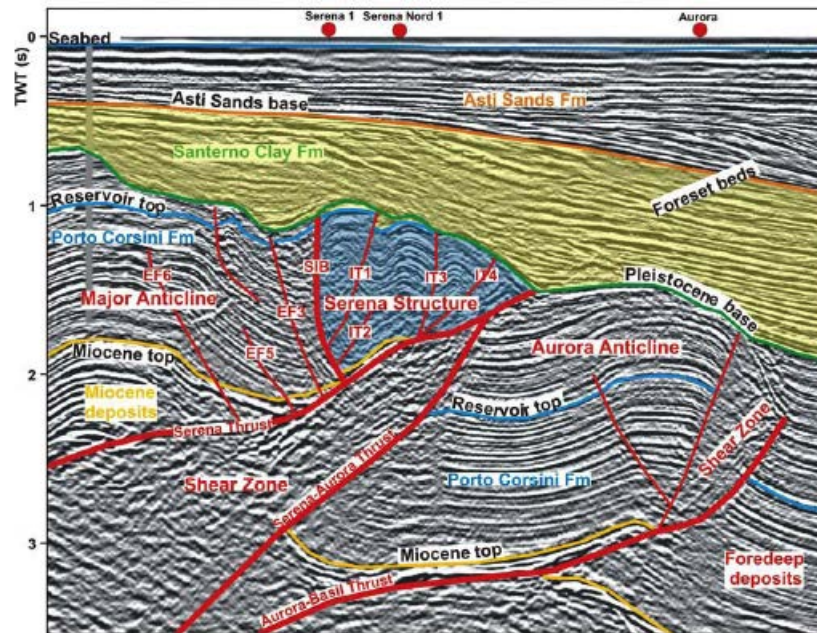
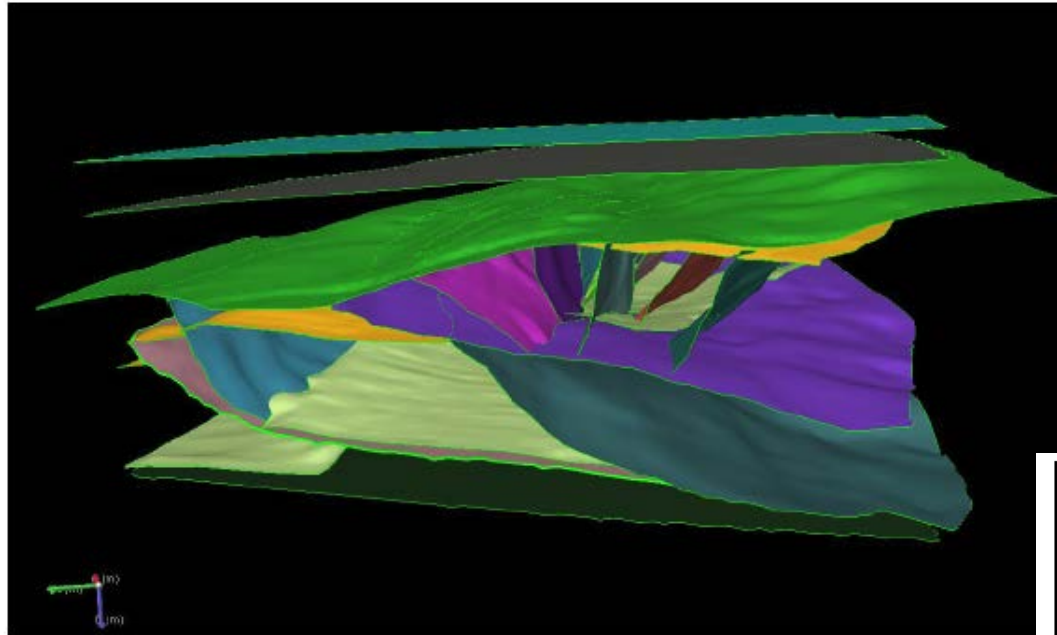


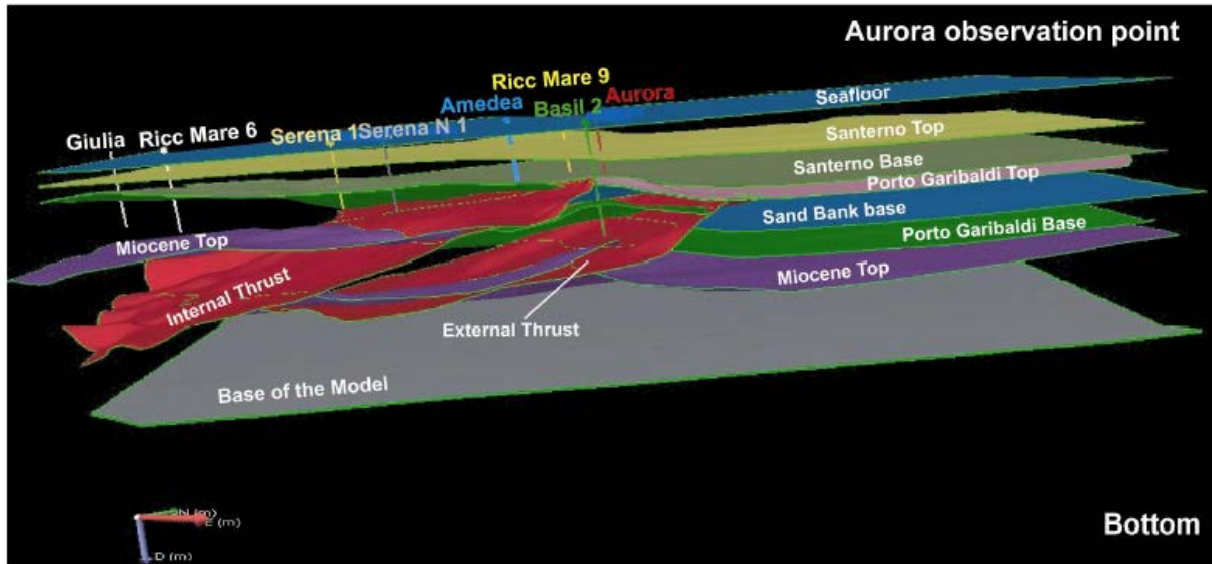
## Seismostratigraphic and structural interpretation of multichannel seismic profiles



## Geological modeling

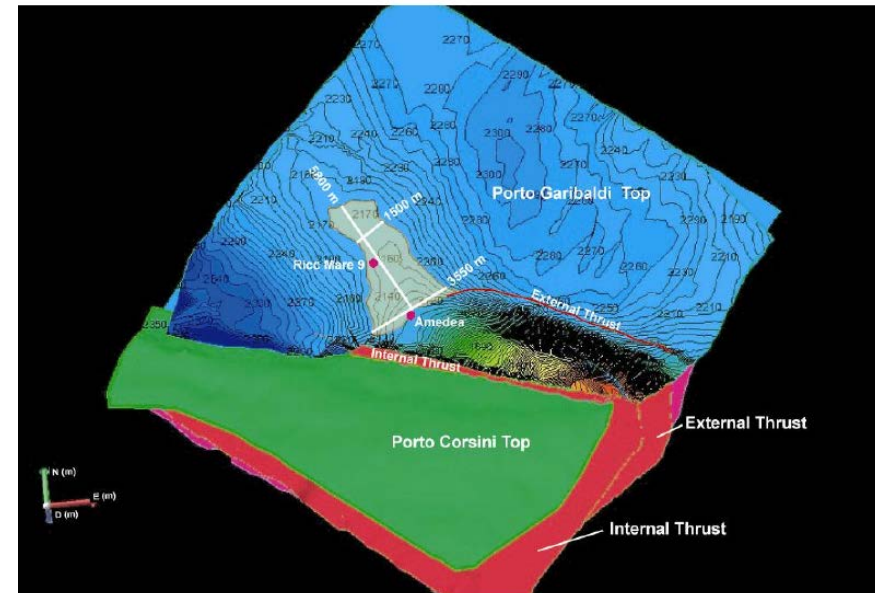
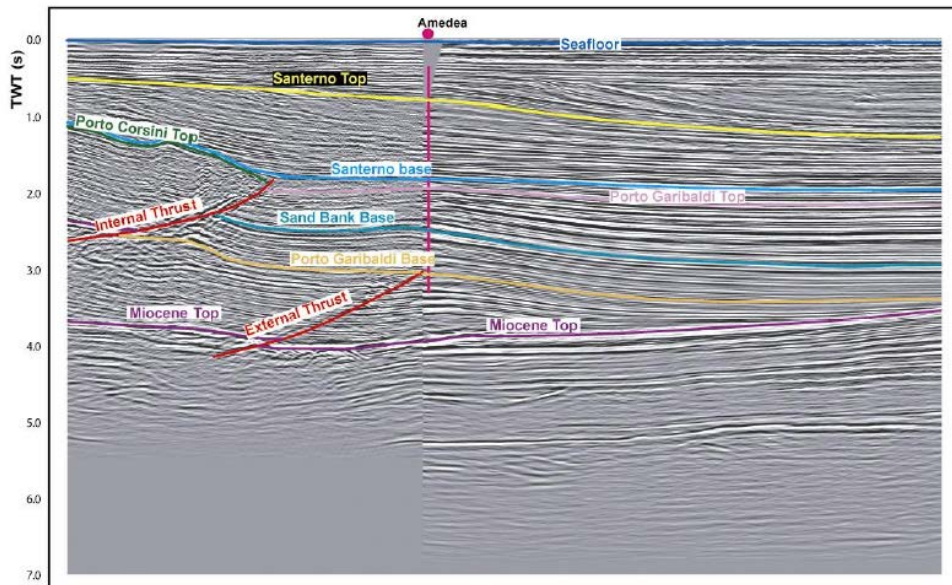
Example of 3D geological model





## Geological modeling

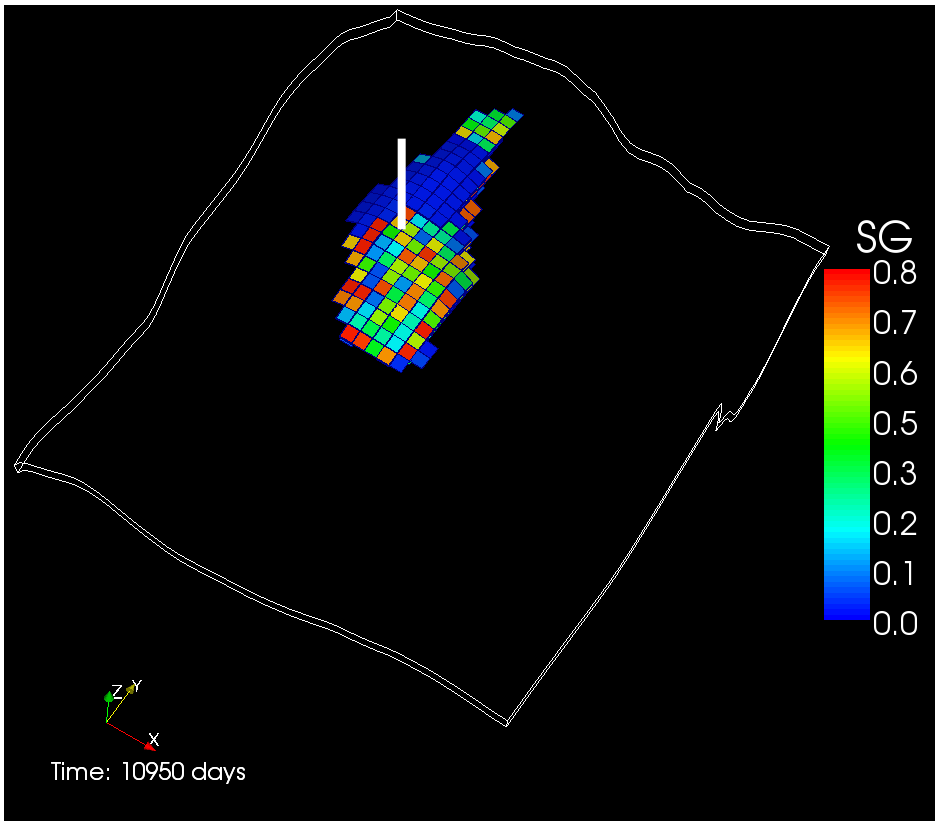
Example of 3D geological model



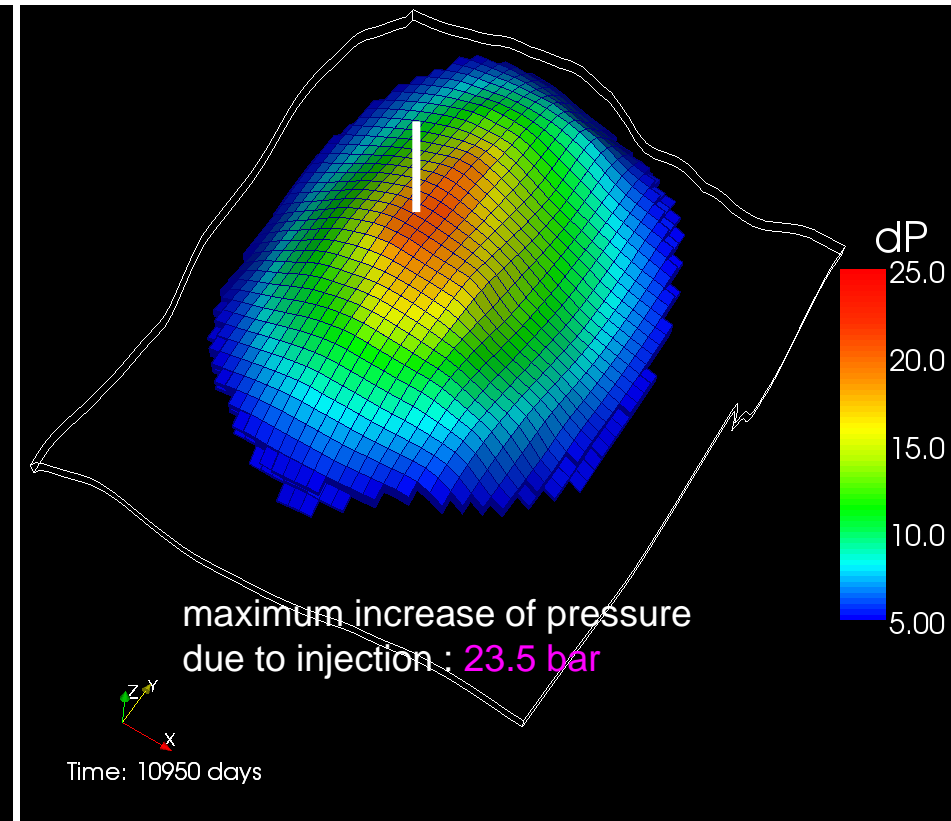


## Modeling of CO<sub>2</sub> Injection

### ONE WELL located on top of the anticline



**Free CO<sub>2</sub> saturation**



**Pressure increase (>5bar) from static conditions**



*Potential areas suitable for CO<sub>2</sub> geological storage in siliciclastic formations*

**PRELIMINARY ESTIMATES OF THE STORAGE CAPACITY: ~ 12 Gt**

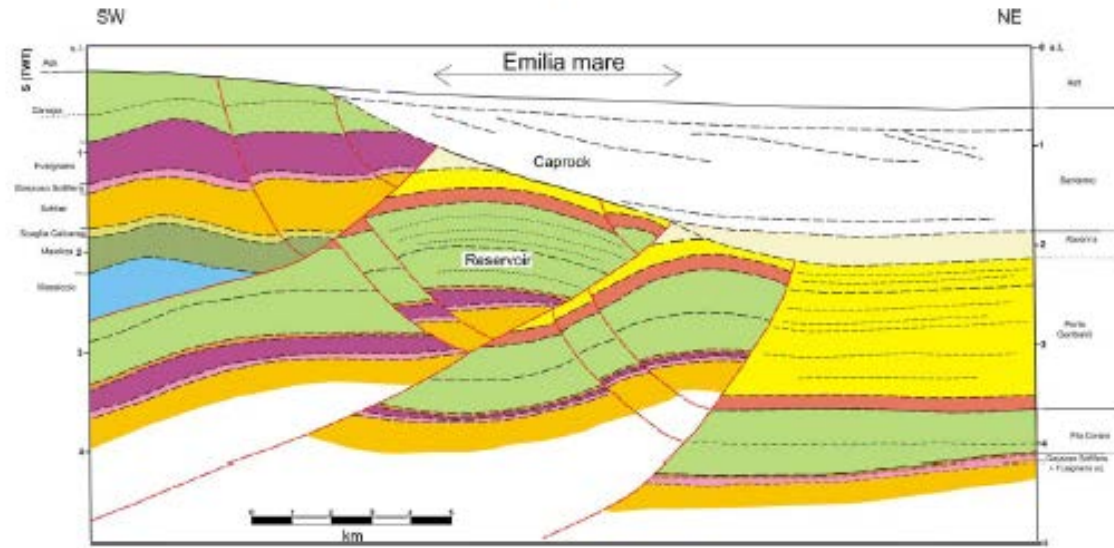
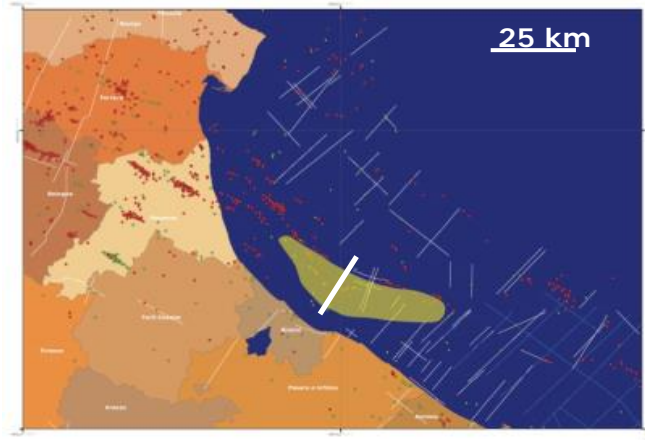
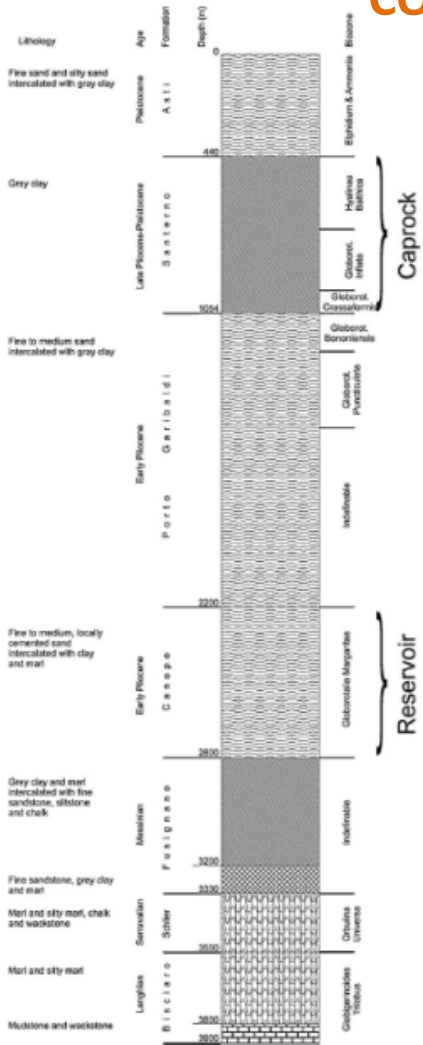


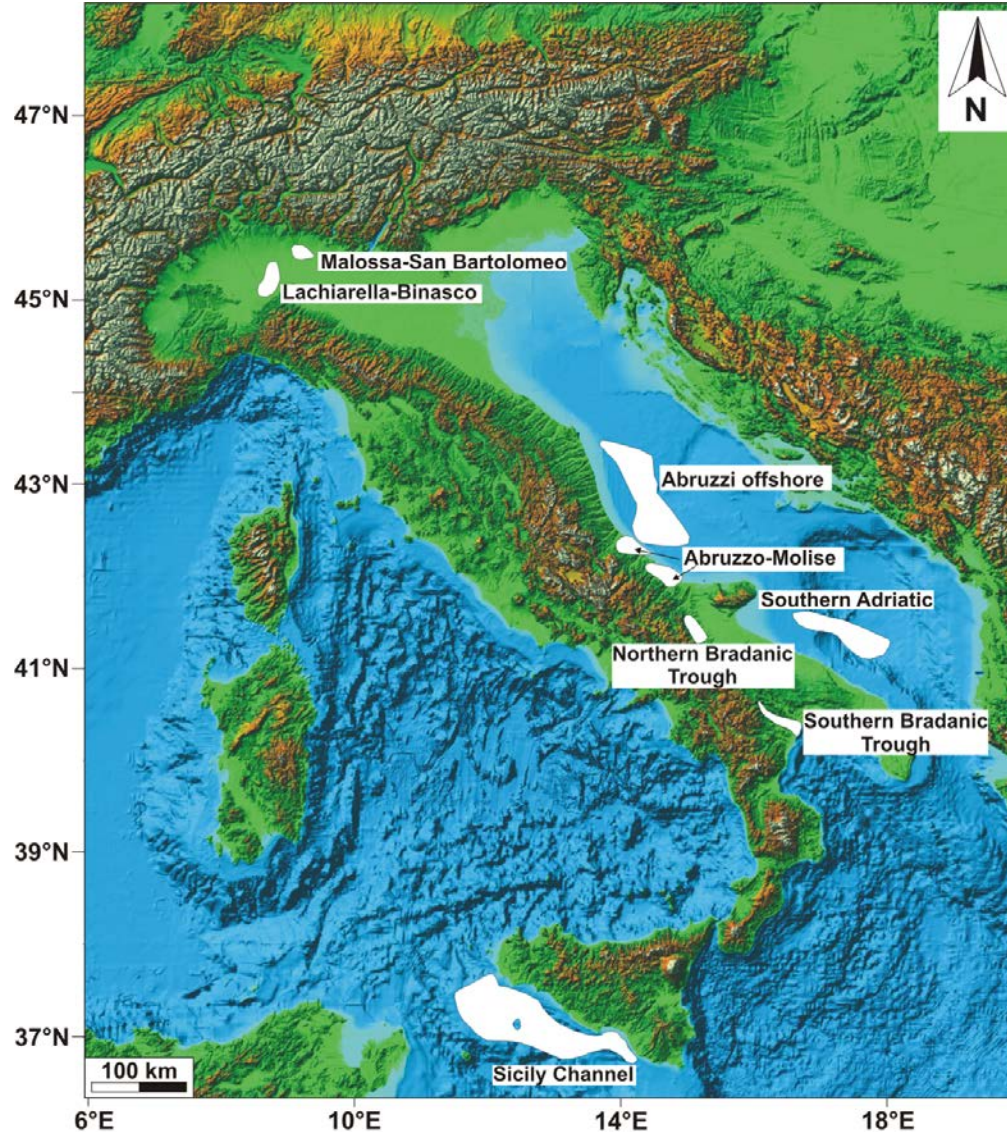
*Storage of Italy's annual CO<sub>2</sub> emissions for the next 50 years*

Donda et al., 2011

# Example of a potential area suitable for CO<sub>2</sub> geological storage in a terrigenous formation

"EMILIA MARE"

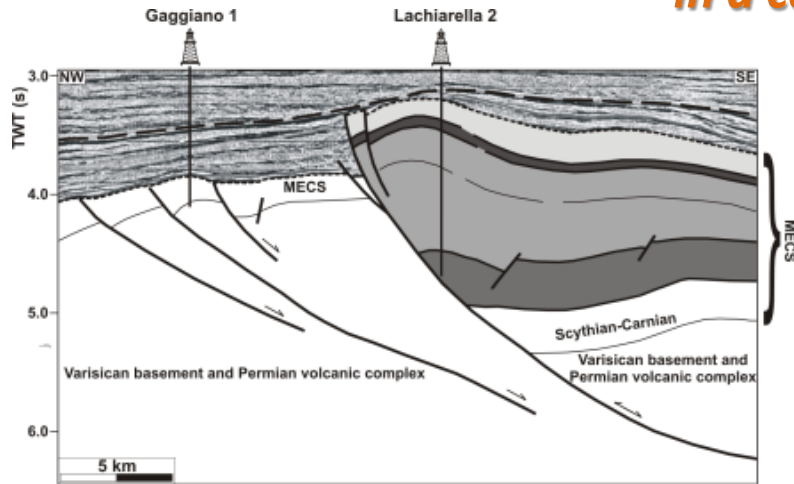




*Potential areas suitable  
for CO<sub>2</sub> geological  
storage in carbonate  
formations*

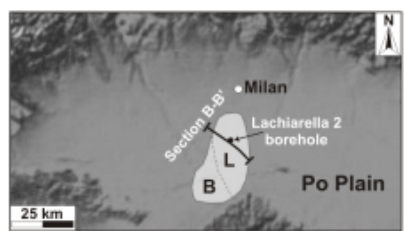
Civile et al., 2013

# Example of a potential area suitable for CO<sub>2</sub> geological storage in a carbonate formation

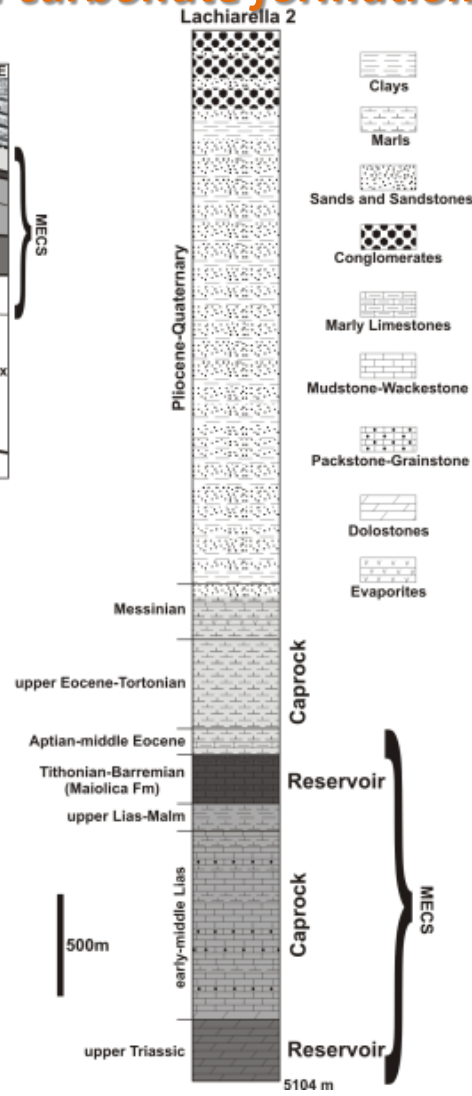


- Caprock**
  - upper Eocene-Messinian siliciclastic succession
  - Cretaceous-middle Eocene pelagic carbonate succession
  - Jurassic pelagic carbonate succession
- Reservoir**
  - Tithonian-Barremian pelagic succession (*Maiolica* Fm)
  - upper Triassic platform carbonate succession

- - - - - Tortonian unconformity  
 - - - - - Top carbonate succession  
 MECS: Mesozoic-middle Eocene carbonate succession  
 Fault

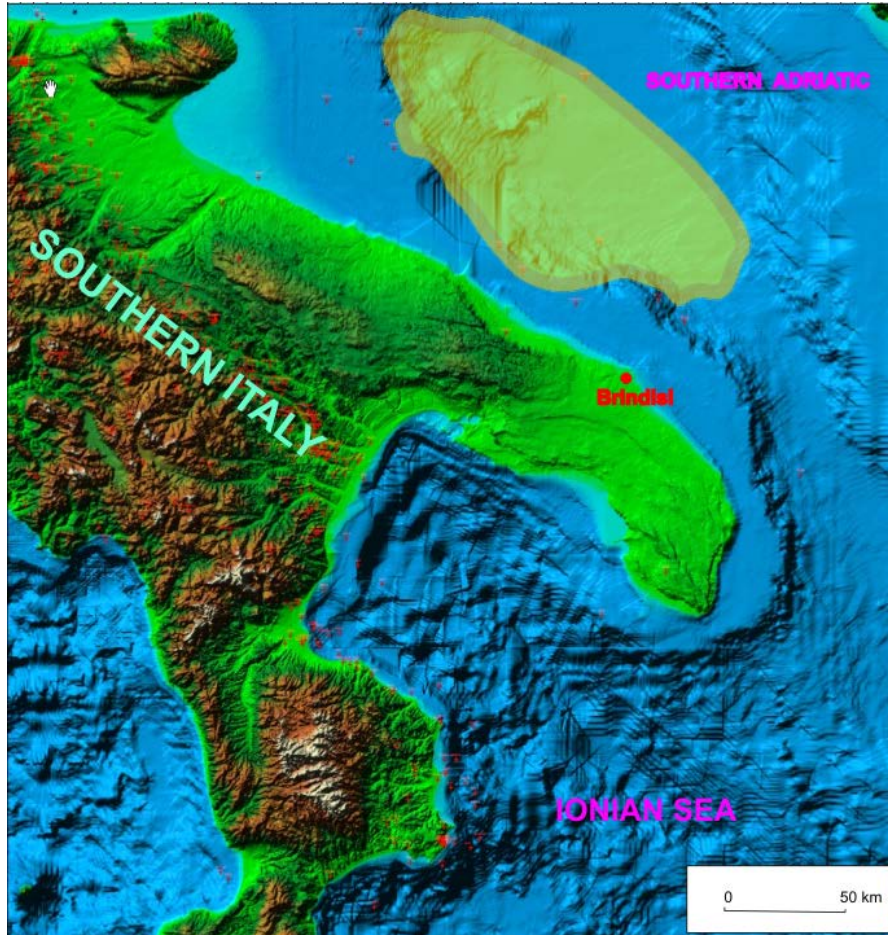


B: Binasco subzone  
 L: Lachiarella subzone



"Lachiarella-Binasco"

## CHARACTERISTICS OF THE SOUTHERN ADRIATIC SITE OPTIONS



### Storage options

- Saline aquifer/structural trap

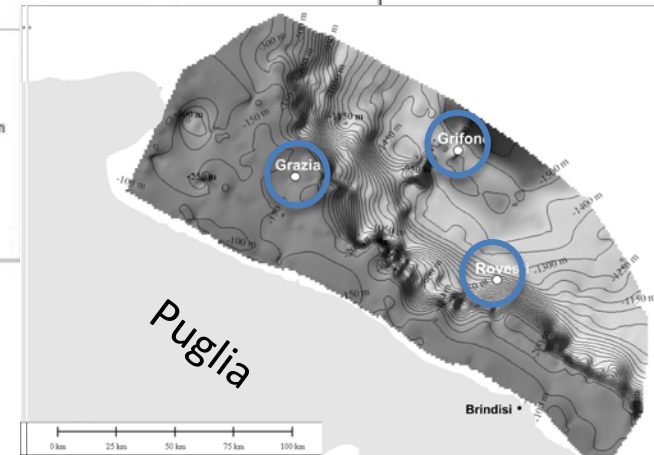
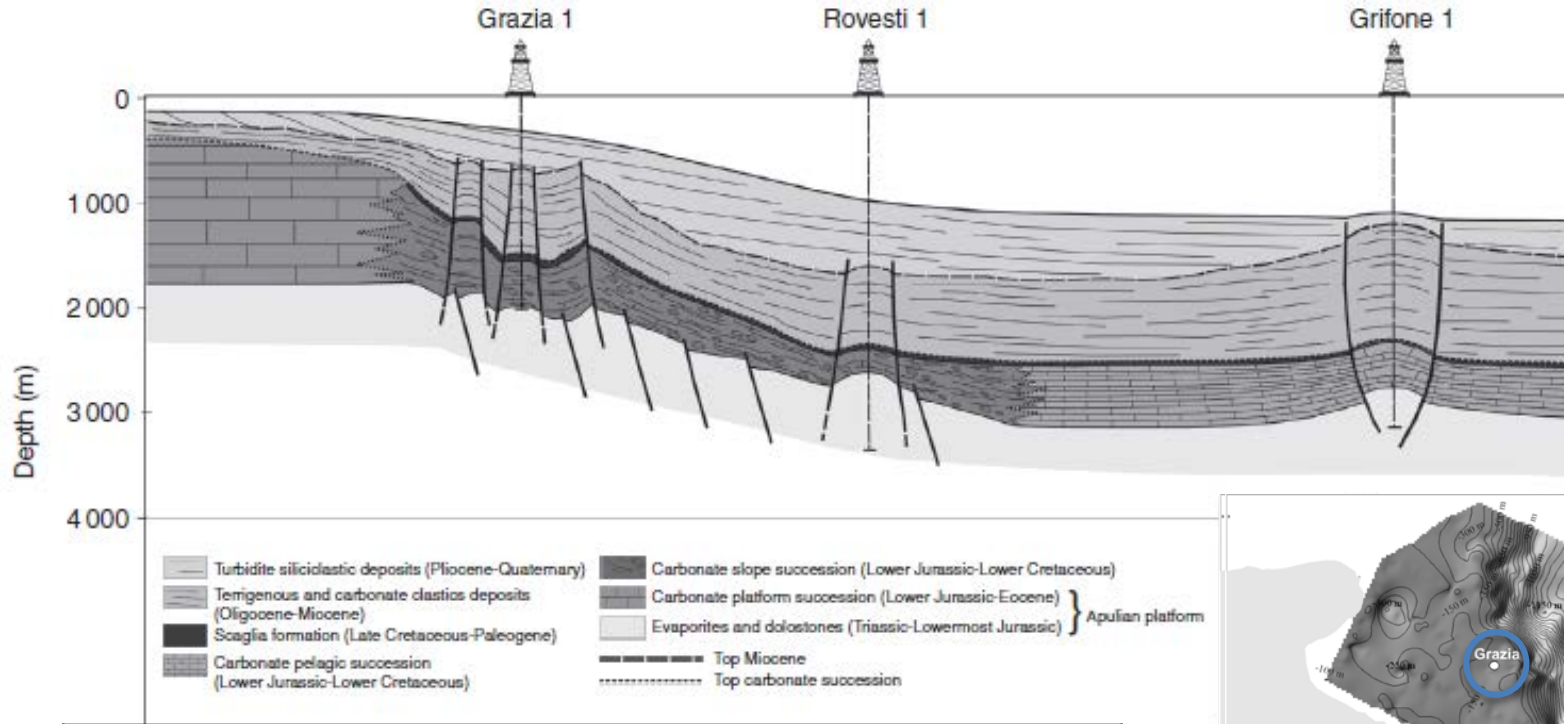
### Location

- Off shore

### Lithology

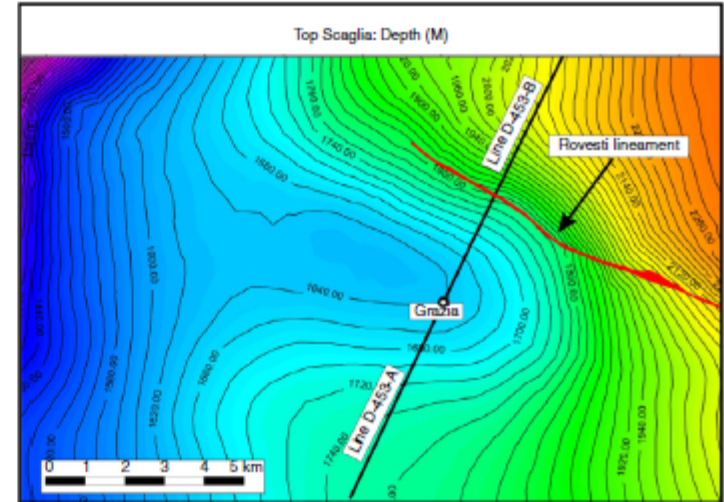
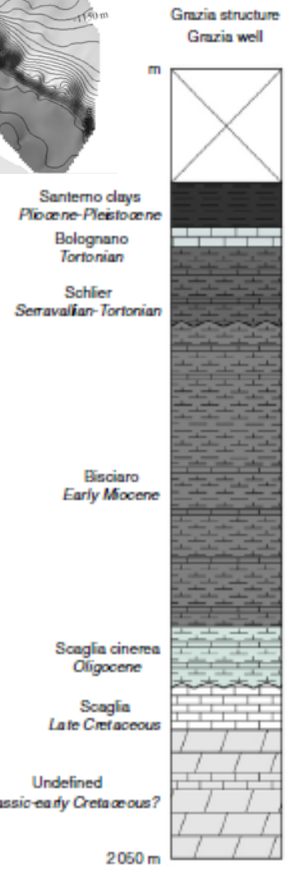
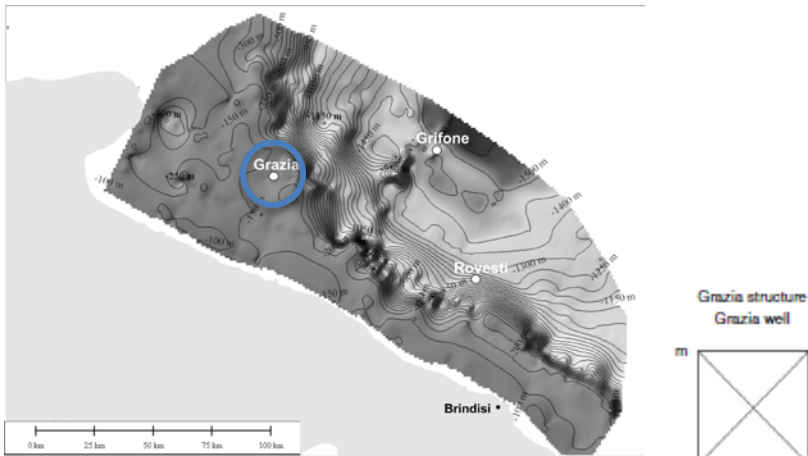
- Carbonate reservoir

## STORAGE SITE IN THE SOUTH ADRIATIC OFFSHORE

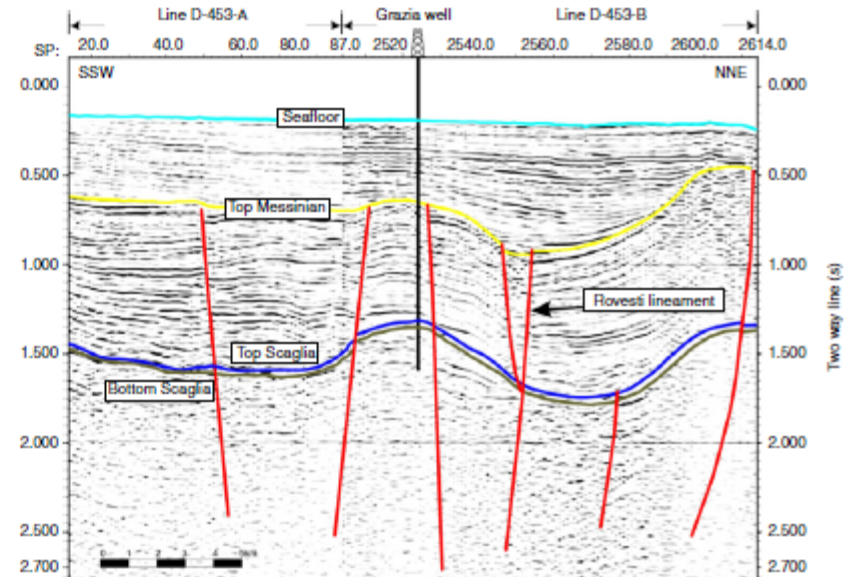


Name	Storage type	Area E+6 (M <sup>6</sup> )	Bulk Volume E+6 (M <sup>6</sup> )	Porosity (Scaglia)
Rovesti	Oil and Gas reservoir	1.7	195	13 - 15 %
Grifone	Saline aquifer	1.0	191	10 - 20%
Grazia	Saline aquifer	1.3	241	2 - 13 %

# GRAZIA STRUCTURE



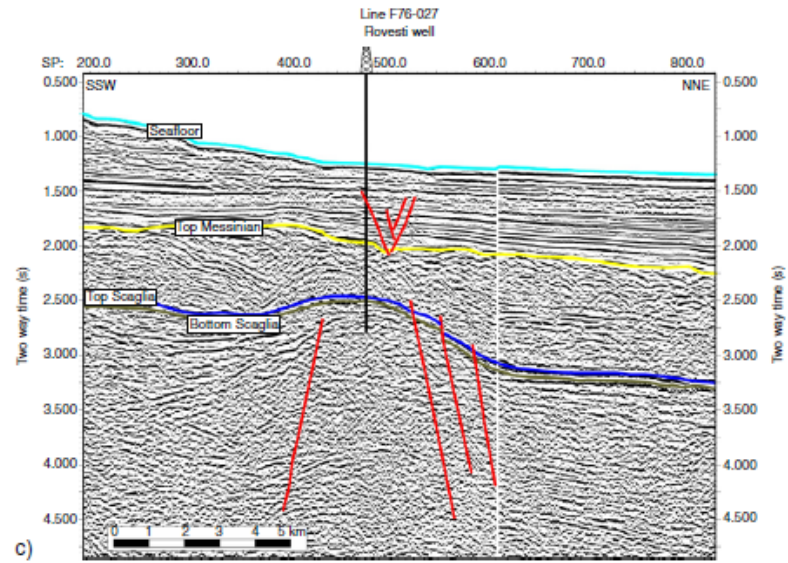
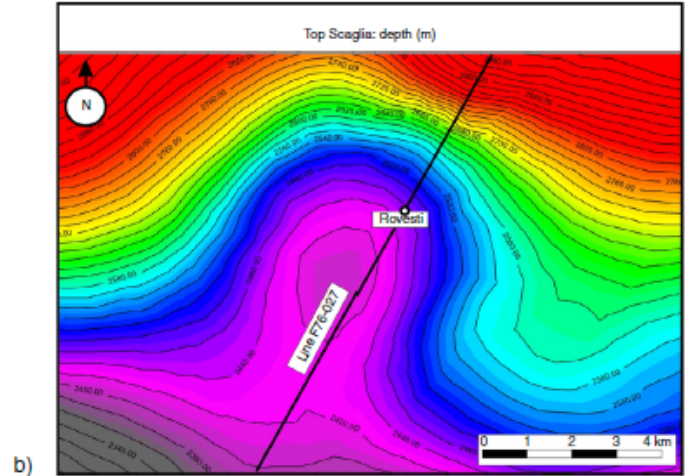
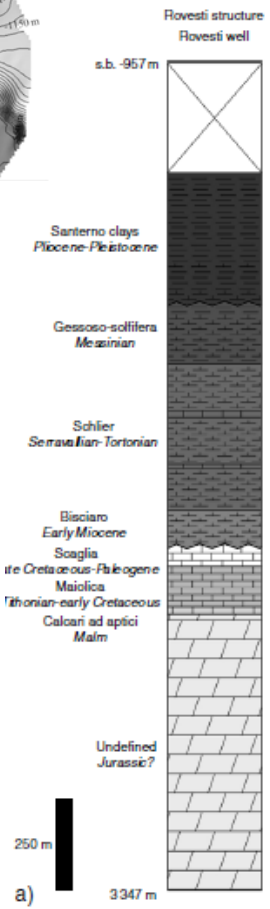
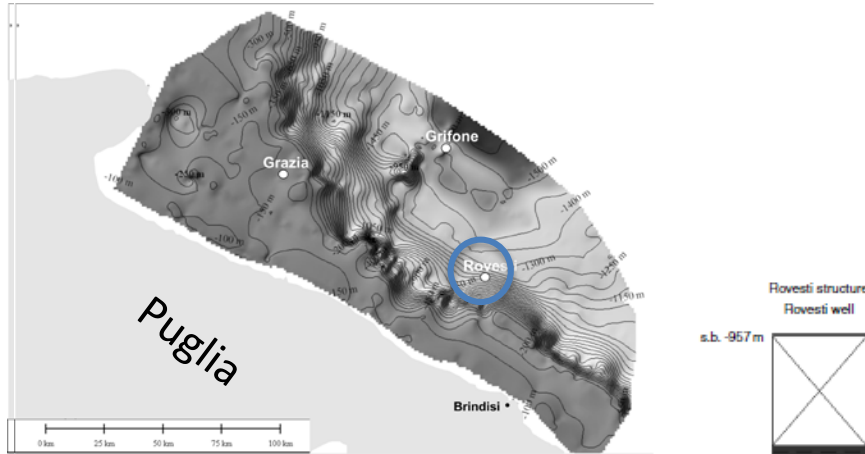
b)



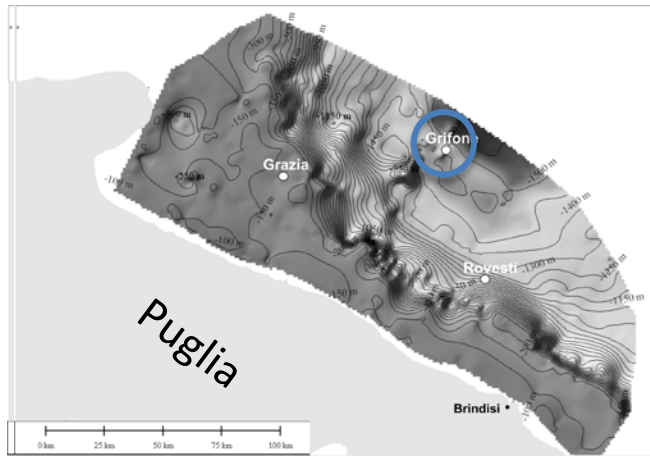
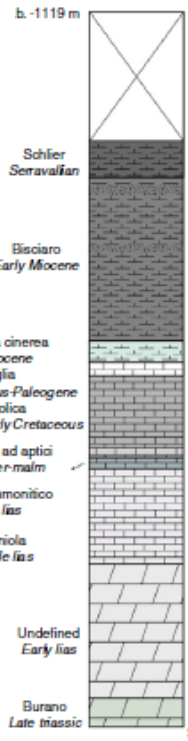
c)



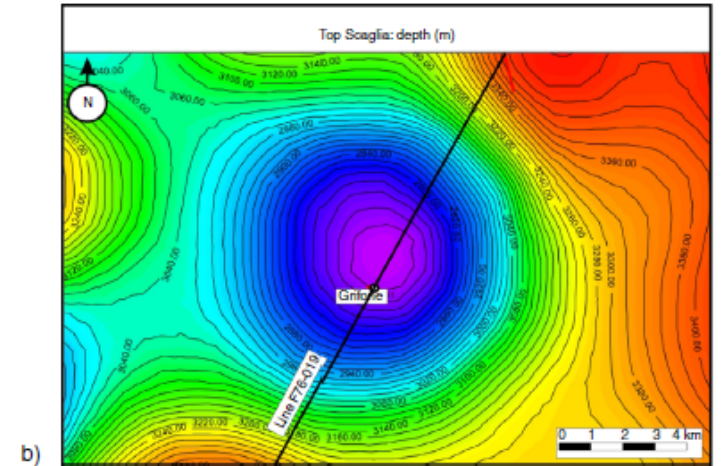
# ROVESTI STRUCTURE



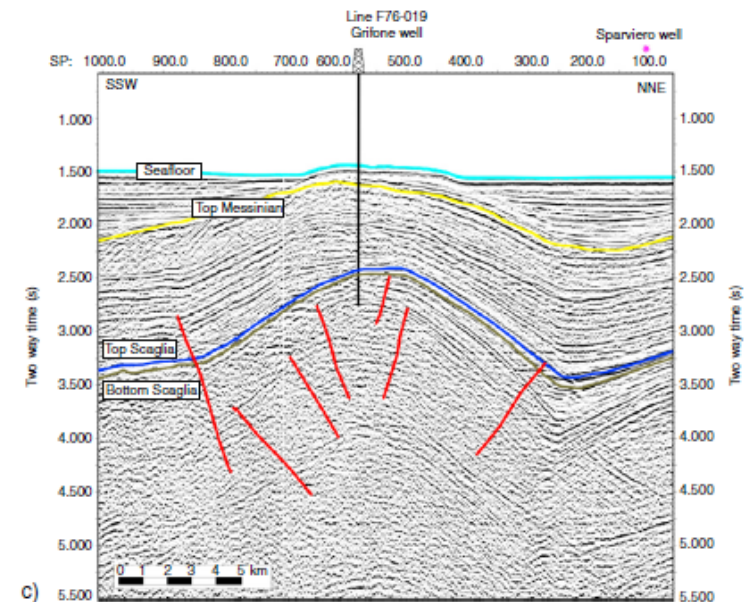
# GRIFONE STRUCTURE


 Grifone structure  
Grifone well


a)



b)

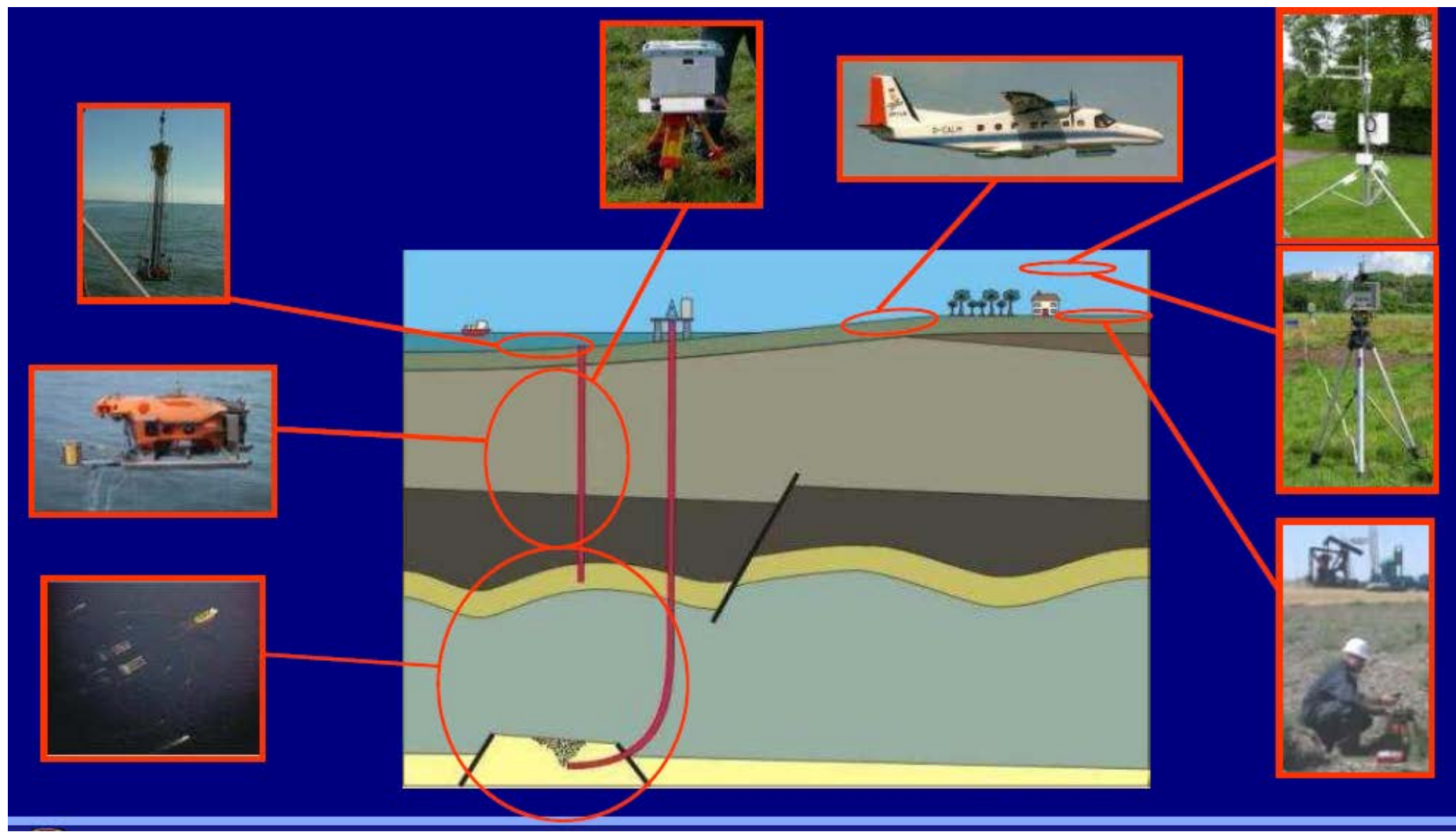


# ***Monitoring of the selected sites***

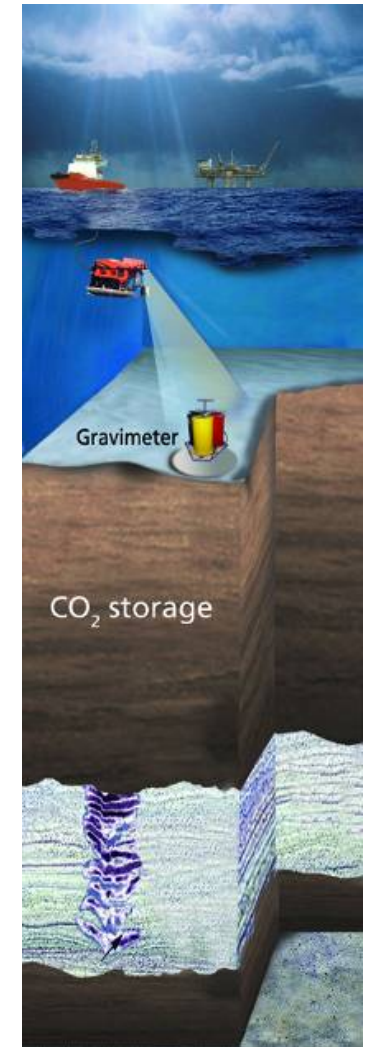
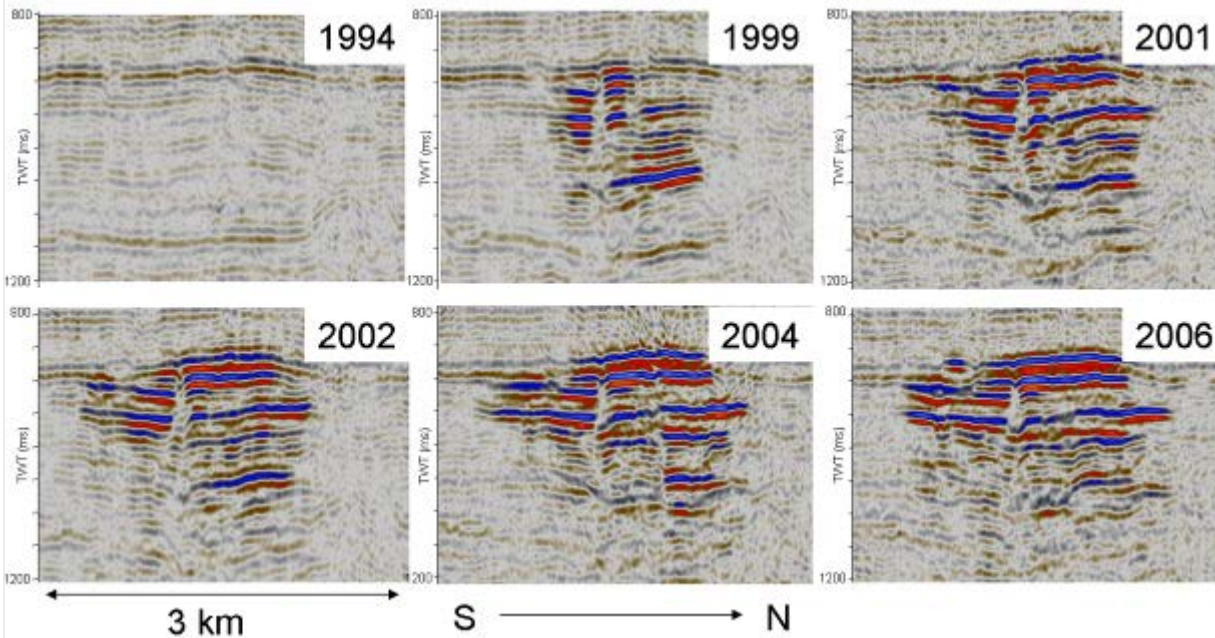
Monitoring is required in order to see whether:

- stored CO<sub>2</sub> behaves as expected
- migration or leakage occurs
- identified leakage damages environment or human health

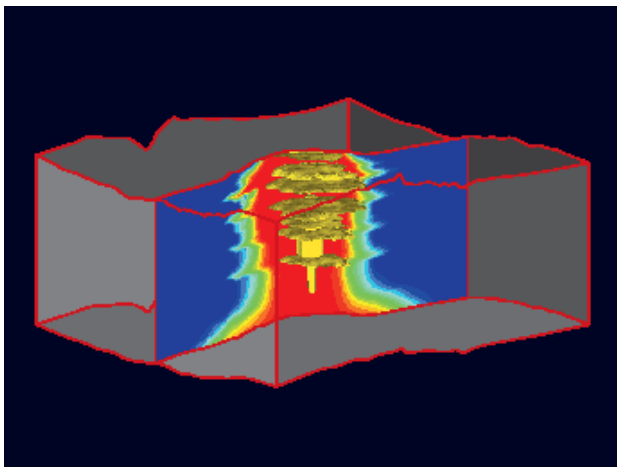
# Monitoring of storage site



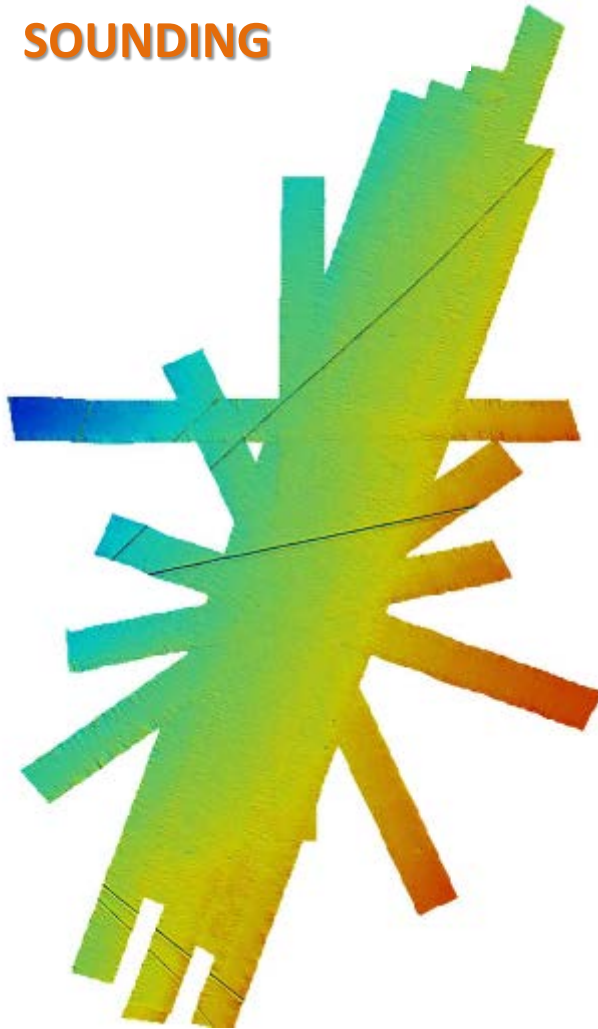
# IDENTIFICATION AND MONITORING OF CO<sub>2</sub> BEHAVIOUR AFTER INJECTION



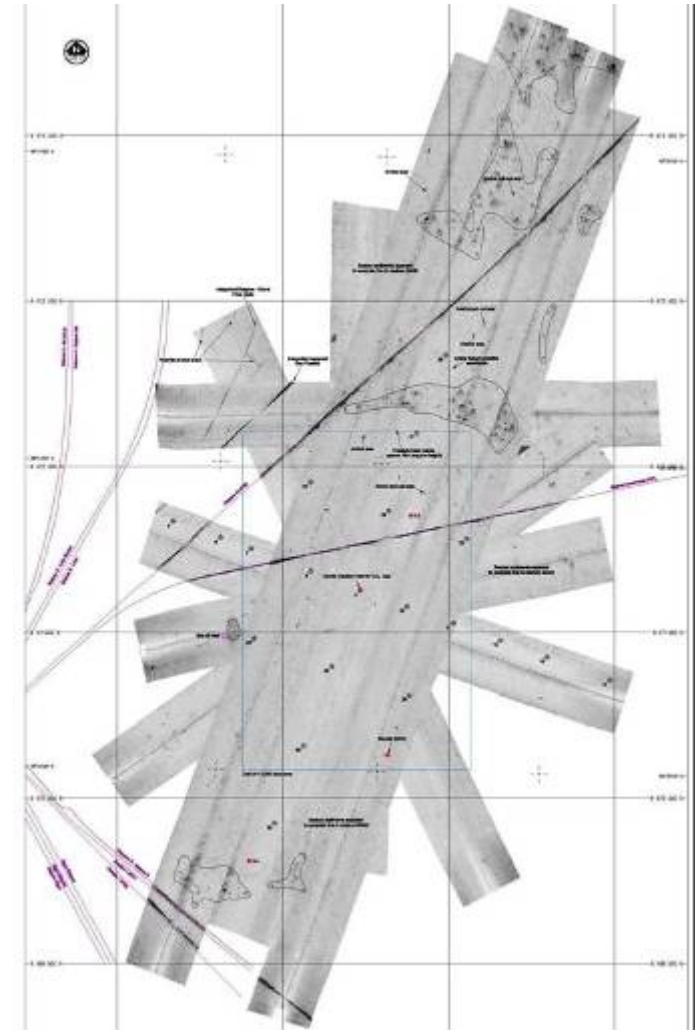
Courtesy Statoil/CO2STORE project



## SEAFLOOR DEPTH, FROM MULTIBEAM ECHO SOUNDING



## MOSAIC OF SIDE SCAN SONAR DATA



# ***Pioneer commercial CCS projects***



**Sleipner**, deep saline aquifer, Norway  
1 Mt CO<sub>2</sub>/y since 1996



**Weyburn**, oil reservoir-EOR Canada  
1 Mt CO<sub>2</sub>/y since 2000



**In-Salah**, gas reservoir, Algeria  
3.8 Mt CO<sub>2</sub> injected from 2004 to 2011



**Snohvit**, deep saline aquifer, Norway,  
3 Mt CO<sub>2</sub> injected since 2008